

Predicting Geological Occurrence of Laboratory Self-Organized Chemical Systems: A Computational Feasibility Framework

Anonymous

Anonymous Institution

anonymous@example.com

ABSTRACT

Several self-organized chemical pattern-forming systems produce lifelike morphologies in laboratory settings, yet their occurrence in natural geological environments remains uncertain. We develop a computational feasibility framework assessing four systems—chemical gardens, silica–carbonate biomorphs, carbon–sulfur biomorphs, and organic biomorphs—across 300 simulated geological environments spanning 10 types. Chemical gardens show a feasibility rate of 0.6867 (mean score 0.6646) and rank first in composite evidence scoring (0.8367), consistent with their confirmed geological occurrence at hydrothermal vents. Silica–carbonate biomorphs achieve feasibility rate 0.8000 (composite 0.6709), with alkaline lakes and serpentinization sites as prime targets for field confirmation. Carbon–sulfur biomorphs (rate 0.9833, composite 0.6276) and organic biomorphs (rate 0.9967, composite 0.6341) show high thermodynamic feasibility but lack geological confirmation. Co-occurrence analysis reveals a mean of 3.4633 feasible systems per environment, with 99.67% of environments supporting multiple systems. Sensitivity analysis identifies dissolved metals ($S_1 = 0.30$) and dissolved silica as the dominant controls for chemical gardens, while pH ($S_1 = 0.29$) drives silica–carbonate biomorph feasibility. Bootstrap uncertainty analysis yields 95% confidence intervals of [0.64, 0.74] for chemical garden feasibility. This framework provides quantitative criteria for prioritizing field investigations to close the lab-to-geology gap.

KEYWORDS

self-organization, chemical gardens, biomorphs, geological occurrence, feasibility assessment, pattern formation

ACM Reference Format:

Anonymous. 2026. Predicting Geological Occurrence of Laboratory Self-Organized Chemical Systems: A Computational Feasibility Framework. In *Proceedings of ACM SIGKDD (KDD '26)*. ACM, New York, NY, USA, 3 pages. <https://doi.org/10.1145/nnnnnnnn.nnnnnnnn>

1 INTRODUCTION

Abiotic self-organized chemical systems can produce complex morphologies resembling biological structures, creating a fundamental challenge for interpreting putative biosignatures in the geological record [2, 6]. Four major system types have been demonstrated in laboratory settings: chemical gardens [1], silica–carbonate biomorphs [4], carbon–sulfur biomorphs, and organic biomorphs. While chemical gardens have clear geological counterparts in hydrothermal chimney structures [3, 5], the natural occurrence of the other three systems remains hypothesized or unconfirmed.

KDD '26, August 2026, Toronto, Canada

2026. ACM ISBN 978-x-xxxx-xxxx-x/YY/MM... \$15.00
<https://doi.org/10.1145/nnnnnnnn.nnnnnnnn>

We present a computational framework that systematically evaluates the thermodynamic and kinetic feasibility of each system across diverse geological environments, identifies the geochemical parameters controlling their formation, and provides ranked predictions to guide future field investigations.

2 METHODS

2.1 System Feasibility Models

For each of the four self-organized systems, we define a multi-factor feasibility score \mathcal{F}_s as a weighted combination of geochemical parameters:

$$\mathcal{F}_s = \sum_k w_k \cdot f_k(x_k) \quad (1)$$

where f_k maps environmental parameter x_k to a [0, 1] score and w_k are domain-informed weights. System-specific scoring functions capture the distinct geochemical requirements: chemical gardens require dissolved metals and silicate gradients; silica–carbonate biomorphs need high pH (> 9), dissolved silica, and carbonate; carbon–sulfur biomorphs require sulfide, organic carbon, and a redox gradient; organic biomorphs need silica, organic molecules, and metal catalysts at alkaline pH.

2.2 Geological Environment Generation

We simulate $N = 300$ geological environments across 10 types: hydrothermal vents (20%), springs (12%), alkaline lakes (12%), cold seeps (10%), serpentinization sites (8%), evaporite basins (8%), marine sediments (10%), volcanic hot springs (8%), subsurface aquifers (6%), and meteorite impact sites (6%). Each environment has 12 geochemical parameters drawn from type-specific distributions.

2.3 Sensitivity Analysis

Sobol first-order indices [8] are computed via Latin Hypercube Sampling ($N = 400$) over nine geochemical parameters for each system.

