

```
In[47]:= << VariationalMethods`
```

Introducing Lagrange's equation:

```
In[48]:= (*Introducing Kinetic energy 'T' and potential energy 'U' for N atoms*)
```

$$T[t_ , N_] := \frac{1}{2} \left(\sum_{i=1}^N \dot{x}_i[t]^2 \right)$$

$$U[t_ , N_] := \frac{1}{2} \omega_0^2 \left(x_1[t]^2 + x_N[t]^2 + \sum_{i=2}^N (x_i[t] - x_{i-1}[t])^2 \right)$$

```
In[50]:= (* Defining the Lagrangian to be L = T - U*)
```

```
L[N_] := T[t, N] - U[t, N]
```

```
In[51]:= (* Using mathematica to apply Euler's  
equation on 'L' to find the equations of motion*)
```

```
EoM[N_, j_] := EulerEquations[L[N], Subscript[x, j][t], t]
```

```
In[52]:= (* Showing the equation of motion for the first,  
middle, and the last atom for a system of 3 atoms*)
```

```
In[53]:= Expand@EoM[3, 1] // TraditionalForm (* the equation of motion of the first atom*)
```

```
Out[53]//TraditionalForm=
```

$$-x_1''(t) - 2x_1(t) + x_2(t) = 0$$

```
In[54]:= Expand@EoM[3, 2] // TraditionalForm (* the equation of motion of the middle atom*)
```

```
Out[54]//TraditionalForm=
```

$$-x_2''(t) + x_1(t) - 2x_2(t) + x_3(t) = 0$$

```
In[55]:= Expand@EoM[3, 3] // TraditionalForm (* the equation of motion of a final atom*)
```

```
Out[55]//TraditionalForm=
```

$$-x_3''(t) + x_2(t) - 2x_3(t) = 0$$

Constructing the matrix D_n and evaluating its eigenmodes:

```
In[56]:= (* Introducing Dn, the matrix of the coefficients of the equations of motion*)
Dn[N_] := Table[Coefficient[-ω₀² (xᵢ₋₁[t] - 2 xᵢ[t] + xᵢ₊₁[t]), xⱼ[t]], {i, N}, {j, N}];

(* Introducing Dω which simply a diagonal matrix with ω² in its diagonal *)
Dω[N_] := DiagonalMatrix[Table[ω², {i, N}], 0, N];

(*Evaluatong Dn[5] and Dn[5]-Dω[5] to show a sample of the matrix we will solve*)
MatrixForm@Dn[5] // TraditionalForm
MatrixForm@ (Dn[5] - Dω[5]) // TraditionalForm
```

Out[58]//TraditionalForm=

$$\begin{pmatrix} 2 & -1 & 0 & 0 & 0 \\ -1 & 2 & -1 & 0 & 0 \\ 0 & -1 & 2 & -1 & 0 \\ 0 & 0 & -1 & 2 & -1 \\ 0 & 0 & 0 & -1 & 2 \end{pmatrix}$$

Out[59]//TraditionalForm=

$$\begin{pmatrix} 2 - \omega^2 & -1 & 0 & 0 & 0 \\ -1 & 2 - \omega^2 & -1 & 0 & 0 \\ 0 & -1 & 2 - \omega^2 & -1 & 0 \\ 0 & 0 & -1 & 2 - \omega^2 & -1 \\ 0 & 0 & 0 & -1 & 2 - \omega^2 \end{pmatrix}$$

```
In[60]:= (* Setting ω₀ = 1 for numerical Computation *)
ω₀ = 1;
```

```
In[61]:= (*Defining the Eigenmodes of Dn*)
EigValSol[n_] := Eigenvalues[N[Dn[n]]];
```

```
In[62]:= (* Solving the eigenvalues for N = 1700 atom, then save it in a variable *)
ListOfSol = EigValSol[1700];
```

Defining the density of state and plot it alongside the plot of eigenmodes' distribution:

