	measure of how easily a fluid		ney have high viscosity,	is the requirement fo while thinner ones li	or the low viscosity of the druke water have low viscosity.	t reflects the interr	nal resistance of a
	liquid's molecules to movement the eye. The European Pharm injection [5]. For this reason the viscosity in under different experiment contains the second s	nacopoeia (EP) provides specisso is a very important measure	cific guidelines regardin	ng the viscosity of int	ravitreal applied biopharmac	euticals to ensure s	Safe and effective Viscosity is tested
	and reduce costs for laborate The data set, which will be excollected as part of a charact DoppelMab), which have diffe	ory equipment and personne xplored in this work consists terization study for various beforent characteristics like mo	I, the long term motivat of viscosity data, where liopharmaceutical produ lecular weight, isoelecti	ion is to predict the veas each observation acts. These products country or extinction	riscosity of every new agent of the data set corresponds consist of different types of coefficient. In order to deter	without any expering to one measurement proteins (IgG2, IgG mine the effect of p	ments in the laborator ent value. The data wa 64, Knob/Hole, product concentration
	on the viscosity, each product - 40°C) to assess the impact Name viscosity_mPas		he data set consists of			respo	e Type Forma
	replicate entered_on instrument temperature_c	Number of replicate. With replicates The date on which the me Instrument, which was use The temperature at which	asurement was conducte	ed ity	nents were conducted as techr	predic	numeric int ctor numeric date ctor nominal catego
	product_concentration_mg_n product protein_format	Internal product name as a	uct in the aqueous solution a unique code stigated product	on in mg/mL	zo of the protoin	predic ID predic	nominal catego
	molecular_weight_kda extinction_coefficient_l_molci isoelectric_point	molecule Isoelectric point of the inv	ne investigated product in	n L·mol ⁻¹ ·cm ⁻¹ . A measu asure of the charge of	re of the light absorption abilit	y of the predic	ctor numeric float ctor numeric float ctor numeric float
	In this work the impact of different product concentration, and the between these explanatory versions. Setup	he molecular weight are cons	sidered as possible pred	dictors and likely hav	e an impact on the response		
566	<pre>import pickle import os from datetime import dat import subprocess from pathlib import Path</pre>						
	<pre>import pandas as pd import altair as alt import numpy as np from sklearn.model_select from sklearn.linear_model from sklearn.model_select</pre>	el import LinearRegressi	on				
	<pre>from sklearn.model_select from sklearn.metrics imp Data</pre>		core				
	The underlying measurement were documented in the intercontain all explanatory data li	rnal Laboratory Information N	Management System (L	MS) along with addit	cional detailed experimental a	and contextual infor	mation, which also
567	# Import data from the code df = pd.read_csv("viscos Data structure						
568 568	viscosity_mPas replicat 0 3.93	1 15.03.2019 VISCOSIMET	_	product_concentrati	10.0 BI655300	IgG2	cular_weight_kDa ex 148830
	2 3.42	 16.03.2019 VISCOSIMET 15.03.2019 VISCOSIMET 15.03.2019 VISCOSIMET 15.03.2019 VISCOSIMET 	TER_02 5 TER_02 5		10.0 BI655300 10.0 BI655300 10.0 BI655300 10.0 BI655300	IgG2 IgG2 IgG2	148830 148830 148830 148830
	<pre>df.info() <class 'pandas.core.frame="" (total="" 11="" 502="" co<="" columns="" data="" entries,="" pre="" rangeindex:=""></class></pre>	0 to 501					
	<pre># Column 0 viscosity_mPas 1 replicate 2 entered_on 3 instrument 4 temperature</pre>	Non-Null Con- 	l float64 l int64 l object l object				
	5 product_concentratio 6 product 7 protein_format 8 molecular_weight_kDa 9 extinction_coefficie 10 isoelectric_point dtypes: float64(4), int64	on_mg_mL 502 non-nul 502 non-nul 502 non-nul 502 non-nul 502 non-nul 502 non-nul	l float64 l object l object l int64 l float64				
	Data corrections According to the literature the	nere are three explanatory vai	•	-			
	temperature generally decreating impact on the viscosity, because such as aggregates or conjugate the moleculare weight of the target variable is exponentell the base-10 logarithm of the	ause molecules in solution int gated proteins, tend to increa product might have an impa I. In order to use the linear re	teract more frequently, lase solution viscosity du ct on the measured visc	leading to increased ue to their size and ir cosity. According to I	resistance to flow [7] Addition nteraction with other molecul iterature the relationship of a	nally larger and mo es in the solution [8 Il three explanatory	ore complex proteins, B], which indicates the v variables and the
