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Battery Charging and Chemistry Detection with the MSP430

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- Battery Charging Basics
- Li-Ion Charging
- NiMH Charging
- Chemistry Detection
- Hardware and Software Overview
- Demo

Battery Chemistries

| Chemistry | Alkaline | Ni-Cad | NiMH | Li-lon |
|----------------|----------|--------|--------|--------|
| Rechargeable | No | Yes | Yes | Yes |
| Memory Effect | N/A | Large | Small | None |
| Self Discharge | Medium | High | High | Low |
| Current Output | Medium | High | High | Low |
| Capacity | Low | Low | Medium | High |
| Cost | Low | Low | Medium | High |

- Li-lon is potentially dangerous to charge
- Ni-Cad is very environmentally unfriendly

Battery Charging Basics

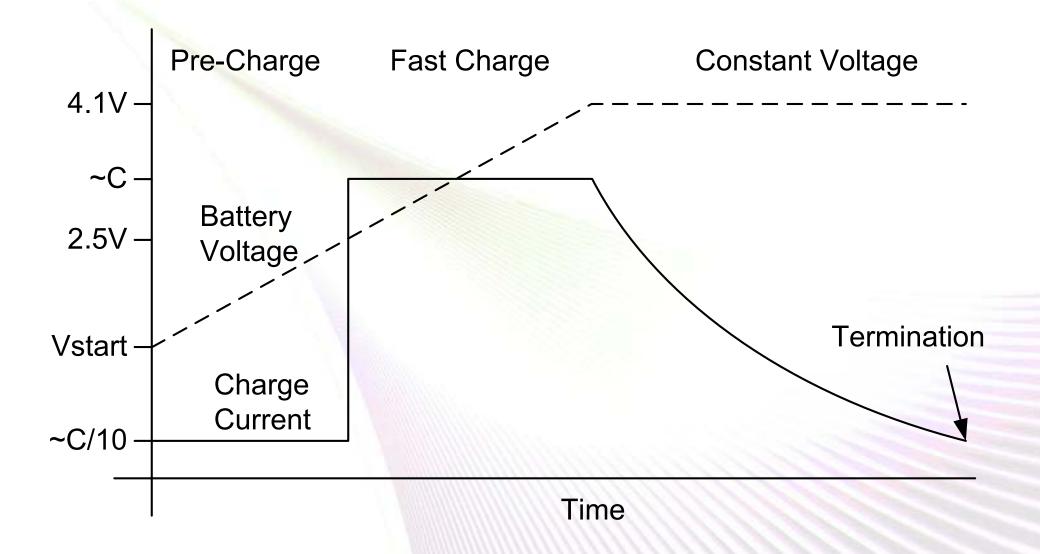
- Battery Capacity is expressed in mAh
- Charge rate is expressed in terms of C
- C = Battery Capacity / 1 hour
- Example: 2500mAh battery
 - C = 2.5A
 - 2C = 5A
 - C/2 = 1.25A
 - C/10 = 250mA
 - Etc.
- A battery charged at 1C should charge in ~1 hour, at C/2 in ~2 hours

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Li-Ion Charging Stages

- Pre-Charge (Conditioning)
 - If battery voltage < 2.5V
 - Constant current of C/10
- Fast Charge
 - Constant current of 1C
 - While battery voltage in between 2.5V and 4.1V
- Constant Voltage Charge
 - Constant voltage at 4.1V
- Termination
 - When current drops below C/10

Li-lon Charging



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NiMH Charging Stages

Pre-Charge (Conditioning)

