

# bcc Lattice and Primitive Unit Cell

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## 1 bcc Lattice (beware of tensors)

The body-centered cubic (BCC) lattice is defined by a conventional cubic unit cell with lattice constant  $a$ . The basis vectors of the conventional unit cell in Cartesian coordinates are:

$$\mathbf{a}_1 = a(1, 0, 0), \quad \mathbf{a}_2 = a(0, 1, 0), \quad \mathbf{a}_3 = a(0, 0, 1) \quad (1)$$

In tensor notation, these vectors are contravariant, expressed as  $\mathbf{a}_i = a_i^\mu \mathbf{e}_\mu$ , where  $\mathbf{e}_\mu$  ( $\mu = 1, 2, 3$ ) are the Cartesian basis vectors, and:

$$a_i^\mu = a\delta_i^\mu \quad (2)$$

where  $\delta_i^\mu$  is the Kronecker delta ( $\delta_i^\mu = 1$  if  $i = \mu$ , else 0).

The BCC crystal structure has a basis of two atoms per conventional unit cell, located at:

$$\mathbf{r}_1 = (0, 0, 0), \quad \mathbf{r}_2 = \frac{a}{2}(1, 1, 1) \quad (3)$$

In tensor form:

$$r_1^\mu = (0, 0, 0), \quad r_2^\mu = \frac{a}{2}(1, 1, 1) \quad (4)$$

The position of any atom in the crystal is:

$$\mathbf{R} = n^i \mathbf{a}_i + \mathbf{r}_j, \quad R^\mu = n^i a_i^\mu + r_j^\mu \quad (5)$$

where  $n^i \in \mathbb{Z}$  and  $j = 1, 2$ . Substituting:

$$R^\mu = n^i a\delta_i^\mu + r_j^\mu = a(n^1\delta_1^\mu + n^2\delta_2^\mu + n^3\delta_3^\mu) + r_j^\mu \quad (6)$$

For  $j = 1$ :

$$R^\mu = a(n^1, n^2, n^3) \quad (7)$$

For  $j = 2$ :

$$R^\mu = a\left(n^1 + \frac{1}{2}, n^2 + \frac{1}{2}, n^3 + \frac{1}{2}\right) \quad (8)$$

This generates all BCC lattice positions.

## 2 Metric Tensor and Volume of Conventional Unit Cell

The metric tensor is:

$$g_{ij} = \mathbf{a}_i \cdot \mathbf{a}_j = a_i^\mu a_j^\nu \delta_{\mu\nu} \quad (9)$$

Since  $\mathbf{a}_i$  are orthogonal:

$$g_{ij} = a^2 \delta_{ij} \quad (10)$$

The volume is given by the scalar triple product:

$$V = |\mathbf{a}_1 \cdot (\mathbf{a}_2 \times \mathbf{a}_3)| = |\epsilon^{\mu\nu\lambda} a_{1\mu} a_{2\nu} a_{3\lambda}| \quad (11)$$

Substituting:

$$V = \epsilon^{\mu\nu\lambda} (a\delta_{1\mu})(a\delta_{2\nu})(a\delta_{3\lambda}) = a^3 \epsilon^{123} = a^3 \quad (12)$$

The volume is  $V = a^3$ .

## 3 Primitive Unit Cell

The primitive unit cell contains one lattice point. Suitable primitive vectors are:

$$\mathbf{b}_1 = \frac{a}{2}(1, 1, -1), \quad \mathbf{b}_2 = \frac{a}{2}(-1, 1, 1), \quad \mathbf{b}_3 = \frac{a}{2}(1, -1, 1) \quad (13)$$

In tensor notation:

$$b_i^\mu = \frac{a}{2} \begin{cases} (1, 1, -1) & i = 1 \\ (-1, 1, 1) & i = 2 \\ (1, -1, 1) & i = 3 \end{cases} \quad (14)$$

Lattice points are generated by:

$$\mathbf{R}^\mu = m^i b_i^\mu, \quad m^i \in \mathbb{Z} \quad (15)$$

## 4 Metric Tensor for Primitive Unit Cell

The metric tensor is:

$$g_{ij} = \mathbf{b}_i \cdot \mathbf{b}_j = b_i^\mu b_j^\nu \delta_{\mu\nu} \quad (16)$$

Computing:

$$g_{11} = \frac{a^2}{4}(1^2 + 1^2 + (-1)^2) = \frac{3a^2}{4} \quad (17)$$

$$g_{12} = \frac{a^2}{4}[(1)(-1) + (1)(1) + (-1)(1)] = -\frac{a^2}{4} \quad (18)$$

$$g_{13} = \frac{a^2}{4}[(1)(1) + (1)(-1) + (-1)(1)] = -\frac{a^2}{4} \quad (19)$$

$$g_{22} = \frac{a^2}{4}[(-1)^2 + 1^2 + 1^2] = \frac{3a^2}{4} \quad (20)$$

$$g_{23} = \frac{a^2}{4}[(-1)(1) + (1)(-1) + (1)(1)] = -\frac{a^2}{4} \quad (21)$$

$$g_{33} = \frac{a^2}{4}[1^2 + (-1)^2 + 1^2] = \frac{3a^2}{4} \quad (22)$$

Thus:

$$g_{ij} = \frac{a^2}{4} \begin{pmatrix} 3 & -1 & -1 \\ -1 & 3 & -1 \\ -1 & -1 & 3 \end{pmatrix} \quad (23)$$

