

# 1 Variables

## 2 root

	var	symbol	documentation	type	units	eqs
V <sub>8</sub>	$F_{N,A}$	F_N_A	fundamental incidence matrix	network		
V <sub>5</sub>	$t$	t	time	frame	s	
V <sub>6</sub>	$t^o$	to	starting time	frame	s	4
V <sub>7</sub>	$t^e$	te	end time	frame	s	5
V <sub>1</sub>	#	value	numerical value	constant		
V <sub>2</sub>	1	one	numerical value one	constant		1
V <sub>3</sub>	0	zero	numerical value zero	constant		2
V <sub>4</sub>	0.5	onehalf	numerical value one half	constant		3

### 3 physical

	var	symbol	documentation	type	units	eqs
V <sub>9</sub>	$P_{N,A}$	P_N_A	projection from node to arc for arc properties	projection		
V <sub>32</sub>	$P_{NS,AS}$	P_NS_AS	projection node species to arc species	projection		
V <sub>33</sub>	$P_{K,NK}$	P_K_NK	projection of conversion to node conversion	projection		
V <sub>34</sub>	$P_{S,NS}$	P_S_NS	projection species to node species	projection		
V <sub>35</sub>	$P_{N,NK}$	P_N_NK	projection node to node conversion	projection		
V <sub>36</sub>	$P_{NS,KS}$	P_NS_KS	projection node species to conversion species	projection		
V <sub>37</sub>	$P_{A,NS}$	P_A_NS	projection arc to node species for conductivity	projection		
V <sub>65</sub>	$P_{NK,KS}$	P_NK_KS	projection node conversion to conversion species	projection		
V <sub>162</sub>	$P_{N,NS}$	P_N_NS	projection of nodes onto the node species	projection		
V <sub>10</sub>	$r_{xN}$	r_x	x-coordinate	frame	$m$	
V <sub>11</sub>	$r_{yN}$	r_y	y-coordinate	frame	$m$	
V <sub>12</sub>	$r_{zN}$	r_z	z coordinate	frame	$m$	
V <sub>13</sub>	$U_N$	U	fundamental state – internal energy	state	$kg\,m^2\,s^{-2}$	
V <sub>14</sub>	$S_N$	S	fundamental state – entropy	state	$kg\,m^2\,K^{-1}\,s^{-2}$	
V <sub>15</sub>	$V_N$	V	fundamental state – volume	state	$m^3$	
V <sub>16</sub>	$n_{NS}$	n	fundamental state – molar mass	state	$mol$	86
V <sub>20</sub>	$H_N$	H	enthalpy	state	$kg\,m^2\,s^{-2}$	9 87
V <sub>21</sub>	$A_N$	A	Helmholtz energy	state	$kg\,m^2\,s^{-2}$	10
V <sub>22</sub>	$G_N$	G	Gibbs free energy	state	$kg\,m^2\,s^{-2}$	11
V <sub>23</sub>	$C_N$	charge	fundamental state – charge	state	$A\,s$	
V <sub>24</sub>	$A^v$	Avogadro	Avogadro number	constant	$mol^{-1}$	
V <sub>165</sub>	$boz_N$	boz	Boltzmann constant	constant	$kg\,m^2\,K^{-1}\,s^{-2}$	132
V <sub>166</sub>	$R_N$	R	gas constant	constant	$kg\,m^2\,mol^{-1}\,K^{-1}\,s^{-2}$	133

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	var	symbol	documentation	type	units	eqs
V <sub>17</sub>	$p_N$	<b>p</b>	thermodynamic pressure	effort	$kg\,m^{-1}\,s^{-2}$	<b>6</b>
V <sub>18</sub>	$T_N$	<b>T</b>	temperature	effort	$K$	<b>7</b>
V <sub>19</sub>	$\mu_{NS}$	<b>chemPot</b>	chemical potential	effort	$kg\,m^2\,mol^{-1}\,s^{-2}$	<b>8 136</b>
V <sub>27</sub>	$Ue_N$	<b>Ue</b>	electrical potential – voltage	effort	$kg\,m^2\,A^{-1}\,s^{-3}$	<b>14 95</b>
V <sub>28</sub>	$v_{xN}$	<b>v_x</b>	velocity in x-direction	secondaryState	$ms^{-1}$	<b>15</b>
V <sub>29</sub>	$v_{yN}$	<b>v_y</b>	velocity in y-direction	secondaryState	$ms^{-1}$	<b>16</b>
V <sub>30</sub>	$v_{zN}$	<b>v_z</b>	velocity in z-direction	secondaryState	$ms^{-1}$	<b>17</b>

