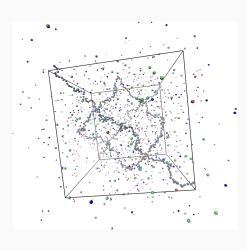
Divalent salt.



Seawater model:

$$c_{
m Cl^-} = 0.54 \;
m mol/l$$
 of negative ions

$$c_{\mathrm{Na^{+}}} \ = \ 0.47 \ \text{mol/l}$$
 of positive ions

$$c_{\mathrm{Ca^{2+}}} = 0.063 \; \mathrm{mol/I}$$
 of positive divalent ions

$$c_{\mathrm{Na^+}} \simeq 0.87 \cdot c_{\mathrm{Cl^-}}$$
 $c_{\mathrm{Ca^{2+}}} \simeq 0.117 \cdot c_{\mathrm{Cl^-}}$

$$\begin{array}{rcl} \mu_{\rm Na^+} & = & \mu_{\rm Cl^-} - 0.139 kT \\ \\ \mu_{\rm Ca^{2+}} & = & \mu_{\rm Cl^-} - 2.03 kT \end{array}$$