Constant pressure, constant chemical pit. Before, me studied const N, v, E Sys N,V,T

Maximize entropy of whole thing

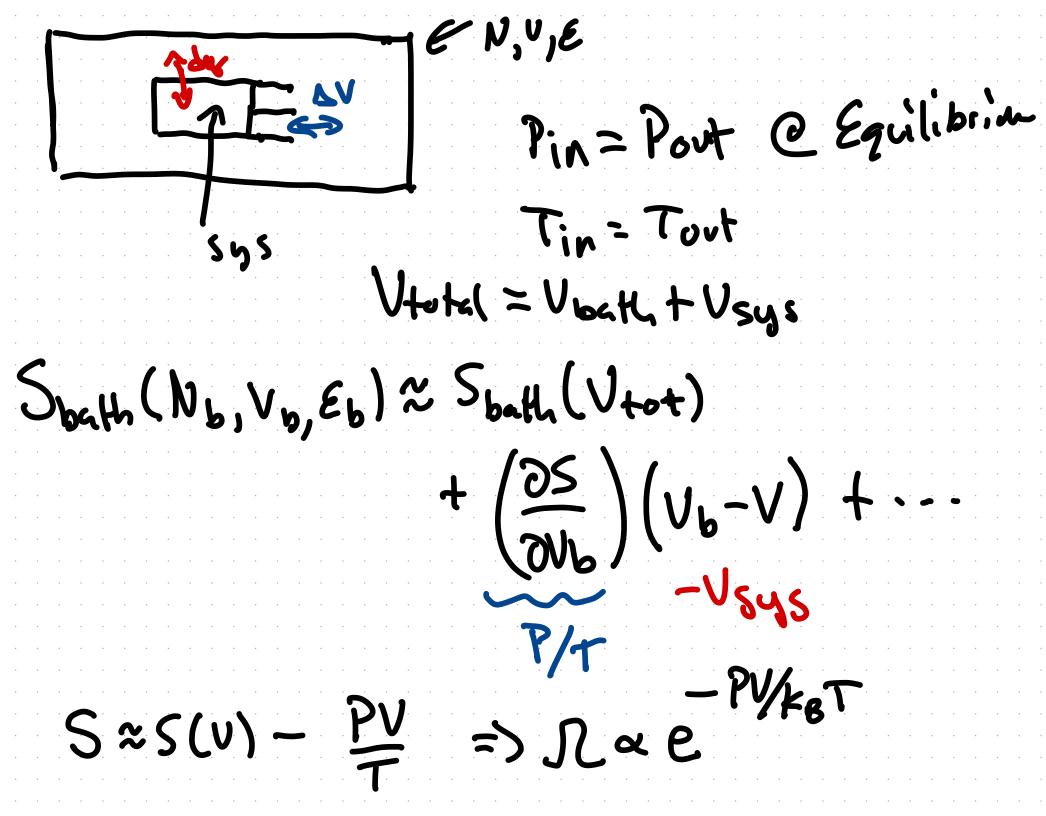
get Tin=Tout P(Xsysten)=e 2

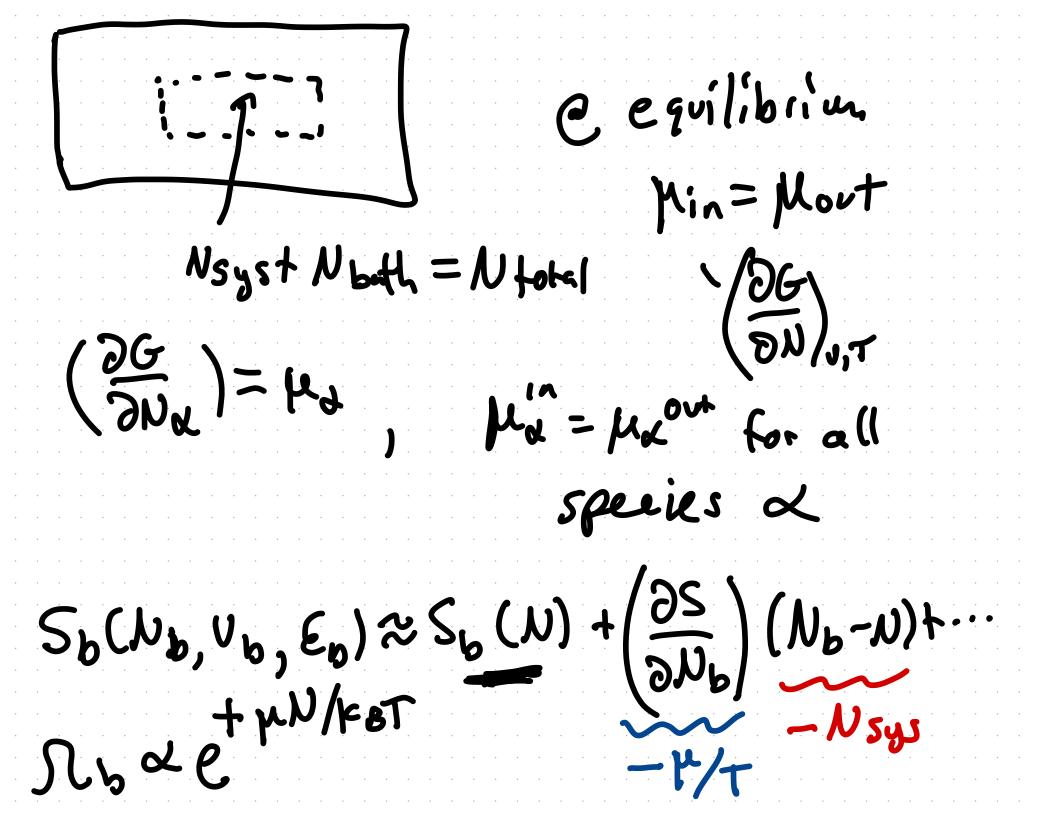
Reminder: E+++= = Eb + Es Stor-Eson 20 Sbath (Nbath, Ubath, Ebath) = Sbath (Etot) _ Esys + (356) (Eb-E+4) Sbath= koln Rbath Sbath/kg - Esys/kgTbah

