

NB: The graded, first version of the report must be returned if you hand in a second time!

H1b: Classical scattering by a central potential

Gustav Hjelmare and Albin Karlsson

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Task N ^o	Points	Avail. points
Σ		

Introduction

How the world works is a question that many strive to answer, and in order to do that we look at the world both at large and small scales. Here we try to describe and evaluate the properties and behavior of aluminium over short times and distances.

Task 1

In implementing our model we have used a lattice spacing of 4.05 \AA (aluminium at room temperature and atmospheric pressure[1]). Before deciding on a suitable timestep, we also implemented the equilibration (Task 2). Thus, the energy data used for determining a suitable timestep is taken from the equilibration process. Plots of energy as a function of time for a few different timesteps can be seen in figure 1.

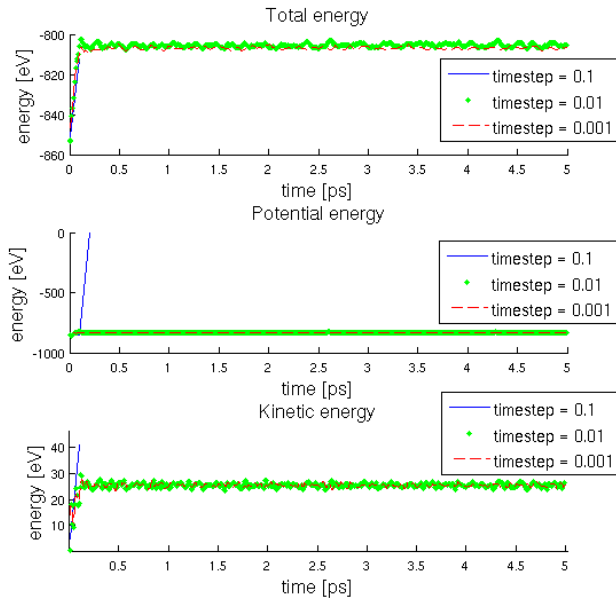


Figure 1: The figure displays how the total, kinetic and potential energy in the system behaves over time, for three different timesteps in the computational model. A timestep that is too large (0.1 ps) makes the model diverge, while a timestep that is too short increases computation time while giving no further information.

The figure shows that 0.01 ps is a suitable timestep. This is what we used for the remainder of the tasks.

Task 2

During equilibration the energy of the system changes, in order to bring the temperature and pressure to their desired values. We found that an equilibration time of 5 ps (corresponding to 500 timesteps at 0.01 ps each) is sufficient to bring the system to the desired state, using time constants $\tau_P = 0.05$ and $\tau_T = 0.02$. The change in total energy is visible in the early timesteps of figure 1.

Task 3 and 4

Figures 2 and 3 show the time evolution of the pressure and temperature. The desired pressure in these runs was $6.32420934 \cdot 10^{-7} \text{ eV/\AA}^3$ (corresponding to 1 atm) and the desired temperature was $500+273=773 \text{ K}$ in fig. 2 and $700+273=973 \text{ K}$ in fig. 3. As can be seen in the figures, the pressure and temperature of the system stabilize around

the desired values. the temperature in figure 2 (after equilibration) has a mean value of 772.95 K and the pressure $4.62 \cdot 10^{-7} \text{ eV/\AA}^3$ while the mean temperature and pressure in figure 3 are 972.8 K and $1.6447 \cdot 10^{-6} \text{ eV/\AA}^3$.

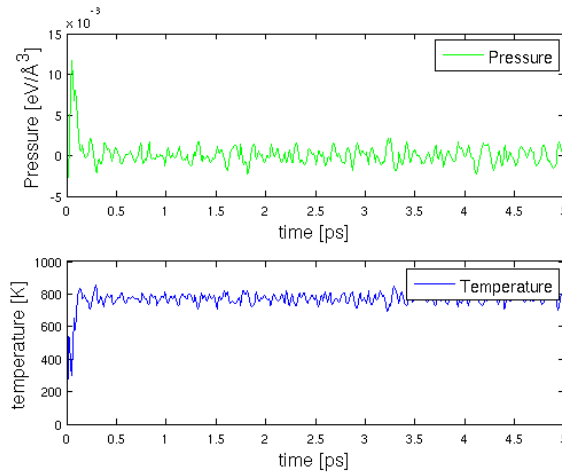


Figure 2: Pressure and temperature during equilibration, showing that the system is stabilized around the temperature 773K and the pressure $6.32420934 \cdot 10^{-7} \text{ eV/\AA}^3$.

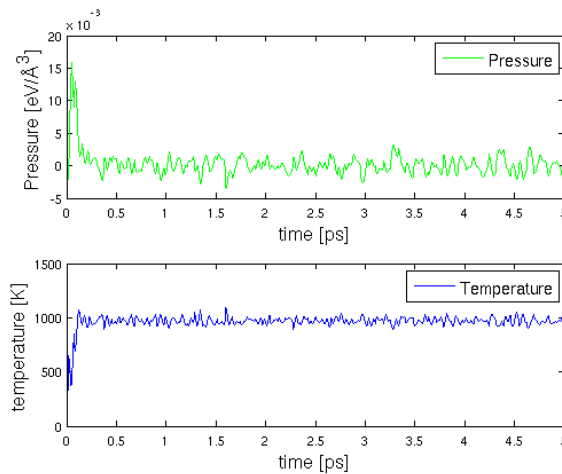


Figure 3: Pressure and temperature during equilibration, showing that the system is stabilized around the temperature 973K and the pressure $6.32420934 \cdot 10^{-7} \text{ eV/\AA}^3$.

