

Chapter 32

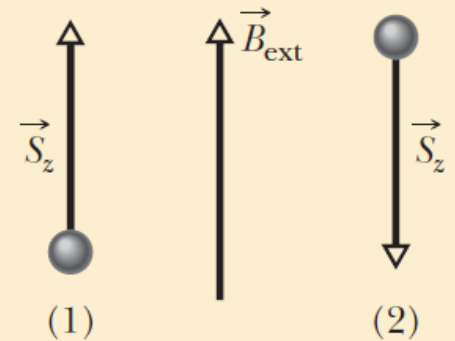
Maxwell Equations; Magnetism of Matter

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Checkpoint 4

The figure here shows the spin orientations of two particles in an external magnetic field \vec{B}_{ext} . (a) If the particles are electrons, which spin orientation is at lower energy? (b) If, instead, the particles are protons, which spin orientation is at lower energy?



(a) 2; (b) 1



Checkpoint 5

The figure shows two diamagnetic spheres located near the south pole of a bar magnet. Are (a) the magnetic forces on the spheres and (b) the magnetic dipole moments of the spheres directed toward or away from the bar magnet? (c) Is the magnetic force on sphere 1 greater than, less than, or equal to that on sphere 2?



1



2



(a) away; (b) away; (c) less

32-7 Paramagnetism



A paramagnetic material placed in an external magnetic field \vec{B}_{ext} develops a magnetic dipole moment in the direction of \vec{B}_{ext} . If the field is nonuniform, the paramagnetic material is attracted *toward* a region of greater magnetic field *from* a region of lesser field.

Paramagnetic materials have atoms with a permanent magnetic dipole moment but the moments are randomly oriented, with no net moment, unless the material is in an external magnetic field \mathbf{B}_{ext} where the dipoles tend to align with that field. The extent of alignment within a volume V is measured as the magnetization M , given by

$$M = \frac{\text{measured magnetic moment}}{V}.$$

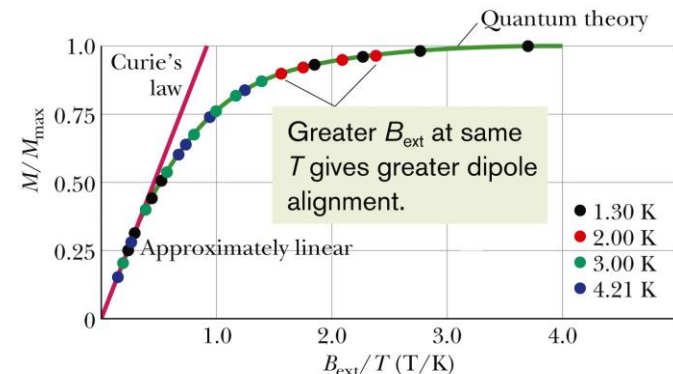
Complete alignment (saturation) of all N dipoles in the volume gives a maximum value $M_{\text{max}} = N\mu/V$.

At low values of the ratio $\mathbf{B}_{\text{ext}}/T$,

$$M = C \frac{B_{\text{ext}}}{T}$$

where T is the temperature (in kelvins) and C is a material's Curie constant.

In a nonuniform external field, a paramagnetic material is attracted to the region of greater magnetic field.



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Richard Megna/Fundamental Photographs
Liquid oxygen is suspended between the two pole faces of a magnet because the liquid is paramagnetic and is magnetically attracted to the magnet.



Checkpoint 6

The figure here shows two paramagnetic spheres located near the south pole of a bar magnet. Are

1

2

S

N

(a) the magnetic forces on the spheres and (b) the magnetic dipole moments of the spheres directed toward or away from the bar magnet?
(c) Is the magnetic force on sphere 1 greater than, less than, or equal to that on sphere 2?

(a) toward; (b) toward; (c) less

Sample Problem 32.03 Orientation energy of a paramagnetic gas in a magnetic field

A paramagnetic gas at room temperature ($T = 300 \text{ K}$) is placed in an external uniform magnetic field of magnitude $B = 1.5 \text{ T}$; the atoms of the gas have magnetic dipole moment $\mu = 1.0\mu_B$. Calculate the mean translational kinetic energy K of an atom of the gas and the energy difference ΔU_B between parallel alignment and antiparallel alignment of the atom's magnetic dipole moment with the external field.

KEY IDEAS

(1) The mean translational kinetic energy K of an atom in a gas depends on the temperature of the gas. (2) The energy U_B of a magnetic dipole $\vec{\mu}$ in an external magnetic field \vec{B} depends on the angle θ between the directions of $\vec{\mu}$ and \vec{B} .

