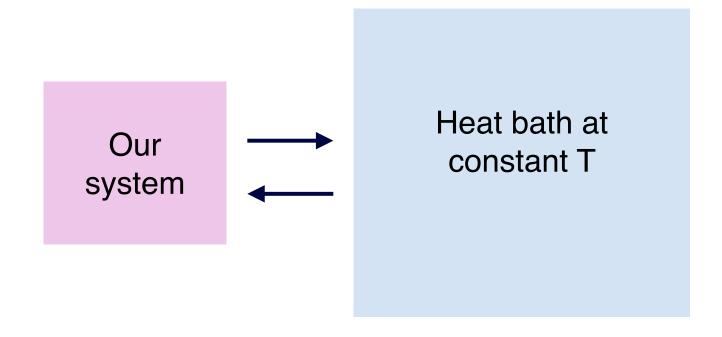
Why do we need to thermostat molecular dynamics?

- Molecular dynamics is a solution to the initial value partial differential equations defined by Newton's equation of motion
- Given x,v (or dx/dt) for all particles
 - F=ma means $d^2x/dt^2 = -1/m dU/dx$
 - We want to know: what are x and dx/dt of each particle at all future times?
- Newton's laws of motion conserve energy
- Therefore this gives constant number of particles, constant volume, and constant energy
- But we want answers at, say 300 K.



We want to simulate properties at constant T



Exchange thermal energy until at equilibrium



How does REAL LIFE thermal equilibration happen?

- Our system is surrounded by other molecules
- They bang into our system, and exchange kinetic energy
 - If they are hotter than the system average, then energy goes into the system
 - If they are cooler than the system average, then energy goes out of the system



How do we get thermal energy into and out of the system?

- We modify the kinetic energies of the particles
- First guess:
 - If the average thermal energy is too high, decrease the velocity of all particles
 - If the average thermal energy is too **low**, increase the velocity of **all** particles
- This is the weak coupling thermostat
- In GROMACS, tcoup1 = berendsen
- Need to set ref_t as well!



How quickly do we pump energy in or out?

- Let's say the system has 110 kJ mol. At the temperature you specified, it should have 100 kJ/mol on average
 - Do you take out 10 kJ/mol each step?
 - Or do you take out 1 kJ/mol each step?
- This will define a rate at which excess kinetic energy is removed

$$dK/dt = k(K-K_0)$$

- k has units of $1/\text{time} = 1/\tau$
- In GROMACS, this is tau t.



Bussi-Parrinello

- A simulation should not have a single kinetic energy.
 It should have a Maxwell-Boltzmann kinetic energy distribution.
- Don't target a fixed kinetic energy K. Target at random K, chosen from the Maxwell-Boltzmann distribution
- This gives a correct distribution of kinetic energies
- Rigorously consistent with NVT and exp(-βU)

 $dK/dt = k(K-K_{Maxwell-Boltzmann})$

In GROMACS, tcoup1 = v-rescale



Nose-Hoover thermostat

- Also a scaling thermostat:
 - If kinetic energy K is too big, make all particles slower
- Math is too complicated for today
- But it is essentially a second-order thermostat

$$dK/dt = k^2(K-K_0)^2$$

- Relaxes to right K faster than Berendsen
- BUT oscillates back and forth
- Pretty much gives the right NVT



Langevin dynamics

$$mdv/dt = F + R - \gamma v$$

- Add thermal energy by adding random noise
 - Like a particle colliding!
 - But, all of the particles are colliding, even ones inside your system
- Suck excess thermal energy awa with a friction term
 - Like moving through a fluid, which do necessarily make sense if you are simulating your fluid already . . .
- Balance the friction loss and random noise to be equal when you are at the right T $\langle R \rangle = 0 \ \langle R^2 \rangle = 2k_BT\gamma$
 - Implemented as integrator = sd in gromacs.
 - Very good stability!
 - If you add too much random noise, and suck out too much velocity, it slows the dynamics.



Do I need a thermostat?

- Unless you are measuring transport properties, almost certainly yes.
- It is VERY EASY to start a simulation at too high a total E. Need to maintain T!
- If your simulation is SLIGHTLY not obeying F=ma, it will keep the simulation from accumulating or losing much energy



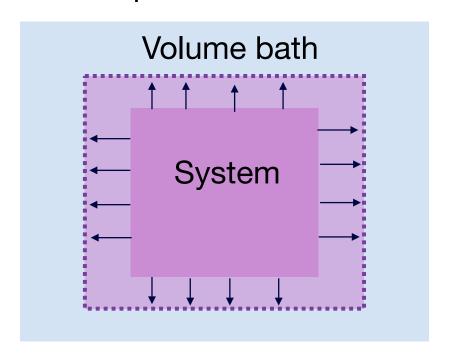
Thermostat recommendations

- Langevin is most stable, but slows system down a bit, especially at large
 - The ONLY one you should use for gases
- Bussi-Parrinello is simple and usually very good
- Parrinello-Rahman is OK for liquids, not so good for gases/solids
- Berendsen is only good to equilibrate a system.



