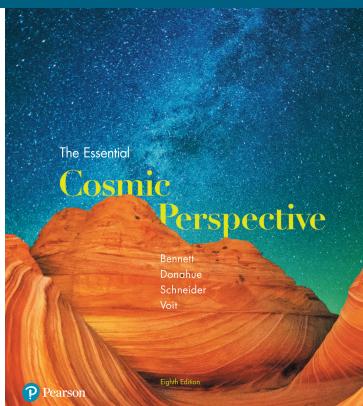


Lecture Outline

Chapter 5: Light The Cosmic Messenger



1

5.1 Basic Properties of Light and Matter

Our goals for learning:

- What is light?
- What is matter?
- How do light and matter interact?

2

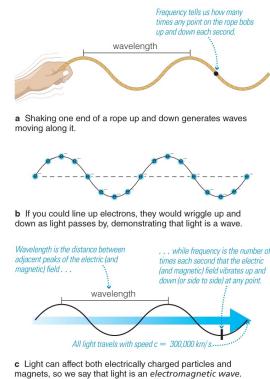
What is light?



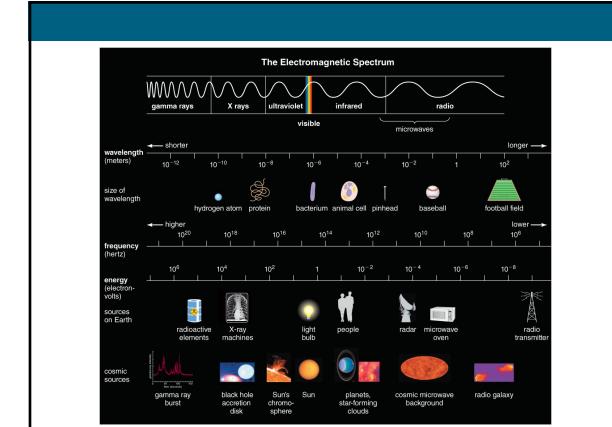
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3

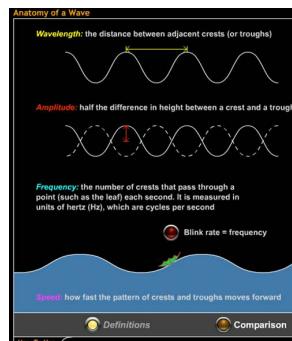
Light is an electromagnetic wave.



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4



PLAY Anatomy of a Wave

5

6

Wavelength and Frequency

- Longer wavelength means lower frequency.

wavelength = 1 cm, frequency = 30 GHz

wavelength = $\frac{1}{2}$ cm, frequency = 2×30 GHz = 60 GHz

wavelength = $\frac{1}{4}$ cm, frequency = 4×30 GHz = 120 GHz

wavelength x frequency = speed of light = constant

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The Electromagnetic Spectrum

Electromagnetic Spectrum

Visible

Gamma rays Ultraviolet Visible Infrared Radio

Frequency 3.00×10^{11} Hz

Wavelength 1.00×10^{-3} m

Speed 3×10^8 m/s

How To Use Credits

PLAY Electromagnetic Spectrum

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8

Particles of Light

- Particles of light are called **photons**.
- Each photon has a wavelength and a frequency.
- The energy of a photon depends on its frequency.

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Wavelength, Frequency, and Energy

$$\lambda \times f = c$$

λ = wavelength, f = frequency
 c = 3.00×10^8 m/s = speed of light

$$E = h \times f = \text{photon energy}$$

$$h = 6.626 \times 10^{-34} \text{ joule} \times \text{s}$$

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Thought Question

The higher the photon energy,

- the longer its wavelength.
- the shorter its wavelength.
- Energy is independent of wavelength.

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Thought Question

The higher the photon energy,

- the longer its wavelength.
- B. the shorter its wavelength.**
- Energy is independent of wavelength.

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What is matter?

The nucleus is nearly 100,000 times smaller than the atom but contains nearly all of its mass.

Ten million atoms could fit end to end across this dot.

Atom: Electrons are "smeared out" in a cloud around the nucleus.

10^{-10} meter

Nucleus: Contains positively charged protons (red) and neutral neutrons (gray).

