

An introduction to force fields

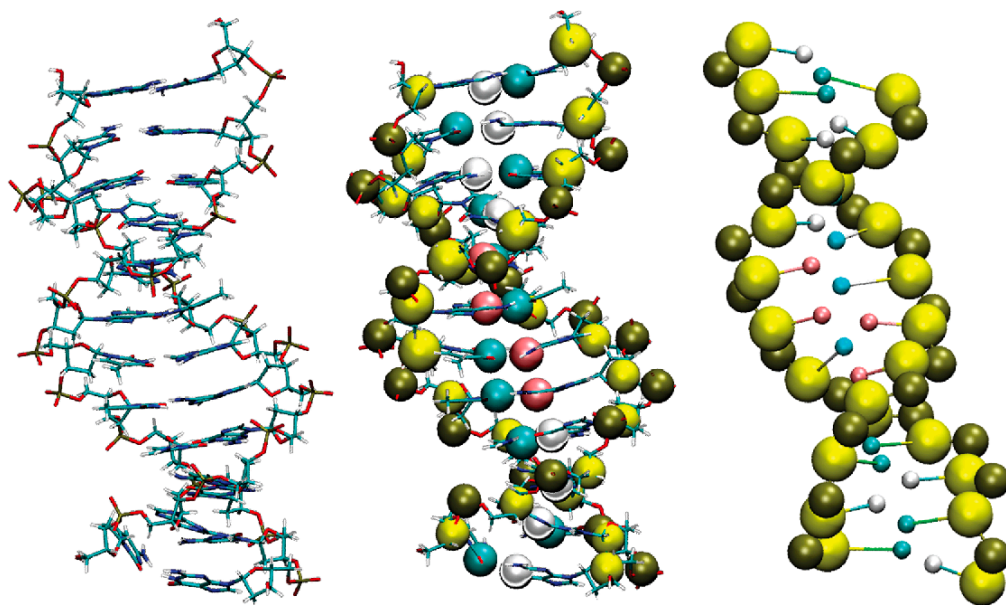
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University of Colorado Boulder
i-CoMSE Summer Workshop
Oklahoma State University, July 2022

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What are we trying to simulate?

- We have some MODEL of our molecule.
 - We have the hypothesis that the physical behavior only depends on SOME of the physics of the molecule
 - We test that hypothesis with a simulation
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- Can be atomistic, quantum mechanical, or coarse grained.
 - Each particle isn't an atom, but a methane particle, or an amino acid.



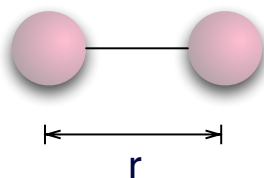
What is a force field?

- A force field is some function $U(x_1, x_2, x_3 \dots x_N)$
 - Input: particle coordinates
 - Output: an energy
 - $-\nabla U$ ($dU/dx_1, dU/dx_2, dU/dx_3 \dots dU/dx_N$) is the **force** on each particle.
 - A force vector is assigned to each set of coordinates.
- Usually, a force field tries to capture the potential energy as a function of atomic nucleus coordinates.
- We AVERAGE OUT the quantum mechanics.
- Let's look at the mathematical and code description of some of the most common force fields (CHARMM/AMBER/GAFF/OPLS/OpenFF)

"Bonded Terms"

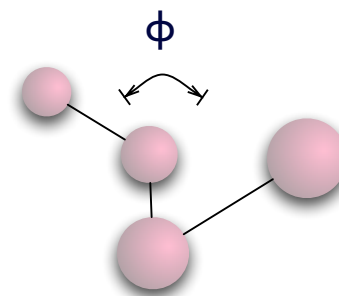
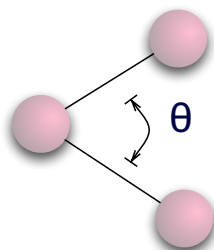
- We replace the electrons in orbitals with simple functional forms that are 10^6 times cheaper (but WAY more approximate).

