

Kinetics & Chemical rates

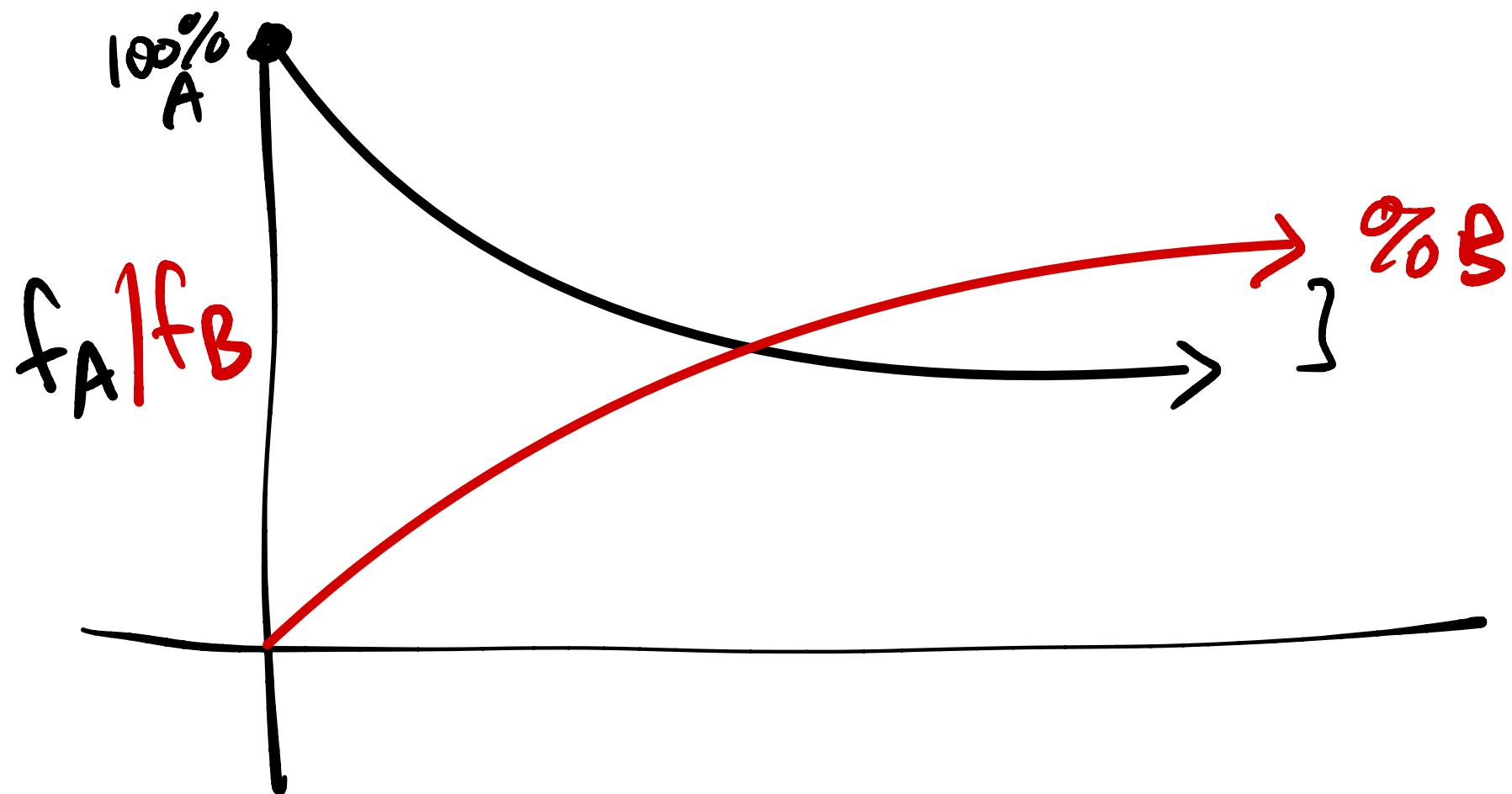
Consider



Chemical equilibrium

Detailed balance:

$$\#A \cdot r_{A \rightarrow B} = \#B \cdot r_{B \rightarrow A}$$



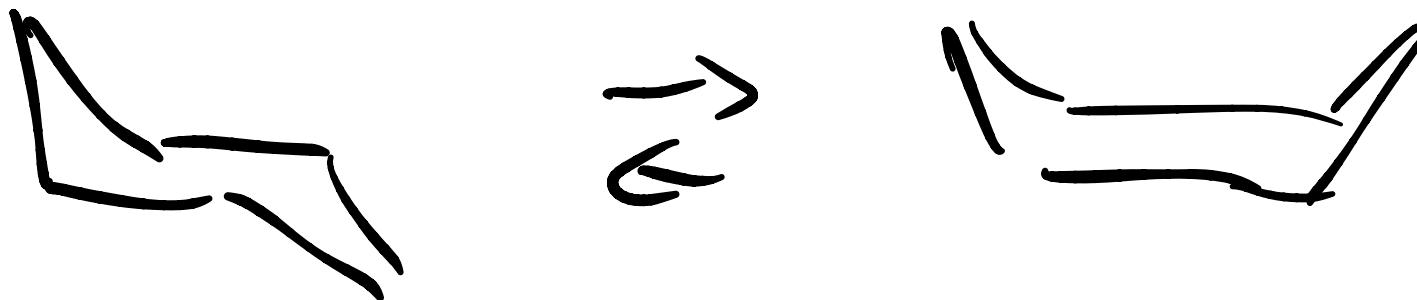
$$[A]_{eq} k_{A \rightarrow B} = [B]_{eq} k_{B \rightarrow A}$$

$$k_{eq} = \frac{[B]_{eq}}{[A]_{eq}} = \frac{k_{A \rightarrow B}}{k_{B \rightarrow A}}$$

