## NATIONAL TAIWAN UNIVERSITY, GRADUATE INSTITUTE OF BIOMEDICAL ENGINEERING AND BIOINFORMATICS

## BEBI5009: Mathematical Modeling of System Biology Homework 4

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Consider the reaction network

$$A \xrightarrow{k_1} B + C$$
$$B + C \xrightarrow{k_{-1}} A$$

- a) Suppose that the system starts with two molecules of A, one molecule of B, and no molecules of C, that is  $(N_A, N_B, N_C) = (2, 1, 0)$ . Determine the set of possible states the system can adopt and write the chemical master equation that describes the corresponding probability distribution.
- b) Take  $k_1 = 1$  ( $time^{-1}$ ) and  $k_{-1} = 1$  ( $time^{-1}$ ), and solve for the steady-state probability distribution.
- c) Simulate sample paths of  $N_A$ ,  $N_B$ , and  $N_C$  using stochastic simulation algorithm (SSA). Set the initial condition of  $N_A$ ,  $N_B$ , and  $N_C$ =(2,1,0) or (10, 5, 0) or (100, 50, 0)

$$R1: A \rightarrow B + C$$
 propensity:  $k_1 N_A$   
 $R2: B + C \rightarrow A$  propensity:  $k_{-1} N_B N_C$ 

Note1: To calculate the waiting time, you will need to generate values of a random variable distributed according to an exponential distribution. You could simply generate a random variable U drawn from the uniform distribution on the unit interval (0, 1), and T=-ln(U)/ $\lambda$  will be an exponential random variable, where  $\lambda$  is the rate parameter of the exponential distribution. In this example,  $\lambda$  would be the sum of the reaction propensities.

Another way to generate exponential random number is to use Matlab function exprnd( $\mu$ ), where  $\mu$  is the mean of the exponential distribution  $\mu$ =1/ $\lambda$ .

Note2: You might need to change the value of  $k_1$  and  $k_{-1}$  for different initial conditions due to the volume changes. Use  $(k_1, k_{-1}) = (1,1)$ , (1, 1/5) and (1, 1/50)  $(time^{-1})$  for I.C.= (2,1,0), (10,5,0), and (100,50,0).

- d) Set the initial condition of  $N_A$ ,  $N_B$ , and  $N_C$ =(2, 1, 0). Analyzed the statistics of your ensemble, and compare to the steady state probability distribution in (b)
- e) Write down a deterministic model for the chemical reaction following the mass action law. Simulate the concentration changes of A, B C with the parameter  $k_1 = 1(s^{-1})$  and  $k_{-1} = 1$  ( $mM^{-1}s^{-1}$ ). Set initial concentrations [A],[B],[C] as 2,1, and 0mM. Compare simulation results in (c)