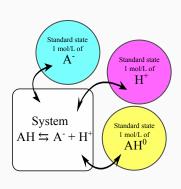
Grand-reaction ensemble.



Reaction ensemble

The reaction of an acidic unit

$$HA \stackrel{\mathcal{K}}{\hookrightarrow} A^- + H^+$$

$$\Omega = E - TS + \sum_{i} \left(\mu_{i} - \mu_{i}^{\ominus}\right) N_{i}$$

Then the change of system free energy during a single reaction step

$$\Delta\Omega = k_B T \ln \left(\prod_i V^{\nu_i \xi} \frac{N_i!}{(N_i + \nu_i \xi)!} \right) + \xi \left(\sum_i \nu_i \mu_i - \sum_i \nu_i \mu_i^{\ominus} \right) + \Delta E$$

accept if $\mathcal{R}^{\xi} < e^{\Delta\Omega/k_BT}$