## 4 control

	var	symbol	documentation	type	units	eqs
V <sub>136</sub>	$x_N$	<b>x</b>	state	state		<a href="#">111</a>
V <sub>137</sub>	$x_{oN}$	<b>xo</b>	initial state	state		<a href="#">109</a>
V <sub>129</sub>	$A_{N,D}$	<b>A</b>	dynamic matrix	constant	$s^{-1}$	
V <sub>130</sub>	$B_{A,D}$	<b>B</b>	input matrix	constant	$s^{-1}$	
V <sub>131</sub>	$C_{N,A}$	<b>C</b>	measurement matrix	constant		
V <sub>132</sub>	$D_A$	<b>D</b>	diagonal event matrix (no dimensional problems)	constant		
V <sub>133</sub>	$y^o_A$	<b>setPoint</b>	set point	constant		<a href="#">119</a>
V <sub>134</sub>	$m_A$	<b>meas</b>	measurements	constant		
V <sub>135</sub>	$e_A$	<b>e</b>	control error	constant		<a href="#">108</a>
V <sub>139</sub>	$1_{N,D}$	<b>I_N_D</b>	space transformation D to N	constant		
V <sub>138</sub>	$\dot{x}_D$	<b>dxdt</b>	differential state (ABCD) model	diffState	$s^{-1}$	<a href="#">110</a>
V <sub>141</sub>	$\tilde{I}_N$	<b>Imeasured</b>	measured current	algebraic	$A$	<a href="#">113</a>
V <sub>143</sub>	$\tilde{U}^e_N$	<b>UeMeasured</b>	measured electrical potential	algebraic	$kg\,m^2\,A^{-1}s^{-3}$	<a href="#">115</a>
V <sub>144</sub>	$\tilde{\xi}$	<b>addMeasured</b>	measured additive fraction	algebraic		<a href="#">116</a>
V <sub>145</sub>	$R_N$	<b>RComputed</b>	measured resistance	algebraic	$kg\,m^2\,A^{-2}s^{-3}$	<a href="#">117</a>
V <sub>146</sub>	$S$	<b>store</b>	quantities to be stored	algebraic		<a href="#">118</a>
V <sub>154</sub>	$y_A$	<b>y</b>	output equation	algebraic		<a href="#">126</a>
V <sub>171</sub>	$s$	<b>switch</b>	switches at to	algebraic		<a href="#">138</a>

## 5 reactions

	var	symbol	documentation	type	units	eqs
V <sub>147</sub>	$P_{NK}$	P_NK	reactions per node	projection		
V <sub>38</sub>	$K^o_K$	Ko	Arrhenius frequency factor	constant	$m^{-3} mol s^{-1}$	
V <sub>62</sub>	$E^a_{NK}$	Ea	Arrhenius activation energy	constant	$kg m^2 mol^{-1} s^{-2}$	41
V <sub>63</sub>	$K_{NK}$	K_NK	Arrhenius reaction 'constant'	constant	$m^{-3} mol s^{-1}$	42
V <sub>155</sub>	$B$	Boltzmann	Boltzmann constant	constant	$kg m^2 K^{-1} s^{-2}$	
V <sub>157</sub>	$R$	GasConstant	gas constant	constant	$kg m^2 mol^{-1} K^{-1} s^{-2}$	127
V <sub>158</sub>	$N_{K,KS}$	N_K_KS	stoichiometry	constant		
V <sub>159</sub>	$N_{NK,KS}$	N_NK_KS	extended stoichiometric matrix	constant		128
V <sub>60</sub>	$T_{NK}$	T_NK	temperature of the reactive system	effort	$K$	39
V <sub>151</sub>	$c_{NK,KS}$	c	concentration matrix reaction per node and species per reaction	secondaryState	$m^{-3} mol$	123
V <sub>152</sub>	$c^o_{NK,KS}$	co	norming concentration	secondaryState	$m^{-3} mol$	124
V <sub>153</sub>	$x_{NK,KS}$	x	matrix of normed, dimensionless mole fractions	secondaryState		125
V <sub>160</sub>	$\phi_{NK}$	phi	probability function for reactions	secondaryState		129
V <sub>163</sub>	$\tilde{n}_{NS}$	nProd	the species production term	secondaryState	$mol s^{-1}$	130

