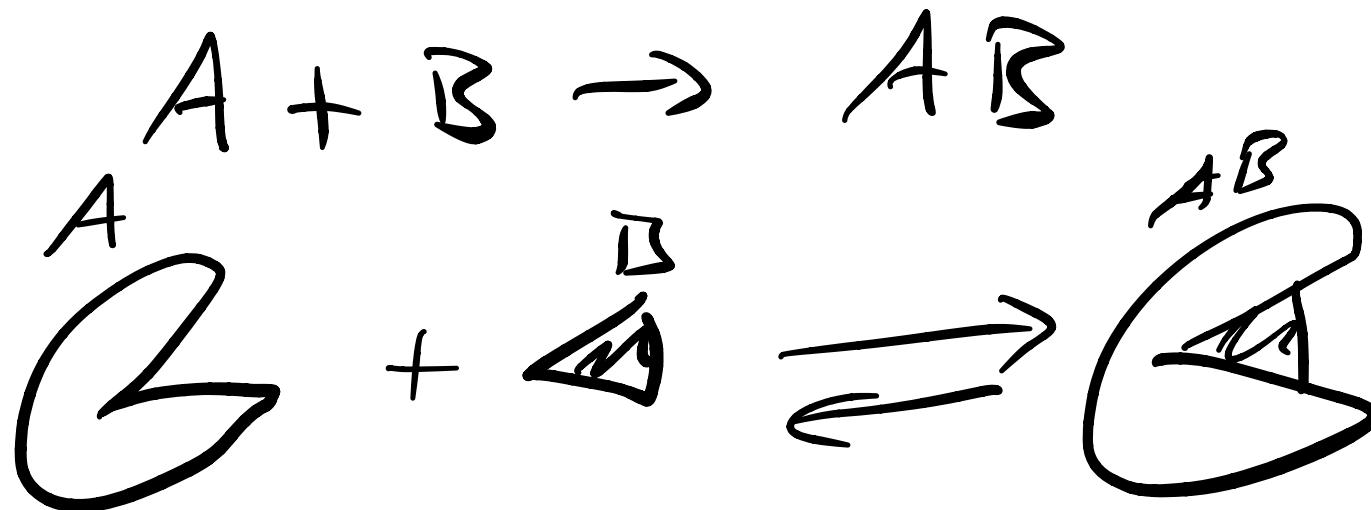
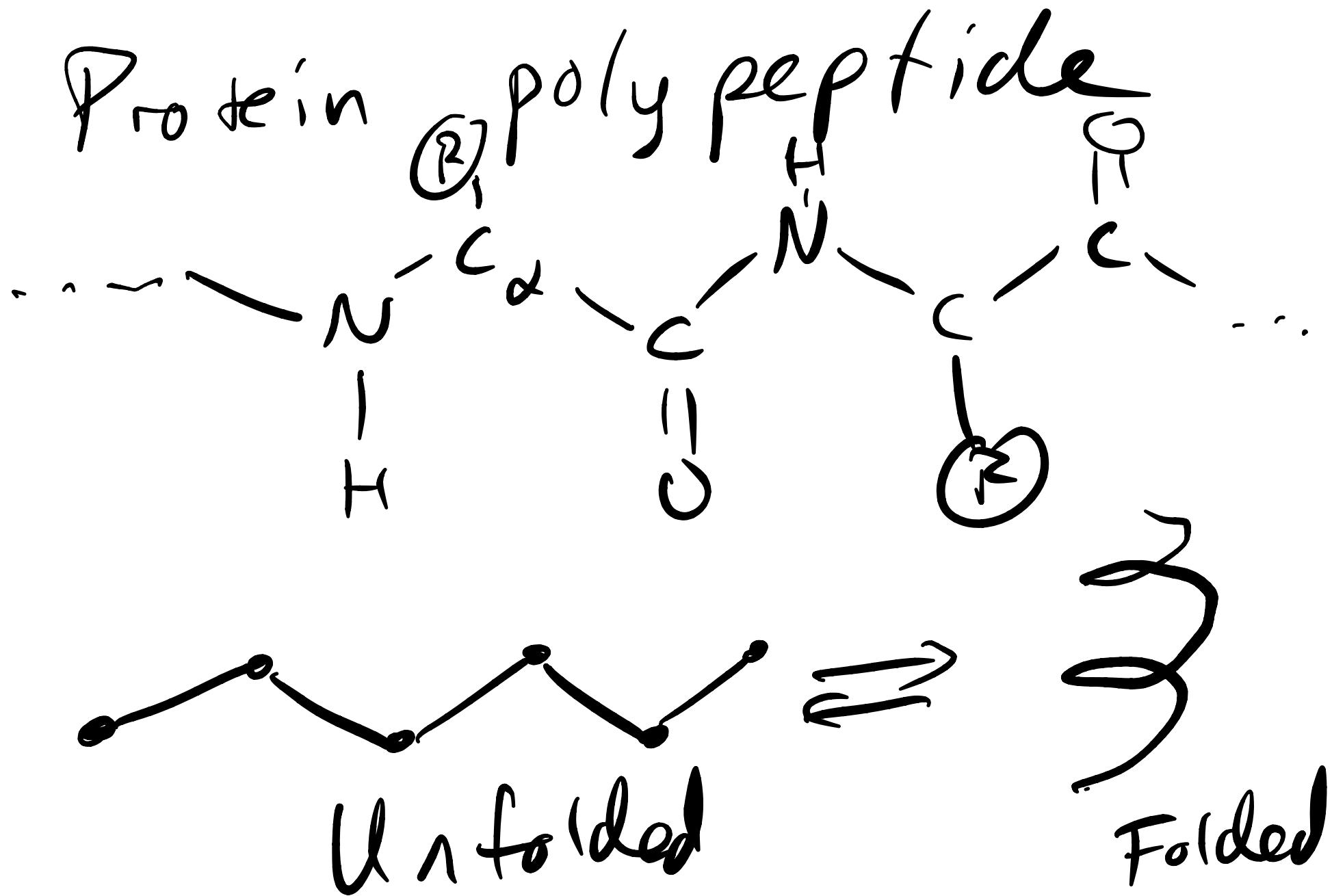