From the task description and other sources[1] it seems like our material is supposed to undergo phase transition into a liquid when taking it to 700 degrees Celcius, but looking at the trajectory of a particle over time (figure 4), this doesn't seem to be the case. If the system is a solid the trajectory is expected to remain in some bounded volume while the trajectory of a particle in a liquid is free to move as it pleases. Figure 4 shows the trajectories of one particle in the system at $T = 500$ degrees Celsius (773 K, blue line) and $T = 700$ degrees Celsius (973 K, green line) which obviously both remain bounded. In order to liquify the system, we raised the temperature to $T = 900$ degrees Celsius (1172 K, red line), which clearly shows a non-bounded behaviour in figure 4).

For the remaining tasks we will use the temperature 1173 K in addition to the specified 773 K in order to liquify our system.

The calculated statistical inefficiency constants are shown in table 1.

What can be seen in the table is that the error of the mean values in our result seem to get larger with larger temperature.

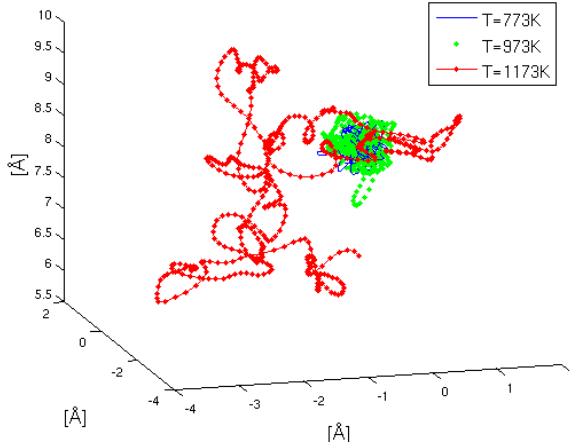


Figure 4: The figure shows the trajectories of one particle in the system at different temperatures. The displacement of the particle at 973 K is larger than the one at 773 K as can be expected, but remains bounded, which is not consistent with our expectations. The trajectory at 1173 K moves away from it's initial position, which is consistent with being in a liquid state.

Table 1: Statistical inefficiency parameters and the errors of our mean value for the temperature, at different temperatures.

	s_T	s_P	$\text{Var}[I_T]$	$\text{Var}[I_P]$
T=773 K	4.2	3.5	6.8	$4.5 \cdot 10^{-9}$
T=973K	3.7	6.1	10.0	$1.8 \cdot 10^{-8}$
T=1173	4.5	20	16.7	$6.5 \cdot 10^{-8}$

Task 5

The mean-squared displacement is shown in figure 5. It is clear that the two lower temperatures result in a bounded mean-squared displacement which confirms that they both are solids, while the highest temperature results in a diverging MSD and therefore should be a liquid. It is also reasonable that the mean-squared displacement is higher at $T = 973$ K than at $T = 773$ K because of more intense vibrations, which can also be seen to some degree in figure 4

The self-diffusion coefficient in the case of the liquid was calculated (using the mean-squared displacement) to be $8.006689 \cdot 10^{-1}$ which is consistent with the self-diffusion coefficient that is calculated using the velocity correlation function in Task 6.

Task 6

The velocity correlation function $\Phi(\Delta t)$ describes, on average, how similar the velocity at time t is to velocities at related times $t + \Delta t$ as t takes all possible values. Since we have identical particles, we use an average over all of them, shown in figure 6. It is clear that for small Δt , velocities are highly correlated, i.e. a particle is unlikely to change velocity completely in a short time span. As Δt grows, the correlation becomes negative, and then positive again after a similar additional time. This is likely due to lattice vibrations, causing many particles to periodically reverse direction after some characteristic time.

The cosine-transform of the velocity correlation function, shown in figure 7, describes the frequency content of the velocity correlation function. The small short-period perturbations are likely an artifact of the simulation, but the fairly clear peaks at 0.3 rad/ps 0.55 rad/ps correspond well with lattice vibration frequencies in other

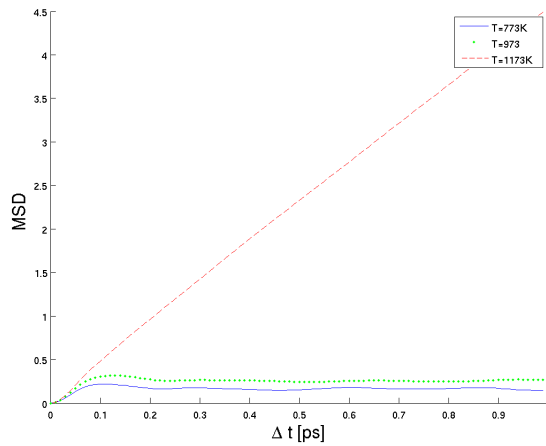


Figure 5: Mean-squared displacement at two different temperatures. At the highest temperature the system is a liquid, as the position appears to diverge over time, while the other two temperatures have bounded positions.

sources[2].

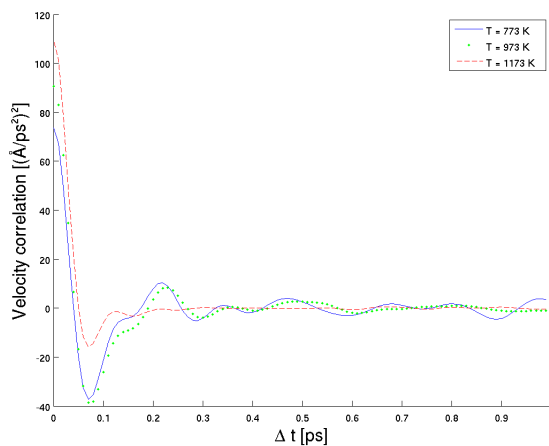


Figure 6: The velocity correlation function, averaged over all particles. For short times there is a significant contribution from the back-and-forth vibration of atoms in the lattice, but as time passes these vibrations get perturbed into other directions and the correlation decays.

