The Emission Spectrum of Hot Water in the Region between 370 and 930 cm⁻¹

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An emission spectrum of the water molecule at a temperature of 1550°C has been recorded in the range from 373 to 933 cm⁻¹. More than 4000 pure rotational lines were observed with the strongest belonging to the ground state (000) and the first excited bending vibrational level (010). Transitions involving rotational quantum numbers J and K_a significantly higher than previously recorded have been assigned. © 1996 Academic Press, Inc.

I. INTRODUCTION

The pure rotational spectrum of water at room temperature has been studied in the millimeter wave, submillimeter wave, and far infrared regions (1-9). Even more studies have been devoted to the rovibrational bands of water (10-21) including overtones with wavenumbers as high as 25 000 cm⁻¹. Among the IR and near IR studies are the classic works by Flaud, Camy-Peyret, and co-workers (10, 11) on the oxyacetvlene flame spectra of water at temperatures as high as 2900 K. Information on the energy levels for high values of the rotational quantum numbers J and K_a are derived mostly from the transitions in these flame spectra. The ground state energy levels up to J = 35 and $K_a = 20$ were derived in Ref. (10) and the levels of the first excited bending state (010) up to J = 32 and $K_a = 14$ were reported in Ref. (11). These energy levels were used for a comprehensive atlas of the water lines (22) and for the assignment of a hot water spectrum (9).

The energy levels derived from the flame spectra have been used in numerous theoretical papers, in which effective Hamiltonians were developed for the theoretical treatment of the highly excited rotational states of water molecule (3, 4, 23-27). Within the framework of the variational approach, the fitting of the potential energy surface of water was accomplished using these energy levels as the input data (28-34).

Experimental information on highly excited rotational levels of water in the ground and lower vibrational states is very important and is widely used in many applications. For example, the infrared spectrum of a sunspot (36, 37) contains thousands of molecular absorption features. Many of these lines belong to the water molecule but most of the lines do not have rovibrational assignments. This spectrum awaits detailed assignment and the obvious step toward this goal is the study of the high temperature water spectrum in the laboratory.

The present study of the hot water spectrum was motivated both by the desire to extend the rotational energy levels higher than those reported in the literature and by the necessity of understanding the sunspot spectrum in more detail.

II. EXPERIMENTAL DETAILS

A high resolution emission spectrum of hot H₂O was recorded with the Bruker IFS 120 HR Fourier transform spectrometer located at the University of Waterloo. The spectrum was recorded using a KBr beamsplitter with a liquid helium cooled Si:B detector in the region 350–1000 cm⁻¹ with a resolution of 0.01 cm⁻¹. The lower limit of the spectrum was set by the transmission of the KBr beamsplitter and the detector response, while the upper limit was determined by a cold filter.

Water vapor flowed continuously from a room temperature glass container into the hot cell. The cell was a 1.2 m long by 5 cm diameter mullite (3Al₂O₃·2SiO₂) tube with KRS-5 windows attached at each end. The mullite tube was placed inside a commercial CM Rapid Temperature Furnace and the pressure maintained at approximately 15 Torr with a vacuum pump. The tube was slowly heated to a temperature of 1550°C. It was necessary to have a constant flow of water vapor since the water vapor reacted rapidly with the hot walls of the tube. Higher temperatures were not obtained because the mullite tube cracked at 1550°C.

III. ANALYSIS

The spectrum was reduced by fitting the water lines with Voigt profiles to determine line position, width, and relative intensity using PC-DECOMP, an interactive spectral analysis program developed by J. W. Brault. The laboratory data were calibrated using J. W. C. Johns' H₂O measurements (8) and a calibration factor of 1.00 004 870 768 was applied to correct the observed line positions.

 ${\bf TABLE\ 1}$ Assigned Line Frequencies of the Hot Water Spectrum with Intensities Higher than 0.1 r.u.