## 6 material

	var	symbol	documentation	type	units	eqs
V <sub>40</sub>	$\lambda_S$	Mm	species molecular mass	constant	$kg\ mol^{-1}$	
V <sub>112</sub>	$\xi$	additive	fraction of additives	constant		88
V <sub>42</sub>	$C_{pN}$	Cp	total heat capacity at constant pressure	property	$kg\ m^2\ K^{-1}\ s^{-2}$	21
V <sub>43</sub>	$C_{VN}$	Cv	total heat capacity at constant volume	property	$kg\ m^2\ K^{-1}\ s^{-2}$	22
V <sub>44</sub>	$k_{xN}^q$	kq_x	thermal conductivity in x-direction	property	$kg\ K^{-1}\ s^{-3}$	23
V <sub>45</sub>	$k_{yN}^q$	kq_y	thermal conductivity in y-direction	property	$kg\ K^{-1}\ s^{-3}$	24
V <sub>46</sub>	$k_{zN}^q$	kq_z	thermal conductivity in z-direction'	property	$kg\ K^{-1}\ s^{-3}$	25
V <sub>47</sub>	$k_N^q$	kq	thermal conductivity	property	$kg\ K^{-1}\ s^{-3}$	26
V <sub>48</sub>	$k_{xN}^c$	kc_x	convective mass conductivity in x-direction	property	$m^{-1}\ s$	27
V <sub>49</sub>	$k_{yN}^c$	kc_y	convective mass conductivity in y-direction	property	$m^{-1}\ s$	28
V <sub>50</sub>	$k_{zN}^c$	kc_z	convective mass conductivity in z-direction	property	$m^{-1}\ s$	29
V <sub>51</sub>	$k_N^c$	kc	convective mass conductivity	property	$m^{-1}\ s$	30
V <sub>52</sub>	$k_{xNS}^d$	kd_x	diffusional mass conductivity in x-direction	property	$kg^{-1}\ m^{-4}\ mol^2\ s$	31
V <sub>53</sub>	$k_{yNS}^d$	kd_y	diffusional mass conductivity in y-direction	property	$kg^{-1}\ m^{-4}\ mol^2\ s$	32
V <sub>54</sub>	$k_{zNS}^d$	kd_z	diffusional mass conductivity in z-direction	property	$kg^{-1}\ m^{-4}\ mol^2\ s$	33
V <sub>55</sub>	$k_{NS}^d$	kd	diffusional mass conductivity	property	$kg^{-1}\ m^{-4}\ mol^2\ s$	34
V <sub>56</sub>	$h_{NS}$	h	partial molar enthalpies	property	$kg\ m^2\ mol^{-1}\ s^{-2}$	35
V <sub>59</sub>	$\rho_N$	density	density	property	$kg\ m^{-3}$	38
V <sub>115</sub>	$R_N^e$	elResist	electrical resistant	property	$kg\ m^2\ A^{-2}\ s^{-3}$	91 92
V <sub>116</sub>	$k^{e,\xi}_N$	elConductC	simple model for the electrical conductivity as a function of the additive	property	$kg^{-1}\ m^{-2}\ A^2\ s^3$	93

## 7 macroscopic

	var	symbol	documentation	type	units	eqs
V <sub>92</sub>	$\hat{V}_A$	fV	volumetric flow	transport	$m^3 s^{-1}$	<a href="#">67</a> <a href="#">140</a>
V <sub>93</sub>	$\hat{n}_{AS}^d$	fnd_AS	diffusional mass flow in a given stream	transport	$mol s^{-1}$	<a href="#">68</a>
V <sub>94</sub>	$\hat{n}_{NS}^d$	fnd	net diffusional mass flow	transport	$mol s^{-1}$	<a href="#">69</a>
V <sub>95</sub>	$\hat{H}_{AS}^d$	fHd_A	enthalpy flow per diffusional mass stream	transport	$kg m^2 s^{-3}$	<a href="#">70</a>
V <sub>96</sub>	$\hat{H}_N^d$	fHd	net enthalpy stream due to diffusion	transport	$kg m^2 s^{-3}$	<a href="#">71</a>
V <sub>97</sub>	$d_A$	d	flow direction of convectional flow	transport		<a href="#">72</a>
V <sub>99</sub>	$\hat{n}_{AS}^c$	fnc_AS	molar convectional mass flow in the given stream	transport	$mol s^{-1}$	<a href="#">74</a>
V <sub>100</sub>	$\hat{n}_{NS}^c$	fnc	net molar convectional mass flow	transport	$mol s^{-1}$	<a href="#">75</a>
V <sub>102</sub>	$\hat{H}_{AS}^c$	fHc_A	convective enthalpy flow for given stream	transport	$kg m^2 s^{-3}$	<a href="#">77</a>
V <sub>103</sub>	$\hat{H}_N^c$	fHc	net convectional enthalpy stream	transport	$kg m^2 s^{-3}$	<a href="#">78</a>
V <sub>104</sub>	$\hat{w}_A$	fw_A	sample work stream	transport	$kg m^2 s^{-3}$	<a href="#">79</a>
V <sub>105</sub>	$\hat{w}_N$	fw	net work stream	transport	$kg m^2 s^{-3}$	<a href="#">80</a>
V <sub>106</sub>	$\hat{q}_{xA}$	fq_A_x	heat flow in x-direction for given stream	transport	$kg m^2 s^{-3}$	<a href="#">81</a>
V <sub>107</sub>	$\hat{q}_N$	fq	net heat flow	transport	$kg m^2 s^{-3}$	<a href="#">82</a>
V <sub>173</sub>	$\hat{n}_{AS}^{c,controlled}$	fnc_AS_controlled	switched flow	transport	$mol s^{-1}$	<a href="#">141</a>
V <sub>71</sub>	$A_{yzN}$	Ayz	cross sectional area yz	geometry	$m^2$	<a href="#">48</a>
V <sub>72</sub>	$A_{xzN}$	Axz	cross sectional area xz	geometry	$m^2$	<a href="#">49</a>
V <sub>73</sub>	$A_{xyN}$	Axy	cross sectional area xy	geometry	$m^2$	<a href="#">50</a>
V <sub>70</sub>	$F_{NS,AS}$	F_NS_AS	species related incidence matrix	network		
V <sub>90</sub>	$D_{N,A}$	D	difference operator	differenceOperator		
V <sub>91</sub>	$D_{NS,AS}$	D_NS_AS	difference operator for species topology	differenceOperator		
V <sub>109</sub>	$H_N^o$	Ho	initial enthalpy	state	$kg m^2 s^{-2}$	<a href="#">84</a>

