Period\_ Name Skill 20.01 Problem 1 Draw Lewis structures for the following molecules: (a) O<sub>3</sub> (b) SO<sub>2</sub> (c) CO<sub>3</sub><sup>2-</sup> (d) ClO<sub>4</sub>-(e) Which of the above molecules undergo "resonance"? How do you know? Skill 20.02 Problem 1 Beryllium can only form two bonds. Explain why using principles of electron structure.

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Name	Period
Draw the Lewis structure for BeF <sub>2</sub> .	
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Skill 20.02 Problem 2	
Sulfur can form up to 6 bonds. Explain why using principles of electron structure.	
Draw the Lewis structure for SF <sub>6</sub> .	

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Name \_\_\_\_\_\_Period \_\_\_\_\_

Skill 20.02 Problem 3				
Draw the following Lewis structures:				
PCl <sub>5</sub>	XeF <sub>4</sub>			

ICl <sub>4</sub> -	I <sub>3</sub> -

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