- If battery voltage < 0.9V
- Constant Current of C/10

Charge

- Constant current of C/2 1C
- When battery voltage > 0.9V

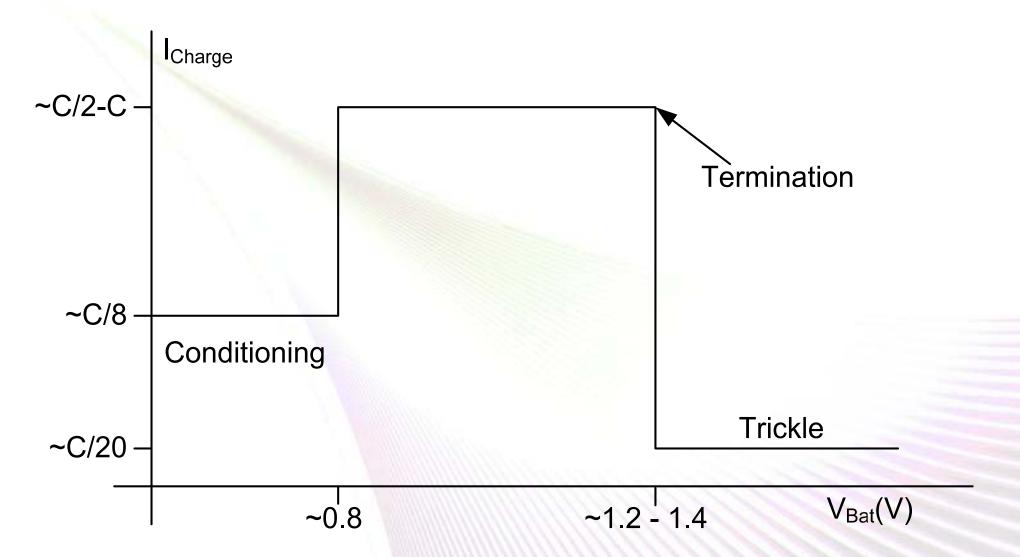
Termination

3 different methods

Trickle Charge

After termination constant current of C/20 to maintain charge

NiMH Charging





NiMH Charge Termination

3 Methods

- -ΔV/Δt (primary)
- ΔT/Δt (secondary)
- Timer (secondary)

Use all 3 for robust charge termination

ΔT/Δt

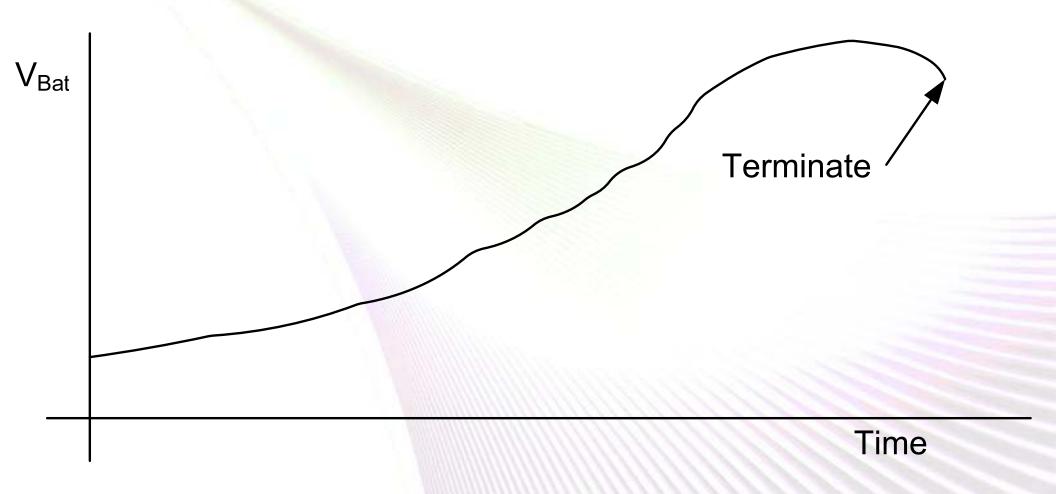
 Look for sharp increase in battery temperature over ambient or absolute safety temperature

Timer

- Set timer for appropriate charge time + 10% to 20% (to account for inefficiencies)
- Example: if charging at C/2, set timer for 2 hours + 12 to 24 minutes

-ΔV/Δt Termination

Look for battery voltage dip





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About Chemistries Detection

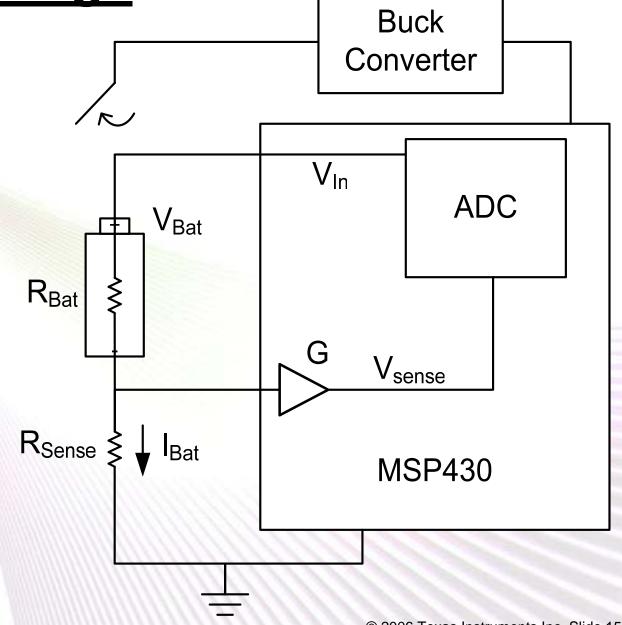
- Different battery chemistries have different internal impedances
- These impedances change over temperature, charge state, and life of the battery
- Goal: only charge the battery if it is definitely a NiMH (or NiCad)
- Never attempt to charge different chemistries (Alkaline, Lithium, etc.)
- NiMH has lowest impedance of all AA battery chemistries
- Impedance = $R_{Bat} = V_{Bat(Closed)} V_{Bat(Open)}$ $I_{Bat(Closed)}$