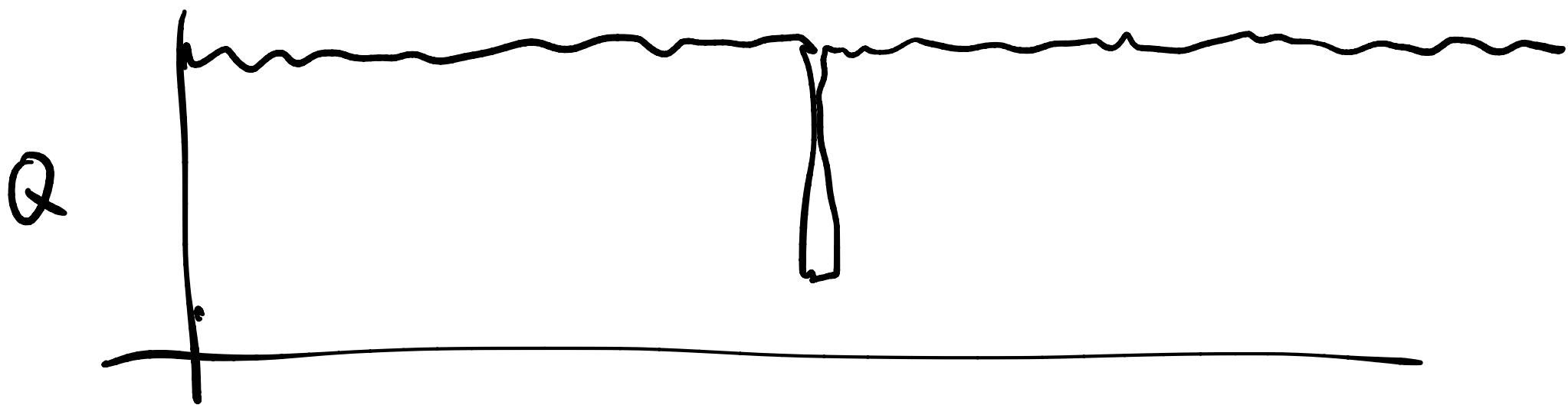
fraction A = $\frac{\# A}{\text{total MC}} = \frac{A/n}{(A+B)/n}$
 $= \frac{[A]}{[A] + [B]}$
 $= \frac{1}{1 + \frac{[B]}{[A]}}$ @ ξ $= \frac{1}{1 + K_{eq}}$

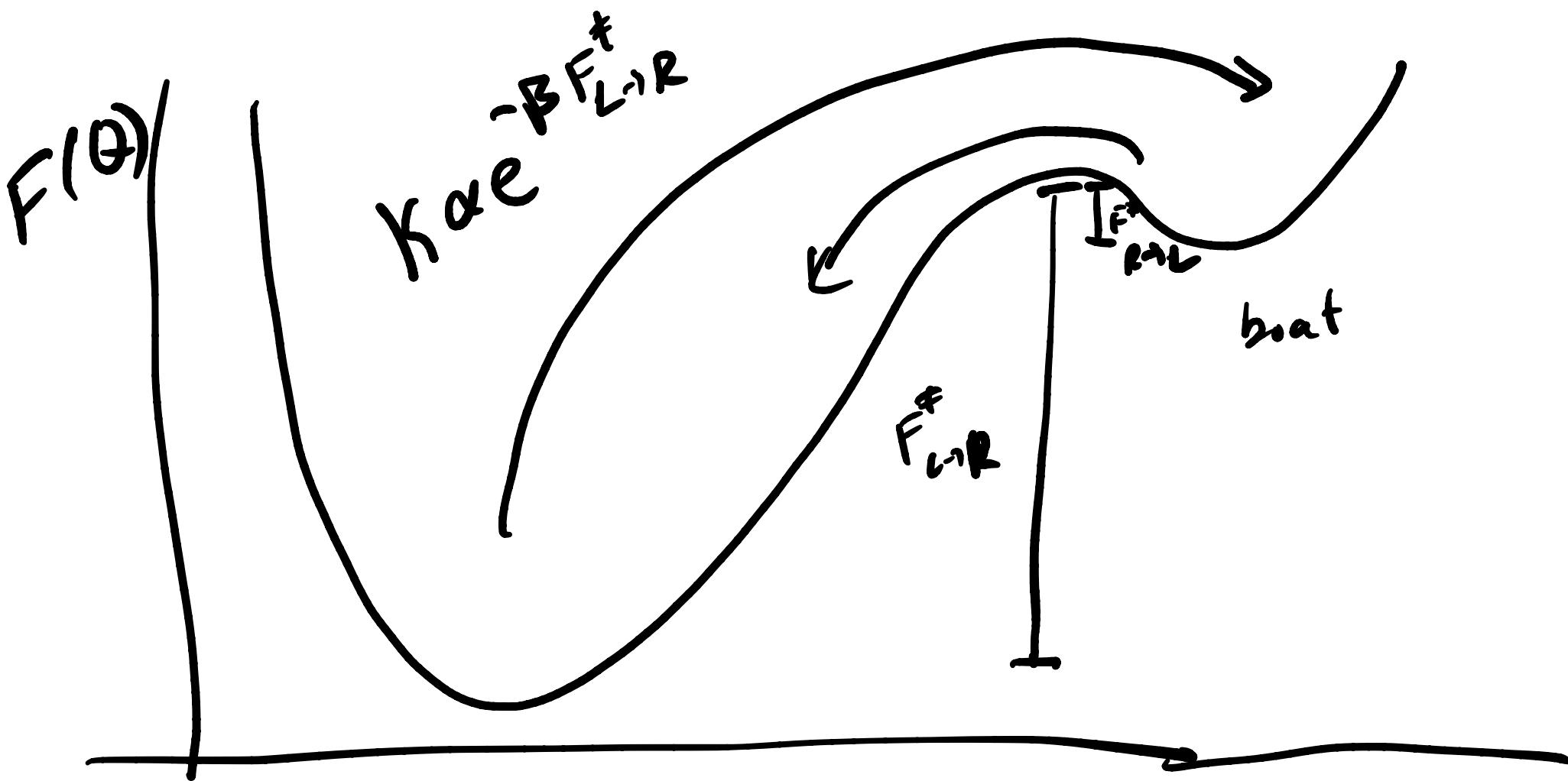
frac B = $1 - \frac{1}{1 + K_{eq}} = \frac{K_{eq}}{1 + K_{eq}} = \frac{1}{1 + 1/K_{eq}}$

