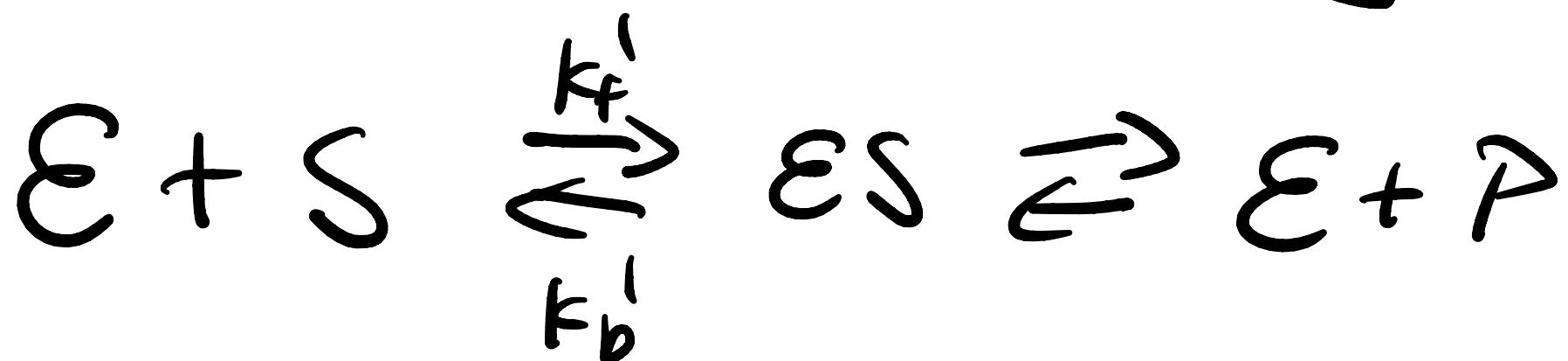
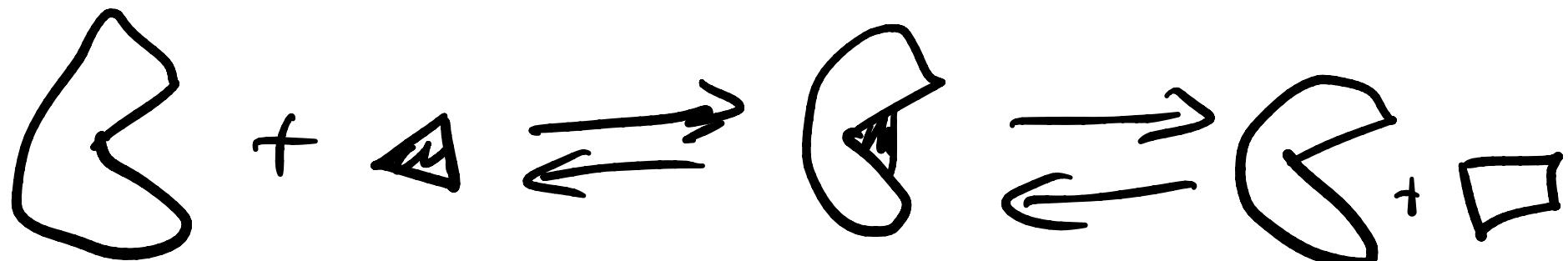


