

Reaction Engineering

Class Test-3, Marks-10, Time: 30 minutes

1. (a) An aqueous feed of A and B(400L/min) with $C_{A0} = 100$ mmol/L and $C_{B0} = 200$ mmol/L is to be converted to product in a plug flow reactor. The reaction kinetic and the stoichiometry are given below.
 $A + B \rightarrow R$, $-r_A = 150 C_A C_B$ mol/(L. min). Estimate the volume of plug flow reactor for 95% conversion of A. [5]
- (b) Explain the procedure for the test of kinetic equation of an elementary gas-phase reaction $A \rightleftharpoons R$ using a plug flow reactor. [5]