frac B(t) = $\frac{1}{1 + \frac{[A]}{[B]}}$



99.99% @ ϵ_g





chain

$$F(Q') = -k_B T \log \int dx S(Q(x)-Q') e^{-\beta u(x)}$$

$$K_{eq} = \frac{k_f}{k_b} \propto \frac{e^{-\beta[F(Q^*) - F(Q_r)]}}{e^{-\beta[F(Q^*) - F(Q_L)]}}$$

$$= \alpha c^{-\beta[\Delta F]} \quad \text{const } N, V, T$$

$$\Delta F = -k_B T \ln(K_{eq}) + \text{const}$$

$$N, P, T \left[\Delta G = -k_B T \ln(K_{eq}) \right]$$





$$\#A_{\text{reg}} - c \quad \#B + c$$

$$C = [A] - [A]_{\text{eq}} = \delta A = -\delta B$$

$$[A] + [B] = \text{const}$$

$$\frac{d[A]}{dt} = \frac{d[A]_{\text{eq}} + C}{dt} = k_{B \rightarrow A}[B] - k_{A \rightarrow B}[A]$$

$$\frac{d[B]}{dt} = \frac{d[B]_{\text{eq}} - C}{dt} = k_{A \rightarrow B}[A] - k_{B \rightarrow A}[B]$$

$$\frac{d[A]}{dt} = \frac{d[\cancel{A_{eq}} + c]}{dt} = k_{B \rightarrow A}[B] - k_{A \rightarrow B}[A]$$

$$\frac{d[B]}{dt} = \frac{d[\cancel{B_{eq}} - c]}{dt} = k_{A \rightarrow B}[A] - k_{B \rightarrow A}[B]$$

Subtract

$$2 \frac{dC}{dt} = 2 [k_{B \rightarrow A}[B] - k_{A \rightarrow B}[A]]$$

$\nwarrow [B_{eq} - c] \qquad \uparrow$

$$k_{B \rightarrow A}[B_{eq}] = k_{A \rightarrow B}[A_{eq}]$$

$[A_{eq} + c]$

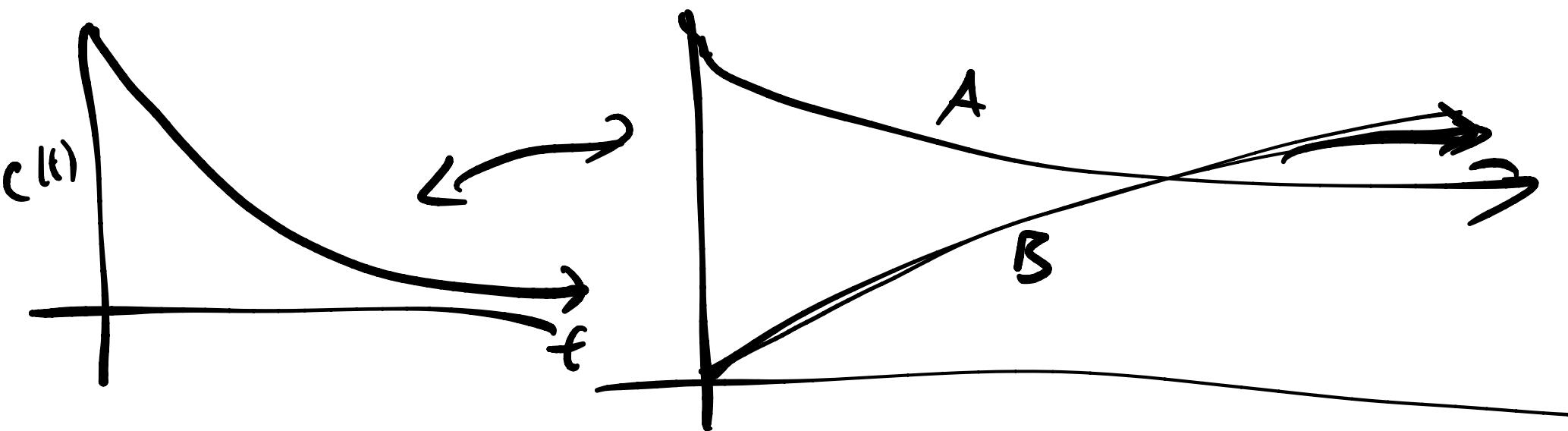
$$= -2(k_{B \rightarrow A} + k_{A \rightarrow B})C$$