When calculating the self-diffusion coefficient, using the velocity correlation we got the result $9.883325 \cdot 10^{-1}$ which is very close to the diffusion coefficient one got using the mean-squared displacement (as it should).

References

- [1] C. Nordling, J. Österman (2006) *Physics Handbook*, 8 ed.
- [2] P. A. Flinn, G. M. McManus (1963) *Lattice Vibrations and Debye Temperatures of Aluminum*. <http://journals.aps.org/pr/pdf/10.1103/PhysRev.132.2458>. (21-11-2014).

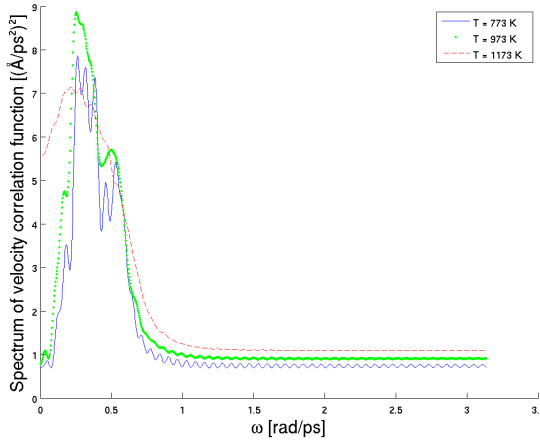


Figure 7: The frequency spectrum of the velocity correlation function. While the data has a significant short-period perturbation, there are also fairly clear peaks around 0.3 rad/ps and 0.55 rad/ps. As can be seen, the highest temperature starts out at rather high value at low frequencies, this means that the velocity is correlated for a long time, which can be expected of a liquid, which is consistent with figure 5 showing a steadily increasing MSD, while the two lower temperatures have no zero-frequency contribution and a limited MSD in figure 5.

A Source Code

A.1 Main file: MD_main.c

```

1 // MD_main.c
2
3 #include <stdio.h>
4 #include <math.h>
5 #include <stdlib.h>
6 #include <time.h>
7 #include "initfcc.h"
8 #include "alpotential.h"
9 #include "MD_functions.h"
10
11 #define PI 3.14159265368979
12
13 /* Main program */
14 int main()
15 {
16     // REMEMBER: \m/ METAL UNITS \m/
17     // simulation settings
18     double productionTime = 5;
19     double equilibrationTime = 5;
20     double timestep = 0.01;
21     // timestep = 0.1;
22     // timestep = 0.001;
23
24     double msdAverageSteps = 50;
25
26     int nSpectrumPoints = 1000;
27     double spectrumInterval = PI;
28
29     double maxCorrelationTime = 1;
30
31     double timeConstantT = 0.02;
32     double timeConstantP = 0.05;
33
34     // physical parameters
35     int dim = 3; // do not change
36     int nCells = 4;
37     int nParticles = 4*pow(nCells,dim);
38     double wantedTemperature = 500+273; // Target temperature
39     // wantedTemperature = 700+273;
40     // wantedTemperature = 900+273;
41     double wantedPressure = 6.32420934* 0.00000001; // 1 atm in metal units
42     double mass = 0.00279636; // 26.9815 u * 1.0364 * 0.0001
43     double latticeParameter = 4.05;
44     double maxDeviation = 0.05;
45

```

```

46 // derived quantities
47 double nEquilibrationSteps = equilibrationTime/timestep;
48 double nProductionSteps = productionTime/timestep;
49 int maxCorrelationSteps = maxCorrelationTime/timestep;
50
51
52 // storage of physical quantities
53 double pos[nParticles][dim];
54 double vel[nParticles][dim];
55 double force[nParticles][dim];
56 double savedValuesT[maxCorrelationSteps];
57 double meanValuesT[maxCorrelationSteps];
58 double savedValuesP[maxCorrelationSteps];
59 double meanValuesP[maxCorrelationSteps];
60 double energy, potentialEnergy, kineticEnergy;
61 double square_root_of_alpha;
62 double alphaT = 1;
63 double alphaP = 1;
64 double currentTemperature;
65 double currentPressure;
66 double sqrtAlphaT;
67 double curAlphaP;
68 double virial;
69 double volume;
70 double nAverageSteps;
71 double meanTemperature, meanSquareTemperature;
72 double meanTemperature_ik[maxCorrelationSteps-1];
73 double meanPressure, meanSquarePressure;
74 double meanPressure_ik[maxCorrelationSteps-1];
75 double phiTemperature[maxCorrelationSteps];
76 double phiPressure[maxCorrelationSteps];
77 double meanTemperatureSquare, meanPressureSquare;
78 double varTemperature, varPressure;
79 double sTemperature, sPressure;
80 double savedPos[maxCorrelationSteps][nParticles][dim];
81 double savedVelocities[maxCorrelationSteps][nParticles][dim];
82 double meanVelocityScalar[maxCorrelationSteps][nParticles];
83 double meanVelocityAverage[maxCorrelationSteps];
84 double spectrum[nSpectrumPoints];
85 double diffusionCoefficient;
86 double meanDistanceScalar[maxCorrelationSteps][nParticles];
87 double meanDistanceAverage[maxCorrelationSteps];
88 double msdDiffCoeff = 0;
89 int MTemperature, MPressure;
90
91 // other stuff
92 int i,j,k,m;
93
94 srand(time(NULL));
95 double random_value;
96
97 // End of variable declarations, start of actual code
98
99 printf("Initializing");
100 init_fcc(pos, nCells, latticeParameter);
101
102 // Applying random perturbations from strict fcc positions
103 for(i = 0; i<nParticles; i++){
104     for (j = 0; j<3; j++){
105         random_value = (double) rand() / (double) RAND_MAX;
106         pos[i][j] = pos[i][j] + (random_value-0.5) * maxDeviation * \
107             latticeParameter;
108     }
109 }
110
111 // Initialize velocities and forces to zero
112 for (i = 0; i<nParticles; i++){
113     for (j = 0; j<3; j++){
114         vel[i][j] = 0;
115         force[i][j] = 0;
116     }
117 }
118
119 // Calculating initial energies
120 potentialEnergy = get_energy_AL(pos, nCells*latticeParameter, nParticles);
121 kineticEnergy = GetKineticEnergy(vel, mass, nParticles);
122 energy = potentialEnergy + kineticEnergy;
123
124 // Saving initial energies on the first line in the file
125 FILE *energyFile;
126 energyFile = fopen("energy.data","w");
127 fprintf(energyFile, "%e \t %e \t %e \t %e \n", 0.0, energy, \
128     potentialEnergy, kineticEnergy);
129
130 FILE *ptFile;
131 ptFile = fopen("pt.data","w");
132
133 FILE *positionFile;
134 positionFile = fopen("position.data","w");
135
136