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Atomic Terminology

- Atomic Number = # of protons in nucleus
- Atomic Mass Number = # of protons + # of neutrons

atomic number = number of protons
 atomic mass number = number of protons + neutrons
 (A neutral atom has the same number of electrons as protons.)

Hydrogen (${}^1\text{H}$)	Helium (${}^4\text{He}$)	Carbon (${}^{12}\text{C}$)
		
atomic number = 1 atomic mass number = 1 (1 electron)	atomic number = 2 atomic mass number = 4 (2 electrons)	atomic number = 6 atomic mass number = 12 (6 electrons)

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Atomic Terminology

atomic number = number of protons
 atomic mass number = number of protons + neutrons
 (A neutral atom has the same number of electrons as protons.)

- Isotope: same # of protons but different # of neutrons (${}^4\text{He}$, ${}^3\text{He}$)
- Molecules: consist of two or more atoms (H_2O , CO_2)

Hydrogen (${}^1\text{H}$)	Helium (${}^4\text{He}$)	Carbon (${}^{12}\text{C}$)
		
atomic number = 1 atomic mass number = 1 (1 electron)	atomic number = 2 atomic mass number = 4 (2 electrons)	atomic number = 6 atomic mass number = 12 (6 electrons)

Different isotopes of a given element contain the same number of protons, but different numbers of neutrons.

Isotopes of Carbon		
		
carbon-12 (6 protons + 6 neutrons)	carbon-13 (6 protons + 7 neutrons)	carbon-14 (6 protons + 8 neutrons)

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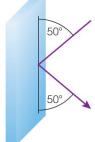
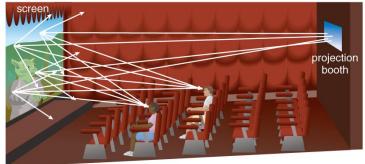
How do light and matter interact?

- Emission
- Absorption
- Transmission
- Transparent objects transmit light.
- Opaque objects block (absorb) light.
- Reflection or scattering

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Reflection and Scattering

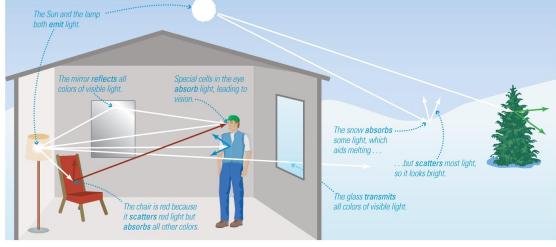
Mirror reflects light in a particular direction.

Movie screen scatters light in all directions.

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Interactions of Light with Matter



The Sun and the lamp both emit light.

The mirror reflects all colors of visible light.

Special cells in the eye absorb light, leading to vision.

The snow absorbs some light, which aids melting... but scatters most light, so it looks bright.

The glass transmits all colors of visible light.

The chair is red because it scatters red light but absorbs all other colors.

Interactions between light and matter determine the appearance of everything around us.

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Thought Question

Why is a rose red?

- A. The rose absorbs red light.
- B. The rose transmits red light.
- C. The rose emits red light.
- D. The rose reflects red light.

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Thought Question

Why is a rose red?

- A. The rose absorbs red light.
- B. The rose transmits red light.
- C. The rose emits red light.
- D. The rose reflects red light.**

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5.2 Learning from Light

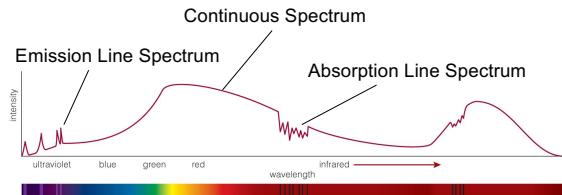
Our goals for learning:

- What are the three basic types of spectra?
- How does light tell us what things are made of?
- How does light tell us the temperatures of planets and stars?
- How does light tell us the speed of a distant object?

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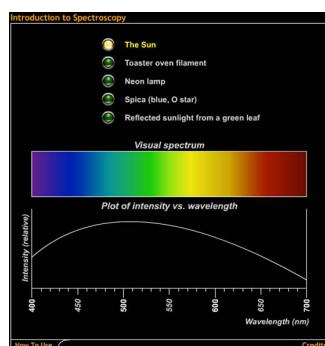
What are the three basic types of spectra?



