Submillimeter Microwave Spectrum and Spectroscopic Constants of the OCS Molecule

Isotopic Species ¹⁶O¹²C³³S and ¹⁸O¹²C³²S

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Microwave spectra of $^{16}O^{12}C^{33}S$ and $^{18}O^{12}C^{32}S$ in their natural isotopic abundances were investigated in the range 280-510 GHz. The effective rotational and centrifugal constants have been determined for all observed vibrational states. An analysis of the rotational spectrum taking account of l- and Fermi-resonances has been performed within each isotopic species.

INTRODUCTION

In our first paper devoted to the investigation of the OCS submillimeter spectrum the less abundant isotopic species of the molecule (natural abundance $i_c < 0.05\%$) were considered (1). The present paper describes the results of studies of the microwave spectrum of OCS isotopic species with abundances $0.05\% < i_c < 1\%$. This interval includes two isotopic species $^{16}\text{O}^{12}\text{C}^{33}\text{S}$ and $^{18}\text{O}^{12}\text{C}^{32}\text{S}$. Their spectra were observed at room temperature in the range 280-510 GHz by a submillimeter microwave spectrometer RAD (2). The experimental procedure including correction of the influence of the wings of neighbouring lines on the measured absorption frequencies ν_{meas} is exactly the same as in Ref. (1). The absorption line frequencies obtained after correction and denoted ν_{exp} are given in Table I together with the correction values $\delta = \nu_{\text{meas}} - \nu_{\text{exp}}$.

PROCESSING OF EXPERIMENTAL DATA

Experimental data for the isotopic species ¹⁸O¹²C³²S obtained in this investigation were processed together with lower frequency data from Ref. (3). Low-frequency data were not used for the isotopic species ¹⁶O¹²C³³S since we did not take into account the quadrupole splitting observed in the low-frequency region. The processing was performed in two steps.

1. The rotational molecular spectrum for each observed vibrational state was described within the model

$$F(J) = B_{\text{eff}}[J(J+1) - l^2] - D_{\text{eff}}[J(J+1) - l^2]^2. \tag{1}$$

All designations are the same as in Ref. (1). The values obtained for $B_{\rm eff}$, $D_{\rm eff}$, and their correlation coefficient R_{BD} as well as the upper limit for the sextic

TABLE I Experimental Frequencies of Rotational Transitions of the Isotopic Species

16O12C33 and 18O12C32

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7 - 3 5 53 21- 15	 		158,555(13) 269,465(26) 233,765(13)	311 533 463	55 - 24 04 - 41	0 004
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ol 9	-	,	954.925(22)	479 502	44 - 41	- 001
2 26- 6 65 \$ 9 £	Ç~	•	754.570(20) 131.63(10) 144.130(60) 144.130(60)	363 155 155 156 156 156 156 156 156 156 156	25 - 25 25 - 25 26 - 25 27 - 25 28 - 25 29 - 25 20	a ⁰ 01
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76 6 065- 1	- - - - -	•	350.073(25) 943.690(40) 224.93(16)	99C 93L 96	3 - 50	

Note. *—Exp. frequency = Meas. frequency – δ correction, a—Ref. (3), b—the present paper.

centrifugal constant \mathbf{H}_{eff} , under which the influence of this constant may be neglected, are presented in Table II. The method of obtaining the upper limit for the neglected constant \mathbf{H}_{eff} is analogous to that used in Refs. (1, 4). The differences vexp- v_{exp} - v_{cale} obtained are given in the first difference column of Table I.

TABLE II

Effective Rotational and Centrifugal Constants for Each Observed Vibrational State of the Isotopic Species $^{16}O^{12}C^{33}S$ and $^{18}O^{12}C^{32}S$

Vibration state	B _{eff} . MHz	D _{eff} . KHz	R_{BD}	Condition on		
	16 _C					
00 ⁰ 0	6 004.916 25(14)	1.270 726(54)	0.939	£1.4 10 ⁻⁵		
01 ^{1c} 0	6 012.263 79(19)	1.289 090(72)	0.987	≉2•0 10 ⁻⁵		
01 ^{1d} 0	6 018,473 72(23)	1.293 132(91)	0.987	≉2•5 10 ⁵		
10 ⁰ 0	5 987.008 85(40)	1.295 83(16)	0.932	≤4.3 10 ⁻⁵		
02 ⁰ 0	6 023.469 25(83)	1.062 19(34)	0.940	≤9.4 10 ⁻⁵		
02 ²⁰ 0	6 025.724 39(80)	1.543 23(33)	0.940	€8.7 10 ⁻⁵		
02 ^{2d} 0	6 025.718 54(75)	1.310 84(31)	0.937	€8.7 10 ⁻⁵		
	18 ₀	12 _C 32 _S				
00 ⁰ 0	5 704.857 19(15)	1.132 977(53)	0.943	≤1.3 10 ⁻⁵		
01 ¹⁰ 0	5 712.160 22(30)	1.149 07(11)	0.926	≤2.8 10 ⁻⁵		
01 ^{1d} 0	5 717.813 17(36)	1.152 64(13)	0.940	≼3.4 10 ⁻⁵		
10 ⁰ 0	5 688,650 0(31)	1.152 9(16)	0.948	≤9.8 10 ⁻⁴		
02 ⁰ 0	5 722.920 2(39)	0.978 0(23)	0.847	≤1.4 10 ⁻³		
02 ^{2c} 0	5 725.029 5(19)	1.355 3(16)	0.668	£9.6 10 ⁻⁴		
02 ^{2d} 0	5 725.027 1(15)	1.172 1(15)	0.625	≤8.9 10 ⁻⁴		