Conformational Equilibrium



$A \rightleftharpoons B$



$$\Delta G_{\text{fold}} = \Delta H - T \Delta S \quad \swarrow$$

< 0 want to be folded

$\Delta H_{\text{fold}} < 0$ more favorable interactions

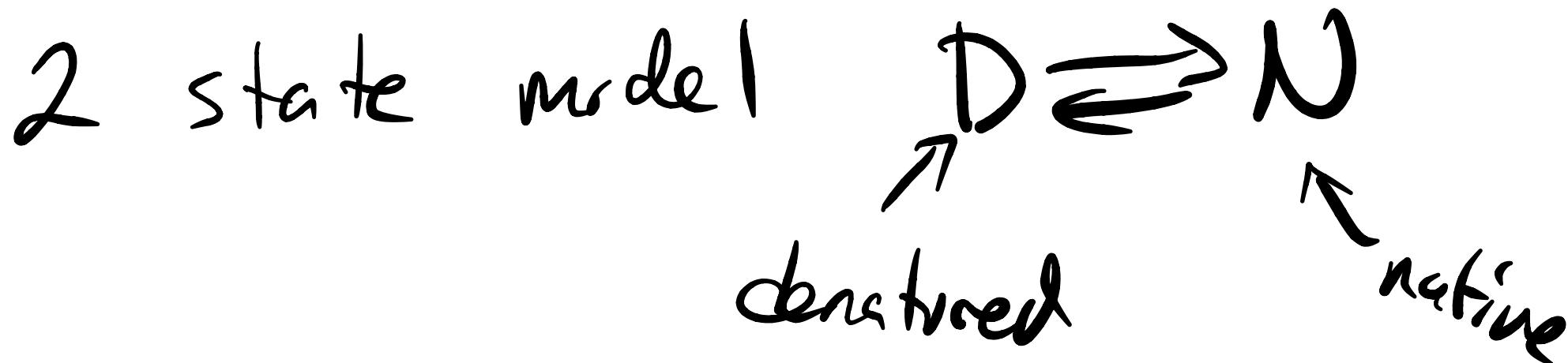
$\Delta S_{\text{fold}} < 0$ "more order"