Spectra of astrophysical objects are usually combinations of these three basic types.

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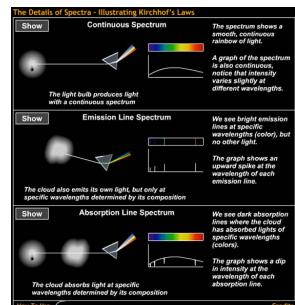


PLAY Introduction to Spectroscopy

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23

Three Types of Spectra

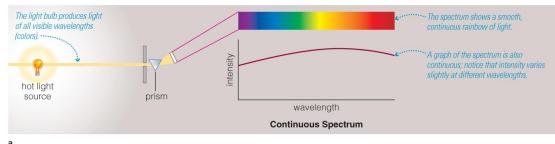


PLAY Illustrating Kirchhoff's Laws

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24

Continuous Spectrum

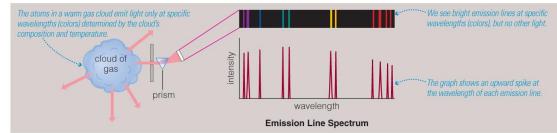


- The spectrum of a common (incandescent) light bulb spans all visible wavelengths, without interruption.

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Emission Line Spectrum

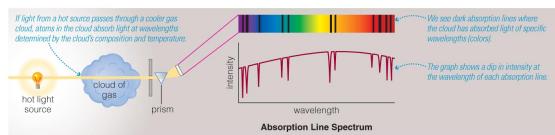


- A thin or low-density cloud of gas emits light only at specific wavelengths that depend on its composition and temperature, producing a spectrum with bright emission lines.

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Absorption Line Spectrum

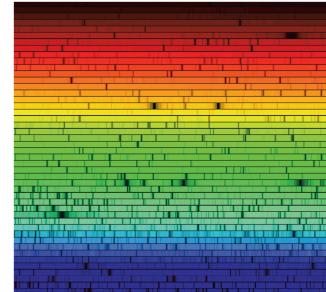


- A cloud of gas between us and a light bulb can absorb light of specific wavelengths, leaving dark absorption lines in the spectrum.

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How does light tell us what things are made of?

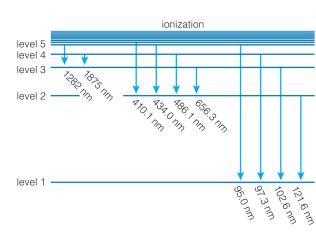


Spectrum of the Sun

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Chemical Fingerprints

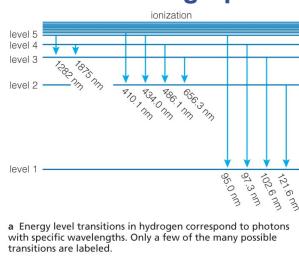


Energy levels of hydrogen

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- Each type of atom has a unique set of energy levels.
- Each transition corresponds to a unique photon energy, frequency, and wavelength.

Chemical Fingerprints



a. Energy level transitions in hydrogen correspond to photons with specific wavelengths. Only a few of the many possible transitions are labeled.

b. This spectrum shows emission lines produced by downward transitions between higher levels and level 2 in hydrogen.

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- Downward transitions produce a unique pattern of emission lines.

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30

Production of Emission Lines

PLAY Production of Emission Lines
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Chemical Fingerprints

a Energy level transitions in hydrogen correspond to photons with specific wavelengths. Only a few of the many possible transitions are labeled.
c This spectrum shows absorption lines produced by upward transitions between level 2 and higher levels in hydrogen.

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- Because those atoms can absorb photons with those same energies, upward transitions produce a pattern of absorption lines at the same wavelengths.

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Production of Absorption Lines

PLAY Production of Absorption Lines
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Chemical Fingerprints

- Each type of atom has a unique spectral fingerprint.