$$+ \sum_{\text{all bonds}} \frac{1}{2} K_b (r - r_0)^2$$



$$+ \sum_{\text{all torsions}} \sum_n K_\phi [1 - \cos(n\phi + \phi_0)]$$

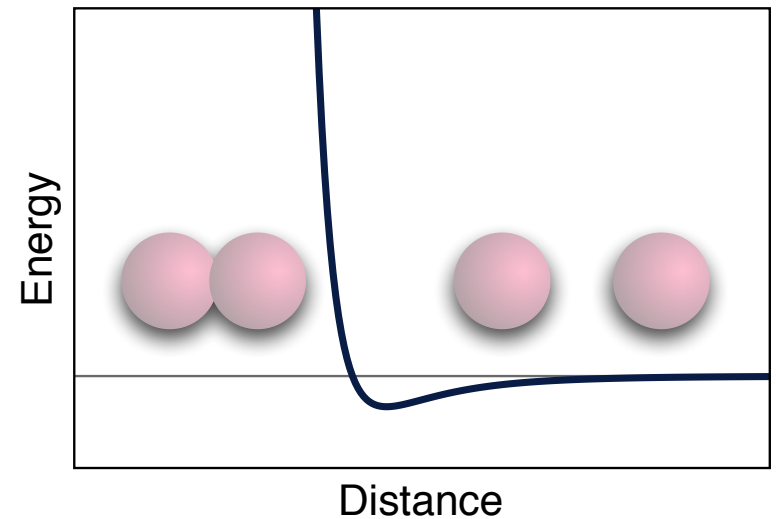
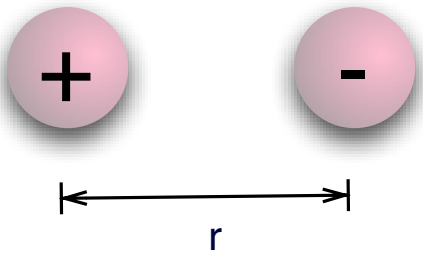
$$+ \sum_{\text{all angles}} \frac{1}{2} K_\theta (\theta - \theta_0)^2$$



"Nonbonded terms"

- We replace the electrostatic / dipole / induced dipole interactions with physically motivated functional forms.

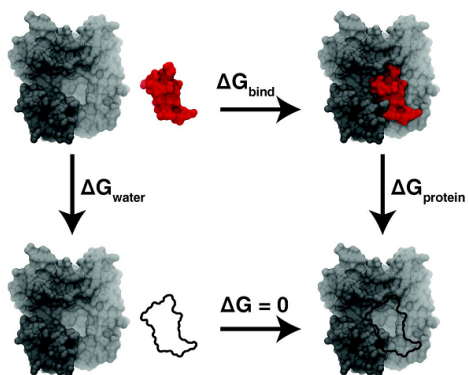
$$U(\mathbf{x}) = \sum_{\text{all pairs}} \frac{q_i q_j}{r} + \sum_{\text{all pairs}} \frac{A_{ij}}{r^{12}} - \frac{B_{ij}}{r^6}$$



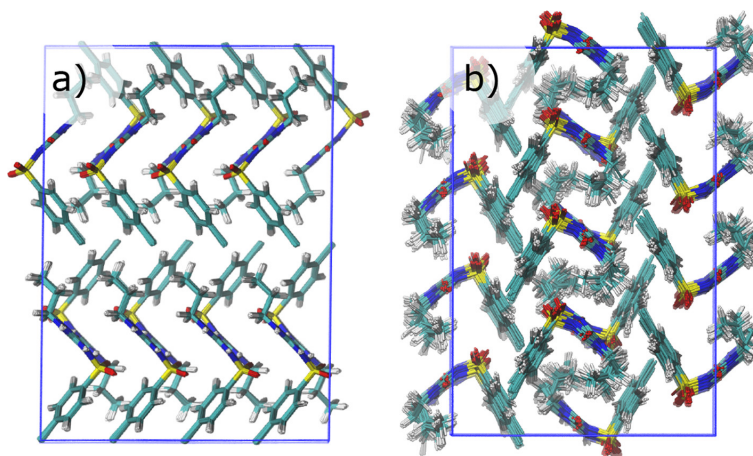
Small molecule force fields are widely-used tools in computational biophysics and soft matter systems

