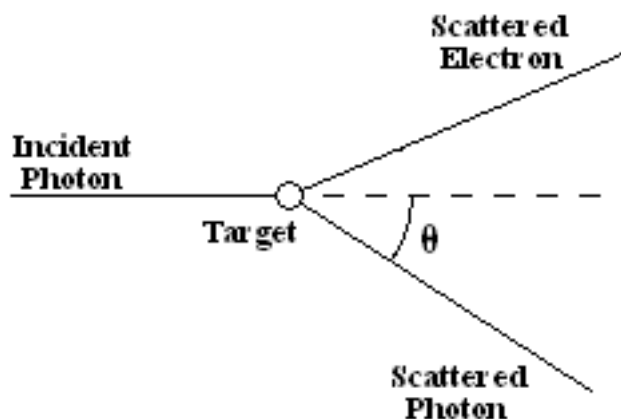


## Compton Scattering

Physicist Arthur Holly Compton won the 1927 Nobel Prize in Physics for observing that a light “wave”, after colliding with a free electron, gained energy. This observation could not be explained by modeling light as a wave, but instead by modeling light as a particle that we call a photon. Thus, Compton’s observation was fundamental in suggesting wave-particle duality. Consider the setup below, where a photon of wavelength  $\lambda$  hits an electron (target) of mass  $m_e$ , and where the scattered photon heads off at an angle of  $\theta$ . Derive a formula for the wavelength of the scattered photon in terms of  $\lambda$ ,  $\theta$ ,  $m_e$ , and any fundamental constants you think you need.



Some possibly useful information:

$$E = \frac{hc}{\lambda}$$

where  $E$  is energy,  $h = 6.63 \times 10^{-34} J \cdot sec$  is Planck’s constant,  $c$  is the speed of light, and  $\lambda$  is wavelength.

$$E^2 = c^2 p^2 + m^2 c^4$$

where  $p$  is [the magnitude of] momentum and  $m$  is mass. By the way, for particles that are moving slowly compared to the speed of light, the usual  $KE = \frac{1}{2}mv^2$  is a good estimate.