```

```

137 // Initialization is done, moving on to equilibration
138
139 printf("\t"); // Progress indicator stuff
140 for (i = 0; i < ( nEquilibrationSteps > nProductionSteps ? \
141 nEquilibrationSteps:nProductionSteps )/100; i++) {
142     printf(".");
143 }
144 printf("\tDone!\nEquilibration\t");
145
146 for (i=0;i<nEquilibrationSteps;i++) { // Start of equilibration loop
147     // Update velocities and positions
148     for (j=0; j<nParticles; j++) {
149         for (k = 0; k<dim; k++) {
150             vel[j][k] = vel[j][k] + 0.5*timestep*force[j][k] / mass;
151             pos[j][k] = pos[j][k] + timestep*vel[j][k];
152         }
153     }
154
155     // Calculate forces so that we can take the next half step
156     get_forces_AL(force, pos, nCells*latticeParameter, nParticles);
157
158     // Update velocities again
159     for (j=0; j<nParticles; j++) {
160         for (k=0; k<dim; k++) {
161             vel[j][k] = vel[j][k] + 0.5*timestep*force[j][k] / mass;
162         }
163     }
164
165     // Calculating energies
166     potentialEnergy = get_energy_AL(pos, nCells*latticeParameter, \
167     nParticles);
168     kineticEnergy = GetKineticEnergy(vel, mass, nParticles);
169     energy = potentialEnergy + kineticEnergy;
170     fprintf(energyFile, "%e \t %e \t %e \t %e \n", (i+1)*timestep, energy, \
171     potentialEnergy, kineticEnergy);
172
173     // Get temperature and pressure
174     currentTemperature = GetInstantTemperature(vel, nParticles, mass, dim);
175     volume = pow(nCells*latticeParameter, 3);
176     virial = get_virial_AL(pos, nCells*latticeParameter, nParticles);
177     currentPressure = GetPressure(currentTemperature, volume, virial, \
178     nParticles);
179     fprintf(ptFile, "%e \t %e \t %e \n", (i+1)*timestep, currentTemperature, \
180     currentPressure);
181
182     // Calculate scaling parameters for equilibration
183     alphaT = GetAlphaT(wantedTemperature, currentTemperature, timestep, \
184     timeConstantT);
185     alphaP = GetAlphaP(wantedPressure, currentPressure, timestep, \
186     timeConstantP);
187     sqrtAlphaT = sqrt(alphaT);
188     curtAlphaP = pow(alphaP, 1.0 / 3);
189
190     // Rescale velocity and position
191     latticeParameter = latticeParameter * curtAlphaP;
192     for (j=0; j<nParticles; j++) {
193         for (k=0; k<dim; k++) {
194             vel[j][k] = vel[j][k] * sqrtAlphaT;
195             pos[j][k] = pos[j][k] * curtAlphaP;
196         }
197     }
198     if( i % 100 == 0) { // Progress indicator
199         printf("I");
200         fflush(stdout);
201     }
202 } // End of equilibration loop
203
204 // End of equilibration, start of production
205
206 printf("\tDone!\nProduction\t");
207
208 for (i=0;i<nProductionSteps;i++) {
209     // Save positions of one particle to check whether the
210     // aluminium is solid or liquid
211     fprintf(positionFile, "%d \t %e \t %e \t %e \n", i, \
212     pos[0][0], pos[0][1], pos[0][2]);
213
214     // Update velocities and positions
215     for (j=0; j<nParticles; j++) {
216         for (k = 0; k<dim; k++) {
217             vel[j][k] = vel[j][k] + 0.5*timestep*force[j][k] / mass;
218             pos[j][k] = pos[j][k] + timestep*vel[j][k];
219         }
220     }
221
222     // Calculate forces so that we can take the next half step
223     get_forces_AL(force, pos, nCells*latticeParameter, nParticles);
224
225     // Update velocities again
226     for (j=0; j<nParticles; j++) {
227         for (k=0; k<dim; k++) {