- Surprisingly, all-atom fixed charge force fields strike an appropriate balance between chemical accuracy and computation efficiency for many systems of interest.

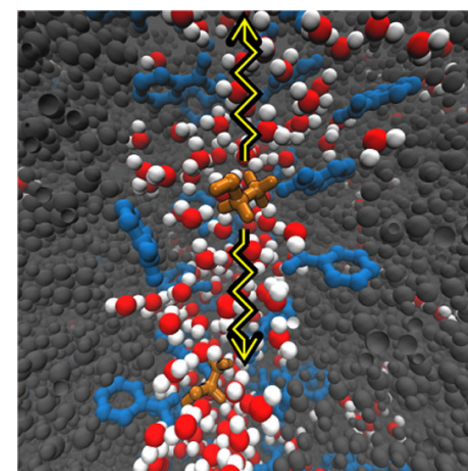
Protein-Ligand Binding



Drug molecule screening



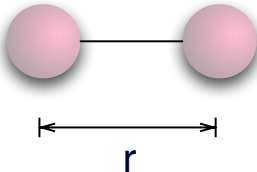
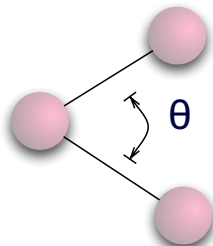
Molecular transport



"Bonded Terms"



- We replace the electrons in orbitals with simple functional forms that are 10^6 times cheaper (but WAY more approximate).

$$+ \sum_{\text{all bonds}} \frac{1}{2} K_b (r - r_0)^2$$

$$+ \sum_{\text{all angles}} \frac{1}{2} K_\theta (\theta - \theta_0)^2$$


The diagram for the bond length term shows two pink spheres connected by a horizontal line. Below the line is a double-headed arrow labeled with the variable r . The diagram for the bond angle term shows three pink spheres arranged in a triangle. The angle between the two lines connecting the bottom-left sphere to the other two is indicated by a curved arrow and labeled with the variable θ .

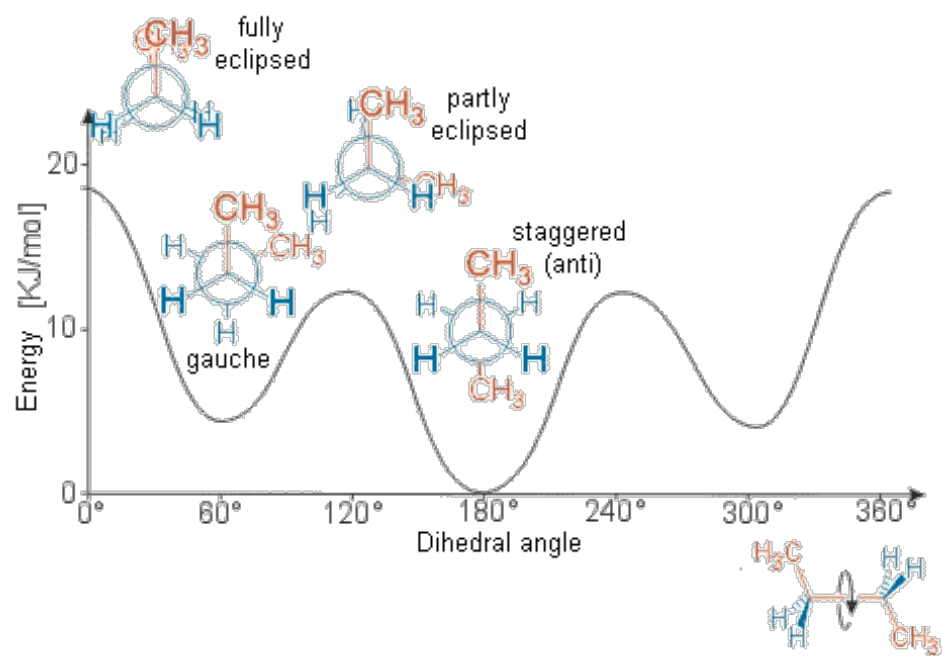
Harmonic bonds and angles
Pretty good approximation to quantum mechanics!