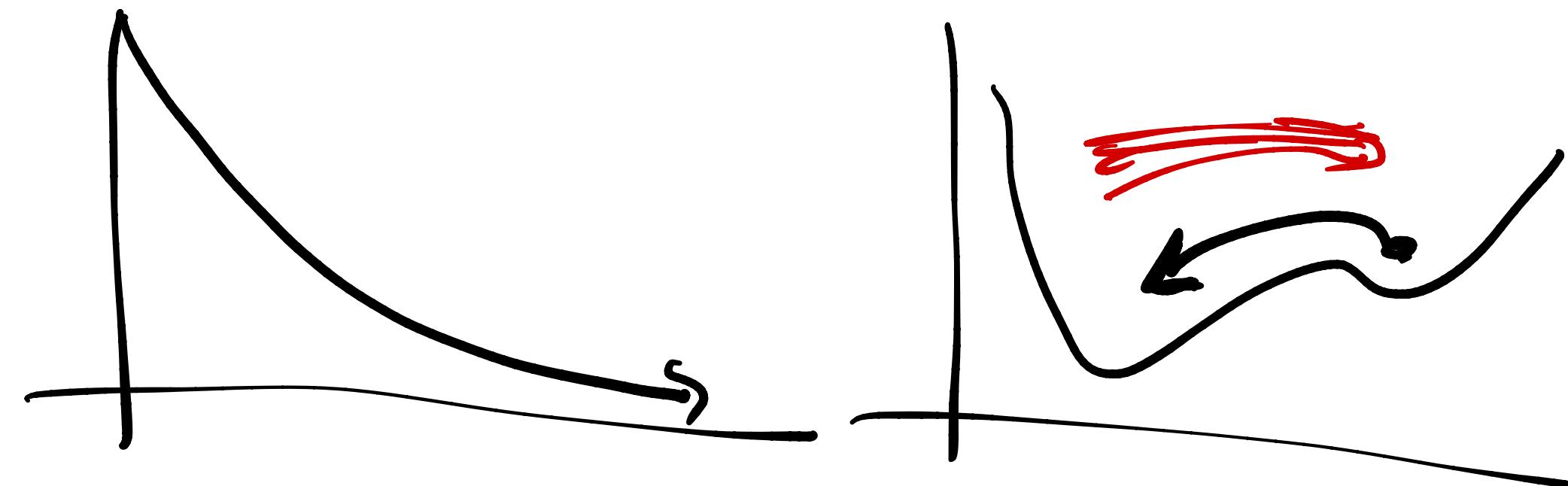
$$\frac{dc}{dt} = - (k_{A \rightarrow B} + k_{B \rightarrow A}) c$$