Calculations: From Eq. 19-24, we have

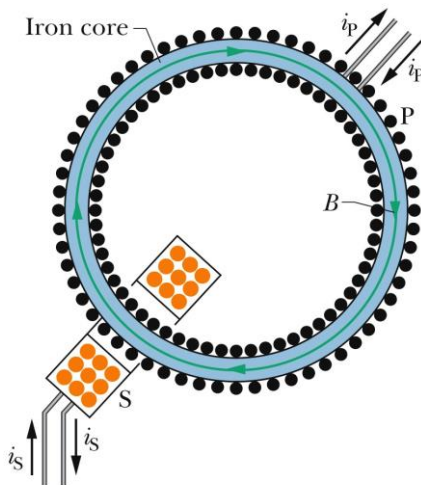
$$\begin{aligned} K &= \frac{3}{2}kT = \frac{3}{2}(1.38 \times 10^{-23} \text{ J/K})(300 \text{ K}) \\ &= 6.2 \times 10^{-21} \text{ J} = 0.039 \text{ eV}. \end{aligned} \quad (\text{Answer})$$

From Eq. 28-38 ($U_B = -\vec{\mu} \cdot \vec{B}$), we can write the difference ΔU_B between parallel alignment ($\theta = 0^\circ$) and antiparallel alignment ($\theta = 180^\circ$) as

$$\begin{aligned} \Delta U_B &= -\mu B \cos 180^\circ - (-\mu B \cos 0^\circ) = 2\mu B \\ &= 2\mu_B B = 2(9.27 \times 10^{-24} \text{ J/T})(1.5 \text{ T}) \\ &= 2.8 \times 10^{-23} \text{ J} = 0.00017 \text{ eV}. \end{aligned} \quad (\text{Answer})$$

Here K is about 230 times ΔU_B ; so energy exchanges among the atoms during their collisions with one another can easily reorient any magnetic dipole moments that might be aligned with the external magnetic field. That is, as soon as a magnetic dipole moment happens to become aligned with the external field, in the dipole's low energy state, chances are very good that a neighboring atom will hit the atom, transferring enough energy to put the dipole in a higher energy state. Thus, the magnetic dipole moment exhibited by the paramagnetic gas must be due to fleeting partial alignments of the atomic dipole moments.

32-8 Ferromagnetism

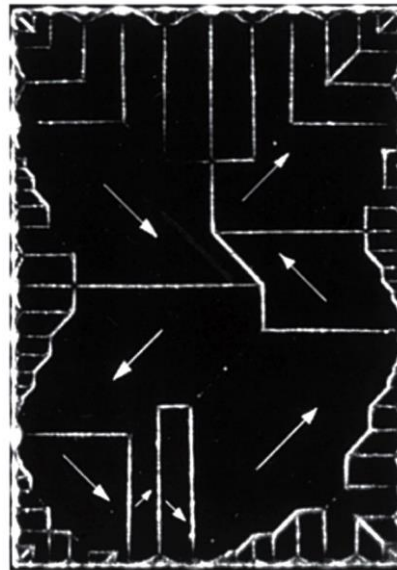


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A ferromagnetic material placed in an external magnetic field \vec{B}_{ext} develops a strong magnetic dipole moment in the direction of \vec{B}_{ext} . If the field is nonuniform, the ferromagnetic material is attracted *toward* a region of greater magnetic field *from* a region of lesser field.

The magnetic dipole moments in a ferromagnetic material can be aligned by an external magnetic field and then, after the external field is removed, remain partially aligned in regions known as **magnetic domains**.

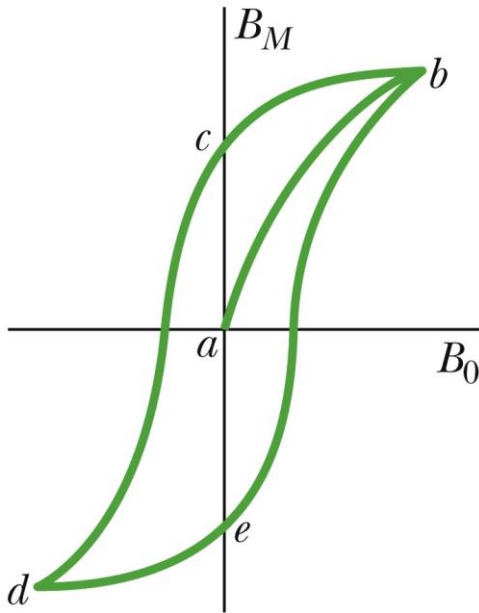


Courtesy Ralph W. DeBlois

A photograph of domain patterns within a single crystal of nickel; white lines reveal the boundaries of the domains. The white arrows superimposed on the photograph show the orientations of the magnetic dipoles within the domains and thus the orientations of the net magnetic dipoles of the domains. The crystal as a whole is unmagnetized if the net magnetic field (the vector sum over all the domains) is zero.