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Composition of a Mystery Gas

PLAY Composition of a Mystery Gas
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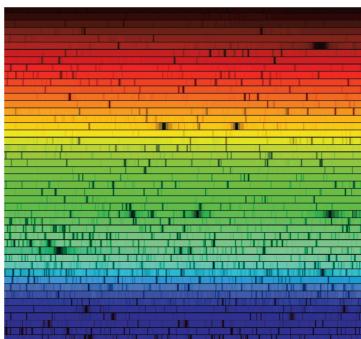
35

Chemical Fingerprints

- Observing the fingerprints in a spectrum tells us which kinds of atoms are present.

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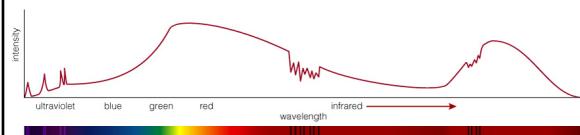
Example: Solar Spectrum

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Thought Question

Which letter(s) label(s) absorption lines?



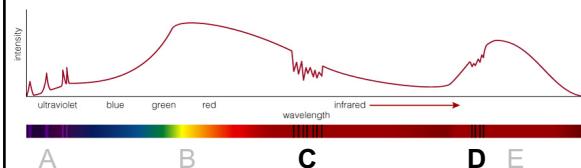
A B C D E

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Thought Question

Which letter(s) label(s) absorption lines?

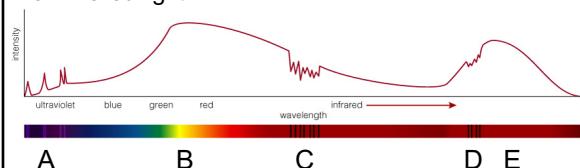


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Thought Question

Which letter(s) label(s) the peak (greatest intensity) of infrared light?



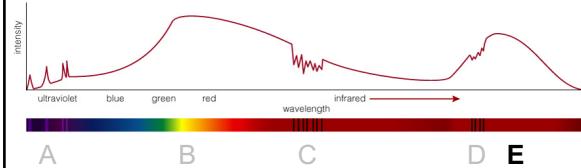
A B C D E

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40

Thought Question

Which letter(s) label(s) the peak (greatest intensity) of infrared light?

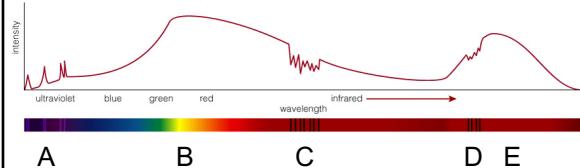


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Thought Question

Which letter(s) label(s) emission lines?



A B C D E

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Thought Question

Which letter(s) label(s) emission lines?

A B C D E

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How does light tell us the temperatures of planets and stars?

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Thermal Radiation

- Nearly all large or dense objects emit thermal radiation, including stars, planets, and you.
- An object's thermal radiation spectrum depends on only one property: its **temperature**.

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Properties of Thermal Radiation

- Hotter objects emit more light at all frequencies per unit area.
- Hotter objects emit photons with a higher average energy.

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Wien's Law

PLAY Wien's Law

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Thought Question

Which is hottest?

- A blue star
- A red star
- A planet that emits only infrared light

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Thought Question

Which is hottest?

- A. A blue star
- B. A red star
- C. A planet that emits only infrared light

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Thought Question

Why don't we glow in the dark?

- A. People do not emit any kind of light.
- B. People only emit light that is invisible to our eyes.
- C. People are too small to emit enough light for us to see.
- D. People do not contain enough radioactive material.

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Thought Question

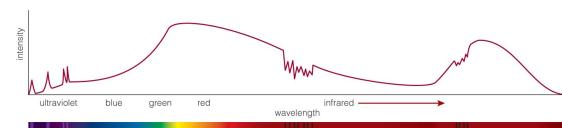
Why don't we glow in the dark?

- A. People do not emit any kind of light.
- B. People only emit light that is invisible to our eyes.**
- C. People are too small to emit enough light for us to see.
- D. People do not contain enough radioactive material.

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Interpreting an Actual Spectrum

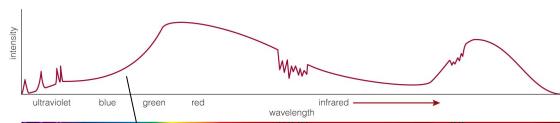


- By carefully studying the features in a spectrum, we can learn a great deal about the object that created it.