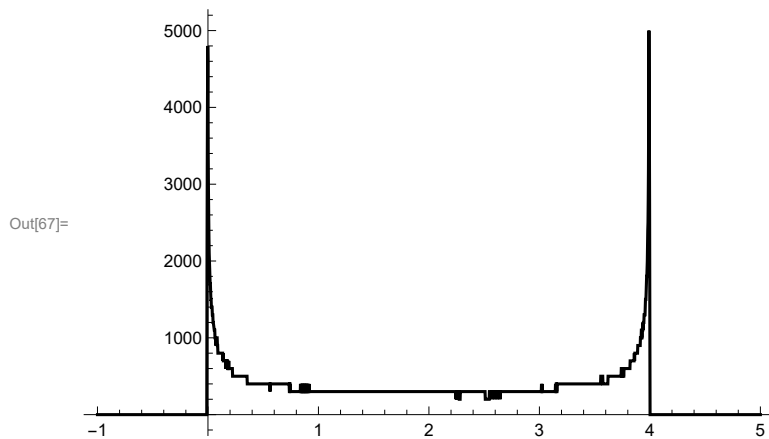
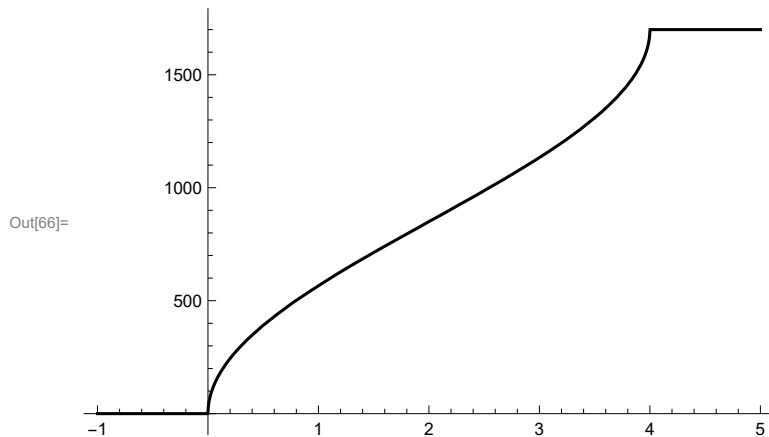
```
In[63]:= (* The length of 'Cases' function will
return the number of eigenmodes that are less than ω*)
NLω[ω_] := Length@Cases[ListOfSol, λ_ /; λ < ω]
```

```
In[64]:= (* Defining the density of state and taking dω=0.01 for smooth curve,
less than this value will cause fluctuations due to the numerical approach *)
ρ[ω_, dω_] := 
$$\frac{NL\omega[\omega + d\omega] - NL\omega[\omega]}{d\omega}$$

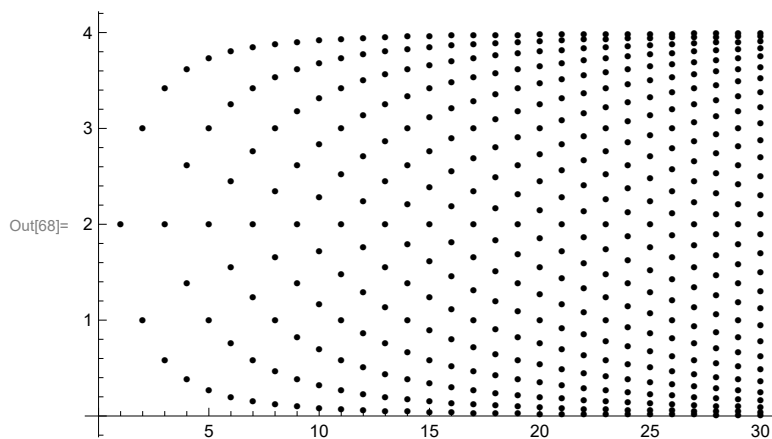
```

```
In[65]:= (* Plotting the Number of Eigenmodes that is
less than  $\omega$  and its derivative: The density of state*)
```

```
In[66]:= Plot[NL $\omega[\omega]$ , { $\omega$ , -1, 4  $\omega_0^2 + 1$ }, PlotPoints → 45,
MaxRecursion → 6, PlotRange → All, PlotStyle → Black]
Plot[ $\rho[\omega, 0.01]$ , { $\omega$ , -1, 4  $\omega_0^2 + 1$ }, PlotPoints → 45,
MaxRecursion → 6, PlotRange → All, PlotStyle → Black]
```



