

Statistical connection between
microscopic & macroscopic properties

Microscopic (atomistic) - molecular properties

x_1, y_1, z_1 of each atom or $v_x, v_y, v_z \dots$

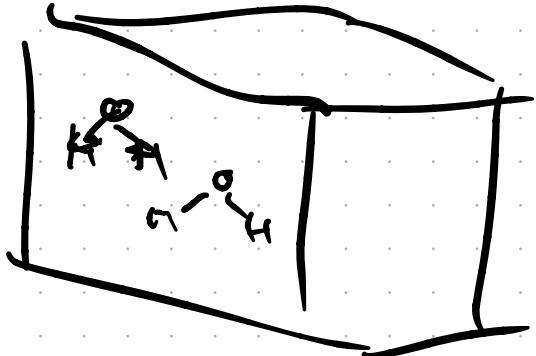
Spin of each of each atom

Macroscopic (Thermodynamic)

T, P , heat capacity, coefficient of thermal expansion

Macrostates - characterizes the thermodynamic system N, V, T N, p, T

Microstate - particular configuration of your system



$$N_{H_2O}, V = 1 \text{ l}$$
$$T = 25^\circ\text{C}$$

$$x_1, y_1, z_1$$

$$x_2, y_2, z_2$$

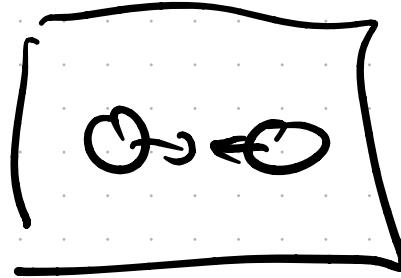
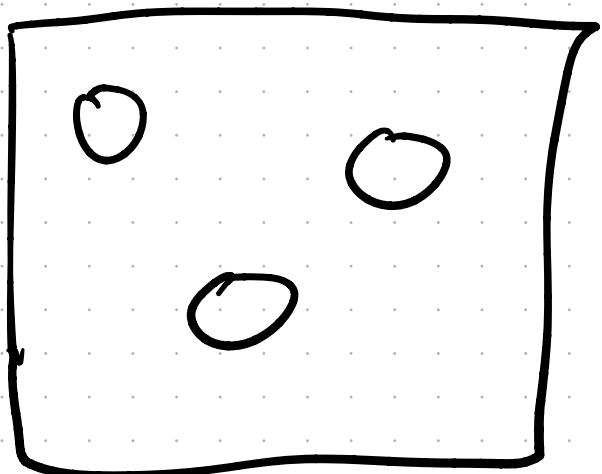
⋮