Wavenumber	Intensity	ي.	$K_a^{'}$	K_c	٦,	Ka	K_c	(0v0)	Wavenumber	Intensity	`~	K. 1	K',	J K	, K	(0^0)	Wavenumber	Intensity	٦,	K'	K'	-	K	Kc	(0^0)
407.30521	0.0772	19	4	16	18	က	15	(000)	499.08242	0.2584 2	23	5 1	18 22	2 6	17	(000)	568.30274	0.4270 1	15		8	4) 6	(010)
418.96138	0.1094	19	4	15	18	2	14	(000)	499.78570							НО	568.94562	0.1504 3		3 2		59	2	_	(000
423.23867	0.0754	20	က	17	19	4	16	(000)	502.19364	0.3375 1	13	9		2 5	œ	(010)	571.28617	0.1647 1					3 10	_	(000)
423.75482	0.1107	20	4	17	19	က	16	(000)	502.61196	0.1098 2	24	•	20 23	3 5	19	(000)	572.88411	0.1932 3	32 (32 3	31	1 31	_	(000)
424.87537	0.1083	23	0	23	22	7	25	(000)	502.93212	0.2643 1				6	-	(010)	572.94282	0.3620 1	. 91				6		(000)
425.16224	0.1165	22	П	21	21	2	20	(010)	503.21357		24		20 2	3 4	19	(000)	573.32095	0.1837 3			_	30	1 29	_	(000)
425.43655	0.1074	22	3	21	21	_	50	(000)	503.97049		6	33		8	œ	(010)	574.53908		14 10	01	5 1	13		_	(000)
432.46246	0.1046							НО	504.38274		13		8 12	2 4	6	(000)	574.55568			0	4 1	13	6	2	(000)
433.39106	0.1044							НО	504.61545		25	3 2	22 24	4 4	21	(000)	575.15347		12 12	2	0 1	11 1	=	_ _	(010)
435.35756	0.1177	11	2	9	10	4	2	(010)	504.64520		25		-	4 3	21	(000)	575.56939	0.3121							art
436.95054	0.1184	20	4	16	19	ಬ	15	(000)	506.68047		56	3 2	-	25 2	23	(000)	575.58194	0.3326 1	3	7	1 1	12 1	_	_	(000)
438.07435	0.1290	24	0	24	23	1	23	(010)	506.74287	0.1175 2	23	6 1	18 22	2 5	17	(000)	576.69533	0.2298 1		7 1	1 1		6 1	01	(010)
438.12334	0.1026	21	9	15	50	7	14	(000)	507.92176	0.2117	14	~		13 6	2	(000)	577.23550	0.3603 1	91	œ	9 1	15	_		(000)
439.88645	0.1673	21	က	18	50	4	17	(000)	508.56081	0.2445 2	82	7	28 27	0 2	27	(000)	577.73289	0.4783 1	13 10	10	4	12			(010)
440.17247	0.1031	21	4	18	20	က	17	(000)	509.04550	0.2584	22	7		26 2	25	(000)	580.21560	0.4751 1	15	6	7	14	- - - -		(000)
441.21862	0.1404	22	2	50	21	က	19	(000)	510.58378	0.1248	12	4	8	1 3	6	(000)	580.22725	0.1264							art
441.22887	0.1535	22	က	20	21	2	19	(000)	513.39309	0.2377	11	6		10 8	3	(010)	580.43439	0.3818 1	5	6	6 1	4	œ	⊃ -	(000)
441.70186	0.1050	22	ဗ	20	21	2	19	(010)	514.76864	0.2043	13	2		2 4	6	(010)	580.73232	0.1318 1	2	က	9 1	_		01	(000)
441.87974	0.2018	24	0	24	23	_	23	(000)	514.91255	0.1692	14	9		13 5	6	_	580.84027*	0.2818 2	. 02	7 1	4 1	61	6	() (3)	(010)
442.21655	0.2051	20	ഹ	16	19	4	15	(000)	517.18934	0.1398	25	4 2	21 2	4 5	20	(000)	581.00694*	0.2796 2	50	7 1	14 1	61		13	(010)
442.25198	0.1257	53	1	22	22	2	21	(010)	517.78161	0.2005	13	-	7 1	12 6	9	(010)	581.14986	0.2195 3	90	4	7 2	29	3 2	.) 92	(000
442.42334	0.1552	23	2	22	22	-	21	(000)	518.41450	0.1591	14	t-	7 1	13 6	œ	(000)	581.25267	0.2889 1	` ∞	7 1	2 1	17	6 1		(010)
446.12087	0.1307	6	œ	_	œ	-1	2	(010)	518.78710	0.1481	24	6 1		23 5	18	_	581.61518	0.3224 1	ੱ ਲ੍ਹ	4	9 1	12	3 1	· 01	(010)
447.41481	0.1008	13	9	œ	12	5	_	(000)	519.14188	0.2550	1.5	∞	5 1	1 7	4	(010)	583.38113	0.3781 1	9	∞	8 1	15		6	(000)
449.27890	0.2260	23	2	16	22	œ	15	(000)	519.20352	0.1916	12	∞	4 1	1 7	3	(010)	584.20394		3 1	3		_	2		(000)
449.69817	0.1511	10	~1	4	6	9	33	(010)	520.13515		12	4		1 3	6	$\overline{}$	584.25344	0.1079 1	, ,	4	8 1				(010)
449.79489	0.1385	10	2	က	6	9	4	(010)	520.35162		56	4	• •	25 3	22	_	584.70802	_	7	2			2	<u> </u>	(000)
454.08393	0.2339	21	4	17	50	J.	16	(000)	520.58546		13			12 6	2	_	585.25851	_	4	6	6 1			_	(010)
456.34126	0.1663	22	က	19	21	4	18	(000)	522.58329		27	7		26 3	24	_	585.29478			6	5 1	13		_	(010)
456.50106	0.2446	55	4	19	21	က	78	(000)	522.74032		15	_		14 6	œ	_	586.62282	_		∞	∞	14		_	(010)
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537.99484 538.27415 538.83496 539.14306 539.56964	539.58613 539.63238 539.82100	540.14835 540.95752	541.03491 541.30157 541.49465	542.47003 542.65487	544.06961 544.26008	550.19349	553.23923	553.73506 554.48878	554.63534 557.03969	557.19993	557.48804 557.54557	557.61690	560.44380	560.66066	561.82984	562.46483	562.76517 563.87255	564.31234	564.86756	565.02529	565.62834	566.27120	567.61545 567.75501
HO (000) (000) (010)	(010) (000) (000)	(010) (010) (010)	(010) (000) (000)	(000) (000)	(010) (000)	(000)	(000)	(010) (010)	(000)	(000)	(000)	(000)	(000)	(010) (000)	(000)	(010)	(010) (010)	(010)	(010)	(010) OH	(010)		HO (000)
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466.85554 468.05342 470.50693 470.64555	471.85437 472.41776 472.61404 479.70433	473.34816	474.29606 474.29606 474.37711	475.49734 475.49734 476.00912	478.14645 480.31431 480.73139	483.28368	480.80013	488.39672 488.52555	488.70675 488.75795	489.57321	490.57922 492.10020	492.59836	494.13252	494.15008	494.28165	495.16398	495.25026 495.26578	496.17999	496.31042	496.39507	497.24411	497.95896	498.16329 498.75866