$$- (k_{A \rightarrow B} + k_{B \rightarrow A}) t$$

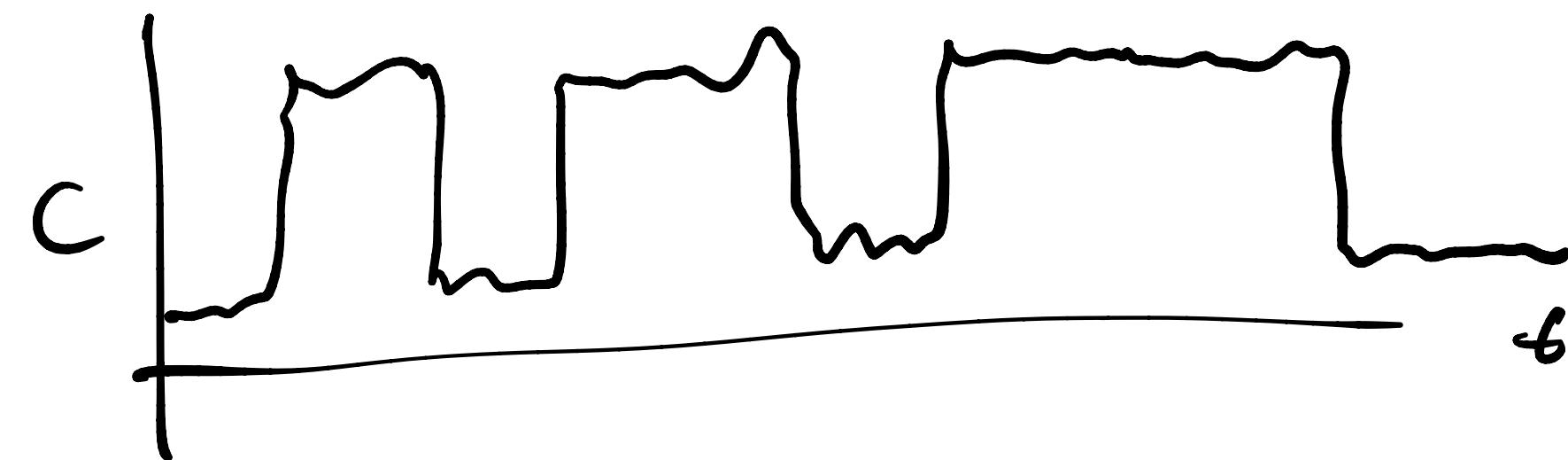
$$c(t) = c(0) e$$

$$\frac{1}{(k_{A \rightarrow B} + k_{B \rightarrow A})} = \tau_{rxn}$$



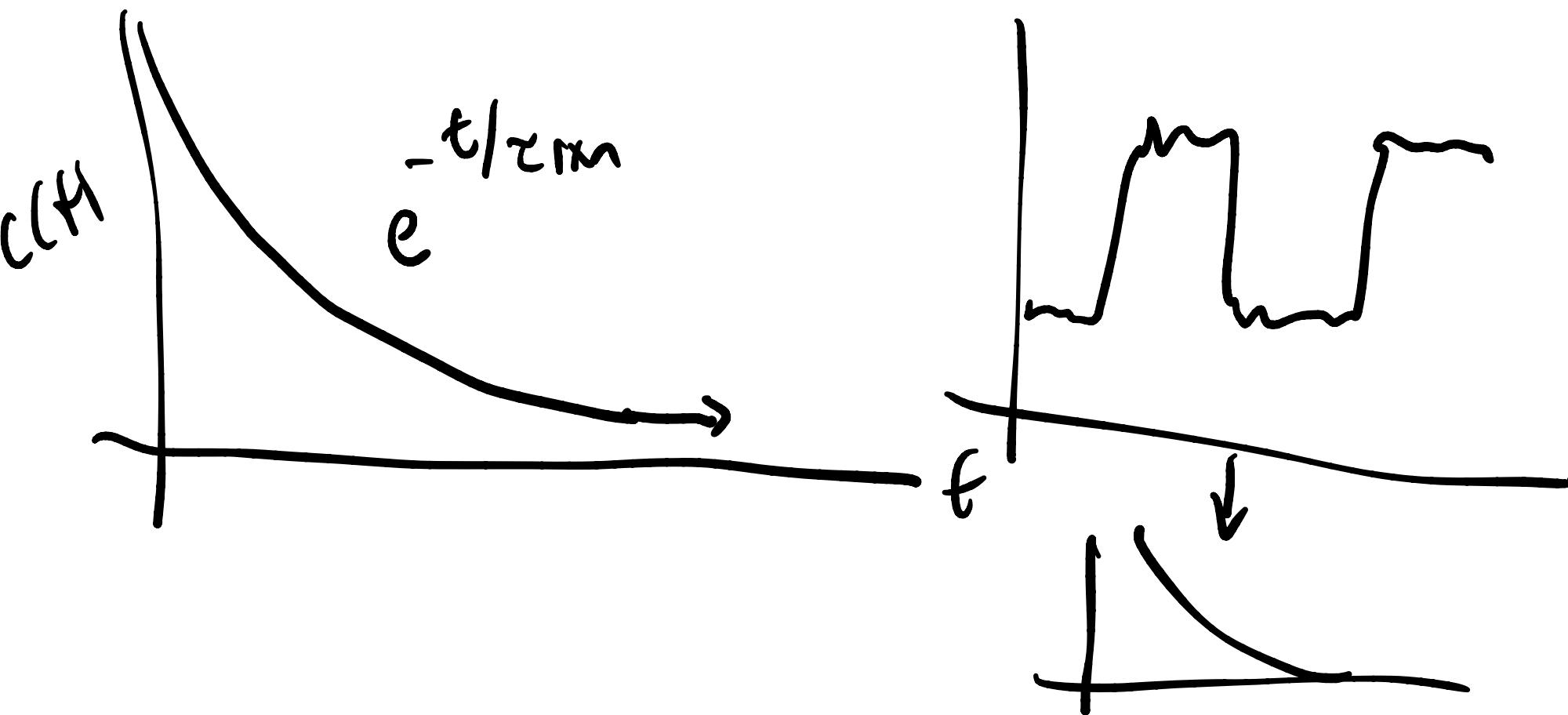


$$\langle c(t)c(t') \rangle = \langle c^2 \rangle_{\text{eq}} e^{-|t-t'|/\zeta}$$



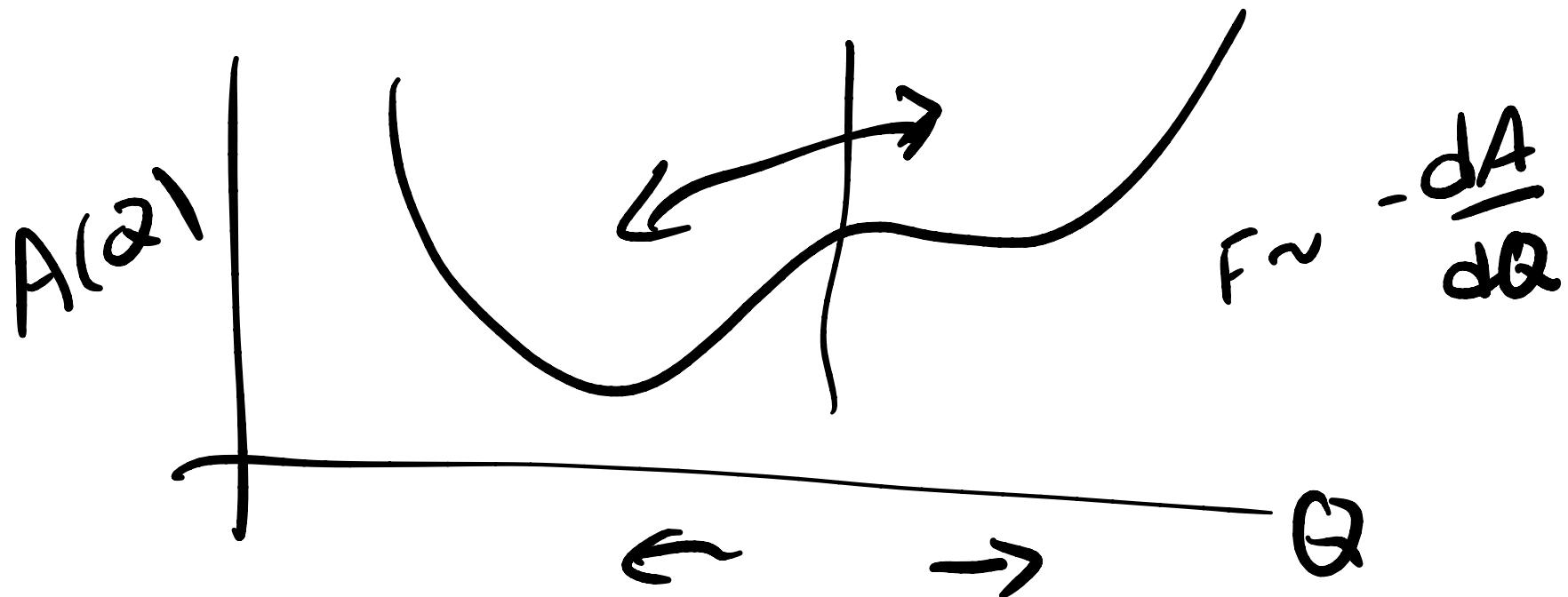
Onsager Regression Hypothesis(1931)

fluctuations relax at the same rate as macroscopic deviations from Eq



$$\langle C(H) C(t') \rangle = \langle \delta C_{\text{eq}}^2 \rangle e^{-(k_f + k_b)(t - t')}$$

$$\frac{dC}{dt} = -(k_f + k_b)C + "SF_{\text{random}}"$$



What is C from microscopic
per species

$$H_A[Q] = \begin{cases} 1 & Q < Q^+ \\ 0 & Q \geq Q^+ \end{cases}$$

$$f_A = \int dQ H_A[Q] e^{-\beta F(Q)} \stackrel{\uparrow}{\langle H_A \rangle} \frac{1}{Z} \equiv \chi_A$$

$$\langle H_A^2 \rangle = \langle H_A \rangle = \chi_A$$

$$\text{Var}_A = \langle \delta H_A^2 \rangle = \langle H_A^2 \rangle - \langle H_A \rangle^2$$

