A Simultaneous Analysis of the Microwave, Submillimeterwave, and Infrared Transitions between the Ground and ν₂ Inversion–Rotation Levels of ¹⁵NH₃

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The submillimeterwave spectra of the pure inversion and inversion-rotation transitions in the ν_2 excited state (79 transitions) and the diode laser spectra of the ν_2 band (83 transitions) of $^{15}\text{NH}_3$ have been measured. A simultaneous least squares analysis has been carried out of these data together with previously published wavenumbers of the pure inversion transitions and inversion-rotation transitions in the ground state measured by the microwave and Fourier spectroscopy, and the ν_2 band transition frequencies obtained by the infrared-microwave two-photon technique. A theory of the $\Delta k = \pm 3n$ interactions in the ground and ν_2 excited states of ammonia (Š. Urban, V. Špirko, D. Papoušek, J. Kauppinen, S. P. Belov, L. I. Gershtein, and A. F. Krupnov, J. Mol. Spectrosc. 88, 274–282 (1981)) has been used in the analysis. The "smoothed" values of the ν_2 band wavenumbers can be used for calibration purposes with better than 1×10^{-3} cm⁻¹ precision.

I. INTRODUCTION

In two previous papers (1, 2), we have described a simultaneous analysis of the microwave, submillimeterwave, Fourier transform infrared, standard diode laser, diode laser heterodyne, and infrared-microwave two-photon measurements of the transition frequencies between the ground and ν_2 inversion-rotation energy levels of ¹⁴NH₃. One of the main results of this work was to establish a calibration standard for the 10- μ m region (ν_2 band of ¹⁴NH₃) with a precision better than 1 × 10⁻⁴ cm⁻¹ (2).

An extension of this work to ¹⁵NH₃ is of importance for theory as well as applied spectroscopy. For example, precise experimental data on the transition frequencies of different isotopomers of the same molecule are important for the determination of a complete anharmonic potential function of the molecule (3). Furthermore, high and very high resolution infrared spectroscopy, especially diode laser spectroscopy, needs calibration standards of precision compatible with the resolution of the individual experimental techniques.

As for the precise transition frequencies for $^{15}NH_3$ which could be used in a simultaneous analysis completely analogous to that described previously for $^{14}NH_3$ (1, 4), there are excellent microwave data on the ground state pure inversion transitions (5) of $^{15}NH_3$ up to J=10 and a precise ground state inversion-rotation transition frequency $J=1\leftarrow0$, K=0 measured in the submillimeterwave region (6). The ground state inversion-rotation transition wavenumbers have been measured in the far-infrared region with a Fourier transform spectrometer (7). Although the resolution was about 0.03 cm⁻¹, these data are useful for determination of the ground state centrifugal distortion constants of $^{15}NH_3$ because they extend up to J=13.

The ν_2 band transition frequencies of ¹⁵NH₃ measured with a precision better than 1×10^{-3} cm⁻¹ are the infrared-microwave two-photon data (8-10), the few laser Stark data indicated in Ref. (11), and the highly precise data obtained by the diode laser heterodyne technique (12, 13).

We have extended these data by measuring 83 ν_2 band transition wavenumbers of ¹⁵NH₃ with a diode laser spectrometer, using ¹⁴NH₃ ν_2 band wavenumbers (2) as a calibration standard.

Furthermore, we have measured for the first time 76 pure inversion transition frequencies in the ν_2 state of ¹⁵NH₃ and 3 inversion-rotation transition frequencies. These transitions appear in the submillimeterwave region; we have also involved into our fit the previously published (14) millimeterwave transition frequency $J = 2 \leftarrow 1$, K = 1 and the submillimeterwave transition frequency (15) $J = 1 \leftarrow 0$, K = 0.

II. EXPERIMENTAL DETAILS

The high-sensitivity submillimeterwave spectrometer RAD (16, 17) at the Institute for Applied Physics of the Academy of Sciences, USSR, in Gorkii has been used (cf. Fig. 1). Besides measurements carried out with the microwave accuracy (rms error about ± 0.00003 cm⁻¹), we measured transition frequencies in the ν_2 state of $^{14}NH_3$ with the accuracy of the standard high resolution infrared spectroscopy (rms error about ± 0.001 cm⁻¹) (Table I). In those measurements, the $^{15}NH_3$ spectrum was recorded simultaneously with the SO₂ spectrum, and the $^{15}NH_3$ frequencies have been determined by interpolating between the SO₂ frequencies. The pressure of the ammonia sample ($^{75}MH_3$) in the acoustic cell was about 130 Pa (1 Torr); the experimental uncertainties of the measured frequencies are given in Table I.