Michaelis-Menten kinetics

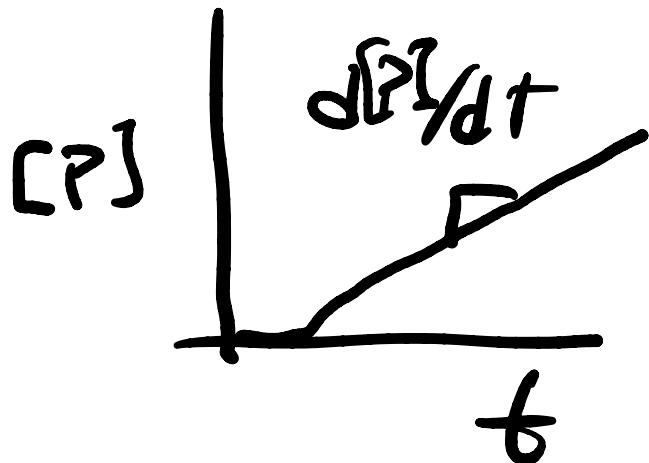


$\frac{d[ES]}{dt} = 0$, Substrate is in excess

$$[E] + [ES] = [E]_0$$

$$\text{rate of product formation} = \frac{(k_f^1 k_f^2 [S]_0 - k_b^1 k_b^2 [P]) [E]_0}{k_f^1 [S]_0 + k_b^1 + k_b^2 [P]_0 + k_f^2}$$

Measure initial rate



Initial conditions : $[P] = 0, [S] \approx [S]_0$

$$r_{\text{initial}} = \frac{k_f^1 k_f^2 [E]_0 [S]_0}{k_f^1 [S]_0 + k_b^1 + k_f^2} = \frac{k_f^2 [E]_0 [S]_0}{[S]_0 + K_M}$$

$$K_M = \frac{k_f^2 + k_b^{-1}}{k_f^{-1}}$$

1 k_f^2 is called
Kcat

$$r_{\text{initial}} = \frac{k_{\text{cat}} [S]_0 [E]_0}{[S]_0 + K_M}$$

If $[S]_0$ really big compared to K_M

$$r_{\text{max}} = k_{\text{cat}} [E]_0 \quad (\text{fit } k_{\text{cat}})$$

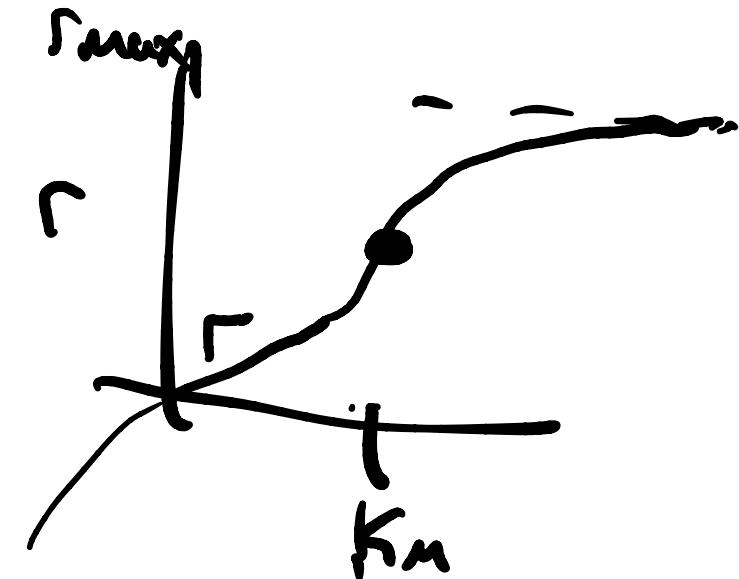
is $[S]_0$ small, expect rate to be linear
in $[S]_0$

$$r = \frac{r_{\max} [S]_0}{[S]_0 + K_M}$$

what if $[S]_0$

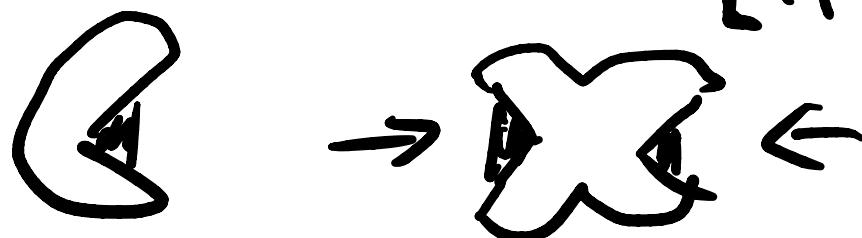
$$= K_M,$$

$$r = r_{\max}/2$$



initial
slope gives $\frac{r_{\max}}{K_M}$

Turnover $\text{ff} = \frac{r_{\max}}{\text{[# active sites]}}$



$$\uparrow [E]_0 \cdot \#/\text{enzyme}$$

Catalytic efficiency

$$\epsilon = k_{\text{cat}} / k_m \quad , \text{ units of } \frac{1}{\mu} \cdot \frac{1}{s}$$

