00_03_1x-TAE-buffer-recipe

FREITAG, 14.5.2021

Goal-Setting

- Manufacture of 1x Tris-acetat-EDTA-buffer

Terms / abbreviations

- EDTA = Ethylenediaminetetraacetic acid
- HCl = Hydrochloric acid
- NaOH = Sodium hydroxide
- PAA = Polyacrylamide
- TAE = Tris-acetat-EDTA
- TRIS = Tris(hydroxymethyl)aminomethan

Risk areas

Hazard symbols

Required materials and / or information

- Chemicals:
  - Acetic acid, AppliChem
  - EDTA Lot, Sigma-Aldrich
  - MilliQ water, Sartorius arium pro VF
  - Tris base, AppliChem
- Materials:
  - Beaker/Schott flask
  - Stirring fish

- All chemicals are solid substances

Templates, devices, software

- Analysis balance, Kern ABJ 220-4NM
- pH meter, Knick Digital pH-Meter 646

Preliminary work

- Calculate the needed masses for specific volumina
- Search for gloves and safety goggles

Operation

- pH meter (ready to use in the chemical room)
  - If it is used for the first time, ask an experienced person around how to use it
  - Regularly calibrate the pH meter according to manufacturer
  - User guide:

        - pH meter should measure about 6.5 when the solution is stored (small Falcon Tube)
        - Take it out and unlock it (small switch at the top)
        - Wash it carefully with some MilliQ water
        - Hold into the solution and measure
        - Wash again after use, turn off the switch and put it back into the small Falcon Tube

The following recipe is standardized to 1 L

1. Weigh the following components into a beaker or schottflask --> Tip use at least a 1500 mL beaker (easier usage/ measurement)
    - Tris base: 121.14 g/L
    - Acetic acid: 60.05 g/L
    - EDTA: 2.92 g/L
2. Fill up to 1 L with MilliQ water
3. Mix by stirring/shaking/inverting
4. Optimal pH = 8.0
    a. If pH is too low, adjust with NaOH
    b. If the pH is too high, do not adjust it with HCl, as this would cause errors with the PAA gel, because the Cl^- ions will migrate first

Disposal

- Observe all federal, state and local environmental regulations
- Here: Black canister on the ground next to trash cans in gel electrophoresis room

Troubleshooting

- None

Follow-up work

- If a chemical runs out or is empty, please fill out the order list on the fridge next to the window
- 01_01_Agarose-gel-preparation
- 01_03_Performing-agarose-gel-electrophoresis