Wavenumber	Intensity	y J'	, K.	, K'	٠,	Ka	Kc	(0^0)	Wavenumber	Intensity	٦,	K_a'	K'	J K.	. Ke	(000)	Wavenumber	Intensity	ند مر	, K,	K	٠,	K,	Kc	(0 _v 0)
	0.1261	12	2	10	11		11	(000)	702.30630	0.3356 1	19 11		9 18	3 10	000	(000)	763.26527	0.1852	22	11	Ξ	21	10	12	(000)
	0.5675	14	13	-	13	12	7	(010)	702.37810	0.4514 1	9 1	_	8 18	3 10	6	(000)	763.49275	0.2520	20	12	6	19	11	∞	(010)
	0.5063	12	13	7	14	12	က	(000)	703.69963	0.4930 1	7 1	_	0 16	91 9	_	(000)	763.78639	0.2249	24	10	15	23	6	14	(000)
	0.2971	12	4	6	= :	-	10	(010)	703.71594	_		4 11	13		12	(000)	766.73727	0.3414	19	9	13	18	v	14	(000)
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	0.3080	2 2	o 00	2 =	17	1 1-	; =	(000)	706 16387	0.9040	0 9	o -		7 0	7 =	(000)	770.09591	0.4464	2 2	C 1	o -	61	10	0 0	(000)
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	0.2664	2 2	9	4 00	1 2	g er	0 ر	(000)	707 44311	0.3799 1	- o	, -	∓ ∓ + or	' -) I	(010)	771 75389	0.0055	10	16	0 ~	0 7	1 L	ים מ	(000)
	0.6405	2 2	0.	0	17) oc	6	(000)	707.45721	0.1447	, ~ , ~	- - —	1 -	10	- ∝	(010)	779 57363	0.1905	61	7	10	0 1 2	3 6	· "	(010)
642.78348	0.5673	17	10	œ	16	6	7	(000)	707.67042*	•	20	9 1	12 19	∞	11	(010)	773.63854	0.2553	19	18	5	2 20	17		(010)
642.91061	0.5273	17	10	2	16	6	œ	(000)	708.26088	0.2546 1	4	5 1	10 13	~	11	(010)	773.83385	0.2216	19	17	က	18	16	2	(010)
	0.3768	18	7	Ξ	17	9	12	(000)	708.33956	0.2448 1	<u>«</u>	6 1	1.	2	13	(000)	774.00120	0.5466	20	16	4	19	15	5	(000)
	0.1442	14	4	10	13	က	11	(010)	709.75494		15	4 1	1 14	1 3	12	(010)	774.53229	0.3054	20	13	∞	19	12	7	(010)
	0.4727	<u>8</u>	6	6	17	œ	10	(000)	712.57127	-	19 1	0 1	0 18	8	6	(010)	778.11966	0.5172	20	17	က	19	16	4	(000)
	0.5352	15	15	0	14	14	,	(000)	712.97183	_	19 1	01	9 18	о Ж	10	(010)	778.70499	0.3984	20	20	-	19	19	0	(000)
	0.4940	16	10	_	15	6	9	(010)	713.05216	0.4932 2	Ξ	9 1	2	« —	13	(000)	778.77158	0.4053	22	12	Ξ	21	Ξ	10	(000)
	0.2888	16	10	9	15	6	۲-	(010)	713.79192	0.2222 1	2		0 1.	4	2	(000)	778.88644	0.2152	22	12	10	21	11	11	(000)
	0.5403	16	12	4	15	Ξ	ಒ	(000)	714.40047	0.4915 1	7	4	3 16	3 13	4	(010)	780.37298	0.4918	20	18	2	19	17	က	(000)
	0.5349	15	12	က	14	=======================================	4	(010)	714.81549	0.6264 1	8	4	4	7 13	r.C	(000)	780.66034	0.4359	50	19	2	19	18	_	(000)
	0.2383	12	6	6	16	œ	œ	(010)	716.55537	0.6760	9 1	2	7	3 11	οκο	_	781.79343	0.5054	21	14	-	20	13	œ	(000)
	0.2059	13	4	10	12	_	11	(000)	718.65746	0.3193 1	r.G	5	<u>-</u>	1 2	12		782.44360	0.3227	23	Ξ	12	22	10	13	(000)
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	0.4983	1.7	9	Ξ	16	5	12	(000)	720.10263	0.4903	∞ 	15		7 11	ဖ	_	783.48504	0.1349	50	14	7	19	13	9	(010)
	0.1258	!		;		1	:	НО	721.23345	$0.2598 ext{1}$	4	9	9	es es	10	(010)	783.76246	0.2464	17	9	12	16	က	13	(000)
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658.69005	0.6255			<u>.</u> و	91	10	<u>-</u> 9	(000)	721.75104	0.6394 1	 ∞_!	15	e d	7	4 1	(000)	785.57960	0.2196	17	2	1	16	4	12	(000)
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	0.5839	16	=	2	15	10	9	(010)	726.00050		'	 	3 20	0	14	(000)	791.97967	0.0010	3 3	- 5	5	2 2	2	10	(000)
	0.5926	<u>x</u>	10	6	17	6	œ	(000)	726.39964	0.5578	21 1	0 1	1 20	5 0	15	(000)	792.01535	0.4598	22	13	10	21	12	6	(000)
664.53574	0.4585	18	10	œ	17	6	6	(000)	726.62928	0.6715	18	9	2 1	.7 15		(000)	793.70547	0.1439	14	က	12	13	0	13	(010)
	0.1175							art	728.28407	0.6706	19	က	6 1	18 12		(000)	793.90849	0.1848	13	9	7	12	က	10	(000)
	0.1161	13	2	6	1.2	2	10	(010)	728.56502	0.1129	14	3 1	1 1	3 2	15	(010)	794.12159	0.2282	<u>×</u>	v	13	17	4	14	(000)
	0.5423	15	14	_	14	13	7	(010)	728.95342	0.1838	19		9 1	8 1(~	(010)	794.67518	0.2074	50	16	ល	19	15	ব	(010)
	0.3017	Ξ.	. ت	10	15	2	Ξ	(010)	728.99050	0.3976	19		8	8 1(·	(010)	795.69543	0.1536	21	13	œ	20	12	6	(010)
	0.5867	91	7	5	15	13	က	(000)	729.41598	0.6410	<u>∞</u>	7	1	7 16		(000)	796.57162	0.4249	21	16	5	20	15	9	(000)
	0.5083	61	ۍ <u>؛</u>	10	<u> </u>	∞ ;	Ξ.	(000)	729.94740	0.6295	. 18 	∞ , ∣	0 1	7 17		(000)	797.18824	0.2277	24	11	14	23	10	13	(000)
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	0.7005	7	7.0	0	10	a	_	(010)	(32.44626	0.3347		2	-	,. D	2	(010)	798.12288	0.1195	53	12	12	7.7	⊒	=	(000)