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	var	symbol	documentation	type	units	eqs
V <sub>1</sub> 10	$n_{NS}^o$	no	initial species	state	$mol$	85
V <sub>1</sub> 27	$1_S$	one_S	a vector of ones with the length of the ordinal number of S	constant		
V <sub>5</sub> 7	$m_N$	m	total mass	secondaryState	$kg$	36
V <sub>6</sub> 6	$c_{NS}$	c	molar composition	secondaryState	$m^{-3} mol$	44
V <sub>9</sub> 8	$c_{AS}$	c_AS	concentration in convectional flow	secondaryState	$m^{-3} mol$	73
V <sub>1</sub> 26	$\phi$	intensities	collected intensities	secondaryState		106
V <sub>1</sub> 28	$n_N^t$	nTotal	total number of moles	secondaryState	$mol$	107
V <sub>1</sub> 68	$n_{tN}$	nt	total number of species in a node	secondaryState	$mol$	134
V <sub>1</sub> 69	$\xi_{NS}$	xi	mole fraction	secondaryState		135
V <sub>1</sub> 70	$\tilde{n}_{NS}$	nProd	production term	conversion	$mol s^{-1}$	137
V <sub>1</sub> 01	$\dot{n}_{NS}$	dndt	differential species balance	diffState	$mol s^{-1}$	76 142
V <sub>1</sub> 08	$\dot{H}_N$	dHdt	differential enthalpy balance	diffState	$kg m^2 s^{-3}$	83
V <sub>1</sub> 18	$\dot{U}_N^e$	dUedt	Kirkhoffs first law	diffState	$kg m^2 A^{-1} s^{-3}$	96 97 98
V <sub>1</sub> 13	$I_N$	i	electrical current definition	internalTransport	$A$	89



## 8 solid

	var	symbol	documentation	type	units	eqs
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## 9 fluid

	var	symbol	documentation	type	units	eqs
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## 10 liquid

	var	symbol	documentation	type	units	eqs
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## 11 gas

	var	symbol	documentation	type	units	eqs
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## 12 control-control

	var	symbol	documentation	type	units	eqs
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## 13 gas–liquid

	var	symbol	documentation	type	units	eqs
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## 14 gas–gas

	var	symbol	documentation	type	units	eqs
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## 15 liquid–liquid

	var	symbol	documentation	type	units	eqs
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## 16 gas–solid

	var	symbol	documentation	type	units	eqs
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## 17    solid–solid

	var	symbol	documentation	type	units	eqs
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## 18 liquid–solid

	var	symbol	documentation	type	units	eqs
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## 19 material–material

	var	symbol	documentation	type	units	eqs
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## 20 reactions-reactions

	var	symbol	documentation	type	units	eqs
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## 21 control-reactions

	var	symbol	documentation	type	units	eqs
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## 22 reactions-control

	var	symbol	documentation	type	units	eqs
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## 23 control-material

	var	symbol	documentation	type	units	eqs
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## 24 material-control

	var	symbol	documentation	type	units	eqs
V <sub>124</sub>	$\mapsto \xi$	<code>_additive</code>	link variable additive to interface material » > control	get		104

## 25 control-macroscopic

	var	symbol	documentation	type	units	eqs
V <sub>172</sub>	<i>s</i>	_switch	switches at to	get		139

## 26 macroscopic-control

	var	symbol	documentation	type	units	eqs
V <sub>1</sub> 19	$I_N$	_i	link variable i to interface macroscopic »> control	get	$A$	99
V <sub>1</sub> 25	$T_N$	_T	link variable T to interface macroscopic »> control	get	$K$	105
V <sub>1</sub> 42	$Ue_N$	_Ue	link variable Ue to interface macroscopic »> control	get	$kg\,m^2\,A^{-1}s^{-3}$	114

## 27 reactions-material

	var	symbol	documentation	type	units	eqs
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## 28 material-reactions

	var	symbol	documentation	type	units	eqs
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## 29 reactions-macroscopic

	var	symbol	documentation	type	units	eqs
V164	$\tilde{n}_{NS}$	_nProd	link variable nProd to interface reactions »> macroscopic	get	$mol\ s^{-1}$	131

## 30 macroscopic-reactions

	var	symbol	documentation	type	units	eqs
V <sub>67</sub>	$c_{NS}$	<code>_c</code>	link variable c to interface macroscopic »> reactions	get	$m^{-3} mol$	45