Folding is a competition

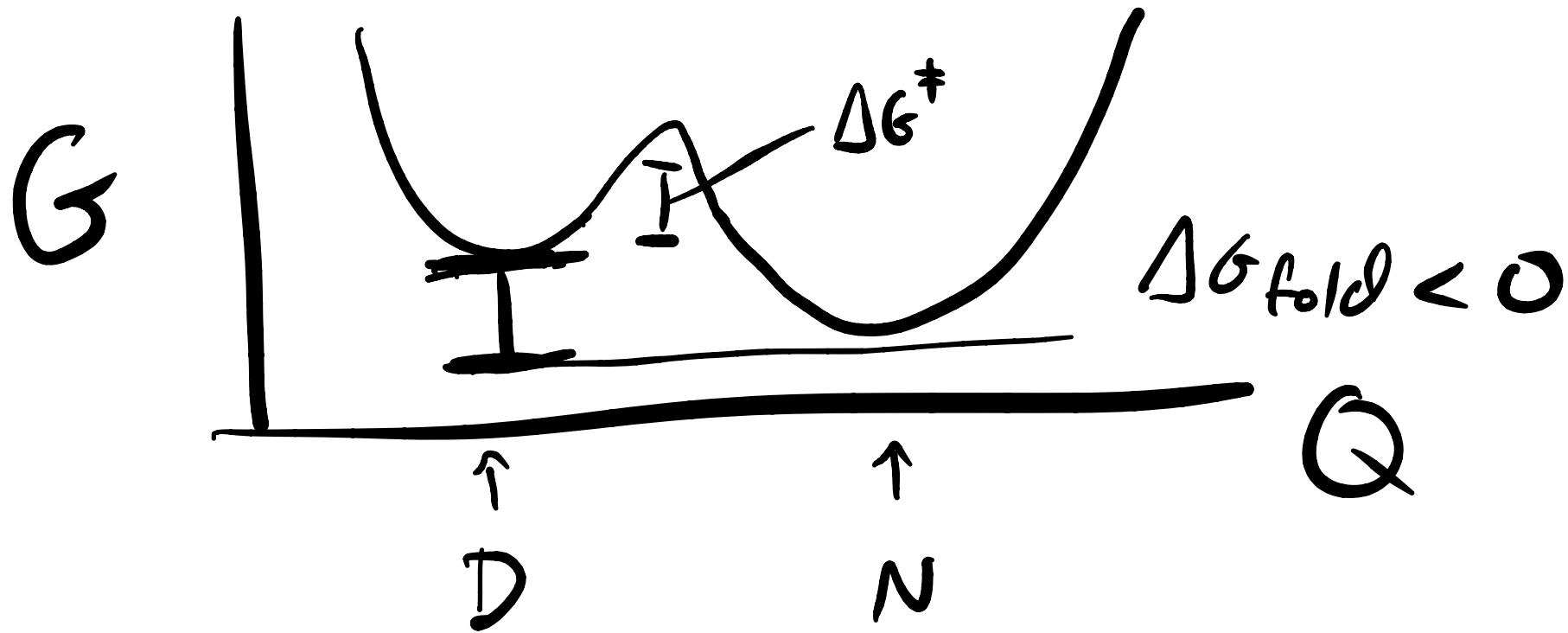
between entropy & enthalpy

$$\Delta G_{\text{folding}} \sim -5 \rightarrow -10 \text{ kcal/mol}$$

Cancellation of a large ΔH
and $-T\Delta S$



$D \rightleftharpoons N$



$Q = \#$ of native contacts

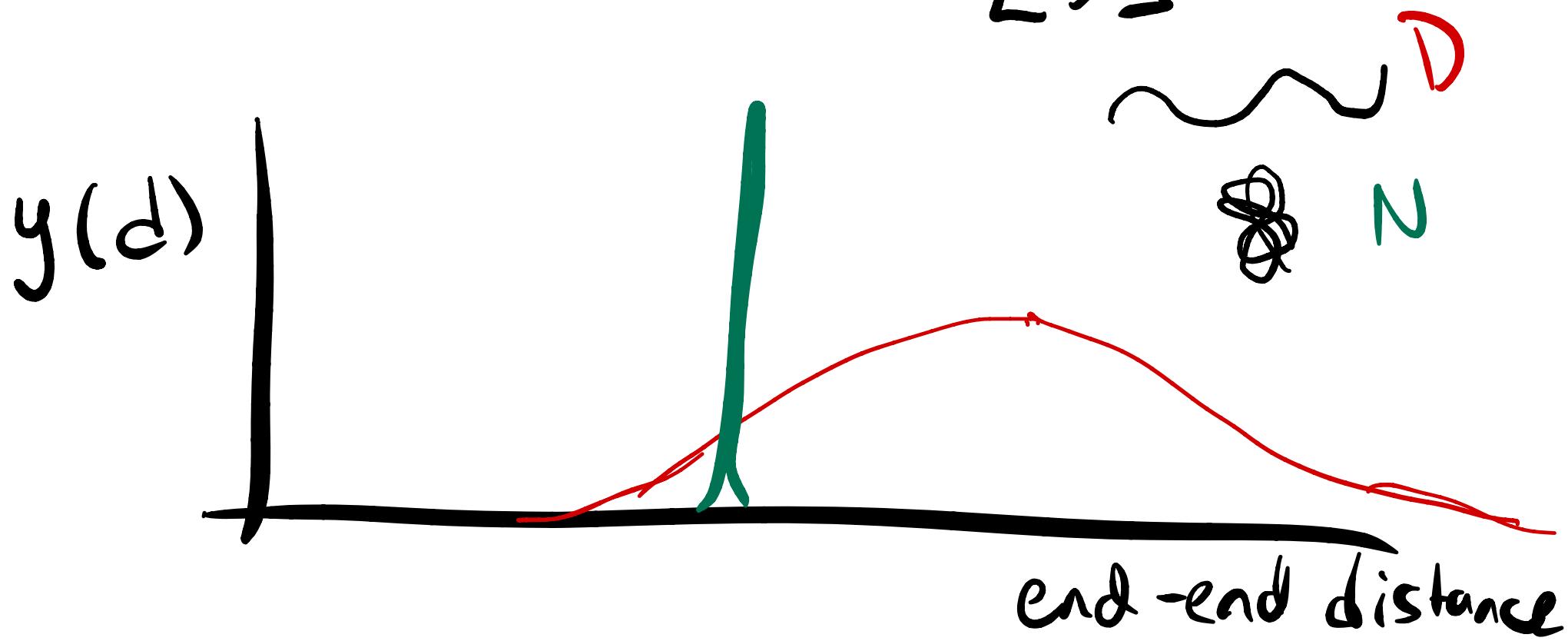
$Q \approx 0$

$Q \approx Q_N$

$$\Delta G_{\text{fold}}^\circ = -RT \ln K_{\text{fold}}$$

$$D \geq N$$

$$K_{\text{fold}} = \frac{\sum N}{[D]}$$



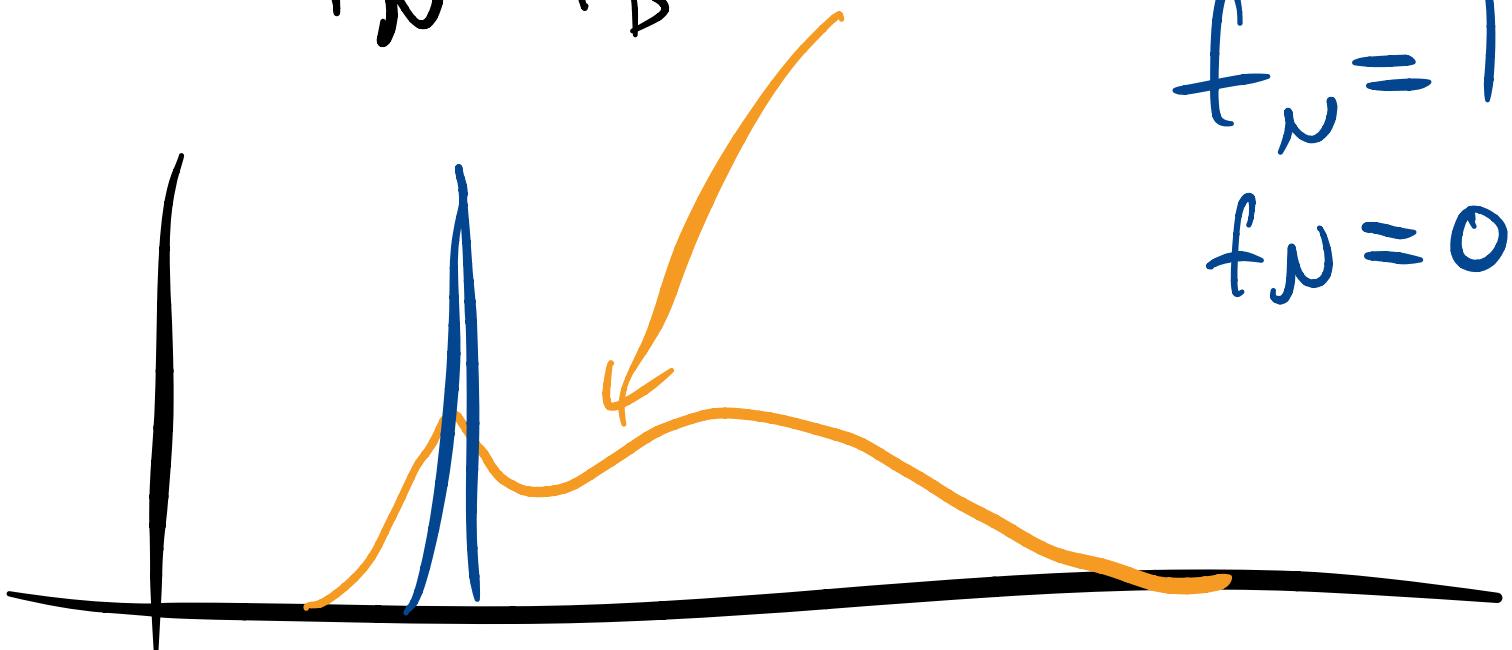