Open Loop Voltage

• $I_{Bat} = 0mA$

• V_{sense} = **0**V

• $V_{Bat} = V_{In}$

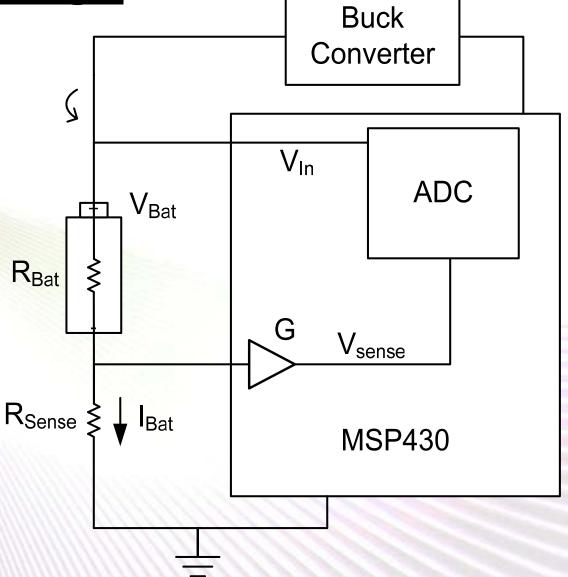


Closed Loop Voltage

• I_{Bat} < 150mA

• V_{sense} = G · I_{Bat} · R_{Sense}

• $V_{Bat} = V_{In} - (I_{Bat} - R_{Sense})$ R_{Bat}



Impedance Calculation

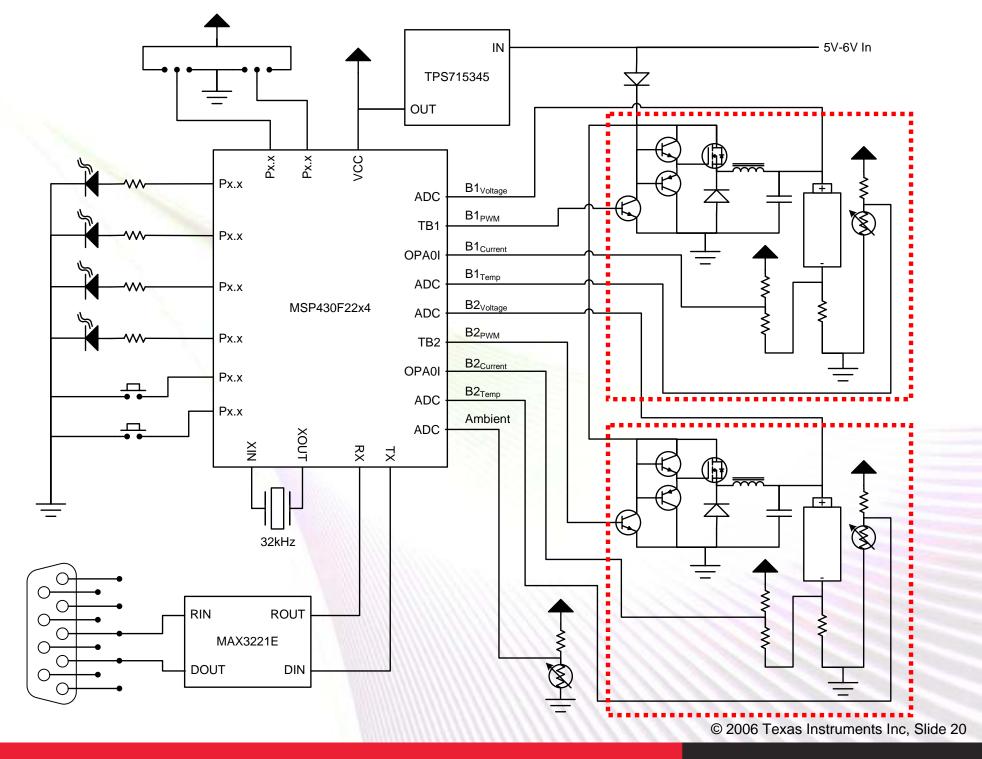
•
$$R_{Bat} = V_{\frac{Bat(Closed)}{I_{Bat(Closed)}}}$$

