## 1 Variables

## 2 root

	var	symbol	documentation	type	units	tokens	eqs
5	$F_{N,A}$	F	incidence matrix of a directed graph	network			
170	$1_N$	ones_N	ones with length number of nodes	network			
171	$1_A$	ones_A	ones with length of number of arcs	network			
6	$\mid t \mid$	t	time	frame	s		
7	$t_o$	to	starting time	frame	s		4
8	$t_e$	te	end time	frame	s		5
1	#	value	numerical value	constant			
2	0	zero	numerical value zero	constant			1
3	1	one	numerical value one	constant			2
4	0.5	onehalf	numerical value one half	constant			3

# 3 physical

	var	symbol	documentation	type	units	tokens	eqs
161	$P_{N,A}$	P_N_A	project node to arc for material properties	projection			
9	$r_{xN}$	r_x	x-coordinate	frame	$\mid m \mid$		
10	$r_{yN}$	r_y	y-coordinate	frame	$\mid m \mid$		
23	$r_{zN}$	r_z	z-coordinate	frame	$\mid m \mid$		
11	$U_N$	U	foundation state – internal energy	state	$kg m^2 s^{-2}$		
12	$S_N$	S	foundation state – entropy	state	$kg  m^2  K^{-1}  s^{-2}$		
13	$V_N$	V	foundation state – volume	state	$m^3$		
18	$H_N$	Н	enthalpy	state	$kg m^2 s^{-2}$	0	9 122 123
19	$A_N$	A	Helmholtz energy	state	$kg m^2 s^{-2}$		10
20	$G_N$	G	Gibbs energy	state	$kg m^2 s^{-2}$		11
42	$n_{NS}$	n	species molar mass	state	mol		116
159	$C_N$	С	charge	state	A s		
26	$A^v$	Avogadro	Avogadro number	constant	$mol^{-1}$		
27	$Bo_N$	Boltzmann	Boltzmann constant	constant	$kg m^2 K^{-1} s^{-2}$		16
28	$R_N$	GasConstant	Gas constant	constant	$kg  m^2  mol^{-1}  K^{-1}  s^{-2}$	<sup>1</sup> []	17
15	$p_N$	p	thermodynamic pressure	effort	$kg  m^{-1}  s^{-2}$		6 115
16	$T_N$	Т	temperature	effort	K		7 113
45	$\mu_{NS}$	chem_pot	chemical potential	effort	$\begin{array}{ c c c c c c c c c c c c c c c c c c c$		32 114
160	$U^e{}_N$	Ue	voltage	effort	$kg  m^2  A^{-1} s^{-3}$		156 177 182
21	$v_{xN}$	v_x	velocity in x-direction	secondaryState	$ms^{-1}$		12

	var	symbol	documentation	type	units	tokens	eqs
22	$v_{yN}$	v_y	velocity in y direction	secondaryState	$ms^{-1}$		13
24	$v_{zN}$	v_z	velocity in z-direction	secondaryState	$ms^{-1}$		14
25	$v_N$	v	velocity vector	secondaryState	$ms^{-1}$		15

## 4 control

	var	symbol	documentation	type	units	tokens	eqs
139	$m_A$	mc	measurements	input			
158	$u_A$	u	data link	input			155
129	$I_{N,D}$	I_N_D	identity to shift from differential space integral space	network			
130	$I_{A,D}$	I_A_D	identity to shift from differential space to arc	network			
137	$x_N$	x	controller state	state			112
138	$x^{o}{}_{N}$	xo	controller initial condition	state			103
133	$C_{N,A}$	Cx	measurement matrix	constant			144
134	$D_{N,D}$	Dx	event matrix (dimension issue)	constant			145
135	$y^o{}_A$	setpoint	set point	constant			146
136	$D_A$	D_A	event diagonal matrix (no dimension problem)	constant			147
142	$A_{N,D}$	Ax	dynamic control matrix	constant	$s^{-1}$		148
143	$B_{A,D}$	Bx	input matrix	constant	$s^{-1}$		149
144	$\dot{x}_D$	dxdt	differential controller state	diffState	$s^{-1}$		106
140	$e_A$	e	control error	algebraic			104
155	$y_A$	У	controller output	algebraic			150 153