Do measurement

$$y^{\text{observed}}(d) = f_N y_N(x) + f_D y_D(x)$$

$$f_N = f_D = 0.5$$

$$f_N = 1.0$$

$$f_N = 0.995$$



Examples

- FRET - infer distances
- Circular Dichroism (CD)
absorption around 220nm
shows secondary structure
- NMR

When can we measure K

$$K_{\text{fold}} = \frac{[N]}{[D]} \Rightarrow [N] = [D]k$$

$$f_N = \frac{[N]}{[N] + [D]}$$

$$= \frac{[D]k}{[D]k + [D]} = \frac{k_{\text{fold}}}{k_{\text{fold}} + 1}$$

$$f_N + f_D = 1 \Rightarrow F_D = \frac{1}{k_{\text{fold}} + 1}$$

$$f_N = 0.1 - 0.9$$

$$f_N = 0.1 = \frac{k_{\text{fold}}}{k_{\text{fold}} + 1}$$

$$0.1k + 0.1 = k \Rightarrow 0.1 = 0.9k$$

$$k_{\min} = 1/9$$

$$f_N = 0.9 \Rightarrow 0.9k + 0.9 = k$$

$$\Rightarrow k_{\max} = 9$$

$$\frac{1}{q} < K_{fold} < q \quad \Delta G^\circ = -RT \ln K$$

$$-\ln q < \ln K_{fold} < \ln q$$

$$-2.2 < \ln K < 2.2 \quad \times -\frac{RT}{}$$

$$1.3 \geq \Delta G^\circ \geq -1.3 \frac{\text{kcal}}{\text{mol}} \quad \approx 0.6 \frac{\text{Kcal}}{\text{mol}}$$

