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# The structure of carbodiimide, HNCNH

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An experimentally determined  $r_s$ -type structure of HNCNH is reported:  $r_{\text{NH}} = 1.0074 \text{ \AA}$ ,  $r_{\text{CN}} = 1.2242 \text{ \AA}$ ,  $\angle \text{HNC} = 118.63^\circ$ ,  $\angle \text{NCN} = 170.63^\circ$ ,  $\angle \text{HN} \cdots \text{NH} = 88.99^\circ$ . The number of digits quoted allow for errors with two significant figures. In order to obtain these values we recorded rotational–torsional spectra of  $\text{HN}^{13}\text{CNH}$ ,  $\text{H}^{15}\text{NC}^{15}\text{NH}$  and  $\text{DNCND}$ , by using isotopically enriched cyanamide. A chemical equilibrium exists between carbodiimide, HNCNH, and the more stable isomer cyanamide,  $\text{H}_2\text{NCN}$ , which strongly favours cyanamide (approximately 1:115 at 110 °C). The expensive C- and N-substituted isotopomers could only be investigated in the millimetre wave region, while for  $\text{DNCND}$  the far infrared spectrum between  $10\text{--}350 \text{ cm}^{-1}$  was also recorded. Rotational constants of the three isotopomers, as well as of the parent species, were determined by fitting the assigned spectral transitions to the Watson Hamiltonian in  $S$  reduction. Using fitting programs written by Schwendeman and Rudolph,  $r_0$ ,  $r_s$  and  $r_m^p$  structures of HNCNH were derived. The experimentally determined structural parameters are compared with an *ab initio*  $r_e$  structure.

## 1. Introduction

Carbodiimide, HNCNH, belongs like  $\text{H}_2\text{O}_2$  and  $\text{H}_2\text{S}_2$  to the  $C_2$ -symmetry group and is a skew-chain molecule, as is shown in figure 1. Molecules of this type are accidentally nearly symmetric tops; for HNCNH Ray's asymmetry parameter  $\kappa$  was found to be  $-0.999\,995\,368(18)$  [1]. The permanent electric dipole moment coincides with the  $C_2$ -symmetry axis, which is perpendicular to the axis of the least principal moment of inertia. For HNCNH the  $C_2$  axis is the  $b$  axis, so that a perpendicular  $b$ -type rotational spectrum dominated by strong  $Q$ -branch absorptions is observed.

The H atoms of carbodiimide undergo internal rotation hindered by *cis* and *trans* potential barriers of nearly equal height, about  $2070 \text{ cm}^{-1}$ , leading to a torsional doublet splitting of all rovibrational transitions [1–5]. The large-amplitude torsional motion is accompanied by significant changes in the HNC and NCN angles, which effectively lower the torsional barriers. On the other side rotation about the  $a$  axis produces a centrifugal force which prevents the molecule from following the minimum energy torsional path, increasing the torsional potential barriers. Therefore the torsional

splitting depends on  $K_a$  and the doublet splitting collapses at high  $K_a$  [1–3]. For  $K_a = 0$  and 1 the torsional splitting depends also on the rotational quantum number  $J$ . Furthermore, HNCNH, like  $\text{H}_2\text{S}_2$ , shows an anomalous  $K_a$  doubling of the  $K_a = 2$  rotational torsional energy levels: the levels are inverted [4].

The goal of the present work was the determination of the molecular structure of HNCNH with spectroscopic methods. The result is compared with *ab initio* calculations performed at high levels of theory parallel to the experimental work. The *ab initio* calculations are reported in a separate article [6].

The experimental structure of HNCNH was not previously determined, due to the experimental difficulties presented by a chemical equilibrium between carbodiimide and cyanamide which is strongly dominated by cyanamide. Because cyanamide also undergoes a large amplitude inversional motion, the spectrum of this species is especially dense [7].

For the experimental evaluation of the structure, rotational–torsional spectra of  $\text{HN}^{13}\text{CNH}$ ,  $\text{H}^{15}\text{NC}^{15}\text{NH}$  and  $\text{DNCND}$  were measured in the millimetre wave region. Fourier transform infrared (FTIR) spectra of the first two expensive isotopomers could not be obtained, because the sample consumption for such measurements is too high. However, the rotational–

† Part of the author's dissertation, Justus-Liebig-Universität (D26).