How to Identify a NiMH AA Battery

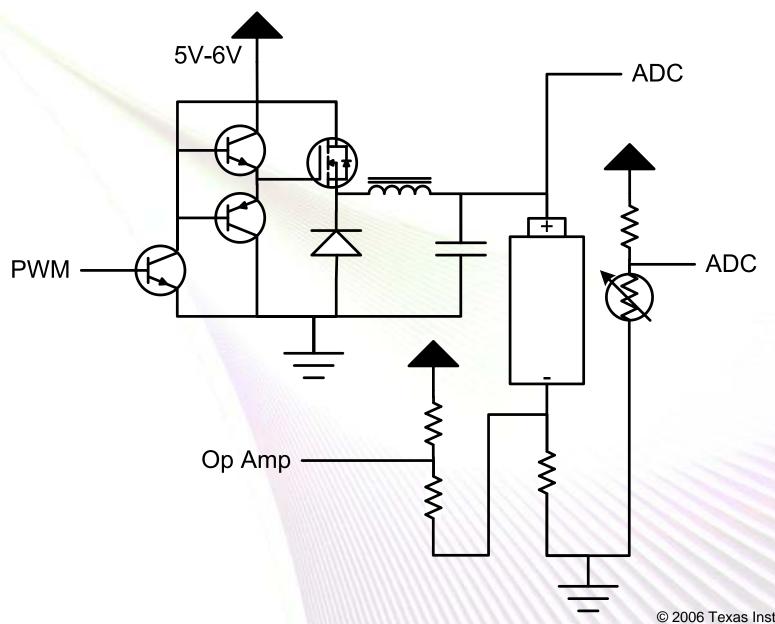
- We use the battery's impedance calculation along with its open circuit voltage (estimate of its charge state) to determine whether it is a NiMH battery
- This method will sometimes fail to identify very old NIMH batteries correctly
 - Impedance increases over the life of a battery
- Conditioning charge of 150mA for some time often allows NiMH to recover to low impedance
- Example Impedance cutoffs for AA batteries

| Voltage | 1.5V | 1.2V | 0.9V |
|---------------------|---------|---------|---------|
| Cutoff Impedance | 100mOhm | 150mOhm | 200mOhm |

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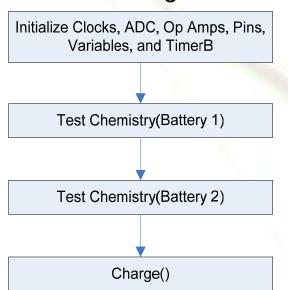


Battery Charge Circuitry

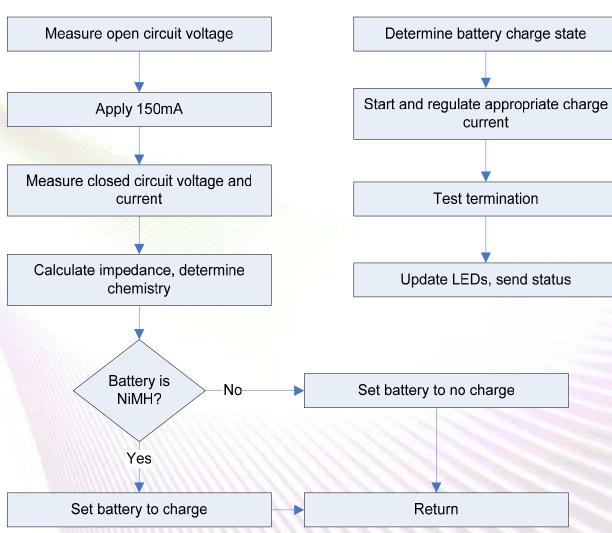


Software Overview

Main Program



Test Chemistry



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Charge

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<u>Demo</u>

- Impedance Demo
- NiMH Charging Demo

Summary

- Li-lon and NiMH batteries can be charged using the analog and digital control capabilities of the MSP430
- Non-Rechargeable batteries can be detected by measuring their open circuit voltage and impedance

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