$$x_N, y_N, z_N$$

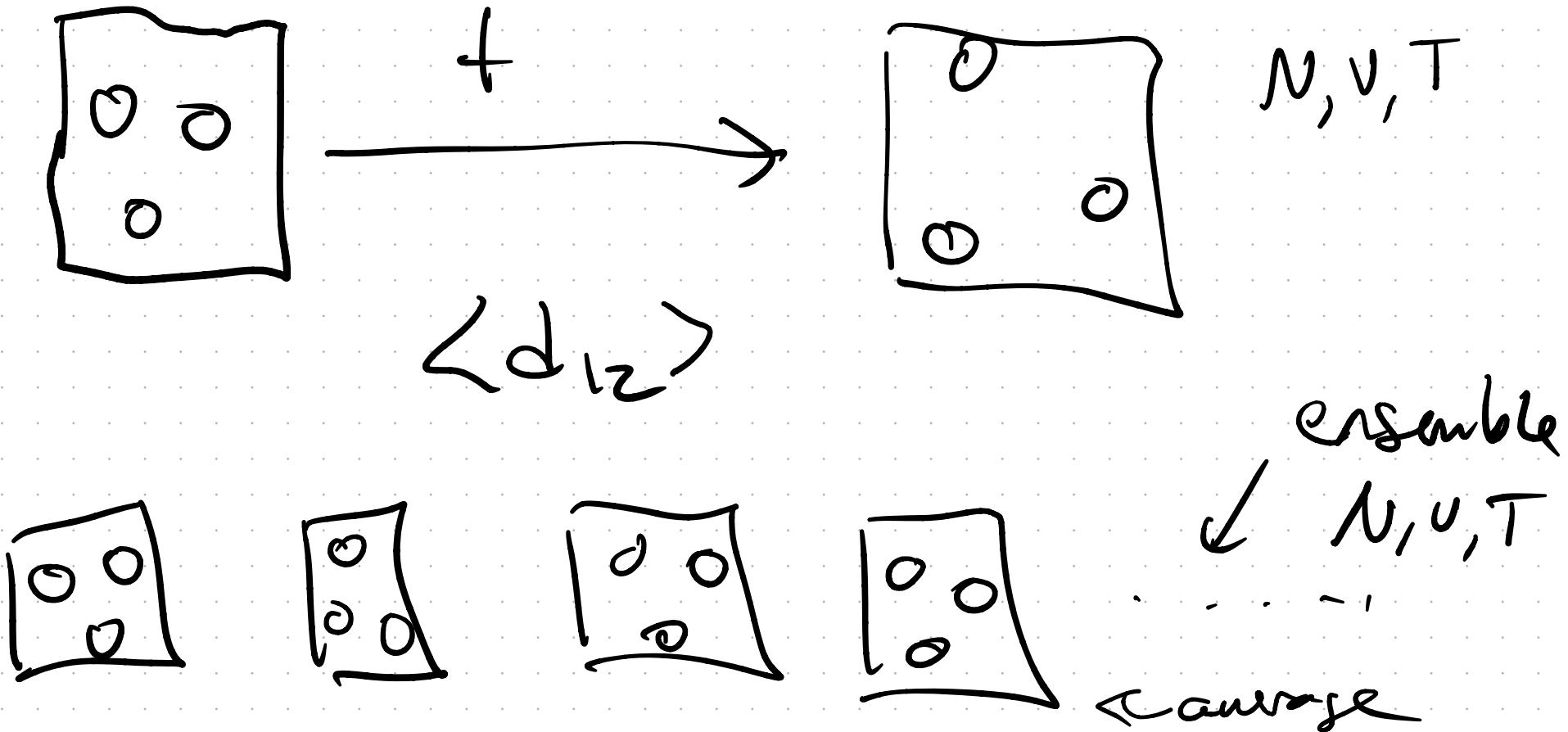
Ergodicity explores all accessible states in ∞ time

Ergodic hypothesis systems are big enough that they're effectively ergodic

Billiard Ball Problem



Ensemble Method If you have many copies
of a system
average a property over that ensemble
Same answer as 1 system for $t \rightarrow \infty$