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What is this object?

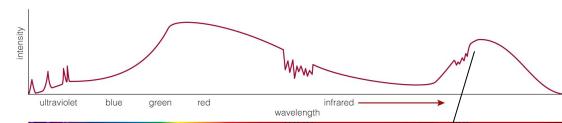


Reflected sunlight:
Continuous spectrum of
visible light is like the
Sun's except that some
of the blue light has
been absorbed—the
object must look red.

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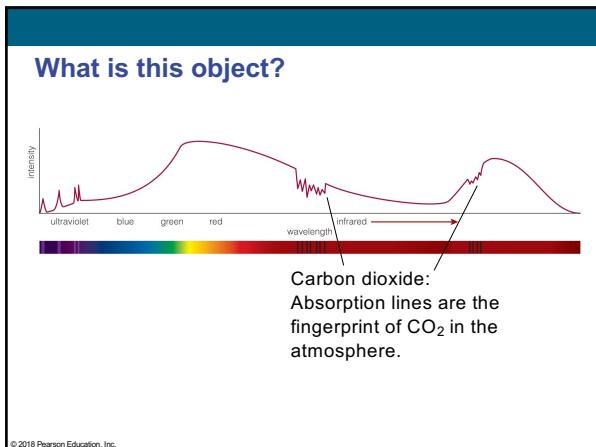
What is this object?



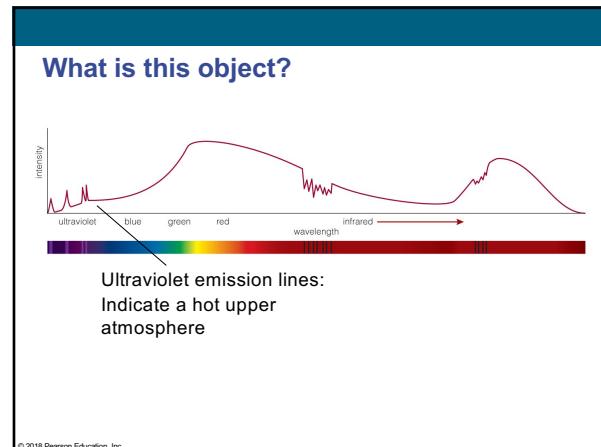
Thermal radiation:
Infrared spectrum peaks
at a wavelength
corresponding to a
temperature of 225 K.

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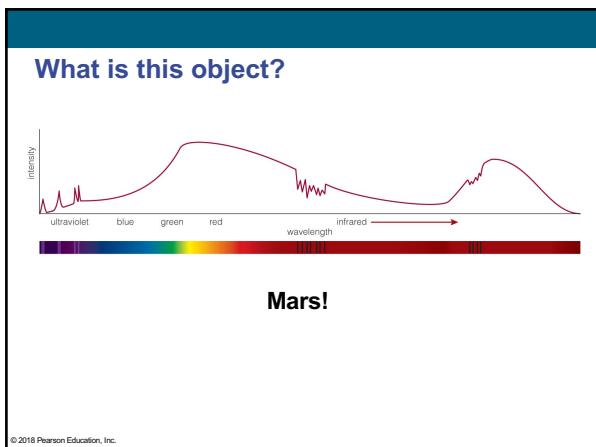
54



