



Nature Chemistry

# Towards Quantum Chemistry on a Quantum Computer

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# Outline

- Problem to solve
- Experiment
- Influence

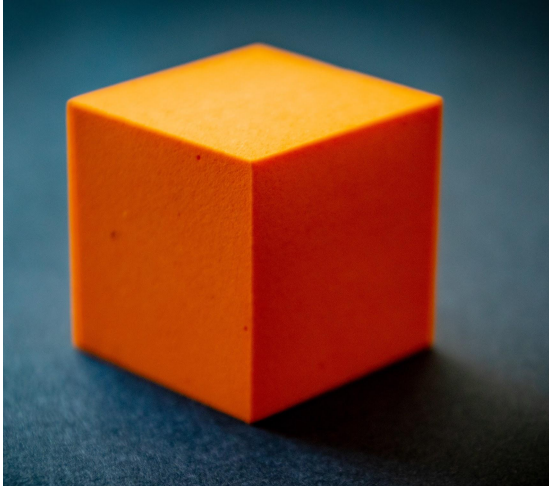


# The problem

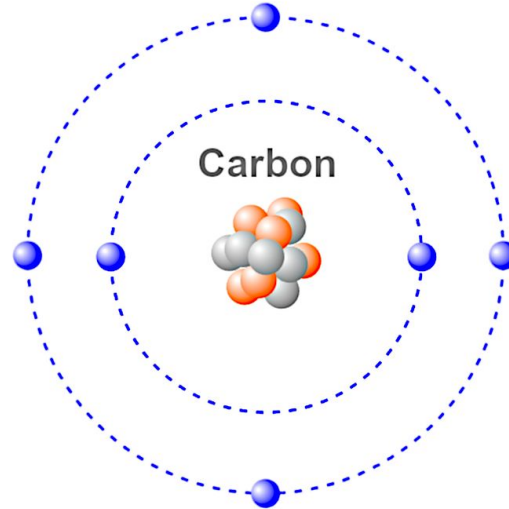
“**Exact first-principles calculations** of molecular properties are currently **intractable** because their **computational cost** grows exponentially with both the number of atoms and basis set size.”



# First Principles Calculations



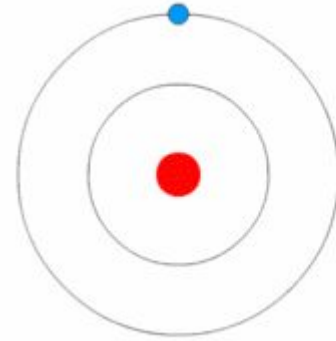
Condensed Matter



Atoms and Molecules

# First Principles Calculations

“For the most part, the way this is done is to pick a particular macroscopic phenomenon, which has been well studied experimentally, and to build empirical, or semi-empirical, models to describe the experimentally observed results.”



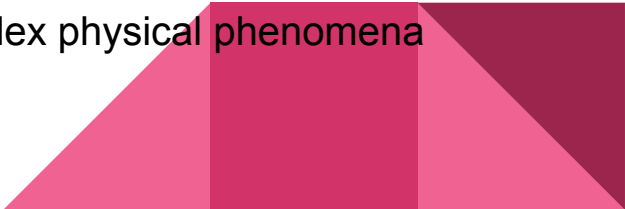
The Niels Bohr model of Hydrogen atom.

# First Principles Calculations

**The first principles approach** to condensed matter theory is entirely different from this. It starts from what we know about all condensed matter systems - that they are made of **atoms**, which in turn are made of **a positively charged nucleus**, and **a number of negatively charged electrons**. The interactions between atoms, such as chemical and molecular bonding, are determined by the **interactions of their constituent electrons and nuclei**.

**All of the physics of condensed matter systems arises ultimately from these basic interactions.**

If we can model these interactions accurately, then all of the complex physical phenomena that arise from them should emerge naturally in our calculations.