#### 5 reactions

	var	symbol	documentation	type	units	tokens	eqs
86	$N_{S,K}$	N	stoichiometric matrix	constant			
87	$E_{aNK}$	Ea	Arrhenius's activation energy	constant	$kg  m^2  mol^{-1}  s^{-2}$		64
88	$K^o{}_K$	Ко	Arrhenius's frequency factor	constant	$m^{-3}mols^{-1}$		
89	$K_{NK}$	K_NK	Arrhenius reaction constant	constant	$m^{-3}mols^{-1}$		65
114	$c_{KS}$	c_KS	molar concentrations in reactive system	constant	$m^{-3}  mol$		90
115	$c^{o}{}_{KS}$	co_KS	norming molar concentrations	constant	$m^{-3}  mol$		91
116	$\phi_{KS}$	phi_KS	probability of species to meet	constant			92

#### 6 material

	var	symbol	documentation	type	units	tokens	eqs
29	$\lambda_S$	Mm	species molecular masses	constant	$kg  mol^{-1}$		142
30	$C_{pN}$	Ср	total heat capacity at constant pressure	property	$kg m^2 K^{-1} s^{-2}$		18
31	$C_{vN}$	Cv	total heat capacity at constant volume	property	$kg m^2 K^{-1} s^{-2}$		19
34	$k_{xN}^q$	kq_x	thermal conductivity in x-direction	property	$kg K^{-1} s^{-3}$		22 131
35	$k_{yN}^q$	kq_y	thermal conductivity in y-direction	property	$kg K^{-1} s^{-3}$		23 132
36	$k_{zN}^q$	kq_z	thermal conductivity in z-direction	property	$kg K^{-1} s^{-3}$	0	24 133
37	$k^q{}_N$	kq	Carthesian thermal conductivity vector	property	$kg K^{-1} s^{-3}$		25
50	$k_{xN}^c$	kc_x	convective mass convectivity in x-direction	property	$m^{-1} s$		37 134
51	$k_{yN}^c$	kc_y	convective mass convectivity in y-direction	property	$m^{-1} s$		38 135
52	$k_{zN}^c$	kc_z	convective mass convectivity in z-direction	property	$m^{-1} s$	0	39 136
53	$k^c{}_N$	kc	Cartesian convective mass convectivity vector	property	$m^{-1} s$		40
54	$k_{xNS}^d$	kd_x	diffusional mass conductivity in x-direction	property	$kg^{-1} m^{-4} mol^2 s$	0	41 137
55	$k_{yNS}^d$	kd_y	diffusional mass conductivity in y-direction	property	$kg^{-1} m^{-4} mol^2 s$		42 138
56	$k_{zNS}^d$	kd_z	diffusional mass conductivity in z-direction	property	$kg^{-1} m^{-4} mol^2 s$		43
57	$k^d{}_{NS}$	kd	Cartesian dffusional mass conductivity vector	property	$kg^{-1} m^{-4} mol^2 s$		44
58	$h_{NS}$	h	partial molar enthalpies	property	$kg m^2 mol^{-1} s^{-2}$		45 139
71	$ ho_N$	density	mass density	property	$kg m^{-3}$		49 154

	var	symbol	documentation	type	units	tokens	eqs
148	$c_{pN}$	ср	specific heat capacity at constant pressure	property	$m^2 K^{-1} s^{-2}$		120 140
149	$c_{vN}$	cv	specific heat capacity at constant volume	property	$m^2 K^{-1} s^{-2}$		121 141
154	$\theta$	data	available data set	property			143
182	$kC_N$	kC	electrical conductiviy	property	$kg^{-1} m^{-2} A^2 s^3$		178
185	$k^{e,\xi}{}_N$	k_e_xi	very simple model for electrical conductivity due to species transort	property	$kg^{-1} m^{-2} A^2 s^3$	0	181