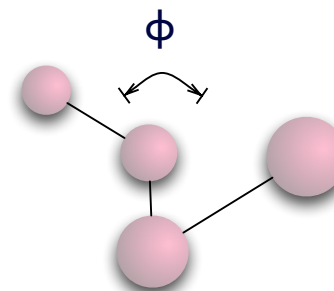
Go To Examples!

Torsions, or the junk bin of force fields

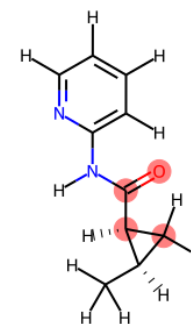
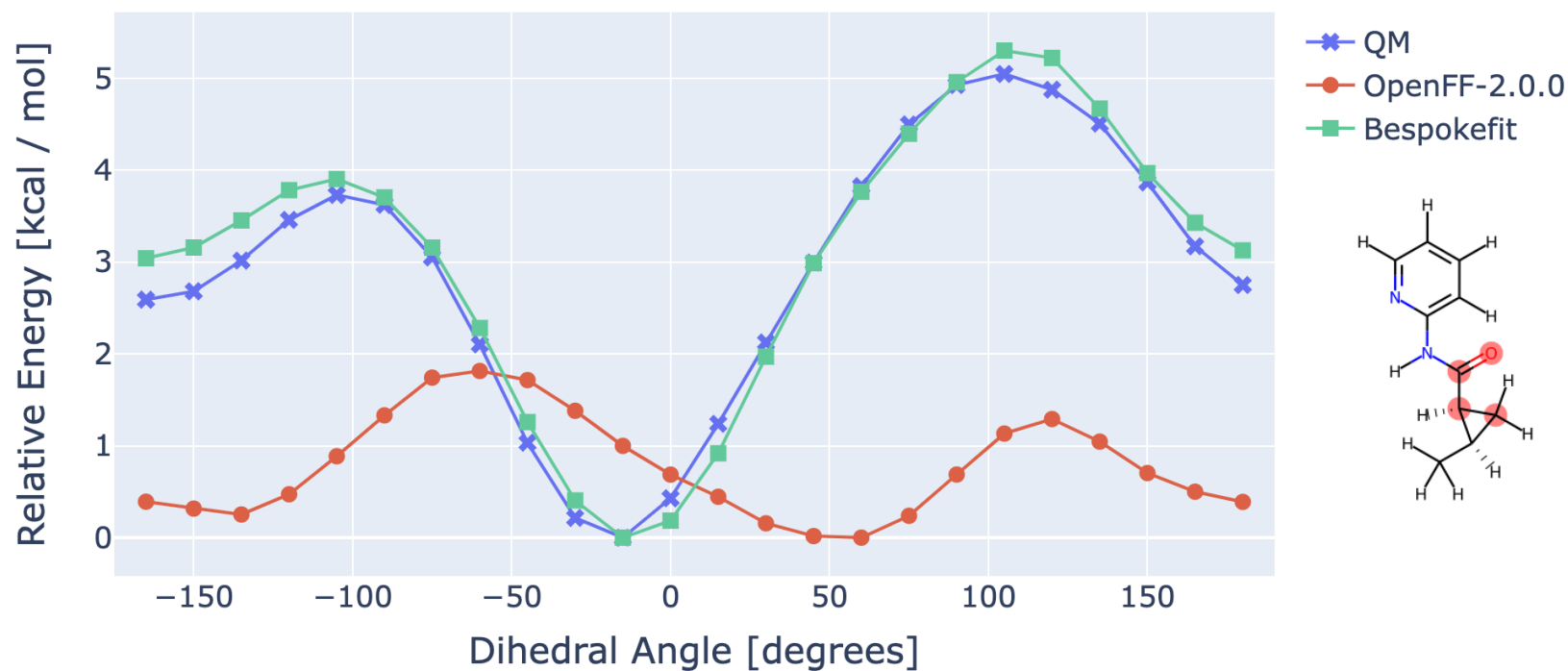
- Replace QM with a sum of periodic trigonometric functions