```



```

228     vel[j][k] = vel[j][k] + 0.5*timestep*force[j][k] / mass;
229 }
230 }
231
232 potentialEnergy = get_energy_AL(pos, nCells*latticeParameter, \
233 nParticles);
234 kineticEnergy = GetKineticEnergy(vel, mass, nParticles);
235 energy = potentialEnergy + kineticEnergy;
236
237 currentTemperature = GetInstantTemperature(vel, nParticles, mass, dim);
238 volume = pow(nCells*latticeParameter, 3);
239 virial = get_virial_AL(pos, nCells*latticeParameter, nParticles);
240 currentPressure = GetPressure(currentTemperature, volume, virial, \
241 nParticles);
242
243 //Saves temp and pressure values in order to calculate s.
244 if (i < maxCorrelationSteps) {
245     savedValuesT[i] = currentTemperature;
246     savedValuesP[i] = currentPressure;
247 } else {
248     // updates the saved values.
249     for (j = 0; j < maxCorrelationSteps - 1; j++){
250         savedValuesT[j] = savedValuesT[j+1];
251         savedValuesP[j] = savedValuesP[j+1];
252     }
253     savedValuesT[maxCorrelationSteps - 1] = currentTemperature;
254     savedValuesP[maxCorrelationSteps - 1] = currentPressure;
255
256     // calculates and saves f_i*f_k
257     for (j = 0; j<maxCorrelationSteps; j++){
258         meanTemperature_ik[j] += savedValuesT[maxCorrelationSteps - 1] \
259 * savedValuesT[maxCorrelationSteps - 1 - j];
260         meanPressure_ik[j] += savedValuesP[maxCorrelationSteps - 1] \
261 * savedValuesP[maxCorrelationSteps - 1 - j];
262     }
263     meanTemperature += currentTemperature;
264     meanPressure += currentPressure;
265     meanSquareTemperature += currentTemperature * currentTemperature;
266     meanSquarePressure += currentPressure * currentPressure;
267 }
268
269 // Saving data for the velocity correlation function
270 if (i < maxCorrelationSteps) {
271     for (m = 0; m < nParticles ; m++) {
272         savedVelocities[i][m][0] = vel[m][0];
273         savedVelocities[i][m][1] = vel[m][1];
274         savedVelocities[i][m][2] = vel[m][2];
275     }
276 } else {
277     for (j = 0; j< maxCorrelationSteps- 1; j++) {
278         for (m = 0; m<nParticles; m++) {
279             savedVelocities[j][m][0] = savedVelocities[j+1][m][0];
280             savedVelocities[j][m][1] = savedVelocities[j+1][m][1];
281             savedVelocities[j][m][2] = savedVelocities[j+1][m][2];
282         }
283     }
284     for (m = 0; m<nParticles; m++) {
285         savedVelocities[maxCorrelationSteps - 1][m][0] = vel[m][0];
286         savedVelocities[maxCorrelationSteps - 1][m][1] = vel[m][1];
287         savedVelocities[maxCorrelationSteps - 1][m][2] = vel[m][2];
288     }
289
290     for (j = 0; j<maxCorrelationSteps; j++) {
291         for (m = 0; m < nParticles; m++) {
292             meanVelocityScalar[j][m] += ScalarProduct( \
293             savedVelocities[0][m], savedVelocities[j][m]);
294         }
295     }
296 }
297
298 if(i < maxCorrelationSteps) {
299     for(j = 0; j<nParticles; j++) {
300         for (k = 0; k<dim; k++) {
301             savedPos[i][j][k] = pos[j][k];
302         }
303     }
304 } else {
305     for (j = 0; j<maxCorrelationSteps-1; j++) {
306         for (k = 0; k<nParticles; k++) {
307             for (m = 0; m<dim; m++) {
308                 savedPos[j][k][m] = savedPos[j+1][k][m];
309             }
310         }
311     }
312     for (j = 0; j<nParticles; j++) {
313         for (k=0; k<dim; k++) {
314             savedPos[maxCorrelationSteps-1][j][k] = pos[j][k];
315         }
316     }
317     for(j = 0; j<maxCorrelationSteps; j++){
318         for(k = 0; k<nParticles; k++){