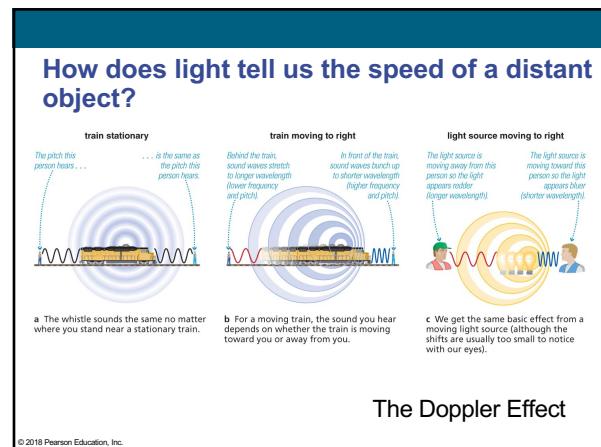
55



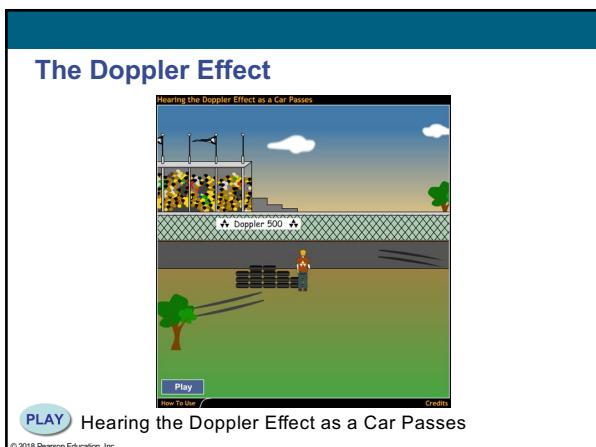
56



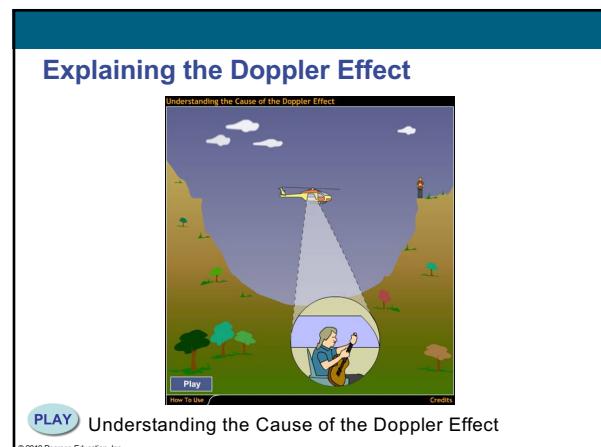
57



58



59



60

Same for light

The Doppler Effect for Visible Light

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Measuring the Shift

Laboratory spectrum
Lines at rest wavelengths.

	Stationary
--	------------

Object 1 Lines redshifted:
Object moving away from us.

	Moving Away
--	-------------

Object 2 Greater redshift:
Object moving away faster than object 1.

	Away Faster
--	-------------

Object 3 Lines blueshifted:
Object moving toward us.

	Moving Toward
--	---------------

Object 4 Greater blueshift:
Object moving toward us faster than object 3.

	Toward Faster
--	---------------

- We generally measure the Doppler effect from shifts in the wavelengths of spectral lines.

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The amount of blue or red shift tells us an object's speed toward or away from us.

The Doppler Shift of an Emission Line Spectrum

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Doppler shift tells us ONLY about the part of an object's motion toward or away from us.

Star 1 is moving directly away from us, so the Doppler shift tells us its full speed.

Star 2 is moving across our line of sight, but not toward or away from us. The Doppler shift measures no speed at all.

Star 3 is moving diagonally away from us. The Doppler shift tells us the part of the star's speed away from us... but not the part of the speed across our line of sight.

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Thought Question

I measure a line in the lab at 500.7 nm. The same line in a star has wavelength 502.8 nm. What can I say about this star?

- It is moving away from me.
- It is moving toward me.
- It has unusually long spectral lines.

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Thought Question

I measure a line in the lab at 500.7 nm. The same line in a star has wavelength 502.8 nm. What can I say about this star?

- It is moving away from me.
- It is moving toward me.
- It has unusually long spectral lines.

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Measuring Redshift

The Doppler Shift of an Emission-Line Spectrum

Speed of Gas Cloud (relative to us): 3000 (Moving toward us 3000 km/s) 3000 (Moving away from us 3000 km/s) Speed = 0 km/s

PLAY The Doppler Shift of an Emission Line Spectrum

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Measuring Redshift

Doppler Shift of Absorption Lines

To Earth

PLAY Doppler Shift of Absorption Lines

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Measuring Velocity

Determining the Velocity of a Gas Cloud

Rest wavelength: 400.0 nm

Observed wavelength: 400.0 nm

Wavelength shift: ----- nm
Relative speed: ----- km/s

PLAY Determining the Velocity of a Gas Cloud

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Measuring Velocity

Determining the Velocity of a Cold Cloud of Hydrogen Gas

Rest wavelength: 400.0 nm

Observed wavelength: 400.0 nm

Wavelength shift: ----- nm
Relative speed: ----- km/s

PLAY Determining the Velocity of a Cold Cloud of Hydrogen Gas

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5.3 Collecting Light with Telescopes

Our goals for learning:

- How do telescopes help us learn about the universe?
- Why do we put telescopes into space?