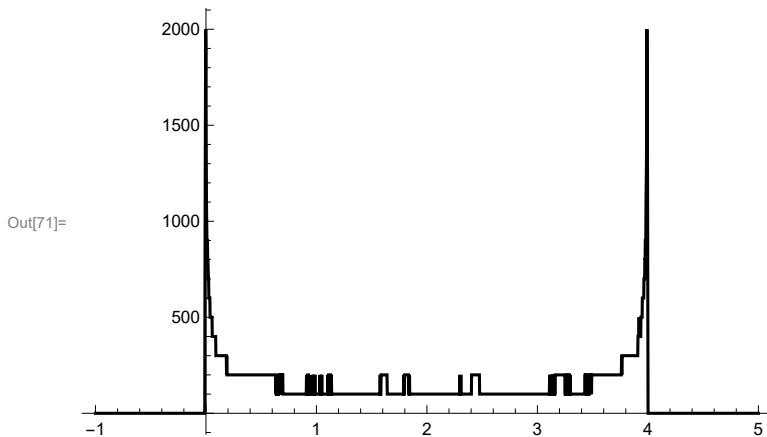
```
In[68]:= (* The distribution of the Eigenmodes of the matrix Dn *)
Show[Table[DiscretePlot[EigValSol[1], {n, 1, 1}, Filling → None, PlotStyle → {Black},
PlotMarkers → {"•", Medium}], {1, 30}], PlotRange → {{0, 30}, {0, 4}}]
```



Investigation the relationship between ω_0^2 and ω_{\max}^2 .

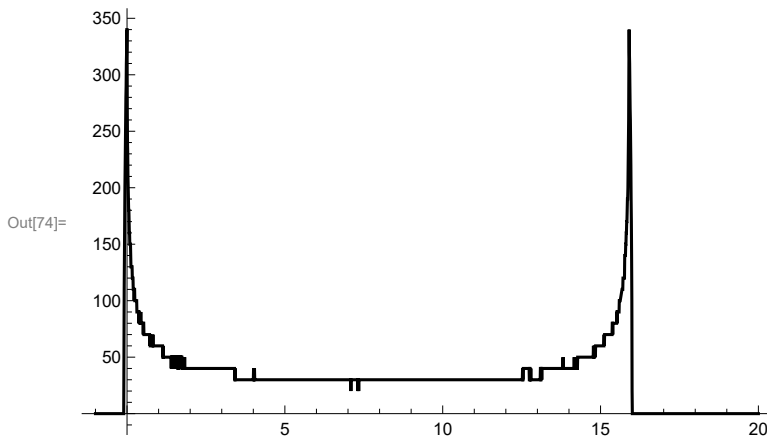
Define ω_{\max}^2 to be the value of ω that the density of state will converge to zero afterwards, for lets take $\omega_0^2 = 1$, lets observe the value of ω_{\max}^2

```
In[69]:=  $\omega_0 = 1;$ 
ListOfSol = EigValSol[700];
Plot[ $\rho[\omega, 0.01]$ , { $\omega$ , -1, 5}, PlotPoints → 45,
MaxRecursion → 6, PlotRange → All, PlotStyle → Black]
```



The value of ω_{\max}^2 for $\omega_0 = 1$ was $\omega_{\max}^2 = 4$, Now we will take $\omega_0 = 2 \rightarrow \omega_0^2 = 4$, and we will observe ω_{\max}^2

```
In[72]:=  $\omega_0 = 2;$ 
ListOfSol = EigValSol[700];
Plot[ $\rho[\omega, 0.1]$ , { $\omega$ , -1, 20}, PlotPoints → 45,
MaxRecursion → 6, PlotRange → All, PlotStyle → Black]
```