## 7 macroscopic

	var	symbol	documentation	type	units	tokens	eqs
65	$d_A$	d	direction of convective flow	transport			46
98	$\hat{V}_A$	fV	volumetric flow in x-direction	transport	$m^3 s^{-1}$		74
104	$\hat{n}^d{}_{AS}$	fnd_AS	diffusional mass transfer per arc	transport	$mol  s^{-1}$		80
105	$\hat{n}^d{}_{NS}$	fnd	net diffusional mass transfer	transport	$mol  s^{-1}$		81
106	$\hat{H}^d{}_A$	fHd_A	enthalpy flow due to mass diffusion per arc	transport	$kg m^2 s^{-3}$		82
107	$\hat{H}^d{}_N$	fHd	net enthalpy flow due to mass diffusion	transport	$kg m^2 s^{-3}$		83
109	$c_{AS}$	c_AS	moler concentration in convective arc	transport	$m^{-3}  mol$		85
110	$\hat{n}^c{}_{AS}$	fnc_AS	convective molar mass flow per arc	transport	$\mod s^{-1}$		86
111	$\hat{n}^c{}_{NS}$	fnc	net convective molar mass flow	transport	$\mod s^{-1}$		87
120	$\hat{H}^c{}_A$	fHc_A	enthalpy flow due to convective mass flow	transport	$kg m^2 s^{-3}$		96
121	$\hat{H}^c{}_N$	fHc	net enthalpy flow due to convective mass flow	transport	$kg m^2 s^{-3}$		97
122	$\hat{w}_A$	fw_A	example of work flow	transport	$kg m^2 s^{-3}$		98
123	$\hat{w}_N$	fw	net work flow	transport	$kg m^2 s^{-3}$		99
124	$\hat{q}_A$	fq_A_x	heat flow in x-direction	transport	$kg m^2 s^{-3}$		100
125	$\hat{q}_N$	fq	net heat flow	transport	$kg m^2 s^{-3}$		101
95	$A_{yzN}$	Ayz	cross sectional area in x-direction	geometry	$m^2$		71
96	$A_{xzN}$	Axz	cross sectional area in y direction	geometry	$m^2$		72
97	$A_{xyN}$	Axy	cross sectional area in z-direction	geometry	$m^2$		73
73	$F_{NS,AS}$	F_NS_AS	incidence matrix of directed graphs for for species NS x AS	network			51
59	$P_{NS,AS}$	P_NS_AS	node species to arc species projection	projection			
60	$P_{K,NK}$	P_K_NK	projection of conversion to node x conversion	projection			
61	$P_{S,NS}$	P_S_NS	projection species to conversion x species	projection			

	var	symbol	documentation	type	units	tokens	eqs
62	$P_{N,NK}$	P_N_NK	projection node to node x conversion	projection			
63	$P_{NK,KS}$	P_NK_KS	projection node x conversion to conversion x species	projection			
64	$P_{NS,KS}$	P_NS_KS	projection node x species to conversion x species	projection			
168	$P_{A,NS}$	P_NS_A	projection of node species to arc for conductivity	projection			
177	$P_{NN}S_{N,NS}$	P_N_NS	projection of node to node x species	projection			
127	$D_{N,A}$	D	difference operator	differenceOperator			
128	$D_{NS,AS}$	D_NS_AS	block difference operator	differenceOperator			
153	phi	phi	state for mass and energy	state			130
186	current	current	curent is really not a state – thus no indexing	state	A		
145	$T_{refN}$	T_ref	refernce temperature	constant	K		117
150	$n^o{}_{NS}$	no	initial condition for species mass in nodes	constant	mol		124
151	$H^o{}_N$	Но	initial condition for enthalpy in nodes	constant	$kg m^2 s^{-2}$		125
164	$1_{NS}$	ones_NS	block vector of ones	constant			
181	$U^{e,circuit}_{N}$	Ue_circuit	closed circuit potential (will be set to zero)	effort	$kg m^2 A^{-1} s^{-3}$		175 176
69	$m_N$	m	mass	secondaryState	kg		47
108	$c_{NS}$	С	molar concentration	secondaryState	$m^{-3}  mol$		84 127
152	$\psi$	intensities	collected intensities	secondaryState			126
165	$n^t{}_N$	cnorm	moles per node	secondaryState	$m^{-3}  mol$		161
166	$\xi_{NS}$	x_frac	mole fractions	secondaryState			162
77	$T_{NK}$	T_NK	temperature in reactive systems	conversion	K		55
93	$N_{NS,NK}$	N_NS_NK	extended stoichiometric matrix	conversion			69

