

from the fingure the C.M.C is between C=8~10 104mol/dm3. C(104mol/dm3)
the C.M.C is about 9.×104 mol/dm3.

$$T = -\frac{1}{kT} \frac{dr}{d\ln c} = \frac{-1}{138 \times 10^{2} \times 98} \times \frac{dr}{d\ln c}$$
at c.m.c.  $\ln 9 = 2.2 \cdot \frac{dr}{d\ln c} = -8 \cdot (10^{-3} \text{ Vm})$ 

$$A = \frac{1}{p} = -1.38 \times 10^{-33} \times 98 \times \frac{1}{-8 \times 10^{-3}} = 50 \times 10^{-30} \frac{1}{\text{molecular}}$$





