

Mechanical Design 444 System Simulation Notes

Rock Bed Heat Storage

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List of symbols

Variables	v volume [m ³]
a surface area [m^2]	<i>W</i> width [m]
A area $[m^2]$ c_p specific heat capacity $[J/(kg \cdot K)]$ d diameter $[m]$ G mass flow rate per unit area $[kg/(s \cdot m^2)]$ h_v volumetric heat transfer coefficient $[W/(m^3 \cdot K)]$ H height $[m]$ k thermal conductivity $[W/(m \cdot K)]$ L length $[m]$ m air mass flow rate $[kg/s]$ Nu Nusselt number $[-]$ Pr Prandtl number $[-]$ q heat transfer $[W/s]$ Q air volume flow rate $[m^3/s]$	f friction factor [—] α thermal diffusivity [m²/s] η loss factor [—] ε void ratio [—] ψ sphericity [—] ρ density [kg/m³] μ dynamic viscosity [kg/(m·s)] ν kinematic viscosity [m²/s] ζ temporary variable Subscripts b packed bed d dimensionless parameter properties based on particle diameter
Re Reynolds number [—]	r air properties through the rock bed heat
T temperature [K]	store
<i>u</i> velocity [m/s]	p particle element

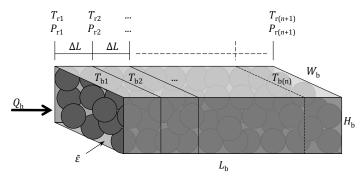


Figure 1: Packed bed heat store

1. Packed bed material properties

Consider the packed bed heat store shown in figure 1. It is filled with spherical stones. Define the average *void fraction* $\bar{\epsilon}$ as

$$\bar{\varepsilon} = \frac{\text{volume of empty space}}{\text{volume of solid media}} \tag{1}$$

Assume that the flow is uniform and one-dimensional through the bed. The projected flow area $A_{\rm b}$ perpendicular to the flow direction is

$$A_{\rm b} = W_{\rm b} H_{\rm b} \tag{2}$$

with W_b and H_b the bed width and height.

Table 1: Fractions of air and solids in packed bed

	Air	Solids
Fraction Volume Mass	$ \frac{\bar{\varepsilon}}{\bar{\varepsilon}} A_{\rm b} L_{\rm b} \bar{\varepsilon} A_{\rm b} L_{\rm b} \rho_{\rm r} $	$ \frac{(1-\bar{\varepsilon})}{(1-\bar{\varepsilon})A_{b}L_{b}} (1-\bar{\varepsilon})A_{b}L_{b}\rho_{p} $

The different fractions of solids and air are given in table 1. Figure 2 gives the average void fraction for same-sized spherical particles in cylindrical container.

Let v_p be the volume and a_p the surface area of a particle element in the bed. For none spherical particle the *equivalent particle diameter* is defined as

$$d_{\rm p} = \left(\frac{6}{\pi}v_{\rm p}\right)^{1/3} \tag{3}$$

The *sphericity* of the particle ψ is defined as the ratio of the particle surface area to that of a sphere with the same volume

$$\psi = \frac{a_{\rm p}}{\pi d_{\rm p}^2} \tag{4}$$

Define the the *superficial air velocity* u'_r in the bed as the flow velocity without any obstacles in its path

$$u_{\rm r}' = \frac{Q_{\rm r}}{A_{\rm b}} \tag{5}$$

with Q_r the volume flow rate. The mass flow rate of air per unit area, G_r , through the bed is

$$G_{\rm r} = \rho_{\rm r} u_{\rm r}' = \frac{\rho_{\rm r} Q_{\rm r}}{A_{\rm b}} \tag{6}$$

with $\rho_{\rm r}$ the air density and $\mu_{\rm r}$ the air kinematic viscosity in the bed.