have been talking about continuous systems
meaning , $\vec{x}, \vec{y}, \vec{z}$

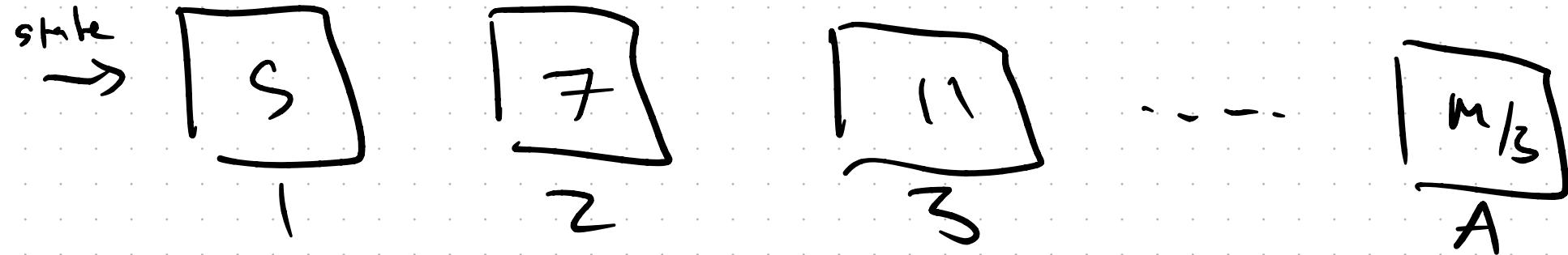
Start with discrete systems

discrete number of possible states

Consider a system that can be

in M microstates

A total copies





N_i is the # in state i

$$\sum_{i=1}^3 N_i = A \quad \leftarrow$$

$$\{N_i\} = \{N_1, N_2, \dots, N_m\}$$

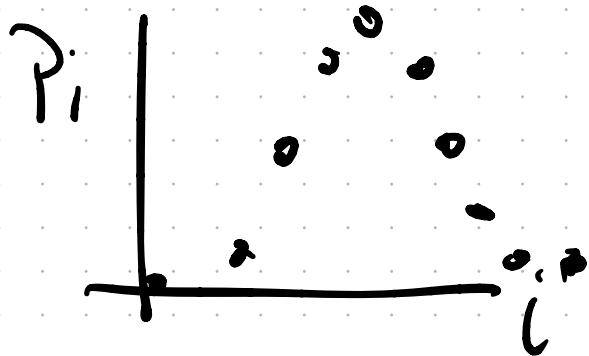
$$\{p_i\} = \{p_1, \dots, p_m\} = \left\{ \frac{N_1}{A}, \frac{N_2}{A}, \dots, \frac{N_m}{A} \right\}$$

$$m = 6$$



Ensemble Average

Property y_i for each i



$$\langle y \rangle_{\text{ensemble}} = \frac{1}{m} \sum_{i=1}^m y_i P_i$$

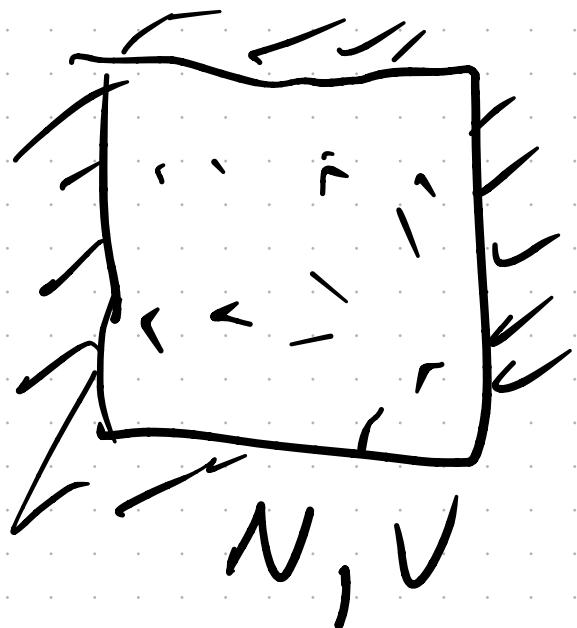
Contrast to a time average

$$y(t_1), y(t_2), \dots, y(t_{N_t})$$

$$\langle y \rangle_{\text{time}} = \frac{1}{N_t} \sum_{i=1}^{N_t} y(t_i)$$

Microcanonical Ensemble

Ensemble where every copy has
the same N, U, E - isolated system



all molecules follow
Newton's equations

$$F = ma$$

or Schrödinger equation

Both case E is constant

Learned previously that entropy
 $S = k \ln \omega$ is a thermodynamic potential for N, V, E systems

S is maximized at equilibrium
Copies of this system maximize entropy

$\Rightarrow P_1, P_2, \dots, P_m$ for our system

$$S = k_B \ln \omega$$

N copies and N_1 in state 1
 N_2 in state 2 ...