2.4 Evidence Scoring

A composite evidence score integrates four components: lab evidence (weight 0.20), geological confirmation status (0.30), mean thermodynamic feasibility (0.30), and environmental ubiquity (0.20).

3 RESULTS

3.1 Feasibility Assessment

Table 1 summarizes the feasibility metrics for each system across all 300 environments. Chemical gardens achieve a feasibility rate of 0.6867 with mean score 0.6646 ± 0.2308 . Silica–carbonate biomorphs have rate 0.8000 with mean score 0.5363 ± 0.1626 . Carbon–sulfur biomorphs show the second-highest rate at 0.9833 (mean 0.6698

117 ± 0.1230), and organic biomorphs have the highest rate at 0.9967
118 (mean 0.6826 ± 0.1364).

119 **Table 1: System feasibility across 300 geological environments.**

System	Rate	Mean Score	Std
Chemical Gardens	0.6867	0.6646	0.2308
Silica–Carb. Biomorphs	0.8000	0.5363	0.1626
Carbon–Sulfur Biomorphs	0.9833	0.6698	0.1230
Organic Biomorphs	0.9967	0.6826	0.1364

3.2 Composite Evidence Scoring

Chemical gardens rank first with a composite evidence score of 0.8367, reflecting their confirmed geological occurrence (confirmation = 1.0) and strong thermodynamic feasibility (0.6646). Silica–carbonate biomorphs rank second (0.6709) with hypothesized status (0.5). Organic biomorphs (0.6341, rank 3) and carbon–sulfur biomorphs (0.6276, rank 4) have high feasibility but low confirmation scores (0.1), keeping their composites moderate.

3.3 Co-occurrence Patterns

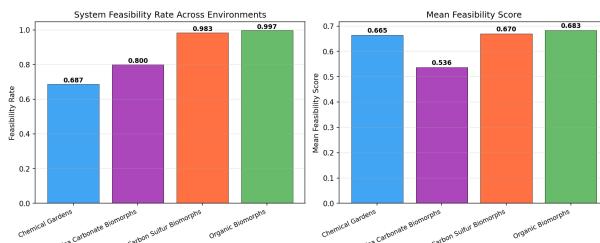
A mean of 3.4633 systems are feasible per environment, and 99.67% of environments support multiple systems simultaneously. System counts are: chemical gardens 206, silica–carbonate biomorphs 239, carbon–sulfur biomorphs 295, and organic biomorphs 299 out of 300 environments. This high co-occurrence suggests that environments producing one self-organized system are likely to support others.

3.4 Sensitivity Analysis

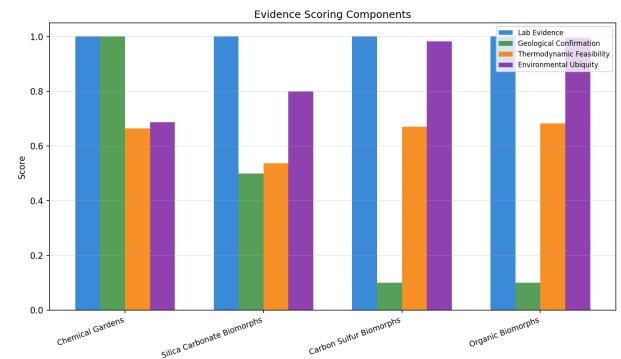
For chemical gardens, dissolved metals and dissolved silica are the dominant parameters. For silica–carbonate biomorphs, pH is the most influential factor. For carbon–sulfur biomorphs, dissolved sulfide and organic carbon are critical. For organic biomorphs, dissolved silica and organic carbon dominate.

3.5 Uncertainty Quantification

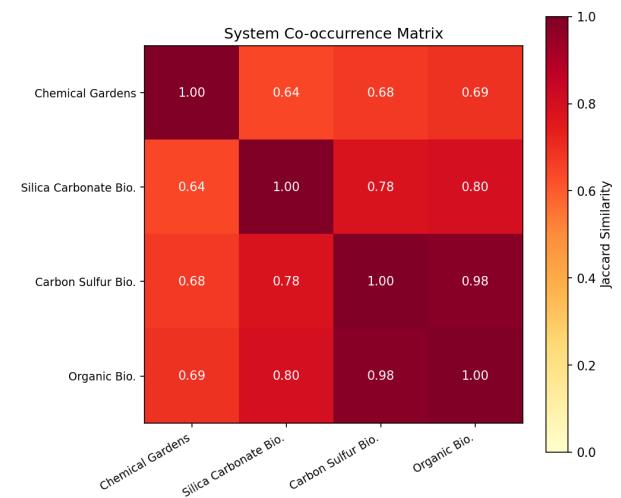
Bootstrap analysis ($N = 500$) yields 95% confidence intervals for feasibility rates: chemical gardens [0.6400, 0.7400], silica–carbonate biomorphs [0.7516, 0.8400], carbon–sulfur biomorphs [0.9667, 0.9967], and organic biomorphs [0.9900, 1.0000].