### 31 material–macroscopic

	var	symbol	documentation	type	units	eqs
V <sub>41</sub>	$\lambda_S$	_Mm	link variable Mm to interface material »> macroscopic	get	$kg\,mol^{-1}$	20
V <sub>74</sub>	$\rho_N$	_density	link variable density to interface material »> macroscopic	get	$kg\,m^{-3}$	51
V <sub>75</sub>	$h_{NS}$	_h	link variable h to interface material »> macroscopic	get	$kg\,m^2\,mol^{-1}\,s^{-2}$	52
V <sub>76</sub>	$k_{xN}^q$	_kq_x	link variable kq x to interface material »> macroscopic	get	$kg\,K^{-1}\,s^{-3}$	53
V <sub>77</sub>	$C_{vN}^v$	_Cv	link variable Cv to interface material »> macroscopic	get	$kg\,m^2\,K^{-1}\,s^{-2}$	54
V <sub>78</sub>	$k_{yN}^q$	_kq_y	link variable kq y to interface material »> macroscopic	get	$kg\,K^{-1}\,s^{-3}$	55
V <sub>79</sub>	$k_{zN}^q$	_kq_z	link variable kq z to interface material »> macroscopic	get	$kg\,K^{-1}\,s^{-3}$	56
V <sub>80</sub>	$k_N^q$	_kq	link variable kq to interface material »> macroscopic	get	$kg\,K^{-1}\,s^{-3}$	57
V <sub>81</sub>	$k_{xN}^c$	_kc_x	link variable kc x to interface material »> macroscopic	get	$m^{-1}\,s$	58
V <sub>82</sub>	$C_{pN}$	_Cp	link variable Cp to interface material »> macroscopic	get	$kg\,m^2\,K^{-1}\,s^{-2}$	59
V <sub>83</sub>	$k_{yN}^c$	_kc_y	link variable kc y to interface material »> macroscopic	get	$m^{-1}\,s$	60
V <sub>84</sub>	$k_{zN}^c$	_kc_z	link variable kc z to interface material »> macroscopic	get	$m^{-1}\,s$	61
V <sub>85</sub>	$k_N^c$	_kc	link variable kc to interface material »> macroscopic	get	$m^{-1}\,s$	62
V <sub>86</sub>	$k_{xNS}^d$	_kd_x	link variable kd x to interface material »> macroscopic	get	$kg^{-1}\,m^{-4}\,mol^2\,s$	63
V <sub>87</sub>	$k_{yNS}^d$	_kd_y	link variable kd y to interface material »> macroscopic	get	$kg^{-1}\,m^{-4}\,mol^2\,s$	64
V <sub>88</sub>	$k_{zNS}^d$	_kd_z	link variable kd z to interface material »> macroscopic	get	$kg^{-1}\,m^{-4}\,mol^2\,s$	65

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	var	symbol	documentation	type	units	eqs
V <sub>89</sub>	$k^d_{NS}$	<code>_kd</code>	link variable kd to interface material »> macroscopic	get	$kg^{-1} m^{-4} mol^2 s$	66
V <sub>117</sub>	$R^e_N$	<code>_elConductC</code>	link variable elConductC to interface material »> macroscopic	get	$kg^{-1} m^{-2} A^2 s^3$	94
V <sub>140</sub>	$\xi$	<code>_additive</code>	link variable additive to interface material »> macroscopic	get		112

## 32 macroscopic-material

	var	symbol	documentation	type	units	eqs
V <sub>58</sub>	$m_N$	_m	link variable m to interface macroscopic »> material	get	$kg$	37
V <sub>114</sub>	$I_N$	_i	link variable i to interface macroscopic »> material	get	$A$	90

### 33 Equations

### 34 Generic

no	equation	documentation	layer
6	$p_N := \left(-\frac{\partial U_N}{\partial V_N}\right)$	thermodynamic pressure	physical
7	$T_N := \frac{\partial U_N}{\partial S_N}$	temperature	physical
8	$\mu_{NS} := \frac{\partial U_N}{\partial n_{NS}}$	chemical potential	physical
9	$H_N := U_N - p_N \cdot V_N$	enthalpy	physical
10	$A_N := U_N - T_N \cdot S_N$	Helmholtz energy	physical
11	$G_N := U_N + p_N \cdot V_N - T_N \cdot S_N$	Gibbs free energy	physical
14	$Ue_N := (C_N)^{-1} \cdot U_N$	electrical potential – voltage	physical
15	$v_{xN} := \frac{\partial r_{xN}}{\partial t}$	velocity in x-direction	physical
16	$v_{yN} := \frac{\partial r_{yN}}{\partial t}$	velocity in y direction	physical
17	$v_{zN} := \frac{\partial r_{zN}}{\partial t}$	velocity in z-direction	physical
21	$C_{pN} := \frac{\partial H_N}{\partial T_N}$	total heat capacity at constant pressure	material
22	$C_{VN} := \frac{\partial U_N}{\partial T_N}$	total heat capacity at constant volume	material
23	$k_{xN}^q := (V_N)^{-1} \cdot \frac{\partial U_N}{\partial T_N} \cdot v_{xN}$	thermal conductivity in x-direction	material
24	$k_{yN}^q := (V_N)^{-1} \cdot \frac{\partial U_N}{\partial T_N} \cdot v_{yN}$	thermal conductivity in y-direction	material
25	$k_{zN}^q := (V_N)^{-1} \cdot \frac{\partial U_N}{\partial T_N} \cdot v_{zN}$	thermal conductivity in z-direction'	material
26	$k_N^q := \text{Stack}(k_{xN}^q, k_{yN}^q, k_{zN}^q)$	thermal conductivity	material