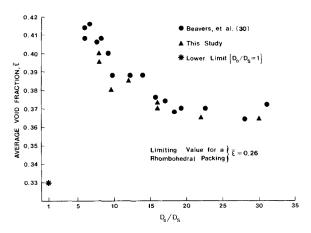


Figure 2: Average void fraction as a function of container diameter to particle diameter for uniform spheres. (Beasley and Clark, 1984)

Define the dimensionless groups based on the particle diameter $d_{\rm p}$

$$Re_{\rm d} = \frac{G_{\rm r}d_{\rm p}}{\mu_{\rm r}}$$
 Reynolds number (7)

$$Nu_{\rm d} = \frac{h_{\rm v}' d_{\rm p}^2}{k_{\rm r}}$$
 Nusselt number (8)

$$f_{\rm d} = \frac{\Delta P}{\Delta L} \frac{\rho_{\rm r} d_{\rm p}}{G_{\rm r}^2} \qquad \text{friction factor}$$
 (9)

Singh *et al.* (2006) performed a wide range of experiments for heat transfer and pressure loss through packed beds and found the following empirical relationships for the Nusselt number and friction factor

$$Nu_{\rm d} = 0.437 \, (Re_{\rm d})^{0.75} \, (\psi)^{3.35} \, (\bar{\epsilon})^{-1.62} \, \exp[29.03 \, (\ln \psi)^2]$$
 (10)

$$f_{\rm d} = 4.466 \, (Re_{\rm d})^{-0.2} \, (\psi)^{0.696} \, (\bar{\varepsilon})^{-2.945} \, \exp[11.85 \, (\ln \psi)^2] \tag{11}$$

The equations above holds for sphericity: $0.55 \le \psi \le 1.00$, void fraction: $0.306 \le \bar{\varepsilon} \le 0.63$ and mass velocity: $0.155 \, \text{kg/(s·m}^2) \le G_r \le 0.266 \, \text{kg/(s·m}^2)$.

2. Heat transfer between the bed and the air

The heat transfer from the air to the bed in control volume i is

$$q_{b(i)} = \dot{m} c_{p_r} (T_{r(i)} - T_{r(i+1)}) = h'_v A_b \Delta L \Delta T_{eff(i)}$$
(12)

with $T_{r(i)}$ and $T_{r(i+1)}$ the average in and outlet air temperatures of control volume i. The logarithmic effective temperature difference is

$$\Delta T_{\text{eff}(i)} = \frac{T_{r(i)} - T_{r(i+1)}}{\ln\left(\frac{T_{r(i)} - T_{b(i)}}{T_{r(i+1)} - T_{b(i)}}\right)} \tag{13}$$

The mass flow rate through the control volume is

$$\dot{m} = \rho_{\rm r} Q_{\rm r} = G_{\rm r} A_{\rm h} \tag{14}$$

The outlet temperature $T_{r(i+1)}$ in equations (12) and (13) is unknown. Solve for $T_{r(i+1)}$, then

$$T_{r(i+1)} = T_{b(i)} - \left(T_{b(i)} - T_{r(i)}\right) \exp\left(-\frac{h_{v}' \Delta L}{c_{p_{r}} G_{r}}\right)$$
(15)

3. Temperature change in the bed

The total mass of particles in control volume i is

$$m_{\rm b} = \rho_{\rm p}(1 - \bar{\varepsilon}) A_{\rm b} \Delta L \tag{16}$$

Assume that there is no temperature gradient inside the individual particles and that all the particle in a control volume is at the average bed temperature, $T_{b(i)}$. The temperature change rate of the particles is then

$$m_{\rm b}c_{p_{\rm p}}\frac{dT_{\rm b(i)}}{dt} = q_{\rm b(i)} = \dot{m}\,c_{p_{\rm r}}\left(T_{\rm r(i)} - T_{\rm r(i+1)}\right) \tag{17}$$

or

$$\frac{dT_{b(i)}}{dt} = \frac{c_{p_r}G_r}{\rho_p c_{p_p}(1 - \bar{\epsilon})\Delta L} (T_{r(i)} - T_{r(i+1)})$$
(18)