A Rowland ring. A primary coil P has a core made of the ferromagnetic material to be studied (here iron). The core is magnetized by a current i_P sent through coil P. (The turns of the coil are represented by dots.) The extent to which the core is magnetized determines the total magnetic field \mathbf{B} within coil P. Field \mathbf{B} can be measured by means of a secondary coil S.

32-8 Ferromagnetism



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A ferromagnetic material placed in an external magnetic field \vec{B}_{ext} develops a strong magnetic dipole moment in the direction of \vec{B}_{ext} . If the field is nonuniform, the ferromagnetic material is attracted *toward* a region of greater magnetic field *from* a region of lesser field.

The lack of retraceability shown in the Figure is called **hysteresis**, and the curve $bcdeb$ is called a **hysteresis loop**. Note that at points c and e the iron core is magnetized, even though there is no current in the toroid windings; this is the familiar phenomenon of permanent magnetism.

Hysteresis can be understood through the concept of magnetic domains. Evidently the motions of the domain boundaries and the reorientations of the domain directions are not totally reversible. When the applied magnetic field B_0 is increased and then decreased back to its initial value, the domains do not return completely to their original configuration but retain some “memory” of their alignment after the initial increase. This memory of magnetic materials is essential for the magnetic storage of information.

Sample Problem 32.04 Magnetic dipole moment of a compass needle

A compass needle made of pure iron (density 7900 kg/m^3) has a length L of 3.0 cm, a width of 1.0 mm, and a thickness of 0.50 mm. The magnitude of the magnetic dipole moment of an iron atom is $\mu_{\text{Fe}} = 2.1 \times 10^{-23} \text{ J/T}$. If the magnetization of the needle is equivalent to the alignment of 10% of the atoms in the needle, what is the magnitude of the needle's magnetic dipole moment $\vec{\mu}$?

KEY IDEAS

(1) Alignment of all N atoms in the needle would give a magnitude of $N\mu_{\text{Fe}}$ for the needle's magnetic dipole moment $\vec{\mu}$. However, the needle has only 10% alignment (the random orientation of the rest does not give any net contribution to $\vec{\mu}$). Thus,

$$\mu = 0.10N\mu_{\text{Fe}}. \quad (32-42)$$

(2) We can find the number of atoms N in the needle from the needle's mass:

$$N = \frac{\text{needle's mass}}{\text{iron's atomic mass}}. \quad (32-43)$$

Finding N : Iron's atomic mass is not listed in Appendix F, but its molar mass M is. Thus, we write

$$\text{iron's atomic mass} = \frac{\text{iron's molar mass } M}{\text{Avogadro's number } N_A}. \quad (32-44)$$

Next, we can rewrite Eq. 32-43 in terms of the needle's mass m , the molar mass M , and Avogadro's number N_A :

$$N = \frac{mN_A}{M}. \quad (32-45)$$

The needle's mass m is the product of its density and its volume. The volume works out to be $1.5 \times 10^{-8} \text{ m}^3$; so

$$\begin{aligned} \text{needle's mass } m &= (\text{needle's density})(\text{needle's volume}) \\ &= (7900 \text{ kg/m}^3)(1.5 \times 10^{-8} \text{ m}^3) \\ &= 1.185 \times 10^{-4} \text{ kg}. \end{aligned}$$

Substituting into Eq. 32-45 with this value for m , and also 55.847 g/mol ($= 0.055847 \text{ kg/mol}$) for M and 6.02×10^{23} for N_A , we find

$$\begin{aligned} N &= \frac{(1.185 \times 10^{-4} \text{ kg})(6.02 \times 10^{23})}{0.055847 \text{ kg/mol}} \\ &= 1.2774 \times 10^{21}. \end{aligned}$$

Finding μ : Substituting our value of N and the value of μ_{Fe} into Eq. 32-42 then yields

$$\begin{aligned} \mu &= (0.10)(1.2774 \times 10^{21})(2.1 \times 10^{-23} \text{ J/T}) \\ &= 2.682 \times 10^{-3} \text{ J/T} \approx 2.7 \times 10^{-3} \text{ J/T}. \end{aligned} \quad (\text{Answer})$$

32 Summary

Gauss' Law for Magnetic Fields

- Gauss' law for magnetic fields,

$$\Phi_B = \oint \vec{B} \cdot d\vec{A} = 0, \quad \text{Eq. 32-1}$$

Maxwell's Extension of Ampere's Law

- A changing electric field induces a magnetic field given by,

$$\oint \vec{B} \cdot d\vec{s} = \mu_0 \epsilon_0 \frac{d\Phi_E}{dt} \quad \text{Eq. 32-3}$$