↑
Stability range where
 K_{eq} can be measured

$$K_{eq} = e^{-\Delta G^\circ / RT}$$

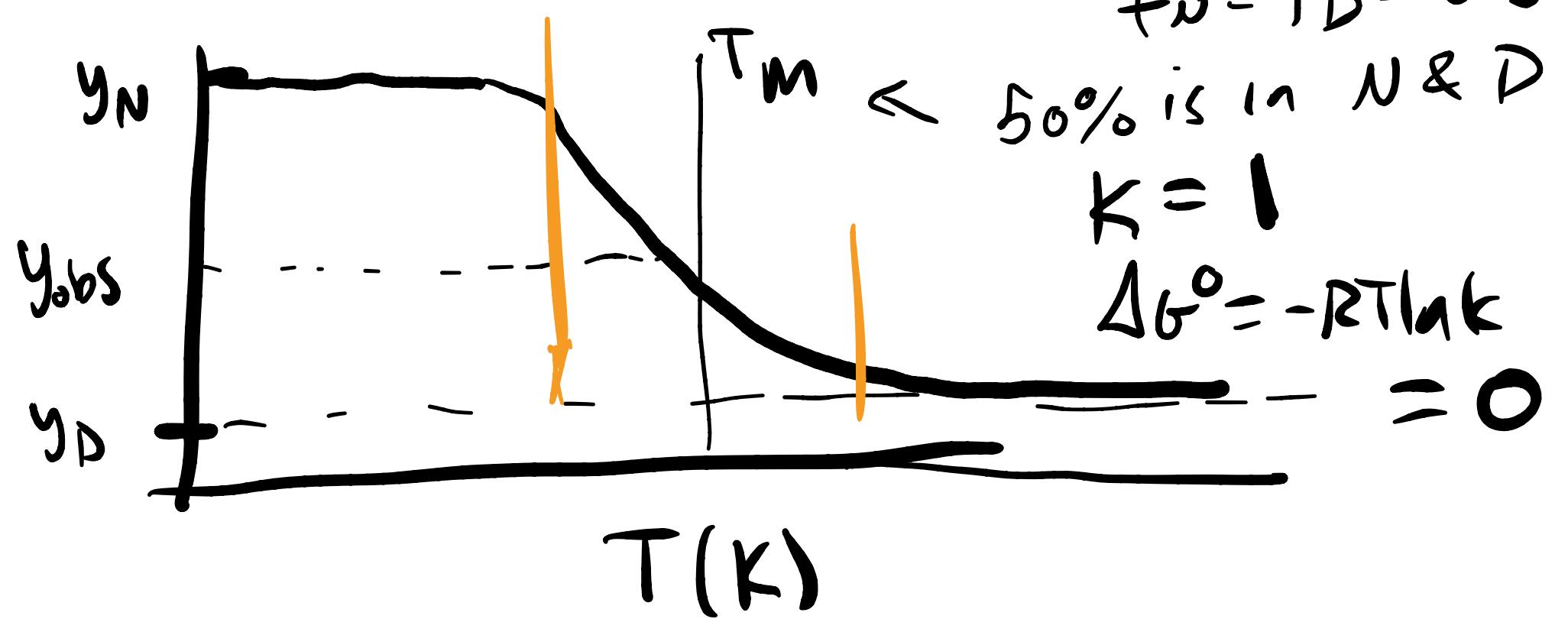
$$RT \approx 0.6 \frac{\text{kcal}}{\text{mol}}$$