# First Principles Calculations

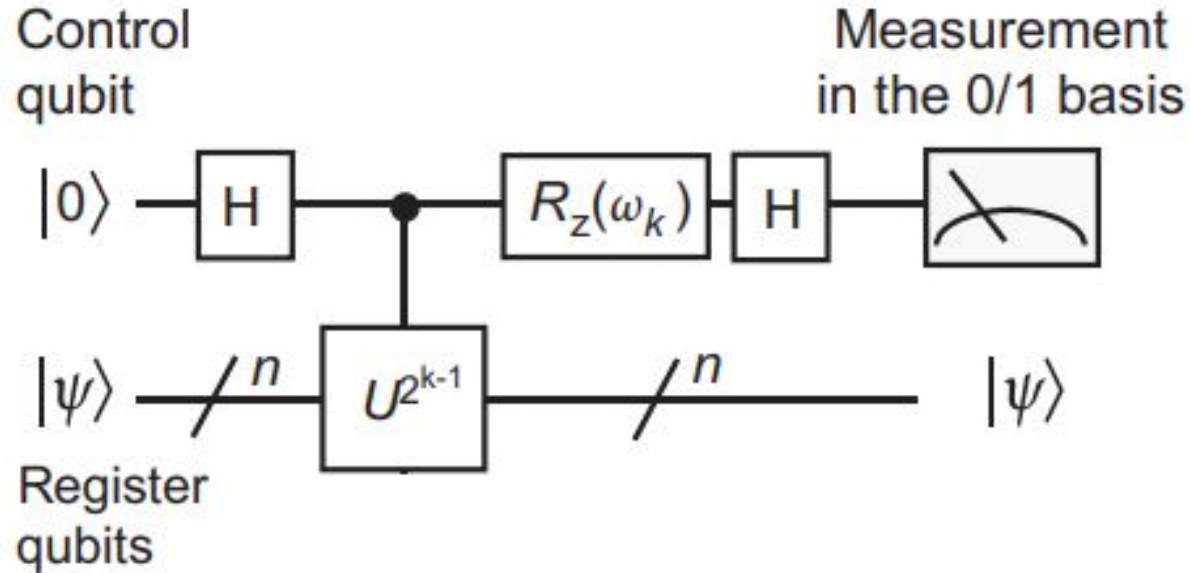
The Many-Electron Schrödinger Equation

$$\left| \hat{T}\Psi_{\text{MB}} + \hat{V}\Psi_{\text{MB}} = -i\frac{d\Psi_{\text{MB}}}{dt}, \right.$$

**Intractable Computational Problem even for Supercomputers**



# Linear Optic Quantum Computer(2 qubits)





# Methodology

1. Encoding a molecular wave function into qubits;
2. Simulating its time evolution using quantum logic gates;
3. Extracting the approximate energy using the phase estimation algorithm.



# Encoding Wave Function into Qubits

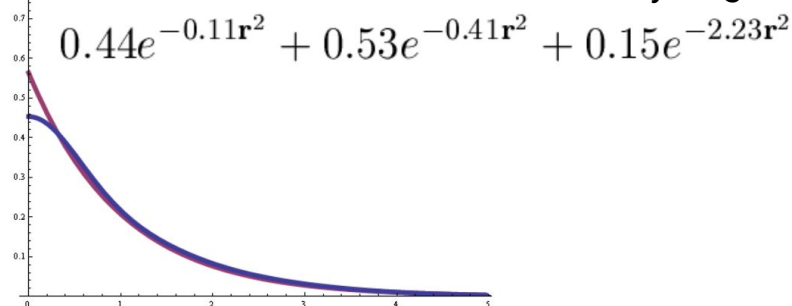
## Basis Set

A set of functions (called basis functions) that is used to **represent the electronic wave function** in the Hartree–Fock method or density-functional theory in order to turn the partial differential equations of the model into algebraic equations **suitable for efficient implementation on a computer.**

