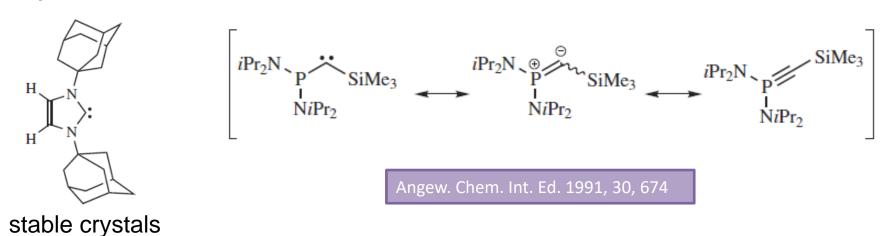
Carbenes

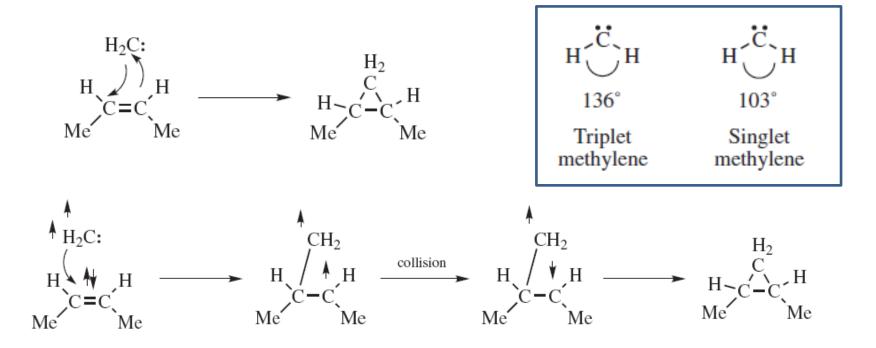
- Highly reactive species with practical lifetimes considerably under 1 s
- Carbenes have been isolated by entrapment in matrices at low temperatures (77 K or less)
- Carbenes that are stable at room temperature have been reported



(m.p. 240-241°C)

Carbenes

 The two nonbonded electrons of a carbene can be either paired (singlet) or unpaired (triplet)





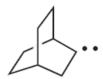
Examples of carbene

- Methylene & dichlorocarbene :CH₂ : CCl₂
- Heterocyclic carbenes

$$\bigcap_{N \in \mathbb{R}}^{R} \cdots \bigcap_{N \in \mathbb{R}}^{N} \cdots$$

Angew. Chem. Int. Ed. 1997, 36, 2607 Angew. Chem. Int. Ed. 2002, 41, 1290 J. Org. Chem. 2002, 67, 9288

• bicyclo[2.2.2]octylidene



Tetrahedron Lett. 2001, 42, 6979

alkylidene carbenes

Chem. Rev. 2004, 104, 3795

• Conformationally restricted cyclopropylcarbene



• β-Silylcarbenes

Tetrahedron Lett. 2001, 42, 8761

J. Org. Chem. 2002, 67, 112



Generation of carbenes

1. α -Elimination

$$CCl_3 - COO^{\ominus} \xrightarrow{\Delta} CCl_2 + CO_2 + Cl^{\ominus}$$

$$\downarrow hv + \cdots + \downarrow hv$$



Generation of carbenes

2. Disintegration of compounds containing certain types of double bonds

$$R_{2}C=Z \longrightarrow R_{3}C: + Z$$

$$CH_{2}=C=O \longrightarrow CH_{2} + {}^{\circ}C\equiv O^{\oplus}$$

$$CH_{2}=N=N^{\circ} \longrightarrow cH_{2} + N\equiv N$$

$$CH_{2}=N=N^{\circ} \longrightarrow cH_{2} + N\equiv N$$

$$R_2C \stackrel{N}{\underset{N}{\subset}} \longrightarrow R_2C: + N \equiv N$$

Diazirines are widely used for its photo-affinity in bioactive probes



♦ Insertion into C—H bonds



♦ Additions to C=C

Cyclopropanation The Skell Rule:

$$^{1}:CH_{2}$$
 + $\binom{R}{R}$ \longrightarrow $\binom{R}{R}$

Singlet carbenes add to olefins stereospecifically;

Triplet carbenes add non-stereospecifically

Skell and Woodworth JACS, 1956, 78, 4496.



Dimerization

$$R_2C$$
: + R_2CN_2 \longrightarrow $R_2C=CR_2$ + N_2

Rearrangement with migration of alkyl or hydrogen



◆ Triplet carbenes can abstract hydrogen or other atoms to give free radicals

Singlet carbenes can also give this reaction, in this case only halogen atoms are abstracted, not hydrogen.

J. Am. Chem. Soc. 1971, 93, 1527, 4935,

Acc. Chem. Res. 1977, 10, 85

