

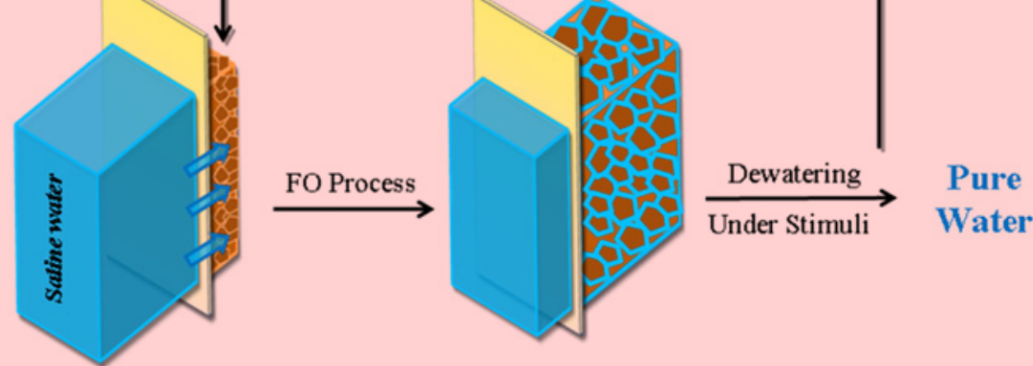
Phase transition in hydrophobic weak polyelectrolyte gel. A computer simulation study.

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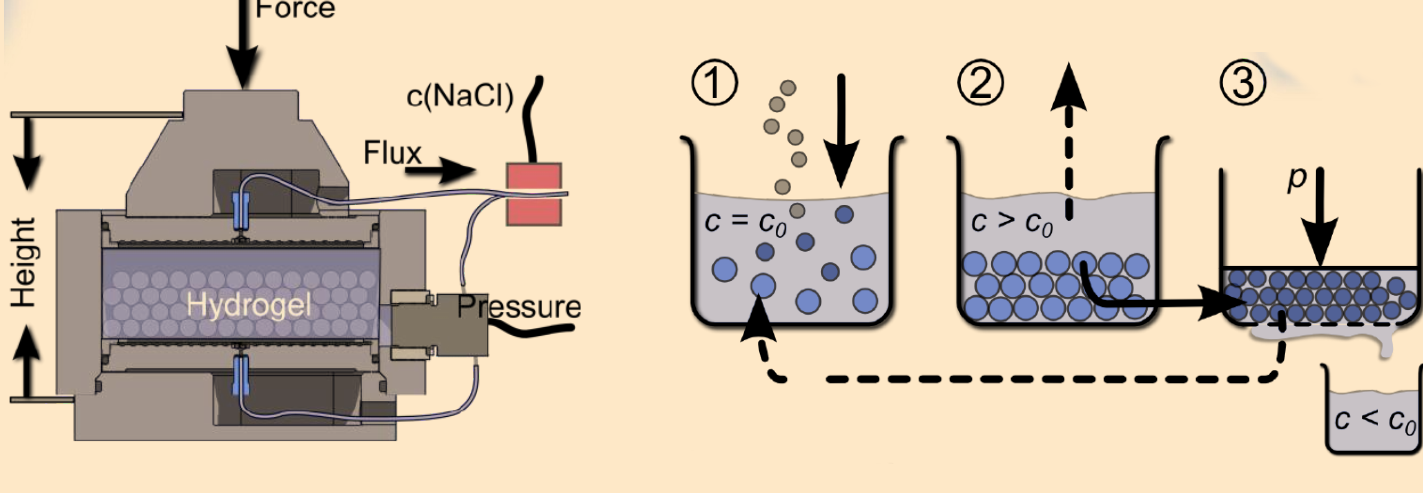
Introduction.

- Hydrogels for desalination
- Forward osmosis
- Various stimuli: thermo-, pH-, electric-, magnetic-, light-induced gel collapse



Dan Li a,c , et al. (2013). **Forward osmosis desalination using polymer hydrogels as a draw agent: Influence of draw agent, feed solution and membrane on process performance.** Water Research

Introduction. Prof. Wilhelm experiment.



Fengler, C., et al (2020). **Desalination of Seawater Using Cationic Poly(acrylamide) Hydrogels and Mechanical Forces for Separation.** Macromolecular Materials and Engineering

Mean field analytical model.