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0.2892 0.1383	0.2254	0.1118	0.2363	0.3852	0.1725	0.2252	0.3531	0.1302	0.1103	0.3552	0.2177	0.4440	0.4137	0.4627	0.2000	0.1856	0.2746	0.1330	0.4788	0.1429	0.2785	0.2957	0.1864		0.2082	0.1162	0.1341	0.2227	0.3953	0.1842	0.1538	0.3396	0.2362	0.4713	0.2857	0.2512	0.2332	0.3132	0.1785
798.36530 798.56397	798.75874	199.00573	801.36109	802.81730	802.98999	803.50603	804.11311 904.10000*	804.19000* 804.42324	804.58205*	805.22471	805.99389	806.69581	808.03805	811.55952	011.33303 811.06458	811.97965	813.75067	814.51706	815.30059	816.15525	816.45026	816.68703	817.15695	818 42382	819.93233*	822.18246*	822.26460*	822.95571	823.53731	825.63552	826.13343	826.97402	827.04029	827.70865	828.75481	828.83165	831.19402	832.31968	835.45046
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0.1988 0.1280	0.5245	0.6404	0.6405	0.3222	0.4628	0.2110	0.4915	0.1029	0.2901	0.4951	0.4469	0.1039	0.3231	0.2416	0.0000	0.4509	0.2705	0.6514	0.2422	0.3938	0.3273	0.3674	0.1273	0.5159	0.6721	0.999.0	0.3086	0.4034	0.4227	0.6575	0.1329	0.5795	0.6026	0.2605	0.3065	0.5374	0.6202	0.3462	0.4249
																							55 45 57	2 8	16	28	95	112	0 29	09	66				898	33	74	94	571
732.51092 733.38132	734.98112	737.85031	737.95872	737.97236	738.04973	738.45629	740 56005	741.20236	742.00289	742.06775	742.12768	742.43352	743.12339	743.34735	743 79640	744.21132	744.81380	745.23143	746.58395	747.31583	747.96251	748.78707	749.74234	749.77550	750.16491	750.69458	751.18495	752.26512	752 88465	754.20060	754.86499	754.92380	755.68307	755.98181	758.71268	758.73833	760.11674	761.00294	762.27571
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0.5213 0.3754	0.4293	0.5751	0.2120	0.5165	0.2964	0.4858	0.5364	0.1517	0.3179	0.5320	0.4064	0.1567	0.4845	0.4766	0.3501	0.4000	0.2254	0.1086	0.4757	0.1578	0.3166	0.2722	0.4528	0.1359	0.1349	0.4157	0.1955	0.3870	0.2483	0.5792	0.3893	0.4286	0.2115	0.4530	0.5178	0.1411	0.1818	0.1111	0.4661
670.31846 671.36049	671.75283	671.86728	673.57662	674.15507	674.53197	675.62922	676.09636	676.55000	679.77495	680.81836	680.84398	682.24652	682.65079	683.29582	604.11933	685 59269	687.47605	687.98667	688.33863	689.03754	689.72957	690.02014	690.05509	690.22063	691.37971	691.79103	691.95960	693.11754	693.52831	694.51611	695.28114	696.52345	697.05479	697.44090	697.56964	697.91839	698.46870	600 65764*	701.77622

TABLE 1—Continued

Wavenumber	Intensity	J'	K_a'	K_c'	J	$K_{\mathbf{a}}$	K_c	(0v0)	Wavenumber	Intensity	, J	K'a	K' _c	J	$K_{\mathfrak{a}}$	K_c	(0v0)
835.55331	0.1698	14	8	7	13	5	8	(000)	866.17192*	0.1237	24	17	7	23	16	8	(000)
839.57208	0.2469	23	16	7	22	15	8	(000)	870.24464	0.1431	24	18	6	23	17	7	(000)
842.89688	0.1814	24	14	10	23	13	11	(000)	871.84116	0.1465	25	15	10	24	14	11	(000)
843.54665*	0.0990	23	23	0	22	22	1	(000)	872.97092*	0.1659	19	6	14	18	3	15	(000)
845.10003	0.2290	23	17	6	22	16	7	(000)	872.97092*	0.1659	24	19	5	23	18	6	(000)
845.73149	0.2045	16	8	9	15	5	10	(000)	873.05353	0.1227	24	22	2	23	21	3	(000)
848.57312	0.1310	23	22	1	22	21	2	(000)	874.39734	0.1094	24	20	4	23	19	5	(000)
849.00507	0.2276	23	18	5	22	17	6	(000)	874.93458	0.1001	19	8	12	18	5	13	(000)
849.63800	0.1830	19	7	13	18	4	14	(000)	879.79967	0.1237	25	16	9	24	15	10	(000)
849.89257	0.1712	25	13	12	24	12	13	(000)	883.07297	0.1501	15	7	8	14	4	11	(000)
850.81073	0.3964	19	5	14	18	4	15	(000)	883.31545	0.2428	21	6	15	20	5	16	(000)
850.97520	0.1951	16	4	13	15	1	14	(010)	886.04332	0.1060	25	17	8	24	16	9	(000)
851.24837	0.1203	15	2	13	14	1	14	(010)	887.40891	0.2749	20	7	14	19	4	15	(000)
851.25820	0.1511	23	21	2	22	20	3	(000)	894.63740	0.2332	20	8	13	19	5	14	(000)
851.34264	0.1973	23	19	4	22	18	5	(000)	902.29342	0.1563	17	3	14	16	2	15	(010)
851.62563	0.1010	17	8	10	16	* 5	11	(000)	903.53332	0.1797	18	5	14	17	2	15	(010)
852.11657	0.1721	23	20	3	22	19	4	(000)	904.62103	0.1895	23	7 1	6 2	2 6	17	(0	00)
852.41199	0.2496	24	15	9	23	14	10	(000)	904.67827	0.1510	20	5 1	5 1	9 4	16	(0)	00)
852.48943	0.1386	13	8	5	12	5	8	(000)	906.22841	0.2622	17	2 1	5 1	6 1	16	(0	00)
852.75443	0.1181	16	2	14	15	1	15	(000)	906.30980	0.1082	17	3 1	5 1	6 0	16	(0)	00)
852.91156	0.2601	16	3	14	15	0	15	(000)	906.75378	0.1767	15	8	7 1	4 5	10	(0	00)
858.54341	0.3281	17	3	14	16	2	15	(000)	910.05108	0.1386	16	3 1	4 1	5 0	15	(0)	10)
859.66033	0.1621	17	4	14	16	1	15	(000)	910.09980	0.1896	18	3 1	5 1	7 2	16	(0)	00)
859.91051	0.1718	18	4	14	17	3	15	(000)	910.71027	0.4266	18	4 1	5 1	7 1	16	(0)	00)
860.02730	0.1827	24	16	8	23	15	9	(000)	911.23428	0.4688	19	4 1	5 1	8 3	16	(0)	00)
860.83946	0.2427	18	8	11	17	5	12	(000)	913.98373	0.1064	13	9	4 1	2 6	7	(0	00)
861.96455	0.1274	25	14	11	24	13	12	(000)	914.60786	0.2387	19	5 1	5 1	8 2	16	(0	00)
864.04819	0.1623	18	6	14	17	3	14	(010)	918.82257	0.4750	20	6 1			16	(0)	00)
864.62304*	0.1437	24	24	0	23	23	1	(000)	919.87598	0.1199	21	8 1	4 2	0 5		,	00)
865.68669	0.3539	18	5	14	17	2	15	(000)	921.39854	0.1148	15	6	9 1	4 3	12	(0	00)
865.92987*	0.1614	24	17	7	23	16	8	(000)									