The diode laser spectra with the Doppler limited resolution have been measured with a spectrometer described in Refs. (18, 19). The ammonia sample containing 95% $^{15}NH_3$ has been measured in a 1-m cell at the pressure $P \le 65$ Pa (0.5 Torr). The $^{15}NH_3$ lines have been calibrated by means of the $^{14}NH_3$ ν_2 band wavenumbers (2). The $^{14}NH_3$ lines have been recorded with sufficient intensity either directly (cf. Fig. 2) or by measuring the $^{15}NH_3$ spectra simultaneously with the 0.6-m cell containing 0.3 Torr of pure $^{14}NH_3$ (Table II).

III. RESULTS AND DISCUSSION

A simultaneous least squares analysis of all the $^{15}NH_3$ transition frequencies mentioned in this paper has been done as described in detail in our previous papers (1, 2, 4). This means that the only vibrational-rotational interaction which has been

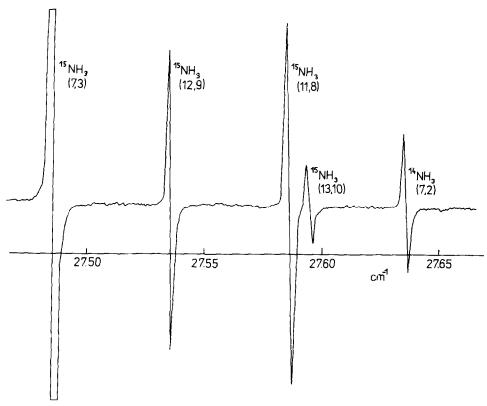


Fig. 1. Part of the submillimeterwave spectrum of ¹⁵NH₃.

considered explicitly was the $\Delta k = \pm 3n$ interaction; all other vibrational-rotational interactions have been "absorbed" into the effective spectroscopic parameters including the octic centrifugal distortion constants (Table III).

The values of the transition wavenumbers calculated from the molecular parameters in Table III are compared with the experimental data in Tables I and II. In the last column of Tables I and II, the intensities of the corresponding lines are given which were calculated as described in Ref. (1).

It is obvious from Tables I and II that we arrived at a quantitative description of the experimental data. The "smoothed" values of the transition wavenumbers of the ν_2 band have a precision estimated to be better than 1×10^{-3} cm⁻¹ for all J' < 10 in the Q-branch lines and for all J' < 8 in the R and P branches. It should be emphasized that the calculated values of the ν_2 band P- and R-branch transition wavenumbers in Table II represent basically a transfer of the experimental information on the ground state, ν_2 state, and Q-branch transition wavenumbers to the $\Delta J = \mp 1$ ν_2 band transitions. Therefore, the precision of the calculated P and R wavenumbers in Table II is determined by the extent of the experimental data available for the aR branch transitions and by the precision of the measured ground state inversion-rotation transition wavenumbers (7).

TABLE I

Pure Inversion and Inversion-Rotation Transition Wavenumbers (cm⁻¹) and Intensities (cm⁻² atm⁻¹) of ¹⁵NH, in the ¹² State*

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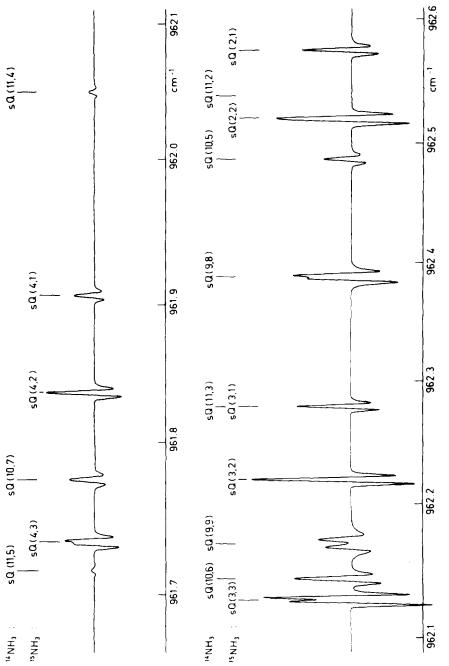


Fig. 2. The diode laser spectrum of the ν_2 band lines of ¹⁵NH₃ and ¹⁴NH₃.