	<pre># Make sure column names df.columns = df.columns. # Log-transformation of df['log_viscosity_mpas']</pre>	s are lower case and eli .str.lower() the target variable	•				
	<pre># For a better overview df = df.iloc[0:502,[4,5, df.head()</pre>	data set is reduced to ,8,11]]	the most interesting		nt to examine		
J	temperature product_co 0 2 1 2 2 5 3 5	10.0 10.0 10.0 10.0 10.0	148830 148830 148830 148830 148830	1.368639 1.453953 1.229641 1.305626			
	4 10 Variable lists	10.0	148830 148830	1.305626 1.061257			
74	<pre>y_log_label = 'log_visco # select features X = df[['temperature','p # create response</pre>	osity_mpas'	_ml','molecular_wei	ght_kda']]			
	<pre># Create response y_log = df[y_log_label] # Create list with numer list_numeric =['temperate Data splitting</pre>		tion_mg_ml','molecu	lar_weight_kda']			
75	<pre># use a test size of 0,2 X_train, X_test, y_train # use your training data df_train = pd.DataFrame(</pre>	n, y_test = train_test_s a to make a pandas dataf		_size=0.2, random	_state=42)		
75	<pre># add your training labe df_train = df_train.join df_train.head(5)</pre>	els to the data		g_viscosity_mpas			
411	temperature product 423 25 19 2 323 40 333 20	62.5 62.5 62.5 10.0	148977 148830 149601 149683	1.401183 1.905088 1.131402 1.160021			
	56 5 Analysis	10.0	149610	0.959350			
76		count mean	std		25% 50%	75%	max
	temperatu	_ml 502.0 36.354582 kda 502.0 161211.103586	26.275976 10 22382.413980 146286	.000000 10.000 .000000 10.000 .000000 148783.000 0.616186 0.788	0000 62.500000	62.500000 62 89.000000 206428	0.000000 2.500000 8.000000 3.327551
	product_concentration_mg_ molecular_weight_k log_viscosity_mp			esponse variable, cha		ining data. Accord	
	molecular_weight_k log_viscosity_mp Exploratory data ana In order to visualize the relation of all three paramination variables take continuous number 1.	ionship between the explanate neters (temperature, product merical values, the dataset c	concentration, molecul contains only a specific i	number of defined va	garithm of the viscosity is ex alues for the independent var	iables. There exist	only nine values of the
77	molecular_weight_k log_viscosity_mp Exploratory data ana In order to visualize the relatic correlation of all three paramination variables take continuous number temperature (2°C, 5°C, 10°C, weight. Therefore box plots where the continuous plots where the continu	ionship between the explanameters (temperature, product merical values, the dataset constants, 15°C, 20°C, 25°C, 30°C, 35°C, were chosen to visualize the	concentration, molecul contains only a specific of 5°C, 40°C), two values of	number of defined va of the product conce	garithm of the viscosity is ex alues for the independent var ntration (10 mg/mL, 62.5 mg	iables. There exist /mL) and 14 values	only nine values of the
77	molecular_weight_k log_viscosity_mp Exploratory data ana In order to visualize the relatice correlation of all three parameter variables take continuous number temperature (2°C, 5°C, 10°C, weight. Therefore box plots weight. Therefore box plots weight. Therefore box plots weight. Therefore decomplete (alt.Chart(df_transmark_boxplot())	ionship between the explanameters (temperature, product merical values, the dataset of 15°C, 20°C, 25°C, 30°C, 35 were chosen to visualize the estore all charts	concentration, molecule contains only a specific of two values of distribution of the response (domain=[0.9*df[x].	number of defined value of the product conce onse variable within the min(), 1.1* df[x]	garithm of the viscosity is ex alues for the independent var ntration (10 mg/mL, 62.5 mg hese values of the predictor	iables. There exist /mL) and 14 values	only nine values of the
77	molecular_weight_k log_viscosity_mp Exploratory data ana In order to visualize the relatice correlation of all three parameter variables take continuous number temperature (2°C, 5°C, 10°C, weight. Therefore box plots where the continuous plots where the continuous is a continuous number temperature (2°C, 5°C, 10°C, weight. Therefore box plots where the continuous is a continuous number temperature (2°C, 5°C, 10°C, weight. Therefore box plots where the continuous is a continuous number temperature (2°C, 5°C, 10°C, weight. Therefore box plots where the continuous is a continuous number temperature (2°C, 5°C, 10°C, weight. Therefore box plots where the continuous number temperature (2°C, 5°C, 10°C, weight. Therefore box plots where the continuous number temperature (2°C, 5°C, 10°C, weight. Therefore box plots where the continuous number temperature (2°C, 5°C, 10°C, weight. Therefore box plots where the continuous number temperature (2°C, 5°C, 10°C, weight. Therefore box plots where the continuous number temperature (2°C, 5°C, 10°C, weight. Therefore box plots where