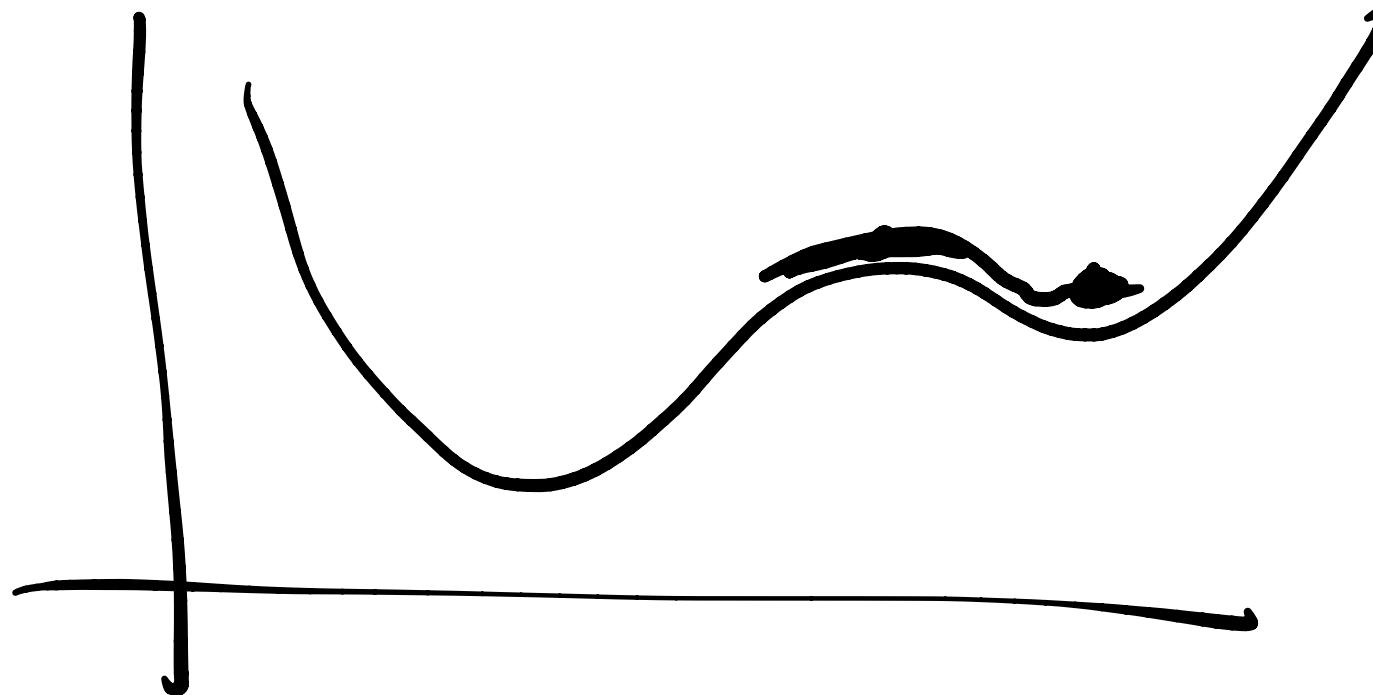
$$\begin{aligned} \uparrow &= \chi_A - \chi_A^2 = \chi_A(1 - \chi_A) \\ &= \chi_A \chi_B \end{aligned}$$

Var_C

$$\begin{aligned} @ \text{Eq} &= \frac{1}{1 + K_{\text{eq}}} \cdot \frac{1}{1 + 1/K_{\text{eq}}} \\ &= K_{\text{eq}} / (1 + K_{\text{eq}})^2 \end{aligned}$$

Transition State Theory

[Chandler]