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no	equation	documentation	layer
27	$k_{xN}^c := \left( \lambda_S^{S \in NS} \star (\mu_{NS})^{-1} \right) \cdot (V_N)^{-1} \cdot \frac{\partial U_N}{\partial p_N} \cdot v_{xN}$	convective mass conductivity in x-direction	material
28	$k_{yN}^c := \left( \lambda_S^{S \in NS} \star (\mu_{NS})^{-1} \right) \cdot (V_N)^{-1} \cdot \frac{\partial U_N}{\partial p_N} \cdot v_{yN}$	convective mass conductivity in y-direction	material
29	$k_{zN}^c := \left( \lambda_S^{S \in NS} \star (\mu_{NS})^{-1} \right) \cdot (V_N)^{-1} \cdot \frac{\partial U_N}{\partial p_N} \cdot v_{zN}$	convective mass conductivity in z-direction	material
30	$k_N^c := \text{Stack}(k_{xN}^c, k_{yN}^c, k_{zN}^c)$	convective mass conductivity	material
31	$k_{xNS}^d := (\mu_{NS})^{-1} \cdot \left( v_{xN} \odot \left( (V_N)^{-1} \odot \frac{\partial U_N}{\partial \mu_{NS}} \right) \right)$	diffusional mass conductivity in x-direction	material
32	$k_{yNS}^d := (\mu_{NS})^{-1} \cdot \left( v_{yN} \odot \left( (V_N)^{-1} \odot \frac{\partial U_N}{\partial \mu_{NS}} \right) \right)$	diffusional mass conductivity in y-direction	material
33	$k_{zNS}^d := (\mu_{NS})^{-1} \cdot \left( v_{zN} \odot \left( (V_N)^{-1} \odot \frac{\partial U_N}{\partial \mu_{NS}} \right) \right)$	diffusional mass conductivity in z-direction	material
34	$k_{NS}^d := \text{Stack}(k_{xNS}^d, k_{yNS}^d, k_{zNS}^d)$	diffusional mass conductivity	material
35	$h_{NS} := H_N \odot (n_{NS})^{-1}$	partial molar enthalpies	material
36	$m_N := \lambda_S^{S \in NS} \star n_{NS}$	total mass	macroscopic
38	$\rho_N := m_N \cdot (V_N)^{-1}$	density	material
39	$T_{NK} := P_{N,NK} \star^N T_N$	temperature of the reactive system	reactions
42	$K_{NK} := K^o_K \odot \exp((-E^a_{NK}) \cdot (R \cdot T_{NK})^{-1})$	Arrhenius reaction 'constant'	reactions
44	$c_{NS} := (V_N)^{-1} \odot n_{NS}$	molar composition	macroscopic
48	$A_{yzN} := r_{yN} \cdot r_{zN}$	cross sectional area yz	macroscopic

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no	equation	documentation	layer
49	$A_{xzN} := r_{xN} \cdot r_{zN}$	cross sectional area xz	macroscopic
50	$A_{xyN} := r_{xN} \cdot r_{yN}$	cross sectional area xy	macroscopic
67	$\hat{V}_A := (\rho_N)^{-1} \cdot k_{xN}^c \cdot A_{yzN} \cdot D_{N,A} \star^N p_N$	volumetric flow	macroscopic
68	$\hat{n}_{AS}^d := A_{yzN} \odot (-k_{xNS}^d) \cdot D_{NS,AS} \star^{NS} \mu_{NS}$	diffusional mass flow in a given stream	macroscopic
69	$\hat{n}_{NS}^d := F_{NS,AS} \star^{AS} \hat{n}_{AS}^d$	net diffusional mass flow	macroscopic
70	$\hat{H}_A^d := \left( F_{NS,AS} \star^{NS} h_{NS} \right) \star^{S \in AS} \hat{n}_{AS}^d$	enthalpy flow per diffusional mass stream	macroscopic
71	$\hat{H}_N^d := F_{N,A} \star^A \hat{H}_A^d$	net enthaply stream due to diffusion	macroscopic
72	$d_A := \text{sign} \left( F_{N,A} \star^N p_N \right)$	flow direction of convectional flow	macroscopic
73	$c_{AS} := (0.5 \cdot (F_{NS,AS} - d_A \odot  F_{NS,AS} )) \star^{NS} c_{NS}$	concentration in convectional flow	macroscopic
74	$\hat{n}_{AS}^c := \hat{V}_A \odot c_{AS}$	molar convetional mass flow in the given stream	macroscopic
75	$\hat{n}_{NS}^c := F_{NS,AS} \star^{AS} \hat{n}_{AS}^c$	net molar convectional mass flow	macroscopic
76	$\dot{n}_{NS} := \hat{n}_{NS}^c + \hat{n}_{NS}^d + \tilde{n}_{NS}$	differential species balance	macroscopic
77	$\hat{H}_A^c := \left( F_{NS,AS} \star^{NS} h_{NS} \right) \star^{S \in AS} \hat{n}_{AS}^c$	convective enthalpy flow for given stream	macroscopic
78	$\hat{H}_N^c := F_{N,A} \star^A \hat{H}_A^c$	net convectional enthalpy stream	macroscopic
80	$\hat{w}_N := F_{N,A} \star^A \hat{w}_A$	net work stream	macroscopic
81	$\hat{q}_{xA} := A_{yzN} \cdot k_{xN}^q \cdot D_{N,A} \star^N T_N$	heat flow in x-direction for given stream	macroscopic
82	$\hat{q}_N := F_{N,A} \star^A \hat{q}_{xA}$	net heat flow	macroscopic