IABLE II

		Tran	Transition Wavenumbers (cm ⁻¹) and Intensities (cm ⁻² atm ⁻¹) in the ν_2 Band of $^{15}\mathrm{NH_3}^{8}$	шреп	cm)	⁻¹) and Inten	sities (Ē	-2 atm ⁻¹) in	the ν_2 Band	d of 15NH	<u>ء</u>	
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4	7	883.10528		;		0.216+01	2	0-	759.34646	*****	1		C.55E-C1
.	n: e	863,00842		:		0.18E+01	11	0	744.69964		ļ		0.47E-01
	n c			1		7.362.01	Ξ:	¢	744.62706		: :		10-342-01
٠ ٧	•					0.166401	Ξ;	~ .	19614.44/	1			10-245-01
	• ~	862.97325		i		0.156401	₹ 5	n .	74.2 54.000				0.276-01
. 7	m	862.79112		l l		0.25E+C1	::		742.92298		. }		0.296-01
S.	4		1	}		0.815+00	=	ۍ د	742,17463		1		C. £15-01
Ţ.	c		1	ł		.0	Ξ	~	741.31988		1		0.33E-C1
9		843.09567		ì		0.116+01	Ξ	Œ.	740,38038	1	{		0.34E-C1
c.	N I	642.96541		1		C+11E+C1		D.	735,38749	1	ļ		C. 646-C1
۰ ۵	· ·	1824/ 248	:	:		0.195.01	11	0	738, 38642	1	1		0.24E-01
0 4	* 4	23221 270	1	1		0.825.00	12	0	725.49810)		}		0.0
0 1	n c	10761 246				0.036+00	12	۰, ۱	725.42136		; ;		0 - ECE - 02
7	-		1	ļ		0.445.00	7.	u r	724 20455				126-02
7	. ~	823.01657		}		0. £3£ + CO		n 4	724.26250				0.536-02
2	3	622,77434		ì		0.12E+01	12	٠.	723,58492		{		0.10E-01
7	*		111111111111111111111111111111111111111	1		0.575+00	2	۵	722.77678		1		0.226-61
7	4			į		0.45E+00	12	~	721.83220	1	1		0.12E-C1
Γ~ -	Φ.			;		0.04340.0	12	œ	720.78663	1	ļ		0.136-01
• •	۰.	~		1		0.		o	719.65963	1	1		0.28E-C1
D 0		803.32341		1		0.33540		10	716.46862	1	1		0.146-01
α	ų "	802-89145				0.44640	~ :	Ξ,	20875 177		!!		10-311-0
م	•	802.51397		;		0.37F+CC		۰.	206.24.405				0.256-02
α	ď	802.05656	*****	}		0.316.00		٠,	704.2075				0.25E-C2
æ	9	801.52911	11111	;		0.54E+00			705.77110		1		C.54E-02
8	7	600,95634		;		00+391.0	13		705.19613				C.296-02
0	0	783.66359		;		0.30E+00	13	5	704.47115		;		0.326-02
0		783.60220	1 1 1 1 1 1	! !		0.15E+C0	13	4	703,59594		ļ		0.7CE-02
<u>.</u>		783.41994		ļ		0.15E+00	13	^	702,58011		ŀ		0.406-02
, (•0 •	163.11506	•	ļ		0.335400	13	æ	701.43729	1	1		0.456-02
,	•	19001 791	1	}		0.165.00	- 13	0	700.18763	1	;		0.1ct-01
۰ ٥	n •	781 58447	1 1 1	; ;		0.335400		s :	698.86111	1	1		0.55E-02
•		780.91028		;		0.146+00	3:	::	£1206.149		; ;		C. 585-02
0	00	760.20423		1		0.996-01			00017.040				
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TABLE II—Continued