	var	symbol	documentation	type	units	tokens	eqs
117	$\xi_{NK}$	xi	extend of reaction per volume	conversion	$m^{-3}  mol  s^{-1}$		93
118	$\tilde{n}_{NS}$	pn	production term	conversion	$mol  s^{-1}$		94
119	$\dot{n}_{NS}$	dndt	differential molar mass balance	diffState	$mol  s^{-1}$		95 129
126	$\dot{H}_N$	dHdt	differential enthalpy balance	diffState	$kgm^2s^{-3}$		102 128
173	$i_N$	i	internal current	internalTransport	A		168
183	$i_{NS}$	i_NS	current due to species transport	internalTransport	A		179

## 8 solid

	var	symbol	documentation	type	units	tokens	eqs
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## 9 fluid

	var	symbol	documentation	type	units	tokens	eqs
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# 10 liquid

	var	symbol	documentation	type	units	tokens	eqs
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# 11 gas

	var	symbol	documentation	type	units	tokens	eqs
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## 12 control-control

	var	symbol	documentation	type	units	tokens	eqs
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# 13 gas-liquid

	var	symbol	documentation	type	units	tokens	eqs
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## 14 gas-gas

	var	symbol	documentation	type	units	tokens	eqs
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# 15 liquid-liquid

	var	symbol	documentation	type	units	tokens	eqs
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# 16 gas-solid

var symbol documentation	type	units	tokens	eqs
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## 17 solid-solid

	var	symbol	documentation	type	units	tokens	eqs
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# 18 liquid-solid

	var	symbol	documentation	type	units	tokens	eqs
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#### 19 material-material

	var	symbol	documentation	type	units	tokens	eqs
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#### 20 reactions—reactions

	var	symbol	documentation	type	units	tokens	eqs
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## 21 Equations

## 22 Generic

no	equation	documentation	layer
1	0 := Instantiate(#, #)	numerical value zero	root
2	1 := Instantiate(#, #)	numerical value one	root
3	0.5 := Instantiate(#, #)	numerical value one half	root
4	$t_o := \text{Instantiate}(t, \#)$	starting time	root
5	$t_e := \text{Instantiate}(t, \#)$	end time	root
6	$p_N := \left( -\frac{\partial U_N}{\partial V_N} \right)$	thermodynamic pressure	physical
7	$T_N := \frac{\partial U_N}{\partial S_N}$	temperature	physical
9	$H_N := U_N - p_N \cdot V_N$	enthalpy	physical
10	$A_N := U_N - T_N \cdot S_N$	Helmholtz energy	physical
11	$G_N := U_N + p_N \cdot V_N - T_N \cdot S_N$	Gibbs energy	physical
12	$v_{xN} := \frac{\partial r_{xN}}{\partial t}$	velocity in x-direction	physical
13	$v_{yN} := \frac{\partial  r_{yN}}{\partial  t}$	velocity in y direction	physical
14	$v_{zN} := \frac{\partial  r_{zN}}{\partial  t}$	velocity in z-direction	physical
15	$v_N := \mathrm{Stack}\left(v_{xN}, v_{yN}, v_{zN}\right)$	velocity vector	physical
16	$Bo_N := \text{Instantiate}(S_N, \#)$	Boltzmann constant	physical
17	$R_N := A^v \cdot Bo_N$	Gas constant	physical