$$+ \sum_{\text{all torsions}} \sum_n K_\phi [1 - \cos(n\phi + \phi_0)]$$

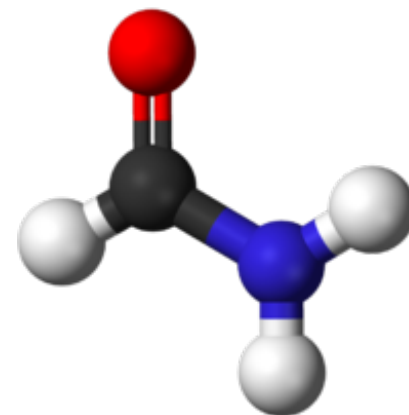
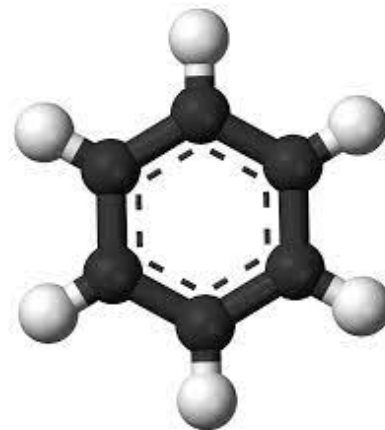
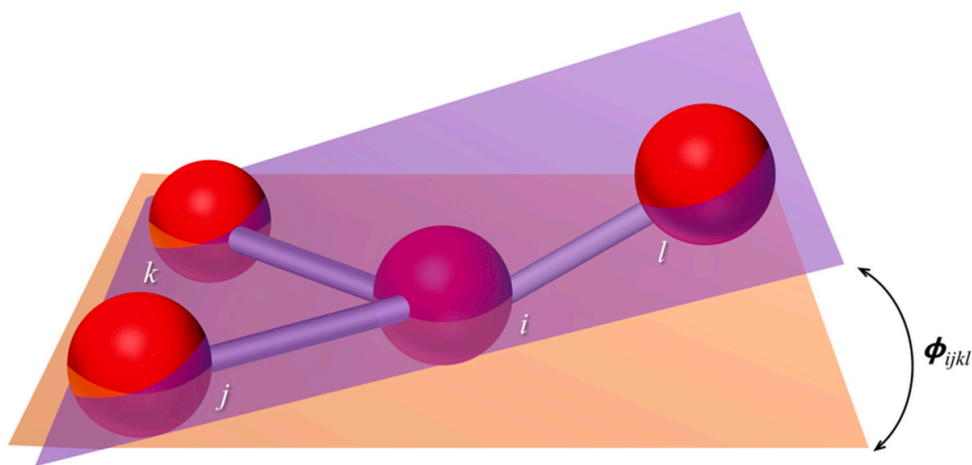


It can be difficult to get these right!