the continuous number temperature (2°C, 5°C, 10°C, weight. Therefore box plots where the continuous number temperature (2°C, 5°C, 10°C, weight. Therefore box plots where the continuous number temperature (2°C, 5°C, 10°C, weight. Therefore box plots where the continuous number temperature (2°C, 5°C, 10°C, weight. Therefore box plots where the continuous number temperature (2°C, 5°C, 10°C, 10°C, weight. Therefore box plots where the continuous number temperature (2°C, 5°C, 10°C, 10°C, weight. Therefore box plots where the continuous number temperature (2°C, 5°C, 10°C, 10°	ionship between the explanameters (temperature, product merical values, the dataset of 1,15°C, 20°C, 25°C, 30°C, 35 were chosen to visualize the estate all charts ain) title=x, scale=alt.Scale g_viscosity_mpas', title act of {x}',	concentration, molecule contains only a specific of two values of distribution of the response (domain=[0.9*df[x].	number of defined value of the product conce onse variable within the min(), 1.1* df[x]	garithm of the viscosity is ex alues for the independent var ntration (10 mg/mL, 62.5 mg hese values of the predictor	iables. There exist /mL) and 14 values	only nine values of th
	molecular_weight_k log_viscosity_mp Exploratory data ana In order to visualize the relatice correlation of all three paraminates variables take continuous number temperature (2°C, 5°C, 10°C, weight. Therefore box plots wheight. Therefore box plots where the continuous of the continuous number temperature (2°C, 5°C, 10°C, weight. Therefore box plots wheight. Therefore box plots where the continuous number temperature (2°C, 5°C, 10°C, weight. Therefore box plots wheight. Therefore box plots wheight and continuous number temperature (2°C, 5°C, 10°C, weight. Therefore box plots wheight and continuous number temperature (2°C, 5°C, 10°C, weight. Therefore box plots wheight and continuous number temperature (2°C, 5°C, 10°C, weight. Therefore box plots wheight and continuous number temperature (2°C, 5°C, 10°C, weight. Therefore box plots wheight and continuous number temperature (2°C, 5°C, 10°C, weight. Therefore box plots wheight and continuous number temperature (2°C, 5°C, 10°C, weight. Therefore box plots wheight and continuous number temperature (2°C, 5°C, 10°C, weight. Therefore box plots wheight and continuous number temperature (2°C, 5°C, 10°C, weight. Therefore box plots wheight and continuous number temperature (2°C, 5°C, 10°C, weight. Therefore box plots wheight and continuous number temperature (2°C, 5°C, 10°C, weight. Therefore box plots wheight and continuous number temperature (2°C, 5°C, 10°C, weight and continuous number temperature (2°C, 5°C, 10°C, 5°C, 10°C, weight and continuous number	ionship between the explanameters (temperature, product merical values, the dataset of 15°C, 20°C, 25°C, 30°C, 35 were chosen to visualize the extension of the store all charts ain) title=x, scale=alt.Scaleg_viscosity_mpas', title act of {x}',	concentration, molecule contains only a specific of two values of distribution of the responsibility of the re	number of defined value of the product conce onse variable within the min(), 1.1* df[x]	garithm of the viscosity is explues for the independent variation (10 mg/mL, 62.5 mg/mLese values of the predictor	iables. There exist /mL) and 14 values	only nine values of the of the molecular
577	molecular_weight_k log_viscosity_mp Exploratory data ana In order to visualize the relatic correlation of all three paraminates take continuous number temperature (2°C, 5°C, 10°C) weight. Therefore box plots with the properties of the properties	ionship between the explanameters (temperature, product merical values, the dataset of 1,15°C, 20°C, 25°C, 30°C, 35 were chosen to visualize the extension of the store all charts ain) title=x, scale=alt.Scale g_viscosity_mpas', title act of {x}', ot) concat(*charts)	impact of proc	min(), 1.1* df[x]as')	garithm of the viscosity is explues for the independent variation (10 mg/mL, 62.5 mg/mbese values of the predictor) .max()])), imax()])),	iables. There exist (mL) and 14 values variables.	only nine values of the of the molecular
	molecular_weight_k log_viscosity_mp Exploratory data ana In order to visualize the relatic correlation of all three paramically variables take continuous number temperature (2°C, 5°C, 10°C, weight. Therefore box plots with the second continuous of the second continuous	ionship between the explanameters (temperature, product merical values, the dataset of 1, 15°C, 20°C, 25°C, 30°C, 35 were chosen to visualize the external charts ain) title=x, scale=alt.Scale g_viscosity_mpas', title act of {x}', ot) concat(*charts) temperature	impact of process. impact of process. impact of process. impact of process.	min(), 1.1* df[x]as')	garithm of the viscosity is explues for the independent variation (10 mg/mL, 62.5 mg/mbese values of the predictor) .max()])), imax()])),	iables. There exist (mL) and 14 values variables.	only nine values of the of the molecular velght_kda
