

Department of Bitechology
Biomolecular simulation BT2123
Fall semester Jan-May 2023
Quiz 1
January 12, 2023

Maximum Marks: 20

Maximum Time: 30 minutes

If required, please use/write the approximate values of the desired constant
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1. What is the main difference between gravitational and electrostatic forces? Can gravitational or electrostatic forces be negative ?
2. What is the MKS unit of electrostatic potential, V. Electron volt eV is unit of which quantity and what is value of 1 eV in MKS units?
3. What is the numerical value of Boltzmann constant
4. Calculate the ratio of gravitational and electrostatic force between a P atom of DNA and Na^+ ion separated by a distance of 10 \AA in vacuum. The masses of P atom and Na^+ ion are 30.97 and 22.99 amu respectively, while the charges are $-2e$ and $+1e$ respectively.

Given that

$$1 e = 1.602 \times 10^{-19} \text{ coulomb.}$$

$$1\text{amu} = 1.66 \times 10^{-27} \text{ kg.}$$

$$\text{permittivity of free space } 8.85 \times 10^{-12} \text{ F/m}$$

$$\text{gravitational force constant is } 6.67 \times 10^{-11} \text{ m}^3 \text{ kg}^{-1} \text{ s}^{-2}$$