4. Calculation procedure

With reference to figure 1, divide the packed bed in n control volumes with lengths $\Delta L = L_{\rm b}/n$. Let $T_{\rm r}(i)$ be the inlet air temperature for control volume i. For every control volume $i=1,2\ldots,n$ and $T_{\rm r}=T_{\rm s}=1$

(a) Calculate average temperature over control volume

$$\bar{T}_i = \frac{1}{2}(T_{r(i)} + T_{r(i+1)}^{\text{old}})$$

(b) Calculate average air and flow properties in control volume

$$\rho_{i} = \rho_{\text{air}}(\bar{T}_{i})$$

$$\mu_{i} = \mu_{\text{air}}(\bar{T}_{i})$$

$$c_{p_{i}} = c_{p_{\text{air}}}(\bar{T}_{i})$$

$$k_{i} = k_{\text{air}}(\bar{T}_{i})$$

$$G_{i} = \rho_{i}Q_{r}/A_{b}$$

(c) Calculate parameters

$$Re_{i} = G_{i}d_{p}/\mu_{i}$$

$$f_{i} = 4.466 (Re_{i})^{-0.2} (\psi)^{0.696} (\bar{\varepsilon})^{-2.945} \exp[11.85 (\ln \psi)^{2}]$$

$$Nu_{i} = 0.437 (Re_{i})^{0.75} (\psi)^{3.35} (\bar{\varepsilon})^{-1.62} \exp[29.03 (\ln \psi)^{2}]$$

$$h'_{vi} = k_{i} Nu_{i}/d_{p}^{2}$$

(d) Calculate air temperature change and pressure drop

$$\begin{split} T_{\mathrm{r}(i+1)} &= T_{\mathrm{b}(i)} - \left(T_{\mathrm{b}(i)} - T_{\mathrm{r}(i)}\right) \exp\left(-\frac{h_{\mathrm{v}}' \Delta L}{c_{p_{\mathrm{r}}} G_{\mathrm{r}}}\right) \\ \Delta P &= -\mathfrak{f}_{i} \frac{G_{i}^{2} \Delta L}{\rho_{i} d_{\mathrm{p}}} \end{split}$$

(e) Calculate rate of change of rock bed temperature

$$\frac{dT_{\mathrm{b}(i)}}{dt} = \frac{c_{p_i}G_i(T_{\mathrm{r}(i)} - T_{\mathrm{r}(i+1)})}{\rho_{\mathrm{p}}c_{p_{\mathrm{p}}}(1 - \bar{\varepsilon})\Delta L}$$

Summate the pressure loss across all the control volumes and add an additional loss factor η_r for the total heat store system. The total pressure loss is then

$$\Delta P_{\rm r} = \eta_{\rm r} \sum_{i=1}^{n} \Delta P_{i} \tag{19}$$

5. Simulation input values

For preliminary simulation purpose use the following input values for rock

 $\begin{array}{ll} \text{Bed void fraction} & \bar{\varepsilon} &= 0.45 \\ \text{Particle diameter} & d_{\text{p}} &= 10 \text{ cm} \\ \text{Particle sphericity} & \psi &= 1.00 \\ \text{Particle density} & \rho_{\text{p}} &= 2640 \text{ kg/m}^3 \\ \text{Particle heat capacity} & c_{p_{\text{p}}} &= 820 \text{ J/(kg·K)} \end{array}$

Pressure loss factor $\eta_r = 1.2$

References

Beasley, D.E. and Clark, J.A. (1984). Transient response of a packed bed for thermal energy storage. *International Journal of Heat and Mass Transfer*, vol. 21, no. 9, pp. 1659–1669.

Singh, R., Saini, R.P. and Saini, J.S. (2006). Nusselt number and friction factor correlations for packed bed solar energy storage system having large sized elements of different shapes. *Solar Energy*, vol. 80, pp. 760–771.