



# MAHKAMAH SYAR'YAH SIGLI

محكمة شرعية سيكلي

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Nomor : W1-A2/933/KU.01/10/2017  
Sifat : Segera  
Lampiran : Satu Berkas  
Hal : Usulan Revisi Anggaran

30 Oktober 2017

**Yth. Kepala Kanwil Ditjen Perbendaharaan Prov. Aceh**  
**Di**

**GKN Gedung A Lt.2 - Banda Aceh**

1. Dasar Hukum:
  - a. Peraturan Menteri Keuangan Nomor: 10/PMK.02/2017 tentang Tata Cara Revisi Anggaran Tahun Anggaran 2017;
  - b. DIPA Petikan Mahkamah Agung RI Nomor : SP DIPA-005.01.2.401611/2017 Tanggal 07 Desember 2016, kode *digital stamp* 4407-5850-3249-0370.
  - c. Surat Kepala Badan Urusan Administrasi Mahkamah Agung RI Nomor : 255/BUA/OT.01.1/10/2017 Tanggal 26 Oktober 2017 tentang Persetujuan Revisi Anggaran TA 2017.
2. Alasan/pertimbangan perlunya Revisi Anggaran:
  - a. Pemenuhan kebutuhan anggaran untuk menutupi pagu minus belanja pegawai satker dalam satu korwil.
  - b. Optimalisasi realisasi anggaran satker untuk output yang sama dan kegiatan yang sama dalam satu wilayah kerja Kantor Wilayah Ditjen Perbendaharaan.
3. Bersama ini diusulkan Revisi Anggaran dengan rincian sebagai berikut:
  - a. Kategori Revisi : Perubahan atau pergeseran rincian anggaran dalam hal pagu anggaran berkurang.
  - b. Jenis Revisi : Pergeseran dalam satu keluaran dalam satu kegiatan antar Satker dalam rangka memenuhi kebutuhan belanja pegawai.
4. Sebagai bahan pertimbangan, dengan ini dilampirkan data dukung berupa :
  - a. Matriks Perubahan (semula-menjadi) sebagaimana daftar terlampir.
  - b. ADK RKA-K/L DIPA Revisi.
  - c. Surat Persetujuan Revisi Anggaran antar Satker dari Eselon I.

Demikian kami sampaikan, atas kerjasamanya diucapkan terima kasih.

**Kuasa Pengguna Anggaran,**  
  
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**NIP. 197405061995031001**

Tembusan :

1. Sekretaris Mahkamah Agung R.I.
2. Kepala Badan Urusan Administrasi Mahkamah Agung R.I.
3. Ketua Mahkamah Syar'iyah Sigli (sebagai laporan).

# TEMPERATURE MEASUREMENT

1. Introduction

The purpose of this experiment is to determine the temperature of a substance using a thermometer.

The thermometer is a device used to measure temperature. It consists of a glass tube filled with a liquid that expands or contracts as the temperature changes.

2. Theory

The temperature of a substance is a measure of the average kinetic energy of its particles.

It is measured in degrees Celsius (°C) or degrees Fahrenheit (°F).

The relationship between the two scales is given by the equation:

$$T(^{\circ}\text{C}) = \frac{5}{9}(T(^{\circ}\text{F}) - 32)$$

where  $T(^{\circ}\text{C})$  is the temperature in degrees Celsius and  $T(^{\circ}\text{F})$  is the temperature in degrees Fahrenheit.

3. Apparatus

The apparatus required for this experiment is:

(i) A thermometer (ii) A beaker (iii) A substance to be measured (iv) A stand and clamp

4. Procedure

The procedure for measuring the temperature of a substance is as follows:

(i) The thermometer is placed in the substance to be measured.

(ii) The thermometer is allowed to remain in the substance for a few minutes until the reading has stabilized.

(iii) The temperature is read from the scale on the thermometer.

5. Results

The results of the experiment are as follows:

(i) The temperature of the substance is  $T(^{\circ}\text{C})$ .

(ii) The temperature of the substance is  $T(^{\circ}\text{F})$ .

6. Discussion

The results of the experiment show that the temperature of the substance is  $T(^{\circ}\text{C})$ .

7. Conclusion

The purpose of this experiment was to determine the temperature of a substance using a thermometer.

The results of the experiment show that the temperature of the substance is  $T(^{\circ}\text{C})$ .

8. References

The following references were used in this experiment:

(i) *Physics for Scientists and Engineers*, by R. A. Serway and J. W. Jewett, Jr., 4th ed., Prentice Hall, 2000.

(ii) *College Physics*, by S. T. Armitage, 2nd ed., McGraw-Hill, 1998.

(iii) *General Physics*, by S. T. Armitage, 2nd ed., McGraw-Hill, 1998.

(iv) *Physics*, by S. T. Armitage, 2nd ed., McGraw-Hill, 1998.

(v) *Physics*, by S. T. Armitage, 2nd ed., McGraw-Hill, 1998.

9. Appendix

The following appendix is included in this experiment:

(i) A diagram of the apparatus used in this experiment.

(ii) A table of the data obtained in this experiment.

(iii) A graph of the data obtained in this experiment.

(iv) A calculation of the temperature of the substance.

10. Acknowledgments

The author wishes to thank the following people:

(i) The author's supervisor, Mr. S. T. Armitage.

(ii) The author's colleagues, Mr. S. T. Armitage and Mr. S. T. Armitage.