2. Within each isotopic molecular species an analysis of the rotational spectrum in all observed vibrational states was performed taking into account l- and Fermiresonances. The theoretical model applied for this purpose is basically analogous to that used in Ref. (5). Slight corrections were made in agreement with Ref. (1). Besides microwave data IR data (6) on the energy differences between Fermiresonance connected states $|02^{0}0, J=0\rangle$ and $|10^{0}0, J=0\rangle$ were used in this stage of processing. The algorithm of the search for spectroscopic constants is given

TABLE III

Spectroscopic Constants of the Isotopic Species ¹⁶O¹²C³³S

							160	12 ₀ 33 _S				
Во	6	004.916	31(14)	MHz	1							
4		20.150	4(10)	MHz	0.044	1						
42		-10.604	32(50)	MHz	0.361	0.101	1					
¥ ₂₂		-50.67(14)	KHz	0.128	0.184	0.933	1				
D ,		1.270	750(53)	KHz	0.937	0.029	0.294	0.070	1			
β1		-8.14(44)	Ηz	-0.012	0.815	-0.285	-0.242	-0.013	1		
P 2		-20.306	(72)	Hz	0.634	-0.160	0.482	0.163	0.677	-0.249	1	
g ₁₁		10.126	2(54779)	GHz	0.129	-0.711	0.345	0.275	0.138	-0.925	0.342	1
-11 11	5	477.89(4243)	GHz	-0.128	0.711	-0.342	-0.271	-0.137	0.922	-0.347	-0.9975
, 40		3.104	93(15)	MHz	0.011	0.076	-0.049	-0.009	0.011	0.081	-0.126	-0.082
		-2.011	(58)	Ηz	-0.013	-0.086	0.041	-0.005	-0.014	-0.093	0.149	0.095
q _J (122 (100)	1	437.607	(80502)	GHz	0.129	-0.713	0.343	0.273	0.138	-0.925	0.342	0.99998
$\lambda_{\rm J}$		-3.916	94(82405)	MHz	0.128	-0.715	0.341	0.271	0.138	-0.925	0.342	0.99997
						Δ1	1					
						q,	0.0	B2 ·	1			
						q.	-0.0	94 -0.9	941	1		
						q _J K(100	-0.9	972 -0.0	082 0.0	95	1	
						$\lambda_{\rm J}^{(100)}$	-0.9	978 -0.0	082 0.0	095 0.	99996	1

TABLE IV										
Spectroscopic Constants of the Isotopic Species ¹⁸ O ¹² C ³² S										

Во		18 ₀ 12 _c 32 _s									
	5	704.857 16(15)	MHz	1							
4		18.320 9(49)	\mathtt{MHz}	0.032	1						
ιģ		-10.265 7(17)	\mathtt{MHz}	0.199	0.215	1					
¥22		-45.35(53)	KHz	0.119	0.228	0.989					
D _o		1.132 965(53)	\mathtt{KHz}	0.943	0.030	0.185	0.110				
P ₁		-5.1(27)	$_{\mathrm{Hz}}$	-0.003	0.725	-0.147	-0.148	-0.003			
)		-17.946(98)	$_{\rm Hz}$	0.501	0.019	0.466	0.339	0.532	-0.063		
ع. ع		12.134(44351)	\mathtt{GHz}	0.026	-0,427	0.193	0.194	0.028	-0.789	0.090	1
۵,	5	532.627(343415)	\mathtt{GHz}	-0.026	0.429	-0.191	-0.192	-0.028	0.789	-0.090	-0.99998
q,		2.826 49(23)	MHz	0.001	0.000	-0.088	-0.088	0.001	0.011	-0.127	-0.013
		-1.790(84)	Hz	-0.001	0.000	0.072	0.055	-0.001	-0.009	0.104	0.011
q _J K(100)	1	505.884(630851)	\mathtt{GHz}	0.026	-0.430	0.191	0.192	0.028	-0.789	0.090	0.99998
$\lambda_{\rm J}^{(100)}$		-3.265 06(576480)	\mathtt{MHz}	0.026	- 0.431	0.190	0.191	0.028	-0.789	0.090	0.9999
					A 1	1					
					q.	0.0	13	1			
					q _J к (122	-0.0	11 -	0.933	1		
					K 122	-0.9	99995 -	0.013	0.011	1	
					$\lambda_{\rm J}^{(10)}$	-0.99	99994 -	0.013	0.011	0.999998	3 1

in Ref. (2). The spectroscopic constants obtained as a result of processing together with the correlation coefficient matrix are presented in Tables III and IV for the isotopic species $^{16}\mathrm{O}^{12}\mathrm{C}^{33}\mathrm{S}$ and $^{18}\mathrm{O}^{12}\mathrm{C}^{32}\mathrm{S}$, respectively. The designations of constants are the same as in Ref. (5) (taking account of the correction performed in Ref. (1)) with the exception of the constants α_1^* and α_2^* which in the present paper are determined as $\alpha_1^* = \alpha_1 - 2\gamma_{11} - \gamma_{13}$, $\alpha_2^* = \alpha_2 - \gamma_{13}$. The differences $\nu_{\rm exp}$ - $\nu_{\rm calc}$ obtained here are presented in the second difference column of Table I.

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