```

```

319         meanDistanceScalar[j][k] += getDistanceSquared( \
320             savedPos[0][k], savedPos[j][k]);
321     }
322 }
323 }
324
325
326 if(i%100 == 0) { // Progress indicator
327     printf("I");
328     fflush(stdout);
329 }
330
331 } // End of production loop
332
333 printf("\tDone!\nCleaning up\n");
334
335 // The production part of the simulation is now done.
336 // The code below forms some averages and other stuff that
337 // is best done with all data on hand
338
339 FILE *valuesFile; // For saving various values (non-array data)
340 valuesFile = fopen("values.data", "w");
341
342 nAverageSteps = nProductionSteps - maxCorrelationSteps;
343 // take averages over this many steps
344
345 FILE *msdFile;
346 msdFile = fopen("msd3.data", "w");
347
348 for( i = 0; i < maxCorrelationSteps; i++) {
349     meanDistanceAverage[i] = 0;
350     for (j = 0; j < nParticles; j++) {
351         meanDistanceScalar[i][j] = meanDistanceScalar[i][j]/nAverageSteps;
352         meanDistanceAverage[i] += meanDistanceScalar[i][j];
353     }
354     meanDistanceAverage[i] = meanDistanceAverage[i]/nParticles;
355     fprintf(msdFile, "%e\t%e\n", i*timestep, meanDistanceAverage[i]);
356 }
357
358 // Diffusion coeff from MSD
359 for (i = maxCorrelationSteps - msdAverageSteps; \
360     i < maxCorrelationSteps; i++) {
361     msdDiffCoeff += 1.0/(6*i*timestep)*meanDistanceAverage[i];
362 }
363 msdDiffCoeff = msdDiffCoeff / msdAverageSteps;
364 fprintf(valuesFile, "%e\tDiffusion coeff (MSD)\n", msdDiffCoeff);
365
366 // Final processing and saving of velocity correlation
367 // function and diffusion coeff from time integral
368 FILE *velcorFile;
369 velcorFile = fopen("velcor.data", "w");
370
371 diffusionCoefficient = 0;
372
373 for( i = 0; i < maxCorrelationSteps; i++) {
374     meanVelocityAverage[i] = 0;
375     for (j = 0; j < nParticles; j++) {
376         meanVelocityScalar[i][j] = meanVelocityScalar[i][j]/nAverageSteps;
377         meanVelocityAverage[i] += meanVelocityScalar[i][j];
378     }
379     meanVelocityAverage[i] = meanVelocityAverage[i]/nParticles;
380     fprintf(velcorFile, "%e\t%e\n", i*timestep, meanVelocityAverage[i]);
381
382     diffusionCoefficient += meanVelocityAverage[i];
383 }
384
385 diffusionCoefficient = (1.0/3.0) * \
386     diffusionCoefficient / maxCorrelationSteps;
387 fprintf(valuesFile, "%e\tDiffusion coeff (time integral)\n", \
388     diffusionCoefficient);
389
390 // Calculating and saving spectrum integral thingy
391 FILE *spectrumFile;
392 spectrumFile = fopen("spectrum.data", "w");
393
394 for( i = 0; i < nSpectrumPoints; i++) {
395     spectrum[i] = 0;
396     for( j = 0; j < maxCorrelationSteps; j++) {
397         spectrum[i] += meanVelocityAverage[j] * cos(spectrumInterval * \
398             i * j / nSpectrumPoints);
399     }
400     spectrum[i] = 2 * spectrum[i]/maxCorrelationSteps;
401     fprintf(spectrumFile, "%e \t %e \n", i*spectrumInterval/nSpectrumPoints, \
402         spectrum[i]);
403 }
404
405 // Setting up for calculating correlation data for temperature and pressure
406 meanTemperature = meanTemperature/nAverageSteps;
407 meanSquareTemperature = meanSquareTemperature/nAverageSteps;
408 meanPressure = meanPressure/nAverageSteps;
409 meanSquarePressure = meanSquarePressure/nAverageSteps;

```

```

410 meanTemperatureSquare = meanTemperature*meanTemperature;
411 meanPressureSquare = meanPressure*meanPressure;
412
413 for(i = 0; i < maxCorrelationSteps+1; i++){
414     meanTemperature_ik[i] = meanTemperature_ik[i]/nAverageSteps;
415     meanPressure_ik[i] = meanPressure_ik[i]/nAverageSteps;
416 }
417
418
419
420 // Correlation data for temperature and pressure
421 FILE *phiTFile;
422 phiTFile = fopen("phiTemperature.data", "w");
423
424 FILE *phiPFile;
425 phiPFile = fopen("phiPressure.data", "w");
426
427 for(i = 0; i<maxCorrelationSteps; i++){
428     phiTemperature[i] = (meanTemperature_ik[i] - meanTemperatureSquare)/ \
429     (meanSquareTemperature - meanTemperatureSquare);
430     phiPressure[i] = (meanPressure_ik[i] - meanPressureSquare)/ \
431     (meanSquarePressure - meanPressureSquare);
432     fprintf(phiTFile, "%e \t %e \n", (i+1)*timestep, phiTemperature[i]);
433     fprintf(phiPFile, "%e \t %e \n", (i+1)*timestep, phiPressure[i]);
434 }
435
436 // Finding M (the longest time for which correlation is above some limit)
437 double limit = exp(-2.0);
438 i = 0;
439 while( (phiTemperature[i] > limit) && (i < maxCorrelationSteps) ) {
440     i++;
441 }
442 MTemperature = i;
443
444 i = 0;
445 while( (phiPressure[i] > limit) && (i < maxCorrelationSteps) ) {
446     i++;
447 }
448 MPressure = i;
449
450 //Sum over phi to get statistical inefficiency for temperature
451 for(i = 0; i<MTemperature; i++){
452     sTemperature += phiTemperature[i];
453 }
454 sTemperature = sTemperature*2;
455 // phi is symmetric and we sum from 0 to M instead of from -M to M
456
457 //Sum over phi to get statistical inefficiency for pressure
458 for(i = 0; i<MPressure; i++){
459     sPressure += phiPressure[i];
460 }
461 sPressure = sPressure*2;
462 // phi is symmetric and we sum from 0 to M instead of from -M to M
463
464 // Calculating variances
465 varTemperature = meanSquareTemperature - meanTemperatureSquare;
466 varPressure = meanSquarePressure - meanPressureSquare;
467
468 varTemperature = varTemperature*sTemperature/nProductionSteps;
469 varPressure = varPressure*sPressure/nProductionSteps;
470 fprintf(valuesFile, "%e\tsTemperature\n", sTemperature);
471 fprintf(valuesFile, "%e\tsPressure\n", sPressure);
472 fprintf(valuesFile, "%e\tvarTemperature\n", varTemperature);
473 fprintf(valuesFile, "%e\tvarPressure\n", varPressure);
474
475 FILE *donefile; // Just an empty file to indicate completion
476 donefile = fopen("done.data", "w"); // useful when using ssh+screen
477 fprintf(donefile, "done");
478
479 close(energyFile);
480 close(ptFile);
481 close(positionFile);
482 close(valuesFile);
483 close(msdFile);
484 close(velcorFile);
485 close(spectrumFile);
486 close(phiTFile);
487 close(phiPFile);
488 close(donefile);
489
490 printf("Done!\n");
491 return 0;
492 }