$$\omega = \frac{A!}{N_1! N_2! N_3! N_4! \dots N_m!} \quad (\text{multinomial distribution})$$
$$\sum_{i=1}^m N_i = A \quad A \& \{N_i\} \text{ large}$$

Sterling's approximation
is okay

$$\omega = \frac{A!}{N_1! N_2! N_3! N_4! \cdots N_m!}$$

$$\ln A! \approx A \ln A - A$$

$$S = k_B \ln \omega = (A \ln A - \sum_{i=1}^m (N_i \ln N_i - N_i)) / k_B$$

$$S/k_B = A \ln A - \underbrace{\sum_{i=1}^m N_i \ln N_i}_{\text{---}}$$

$$\frac{\partial S/k_B}{\partial N_i} = 0$$

$$\frac{S}{k_B} \approx A \ln A - \underbrace{\sum_{i=1}^m N_i \ln N_i}_{A}$$

$$A = \sum_{i=1}^m N_i$$

$$\frac{\partial S/k_B}{\partial N_j} = A \frac{\partial \ln A}{\partial N_j} + \frac{\partial A}{\partial N_j} \ln A$$

$$- \frac{N_j}{N_j} - 1 \cdot \ln N_j$$

$$\frac{\partial A}{\partial N_j} = 1$$

$$= 1 + \ln A - 1 - \ln N_j$$

$$= \ln A - \ln N_j = \ln \left(\frac{A}{N_j} \right)$$

Maximize
 N_j to get

$$S \rightarrow 0$$

Want to maximize all N_j

but $\sum_{j=1}^m N_j = A \leftarrow \text{constraint}$

Method of Lagrange Multipliers

$f(x)$ ← want to find max or min
with a constraint

Max
or min $I = f(x) - \alpha(\text{constraint})$

$$S = k \ln \omega \quad \text{w/ constraint}$$

$$\sum_{i=1}^m N_i = A \Rightarrow \left(\sum_{i=1}^m N_i - A \right) = 0$$

$$\bar{I} = k \ln \omega - \alpha \left(\sum_{i=1}^m N_i - A \right)$$

$$\frac{\partial \bar{I}}{\partial N_j} = \ln A - \ln N_j - \alpha = 0 \quad \text{for every } N_j$$