range $1 - 10^{10} / (\mu s)$

$10^8 - 10^{10}$ diffusion limited

$$\frac{k_{\text{cat}}}{k_m} = \frac{k_{\text{cat}}}{\left(\frac{k_{\text{cat}} + k_b'}{k_{f'}} \right)} = \frac{k_{f'}}{1 + \frac{k_b'}{k_{\text{cat}}}}$$

biggest when $k_{\text{cat}}, k_{f'}$ big + k_b' small

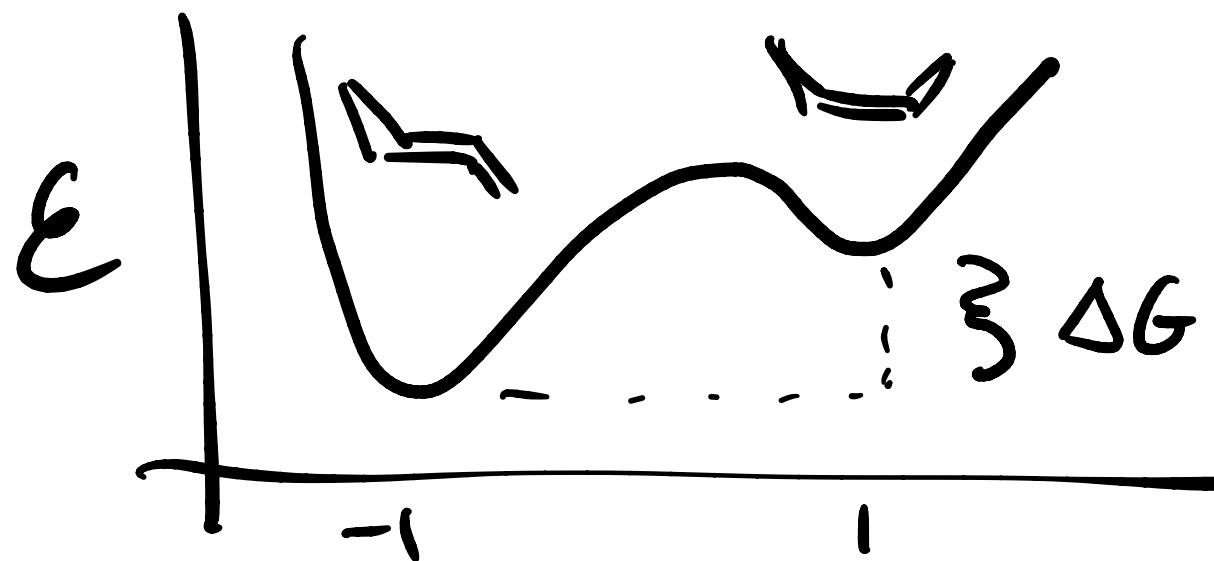
Introduction to Statistical Thermodynamics (ch 9)