Improper torsions

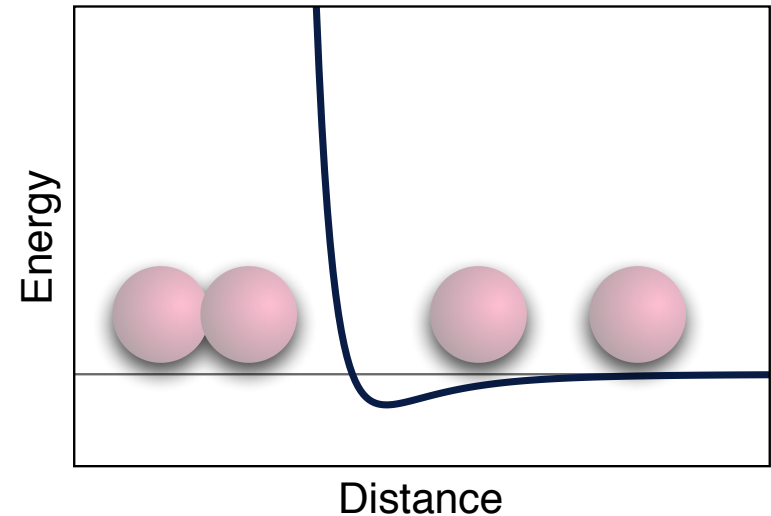
- Ways to keep things with conjugated double bonds planar



Go To Examples!

Lennard-Jones terms

$$U(r_{ij}) = 4\epsilon \left(\left(\frac{\sigma_{ij}}{r_{ij}} \right)^{12} - \left(\frac{\sigma_{ij}}{r_{ij}} \right)^6 \right)$$



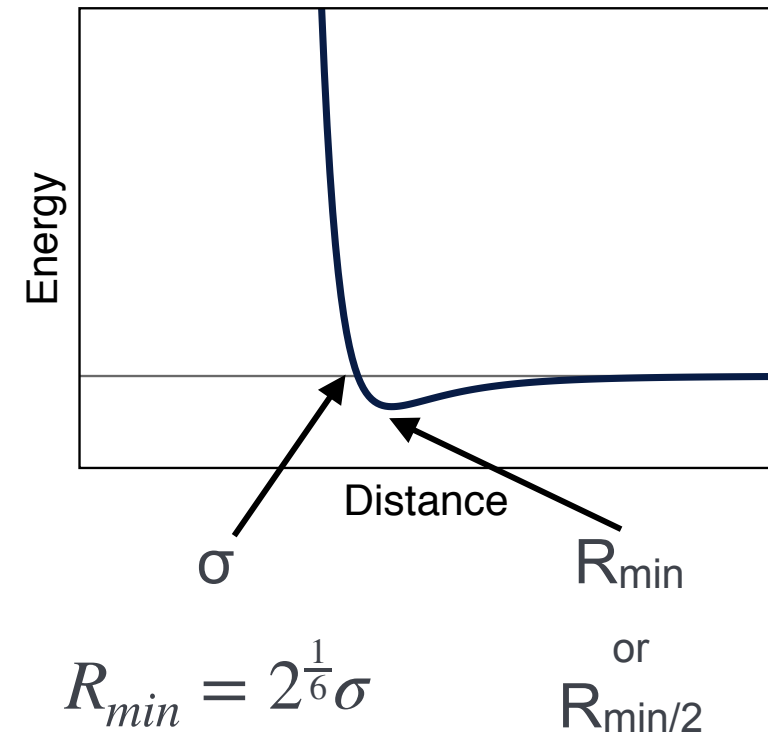
```
[ atomtypes ]
; atom type name type      mass      charge      sigma      epsilon
opls_111      OW      15.99940    0.000      A      3.12171e-01  7.94960e-01
opls_112      HW       1.00800    0.000      A      0.00000e-01  0.00000e-01
```

- This just has sigma and epsilon for O and H. What about the O-H interaction???