## Minimal Basis Sets

**STO-3G**, 3 Gaussian primitive functions comprising a single basis function.

STO-3G functions for the 1s orbital of a hydrogen atom



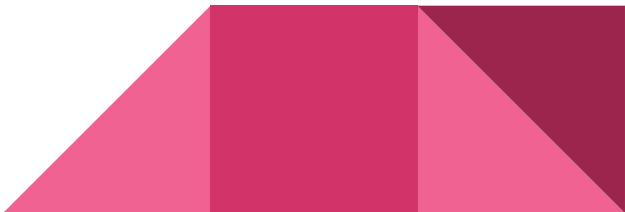
# Encoding Wave Function into Qubits

Six two-electron configurations

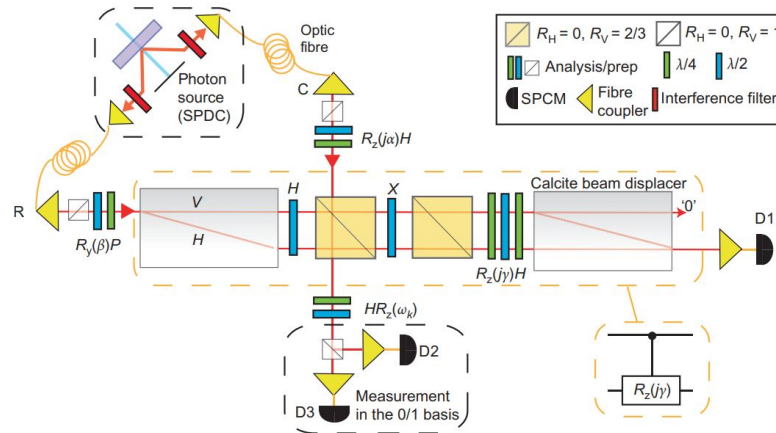
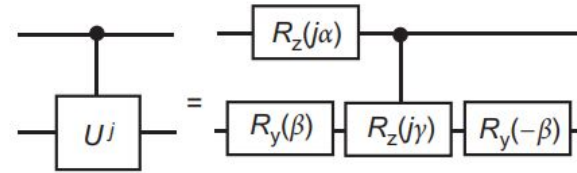
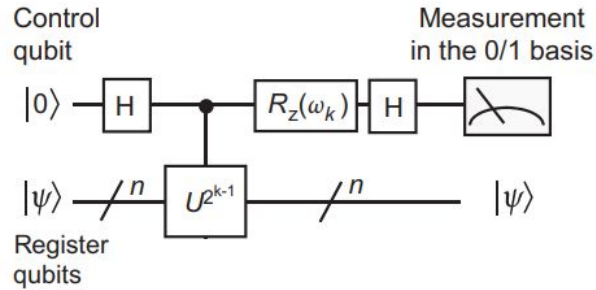
$$\{\Phi_i\}_{i=1}^6 \rightarrow \{\Phi_1, \Phi_6\} \quad \{\Phi_2\} \quad \{\Phi_3, \Phi_4\} \quad \{\Phi_5\}$$

Finding the **eigenvalues** of the two  $2 \times 2$  submatrices in the Hamiltonian,  $H^{(1,6)}$  and  $H^{(3,4)}$ , is what we need.

Estimating the eigenvalues of  $2 \times 2$  matrices is the simplest problem for the IPEA.



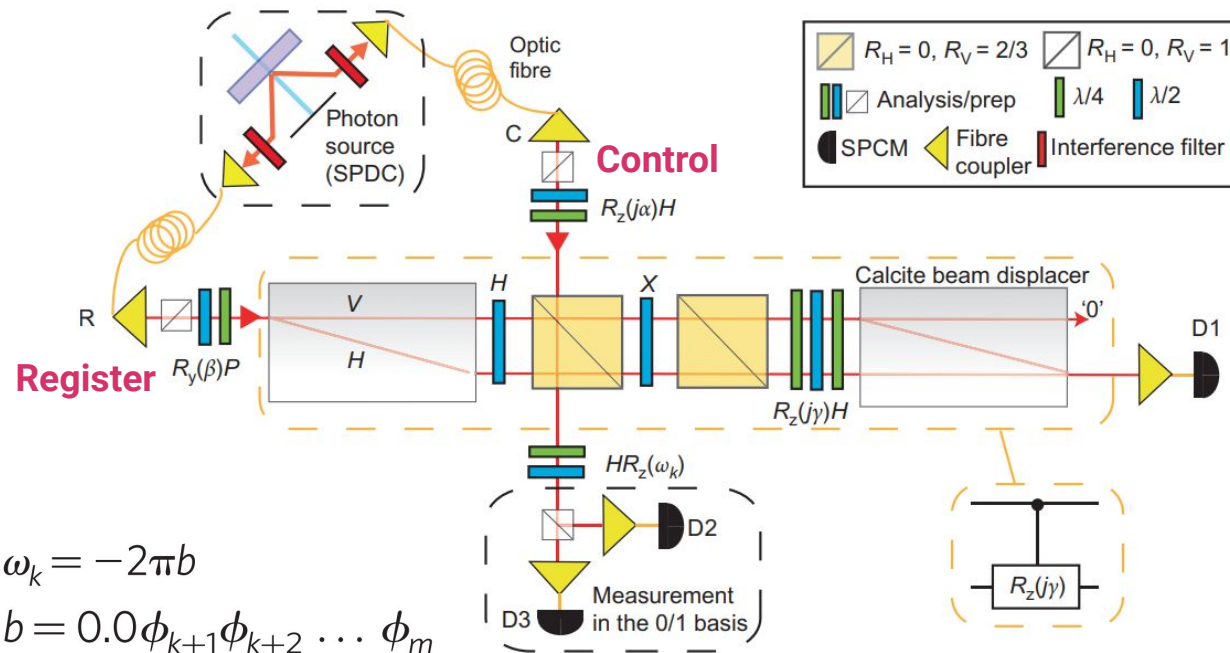
# Experiment Design



$$\hat{U} = e^{i\alpha} \hat{R}_y(\beta) \hat{R}_z(\gamma) \hat{R}_y(-\beta)$$

A series of rotations of the one-qubit Hilbert space.