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0.00000000000000000000000000000000000		CALC	EXP	δ×10 ⁵	e e	TNI
96.2.77228 9		958.01630			ں	0.60E+00
962.57328		957.64085			I	0.51E+00
962.22014 962.22034 20. 24 H DL 0.14400 10		957.27363	957.274211 80	80) 58	7 i	0. BEE+00
962.20067 962.20381 20 24 h DI 0.438+01 10 10 10 10 10 10 10 10 10 10 10 10 1		958-69124		÷ :	E	0.305401
962,20013 9 962,20014 CD		958,64965		;		0.166-02
962.20076 962.20034 200 1 H PI 0.546400 10 962.2007 962.20034 200 1 H PI 0.23641 10 10 962.2007 962.20034 200 1 H PI 0.23641 10 10 962.2007 962.20034 200 1 H PI 0.23641 10 10 10 962.2007 962.20034 200 1 H PI 0.23641 10 10 10 962.2007 962.20034 200 1 H PI 0.23641 10 10 10 962.2007 962.20034 10 10 10 10 10 10 10 10 10 10 10 10 10		558.52969	1	-		0.45E-02
962.20070 962.200541 (20) 21 A LU B. 0.235.01		958, 33374	1	1		0.326-01
962.22076 962.22076 97 0		958.03323				0.326-61
962.12807 2017 2017 2017 2017 2017 2017 2017 20		957.68719	957.487091 20	5- 107		0.556-01
962.12887 962.1288801 200 -5c H 718 962.128891 200 -5c H 718 962.1288801 200 -5c H 718 962.128881 200 -5c H 718 962.12881 400 -5c H 718 962.1288		957.28682			I	0.216+60
061,92725 952,1828(11C5 52		ě			I	0.18f+LC
061194229 961194788 501 45 H DL 0.2EE+CO		956.40271		7	U	0 - 30£ +00
961,19219 961,733941 200 94 10 0.588401 10 10 10 10 10 10 10 10 10 10 10 10 1					Ι,	
961,77331 961,77354 501 43 H DL 0586401 110 110 110 110 110 110 110 110 110		45094.004	15 9386 566		נע	10+101+01
961,66229 961,66176 201 -52 1 18 0 1916,6629 961,6628 961	-	055,63304				0.00
961,48420) 961,48420) 961,48420) 961,48420) 961,48420] 961,48420] 961,48420] 961,48420] 961,48420] 961,48420] 962,48420]	•	058.105801	1			
96.1.48828.		058.06151				F. 45E-03
961.26168 961.26197 501 31 H DL 0.266401 111		957.93739				216107
961,20168 961,221997 501 31 H DL 0,267601 111 111 111 111 111 111 111 111 111		067 65626	10000			
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90.09336 90.08705 90.08705 90.08705 90.08705 90.08705 90.08705 90.08705 90.08705 90.08706 90.087		2570775		1		0.656-01
960,94595		056.0750	256-076001			1,546
960,3259 960,3256 100 100 100 100 100 100 100 100 100 10					- 3	10.10
960,3565 96.000000000000000000000000000000000000		055.55304	055 542631 005	2016 - 545	, L	0.656-01
966,33297 96.13526 96.1 571 C C 6.16641 111 111 111 111 111 111 111 111 11		055.03178		200	_ د	336+00
\$\\ \text{\$60,132.97} \text{\$60,135.26} \text{\$60,000.134} \$60,0		0.700000		3		001110
960,13345 96C,14221 (40) -23 H CL 0,970 (1) 17 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1		054.18540				0.10110
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965,40977 960,31512 60, -3 + 01 0.218-00 175 60,03751 96,		77007				
\$66.31516 \$66.313121 (c) -2 # D1 G.12E+CC 12		057 24016		1		415-03
966.113451 960.1134 91 0.6616.00 1174 91 0.6616.		062 16906				20110