	molecular_weight_k log_viscosity_mp Exploratory data ana In order to visualize the relatice correlation of all three paraminate variables take continuous number temperature (2°C, 5°C, 10°C, weight. Therefore box plots where the possible of the possible	ionship between the explanameters (temperature, product merical values, the dataset of 1,15°C, 20°C, 25°C, 30°C, 35 were chosen to visualize the store all charts ain) title=x, scale=alt.Scale g_viscosity_mpas', title act of {x}', temperature temperature	impact of product impact of pro	min(), 1.1* df[x] min(), 1.1* df[x] as') duct_concentration_m concentration_mg_ml	garithm of the viscosity is explues for the independent variation (10 mg/mL, 62.5 mg hese values of the predictor .max()])), imax()])), imax()]) imax()])), imax()])), imax()])), imax()])), imax()])), imax()]) imax()] imax()] imax()]) imax()] imax()] imax()] imax()] imax()] imax()] imax()] imax()] imax()	pact of molecular_v pact of molecular_v 160,000 180,000 molecular_weight	veight_kda veight_kda veight_kda 200,000 220,000
	molecular_weight_k log_viscosity_mp Exploratory data ana In order to visualize the relatic correlation of all three param variables take continuous nurtemperature (2°C, 5°C, 10°C, weight. Therefore box plots volume to the perature (2°C, 5°C, 10°C, weight. Therefore box plots volume to the perature (2°C, 5°C, 10°C, weight. Therefore box plots volume to the perature (2°C, 5°C, 10°C, weight. Therefore box plots volume to the perature (2°C, 5°C, 10°C, weight. Therefore box plots volume to the perature (2°C, 5°C, 10°C, weight. Therefore box plots volume to the perature (2°C, 5°C, 10°C, weight. Therefore box plots volume to the perature (2°C, 5°C, 10°C, weight. Therefore box plots volume to the perature (2°C, 5°C, 10°C, weight. Therefore box plots volume to the perature (2°C, 5°C, 10°C, weight. Therefore box plots volume to the perature (2°C, 5°C, 10°C, weight. Therefore box plots volume to the perature (2°C, 5°C, 10°C, weight. Therefore box plots volume to the perature (2°C, 5°C, 10°C, weight. Therefore box plots volume to the perature (2°C, 5°C, 10°C, weight. Therefore box plots volume to the perature (2°C, 5°C, 10°C, weight. Therefore box plots volume to the perature (2°C, 5°C, 10°C, weight. Therefore box plots volume to the perature (2°C, 5°C, 10°C, weight. Therefore box plots volume to the perature (2°C, 5°C, 10°C, weight. Therefore box plots volume to the perature (2°C, 5°C, 10°C, weight. The perature (2°C, 5°C, 10°C, weight. Therefore box plots volume to the perature (2°C, 5°C, 10°C, weight. The perature (2°C, 5°C, 10°C, weight. The perature (2°C, 5°C, 10°C, weight. Therefore box plots volume to the perature (2°C, 5°C, 10°C, weight. Therefore box plots volume to the perature (2°C, 5°C, 10°C, weight. Therefore box plots volume to the perature (2°C, 5°C, 10°C, weight. Therefore box plots volume to the perature (2°C, 5°C, 10°C, weight. Therefore box plots volume to the perature (2°C, 5°C, 10°C,	ionship between the explanare neters (temperature, product merical values, the dataset of 1,15°C, 20°C, 25°C, 30°C, 35° were chosen to visualize the store all charts ain) title=x, scale=alt.Scale g_viscosity_mpas', title act of {x}', ot) concat(*charts) temperature onship between the logarithm apprature increases, the logarithm apprature increases, the logarithm apprature increases, the logarithm apprature increases, the logarithm approximately including approximately trained including are consistent with the logarithm and including are consistent with the logarithm and including are consistent with the logarithm and	impact of product impact of product impact of product in of the measured visco arithm of the	min(), 1.1* df[x] min(), 1.1* df[x] as') duct_concentration_m psity in mPas and the viscosity decreases, onse variable appearationship between temperature and the viscosity of the viscosity	garithm of the viscosity is explues for the independent variation (10 mg/mL, 62.5 mg) hese values of the predictor in max()])), max()])), sequence of the predictor in max()])), temperature in °C is shown. indicating a negative associate to be symmetric. This suggingerature and viscosity (without a fluid decreases exponential in max() in the max() in t	pact of molecular_v A clear negative traction between the traction between the relation traction as temperature.	veight_kda veight_kda veight_kda end is evident from the wo variables. Although onship between the two variables and the veight of the vold follow and the verses [6].