no	equation	documentation	layer
18	$C_{pN} := \frac{\partial H_N}{\partial T_N}$	total heat capacity	material
19	$C_{vN} := rac{\partial U_N}{\partial T_N}$	total heat capacity at constant volume	material
22	$k_{xN}^q := (V_N)^{-1} \cdot \frac{\partial U_N}{\partial T_N} \cdot v_{xN}$	thermal conductivity in x-direction	material
23	$k_{yN}^q := (V_N)^{-1} \cdot \frac{\partial U_N}{\partial T_N} \cdot v_{yN}$	thermal conductivity in y-direction	material
24	$k_{zN}^q := (V_N)^{-1} \cdot \frac{\partial U_N}{\partial T_N} \cdot v_{zN}$	thermal conductivity in z-direction	material
25	$k^{q}_{N} := \operatorname{Stack}\left(k_{xN}^{q}, k_{yN}^{q}, k_{zN}^{q}\right)$	Carthesian thermal conductivity vector	material
32	$\mu_{NS} := \frac{\partial U_N}{\partial n_{NS}}$	chemical potential	physical
37	$k_{xN}^c := \left(\lambda_S \overset{S \in NS}{\star} (\mu_{NS})^{-1}\right) \cdot (V_N)^{-1} \cdot \frac{\partial U_N}{\partial p_N} \cdot v_{xN}$	convective mass convectivity in x-direction	material
38	$k_{yN}^c := \left(\lambda_S \overset{S \in NS}{\star} (\mu_{NS})^{-1}\right) \cdot (V_N)^{-1} \cdot \frac{\partial U_N}{\partial p_N} \cdot v_{yN}$	convective mass convectivity in y-direction	material
39	$k_{zN}^c := \left(\lambda_S \overset{S \in NS}{\star} (\mu_{NS})^{-1}\right) \cdot (V_N)^{-1} \cdot \frac{\partial U_N}{\partial p_N} \cdot v_{zN}$	convective mass convectivity in z-direction	material
40	$k^{c}_{N} := \operatorname{Stack}\left(k^{c}_{xN}, k^{c}_{yN}, k^{c}_{zN}\right)$	Cartesian convective mass convectivity vector	material
41	$k_{xNS}^d := (\mu_{NS})^{-1} \cdot \left( v_{xN} \odot \left( (V_N)^{-1} \odot \frac{\partial U_N}{\partial \mu_{NS}} \right) \right)$	diffusional mass conductivity in x-direction	material
42	$k_{yNS}^d := (\mu_{NS})^{-1} \cdot \left( v_{yN} \odot \left( (V_N)^{-1} \odot \frac{\partial U_N}{\partial \mu_{NS}} \right) \right)$	diffusional mass conductivity in y-direction	material
43	$k_{zNS}^d := (\mu_{NS})^{-1} \cdot \left( v_{zN} \odot \left( (V_N)^{-1} \odot \frac{\partial U_N}{\partial \mu_{NS}} \right) \right)$	diffusional mass conductivity in z-direction	material

