

## Homework Assignment 1

**Due @ 11:59pm Monday 23th February by email [serdal@nyu.edu](mailto:serdal@nyu.edu) or to ERB 114**



**This sign means python or any kind of scientific computing is needed to get the answer. Please provide your answer with your python code.**

1. A classical liquid of  $N$  molecules of Oxygen and  $N$  molecules of Chlorine molecules are in a closed container. Each interacting with the rest with a pairwise Lennard-Jones potential which is given as:

$$V^{LJ}(r_{ij}) = 4\epsilon_{ij} \left[ \left( \frac{\sigma_{ij}}{r_{ij}} \right)^{12} - \left( \frac{\sigma_{ij}}{r_{ij}} \right)^6 \right]$$

where  $i$  and  $j$  are particle indices and  $r_{ij}$  is the distance in between,  $(\epsilon, \sigma)$  are van der Waals parameters that accounts for the polarizability of the particle and effective size respectively. When two different type of particles interact their  $(\epsilon, \sigma)$  is computed by Lorents-Bertholet mixing rule. To compute the  $(\epsilon, \sigma)$  parameter for particle  $i$  and  $j$  of different kind we use the following relations:

$$\sigma_{ij} = \frac{1}{2}[\sigma_i + \sigma_j] \quad \text{and} \quad \epsilon_{ij} = [\epsilon_i \epsilon_j]^{1/2}.$$

	$\epsilon / k_B$	$\sigma (nm)$
Oxygen	61.6	0.295
Chlorine	173.5	0.335

Table1. van der Walls parameters of oxygen and chlorine molecules, epsilon is in units of Boltzmann constant.

- a. Plot the potential energy as a function of distance ( $2.85 < r < 10$  angstrom) between two oxygen molecules, two chlorine molecules and one oxygen and one chlorine using the values in table 1. Discuss the difference between the three.
- b. Find the point where  $V^{LJ}$  is at minimum for oxygen analytically and compute the force at that point for oxygen-oxygen Lennard-Jones interaction.
- c. Write down the Newtons' equation of motion explicitly for one oxygen molecule due to the  $2N-1$  molecules of oxygen and chlorine.
- d. The positions of atoms for  $N=100$  particles is given in the file "O2Cl2.txt". Calculate the total force on particle 7 and 25. Please provide your answer with your python code.

- e. Assuming that each force calculations for one pairwise interaction is six floating point operation. How long does it take to compute the forces for  $N=10^{23}$  particles if all BuTinah supercomputer is allocated for your research. How long would it take if you have only  $10^4$  particles? Please as a unit of time use days.


2. The potential energy of a particle in a single well is given as  $U_0(x^2 - 2)$  where, the strength of the potential is  $U_0=2.0$ ,  $m=1$ . Everything is dimensionless so don't worry about the units.

- a. Show that the equations of motion for the particle by using Newton's second Law for a general  $x_0$  and  $v_0$  is  $x(t) = x_0 \cos \omega t + (v_0 / \omega) \sin \omega t$  where  $\omega^2 = (U_0 / m)$  and also derive the equation for  $p(t)$ . (Hint. Find the general solution of  $\frac{d^2 x}{dt^2} = -\omega^2 x$  you will have two initial conditions to satisfy)
- b. Numerical solution of ordinary differential equations is achieved by finite difference methods. The idea is using the information at an earlier step  $x(0), p(0)$  to solve the equations at a later time step  $\Delta t$  and obtain  $x(\Delta t), p(\Delta t)$ . Here  $\Delta t$  must be a very small number to get a correct solution. If you need the positions at time  $t$  then you need to repeat the calculation for  $n=t / \Delta t$  times to get the answer. Using Taylor series expansion one can write

$$x(t + \Delta t) = x(t) + \Delta t v(t) + \frac{1}{2} \Delta t^2 a(t) + \frac{1}{6} \Delta t^3 b(t) + \dots$$

if we truncate the function after the second order in  $\Delta t$  then the numerical errors will be in the order of  $\approx O(\Delta t^2)$ . Write a code and compute the position at  $x(\Delta t)$  each of the time steps  $\Delta t = \{10^{-4}, 10^{-3}, 10^{-2}, 10^{-1}, 1\}$ . Compare your result with the exact

result by  $Error(\Delta t) = \frac{x(\Delta t)^{SIM} - x(\Delta t)^{EXACT}}{\Delta t^2}$ . Plot Error versus  $\Delta t$ . Do

you see a linear dependence? As initial conditions use  $p(0)=2$ ,  $x(0)=0$ . 

- c. Repeat the same calculation for a widely used algorithm in molecular dynamics simulations called velocity Verlet and compare your result for  $\Delta t = \{10^{-4}, 10^{-3}, 10^{-2}, 10^{-1}, 1\}$  with the exact solution just like as in b.

$$x(\Delta t) = x(0) + \Delta t \frac{p(0)}{m} + \frac{\Delta t^2}{2m} F(x(0))$$

$$p(\Delta t) = p(0) + \frac{\Delta t}{2} [F(x(0)) + F(x(\Delta t))]$$



- d. Repeat velocity Verlet algorithm for  $n=1000$  steps every time using the position and momentum from the one previous step as your initial phase space points and plot Hamiltonian as a function of time. Repeat this for step sizes



$\Delta t = \{10^{-4}, 10^{-3}, 10^{-2}, 10^{-1}\}$ . Also plot  $\frac{\delta H}{\delta t}(t)$  versus time for each time step sizes. What is the optimum step size to use? And explain why? (hint: to compute  $\frac{\delta H}{\delta t}$  you can use the finite difference :

$\frac{\delta H}{\delta t} \equiv \frac{H(t + \Delta t) - H(t)}{\Delta t}$ ). Plot them to the same graph with different colors and label x and y axis accordingly.

- e. Using the phase space points recorded for initial condition  $p(0)=2$  and  $x(0)=0$  plot the accessible phase space during your numerical simulations of  $n=10000$  steps. Repeat the same for another initial conditions,  $p(0)=4$  and  $x(0)=1$ . Are they the same? Plot them to the same graph with different colors and label the x and y axis accordingly.