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How do telescopes help us learn about the universe?

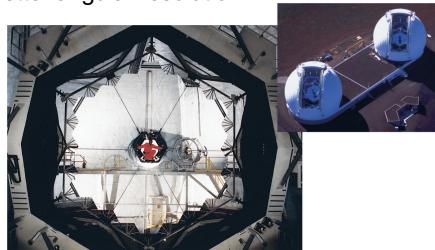
- Telescopes collect more light than our eyes
⇒ **light-collecting area**
- Telescopes can see more detail than our eyes ⇒ **angular resolution**
- Telescopes/instruments can detect light that is invisible to our eyes (e.g., infrared, ultraviolet)

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Bigger is better

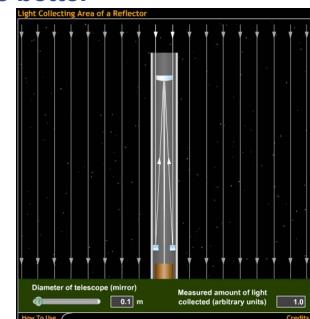
1. Larger light-collecting area
2. Better angular resolution



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Bigger is better



PLAY Light Collecting Area of a Reflector

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Angular Resolution

- The *minimum* angular separation that the telescope can distinguish



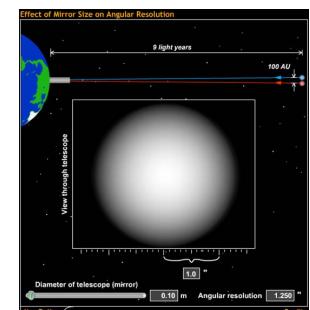
PLAY

Angular Resolution Explained Using Approaching Car Lights

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Angular Resolution: Smaller Is Better



PLAY

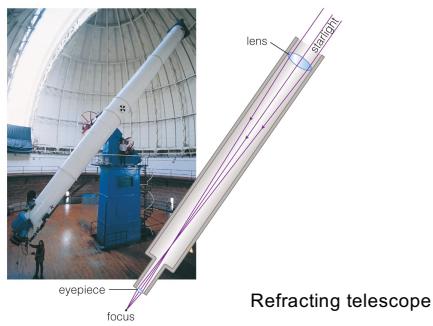
Effect of Mirror Size on Angular Resolution

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Basic Telescope Design

- Refracting: lenses



Refracting telescope

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Basic Telescope Design

- Reflecting: mirrors
- Most research telescopes today are reflecting

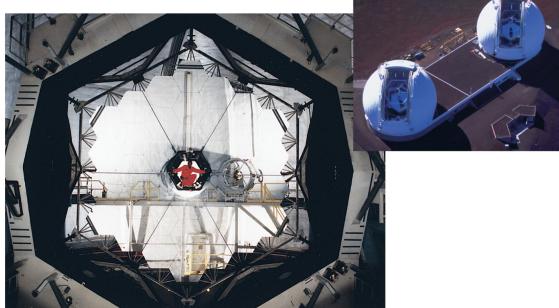


Reflecting telescope

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Keck I and Keck II Mauna Kea, Hawaii



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Mauna Kea, Hawaii

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Different designs for different wavelengths of light



500 meter radio telescope (Guizhou Province, China)

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Interferometry

- This technique allows two or more small telescopes to work together to obtain the *angular resolution* of a larger telescope.