no	equation	documentation	layer
44	$k^{d}_{NS} := \operatorname{Stack}\left(k^{d}_{xNS}, k^{d}_{yNS}, k^{d}_{zNS}\right)$	Cartesian dffusional mass conductivity vector	material
45	$h_{NS} := H_N \odot \left( n_{NS} \right)^{-1}$	partial molar enthalpies	material
46	$d_A := \operatorname{sign}\left(F_{N,A} \stackrel{N}{\star} p_N\right)$	direction of convective ow	macroscopic
47	$m_N := \lambda_S \overset{S \in NS}{\star} n_{NS}$	mass	macroscopic
49	$\rho_N := m_N \cdot (V_N)^{-1}$	mass density	material
51	$F_{NS,AS} := F_{N,A} \odot P_{NS,AS}$	incidence matrix of directed graphs for for species NS x AS	macroscopic
55	$T_{NK} := P_{N,NK} \stackrel{N}{\star} T_N$	temperature in reactive systems	macroscopic
64	$E_{aNK} := \text{Instantiate}(P_{N,NK} \stackrel{N}{\star} R_N . T_{NK}, \#)$	Arrhenius's activation energy	reactions
65	$K_{NK} := K^o{}_K \odot exp((-E_{aNK}) \cdot \left(R_N * P_{N,NK} \cdot T_{NK}\right)^{-1})$	Arrhenius reaction constant	reactions
69	$N_{NS,NK} := P_{S,NS} \stackrel{S}{\star} \left( \left( P_{K,NK} . T_{NK} . \left( T_{NK} \right)^{-1} \right) \stackrel{K}{\star} N_{S,K} \right)$	extended stoichiometric matrix	macroscopic
71	$A_{yzN} := r_{yN} \cdot r_{zN}$	cross sectional area in x-direction	macroscopic
72	$A_{xzN} := r_{xN} \cdot r_{zN}$	cross sectional area in y direction	macroscopic
73	$A_{xyN} := r_{xN} \cdot r_{yN}$	cross sectional area in z-direction	macroscopic
74	$\hat{V}_A := (\rho_N)^{-1} \cdot k_{xN}^c \cdot A_{yzN} \cdot D_{N,A} *^{N} p_N$	volumetric flow in x-direction	macroscopic
80	$\hat{n}^d{}_{AS} := A_{yzN} \odot \left( -k^d_{xNS} \right) . D_{NS,AS} \overset{NS}{\star} \mu_{NS}$	diffusional mass transfer per arc	macroscopic
81	$\hat{n}^d{}_{NS} := F_{NS,AS} \stackrel{AS}{\star} \hat{n}^d{}_{AS}$	net diffusional mass transfer	macroscopic

no	equation	documentation	layer
82	$\hat{H}^d{}_A := \left(F_{NS,AS} \overset{NS}{\star} h_{NS}\right) \overset{S \in AS}{\star} \hat{n}^d{}_{AS}$	enthalpy flow due to mass diffusion per arc	macroscopic
83	$\hat{H}^d{}_N := F_{N,A} \stackrel{A}{\star} \hat{H}^d{}_A$	net enthalpy flow due to mass diffusion	macroscopic
84	$c_{NS} := (V_N)^{-1} \odot n_{NS}$	molar concentration	macroscopic
85	$c_{AS} := (0.5 \cdot (F_{NS,AS} - d_A \odot  F_{NS,AS} )) \overset{NS}{\star} c_{NS}$	moler concentration in convective arc	macroscopic
86	$\hat{n}^c{}_{AS} := \hat{V}_A \odot c_{AS}$	convective molar mass flow per arc	macroscopic
87	$\hat{n}^c{}_{NS} := F_{NS,AS} \stackrel{AS}{\star} \hat{n}^c{}_{AS}$	net convective molar mass flow	macroscopic
90	$c_{KS} := c_{NS} \overset{NS}{\star} P_{NS,KS}$	molar concentrations in reactive system	reactions
91	$c^o{}_{KS} := \text{Instantiate}(c_{KS}, \#)$	norming molar concentrations	reactions
92	$\phi_{KS} := \prod \left( c_{KS} \cdot \left( c^o_{KS} \right)^{-1} \right)$	probability of species to meet	reactions
93	$\xi_{NK} := K_{NK} \cdot P_{NK,KS} \overset{KS}{\star} \phi_{KS}$	extend of reaction per volume	macroscopic
94	$\tilde{n}_{NS} := V_N \odot \left( N_{NS,NK} \stackrel{NK}{\star} \xi_{NK} \right)$	production term	macroscopic
95	$\dot{n}_{NS} := \hat{n}^c{}_{NS} + \hat{n}^d{}_{NS} + \tilde{n}_{NS}$	differential molar mass balance	macroscopic
96	$\hat{H}^{c}{}_{A} := \left(F_{NS,AS} \overset{NS}{\star} h_{NS}\right) \overset{S \in AS}{\star} \hat{n}^{c}{}_{AS}$	enthalpy flow due to convective mass flow	macroscopic
97	$\hat{H}^c{}_N := F_{N,A} \stackrel{A}{\star} \hat{H}^c{}_A$	net enthalpy flow due to convective mass flow	macroscopic
98	$\hat{w}_A := \text{Instantiate}(\hat{H}^c{}_A, \#)$	example of work flow	macroscopic
99	$\hat{w}_N := F_{N,A} \overset{A}{\star} \hat{w}_A$	net work flow	macroscopic