	molecular_weight_k log_viscosity_mp Exploratory data ana In order to visualize the relatice correlation of all three paramination variables take continuous nuttemperature (2°C, 5°C, 10°C, weight. Therefore box plots where the continuous numbers are the continuous	ionship between the explanameters (temperature, product merical values, the dataset of 1,15°C, 20°C, 25°C, 30°C, 35° were chosen to visualize the extension of the store all charts ain) title=x, scale=alt.Scale g_viscosity_mpas', title act of {x}', temperature onship between the logarithm apperature increases, the logarithm appearance in the logarit	impact of product of the measured viscolarithm of the measured viscolarith	min(), 1.1* df[x] min(), 1.1* df[x] min(), 1.1* df[x] as') duct_concentration_m psity in mPas and the viscosity decreases, onse variable appearationship between temperature that the viscosity decreases are the transfer of the response variable appearation are the response variable appearation are the response variable appearations are the response variable appea	garithm of the viscosity is explues for the independent varintration (10 mg/mL, 62.5 mg hese values of the predictor shown indicating a negative associate to be symmetric. This suggester and viscosity (without a fluid decreases exponent mPas) and the product condent between the logarithm of the appears to be symmetric. The conversely, the relationship in the conversely in the relationship in the conversely.	pact of molecular_veight A clear negative tration between the tration tially as temperature the viscosity and the This symmetry sugo with the untransform	veight_kda veight_kda veight_kda end is evident from the wo variables. Although onship between the two vold follow and re rises [6]. L). Although only two expressions are rises reproduct concentrating gests that the permed viscosity would to the vold viscosity would reconcentrate the permed viscosity would reconcentrat
	molecular_weight_k log_viscosity_mp Exploratory data ana In order to visualize the relatice correlation of all three paramy variables take continuous nuttemperature (2°C, 5°C, 10°C, weight. Therefore box plots weight. Chart(df_tra	ionship between the explanameters (temperature, product merical values, the dataset of 1,15°C, 20°C, 25°C, 30°C, 35 were chosen to visualize the external charts ain) title=x, scale=alt.Scale g_viscosity_mpas', title act of {x}', ot) concat(*charts) temperature onship between the logarithm inperature increases, the logarithm inperature increases, the logarithm increases, the logarithm increases increases, the logarithm increases increase	impact of product in of the measured viscos arithm of the	min(), 1.1* df[x] min(), 1.1* df[x] min(), 1.1* df[x] as') duct_concentration_m posity in mPas and the viscosity decreases, onse variable appearationship between terms that the viscosity (interpretation of the response variable appearation of the response variable	garithm of the viscosity is explues for the independent variation (10 mg/mL, 62.5 mg hese values of the predictor leaves are to be symmetric. This suggester to be symmetric. This suggester to be symmetric. This suggester and viscosity (without a fluid decreases exponent mPas) and the product concern between the logarithm of the appearance of the product concern the symmetric. The symmetric of protein solutions increased the concern that is a larger cluster between 2000 and a la	pact of molecular_v pact of molecular_v tion between the transfer symmetry sught the viscosity and the rise symmetry sught with the untransfer se exponentially with the untransfer se exponentially with the relation of the control of the contr	veight_kda veight_kda veight_kda end is evident from the wo variables. Although onship between the two variables are rises [6]. L). Although only two e product concentrating gests that the product concentrating the rising protein iraph, is less clear. It the molecular weigh 000 kDa. The
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Draft analysis

Name: Samuel Hempelt

Introduction

Diabetic retinopathy is a serious illness, which is expected to affect > 200 million people by the year 2025 [1]. It is an eye desease resulting in blindness for over 10000

new active ingredients, which has the potential to slow or even stop the progression of this desease [3]. A unique characteristic of these medications is the intravitreal

application, which means that the drug product is injected directly into the vitreous humor, the gel-like substance inside the eye (see picture below).