```

A.2 Functions header: MD_functions.h

```

1  #ifndef _MD_functions_h
2  #define _MD_functions_h
3
4  extern double getDistanceSquared(double *, double *);
5  extern double ScalarProduct(double *, double *);
6  extern double GetPressure(double, double, double, int);
7  extern double GetInstantTemperature(double[][3], int, double, int);
8  extern double GetKineticEnergy(double[][3], double, int);
9  extern double GetAlphaT(double, double, double, double);
10 extern double GetAlphaP(double, double, double, double);
11
12
13 #endif

```

A.3 Functions code: MD_functions.c

```

1  #include <stdio.h>
2  #include <math.h>
3  #include <stdlib.h>
4
5  #define BOLTZMANN 8.6173324*0.00001 // in eV/K
6
7
8  double getDistanceSquared(double vec1[3], double vec2[3])
9  {
10     double sum = 0;
11     double temporary;
12     int i;
13
14
15     for(i = 0; i<3; i++){
16         temporary = vec1[i] - vec2[i];
17         sum += pow(temporary,2);
18     }
19     return sum;
20 }
21
22 double ScalarProduct(double vec1[3], double vec2[3])
23 {
24     double product = 0;
25     int i;
26
27     for(i = 0 ; i < 3 ; i++) {
28         product += vec1[i]*vec2[i];
29     }
30
31     return product;
32 }
33
34
35 double GetPressure(double temperature, double volume, double virial, \
36 int nParticles)
37 {
38     double pressure = 0;
39
40     pressure = (nParticles * BOLTZMANN * temperature + virial)/volume;
41
42     return pressure;
43 }
44
45 double GetInstantTemperature(double vel[][3], int nParticles, \
46 double mass, int dim)
47 {
48
49     {
50         double temperature = 0;
51         int i,j;
52
53         double factor = 2/(3*nParticles*BOLTZMANN);
54
55         for (i=0;i<nParticles;i++) {
56             for (j=0;j<dim;j++) {
57                 temperature += factor * pow(vel[i][j],2)*mass/2;
58             }
59         }
60
61         return temperature;
62     }
63
64     // Calculates the kinetic energy
65     double GetKineticEnergy(double vel[][3], double mass, int nParticles)
66     {
67         int i,j;
68         double sum;
69         double energy = 0;
70

```

```

71 for(i = 0; i<nParticles; i++) {
72     sum = 0;
73     for(j = 0; j<3; j++) {
74         sum +=vel[i][j]*vel[i][j];
75     }
76     energy += sum*mass/2;
77 }
78 return energy;
79 }
80
81 // Calculate alphaT (the modifier to get the right temperature)
82 double GetAlphaT(double wantedTemp, double currentTemp, \
83                 double timestep, double timeConstant)
84 {
85     double alpha;
86     alpha = 1 + timestep/timeConstant * (wantedTemp-currentTemp)/currentTemp;
87
88     return alpha;
89 }
90
91
92
93 // Calculate alphaP (the modifier to get the right pressure)
94 double GetAlphaP(double wantedPressure, double currentPressure, \
95                 double timestep, double timeConstant)
96 {
97     double kappa = 2.21901454; // \AA^3/eV (at 300K)
98     double alpha;
99
100     alpha = 1 - kappa*timestep/timeConstant * \
101             (wantedPressure - currentPressure);
102
103     return alpha;
104 }

```

A.4 Plotting in matlab: plot_results.m

```

1  % in some places, data for several temperatures is required
2  % so you need to run the main program several times with
3  % different temperature settings and manually rename data files
4
5  clear all
6  clc
7
8  data = importdata('energyT1.data');
9
10 time1 = data(:,1);
11 energy1 = data(:,2);
12 potentialEnergy1 = data(:,3);
13 kineticEnergy1 = data(:,4);
14
15 data = importdata('energyT2.data');
16 time2 = data(:,1);
17 energy2 = data(:,2);
18 potentialEnergy2 = data(:,3);
19 kineticEnergy2 = data(:,4);
20
21 data = importdata('energyT3.data');
22 time3 = data(:,1);
23 energy3 = data(:,2);
24 potentialEnergy3 = data(:,3);
25 kineticEnergy3 = data(:,4);
26
27
28
29 %%
30 textStorlek = 14;
31 legendStorlek = 11;
32 subplot(3,1,1)
33 hold on
34 plot(time1, energy1, 'b');
35 plot(time2, energy2, 'g');
36 plot(time3, energy3, 'r--');
37 text = legend('timestep = 0.1', 'timestep = 0.01', 'timestep = 0.001');
38 set(text, 'FontSize', legendStorlek);
39 title('Total energy', 'FontSize',textStorlek)
40 ylabel('energy [eV]', 'FontSize',textStorlek);
41 xlabel('time [ps]', 'FontSize',textStorlek);
42 subplot(3,1,2)
43 hold on
44 plot(time1, potentialEnergy1, 'b');
45 plot(time2, potentialEnergy2, 'g');
46 plot(time3, potentialEnergy3, 'r--');
47 text = legend('timestep = 0.1', 'timestep = 0.01', 'timestep = 0.001');
48 set(text, 'FontSize', legendStorlek);
49 title('Potential energy', 'FontSize',textStorlek);

