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| **Skill 11.01 Exercise 1** |
| Refer to the figure shown to the right. Identify the range of wavelengths associated with each color. The “Reds” have already been filled in. |
| |  |  | | --- | --- | | **Color Range** | **Wavelengths** | | Reds | 680–740 | | Oranges |  | | Yellows |  | | Greens |  | | Blues |  | | Violets |  | |

**Skill 11.01 Exercise 2**

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| Navigate to the wave on a string simulator.  <http://phet.colorado.edu/sims/html/wave-on-a-string/latest/wave-on-a-string_en.html>  Once there, select the “No end” option and the “Oscillate” option |
| 1. Locate the frequency slider. Move it back and forth and observe how the wavelength changes. 2. When you increase the frequency, what happens to the wavelength? Does it increase or decrease? 3. When you decrease the frequency, what happens to the wavelength? Does it increase or decrease? 4. What is the relationship between frequency and wavelength? Is it inverse or direct? |

**Skill 11.01 Exercise 3**

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| Refer to the colors below. Sort the colors from low to high with respect to frequency.   |  |  | | --- | --- | | **Color Range** | **Order of frequency (1 = lowest)** | | Reds |  | | Oranges |  | | Yellows |  | | Greens |  | | Blues |  | | Violets |  | |

**Skill 11.02 Exercise 1**

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| For each of the colors below,   1. Indicate the average wavelength in nanometers (nm) 2. Convert each wavelength to nanometers (1 x 10-9 m = 1 nm) 3. Calculate the frequency |
| |  |  |  |  | | --- | --- | --- | --- | | **Color Range** | **Average λ (nm)** | **λ (m)** | **ν (/s)** | | Reds | 710 nm |  |  | | Oranges |  |  |  | | Yellows |  |  |  | | Greens |  |  |  | | Blues |  |  |  | | Violets |  |  |  | |
| * 1. Are the wavelengths of ultra-violet light longer or shorter than that of visible light? |
| * 1. Are the frequencies of ultra-violet light longer or shorter than that of visible light? |
| * 1. Are the wavelengths of infra-red light longer or shorter than that of visible light? |
| * 1. Are the frequencies of infra-red light longer or shorter than that of visible light? |

**Skill 11.03 Exercise 1**

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| Now return to the simulator (<http://phet.colorado.edu/sims/html/wave-on-a-string/latest/wave-on-a-string_en.html>) Select the “Manual” option.  Move the wrench up and down as fast as you can and observe the wavelength. Now move the wrench up and down slowly and observe the wavelength.   1. When you moved the wrench up and down quickly (high energy), what happened to the wavelength? Did it increase or decrease?    1. When you moved the wrench up and down slowly (low energy), what happened to the wavelength? Did it increase or decrease? 2. What is the relationship between energy and wavelength? Is it inverse or direct? |

**Skill 11.03 Exercise 2**

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| Refer to the colors below. Sort the colors from low to high with respect to energy.   |  |  | | --- | --- | | **Color Range** | **Order of energy (1 = lowest)** | | Reds |  | | Oranges |  | | Yellows |  | | Greens |  | | Blues |  | | Violets |  | |

**Skill 11.04 Exercise 1**

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| Refer to problem 8.02 Problem 1   1. For each color, copy the corresponding frequency 2. Calculate the energy in joules |
| |  |  |  | | --- | --- | --- | | **Color Range** | **frequency** ν | **Energy (J)** | | Reds |  |  | | Oranges |  |  | | Yellows |  |  | | Greens |  |  | | Blues |  |  | | Violets |  |  | |
| 1. How do the energies of ultra-violet compare to visible light? |
| 1. How do the energies of infra-red light compare to visible light? |