microscopic \rightarrow Macroscopic observables

Key idea: Properties are universal
can't depend on arrangement
of molecules

All thermodynamic properties are avg.s

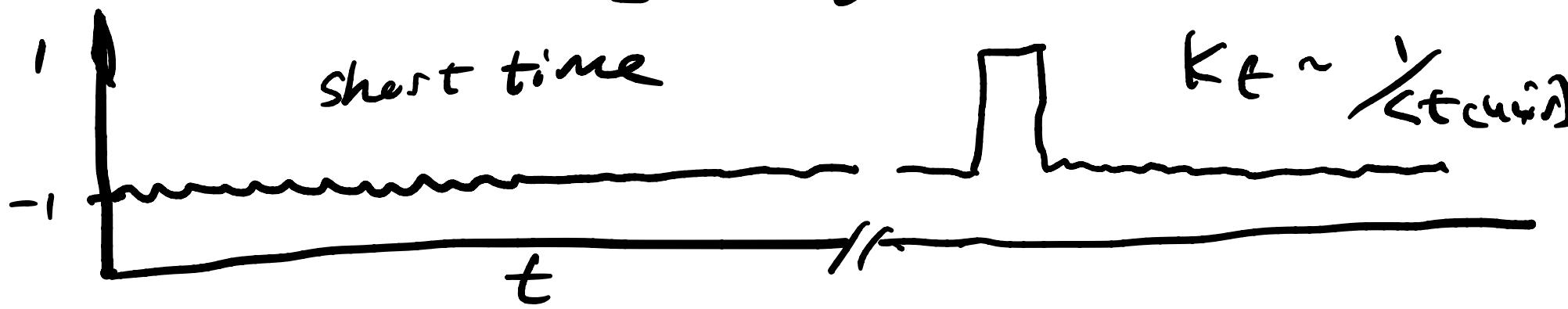




$$\frac{k_f}{k_b} = k_{eq} = e^{-\Delta G^\circ / RT}$$

$$= \frac{\sum [\text{Boat}]}{[\text{Chair}]}$$

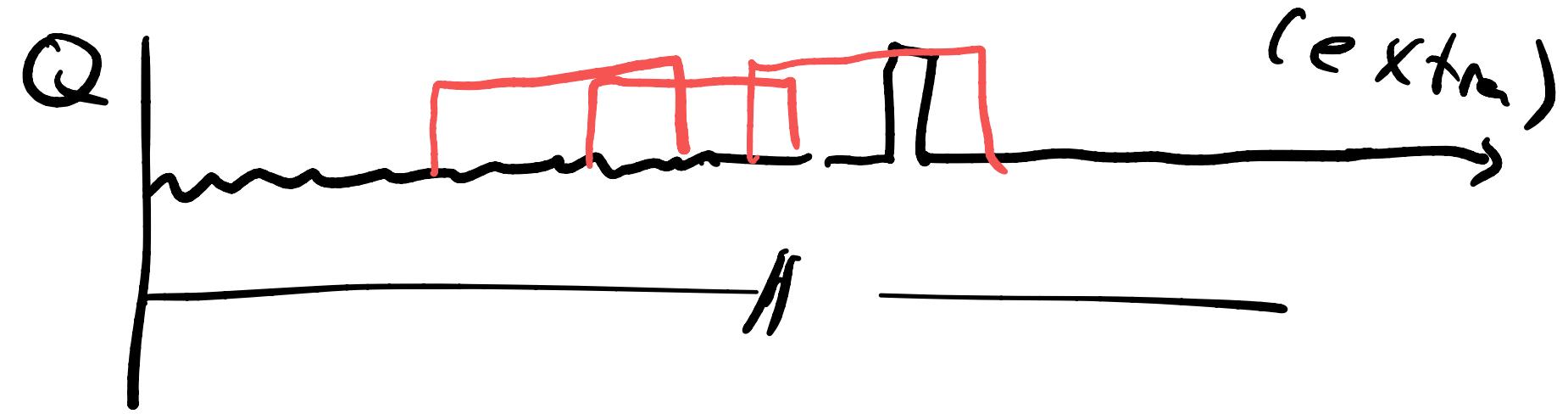
$$k_b \sim \frac{1}{\langle t_{\text{boat}} \rangle}$$



- Single molecule
observe many transitions \rightarrow avg
 eq
- Many molecules
observe few transitions \rightarrow avg
 eq