$$\alpha = \ln \left(\frac{A}{N_j} \right) \Leftrightarrow N_j = A e^{-\alpha}$$

$$N_j = A e^{-\alpha} \quad \rightarrow$$

$$\sum_{i=1}^m N_i - A = 0 \leftarrow \text{constraint}$$

$$\sum_{i=1}^m A e^{-\alpha} = A \Rightarrow m A e^{-\alpha} = A$$

$$e^\alpha = m \quad \alpha = \ln(m)$$

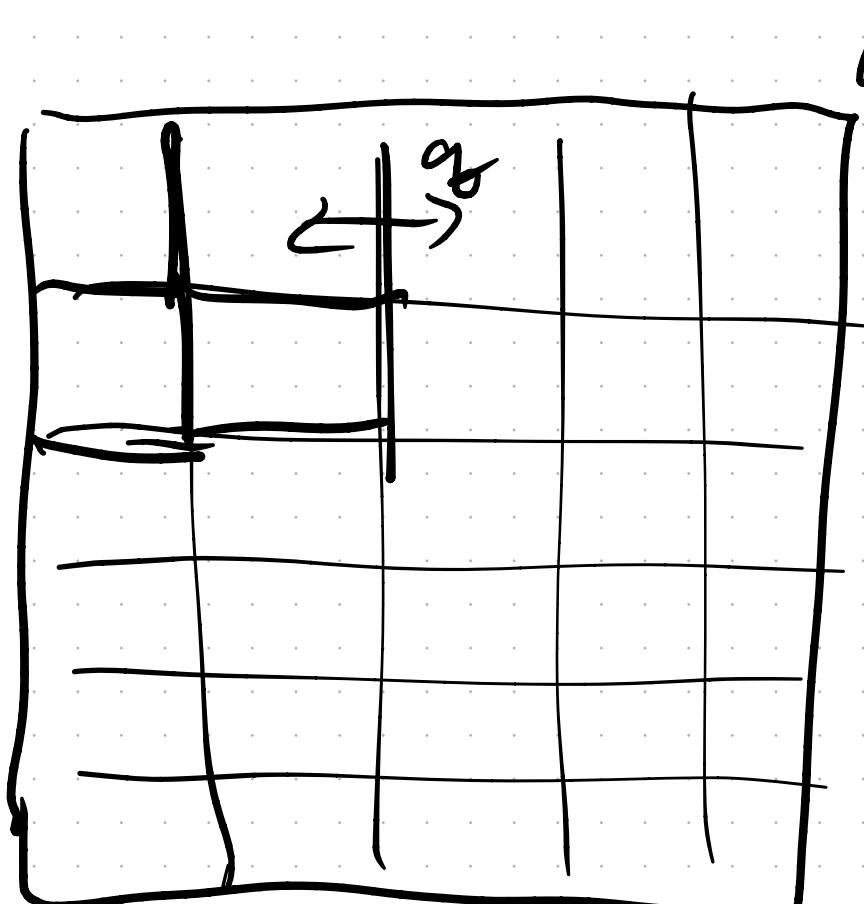
$$P_i = \frac{N_i}{A} = \frac{A e^{-\alpha}}{A} = \frac{1}{e^\alpha} = \frac{1}{m}$$

Chapter 10 Canonical Ensemble

a system of interest has constant

$$N, V, T$$

isolated



$$N = A \tilde{N} \quad V = A \tilde{V} \quad \varepsilon = \sum_i^A \varepsilon_i$$

N, V, ε overall
heat transfers
until each
System has same T

one system @ eq

$$\tilde{N}, \tilde{V}, T$$

Whole system is N, V, E

Maximize $S = k \ln \omega$

constraints

$$(1) \sum_{i=1}^m N_i = A$$

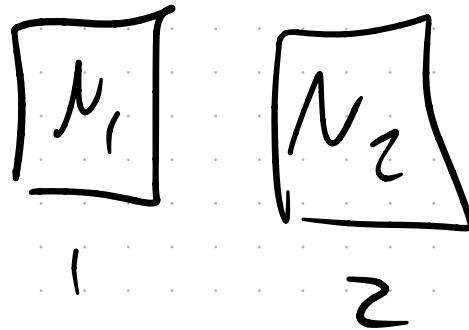
$$(2) \sum_{i=1}^m \varepsilon_i = \varepsilon_{\text{total}} \quad \text{constant}$$

Lagrange multipliers (2 constraints)

$$S = k \ln \omega - \alpha (\sum N_i - A) - \beta (\sum \varepsilon_i - \varepsilon_{\text{total}})$$

Energy constraint

$$E_{\text{total}} = \sum_{i=1}^A N_i \varepsilon_i$$



$$E_{\text{total}} = \sum_{i=1}^M N_i \varepsilon_i$$

$$S = k \ln \omega - \alpha (\sum N_i - A) - \beta (\sum N_i \varepsilon_i - E_{\text{total}})$$

$$\frac{\partial S}{\partial N_j} = \ln(A/N_j) - \alpha - \beta \varepsilon_j = 0$$