- Maxwell's law and Ampere's law can be written as the single equation

$$\oint \vec{B} \cdot d\vec{s} = \mu_0 \epsilon_0 \frac{d\Phi_E}{dt} + \mu_0 i_{\text{enc}} \quad \text{Eq. 32-5}$$

Displacement Current

- We define the fictitious displacement current due to a changing electric field as

$$i_d = \epsilon_0 \frac{d\Phi_E}{dt}. \quad \text{Eq. 32-10}$$

- Equation 32-5 then becomes

$$\oint \vec{B} \cdot d\vec{s} = \mu_0 i_{d,\text{enc}} + \mu_0 i_{\text{enc}} \quad \text{Eq. 32-11}$$

Maxwell's Equations

- Four equations are as follows:

$$\oint \vec{E} \cdot d\vec{A} = q_{\text{enc}}/\epsilon_0$$

$$\oint \vec{B} \cdot d\vec{A} = 0$$

$$\oint \vec{E} \cdot d\vec{s} = -\frac{d\Phi_B}{dt}$$

$$\oint \vec{B} \cdot d\vec{s} = \mu_0 \epsilon_0 \frac{d\Phi_E}{dt} + \mu_0 i_{\text{enc}}$$

32 Summary

Spin Magnetic Dipole Moment

- Spin angular momentum of electron is associated with spin magnetic dipole momentum through,

$$\vec{\mu}_s = -\frac{e}{m} \vec{S}. \quad \text{Eq. 32-22}$$

- For a measurement along a z axis, the component S_z can have only the values given by

$$S_z = m_s \frac{h}{2\pi}, \quad \text{for } m_s = \pm \frac{1}{2}, \quad \text{Eq. 32-23}$$

- Similarly,

$$\mu_{s,z} = \pm \frac{eh}{4\pi m} = \pm \mu_B, \quad \text{Eq. 32-24 \& 26}$$

- Where the Bohr magneton is

$$\mu_B = \frac{eh}{4\pi m} = 9.27 \times 10^{-24} \text{ J/T}. \quad \text{Eq. 32-25}$$

- The energy U

$$U = -\vec{\mu}_s \cdot \vec{B}_{\text{ext}} = -\mu_{s,z} B_{\text{ext}}. \quad \text{Eq. 32-27}$$

Orbital Magnetic Dipole Momentum

- Angular momentum of an electron is associated with orbital magnetic dipole momentum as

$$\vec{\mu}_{\text{orb}} = -\frac{e}{2m} \vec{L}_{\text{orb}}. \quad \text{Eq. 32-28}$$

- Orbital angular momentum is quantized,

$$L_{\text{orb},z} = m_\ell \frac{h}{2\pi},$$

for $m_\ell = 0, \pm 1, \pm 2, \dots, \pm (\text{limit}).$ Eq. 32-29

- The associated magnetic dipole moment is given by

$$\mu_{\text{orb},z} = -m_\ell \frac{eh}{4\pi m} = -m_\ell \mu_B. \quad \text{Eq. 32-30\&31}$$

- The energy U

$$U = -\vec{\mu}_{\text{orb}} \cdot \vec{B}_{\text{ext}} = -\mu_{\text{orb},z} B_{\text{ext}}. \quad \text{Eq. 32-32}$$

32 Summary

Diamagnetism

- Diamagnetic materials exhibit magnetism only when placed in an external magnetic field; there they form magnetic dipoles directed opposite the external field. In a nonuniform field, they are repelled from the region of greater magnetic field.

Paramagnetism

- Paramagnetic materials have atoms with a permanent magnetic dipole moment but the moments are randomly oriented unless the material is in an external magnetic field. The extent of alignment within a volume V is measured as the magnetization M , given by

$$M = \frac{\text{measured magnetic moment}}{V}. \quad \text{Eq. 32-28}$$

- Complete alignment (saturation) of all N dipoles in the volume gives a maximum value $M_{\max} = N\mu/V$. At low values of the ratio $\mathbf{B}_{\text{ext}}/T$,

$$M = C \frac{B_{\text{ext}}}{T} \quad \text{Eq. 32-39}$$

Ferromagnetism

- The magnetic dipole moments in a ferromagnetic material can be aligned by an external magnetic field and then, after the external field is removed, remain partially aligned in regions (domains). Alignment is eliminated at temperatures above a material's Curie temperature. In a nonuniform external field, a ferromagnetic material is attracted to the region of greater magnetic field.