```

```

50 ylabel('energy [eV]', 'FontSize',textStorlek);
51 xlabel('time [ps]', 'FontSize',textStorlek);
52 subplot(3,1,3)
53 hold on
54 plot(time1, kineticEnergy1, 'b');
55 plot(time2, kineticEnergy2, 'g. ');
56 plot(time3, kineticEnergy3, 'r--');
57 text = legend('timestep = 0.1', 'timestep = 0.01', 'timestep = 0.001');
58 set(text, 'FontSize', legendStorlek);
59 title('Kinetic energy', 'FontSize',textStorlek);
60 ylabel('energy [eV]', 'FontSize',textStorlek);
61 xlabel('time [ps]', 'FontSize',textStorlek);
62 %% Equilibration - T & P
63 data = importdata('pt500.data');
64 temp500 = data(:,2);
65 pressure500 = data(:,3);
66
67 data = importdata('pt700.data');
68 temp700 = data(:,2);
69 pressure700 = data(:,3);
70
71 textStorlek = 14;
72 legendStorlek = 11;
73
74
75 hold on
76 subplot(2,1,1);
77 plot(data(:,1), pressure500, 'g');
78 text = legend('Pressure');
79 set(text, 'FontSize', legendStorlek);
80 ylabel('Pressure [eV/ ^{3}]', 'FontSize',textStorlek);
81 xlabel('time [ps]', 'FontSize',textStorlek);
82
83 hold on
84
85 subplot(2,1,2);
86 plot(data(:,1), temp500, 'b');
87 text = legend('Temperature');
88 set(text, 'FontSize', legendStorlek);
89 ylabel('temperature [K]', 'FontSize',textStorlek);
90 xlabel('time [ps]', 'FontSize',textStorlek);
91
92 figure
93 hold on
94 subplot(2,1,1);
95 plot(data(:,1), pressure700, 'g');
96 text = legend('Pressure');
97 set(text, 'FontSize', legendStorlek);
98 ylabel('Pressure [eV/ ^{3}]', 'FontSize',textStorlek);
99 xlabel('time [ps]', 'FontSize',textStorlek);
100
101 hold on
102
103 subplot(2,1,2);
104 plot(data(:,1), temp700, 'b');
105 text = legend('Temperature');
106 set(text, 'FontSize', legendStorlek);
107 ylabel('temperature [K]', 'FontSize',textStorlek);
108 xlabel('time [ps]', 'FontSize',textStorlek);
109
110 useFrom = 0.6;
111 start = fix(length(energy)*useFrom);
112
113 meanTemp = mean(temp(start:end))
114 meanPress = mean(pressure(start:end))
115
116 %% Velocity correlation function
117
118 clear all
119 clc
120 clf
121
122 textStorlek = 14;
123 legendStorlek = 11;
124
125 data_500 = importdata('velcor_500.data');
126 data_700 = importdata('velcor_700.data');
127 data_900 = importdata('velcor_900.data');
128
129 hold on
130 plot(data_500(:,1),data_500(:,2),'b')
131 plot(data_700(:,1),data_700(:,2),'g. ')
132 plot(data_900(:,1),data_900(:,2),'r--')
133
134 ylabel('Velocity correlation [( /ps^2)^2]', 'FontSize',textStorlek)
135 xlabel('\Delta t [ps]', 'FontSize',textStorlek)
136
137 text = legend('T = 773 K', 'T = 973 K', 'T = 1173 K');
138 set(text, 'FontSize', legendStorlek);
139
140 saveas(gcf, 'velcor.png', 'png')

```

```

141
142 %% Spectrum analysis
143
144 clear all
145 clc
146 clf
147
148 textStorlek = 14;
149 legendStorlek = 11;
150
151 data_500 = importdata('spectrum_500.data');
152 data_700 = importdata('spectrum_700.data');
153 data_900 = importdata('spectrum_900.data');
154
155 hold on
156 plot(data_500(:,1),data_500(:,2),'b')
157 plot(data_700(:,1),data_700(:,2),'g.')
158 plot(data_900(:,1),data_900(:,2),'r--')
159
160 ylabel('Spectrum of velocity correlation function [( /ps^2)^2]', ...
161 'FontSize',textStorlek)
162 xlabel('\omega [rad/ps]', 'FontSize',textStorlek)
163
164 text = legend('T = 773 K', 'T = 973 K', 'T = 1173 K');
165 set(text, 'FontSize', legendStorlek);
166
167 saveas(gcf, 'spectrum.png', 'png')
168
169 %% MSD function
170
171 clear all
172 clc
173 clf
174
175 data = importdata('msd.data');
176 data2 = importdata('msd2.data');
177 data3 = importdata('msd3.data');
178
179 textStorlek = 14;
180 legendStorlek = 11;
181
182 hold on
183 plot(data(:,1),data(:,2), 'b');
184 plot(data3(:,1),data3(:,2), 'g. ');
185 plot(data2(:,1),data2(:,2), 'r--');
186
187
188 ylabel('MSD ', 'FontSize',textStorlek)
189 xlabel('\Delta t [ps]', 'FontSize',textStorlek)
190 text = legend('T=773K', 'T=973', 'T=1173K');
191 set(text, 'FontSize', legendStorlek);
192
193 saveas(gcf, 'MSD.png', 'png')
194
195 %% Plot positions (3D-plot) of one particle
196 clear all
197 clc
198 pos1 = importdata('position1.data');
199 pos2 = importdata('position2.data');
200 pos3 = importdata('position3.data');
201
202
203 textStorlek = 14;
204 legendStorlek = 11;
205
206 hold on
207 plot3(pos1(:,2), pos1(:,3),pos1(:,4), 'b');
208 plot3(pos2(:,2), pos2(:,3),pos2(:,4), 'g. ');
209 plot3(pos3(:,2), pos3(:,3),pos3(:,4), 'r. ');
210 text = legend('T=773K', 'T=973K', 'T=1173K');
211 set(text, 'FontSize', legendStorlek);
212
213 xlabel('[ ]', 'FontSize',textStorlek);
214 ylabel('[ ]', 'FontSize',textStorlek);
215 zlabel('[ ]', 'FontSize',textStorlek);

```