# Experiment Design



$$\omega_k = -2\pi b$$

$$b = 0.0\phi_{k+1}\phi_{k+2} \dots \phi_m$$

starting from the least significant ( $\phi_m$ )

To produce an m-bit approximation to the phase of the eigenstate, we need to repeat this algorithm m times.

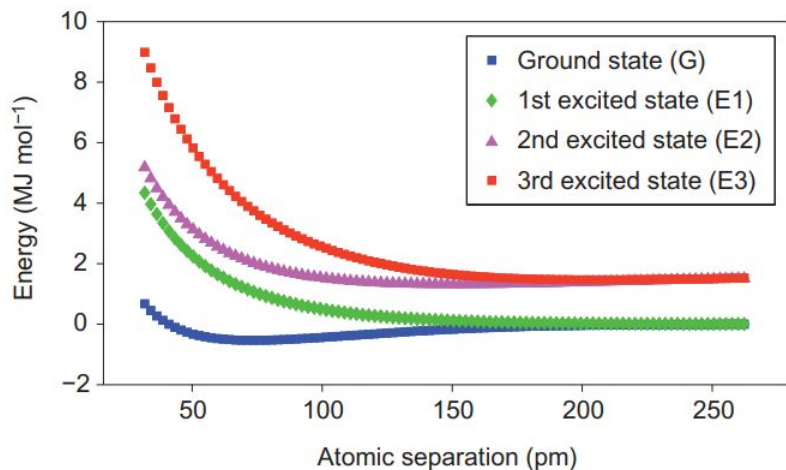
IF **control** **horizontally** polarized:

Nothing is changed

IF **control** **vertically** polarized:

$\hat{U}^{2^{k-1}}$  is implemented on the **register**

# Result



Each point was calculated with a 20-bin iterative phase estimation algorithm, using  $n=31$  samples per bit.

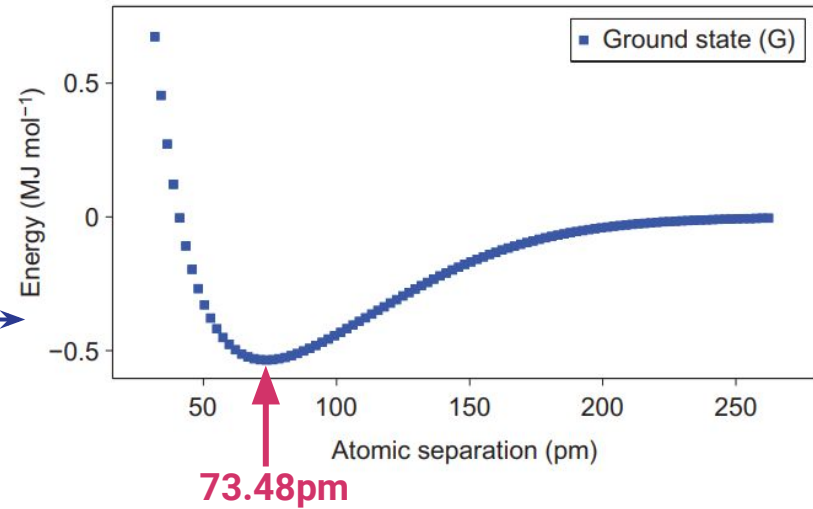
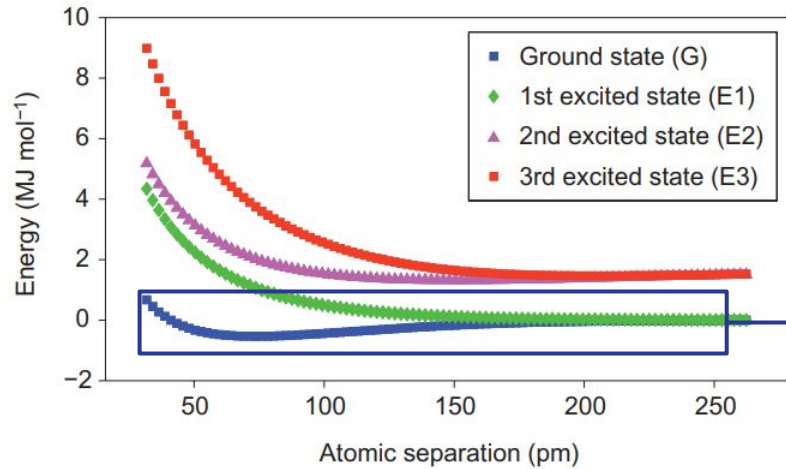
- Ground state (G) – Low Eigenvalue of  $H^{(1,6)}$
- ◆ 1st excited state (E1) – Low Eigenvalue of  $H^{(3,4)}$
- ▲ 2nd excited state (E2) – High Eigenvalue of  $H^{(3,4)}$
- 3rd excited state (E3) – High Eigenvalue of  $H^{(1,6)}$