The strongest lines belong to the pure rotational spectrum of the ground (000) and first excited bending level (010). Except for some OH lines, the rest of the transitions belong to the pure rotational spectrum of higher excited vibrational levels such as (100), (020), (001) and to the rovibrational bands ν_2 , $2\nu_2 - \nu_2$. About 250 lines belong to the pure rotational spectrum of water in the ground and (010) vibrational states. These lines have an intensity between 0.7 and 0.3 in relative units (r.u.). At least half of the approximately 1000 lines with an intensity between 0.3 and 0.1 also belong to these states. The lowest observable intensity was about 0.003 relative units. Some of the more than 3000 lines with intensity between 0.1 and 0.003 r.u. also belong to the ground and (010) states, but the majority of them are due to the pure rotational transitions of higher vibrational states and rovibrational bands.

Lines with *J* rotational quantum numbers between 9 and 13, which are quite strong in room temperature spectra, suffered from self absorption caused by cooler water at the ends of the cell, atmospheric water and trace amounts of water in the spectrometer. Because of the Doppler effect, the high temperature emission lines are broader than the cooler absorption lines. The cooler water thus removes the line center

from the emission lines, resulting in doublet line artifacts. As a rule, the line centers of these doublets were found to be shifted to the right and to the left by approximately 0.005 cm⁻¹ from the line center of the room temperature transition. For completeness they are listed in Table 1 and the weaker components of such artifact doublets are marked by the letters "art."

A comprehensive analysis of the present spectrum would require a reliable calculation of the energy levels for the first few vibrational states up to very high values of the rotational quantum numbers. This could be done using variational calculations, since this approach allows the simultaneous calculation of energy levels and intensities of the lines. The only problem is the accuracy of the potential energy and dipole moment surfaces. Recently (34) a potential energy surface good enough to reliably predict highly excited rotational states has been obtained. Reliable dipole moment surfaces are also available (35). Work on the comprehensive assignment of all the lines in the present spectrum is underway.

Nevertheless, the assignment of the strongest lines of the spectrum, belonging to the ground and (010) states, can be achieved using both the available experimental energy levels

 $TABLE \ 2 \\ Energy \ Levels \ of \ Water \ (in \ cm^{-1}) \ Obtained \ from \ the \ Wavenumbers \ of \ the \ Hot \ Water \ Spectrum \ for \ the \ Ground \ Vibrational \ State$

15 11 4 4534.9521 c 17 14 3 6147.0802 c 19 13 6 6747.4635 c 21 8 13 6465.2352 d 15 12 4 4796.9684 c 17 15 3 6430.7181 c 19 14 6 7034.6919 c 21 9 13 6654.3026 d d 15 12 3 4796.9684 c 17 15 2 6430.7181 c 19 14 5 7034.6919 c 21 9 12 6660.321 d 15 13 3 5066.2239 c 17 16 1 6714.1226 c 19 15 4 7326.2778 c 21 10 11 6893.2154 d 15 15 15 3339.6727 c 17 17 0 6993.8964 c 19 16 4 7619.5258 c 21		- 0.				• ,														
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14	14	5	9	2983.3948		16	10	6	4665.9822		18	11	7	5750.8614		20	8	12	5966.8266	d
14	14	6	9	3084.8380		16	11	6	4919.2577	c	18	12	7	6019.1794	c	20	9	12	6167.7135	
14				3101.4312		16	11	5	4919.2577	С	18	12	6	6019.1794	С	20	9	11	6170.8317	
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14 13	14	12	3	4431.6326	c	17	0	17	2981.3630	a	18	18	1	7723.8438	c	20	15	6	7802.7130	c
14	14	12	2	4431.6326	c	17	1	17	2981.3631	a	18	18	0	7723.8438	c	20	15	5	7802.7130	c
14	14	13	2	4697.6622	c	17	1	16	3291.1485	a	19	0	19	3675.1160	a	20	16	5	8100.2790	c
14	14	13	1	4697.6622	с	17	2	16	3291.1522	a	19	1	19	3675.1160	a	20	16	4	8100.2790	С
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15	15	6	10	3443.1905		17	9	9	4830.6054		19	8	12	5481.1021	d	21	3	19	5163.0891	a
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TABLE 2—Continued