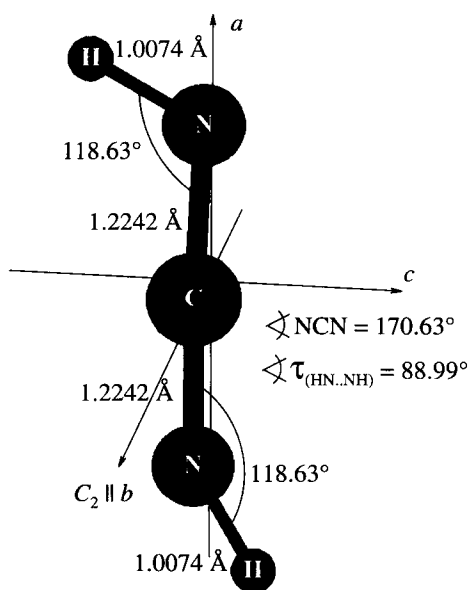


Figure 1. The carbodiimide molecule. The  $C_2$ -symmetry axis coincides with the  $b$  axis. The internuclear distances and bond angles of structure number 9 in table 8 (see section 5) are given. The errors of the distances are estimated to be in the range from 0.002 to 0.01 Å, and the errors of the bond angles in the range from 0.1 to 0.2° (see text).

torsional spectrum of DNCND was successfully recorded in the far infrared (FIR) region.

## 2. Experimental details

In order to determine the structure of HNCNH from experimental data, spectra were recorded with three different spectrometers.

### 2.1. Measurements with the Cologne millimetre/submillimetre wave/terahertz spectrometer

The essential elements of the Cologne spectrometer are high-frequency, broad-band tunable backward wave oscillators (BWOs) obtained from the ISTOK Research and Production Company, together with a subharmonic mixer, and a low-noise HEMT (high-electron mobility transistor) amplifier circuit for the intermediate frequency (IF) path. A tunable millimetre wave KVARZ synthesizer which covers the frequency region 78–118 GHz [8–10] is used to stabilize the BWO source frequency. In this study two BWOs were used, one with a frequency coverage from 258 to 375 GHz, the other one operating in the frequency region from 530 to 714 GHz. In the first region, we observed the  ${}^rQ_0$  branches and some  ${}^rP_0$  transitions of  $H^{15}NC^{15}NH$  and  $HN^{13}CNH$ , while in the second region the  ${}^rQ_1$  branch of DNCND was observed and measured.

The power output of the high frequency BWO is divided: a small fraction is used to drive the frequency

stabilization system by using suitable harmonics of the output frequency of a KVARZ synthesizer. The major portion of the BWO radiation power is focused through a free space absorption cell 4 m in length onto a He-cooled InSb hot-electron bolometer. Source modulation (modulation frequency = 7 kHz, modulation deviation = 480 kHz) with second derivative detection was employed. A sample pressure of 2 Pa was maintained in the cell, at room temperature, by pumping on a commercial sample of cyanamide (melting point: 46 °C) stabilized with  $NaH_2PO_4$ . Isotopically enriched samples of  $H^{15}NC^{15}N$  and  $H_2N^{13}CN$  were obtained from Icon Services Inc. with specified enrichment of 99%. Deutero-cyanamide was prepared according to a description given by Fletcher and Brown [11]: repeated solution of freshly sublimated  $H_2NCN$  in  $D_2O$  followed by evaporation under vacuum from  $D_2O$  over  $P_2O_5$  yielded  $D_2NCN$  of approximately 90% deuteration after six exchanges. Before dissolving the  $H_2NCN$  sample for the first time in  $D_2O$ , a small amount of  $KD_2PO_4$  was added to stabilize the deutero-cyanamide.

In the gaseous phase cyanamide is in equilibrium with carbodiimide. At room temperature only  $\approx 0.25\%$  of the sample exists as carbodiimide [12], so that the partial pressure of carbodiimide was of the order of 0.005 Pa. Consequently the weak spectrum of carbodiimide is interspersed among the strong absorption lines of cyanamide. The absolute accuracy of the line positions is better than 30 kHz for strong and unblended lines and 300 kHz for weak and partly blended lines.

### 2.2. Measurements with the millimetre wave spectrometer in Gießen

The AM-MSP spectrometer designed by the Analytik & Meßtechnik GmbH, Chemnitz, covers a frequency region between 52 and 179 GHz. The spectrometer consists of a millimetre wave synthesizer, a modulation unit and a receiver module. The millimetre wave power is generated by BWOs with output power levels between 1 and 25 mW. The radiation is guided through a 2.5 m glass absorption cell onto a detector. The detector consists of a silicon Schottky barrier diode at room temperature. Source modulation (modulation frequency = 200 kHz, modulation deviation = 300 kHz) was used. A detailed description of the spectrometer is given by Winnewisser *et al.* [13]. The accuracy of line positions is estimated to be  $\pm 10$  kHz for strong and unblended lines and  $\pm 100$  kHz for weak and partly blended lines.