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\$55.71741 \$45.71765 \$1.00 \$1.0		*CT/ 2006	,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,		:	20-30-0
995,46071 957,71764(20) -26 C TT		0770 9070		100	5 6	201101
955,21299 955,21256 (40) -15 H 010 0,346+011 17 955,21299 959,21256 (120) -15 H 010 0,346+011 17 959,21256 (120) -15 H 010 0,346+011 17 959,31259 959,72126 (1100) -15 H 010 0,256+010 959,32739 959,73105 (100) -10 H 010 0,256+010 958,7329 959,732813 H 010 0,000 H 010 0,000 H 010 958,7329 959,7409 0,000 H 010		955 32011	77.701.101.101.101		c	101111111111111111111111111111111111111
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(559,870,48) 95,8126,01001 -38 A TN C		369717656		!		0.546-01
1 1 1 1 1 1 1 1 1 1		75875.567		1		0-279-0
99.83508 955.63465 551 -62 H DL 0.12E-01 13 555.7502 959.731055 501 -62 H DL 0.5EE-01 13 555.5527 955.5221		553.00037	-	!		0 - 1 4 - + CC
557.7307 979.7305.501 -42. H D1 0.156-01 13 557.7307 979.73105.501 -2. H D1 0.256-00 13 557.3727 979.73105.501 -2. H D1 0.256-00 13 557.7307 979.73105.501 -10. H D1 0.256-00 13 557.7307 979.73011 800 -10. H D1 0.256-00 13 557.7307 979.73011 800 -10. H D1 0.266-00 13 557.7307 979.73011 800 -10. H D1 0.266-00 13 557.7307 979.73011 800 -10. H D1 0.266-00 13 557.7307 979.7301 10. H D1 0.756-00 13 5		952.61384	-	!		0.526+00
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55.32753 954.32813 80) CC H DL C.225400 13 554.32753 954.32813 80) CC H DL C.425400 13 554.4305 954.43612 00 10 H DL C.46540 13 558.13675 955.14996 [20] 15 C IC C.16540 13 558.13675 955.14996 [20] 10 C IC C.255400 13 958.13676 52.15678 [20] 10 C IC C.255400 13 959.15678 52.15678 [20] 10 C IC C.255400 13 959.15678 52.15678 [20] 10 C IC C.255400 13 10 C IC C.256400 13 10 C IC C.266400 13 10 C IC C		626.94910	1 1 1 1 1 1	1		0.35E-04
954,2759 954,05171 50) 16 H DL (.258+00 13 954,0456 955,05171 50) 15 CL (.168+01 13 958,1507 958,144961 20) 15 CL (.168+01 13 958,1507 958,149961 20) 10 CL (.168+01 13 958,1507 958,15071 20) 11		956,84258		!		C.166-03
554.7034 995.74509 [20] 10 H D C.46640 13 958.77454 995.74509 [20] 15 C C C.166401 13 958.7456		956.43746	1	!		C. 616-03
938.43494 938.44609(20) 15 CTC 6.166.01 13 958.43405		956,17927	1	!		C. E1E-C3
958.4406 958.14996 (20) -10 0.144401 13 958.4406 958.14996 (20) -10 C 1C 0.244401 13 959.28304) 0. 1 LS 0.244401 13		955.73949	,	;		0.15E-C2
928.15607 958.149961 20) -10 C 1C 0.244461 13 929.28304) 0.		955.21565	1	-		C. 51E-C2
(959.28304) 0. 13		954.61848	1	1		0.426-62
(959.28304) 0.		553,96249		1		C. 74E-62
		553.26RC1	1			10-346.0
959.24445 959.24£00(50) 55 H GL 0.45E-02		952.56424	1			10120
959-1316C 959,13140(50) -19 H DL 0,19E-Cl		951.3936.8		-		10-32-0
958.92040 95F.91999(50) -40 H DI 0.93F-01		30415 150				104071
958.08184 958.081601 601 -23 H 01 0.938-01				•		70.11.10
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X	CALC (982-63388) 11052-25426 11052-25426 11052-174321) 1021-73208 11021-707315 11021-707315 11041-004976 11041-004976 11041-004976		δ _{×10} 5	dcI è	F		ν.		d × d	δ×10 ⁵	qui	FNF