Barostats

- So: We know how to handle NVT
- What about NPT
 - Where the volume is not constant?
 - System exchanges volume with the surroundings until in mechanical equilibrium with its environment.





How does a real life barostat work?

- Particles collide with your system.
 - If there are lots of hard collisions with your system, it pushes your system together
 - If there are few hard collisions with your system, your system pushes out.



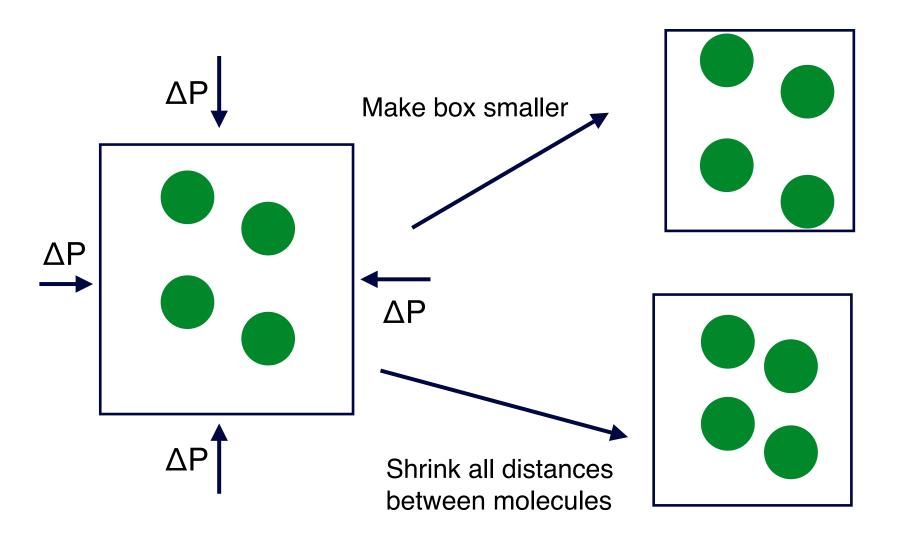
How does an MD barostat work?

$$P = \left\langle \frac{1}{V} \left(2K + \frac{1}{3} \sum_{i>j} r_{ij} \cdot F(r_{ij}) \right) \right\rangle$$

- We compare the quantity in the brackets to P
- It can oscillate wildly (like, several 1000 atm)
- Again, there is no such thing as an instantaneous pressure.
 - Just frames where more of the system wants to breath out, and frames where more of the system wants to breath in.

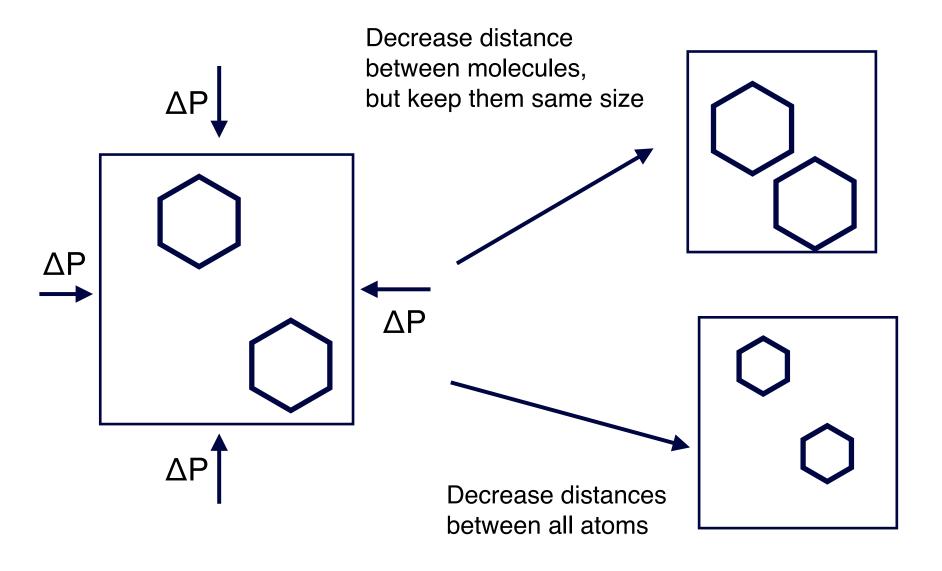


How does an MD barostat work?





