Coordinated In-Situ Analysis of Meteoritic Nanodiamonds



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1.0 1 Energy (keV)

ND

amor-OM

Arizona State University

Α

В

Introduction

Meteoritic nanodiamonds (NDs)

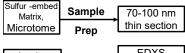
- ❖ carbon nanograins (2–5 nm) trapped in carbonaceous meteorites thought to be presolar
- carry stellar nucleosynthetic anomalies and encode processes predating the Solar System.

Traditional bulk analyses mask ND subpopulation variation, preventing origin(s) identification.

Objective

Differentiate presolar stardust vs. solar system ND formation in carbonaceous chondrites and returned asteroid samples

Methods



EDXS In-situ (composition) aberration Imaging **EELS** corrected (bonding) STEM/TEM

Isotopic NanoSIMS C,N Isotopes **Imaging**

Technical Approach

Locate ND clusters embedded in organic matter in carbonaceous chondrites and returned asteroid

Measure C and N isotopes of clusters to check for multiple formation histories

References

- S. J. Desch and N. Miret-Roig, "The Sun's birth environment: Context for meteoritics," Space Sci. Rev., vol. 220, no. 7, Art. no. 76,
- G. R. Huss and R. J. Lewis, "Presolar diamond, SiC, and graphite in meteorites," Geochim. Cosmochim. Acta, vol. 60, pp. 331–346, 1996.
 G. R. Huss, S. T. Husa, and S. A. Sandford, "IV processing of
- presolar organic ices," Astrophys. J., vol. 849, p. 75, 2017.

 4. A. Kouchi et al., "Novel routes for diamond formation in interstell ices and meteoritic parent bodies," Astrophys. J. Lett., vol. 626,
- A. Kumar et al., "Formation of nanodiamonds at near-ambient conditions via microplasma dissociation of ethanol vapour," Nat. Commun., vol. 4, Art. no. 2618, 2013.
- J. Llorca and I. Casanova, "Co + H₂ reaction over Fe-Ni metal at low pressures: Implications for nebular diamond formation," Geochim. Cosmochim. Acta, vol. 64, no. 15, pp. 2673–2684, 2000. C. L. R. Nittler and F. Ciesla, "Astrophysics with extraterrestrial materials," Annu. Rev. Astron. Astrophys., vol. 54, pp. 53–93, 2016.
 G. Pareras et al., "Single-atom catalysis in space – II. Ketene
- acetaldehyde-ethanol and methane synthesis via Fischer-Tropsch chain growth," Astron. Astrophys., vol. 687, p. A230, 2024.
- R. M. Stroud et al., "Electron microscopy observations of the diversity of Ryugu organic matter and its relationship to minerals at the micro-to nano-scale," Meteori. & Planeta. Sci. 59 (8), 2023-2043

Initial STEM-EDXS-EELS Results

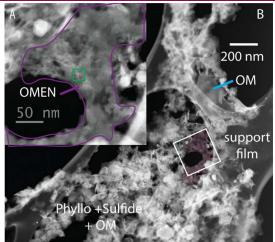


Figure 1 (left) STEM HAADF of microtomed Ryugu particle on lacey carbon grid.

(A) Organic matter (OM) with embedded nanodiamonds (OMEN) (magenta)

(B) Petrographic context OMEN in fine grained matrix \$\oint_{2.5}\$ (phyllosilicates/ sulfides/diamondfree OM). Elliptical void in OMEN suggests NDs form as an interstellar

icy particle rind.

Figure 2 (A) Spectrum image ROI (inset) and Extracted EDXS spectra from Amorphous C is N-richer than ND (blue).