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23 5 19 6707.7192 d 24 16 8 10201.9933 c 23 5 18 6980.3242 d 24 17 8 10516.4286 c,d 23 6 18 6983.5764 d 24 17 7 10516.4286 c,d 23 6 17 7211.3501 d 24 18 7 10831.4770 c,d 23 7 16 24 19 6 10831.4770 c,d 23 7 16 24 19 6 11144.7979 c,d 23 8 16 24 19 5 11144.7979 c,d 23 8 15 7530.9297 d 24 20 5 11454.2127 c,d					d		c
23 5 18 6980.3242 d 24 17 8 10516.4286 c,d 23 6 18 6983.5764 d 24 17 7 10516.4286 c,d 23 6 17 7211.3501 d 24 18 7 10831.4770 c,d 23 7 16 24 18 6 10831.4770 c,d 23 8 16 24 19 6 11144.7979 c,d 23 8 15 7530.9297 d 24 20 5 11454.2127 c,d							c
23 6 18 6983.5764 d 24 17 7 10516.4286 c,d 23 6 17 7211.3501 d 24 18 7 10831.4770 c,d 23 7 17 24 18 6 10831.4770 c,d 23 7 16 24 19 6 11144.7979 c,d 23 8 16 24 19 5 1144.7979 c,d 23 8 15 7530.9297 d 24 20 5 11454.2127 c,d							
23 6 17 7211.3501 d 24 18 7 10831.4770 c,d 23 7 17 24 18 6 10831.4770 c,d 23 7 16 24 19 6 11144.7979 c,d 23 8 16 24 19 5 11144.7979 c,d 23 8 15 7530.9297 d 24 20 5 11454.2127 c,d							
23 7 17 24 18 6 10831.4770 c,d 23 7 16 24 19 6 11144.7979 c,d 23 8 16 24 19 5 11144.7979 c,d 23 8 15 7530.9297 d 24 20 5 11454.2127 c,d							
$\begin{array}{cccccccccccccccccccccccccccccccccccc$				7211.3501	d		
23 8 16 23 8 15 7530.9297 d 24 19 5 11144.7979 c,d 24 20 5 11454.2127 c,d							
23 8 15 7530.9297 d 24 20 5 11454.2127 c,d							
,				ww.o.o			_
		8	15	7530.9297	d	24 20 5 11454.2127	$^{\rm c,d}$
⁸ Energy levels taken from Def. 10. ^c Energy levels obtained using							

^a Energy levels taken from Ref. 10. ^c Energy levels obtained using unresolved doublets. ^d Energy levels which were not known previously.

(10, 11) and by extrapolation these levels and other available experimental data (1-11) using an effective Hamiltonian model.

In this work, the Padé–Borel (4, 24) method for the summation of the divergent perturbative effective Hamiltonian of the water molecule has been employed. A model with 30 parameters for fitting the available ground state data and 24 parameters for the (010) data was used. Recently very accurate fits to the available data on the ground and (010) states of water were achieved by Coudert (27). Accuracy close to the experimental error was obtained using a modified, exactly solvable model by Makarevicz (38) with 36 parameters for the (000) state and 31 for (010) state respectively. Our use of a 30 (000) and 24 (010) constant Padé-Borel model resulted in somewhat lower accuracy—up to a few wavenumbers for the calculation of the highest rotational quantum numbers levels obtained in this work. One of the reasons for this is the smaller number of fitted parameters. However, in most cases, this accuracy was enough for the assignment of the lines. Only in a few cases, involving the highest J and K_a quantum numbers, was the assignment ambiguous (see below). The transitions involving the rotational levels of the (000) and (010) vibrational states with rotational quantum numbers significantly higher than previously observed (10, 11) were assigned. In particular, transitions involving high J and high K_a ($\approx J$) were assigned up to J=25 for the (000) state and for J = 21 for the (010) state. Levels up to $K_a = 20 (000)$ and $K_a = 14 (010)$ were previously known (10, 11). Some higher J, K_a transitions were also assigned; however, their intensities are lower than 0.1 r.u. and their assignment is less certain.

The measured values of the most intense (down to 0.1 r.u.) transition frequencies, together with their rotational quantum numbers, are presented in Table 1. In a few cases, indicated by an asterisk, there is more than one candidate line for an assignment. More accurate variational calculations and/or the measurement of lines in other spectral regions will lead to the elimination of the ambiguity in the assignment as well as to the assignment of the weaker lines in the spectrum. The assignment of the lines of the (020) state and of the ν_2 vibration–rotation lines was made using the available literature data (22). These lines and the additional lines of the (000) and (010) levels weaker than 0.1 r.u. will be presented in a separate publication.

The water energy levels derived from the new transitions of Table 1 are presented in Tables 2 and 3. These energy levels were obtained from the observed lines by the addition of the lower state energy levels either taken from the work of Flaud *et al.* (10) for the 000 vibrational level and Camy-Peyret *et al.* (11) for the 010 level or, if available, from the energy levels obtained in this work. We do not list the lower J energy levels since they are

¹ The full line list is available from the Depository of Unpublished Data and from the authors on request.

TABLE 3 Energy Levels of Water (in ${\rm cm}^{-1}$) Obtained from the Wavenumbers of the Hot Water Spectrum for the Excited Vibrational State (010)