In order to extend the available frequency range up to 600 GHz a frequency multiplier based on a planar Schottky diode was developed in the Cologne laboratory [14]. In this work the multiplier was used to generate second harmonic power to record the  ${}^rQ_0$  branch of DNCND at 202 GHz. The radiation was detected by

Table 1. Measurement parameters for the FIR spectra of D<sub>2</sub>NCN/DNCND.

Region /cm <sup>-1</sup>	1/(MOPD) /cm <sup>-1</sup>	Windows	Source	Detector	Beamsplitter	Opt. filter /cm <sup>-1</sup>	Aperture /mm	$v_{\text{mir}}^a$ /(cm/s)	Electr. filter /cm <sup>-1</sup>	No. of scans	Calib. gas [15]
9–45	0.0018	Polyethylene	Hg-Lamp	Si-bolom.	Mylar, 75 $\mu\text{m}$	0–100	8.0	1.266	39–237	93	H <sub>2</sub> O
20–100	0.0018	Polyethylene	Hg-Lamp	Si-bolom.	Mylar, 23 $\mu\text{m}$	0–100	4.0	1.266	39–237	20	H <sub>2</sub> O
90–350	0.0018	Polyethylene	Hg-Lamp	Si-bolom.	Mylar, 6 $\mu\text{m}$	0–370	2.5	1.266	39–474	75	H <sub>2</sub> O

<sup>a</sup> $b_{\text{mir}}$ : velocity of the mirror.

a liquid-helium-cooled InSb bolometer. The modulation frequency was 10 kHz and the modulation deviation 250 kHz.

### 2.3. Measurements with the high resolution FTIR spectrometer in Gießen

D<sub>2</sub>NCN/DNCND spectra were measured using a Bruker IFS 120 HR Fourier transform interferometer in three overlapping regions from 9 to 350 cm<sup>-1</sup>, operating with a resolution defined by 1/MOPD (maximum optical path difference) of 0.0018 cm<sup>-1</sup>. Table 1 gives the chosen instrumental parameters.

The sensitivity of a FTIR spectrometer is considerably less than the sensitivity of a millimetre wave spectrometer. Due to the low vapour pressure of cyanamide (16 Pa at 60°C [11]), we therefore raise the sample and the absorption cell temperatures above room temperature in order to obtain sufficient column density of H<sub>2</sub>NCN and HNCNH in the gaseous phase. The increased temperature also shifts the carbodiimide  $\rightleftharpoons$  cyanamide equilibrium to the HNCNH side ( $p(\text{HNCNH}) : p(\text{H}_2\text{NCN}) = 1 : 115$  at 110°C [12]). During the measurements the cyanamide sample was kept at  $\approx 100^\circ\text{C}$ , and the variable-temperature 3 m absorption cell, developed by Schermaul [16], was maintained at  $\approx 130^\circ\text{C}$ . However, cyanamide decomposes and/or polymerizes quite fast at these temperatures, so that it was necessary to work with a constant mass flow system which led to high sample consumption ( $\approx 60$  g D<sub>2</sub>NCN for the spectra reported here). The D<sub>2</sub>NCN spectra will be analysed separately.

### 3. Spectra, assignment and fits

The NH groups in carbodiimide rotate relative to one another, following a potential function with maxima when the dihedral angle  $\tau$  between the two hydrogens is either  $0^\circ$  (*cis* conformation) or  $180^\circ$  (*trans* conformation). The *cis* and *trans* maxima of the torsional potential function are nearly of equal height ( $V_{\text{cis}} = 2061$  and  $V_{\text{trans}} = 2078$  cm<sup>-1</sup> [3]). This is due in part to the flexing of the bond angles in HNCNH during the torsional motion. In order to classify the torsional rotational energy levels one can follow the group-theoretical treatment developed by Hougen [17] for H<sub>2</sub>X<sub>2</sub> molecules

having both a *cis* and *trans* feasible torsional tunnelling. Each rovibrational energy level is split in this case and the resulting levels can be represented by two independent doublet splittings. One doubling of levels results for the *cis* torsional tunnelling and the other doubling for the *trans* torsional tunnelling. The torsional energy levels can be labelled according to the  $G_4$ -extended molecular symmetry group [18], with  $A_{1s}$ ,  $A_{1d}$ ,  $A_{2d}$  and  $A_{2s}$ . The lowest energy level belongs to the symmetry species  $A_{1s}$  while the uppermost level has the symmetry designation  $A_{2s}$ . The relative order of the inner energy levels depends on the relative height of the *cis* and *trans* potential barriers and it was found that the  $A_{1d}$  energy level is below the  $A_{2d}$  level [1, 2]. The spacing between these two levels is small, because the difference between the *cis* and *trans* potential barriers is small. This is indicated schematically in figure 2.

According to the Pauli principle, the sign of the total wavefunction of the molecule HNCNH must change under the symmetry operation (12) (34), which is the permutation of both the hydrogen and nitrogen nuclei.