	982.63388) CC2.25457 CC2.27426 CC1.743218 CC1.73208 CC1.08966 CC1.07915 CC1.06916					,		CALC	Š			
- *	CC2.25457 CC2.24226 CC1.73218 C21.73208 C21.76696 C41.08966 C41.08976 C41.0811		1				1 -	(1172,09808)		;		
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•	041.00911 041.04956 041.07915 041.00811 060.286593		: :					87637	1	;		12F+06
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•	C41.07915 C41.04976 O41.00811 C6C.28655)	1041-056461	201		0 755 +01			66735		1 6		1 1 5 2 4
•	041.04976 041.00811 060.28655)	1041.075400	20) 25	- E	0.37E+C1			1. 6CE26	-	1		C. 15E+00
•	041.00811	1041.049970	301 2		C. 32E+C1			1.57537		1		0.16E+00
~	C6C.286551	1041,006100	301		0.406+01			1171,62465		1		0-306+00
		-	1					. 80295		1		•11E+0
	000.27616	1	-		C . 33£ +01			1.14542		1		0-846-61
	CE C. 247C4		1		0.21E+C1			13081		1		426-0
	106C.20367		1		0.51E+C1			. 10564	1	ţ	_	0.44E-C1
	066.16190	***	-		C.16E+01			1190,02263		1		0.92E-C1
	079.32856	1 1 1 1	ì		C. 51£+01	11		.96208		;		0-49E-C
	079.31792	1 1	1		C. 25E+01			1169.79110	1	;		0.536-03
	679,28795		1		C. 246+C1	11		67605		ļ		1.12E+00
	079.24408		-		0.458+61	11		1189.57286		}		• £ 5 E - C
	079.19605				0.156+01	-1		. 50310		1		0.71E-C
,	1079,16109		-		0.126+01	11		1189.45665	11000	1 7		0.158+00
J	098.211831	1 1 1				11 11		1.59532		•	•	0.756-C
-	058.20066				0-176+01			5.85772	-	1 2		C.57E-0
	056.16914	•			0-17E+01		~	. (2222)		;		
	1098,11537		:		G-32E+C1			689000		;		0.156-01
	.98.C6622	•	1		0.156+01			1.00314		1 5 7		0.15E-C
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	1098.00435	1098.004750	201 40	5 7	C.17E+01		-	1.75468	1	;		0.17E-C
	1116.93324		-		0.2CE+C1	1.5		1207.63009		1		0-351-0
	116.92138	1	1		0.1CE+C1			4.49639		;		0-435-01
	116.88809		-		0.10E+01			.36/48		1 1		10-157-0
	116.83576	•	•		C.20£+01			.26189	1	;		0-285-01
	116.77093	1 1 1 1 1	;		0.106+01			. 20488	1	;		. 63E-(
	116.71317	1	;		0.556+00			.23165	-	,		0.346-01
	116.67755	-			C+17E+01			7.39173	-	}		C. 35E-0
	116.69071	1	į		0.576+00			1.75497	111111	1		0.56E-C
	135,48993)		1		•			1225.73480	1 1 1 1	;		20-145-0
	135,47733	1	!		0.535+00			11840	1	;		
	135.44263	•	1		0.54E+00			19286		(1)		
	135,36802		1		0.116+01			5. 60588		;		0.10E-01
	135,30877		:		0.56E+00			.43754		,		3ct-0
	135.2376¢	1 1 1	;		0.578+00			5.29736		t i		0.626-02
	135.18170	-			0.11E+C1			14359		i 1		7
	135, 16463		-		0.506+00			. 98895	-	;		C-81E-0
	135.21948	1	!		0.35E+C0			1224.84938		1		0.54E-02
	153,87906		1		0.5CE+00			4.7462t		;		+226-C1
	1153.86576		-		0.258+00	13 10		224,70887		;		0.13E-01
	153,83082		1		0.24E+00			1224,77831	1	!		0-151-0
	153.76684		!		0.54E+00	13 1		0.01282		;		0.316-C1
	153.67796	4 5 7 1 1 1	!		0.28E+C0			. 49524	1	1		.136-0
2	1153.59334	447.5.	ļ		0.25E+00							
	1153.51695		!		0.61E+CC							
	153.46975		1		0.316+60							
8	1153,48073	1	ł ł		0.286+00							
; ;	153.59026		-		0.41E+C0	_						