Cu peak = system artifact; i/AI/Mg/Fe/O from heteroatoms in OM: residual from acid dissolution

(B) Extracted EELS C K-edge From ND (sp3) amor-OM (sp2)

Potential Formation Mechanisms & Distinct Isotopic Fingerprints

Cold Formation Mechanisms Radical Mediated Nucleation

Microplasma Alcohol Vapor Dissociation $n(C-based radicals) \cdot +0H + H_2$ Lvα Dominated (88%)

n (C-based radicals) \bullet association H-terminated [C_nH_m] Ethanol C-C Bonds are ideal Atmospheric pressure microplasma

UV/H-radical Exposure

UV Photolysis of Organic-Rich Ices

 $H_2O: CO: CH_4ice + h\nu (< 200nm) \xrightarrow[]{UV} [R] (CH_3, CH_2 \bullet)$

Organic ice mantles absorb FUV photons

 $(\lambda < 200 \text{ nm})$

H₂O-rich ice + minor CH₄, CH₃OH, CO, NH₃

(1-5%)

T≈100°C Residence time ≈1 ms

Hydrocarbon

Feedstock

Carbon Cluster

Formation

 (C_nH_{2n})

Flux sp² Hybridization Alkvl radicals $CH_2, CH_3 \bullet)$ Meteoritic Nanodiamond Formation Diamond

Nucleation

Outer Nebula cluster-mediated CO-H2 ice reactions FUV Flux UV (erg cm $^{-2}$ s $^{-1}$) G_0 UV (912-1700Å) Flux (912-1700Å)

0.05-5

Circumstellar Outflow

(CTTS at 1 AU)

Early Solar System

active accretion

FUV Flux

(912-1700Å)

~1.6 × 104

(erg cm⁻² s⁻¹) **G**₀

~107

Diffuse ISM & Cold Molecular Cloud Photon-processed ice grains FUV Flux IIV (erg cm⁻² s⁻¹) G₀ Flux (912-1700Å)

> Inner Nebula Hydrocarbon Catalysis on Fe-Ni grains

0.05-5

under solar nebula condition **FUV Flux** UV (erg cm⁻² s⁻¹) **G**₀ Flux (912-1700Å) ~107 ~1.6 × 104

~ Solar Viable Environment $\delta^{13}C$ $\delta^{15}N$ Solar ~ Solar C star outflow Diffuse ISM ✓ Cold Molecular Cloud Outer Nebula Inner Nebula

Hot Formation Mechanisms

HT/HP Processes

Vapor Condensation from Supernovae Ejecta Precursor Flash Heating/Collisional Shock flash heating/shock C (v)

Carbon Vapor Supersaturation C(v) supersaturation C(v) \rightarrow C(v)

r-process enriched Carbon vapor Extreme initial condition (IC) (P > 10 GPa) $(T \gg 1000 \text{ K})$

Cooling ejecta under low-pressure conditions (≤10⁻⁴ atm) forms NDs to later enter presolar cloud

Fischer-Tropsch Fe-Ni Catalysis M = Fe-Ni catalytic site (metal grain surface) Nebular conditions (~300-700K, 10⁻⁴ atm)

CO adsorption and activation

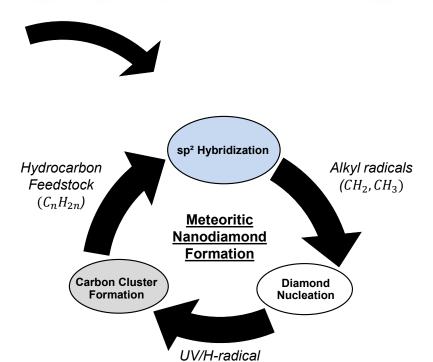
 $CO(g) + M \stackrel{k_1}{\rightarrow} M - C + M - O$ H₂ activation

 $H_2(ads) + 2M \stackrel{k_H}{\rightarrow} 2M - H$ CH₂ initiation

 $M-C+2M-H \stackrel{k_{\alpha}}{\rightarrow} M-CH_2+M$ Carbene insertion

 $M - CH_2 + CO(g) + M - H \stackrel{k_{ins}}{\rightarrow} M - C_2H_4 + M - O$ Chain growth

repeat CH2 initiation + Carbene insertion build $M - C_n H_{2n}$



Exposure



