Midterm quiz

1 Magnetic versus electric dipolar energy (10 pts.)

- a) By any method you like, e.g. considering kinetic and potential energy plus Heisenberg's uncertainty, find the Bohr radius a_0 of hydrogen (assuming a fixed proton) in terms of the electron's mass m and charge e, and \hbar .
- b) Write down a typical "atomic value" for an electric dipole moment d.
- c) Find a typical "atomic" value for the electric field strength E.
- d) Considering the electron as a current loop of radius a_0 , find a typical atomic value for the magnetic moment μ (ignore numerical factors).
- e) Find the magnetic field strength B created by this current loop, again no prefactors required.
- f) Find the ratio of typical magnetic dipolar energy μB to electric dipolar energy dE in terms of the fine structure constant $\alpha = e^2/\hbar c$ (in cgs units. SI will make your life harder).

2 Fine structure in a magnetic field (20 pts)

An atom has orbital angular momentum L, spin angular momentum S, and no nuclear spin. The fine structure Hamiltonian is $H_{\rm FS} = A \boldsymbol{L} \cdot \boldsymbol{S}$. It is interacting with a magnetic field according to the Zeeman Hamiltonian $H_{\rm Z} = -\mu_B \left(g_S \boldsymbol{S} + \boldsymbol{L}\right) \cdot \boldsymbol{B}$. The total angular momentum is denoted $\boldsymbol{J} = \boldsymbol{L} + \boldsymbol{S}$.

- a) What are the "good" quantum numbers at low magnetic field ($\mu_B B \ll A$)?
- b) What are the "good" quantum numbers at high magnetic fields $(\mu_B B \gg A)$?
- c) What are the energy levels of the atom in zero magnetic field B = 0?
- d) What are the energy levels at very high field?
- e) At low magnetic fields one observes a linear Zeeman effect whose description involves the Landé g-factor. Derive the Landé g-factor in terms of g_s , S, L and other good quantum number(s) at low field. In terms of g, and relevant quantum number(s), write down the expression for the Zeeman energies at low field.

3 Microwave dressing of molecules (20 pts)

The rotational Hamiltonian of a molecule is $H = BJ^2$, where B is the rotational constant. In the molecule frame, the "permanent" dipole moment of the molecule is d. We would like to induce a strong dipole moment in the lab frame, but instead of using a large static electric field we will use microwaves tuned to rotational transitions. For simplicity, we will consider π -polarized microwaves, i.e. the electric field will oscillate along the z-direction, our quantization axis. So the perturbation is

$$V = -\mathbf{d} \cdot \mathbf{E}(t) = -dE \cos \theta \cos(\omega t)$$

- a) Starting with the molecule in the rotational ground state $|J=0,m_J=0\rangle$, which excited state can the microwave reach and what is the resonant frequency ω_0 for this transition?
- b) Derive the coupling matrix element between $|0,0\rangle \equiv |\downarrow\rangle$ and the excited state (denoted $|\uparrow\rangle$). This can be written as $\hbar\omega_R\cos\omega t$ with Rabi frequency ω_R that you will find. Note that $\cos\vartheta = \sqrt{\frac{4\pi}{3}} Y_{10}(\theta,\phi)$.
- c) Using the rotating wave approximation, derive a time-independent 2×2 Hamiltonian matrix describing the evolution in the presence of the microwave drive.
- d) Find the eigenstates (in the rotating frame, i.e. time-independent) in the case of a resonant drive $\omega = \omega_0$.
- e) Find the dipole moment in these two eigenstates.
- f) Starting in the state $|0,0\rangle$, sketch the dipole moment versus time. Choose a scale where dE is a good fraction of ω_0 , so that all relevant behavior is visible. At what times is the dipole moment at its largest magnitude? Note: Just a proper sketch suffices, you don't need to give the wavefunction as a function of time, although you might want to write that down if you are stuck, and derive the dipole moment vs. time.
- g) Instead of the behavior in f), we would like to find a microwave dressing scheme that smoothly "turns on" the molecule's dipole moment permanently. Can you propose a method that achieves this? Is there an adiabaticity criterion you need to fulfill, and if so, what is it?