$$e^{2.3} \approx 10$$

$$\Delta G = -2.3RT \approx -1.4 \text{ kcal/mol}$$

$K_{fold} > 10^3 \approx \text{~factor of } 10^6$
at room temperature

Do a denaturing experiment
increasing temperature
"melt" protein



fit curve to a model to
get ΔG° , ΔH° & ΔS°

$$y_{\text{obs}} = f_N y_N + (1-f_N) y_D$$

$$= y_D + f_N(y_N - y_D)$$

$$f_N = \frac{y_{\text{obs}} - y_D}{y_N - y_D}$$

$$y_{obs} = \frac{\kappa}{1+\kappa} y_N + \frac{1}{1+\kappa} y_D$$

$$= \frac{y_D + \kappa y_N}{1 + \kappa} <$$

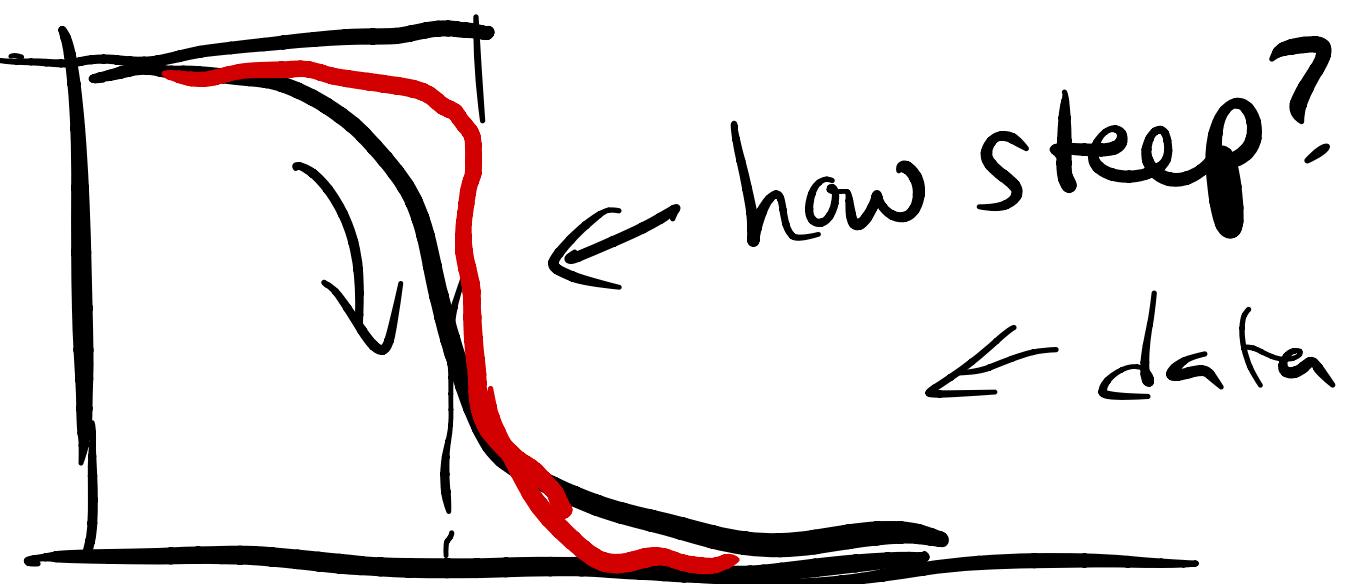
$$= y_D - (y_D - y_N) \frac{\kappa}{1 + \kappa}$$

$$y_{obs} = \frac{y_D + K y_N}{1 + K} = \frac{y_D + y_N e^{-\bar{\Delta G}^{\circ}/RT}}{1 + e^{-\bar{\Delta G}^{\circ}/RT}}$$

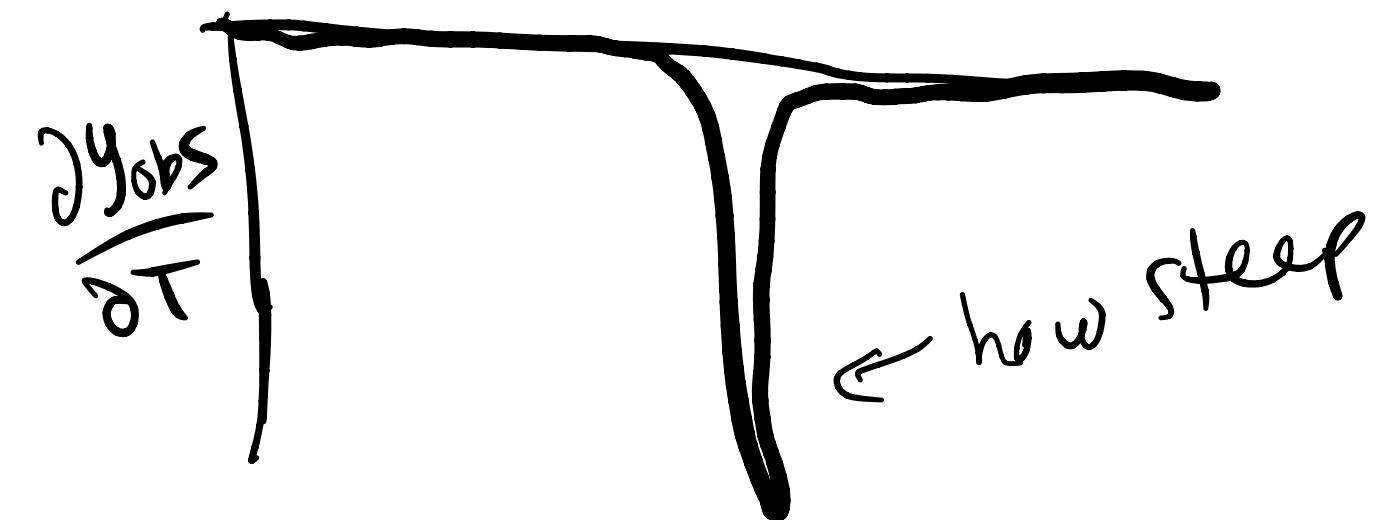
$$\bar{\Delta G}^{\circ} = \Delta H^{\circ} - T \bar{\Delta S}^{\circ}$$