Const

Const

Const





Sb(N₁, V₃, E₆)
$$\approx S(N, V, E) + \begin{pmatrix} \partial S \\ \partial N \end{pmatrix}(N-N_0)$$

$$+ \begin{pmatrix} \partial S \\ \partial E \end{pmatrix} (E-E_6) = \\ + \begin{pmatrix} \partial S \\ \partial V \end{pmatrix} (V-V_6) + ...$$
Get N, P, T
$$Sb = PV/k_8T - E/k_8T$$
Prob in NPT 1S
$$Prob = NPT + NP/k_8T$$

$$Prob = PV/k_8T + NP/k_8T$$

Const N, PT
$$\Delta(N,P,T) = C \int_{0}^{\infty} dV \int_{0}^{\infty} dX e^{-\beta(H(K)+PV)}$$

$$C = L^{2N}V_{0} \qquad \frac{1}{V_{0}} \int_{0}^{\infty} dV e^{\beta H} \mathcal{I}(U,U,T)$$

$$P(X,V) \sim e^{-\beta PU+H(H)} \int_{0}^{\infty} I(U,U,T)$$

$$G = E-TS+PU = A+PV \int_{0}^{\infty} I(U,U,T) \int_{$$

Grand Canonical (µ, V, T) $Z = \sum_{N=0}^{\infty} \int_{0}^{\infty} dx e^{-\beta \mathcal{X}(x) + \beta \mu N}$ Grand $\Omega(X) = -k_BT \log Z$. Remove constraint on Nor V, can have fluctuations in NorV

$$N_i P_i T$$
, V can vary
$$Var(V)$$

$$\langle V \rangle = \left(\frac{\partial f}{\partial P}\right)_{N_i, T} = -k_B T \left(\frac{\partial}{\partial P} \log \mathcal{L} V e^{-\frac{\beta^2 V}{N_i}} V_{N_i, V_i, T_i} \right)$$

$$= -k_{S}T \cdot \frac{1}{\Delta} \cdot \frac{\partial \Delta}{\partial P} = \frac{-k_{S}T}{\Delta} \int_{0}^{\infty} U(-\beta v) e^{-\beta PV} dv$$

$$= \int_{0}^{\infty} \frac{1}{\Delta} \left[Ve^{-\beta PV} \mathcal{L}(N, V, T) \right] = \langle V \rangle$$

Grand canonical
$$(\mu, \nu, T)$$

 $iZ_1 = \sum_{n=0}^{\infty} e^{i\beta\mu N} \mathcal{N}(N, \nu, T)$
 $N=0$
 $(N) = \sum_{n=0}^{\infty} Ne^{i\beta\mu N} \mathcal{N}(N, \nu, T)$
 $N=0$
 $iZ_1 = \sum_{n=0}^{\infty} Ne^{i\beta\mu N} \mathcal{N}(N, \nu, T)$
 $iZ_1 = \sum_{n=0}^{\infty} Ne^{i\beta\mu N} \mathcal{N}(N, \nu, T)$

PV = KBT log Z = <N>KBT
grand potential (-70)
for ideal gas

for an ideal gas
$$Q(\lambda, V, T) = \int d\vec{p} e^{-\vec{p}\cdot\vec{p}^2/2m} \cdot \int_{N!}^{N!} L^{3N}$$