								LACITU	V IDI	ational 5	riai	(U	10)								_
J	K_a	K_c	Position		J	K_a	K_c	Position	n		\boldsymbol{J}	K_a	K_c	Position		J	K_a	K_c	Position	ı	
10	0	10	2144 5729	_	14	10	5	5721.1016	c	Ī	16	15	1	7928.9162	c,d	18	18	1	9695.4142	$_{\rm c,d}$	
12		12	3144.5732	a	14	10	4	5721.1016	c	1	16	16	1	8225.0796	$_{\mathrm{c,d}}$	18	18	0	9695.4142	$_{\rm c,d}$	
12	1	12	3144.5794	a	14	11	4	5996.6928	c	1	16	16	0	8225.0796	$_{\mathrm{c,d}}$	19	0	19	5241.7443	a	
12	1	11	3386.0534	\mathbf{a}	14	11	3	5996.6928	c	1	1.7	0	17	4554.6252	a	19	1	19	5241.7443	a	
12	2	11	3386.3820	a	14	12	3	6280.5557	c	1	١7	1	17	4554.6255	a	19	1	18	5632.0440	a	
12	2	10	3587.6693	a	14	12	2	6280.5557	c	1	17	1	16	4902.6111	a	19	2	18	5632.0475	a	
12	3	10	3592.4256		14	13	2	6569.4213	$_{\rm c,d}$	1	17	2	16	4902.6230	a	19	2	17	5970.9398	a	
12	3	9	3738.5437	a	14	13	1	6569.4213	$_{c,d}$	1	17	2	15	5204.7549	\mathbf{a}	19	3	17	5970.9971	a	
12	4	9	3770.8807		14	14	1	6859.8792	c	1	17	3	15	5204.9945	a	19	3	16			
12	4	8	3843.4063		14	14	0	6859.8792	c	1	17	3	14	5466.4074		19	4	16			
12	5	8	3940.5212		15	0	15	3937.5753	\mathbf{a}	1	17	4	14			19	4	15	6525.0581	a	
12	5	7	3959.2550		15	1	15	3937.5763	a		17	4	13	5680.5545		19		15			
12	6	7	4123.2876		15	1	14	4243.1121	a		17		13			19		14			
12	6	6	4125.5965		15	2	14	4243.1622	a		17		12	5835.2897	d	19	6	14			
12	7	6	4329.3258		15	2	13	4506.7671			17		12			19		13			
12	7	5	4329.4988		15	3	13	4507.5223	a		17		11	5965.7487		19		13			
12	8	5	4557.5457		15	3	12	4728.3037			17	7	11	6123.4382		19		12	7023.9820	d	
12	8	4	4557.5585		15	4	12				17	7	10	6134.9017		19		12	102010020	· ·	
12	9	4	4803.8217	c	15	4	11	4894.5909			17	8	10	6351.8983		19	8	11	7226.7763	d	
12	9	3	4803.8217	c	15	5	11	4938.2569			17	8	9	6353.4060	d	19		11	1220.1100	u	
12	10	3	5064.1338	С	15	5	10	5015.7065			17	9	9	6601.3836	d	19	9	10			
12	10	2	5064.1338	c	15	6	10	5132.5025					8						7729 0469	A	
12	11	2	5334.8713	c	15	6	9	5152.9667			L7	9		6601.5198	d	19		10	7738.0462	d	
12	11	1	5334.8713	c	15	7	9	5339.4931				10	8	6868.8823		19	10	9	7738.1306	d	
12	12	1	5612.4869	c				5342.3487				10	7	6868.8932	1	19	11	9	8022.2966	d	
12	12	0	5612.4869	c	15	7	8					11	7	7150.4108	c,d	19	11	8	8022.3013	d	
13	0	13	3391.1305	a	15	8	8	5568.0938				11	6	7150.4108	c,d	19	12	8	8318.4672	$_{\mathrm{c,d}}$	
13	1	13	3391.1349	a	15	8	7	5568.3568				12	6	7442.3418	c,d	19	12	7	8318.4672	c,d	
13	1	12	3654.0504	a	15	9	7	5817.0792			17	12	5	7442.3418	$_{\mathrm{c,d}}$	19	13	7	8623.2247	$_{\rm c,d}$	
13	2	12	3654.2181	a	15	9	6	5817.1001		1	17	13	5	7741.4383	$_{\mathrm{c,d}}$	19	13	6	8623.2247	$_{\mathrm{c,d}}$	
13	2	11	3877.0903	а	15	10	6	6082.4127	с	1	17	13	4	7741.4383	$_{\mathrm{c,d}}$	19	14	6	8933.5679	$^{\mathrm{c,d}}$	
13	3	11	3879.7199		15	10	5	6082.4127	С	1	17	14	4	8044.6529	$_{\mathrm{c,d}}$	19	14	5	8933.5679	$^{\mathrm{c,d}}$	
				a	15	11	5	6360.2002	c	1	17	14	3	8044.6529	$_{\mathrm{c,d}}$	19	15	5	9246.6922	$^{\mathrm{c,d}}$	
13	3	10	4052.8140		15	11	4	6360.2002	c	1	17	15	3	8348.9761	$^{\mathrm{c,d}}$	19	15	4	9246.6922	$^{\mathrm{c,d}}$	
13	4	10	4074.0401		15	12	4	6646.9566	c	1	17	15	2	8348.9761	$_{\mathrm{c,d}}$	19	16	4	9560.1332	$^{\mathrm{c,d}}$	
13	4	9	4174.0408		15	12	3	6646.9566	c	1	17	16	2	8651.8329	$_{\mathrm{c,d}}$	19	16	3	9560.1332	ϵ ,d	
13	5	9	4252.4526		15	13	3	6939.4274	$^{\mathrm{c,d}}$	1	17	16	1	8651.8329	c,d	19	17	3	9870.1257	$_{\rm c,d}$	
13	5	8	4285.6493		15	13	2	6939.4274	c,d	1	17	17	1	8947.4517	c,d	19	17	2	9870.1257	$_{c,d}$	
13	6	8		d	15	14	2	7234.4307	$^{\mathrm{c,d}}$	1	17	17	0	8947.4517	c,d	19	18	2	10174.2585	c,d	
13	6	7	4442.7148		15	14	1	7234.4307	$_{c,d}$		18	0	18	4889.4911	a	19	18	1	10174.2585	c,d	
13	7	7	4643.3781		15	15	1	7528.5561	$_{c,d}$		18	1	18	4889.4912	a	19	19	1	10465.5004	c,d	
13	7	6	4643.8731		15	15	0	7528.5561	c,d		18	1	17	5258.6278	a	19	19	0	10465.5004	c,d	
13	8	6	4871.9516		16	0	16	4237.3240	a		18	2	17	5258.6339	a	20	0	20	5611.3316	a	
13	8	5	4871.9807		16	1	16	4237.3245	a		18	2	16	5579.3641	a	20	1	20	5611.3316	a	
13	9	5	5119.3779	C	16	1	15	4564.0860	a		18	3	16	5579.4993	a	20		19	6022.7991	a	
13	9	4	5119.3779	c	16	2	15	4564.1140	a		18	3	15	0013.4330	a.	20		19	6022.8011	a	
13	10	4	5381.5546	c	16	2	14	4847.1880	a		18	4	15	5861.2418	2	20		18	6379.3666	a.	
13	10	3	5381.5546	c	16	3	14	4847.6264	u		18	4	14		a	20		18	6379.4036	a	
13	11	3	5654.7733	c	16	3	13	5089.3832	a					6095.5060	a	20		17	0019.4000	a	
13	11	2	5654.7733	c	16	4	13	5094.0873	а		8	5 5	14			$\frac{20}{20}$		17			
13	12	2	5935.6106	c		4		5280.0959			18		13	0000 4507	1	20		16			
13	12	1	5935.6106	c	16		12				18	6	13	6330.4537	d	20		16			
13	13	1	6220.6209	c	16	5	12	5310.2731			18	6	12	05150011	,	20		15			
13	13	0	6220.6209	c	16	5	11	EE19.0160			18	7	12	6547.0014	d	20		15			
14	0	14	3655.5180	a	16	6	11	5512.0169	,		18	7	11			20		14			
14	1	14	3655.5187	a	16	6	10	5546.7429	d		18	8	11	6775.7904	d	20		14			
14	1	13	3939.8912	a	16	7	10	5720.7217			18	8	10					13			
14	2	13	3939.9141	a	16	7	9	5726.6901		1	18	9	10	7025.1588	d	20					
14	2	12	4183.3919	a	16	8	9	5949.2186			18	9	9	7025.4749	d	20		13			
14	3	12	4184.8360	a	16	8	8	5949.8748		1	18	10	9	7293.3108	d	20		12			
					16	9	8	6198.5747		1	8	10	8	7293.3432	d	20		12			
14	3	11	4382.7831		16	9	7	6198.6277		1	18	11	8	7576.3363	\mathbf{d}	20		11			
14	4	11	4396.0533		16	10	7	6465.1300		1	18	11	7	7576.3395	d	20		11			
14	4	10	4525.2385		16	10	6	6465.1314		1	18	12	7	7870.5134	$_{\mathrm{c,d}}$	20		10			
14	5	10	4585.3512		16	11	6	6744.9009	c	1	8	12	6	7870.5134	$_{\mathrm{c,d}}$	20		10	8487.9069	d	
14	5	9	4638.3524		16	11	5	6744.9009	c	1	8	13	6	8172.5650	$_{\mathrm{c,d}}$	20	11	9	8487.9153	d	
14	6	9	4774.0460		16	12	5	7034.3553	c		8	13	5	8172.5650	$_{\mathrm{c,d}}$	20	12	9	8785.7941	$^{\mathrm{c,d}}$	
14	6	8	4785.0043		16	12	4	7034.3553	c		8	14	5	8479.4880	c,d	20	12	8	8785.7941	$^{\mathrm{c,d}}$	
14	7	8	4980.2250		16	13	4	7330.2524	$_{c,d}$		8	14	4	8479.4880	c,d	20	13	8	9092.9995	$^{\mathrm{c,d}}$	
14	7	7	4981.4710	d	16	13	3	7330.2524	$_{\rm c,d}$		8	15	4	8788.3794	c,d	20	13	7	9092.9995	$^{\mathrm{c,d}}$	
14	8	7	5208.8984		16	14	3	7629.4825	c,d		18	15	3	8788.3794	c,d	20	14	7	9406.7097	c,d	
14	8	6	5208.9907		16	14	2	7629.4825	c,d		18	16	3	9096.2919	c,d						
14	9	6	5457.2392		16	15	2		$_{\rm c,d}$		18	16	2	9096.2919	c,d						
14	9	5	5457.2464						,		8	17	2	9400.6200	c,d						
											18	17	1	9400.6200	c,d						
										1	_		-	2.20.0200							