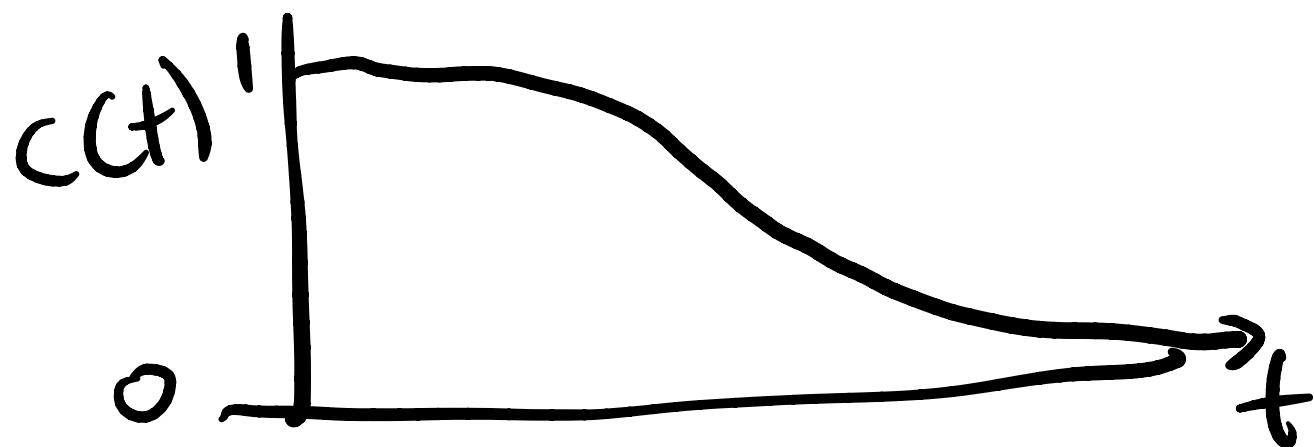
time avg = ensemble avg
(assumption / principle)

\rightarrow true : ergodic / ergodicity



Time correlation function:

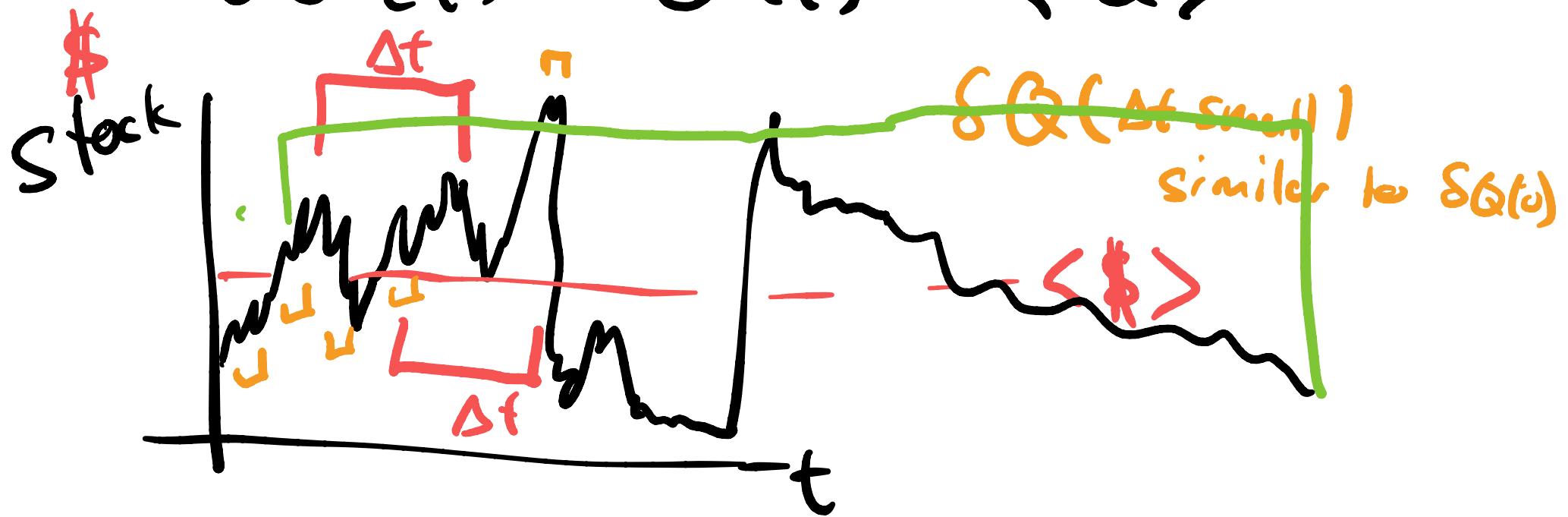
If you know something now,
do you know it later