,Suspensory

people with diabetes per year [2]. In order to help these patients Boehringer Ingelheim is investing in Research and Development of biopharmaceuticals and is screening for

In [587... #Retransformation of the prediction y_test_exp = np.exp(y_test) y_pred_exp = np.exp(y_pred) In [588... mse = mean_squared_error(y_test_exp, y_pred_exp) print("Mean Squared Error after reverse log transformation:", mse) Mean Squared Error after reverse log transformation: 1.8713359627143917 Save model Save your model in the folder models/ . Use a meaningful name and a timestamp. In [589... # timestamp and name now = datetime.now() timestamp = now.strftime("%Y%m%d%H%M%S") model_name = timestamp + "_viscosity_model.pkl" # give out target directory repo_directory = subprocess.check_output(["git", "rev-parse", "--show-toplevel"]).strip().decode() directory = repo_directory + "/models" # complete path for file model_path = os.path.join(directory, model_name) # save model with open(model_path, "wb") as file: pickle.dump(reg, file) print(f"Model was saved in {model_path} ") Model was saved in /Users/snowwhite/Desktop/DataAnalyticswithStatistics/Project/project/models/20241216163517_viscosity_model.pkl Conclusions The model demonstrates a relatively low error (MSE = 1.87), suggesting reasonable prediction accuracy. This corresponds to an average deviation of about $\sqrt{1.87} \approx 1.37$ (in the same units as the target variable) between predicted and actual values. However, the adequacy of this error depends largely on the target variable's scale. With the target variable ranging from approximately 0.54 to 27.87, an average error of 1.37 is relatively small. Notably, in some cases, the difference between technical replicates (measurements under identical conditions) exceeds the model's average error, highlighting its potential utility. To enhance the model further, additional variables in the dataset, such as the extinction coefficient, isoelectric point, or protein format of the molecule, should be investigated and incorporated into the regression model where relevant. Bibliography [1] Karin S Coyne, Mary Kay Margolis, Tessa Kennedy-Martin, Timothy M Baker, Ronald Klein, Matthew D Paul, Dennis A Revicki, The impact of diabetic retinopathy: perspectives from patient focus groups, Family Practice, Volume 21, Issue 4, August 2004, Pages 447–453. [2] Donald S. Fong, Frederick L. Rerres, Lloyd P. Aiello, Ronald Klein, Diabetic Retinopathy, Diabetes Care, Volume 27, Number 10, October 2004 [3] Produkt Portfolio Boehringer Ingelheim, 2023, https://unternehmensbericht.boehringer-ingelheim.de/2023/download/BOE_GB23_Produktportfolio_DE_safe.pdf [4] Parenky AC, Wadhwa S, Chen HH, Bhalla AS, Graham KS, Shameem M. Container Closure and Delivery Considerations for Intravitreal Drug Administration. AAPS PharmSciTech. 2021 Mar 11;22(3):100. doi: 10.1208/s12249-021-01949-4. PMID: 33709236; PMCID: PMC7952281. [5] European Directorate for the Quality of Medicines & HealthCare. European Pharmacopoeia. 10th ed., Council of Europe, 2020. www.edqm.eu/en/europeanpharmacopoeia-pharmacopoeia-europe. [6] Arrhenius S. The Viscosity of Solutions. Biochem J. 1917 Aug;11(2):112-33. doi: 10.1042/bj0110112. PMID: 16742728; PMCID: PMC1258811. [7] Wozniak, Spencer, and Michael Feig. "Diffusion and Viscosity in Mixed Protein Solutions." The Journal of Physical Chemistry B, vol. 128, no. 47, 2024 [8] Woldeyes, M. A., Josephson, L. L., Leiske, D. L., Galush, W. J., Roberts, C. J., & Furst, E. M. (2018). Viscosities and protein interactions of bispecific antibodies and their monospecific mixtures. Molecular Pharmaceutics, 15(10), 4252-4261.