TABLE II—Continued

TABLE II—Continued

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027-10577 022-1057 021	>				l		0.	^	-	934.17133)			1	
Control Cont	٥.			-			•0	٠		933,94108	933.94143(35	
728-21037	٠,	927.69675		ļ			0.23E+01	<u></u>		933.25804	933.25821(7.	
Q25.13288 Q25.13288 Q25.13289 Q25.13289 Q25.13289 Q25.13281 Q25.13281 <t< td=""><td>,</td><td>928,21057</td><td>928.216426</td><td></td><td></td><td>ā</td><td>0.105+01</td><td>, ,</td><td></td><td>436404364</td><td>936-035981</td><td></td><td>1 1</td><td></td></t<>	,	928,21057	928.216426			ā	0.105+01	, ,		436404364	936-035981		1 1	
\$22,4038 \$27,40346 \$20 \$34 \$45 \$			928,21057		_	, z	•	٠,		928,19279	928.192681		9	
727.4038 927.40234 (20) -35 C TN 0.44E+01 7 7 921.7534 921.7534 120) -30 L LS C C C C C C C C C C C C C C C C C					_	ž		1		78181.679	925.3324161		24	
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Q2E-18148 Q2E-						2 5	0011100							C. 12F-01
926.687796 926.818781 (20) -134 H 10	~	P.1814				2 Z	0.23F+01	· c	٠,	934.85134	934.85127(201	9-	
926.187796 926.187371 93.18640 93.74476 50 93.1869 93.74476 50 93.1869 93.74476 50 95.0000 95.						Z		_	,		934.85121(203	-12	
924.817012 926.837711 00 -24 H DD			928,18184(100)		_	ă		œ	۳	933.74686	933,746766	20)	ŗ	
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727-7578 927-7578 927-7578 9 10 0-286-00			_			7					930.046710		U,	
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TABLE II—Continued

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TABLE II—Continued

						" CALC and EXP are calculated and observed	"	II K		Symbole given in this column identify the		suthors of the corresponding measurement and	the technique used: A ref.(8). B ref.(9):		Cref.(10); Erefs.(12,13); Fref.(20);		H our massursment; L ref.(11). Experimental	the design of the state of the	techniques are denoted by the following symmot	HD diade laser heterodyne. DL standard diode		laser, LS laser Stark spectroscopy; IC and	TN are two photon infrared-microwave technique		with ${\rm CO}_2$ and ${\rm N}_2{\rm O}$ lasars, respectively; LC	and IN are two-photon infrared-microwave		techniques using the Lamb-dip effect.		This value has not been taken into the fit,																
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TABLE III
Ground State and ν_2 Excited State Molecular Parameters of ¹⁵ NH ₃ (cm ⁻¹) ^a