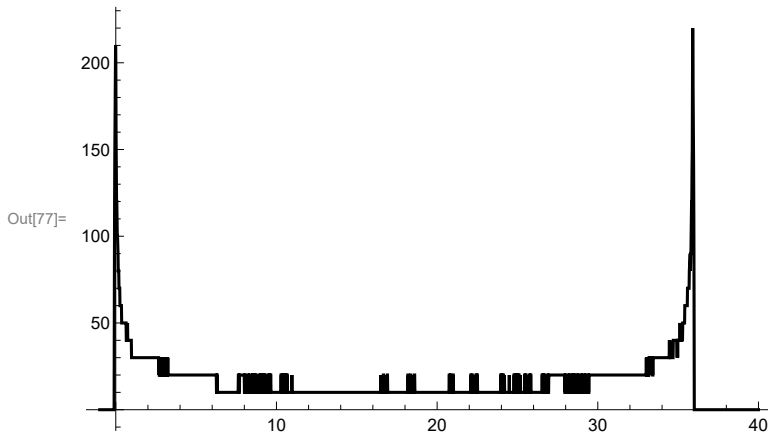


The value of ω_{\max}^2 for $\omega_0 = 2$ was $\omega_{\max}^2 = 16$, Now we will take $\omega_0 = 3 \rightarrow \omega_0^2 = 9$, and we will observe ω_{\max}^2

```

In[75]:=  $\omega_0 = 3$ ;
ListOfSol = EigValSol[700];
Plot[ $\rho[\omega, 0.1]$ , { $\omega$ , -1, 40}, PlotPoints → 45,
MaxRecursion → 6, PlotRange → All, PlotStyle → Black]

```

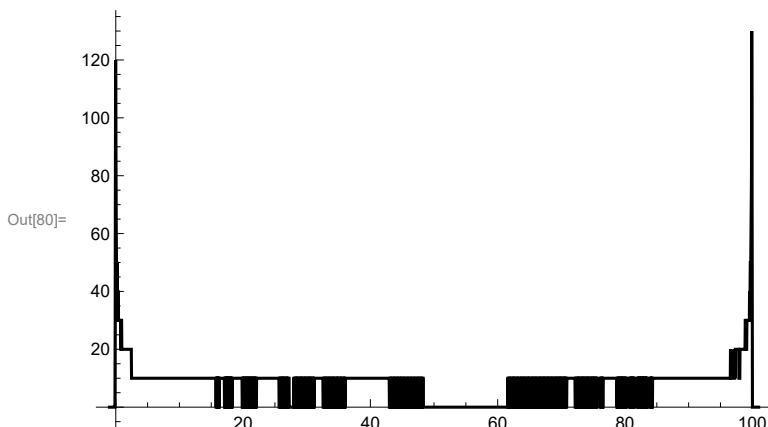


The value of ω_{\max}^2 for $\omega_0 = 3$ was $\omega_{\max}^2 = 36$. We can now describe the relationship between ω_{\max}^2 and ω_0^2 by this empirical formula: $\omega_{\max}^2 = 4 \omega_0^2$, we can verify it by choosing a value of ω_0^2 then predict the value of ω_{\max}^2 and check if we were accurate. Lets take $\omega_0 = 5 \rightarrow \omega_0^2 = 25$, we predict ω_{\max}^2 to be 100:

```

In[78]:=  $\omega_0 = 5$ ;
ListOfSol = EigValSol[700];
Plot[ $\rho[\omega, 0.1]$ , { $\omega$ , -1,  $4 \omega_0^2 + 1$ }, PlotPoints → 45,
MaxRecursion → 6, PlotRange → All, PlotStyle → Black]

```



The empirical formula $\omega_{\max}^2 = 4 \omega_0^2$ appears to work fine for this system of atoms.