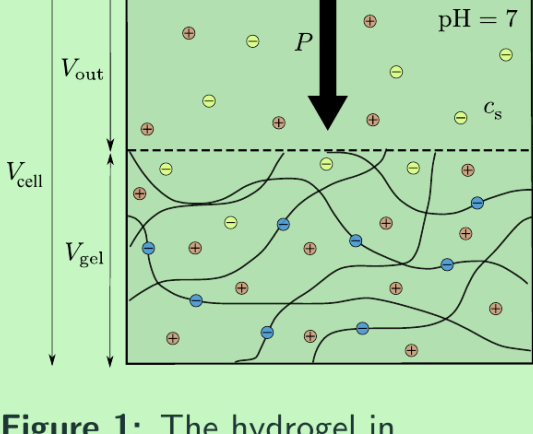


Figure 1: The hydrogel in equilibrium with a bath of aqueous solution

- Free energy of a hydrogel chain

$$F = F_{\text{conf}} + F_{\text{int}} + F_{\text{ion}}$$

- Conformational entropy

$$F_{\text{conf}} = \frac{3}{2} \frac{R^2}{1 - R^2/(b^2 N)} - \frac{3}{2} \ln \left(\frac{R^2}{b^2 N} \right)$$

- Steric interactions

$$F_{\text{int}} = \frac{N}{c_p} \left[(1 - c_p) \ln (1 - c_p) - \chi c_p^2 \right]$$

$$F_{\text{ion}} = \frac{N}{c_p} \sum_i \left(c_i^{\text{in}} \ln \frac{c_i^{\text{in}}}{c_i^{\text{out}}} + c_i^{\text{out}} - c_i^{\text{in}} \right)$$

Mean field analytical model. Donnan potential.

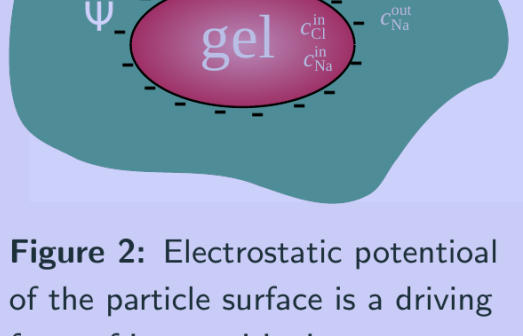


Figure 2: Electrostatic potential of the particle surface is a driving force of ion partitioning

- Donnan electrostatic potential, ψ

$$e^\psi = \xi = \frac{c_{\text{H}^+}^{\text{out}}}{c_{\text{H}^+}^{\text{in}}} = \frac{c_{\text{Na}^+}^{\text{out}}}{c_{\text{Na}^+}^{\text{in}}} = \frac{c_{\text{Cl}^-}^{\text{in}}}{c_{\text{Cl}^-}^{\text{out}}} = \frac{c_{\text{OH}^-}^{\text{in}}}{c_{\text{OH}^-}^{\text{out}}}$$

- Local electroneutrality condition

$$\alpha c_p + c_{\text{Cl}^-}^{\text{in}} + c_{\text{OH}^-}^{\text{in}} = c_{\text{Na}^+}^{\text{in}} + c_{\text{H}^+}^{\text{in}}$$

$$\xi(c_p, c_s) = \sqrt{1 + \left(\frac{\alpha c_p}{2 c_s} \right)^2} \pm \frac{\alpha c_p}{2 c_s}$$

pH sensitive hydrogel.

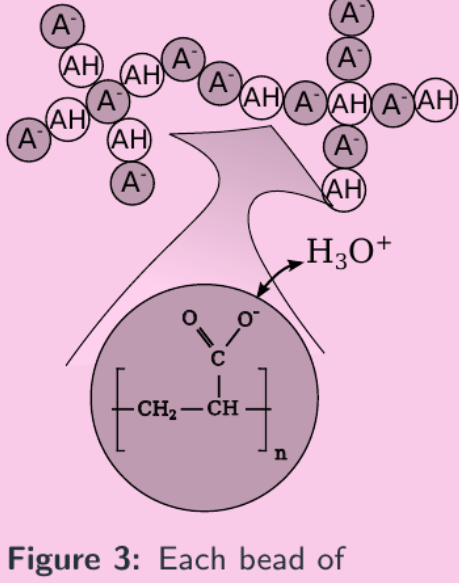


Figure 3: Each bead of hydrogel is acidic. It changes its charge depending on pH.

- ionization reaction



- ionization equilibrium

$$\frac{\alpha}{1 - \alpha} = \frac{c_{\text{H}^+}^{\text{in}}}{K} = \frac{c_{\text{H}^+}^{\text{out}}}{K} \frac{c_{\text{H}^+}^{\text{in}}}{c_{\text{H}^+}^{\text{out}}} = 10^{pK - pH} \xi^{-1}$$

$$\frac{\alpha}{1 - \alpha} = 10^{pK - pH} \left(\sqrt{1 + \left(\frac{\alpha c_p}{2 c_s} \right)^2} \mp \frac{\alpha c_p}{2 c_s} \right)$$

- Free energy ionization term

$$F_\alpha = \alpha N \left(\ln \alpha + \ln(1 - \alpha) + \ln c_{\text{H}^+}^{\text{in}} - \ln K \right)$$

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