Parameter ^b	Ground-State Value	ν ₂ -State Value	Parameter ^b	Ground-State Value	ν_2 -State Value
ΔE _o	0,7576868(13)	34,437785(16)	(s) _E _o	o. ^c	928,456683(76) ^d
Δ B	-4.92922(13)x10 ⁻³	-0.1777943(51)	(s) _B	9.9223556(24)	10.0489023(65)
$\Delta(\text{C-B})$	6.85128(16)×10 ⁻³	0.2474842(80)	(s) _{C-} (s) _B	-3,700(13)	-3,964(14)
$\Delta o_{\mathtt{J}}$	-1,67575(45)x10 ⁻⁵	-0,43867(21)x10 ⁻³	(s) _D	0.85167(30)×10 ⁻³	1.13920(38)×10 ⁻³
Δ D _{DK}	4.60854(100)×10 ⁻⁵	1,19920(55)×10 ⁻³	(s) _D _D K	-1.5832(12)×10 ⁻³	-2.442/(12)×10 ⁻³
Δo _K	-3,14644(61)x10 ⁻⁵	-0.81075(53)×10 ⁻³	(s) _D ,	1.x10 ^{-3 c}	1.61535(70)×10 ⁻³
Δ H _{JJJ}	-0.40575(65)x10 ⁻⁷	-0.6501(29)×10 ⁻⁶	CCCH(e)	0.2546(35)x10 ⁻⁶	0.5534(53)x10 ⁻⁶
Δ H _{JJK}	1.6538(22)×10 ⁻⁷	2.599(11)×10 ⁻⁶	(s)H _{77K}	-0.907(17)×10 ⁻⁶	-2.236(24)×10 ⁻⁶
$^{\Delta}$ н $_{JKK}$	-2,2307(26)x10 ⁻⁷	-3.396(17)×10 ⁻⁶	(s) _H akk	1.125(35)×10 ⁻⁶	2,978(41)×10 ⁻⁶
Δ H _{KKK}	0,99622(101)×10 ⁻⁷	1.463(12)×10 ⁻⁶) (a)HKKK	-0.5×10 ^{-6 c}	-1.320(14)×10 ⁻⁶
ΔG_{JJJJ}	0.7608(42)×10 ⁻¹⁰	0.564(13)×10 ⁻⁹	(\$) _{G_7777}	-0.982(102)×10 ⁻¹⁰	-0.183(24)×10 ⁻⁹
$\Delta_{G_{JJJK}}$	-4.110(20)×10 ⁻¹⁰	-2.877(70)×10 ⁻⁹	(s) _G	5.68(60)×10 ⁻¹⁰	1.20(13)×10 ⁻⁹
$^{\Delta G}_{JJKK}$	8,248(36)×10 ⁻¹⁰	5.29(16)×10 ⁻⁹	(s)G _{TDKK}	-12.1(12)×10 ⁻¹⁰	-2.60(28)×10 ⁻⁹
$\Delta {\sf G}_{\sf JKKK}$	-7,296(27)×10 ⁻¹⁰	-4.23(19)×10 ⁻⁹	(s)GJKKK	10.0(11)×10 ⁻¹⁰	2.17(26)×10 ⁻⁹
ΔG_{KKKK}	2,4013(78)×10 ⁻¹⁰	1,253(100)×10 ⁻⁹	(s) _{GKKKK}	-1.×10 ^{-10 c}	-0.436(90)×10 ⁻⁹
ΔL_{33333}	-0.8737(102)×10 ⁻¹³	o. c	ηο 3	4.2650(38)×10 ⁻⁹	1,709(38)×10 ⁻⁹
Δ L _{JJJJK}	5.909(64)×10 ⁻¹³	o. c	73	-4.947(15)×10 ⁻¹²	o. c
$\Delta L_{\rm JJJKK}$	-15.81(15)×10 ⁻¹³	o. c	η 6	1.73(24)×10 ⁻¹⁹	o. c
Δ L _{JJKKK}	20,92(18)×10 ⁻¹³	o. c	α°	8,0x10 ⁻⁵ c	1.2940(49)×10 ⁻⁴
Δ L _{JKKKK}	-13.715(104)×10 ⁻¹³	0. c			•
Δ ^L KKKKK	3.565(24)×10 ⁻¹³	o. c			

a Values in parentheses are standard deviations of the parameters, given in units of the last digit quoted.

It should also be noted that the diode laser data of $^{15}NH_3$ in Table II (obtained by using $^{14}NH_3$ as a calibration standard) are highly consistent with the other precise ν_2 band data. This confirms that the $^{14}NH_3$ ν_2 band wavenumbers (2) can be used as a reliable calibration standard.

It is obvious from a comparison of the spectroscopical parameters of ¹⁴NH₃ [Table III in Ref. (2)] and ¹⁵NH₃ (Table III in this paper), that there is a close correspondence between all of the ¹⁴NH₃ and ¹⁵NH₃ molecular parameters. Such a correspondence can be expected but the fact that we did arrive at it seems to indicate that the ¹⁴NH₃ and ¹⁵NH₃ data sets are consistent.

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b If X denotes a general parameter, Δx is defined as $\Delta x = {}^{(a)}X - {}^{(c)}X$, where ${}^{(a)}X$ is the parameter referring to the upper level of the inversion doublet; ΔE_0 is the inversion splitting in the ground and Y_2 excited state for $\Im = K = 0$; \Re and \Re are parameters describing the $\Delta k = \frac{1}{2}$ 3n interactions $\{ \text{refs.}(1,2,4) \}$.

Constrained value,

^d Band origin for the $a \leftarrow s$ transition is 962.894468(92)cm⁻¹, for the $s \leftarrow a$ transition is 927.698996(77)cm⁻¹.

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