$$\ln(A/N_j) - \alpha - \beta \varepsilon_j = 0$$

$$N_j = e^{-\alpha} e^{-\beta \varepsilon_j} \cdot A$$

$$\sum_{j=1}^m N_j = A = A e^{-\alpha} \sum_{j=1}^m e^{-\beta \varepsilon_j}$$

$$1 = e^{-\alpha} \sum_{j=1}^m e^{-\beta \varepsilon_j}$$

$$e^\alpha = \sum_{j=1}^m e^{-\beta \varepsilon_j}$$

$$P_j = \frac{N_j}{A} = e^{-\alpha - \beta \varepsilon_j}$$

$$= e^{-\beta \varepsilon_j} / \sum_{j=1}^m e^{-\beta \varepsilon_j}$$

$$P_j = \frac{e^{-\beta \epsilon_j}}{\sum_{j=1}^m e^{-\beta \epsilon_j}}$$

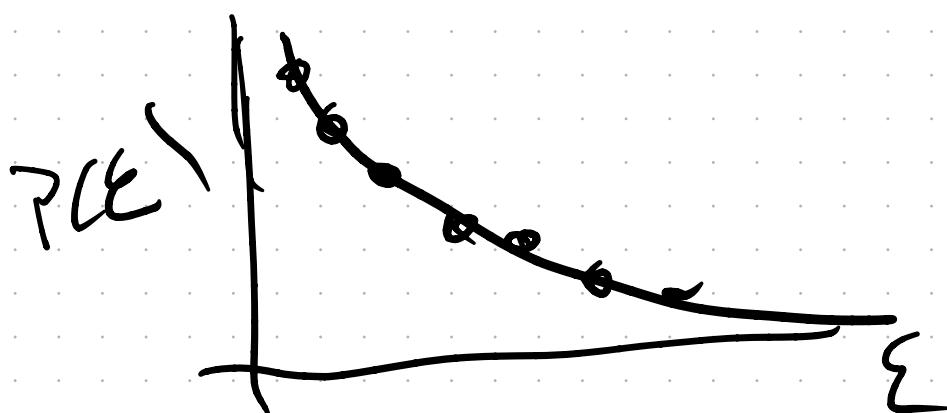
*

$$Q = \sum_{j=1}^m e^{-\beta \epsilon_j}$$

"partition function"

$e^{-\beta \epsilon_j}$ is called

Boltzmann factor
(relative weight)



for Micro Canonical

Boltzmann factor = 1 like $\beta = 0$

$$Q = \sum_{i=1}^m \beta F_i = m$$

$$\langle g \rangle = \sum_{i=1}^m g_i P_i = \sum_{i=1}^m g_i e^{-\beta E_i} / Q$$

$$\langle E \rangle = \sum_{i=1}^m \epsilon_i e^{-\beta \epsilon_i} / Q$$

$$\beta = 1/k_B T$$

$$Q = \sum_{i=1}^m e^{-\beta \epsilon_i}$$

can derive thermodynamic quantities
from the partition function

$$\langle \epsilon \rangle = \sum_{i=1}^m \epsilon_i e^{-\beta \epsilon_i} / Q$$

$$-\frac{\partial Q}{\partial \beta} = \sum_{i=1}^m \epsilon_i e^{-\beta \epsilon_i}$$

$$\frac{\partial \log Q}{\partial x} = \frac{1}{Q} \frac{\partial Q}{\partial x}$$

$$-\frac{\partial \ln Q}{\partial \beta} = \sum_{i=1}^m e_i e^{-\beta e_i} = \langle e \rangle$$

$$\langle e \rangle = -\frac{\partial \ln Q}{\partial \beta}$$

Gibbs Entropy

$$S = k \ln W_{\text{micro}}$$

$$S = -k_B \sum_{i=1}^m p_i \ln p_i$$

$$\sum p_i y_i = \langle y \rangle$$

$$p_i = e^{-\beta E_i} / Z$$

$$= -k_B \sum p_i [-\beta E_i - \ln Q]$$

$$= k_B \beta \sum p_i E_i + k_B \ln Q$$

$$= k_B \beta \langle E \rangle + k_B \ln Q \quad \text{← entropy}$$

$$S = k_B \beta \langle \epsilon \rangle + k_B \ln \theta$$

$$T = \left(\frac{\partial U}{\partial S} \right)_V = -$$