$$E = 2\pi\phi E_h + \frac{(e^2/4\pi\epsilon R)}{E_\infty}$$

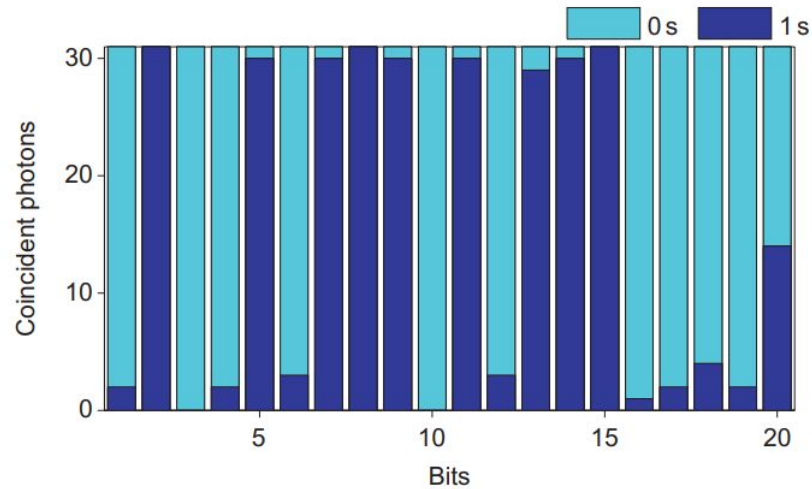
Proton-proton  
Coulomb energy  
at atomic  
separation  $R$

Ground state energy of  
two hydrogen atoms  
at infinite separation

# Result



# Result



Equal to the exact value  
of minimal basis, to a  
precision of  $\pm 2^{-20} E_h$

0.01001011101011100000



# Impact

## Previous Idea:

- [1] Feynman, R.P. Simulating physics with computers. Int J Theor Phys 21, 467–488 (1982).  
<https://doi.org/10.1007/BF02650179>
- [2] Lloyd, Seth. "Universal quantum simulators." Science (1996): 1073-1078.
- [3] Abrams, Daniel S., and Seth Lloyd. "Simulation of many-body Fermi systems on a universal quantum computer." Physical Review Letters 79.13 (1997): 2586.

## Previous Algorithm:

- [1] Brown, Kenneth R., Robert J. Clark, and Isaac L. Chuang. "Limitations of quantum simulation examined by simulating a pairing Hamiltonian using nuclear magnetic resonance." Physical review letters 97.5 (2006): 050504.
- [2] Somaroo, S., et al. "Quantum simulations on a quantum computer." Physical review letters 82.26 (1999): 5381.
- [3] Yang, Xiaodong, et al. "Experimental simulation of a pairing Hamiltonian on an NMR quantum computer." Chemical physics letters 422.1-3 (2006): 20-24.
- [4] Friedenauer, Axel, et al. "Simulating a quantum magnet with trapped ions." Nature Physics 4.10 (2008): 757-761.
- [5] Gerritsma, Rene, et al. "Quantum simulation of the Dirac equation." Nature 463.7277 (2010): 68-71.

**Experimentally** proved the feasibility of quantum computer simulating particles.

# Is it believable?

Yes.

Nature Chemistry

582 citations in 10 years.



# Limitations

- 2-qubits quantum computer
- STO-3G H<sub>2</sub> model
- Non-scalable implementation of IPEA



# Questions?





Thank you