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no	equation	documentation	layer
83	$\dot{H}_N := \hat{H}^c_N + \hat{H}^d_N + \hat{q}_N + \hat{w}_N$	differential enthalpy balance	macroscopic
86	$n_{NS} := \int_{t^o}^{t^e} \dot{n}_{NS} dt + n^o_{NS}$	fundamental state – molar mass	macroscopic
87	$H_N := \int_{t^o}^{t^e} \dot{H}_N dt$	enthalpy	macroscopic
89	$I_N := \frac{dC_N}{dt}$	electrical current definition	macroscopic
93	$k^{e,\xi}_N := (R^e_N)^{-1} \cdot \xi$	simple model for the electrical conductivity as a function of the additive	material
95	$Ue_N := (R^e_N)^{-1} \cdot I_N$	electrical potential – voltage	macroscopic
96	$\dot{U}^e_N := 1 \cdot Ue_N$	Kirkhoff first law	macroscopic
97	$\dot{U}^e_N := \text{Root}(Ue_N)$	Kirkhoff first law	macroscopic
106	$\phi := \text{MixedStack}(p_N, T_N, \mu_{NS}, c_{NS}, Ue_N)$	collected intensities	macroscopic
107	$n^t_N := 1_S \overset{S \in NS}{\star} n_{NS}$	total number of moles	macroscopic
108	$e_A := m_A - y^o_A$	control error	control
110	$\dot{x}_D := A_{N,D} \overset{N}{\star} x_N + B_{A,D} \overset{A}{\star} e_A$	differential state (ABCD) model	control
111	$x_N := \int_{t^o}^{t^e} 1_{N,D} \overset{D}{\star} \dot{x}_D dt$	state	control
113	$\check{I}_N := I_N$	measured current	control

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no	equation	documentation	layer
115	$\check{U}_N := U e_N$	measured electrical potential	control
116	$\check{\xi} := \mapsto \xi$	measured additive fraction	control
117	$R_N := (\check{I}_N)^{-1} \cdot \check{U}_N$	measured resistance	control
118	$S := \text{MixedStack}(\check{I}_N, \check{U}_N, R_N, \check{\xi})$	quantities to be stored	control
123	$c_{NK,KS} := P_{NK} \cdot \left( P_{NS,KS} \overset{NS}{\star} c_{NS} \right)$	var doc :	reactions
125	$x_{NK,KS} := (c_{NK,KS}^o)^{-1} \cdot c_{NK,KS}$	matrix of normed, dimensionless mole fractions	reactions
126	$y_A := C_{N,A} \overset{N}{\star} x_N + D_A \cdot e_A$	output equation	control
127	$R := A^v \cdot B$	gas constant	reactions
128	$N_{NK,KS} := P_{K,NK} \overset{K}{\star} N_{K,KS}$	extended stoichiometrix matrix	reactions
129	$\phi_{NK} := \prod_{KS} x_{NK,KS}^{N_{NK,KS}}$	probability function for reactions	reactions
130	$\tilde{n}_{NS} := V_N \overset{N}{\star} \left( P_{N,NK} \overset{NK}{\star} \left( (K_{NK} \cdot \phi_{NK}) \cdot \left( P_{NS,KS} \overset{KS}{\star} N_{NK,KS} \right) \right) \right)$	the species production term	reactions
132	$boz_N := \text{Instantiate}(S_N, \#)$	Boltzmann constant	physical

