

bookdown+: Authoring Articles, Mails,  
Guitar Chords, Chemical Molecular  
Formulae and Equations with R  
bookdown

*Peng Zhao*

*2017-05-09*



# Contents

|          |                               |          |
|----------|-------------------------------|----------|
| <b>1</b> | <b>Introduction</b>           | <b>5</b> |
| <b>2</b> | <b>Demos</b>                  | <b>7</b> |
| 2.1      | Chemical formulae . . . . .   | 7        |
| 2.2      | Chemical equations . . . . .  | 7        |
| 2.3      | Structural formulae . . . . . | 8        |



# Chapter 1

## Introduction



# Chapter 2

## Demos

### 2.1 Chemical formulae

$\text{H}_2\text{O}$ ,  $\text{Sb}_2\text{O}_3$

$\text{H}^+$ ,  $\text{CrO}_4^{2-}$ ,  $[\text{AgCl}_2]^-$ ,  $\text{Y}^{99+}$ ,  $\text{Y}^{99+}$

$\text{Fe}^{\text{II}}\text{Fe}^{\text{III}}_2\text{O}_4$ ,  $2\text{H}_2\text{O}$ ,  $2\text{H}_2\text{O}$ ,  $0.5\text{H}_2\text{O}$ ,  $\frac{1}{2}\text{H}_2\text{O}$ ,  $(\frac{1}{2})\text{H}_2\text{O}$ ,  $n\text{H}_2\text{O}$

$^{227}_{90}\text{Th}^+$ ,  $^{227}_{90}\text{Th}^+$ ,  $^0_{-1}\text{n}^-$ ,  $^0_{-1}\text{n}^-$

$\text{H}^3\text{HO}$ ,  $\text{H}^3\text{HO}$

$(\text{NH}_4)_2\text{S}$ ,  $[\{(\text{X}_2)_3\}_2]^{3+}$

$\text{H}_2(\text{aq})$ ,  $\text{CO}_3^{2-}(\text{aq})$ ,  $\text{NaOH}(\text{aq}, \infty)$

$\text{OCO}^-$ ,  $\text{NO}^{(2)-}$

$\text{NO}_x$ ,  $\text{Fe}^{n+}$ ,  $\text{Fe}^{n+}$

$\text{Pt}(\eta^2\text{-C}_2\text{H}_4)_2\text{Cl}_3$

$\text{KCr}(\text{SO}_4)_2 \cdot 12\text{H}_2\text{O}$ ,  $\text{KCr}(\text{SO}_4)_{2.12}\text{H}_2\text{O}$ ,  $\text{KCr}(\text{SO}_4)_2 * 12\text{H}_2\text{O}$

$\text{C}_6\text{H}_5\text{-CHO}$ ,  $\text{A-B=C=D}$ ,  $\text{A-B=C=D}$

$\text{A-B=C=D}$ ,  $\text{A-B-C}$ ,  $\text{A=B=C=D}$ ,  $\text{A}\cdots\text{B}\cdots\text{C}$ ,  $\text{A}\rightarrow\text{B}\leftarrow\text{C}$

### 2.2 Chemical equations

$\text{A} \longrightarrow \text{B}$

$\text{A} \longleftarrow \text{B}$

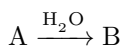
$\text{A} \longleftrightarrow \text{B}$

$$A \leftarrow - > B$$

$$A \rightleftharpoons B$$

$$A \rightleftharpoons B$$

$$A \rightleftharpoons B$$



$$A \xrightarrow[\text{textbelow}]{\text{textabove}} B$$

$$A \xrightarrow[x_i]{x} B$$

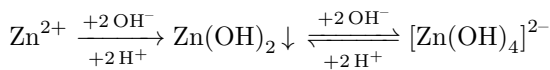
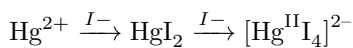
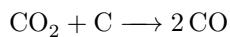
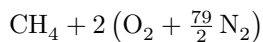
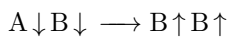
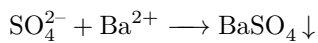
$$A \xrightarrow{x} B$$

$$A + B$$

$$A - B$$

$$A = B$$

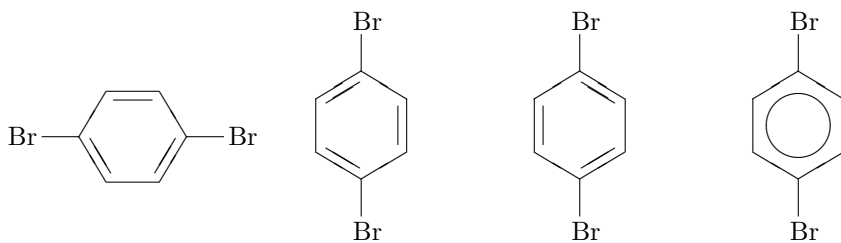
$$A \pm B$$



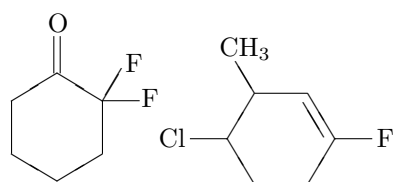
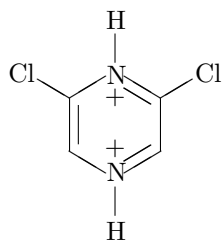
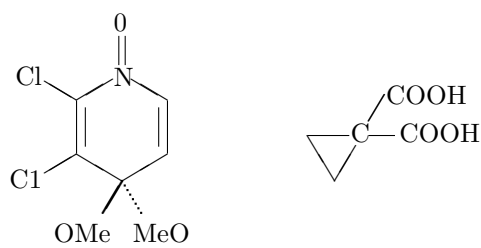
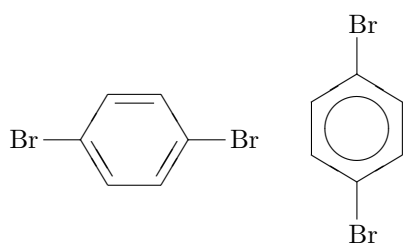
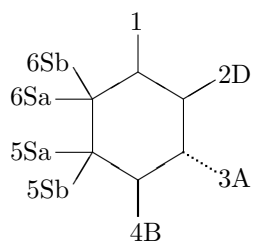
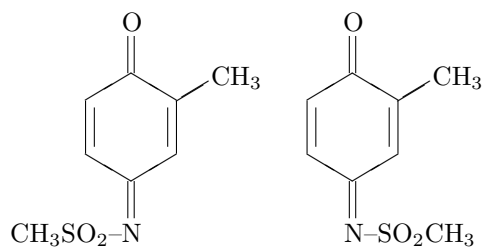
$$K = \frac{[\text{Hg}^{2+}][\text{Hg}]}{[\text{Hg}_2^{2+}]}$$

$$K = \frac{[Hg^{2+}][Hg]}{[Hg^{22+}]}$$

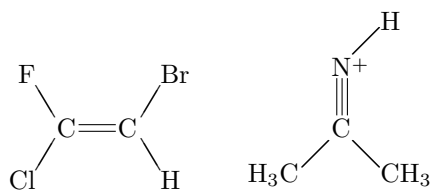
## 2.3 Structural formulae













# Bibliography