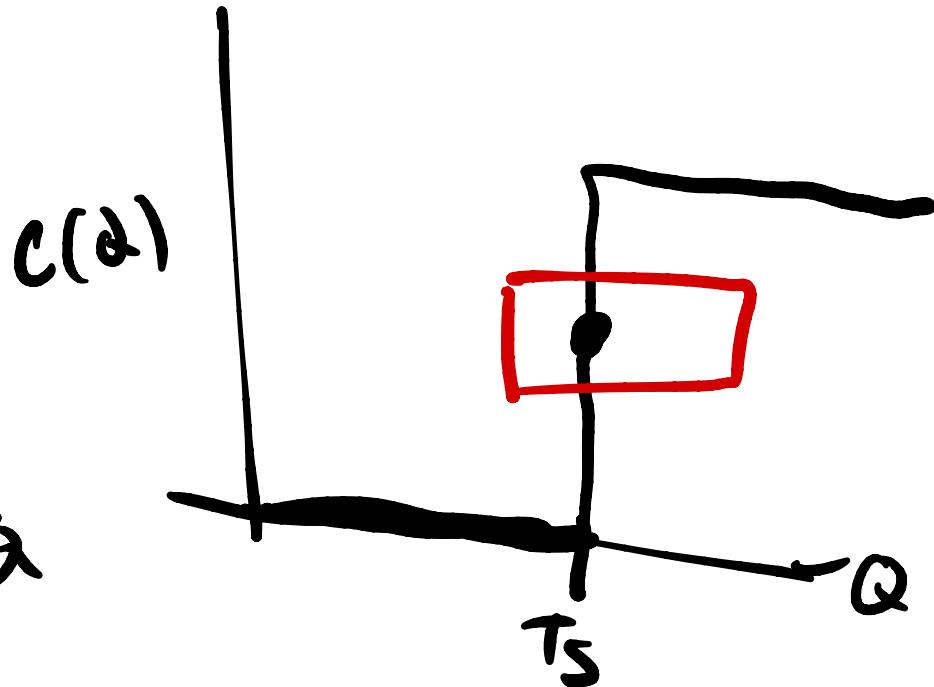
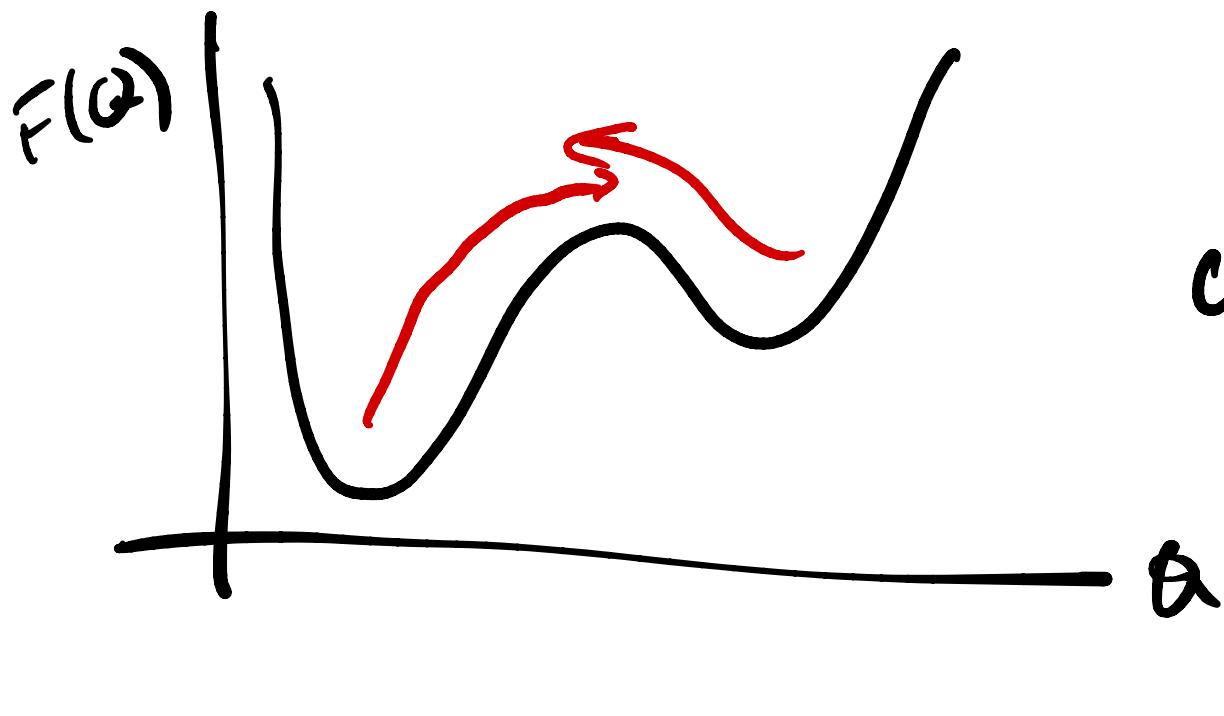
$$K_f = \frac{1}{\chi_A} \left\langle \frac{\partial}{\partial Q} \delta(Q(0) - \bar{Q}) \delta H_B(Q_{\text{TS}}) \right\rangle$$