no	equation	documentation	layer
100	$\hat{q}_A := A_{yzN} \cdot k_{xN}^q \cdot D_{N,A} \stackrel{N}{\star} T_N$	heat flow in x-direction	macroscopic
101	$\hat{q}_N := F_{N,A} \overset{A}{\star} \hat{q}_A$	net heat flow	macroscopic
102	$\dot{H}_N := \hat{H}^c{}_N + \hat{H}^d{}_N + \hat{q}_N + \hat{w}_N$	differential enthalpy balance	macroscopic
103	$x^o_N := \text{Instantiate}(x_N, \#)$	controller initial condition	control
104	$e_A := m_A - y^o{}_A$	control error	control
106	$\dot{x}_D := A_{N,D} \stackrel{N}{\star} x_N + B_{A,D} \stackrel{A}{\star} e_A$	differential controller state	control
112	$x_N := \int_{t_o}^{t_e} I_{N,D} \stackrel{D}{\star} \dot{x}_D \ dt$	controller state	control
113	$T_N := \text{Instantiate}(T_N, \#)$	temperature	physical
114	$\mu_{NS} := \text{Instantiate}(\mu_{NS}, \#)$	chemical potential	physical
115	$p_N := \text{Instantiate}(p_N, \#)$	thermodynamic pressure	physical
116	$n_{NS} := \int_{t_o}^{t_e} \dot{n}_{NS} \ dt + n^o{}_{NS}$	species molar mass	macroscopic
117	$T_{refN} := \text{Instantiate}(T_N, \#)$	refernce temperature	macroscopic

no	equation	documentation	layer
120	$c_{pN} := C_{pN} \cdot (m_N)^{-1}$	specific heat capacity at constant pressure	material
121	$c_{vN} := C_{vN} \cdot \left(m_N\right)^{-1}$	specific heat capacity at constant volume	material
122	$H_N := m_N \cdot \int_{T_{ref}N}^{T_N} c_{pN} \ dT_N$	enthalpy	macroscopic
123	$H_N := \int_{t_o}^{t_e} \dot{H}_N \ dt + H^o{}_N$	enthalpy	macroscopic
124	$n^o{}_{NS} := \text{Instantiate}(n_{NS}, \#)$	initial condition for species mass in nodes	macroscopic
125	$H^o{}_N := \text{Instantiate}(H_N, \#)$	initial condition for enthalpy in nodes	macroscopic
126	$\psi := \operatorname{MixedStack}\left(p_N, T_N, \mu_{NS}, c_{NS}\right)$	collected intensities	macroscopic
127	$c_{NS} := \text{Instantiate}(c_{NS}, \#)$	molar concentration	macroscopic
128	$\dot{H}_N := \text{Instantiate}(\dot{H}_N, 0)$	differential enthalpy balance	macroscopic
129	$\dot{n}_{NS} := \operatorname{Instantiate}(\dot{n}_{NS}, 0)$	differential molar mass balance	macroscopic
130	$phi := MixedStack\left(n_{NS}, H_N\right)$	state for mass and energy	macroscopic
131	$k_{xN}^q := \text{Instantiate}(k_{xN}^q, \#)$	thermal conductivity in x-direction	material