$$A = \int_{2\pi m keT}^{N^2} = \frac{1}{N!} \left(\frac{V}{N^3} \right)^N = \frac{1}{N!} \left(\frac{V}{N^3} \right)^N = \frac{1}{N!} \left(\frac{V}{N^3} \right)^N = \exp \left[e^{\frac{2\pi m}{N^3}} \left(\frac{V}{N^3} \right) \right]^N = \exp \left[e^{\frac{2\pi m}{N^3}} \left(\frac{V}{N^3} \right) \right]^N = \exp \left[e^{\frac{2\pi m}{N^3}} \left(\frac{V}{N^3} \right) \right]^N = \exp \left[e^{\frac{2\pi m}{N^3}} \left(\frac{V}{N^3} \right) \right]^N = \exp \left[e^{\frac{2\pi m}{N^3}} \left(\frac{V}{N^3} \right) \right]^N = \exp \left[e^{\frac{2\pi m}{N^3}} \left(\frac{V}{N^3} \right) \right]^N = \exp \left[e^{\frac{2\pi m}{N^3}} \left(\frac{V}{N^3} \right) \right]^N = \exp \left[e^{\frac{2\pi m}{N^3}} \left(\frac{V}{N^3} \right) \right]^N = \exp \left[e^{\frac{2\pi m}{N^3}} \left(\frac{V}{N^3} \right) \right]^N = \exp \left[e^{\frac{2\pi m}{N^3}} \left(\frac{V}{N^3} \right) \right]^N = \exp \left[e^{\frac{2\pi m}{N^3}} \left(\frac{V}{N^3} \right) \right]^N = \exp \left[e^{\frac{2\pi m}{N^3}} \left(\frac{V}{N^3} \right) \right]^N = \exp \left[e^{\frac{2\pi m}{N^3}} \left(\frac{V}{N^3} \right) \right]^N = \exp \left[e^{\frac{2\pi m}{N^3}} \left(\frac{V}{N^3} \right) \right]^N = \exp \left[e^{\frac{2\pi m}{N^3}} \left(\frac{V}{N^3} \right) \right]^N = \exp \left[e^{\frac{2\pi m}{N^3}} \left(\frac{V}{N^3} \right) \right]^N = \exp \left[e^{\frac{2\pi m}{N^3}} \left(\frac{V}{N^3} \right) \right]^N = \exp \left[e^{\frac{2\pi m}{N^3}} \left(\frac{V}{N^3} \right) \right]^N = \exp \left[e^{\frac{2\pi m}{N^3}} \left(\frac{V}{N^3} \right) \right]^N = \exp \left[e^{\frac{2\pi m}{N^3}} \left(\frac{V}{N^3} \right) \right]^N = \exp \left[e^{\frac{2\pi m}{N^3}} \left(\frac{V}{N^3} \right) \right]^N = \exp \left[e^{\frac{2\pi m}{N^3}} \left(\frac{V}{N^3} \right) \right]^N = \exp \left[e^{\frac{2\pi m}{N^3}} \left(\frac{V}{N^3} \right) \right]^N = \exp \left[e^{\frac{2\pi m}{N^3}} \left(\frac{V}{N^3} \right) \right]^N = \exp \left[e^{\frac{2\pi m}{N^3}} \left(\frac{V}{N^3} \right) \right]^N = \exp \left[e^{\frac{2\pi m}{N^3}} \left(\frac{V}{N^3} \right) \right]^N = \exp \left[e^{\frac{2\pi m}{N^3}} \left(\frac{V}{N^3} \right) \right]^N = \exp \left[e^{\frac{2\pi m}{N^3}} \left(\frac{V}{N^3} \right) \right]^N = \exp \left[e^{\frac{2\pi m}{N^3}} \left(\frac{V}{N^3} \right) \right]^N = \exp \left[e^{\frac{2\pi m}{N^3}} \left(\frac{V}{N^3} \right) \right]^N = \exp \left[e^{\frac{2\pi m}{N^3}} \left(\frac{V}{N^3} \right) \right]^N = \exp \left[e^{\frac{2\pi m}{N^3}} \left(\frac{V}{N^3} \right) \right]^N = \exp \left[e^{\frac{2\pi m}{N^3}} \left(\frac{V}{N^3} \right) \right]^N = \exp \left[e^{\frac{2\pi m}{N^3}} \left(\frac{V}{N^3} \right) \right]^N = \exp \left[e^{\frac{2\pi m}{N^3}} \left(\frac{V}{N^3} \right) \right]^N = \exp \left[e^{\frac{2\pi m}{N^3}} \left(\frac{V}{N^3} \right) \right]^N = \exp \left[e^{\frac{2\pi m}{N^3}} \left(\frac{V}{N^3} \right) \right]^N = \exp \left[e^{\frac{2\pi m}{N^3}} \left(\frac{V}{N^3} \right) \right]^N = \exp \left[e^{\frac{2\pi m}{N^3}} \left(\frac{V}{N^3} \right) \right]^N = \exp \left[e^{\frac{2\pi m}{N^3}} \left(\frac{V}{N^3} \right) \right]^N = \exp \left[e^{\frac{2\pi m}{N^3}} \left(\frac{V}{N^3} \right) \right]^N = \exp \left[e$$

PV =
$$+ k_B T \log \left[e^{RM} \left(\frac{1}{N^3} \right) \right]$$

= $+ k_B T \left[e^{RM} \left(\frac{1}{N^3} \right) \right]$
 $< N > < check$
PV = $< N > k_B T$

Var(V) = kBT 2 <V> N,PT 2P FDT $V_{ar}(n) = k_{8}T \frac{3 < n}{3 n} < \mu_{i}v_{i}T$ = KgT · (N)² KT per atom

| The per atom
| Isothermal compressibility
| Line | Compressibility