Do we shrink molecules??





How a barostat works

- "P"estimate > Pdesired
 - Shrink box by some factor s
- "P"estimate < Pdesired
 - Expand box by some factor s
- HOW BIG should s be?
- Assume s is proportional to P_{estimate} -P_{desired}
- What is the constant of proportionality, i.e. how QUICKLY does one respond to differences in pressure?
- Governed by tau_p
- If we say: $V_{new}/V_{old} = k(P_{estimate}-P_{desired})$
- Berendsen's weak coupling barostat
- pcouple=Berendsen, and set ref_p as well!



Parinello-Rahman barostat

- Parrinello-Rahman barostat
 - Similar theory to Nose-Hoover thermostat
 - 2nd order, so approaches true answer with some oscillations
 - Common in most MD packages
 - Rigorously* gives NPT
 - Has some minor errors if you have constraints, like rigid water, hydrogen bond constraints
 - Tends to be somewhat numerically unstable, especially for small systems
 - pcoupl = Parrinello-Rahman



Parrinello-Rahman barostat

- Cell-resizing thermostat
 - Very new!
 - Similar theory to Bussi-Parrinello barostat
 - Like Berendsen, but with added fluctuations to get NPT right.
 - Rigorously gives NPT
 - Not that well understood where it might fail, but in theory it might be the best wone.
 - •pcoupl = c-rescale



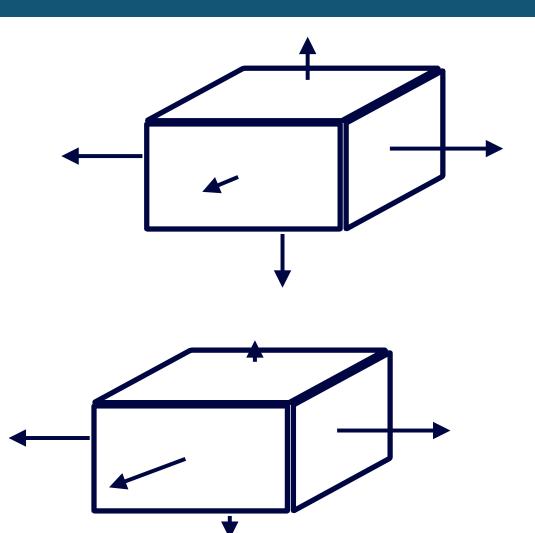
Monte Carlo Barostat

- Change the system volume by a small amount
- If U+PV is lower, then accept the change
- If U+PV is higher, then roll a random number to accept
 - If accepted, then shrink/expand the box
- Gives correct NPT volume distributions!
- Breaks the dynamics
- Barostats already break the dynamics, since particles don't magically move together and apart in nature



Other possibilities

- Isotropic
 - Box side ratios do not change
- Anisotropic:
 - All box sides can move independently.
 - The ratio of the sides can change
- Semiisotropic
 - x and y isotropically, but not z
 - Common for bilayer/ lamellar simulations



• In GROMACS, pcoupltype = anisotropic



What can go wrong with a barostat

- The systems of equations are significantly less stable than MD alone or MD + barostat.
- More likely to crash.
 - If you start from a bad configuration.
 - If the system is poorly equilibrated
 - If the system is very small.
 - If there is some sort of phase separation or self-assembly.



Do I need a barostat?

- You almost certainly need a barostat to relax your simulation to the equilibrium volume.
 - You don't want to run at a density that is too high or too low
 - Like running at 1000 atm, or bubbles could form.
- BUT if you system is highly incompressible, you may not need to run your simulation at NPT once it reaches the right P
 - You could average the volume over the last 100 ps, find the average volume, and pick a frame with that average volume.
- OFTEN results are OK with the Berendsen barostat



Some guidelines

• Thermostats:

- For equilibration, have a SHORT tau_t
 - 0.1 to 1 ps?
- For production, have a LONGER tau t
 - 5 or 10 ps?

• Barostats:

- Pressure will frequently oscillate between +/- 1000 atm for a low compressibility fluid. THIS IS NORMAL!
- Generally, keep tau_p about 5x tau_t
- If the simulation crashes because of volume fluctuations, try a somewhat longer (2x-5x). Results will be valid as long as tau_p is < 0.1-1% of the simulation run.