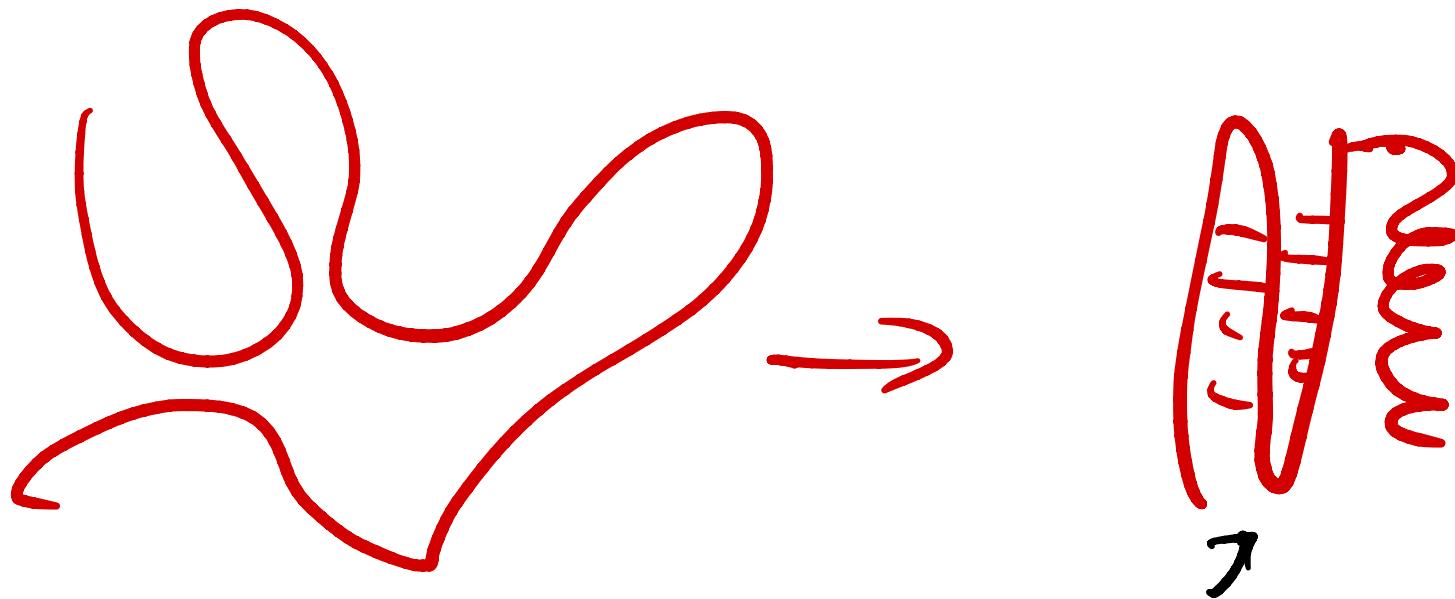
↑
in state b
after some time

Fraction of trajectories that start at
 Q^* and end up in B

before A

Committee $C(Q^*)$

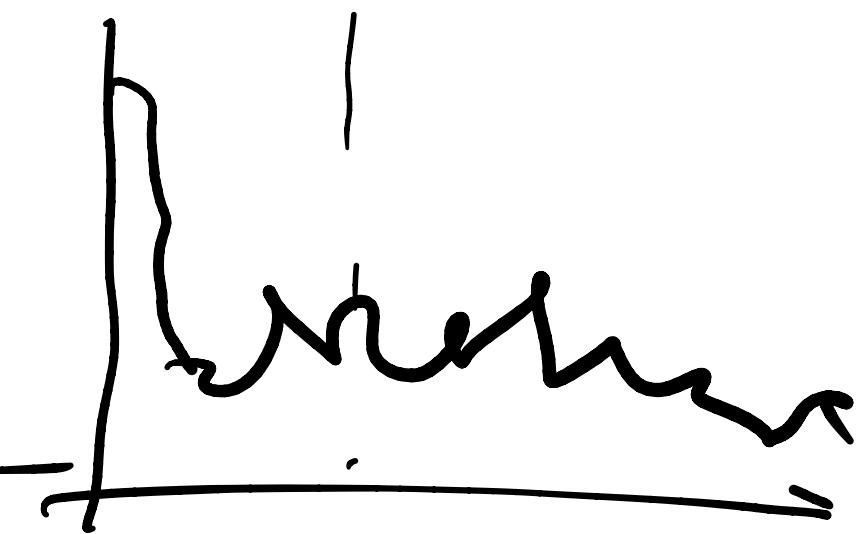


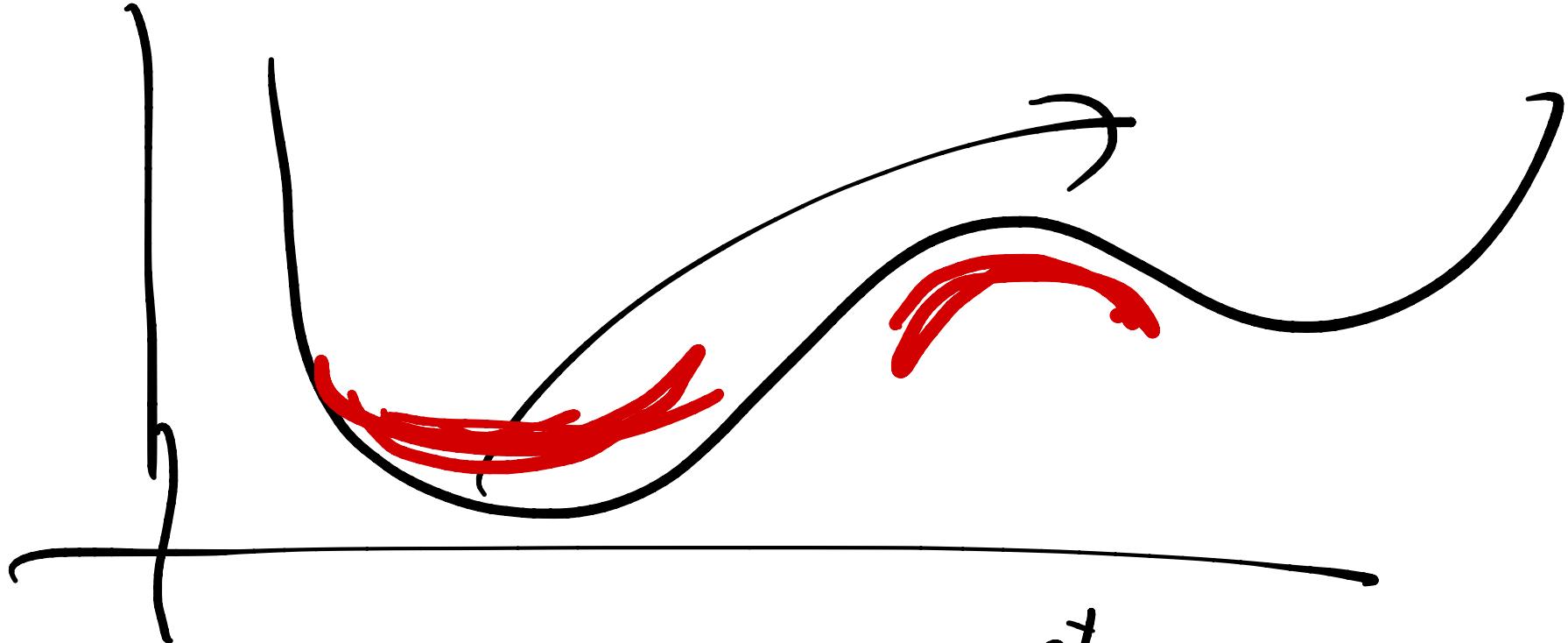


$Q_1 = \text{RMSD to native struct}$

$Q_2 = \text{fraction of native contacts}$







$$k_{A \rightarrow B} \propto e^{-\beta E^+}$$

Fokker-Planck Equations
Smoluchowski