TABLE 3—Continued

J	K_a	K	c Position	n	J	K_a	K_c	Position	ı
20	14	6	9406.7097	c,d	21	7	14		
20	15	6	9723.5135	$_{\mathrm{c,d}}$	21	8	14		
20	15	5	9723.5135	$^{\mathrm{c,d}}$	21	8	13		
20	16	5	10041.3674	$_{\rm c,d}$	21	9	13		
20	16	4	10041.3674	$^{\mathrm{c,d}}$	21	9	12		
20	17	4	10357.6841	$_{\rm c,d}$	21	10	12		
20	17	3	10357.6841	$_{\rm c,d}$	21	10	11		
20	18	3	10668.4944	$_{\mathrm{c,d}}$	21	11	11		
20	18	2	10668.4944	$_{\mathrm{c,d}}$	21	11	10		
20	19	2	10972.0620	$^{\mathrm{c,d}}$	21	12	10		
20	19	1	10972.0620	$^{\mathrm{c,d}}$	21	12	9	9271.1998	d
20	20	1			21	13	9	9581.4895	$^{\rm c,d}$
20	20	0			21	13	8	9581.4895	$^{\mathrm{c,d}}$
21	0	21	5998.1661	\mathbf{a}	21	14	8	9897.1895	$^{\rm c,d}$
21	1	21	5998.1661	\mathbf{a}	21	14	7	9897.1895	$^{\mathrm{c,d}}$
21	1	20	6430.7950	a	21	15	7	10218.6894	$^{\rm c,d}$
21	2	20	6430.7950	a	21	15	6	10218.6894	$^{\rm c,d}$
21	2	19	6804.4702	a	21	16	6	10539.6687	$^{\mathrm{c,d}}$
21	3	19	6804.4900	a	21	16	5	10539.6687	$_{\mathrm{c,d}}$
21	3	18			21	17	5	10861.2997	$^{\mathrm{c,d}}$
21	4	18			21	17	4	10861.2997	$^{\mathrm{c,d}}$
21	4	17			21	18	4	11179.9487	$^{\mathrm{c,d}}$
21	5	17			21	18	3	11179.9487	$^{\mathrm{c,d}}$
21	5	16			21	19	3	11490.6769	$^{\mathrm{c,d}}$
21	6	16			21	19	2	11490.6769	$^{\mathrm{c,d}}$
21	6	15			21	20	2		
21	7	15			21	20	1		

^a Energy levels taken from Ref. 11.

well known and, moreover, they are poorly determined from our emission spectra. For completeness we have added all known high J energy levels to Tables 2 and 3, even when our spectra do not provide new values. Note that possible confusion in the assignment of lines in Table 1 leads to ambiguities in the energy levels with the highest rotational quantum numbers of Tables 2 and 3.

IV. CONCLUSION

The emission spectrum of hot water (1550°C) was observed in the spectral region between 373 and 931 cm⁻¹. The wavenumbers of more than 4000 lines with an intensity from 0.7 down to 0.003 relative units were measured. More than 600 of the strongest lines belonging to the pure rotational spectrum in the ground (000) and first excited bending (010) vibrational levels were assigned. From the line positions, energy levels of the (000) and (010) states were derived with significantly higher rotational quantum numbers than previously available.

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^d Energy levels which were not known previously.

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