$$\bar{\Delta G}^{\circ} = 0 \Rightarrow \Delta \bar{H}^{\circ} - T_m \bar{\Delta S}^{\circ} = 0$$

$$T_m = \bar{\Delta H}^{\circ} / \bar{\Delta S}^{\circ}$$



T_m



Simple model $\Delta H^\circ, \Delta S^\circ$ are constant

$$\frac{\partial y}{\partial T} = (y_N - y_D) \frac{K}{(1+K)^2} \frac{\Delta \bar{H}^\circ}{R T^2}$$

$\rightarrow 0$

$$\Delta \bar{H}^\circ < 0 \quad T_m = \Delta \bar{H}^\circ / \Delta \bar{S}^\circ$$

$$\Delta \bar{S}^\circ = \Delta \bar{H}^\circ / T_m$$

$$y_{\text{obs}} = y_D + (y_N - y_D) \frac{k}{1+k}$$

$$\xrightarrow{k} K = e^{-\Delta \bar{G}^{\circ}/RT} = e^{\frac{(-\Delta \bar{H}^{\circ}/RT + \Delta S^{\circ}/R)}{C}}$$

$$\begin{aligned}\Delta G^{\circ} &= \Delta H^{\circ} - T \Delta S^{\circ} = \frac{\Delta \bar{H}^{\circ} - T \Delta \bar{S}^{\circ}}{T_m} \\ &= \Delta H^{\circ} (T_m - T) / T_m\end{aligned}$$

$$K = e^{\Delta H^\circ(T - T_m) / (RTm)}$$

Can fit $y_{\text{obs}}(T)$ to

$$\Delta H^\circ, T_m,$$

A better model

Constant heat capacity

C_p for Denatured & Native

Native:

$$d\bar{H}_N = \bar{C}_p dT$$

$$\bar{H}_N = \underline{\bar{H}_N^{\text{ref}}} + \bar{C}_p (T - T_{\text{ref}})$$

$$\bar{H}_N = \bar{H}_N^{\text{ref}} + \bar{C}_P^N(T - T_{\text{ref}})$$

$$-\left(\bar{H}_D = \bar{H}_D^{\text{ref}} + \bar{C}_P^D(T - T_{\text{ref}}) \right)$$

$$\Delta \bar{H} \approx \underline{\Delta \bar{H}_{\text{ref}}} + \underline{\Delta \bar{C}_P(T - T_{\text{ref}})}$$

\bar{C}_{T_m}

Can do same for dS

$$d\bar{S} = \bar{C}_P / T \, dT$$

$$S_{N,D}^{N,D} = S_{N,D}^{\text{ref}} + \bar{C}_P^{N,D} \ln(T/T_{\text{ref}})$$

$$\Delta \bar{S} = \Delta \bar{S}_{\text{ref}} + \Delta \bar{C}_P \ln(T/T_{\text{ref}})$$

$$\Delta G^\circ = \Delta H^\circ - T \Delta S^\circ$$

$$K = e^{-\Delta G^\circ / RT} \rightarrow \gamma_N$$