## 5 Volume of Primitive Unit Cell

The volume is:

$$V = |\mathbf{b}_1 \cdot (\mathbf{b}_2 \times \mathbf{b}_3)| = |\varepsilon^{\mu\nu\lambda} b_{1\mu} b_{2\nu} b_{3\lambda}| \quad (24)$$

Compute:

$$\mathbf{b}_2 \times \mathbf{b}_3 = \frac{a^2}{4} \begin{vmatrix} \hat{\mathbf{e}}_1 & \hat{\mathbf{e}}_2 & \hat{\mathbf{e}}_3 \\ -1 & 1 & 1 \\ 1 & -1 & 1 \end{vmatrix} = \frac{a^2}{4} [2\hat{\mathbf{e}}_1 + 2\hat{\mathbf{e}}_2 + 0\hat{\mathbf{e}}_3] = \frac{a^2}{2} (1, 1, 0) \quad (25)$$

$$V = \left| \frac{a}{2} (1, 1, -1) \cdot \frac{a^2}{2} (1, 1, 0) \right| = \frac{a^3}{4} |1 \cdot 1 + 1 \cdot 1 + (-1) \cdot 0| = \frac{a^3}{4} \cdot 2 = \frac{a^3}{2} \quad (26)$$

The volume is  $\frac{a^3}{2}$ , half the conventional cell volume, confirming one lattice point.

## 6 Basis in Primitive Unit Cell

The primitive unit cell has one lattice point with a basis of one atom at:

$$\mathbf{r}^\mu = (0, 0, 0) \quad (27)$$

Positions are:

$$\mathbf{R}^\mu = m^i \mathbf{b}_i^\mu \quad (28)$$

## 7 Transformation Between Conventional and Primitive Cells

Express primitive vectors as:

$$\mathbf{b}_1 = \frac{1}{2}(\mathbf{a}_1 + \mathbf{a}_2 - \mathbf{a}_3) \quad (29)$$

$$\mathbf{b}_2 = \frac{1}{2}(-\mathbf{a}_1 + \mathbf{a}_2 + \mathbf{a}_3) \quad (30)$$

$$\mathbf{b}_3 = \frac{1}{2}(\mathbf{a}_1 - \mathbf{a}_2 + \mathbf{a}_3) \quad (31)$$

In tensor form:

$$b_i^\mu = T_i^j a_j^\mu, \quad T_i^j = \frac{1}{2} \begin{pmatrix} 1 & 1 & -1 \\ -1 & 1 & 1 \\ 1 & -1 & 1 \end{pmatrix} \quad (32)$$

The volume ratio is:

$$\det(T) = \frac{1}{8} \begin{vmatrix} 1 & 1 & -1 \\ -1 & 1 & 1 \\ 1 & -1 & 1 \end{vmatrix} = \frac{1}{8} [1(1 \cdot 1 - 1 \cdot (-1)) - 1((-1) \cdot 1 - 1 \cdot 1) + (-1)((-1) \cdot (-1) - 1 \cdot 1)] = \frac{1}{2} \quad (33)$$

Thus:

$$V_{\text{primitive}} = |\det(T)| V_{\text{conventional}} = \frac{1}{2} \cdot a^3 = \frac{a^3}{2} \quad (34)$$

## 8 Conclusion

The BCC lattice is described by:

- **Conventional unit cell:** Basis vectors  $a_i^\mu = a\delta_i^\mu$ , two-atom basis at  $r_1^\mu = (0,0,0)$ ,  $r_2^\mu = \frac{a}{2}(1,1,1)$ . Volume  $a^3$ , metric tensor  $g_{ij} = a^2\delta_{ij}$ .
- **Primitive unit cell:** Basis vectors  $b_1^\mu = \frac{a}{2}(1,1,-1)$ ,  $b_2^\mu = \frac{a}{2}(-1,1,1)$ ,  $b_3^\mu = \frac{a}{2}(1,-1,1)$ , single-atom basis at  $r^\mu = (0,0,0)$ . Volume  $\frac{a^3}{2}$ , metric tensor  $g_{ij} = \frac{a^2}{4} \begin{pmatrix} 3 & -1 & -1 \\ -1 & 3 & -1 \\ -1 & -1 & 3 \end{pmatrix}$ .

The transformation  $T_i^j$  and volume calculations prove the primitive cell contains one lattice point, consistent with the BCC structure.