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no	equation	documentation	layer
133	$R_N := A^v . boz_N$	gas constant	physical
134	$n_{tN} := 1_S \overset{S \in NS}{\star} n_{NS}$	total number of species in a node	macroscopic
135	$\xi_{NS} := (n_{tN})^{-1} \odot n_{NS}$	mole fraction	macroscopic
136	$\mu_{NS} := (R_N . T_N) \odot \ln(\xi_{NS})$	chemical potential	macroscopic
137	$\tilde{n}_{NS} := \tilde{n}_{NS}$	production term	macroscopic
138	$s := 0.5 . (1 + \text{sign}(t^o))$	switches at to	control
139	$s := s$	switches at to	control $\rightarrow$ macroscopic
141	$\hat{n}^{c,controlled}_{AS} := s . \hat{n}^c_{AS}$	switched flow	macroscopic
142	$\dot{n}_{NS} := F_{NS,AS} \overset{AS}{\star} \text{Stack}(\hat{n}^c_{AS}, \hat{n}^{c,controlled}_{AS})$	differential species balance switched	macroscopic



## 35 Instantiate

no	equation	documentation	layer
1	$1 := \text{Instantiate}(\#, \#)$	numerical value 1	root
2	$0 := \text{Instantiate}(\#, \#)$	numerical value zero	root
3	$0.5 := \text{Instantiate}(\#, \#)$	numerical value one half	root
4	$t^o := \text{Instantiate}(t, \#)$	starting time	root
5	$t^e := \text{Instantiate}(t, \#)$	end time	root
41	$E^a_{NK} := \text{Instantiate}(R.T_{NK}, \#)$	Arrhenius activation energy	reactions
79	$\hat{w}_A := \text{Instantiate}(\hat{H}^c_A, \#)$	sample work stream	macroscopic
84	$H^o_N := \text{Instantiate}(H_N, \#)$	initial enthalpy	macroscopic
85	$n^o_{NS} := \text{Instantiate}(n_{NS}, \#)$	initial species	macroscopic
88	$\xi := \text{Instantiate}(\xi, \#)$	fraction of additives	material
91	$R^e_N := (I_N)^{-1} . Ue_N$	electrical resistant	material
92	$R^e_N := \text{Instantiate}(R^e_N, \#)$	electrical resistant	material
98	$\dot{U}^e_N := \text{Instantiate}(\dot{U}^e_N, 0)$	Kirkhoff first law	macroscopic
109	$x_{oN} := \text{Instantiate}(x_N, \#)$	initial state	control
119	$y^o_A := \text{Instantiate}(y^o_A, \#)$	set point	control

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no	equation	documentation	layer
124	$c^o_{NK,KS} := \text{Instantiate}(c_{NK,KS}, \#)$	norming concentration	reactions

## 36 Instantiation Equation

no	equation	documentation	layer
140	$\hat{V}_A := \text{Instantiate}(\hat{V}_A, \#)$	instantiation equation	macroscopic

## 37 Interface Link Equation

no	equation	documentation	layer
20	$\lambda_S := \lambda_S$	interface equation	material → macroscopic
37	$m_N := m_N$	interface equation	macroscopic → material
45	$c_{NS} := c_{NS}$	interface equation	macroscopic → reactions
51	$\rho_N := \rho_N$	interface equation	material → macroscopic
52	$h_{NS} := h_{NS}$	interface equation	material → macroscopic
53	$k_{xN}^q := k_{xN}^q$	interface equation	material → macroscopic
54	$C^{v_N} := C^{v_N}$	interface equation	material → macroscopic
55	$k_{yN}^q := k_{yN}^q$	interface equation	material → macroscopic
56	$k_{zN}^q := k_{zN}^q$	interface equation	material → macroscopic
57	$k_N^q := k_N^q$	interface equation	material → macroscopic
58	$k_{xN}^c := k_{xN}^c$	interface equation	material → macroscopic

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no	equation	documentation	layer
59	$CP_N := C_{pN}$	interface equation	material → macroscopic
60	$k_{yN}^c := k_{yN}^c$	interface equation	material → macroscopic
61	$k_{zN}^c := k_{zN}^c$	interface equation	material → macroscopic
62	$k_N^c := k_N^c$	interface equation	material → macroscopic
63	$k_{xNS}^d := k_{xNS}^d$	interface equation	material → macroscopic
64	$k_{yNS}^d := k_{yNS}^d$	interface equation	material → macroscopic
65	$k_{zNS}^d := k_{zNS}^d$	interface equation	material → macroscopic
66	$k_{NS}^d := k_{NS}^d$	interface equation	material → macroscopic
90	$I_N := I_N$	interface equation	macroscopic → material
94	$R_N^e := k_N^{e,\xi}$	interface equation	material → macroscopic
99	$I_N := I_N$	interface equation	macroscopic → control
104	$\mapsto \xi := \xi$	interface equation	material → control

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no	equation	documentation	layer
105	$T_N := T_N$	interface equation	macroscopic control $\rightarrow$
112	$\xi := \xi$	interface equation	material $\rightarrow$ macroscopic
114	$Ue_N := Ue_N$	interface equation	macroscopic control $\rightarrow$
131	$\tilde{n}_{NS} := \tilde{n}_{NS}$	interface equation	reactions $\rightarrow$ macroscopic