Lots of different, equivalent ways to write Lennard-Jones parameters!

$$U(r_{ij}) = 4\epsilon \left(\left(\frac{\sigma_{ij}}{r_{ij}} \right)^{12} - \left(\frac{\sigma_{ij}}{r_{ij}} \right)^6 \right)$$

$$U(r_{ij}) = \frac{A_{ij}}{r^{12}} - \frac{B_{ij}}{r^6}$$



Combining rules

```
[ defaults ]  
; nbfunc      comb-rule      gen-pairs      fudgeLJ  fudgeQQ  
1             3              yes              0.5      0.5
```

Geometric

$$\epsilon_{ij} = \sqrt{\epsilon_i \epsilon_j} \qquad \sigma_{ij} = \sqrt{\sigma_i \sigma_j}$$

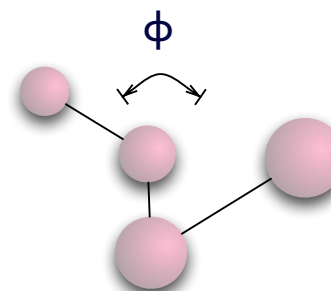
Lorentz-Berthelot

$$\epsilon_{ij} = \sqrt{\epsilon_i \epsilon_j} \qquad \sigma_{ij} = \frac{1}{2} (\sigma_i + \sigma_j)$$

Works amazingly well! (but not perfect)

Exclusions?

- What about atoms separated by no bonds (1,2)?
- What about atoms separated by 1 bond (1,3)?
- What about atoms separated by 2 bonds (1,4)?



```
[ defaults ]  
; nbfunc      comb-rule      gen-pairs      fudgeLJ fudgeQQ  
1             3              yes              0.5      0.5
```


Go To Exercise!

Other terms you could put in:

- Hydrogen bonding terms
- Polarizability: The ability for charge distributions to rearrange.
- Other many body terms (charge penetration)
- Force field that break bonds (ReaxFF)
- Machine learning terms (Ani neural net potentials, others)

- BUT pairwise nonbonded potentials (plus some multibody bonded terms) are **much** faster than general multibody terms for both MD and MC.

- As long as you have a function $U(x_1, x_2, x_3 \dots x_N)$, an energy output with an coordinate input, you can do molecular dynamics (or Monte Carlo!)

What sorts of force fields are out there?

- OPLS-AA / OPLS-UA
 - Small molecules, proteins
 - Evolved out of Bill Jorgensen's work
 - Now run by Schrodinger
 - Biomolecules and small molecule ligands
- TRAPPE
 - Coming mostly from Ilja Siepmann's group
 - Designed for phase equilibrium

Other Force Fields

- CHARMM
 - Evolved from the work in Martin Karplus's group at Harvard over ~50 years.
 - Alex MacKerrell (Maryland) and Benoit Roux (Chicago) are longest contributing developers but MANY more.
 - Proteins, lipids, nucleic acids small molecules
- AMBER
 - Evolved from the work in Peter Kollman's group at UCSF, and his academic descendants.
 - GAFF small molecule force field
 - Proteins and nucleic acids

Other Force Fields

- Open Force Field
 - <http://www.openforcefield.org>
 - New effort, focused on open software as well as open force fields
 - Small molecule based, in the process of adding biopolymers
 - Can build from SMILES or mol2
 - Difference is: it matches molecular parameters in a more general way than foyer



Other Force Fields

- What about polymers?
 - OPLS often used.
 - COMPASS
 - MMFF
- What about metals?
 - Lots of metal specific functional forms.

Force Field Parameterization Tools

- AmberTools/tleap
- CHARMM-GUI/CGENFF
- OpenFF toolkit
- foyer
- pysimm (polymers)
- OpenKIM (metals)

But What Force Field Should I Use?

- The one that allows you to test the hypothesis that you want to

OR

- The one that is validated as an effective screening mechanism to predict the property of interest to other researchers.
- Has the force field I want to use been validated on sufficiently similar systems?
- "All models are wrong. Some models are useful."
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