Karl G. Jansky Very Large Array (JVLA), New Mexico

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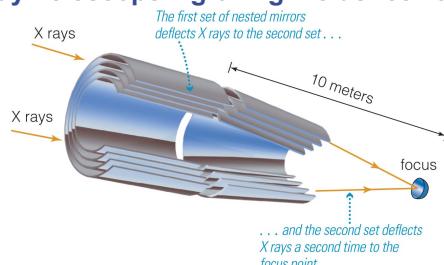


The Atacama Large Millimeter/submillimeter Array (ALMA) in Chile

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X-ray Telescope: "grazing incidence" optics



- a* This diagram shows the arrangement of Chandra's nested, cylindrical X-ray mirrors. Each mirror is 0.8 meter long and between 0.6 and 1.2 meters in diameter.

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Want to buy your own telescope?

- Buy binoculars first (e.g., 7 x 35) — you get much more for the same amount of money.
- Ignore magnification (sales pitch!).
- Notice: aperture size, optical quality, portability.
- Consumer research: *Astronomy*, *Sky & Telescope*, *Mercury* magazines; astronomy clubs.

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Looking Beyond Light



Shown here is one of the gravitational wave detectors that are part of Advanced Laser Interferometer Gravitational-Wave Observatory (LIGO).

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Why do we put telescopes into space?

It is NOT because they are closer to the stars!

Recall our 1-to-10 billion scale:

- Sun size of grapefruit
- Earth size of a tip of a ballpoint pen, 15 m from Sun
- Nearest stars 4000 km away
- Hubble orbit microscopically above tip of a ballpoint pen-size Earth



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Observing problems due to Earth's atmosphere

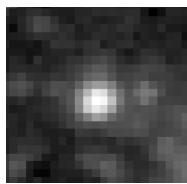
1. Light pollution



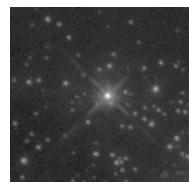
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2. Turbulence causes twinkling ⇒ blurs images



Star viewed with ground-based telescope

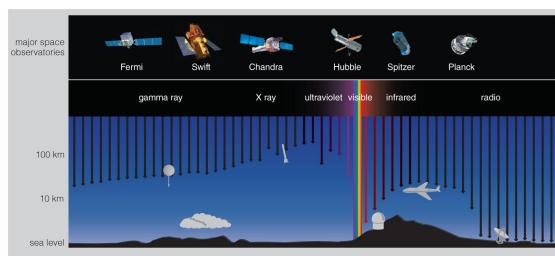


View from Hubble Space Telescope

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3. Atmosphere absorbs most of EM spectrum, including all UV and X ray and most infrared.



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Telescopes in space solve all three problems.

- Location/technology can help overcome light pollution and turbulence.
- Nothing short of going to space can solve the problem of atmospheric absorption of light.



a Artist's illustration of the Chandra X-Ray Observatory, which orbits Earth.
Chandra X-Ray Observatory

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James Webb Space Telescope



The James Webb Space Telescope, a full-scale model of which is shown here, is set to launch in March 2021.

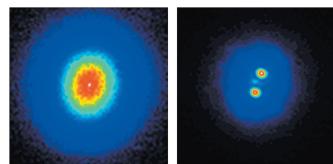
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Improvements for ground-based telescopes

Adaptive optics

- Rapid changes in mirror shape compensate for atmospheric turbulence.



a Atmospheric distortion makes this ground-based image of a double star look like a single star.

Without adaptive optics

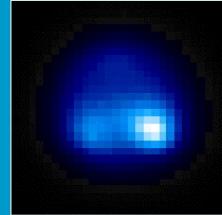
With adaptive optics

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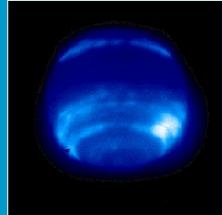
93

Adaptive optics: Neptune

without



with



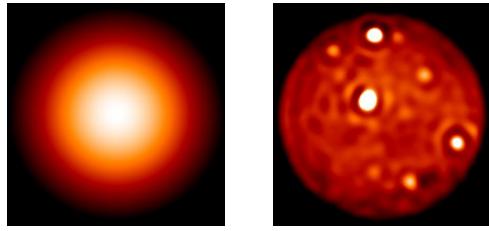
Center for Adaptive Optics, Univ. of California

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Adaptive Optics

- Jupiter's moon Io observed with the Keck telescope



without adaptive optics

with adaptive optics

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The Moon would be a great spot for an observatory (but at what price?).

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