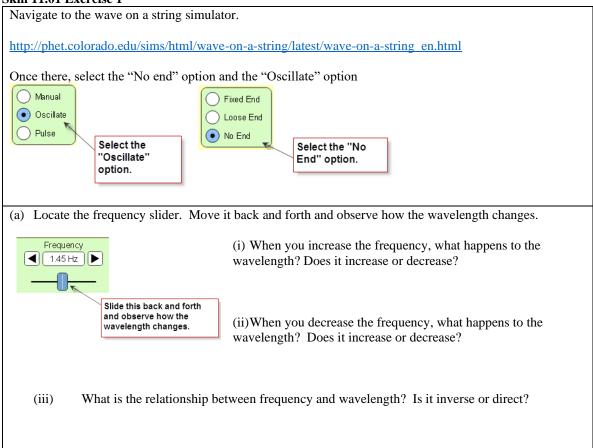
Name	Period
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### Skill 11.01 Exercise 1



# Skill 11.01 Exercise 2

Refer to the colors below. Sort the colors from low to high with respect to frequency.

Color Range	Order of frequency (1 = lowest)
Reds	
Oranges	
Yellows	
Greens	
Blues	
Violets	

Set 11: Relationship between wavelength, frequency, energy, and light

Name	Period
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### Skill 11.02 Exercise 1

1	For	each	of the	colors	below.
п	гог	еасп	or me	COIOIS	Delow.

- (a) Indicate the average wavelength in nanometers (nm)
- (b) Convert each wavelength to nanometers  $(1 \times 10^{-9} \text{ m} = 1 \text{ nm})$
- (c) Calculate the frequency

Color Range	Average λ (nm)	λ (m)	v (/s)
Reds	710 nm	$710 \text{nm} \times \frac{1 \text{m}}{1 \times 10^9 \text{ nm}} = 7.10 \times 10^{-7} \text{ m}$	$\frac{3.00 \times 10^8 \text{m/s}}{7.10 \times 10^{-7} \text{m}} = 4.23 \times 10^{14} /\text{s}$
Oranges			
Yellows			
Greens			
Blues			
Violets			

- (a) Are the wavelengths of ultra-violet light longer or shorter than that of visible light?
- (b) Are the frequencies of ultra-violet light longer or shorter than that of visible light?
- (c) Are the wavelengths of infra-red light longer or shorter than that of visible light?
- (d) Are the frequencies of infra-red light longer or shorter than that of visible light?

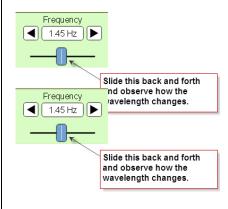
Name Period Period	
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### Skill 11.03 Exercise 1

Now return to the simulator (<a href="http://phet.colorado.edu/sims/html/wave-on-a-string/latest/wave-on-a-string\_en.html">http://phet.colorado.edu/sims/html/wave-on-a-string/latest/wave-on-a-string/latest/wave-on-a-string\_en.html</a>) Select the "Manual" option.



Move the wrench up and down as fast as you can and observe the wavelength. Now move the wrench up and down slowly and observe the wavelength.



- (i) When you moved the wrench up and down quickly (high energy), what happened to the wavelength? Did it increase or decrease?
- (ii) When you moved the wrench up and down slowly (low energy), what happened to the wavelength? Did it increase or decrease?
- (i) What is the relationship between energy and wavelength? Is it inverse or direct?

## Skill 11.03 Exercise 2

Refer to the colors below. Sort the colors from low to high with respect to energy.

Reds Oranges Yellows	st)
Yellows	
C	
Greens	
Blues	
Violets	

Chemistry
Ticket Out the Door
Set 11: Relationship between wavelength, frequency, energy, and light

lor Range	frequency v	Energy (J)	
eds	4.23×10 <sup>14</sup> /s	$(6.63 \times 10^{-34} \text{ J} \bullet \text{s})(4.23 \times 10^{14} / \text{s})$	
		$= 2.80 \times 10^{-19} \text{J}$	
Oranges			
Yellows			
~			
Greens			
Blues			
biues			
Violets			
VIOICIS			
) How do	41	olet compare to visible light?	