Calorimetry

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Heat exchange ; relation between mass, time and energy. $\,$

Heating a mass of water of 1 kg from 20°C to 30°C demands 1/2 less energy than required for a mass of 2 kg from 20°C to 30°C; and 1/2 in energy than needed to raise 1 kg from 20°C to 40°C.