no	equation	documentation	layer
132	$k_{yN}^q := \text{Instantiate}(k_{yN}^q, \#)$	thermal conductivity in y-direction	material
133	$k_{zN}^q := \text{Instantiate}(k_{zN}^q, \#)$	thermal conductivity in z-direction	material
134	$k_{xN}^c := \text{Instantiate}(k_{xN}^c, \#)$	convective mass convectivity in x-direction	material
135	$k_{yN}^c := \text{Instantiate}(k_{yN}^c, \#)$	convective mass convectivity in y-direction	material
136	$k_{zN}^c := \text{Instantiate}(k_{zN}^c \cdot \#, -)$	convective mass convectivity in z-direction	material
137	$k_{xNS}^d := \text{Instantiate}(k_{xNS}^d, \#)$	diffusional mass conductivity in x-direction	material
138	$k_{yNS}^d := \text{Instantiate}(k_{yNS}^d, \#)$	diffusional mass conductivity in y-direction	material
139	$h_{NS} := \text{Instantiate}(h_{NS}, \#)$	partial molar enthalpies	material
140	$c_{pN} := \text{Instantiate}(c_{pN}, \#)$	specific heat capacity at constant pressure	material
141	$c_{vN} := \text{Instantiate}(c_{vN}, \#)$	specific heat capacity at constant volume	material
142	$\lambda_S := \operatorname{Instantiate}(\lambda_S, \#)$	species molecular masses	material
143	$\theta := \text{MixedStack}\left(k^q{}_N, k^c{}_N, k^d{}_{NS}, h_{NS}, c_{pN}, c_{vN}, \lambda_S, \rho_N\right)$	available data set	material

no	equation	documentation	layer
144	$C_{N,A} := \text{Instantiate}(C_{N,A}, \#)$	measurement matrix	control
145	$D_{N,D} := \text{Instantiate}(D_{N,D}, \#)$	event matrix (dimension issue)	control
146	$y^o{}_A := \operatorname{Instantiate}(y^o{}_A, \#)$	set point	control
147	$D_A := \text{Instantiate}(D_A, \#)$	event diagonal matrix (no dimension problem)	control
148	$A_{N,D} := \text{Instantiate}(A_{N,D}, \#)$	dynamic control matrix	control
149	$B_{A,D} := \text{Instantiate}(B_{A,D}, \#)$	input matrix	control
150	$y_A := C_{N,A} \overset{N}{\star} x_N + D_A \cdot e_A$	controller output	control
153	$y_A := D_A . e_A$	controller output	control
154	$ \rho_N := \operatorname{Instantiate}(\rho_N, \#) $	mass density	material
155	$u_A := 1 \cdot y_A$	data link	control
156	$U^e{}_N := \left(C_N\right)^{-1} . U_N$	voltage	physical
161	$n^t{}_N := 1_{NS} \overset{S \in NS}{\star} c_{NS}$	moles per node	macroscopic

no	equation	documentation	layer
162	$\xi_{NS} := \left(n^t{}_N\right)^{-1} \odot c_{NS}$	mole fractions	macroscopic
168	$i_N := rac{dC_N}{dt}$	internal current	macroscopic
175	$U^{e,circuit}{}_{N} := 1_{N} \cdot U^{e}{}_{N}$	closed circuit potential (will be set to zero)	macroscopic
176	$U^{e,circuit}_{N} := \text{Instantiate}(U^{e,circuit}_{N}, 0)$	closed circuit potential (will be set to zero)	macroscopic
177	$U^e{}_N := \text{Instantiate}(U^e{}_N, \#)$	voltage	macroscopic
178	$kC_N := i_N \cdot (U^e{}_N)^{-1}$	electrical conductiviy	material
179	$i_{NS} := P_{NN} S_{N,NS} \overset{N}{\star} i_{N}$	current due to species transport	macroscopic
181	$k^{e,\xi}_N := (U^e_N)^{-1} \cdot \left(i_{NS} \overset{S \in NS}{\star} ln(\xi_{NS})\right)$	very simple model for electrical conductivity due to species transort	material
182	$U^e{}_N := \left(k^{e,\xi}{}_N\right)^{-1}  .  i_N$	voltage	macroscopic