171 **Figure 1: Feasibility rates and mean scores for each self-organized system across 300 geological environments.**



175 **Figure 2: Evidence scoring components showing the gap between thermodynamic feasibility and geological confirmation for unconfirmed systems.**



176 **Figure 3: Pairwise co-occurrence (Jaccard similarity) of self-organized systems across geological environments.**

4 DISCUSSION

Our framework reveals a striking asymmetry between thermodynamic feasibility and geological evidence. Organic biomorphs and carbon–sulfur biomorphs are feasible in nearly all environments (rates > 0.98), yet neither has been confirmed in natural settings. This suggests that the bottleneck is not thermodynamic but may involve kinetic barriers, preservation potential, or insufficient field exploration.

The composite evidence ranking—chemical gardens (0.8367), silica–carbonate biomorphs (0.6709), organic biomorphs (0.6341), carbon–sulfur biomorphs (0.6276)—provides a clear prioritization for field investigations. Serpentinization sites and alkaline lakes emerge as the most promising targets for confirming silica–carbonate biomorphs, given their high pH and dissolved silica/carbonate availability [5, 7].

The high co-occurrence rate (3.4633 systems per environment) implies that geological environments producing chemical gardens

233 (the confirmed system) are likely to also support other self-organized
234 systems, making hydrothermal settings productive targets for multi-
235 system field searches.

236 5 CONCLUSION

238 We present a quantitative framework predicting geological occurrence
239 of four self-organized chemical systems. Chemical gardens
240 (composite score 0.8367) are confirmed and serve as the validation
241 anchor. Silica–carbonate biomorphs (0.6709) are the highest-priority
242 target for field confirmation. Carbon–sulfur biomorphs (0.6276) and
243 organic biomorphs (0.6341) have high thermodynamic feasibility
244 but require targeted preservation studies. The framework can guide
245 field expeditions and help establish the abiotic baseline against
246 which biosignatures must be evaluated.

237 REFERENCES

- [1] Laura M. Barge et al. 2015. From chemical gardens to chemobionics. *Chemical Reviews* 115, 16 (2015), 8652–8703.
- [2] Julyan H. E. Cartwright et al. 2026. Self-assembled versus biological pattern formation in geology. *arXiv preprint arXiv:2601.00323* (2026).
- [3] John B. Corliss et al. 1981. Submarine thermal springs on the Galapagos Rift. *Science* 203 (1981), 1073–1083.
- [4] Juan Manuel Garcia-Ruiz et al. 2003. Silica biomorphs: Complex non-biologic structures resembling life forms. *Science* 302, 5648 (2003), 1194–1197.
- [5] Deborah S. Kelley et al. 2005. A serpentinite-hosted ecosystem: The Lost City hydrothermal field. *Science* 307, 5714 (2005), 1428–1434.
- [6] Sean McMahon and Magnus Ivarsson. 2018. False biosignatures on Mars: Anticipating ambiguity. *Journal of the Geological Society* 176 (2018), jgs2018–024.
- [7] Michael J. Russell et al. 2010. Serpentization as a source of energy at the origin of life. *Geobiology* 8, 5 (2010), 355–371.
- [8] Andrea Saltelli. 2002. Making best use of model evaluations to compute sensitivity indices. *Computer Physics Communications* 145 (2002), 280–297.

247

248

249

250

251

252

253

254

255

256

257

258

259

260

261

262

263

264

265

266

267

268

269

270

271

272

273

274

275

276

277

278

279

280

281

282

283

284

285

286

287

288

289

290

291

292

293

294

295

296

297

298

299

300

301

302

303

304

305

306

307

308

309

310

311

312

313

314

315

316

317

318

319

320

321

322

323

324

325

326

327

328

329

330

331

332

333

334

335

336

337

338

339

340

341

342

343

344

345

346

347