$C(t)$ is relative to avg

$$C(\Delta t) = \frac{\langle \delta Q(t + \Delta t) \delta Q(t) \rangle}{\langle \delta Q(t) \rangle^2}$$

$$\delta Q(t) = Q(t) - \langle Q \rangle$$



$$C(\Delta t) = \frac{\langle \delta Q(t + \Delta t) \delta Q(t) \rangle}{\langle \delta Q(t) \rangle^2} \leftarrow \text{variance}$$

$$\begin{aligned}\langle \delta Q \rangle &= 0 = \langle Q - \langle Q \rangle \rangle \\ &= \langle Q \rangle - \langle \langle Q \rangle \rangle \\ &= \langle Q \rangle - \langle Q \rangle = 0\end{aligned}$$

e $\Delta t = 0 \quad C(t) = 1$

e Δt really big, no correlation

$$\langle \delta Q(t + \Delta t) \delta Q(t) \rangle = \langle \delta Q \rangle \langle \delta Q \rangle = 0$$



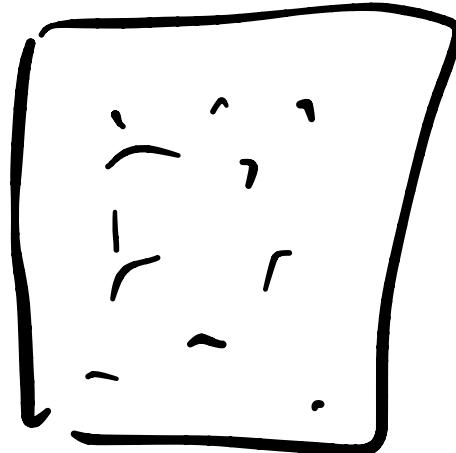
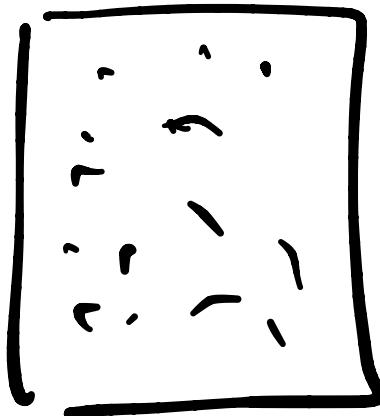
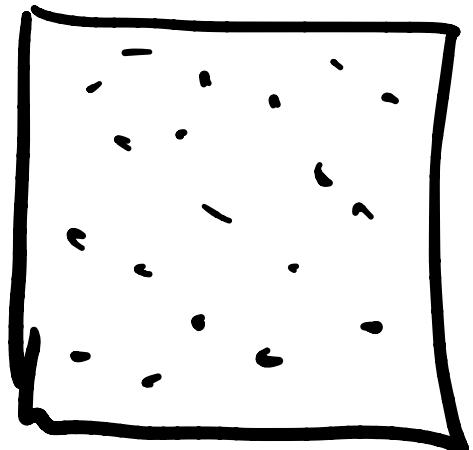
what is the avg time
for rxn

$$\tau_{rxn} \equiv \int_0^\infty C(st) d\Delta t$$

if exponential, $C(\Delta t) = e^{-t/\tau_{rxn}}$

$$\text{rate} = \gamma \tau_{rxn}$$

Statistical Ensembles



M copies

N, V, E

all of these configs
are equally likely

Calculate averages over

$$\langle O(x_i) \rangle = \frac{1}{M} \sum_{i=1}^M O(x_i)$$

$$P(x_i) = \frac{1}{M}$$

$$\langle O \rangle = \sum_{i=1}^m O_i P_i \quad \text{count states}$$

$$= \int dx_1 dx_2 dx_3 \dots dx_N O(\mathbf{x}) P(\mathbf{x})$$

Next time, if const T

$$P(x) = e^{-U(x)/k_B T}$$

and similar for other "ensembles"