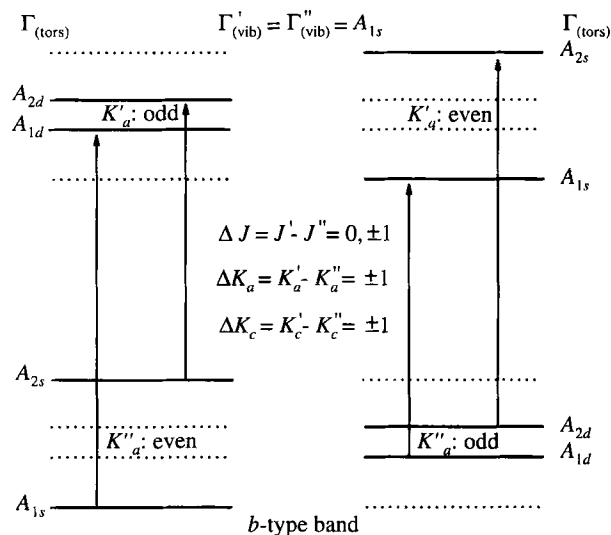


Figure 2. Torsional-rotational selection rules for the vibrational ground state of HNCNH. The levels entered as dotted lines do not exist due to the Pauli principle. The order of the energy levels  $A_{1d}$  and  $A_{2d}$  implies that  $V_{\text{cis}} < V_{\text{trans}}$ .

Table 2. Allowed combinations of symmetry species of the torsional ( $\Gamma_{\text{tors}}$ ) rotational ( $\Gamma_{\text{rot}}$ ), and nuclear spin wavefunctions ( $\Gamma_{\text{ns}}$ ) for HNCNH, HN<sup>13</sup>CNH, H<sup>15</sup>NC<sup>15</sup>NH, and DNCND in a vibrational state with  $A_{1s}$  symmetry.

$\Gamma_{\text{tors}}$	$\Gamma_{\text{rot}}$	$K_a$	$K_c$	Fermions			Bosons			
				HNCNH, HN <sup>13</sup> CNH			H <sup>15</sup> NC <sup>15</sup> NH, DNCND			
				$\Gamma_{\text{ns}}$	$\Gamma_{\text{tot}}$	Stat. weight	$\Gamma_{\text{ns}}$	$\Gamma_{\text{tot}}$	Stat. weight <sup>15</sup> N	Stat. weight D
$A_{1s}$	$A_{1s}$	even	even	$B_{2s}$	$B_{2s}$	15	$A_{1s}$	$A_{1s}$	10	45
$A_{1s}$	$B_{1s}$	even	odd	$A_{1s}$	$B_{1s}$	21	$B_{2s}$	$A_{2s}$	6	36
$A_{2s}$	$A_{1s}$	even	even	$B_{2s}$	$B_{1s}$	15	$A_{1s}$	$A_{2s}$	10	45
$A_{2s}$	$B_{1s}$	even	odd	$A_{1s}$	$B_{2s}$	21	$B_{2s}$	$A_{1s}$	6	36
$A_{1d}$	$A_{2d}$	odd	odd	$B_{2s}$	$B_{1s}$	15	$A_{1s}$	$A_{2s}$	10	45
$A_{1d}$	$B_{2d}$	odd	even	$A_{1s}$	$B_{2s}$	21	$B_{2s}$	$A_{1s}$	6	36
$A_{2d}$	$A_{2d}$	odd	odd	$B_{2s}$	$B_{2s}$	15	$A_{1s}$	$A_{1s}$	10	45
$A_{2d}$	$B_{2d}$	odd	even	$A_{1s}$	$B_{1s}$	21	$B_{2s}$	$A_{2s}$	6	36

The total wavefunction belongs therefore to the irreducible representations denoted by  $B$ . Furthermore the sign must not change by an internal rotation of  $2\pi$ , so that the total wavefunction belongs to the irreducible representations  $B_{1s}$  or  $B_{2s}$ . This constrains the possible combinations of torsional, rotational, and nuclear spin wavefunctions in a vibrational state with  $A_{1s}$  symmetry to those which are given in table 2 [2]. In consequence, for even  $K_a$  quantum numbers only the ‘outer’ torsional energy levels exist, labelled with  $A_{1s}$  and  $A_{2s}$ , while for odd  $K_a$  only the ‘inner’ energy levels, labelled with  $A_{1d}$  and  $A_{2d}$ , exist, as displayed in figure 2. We conclude that the torsional motion in carbodiimide leads to a splitting of each rovibrational state into a doublet. According to the rotational state of the molecule, the doublet exhibits either  $s$  or  $d$  symmetry.

For HNCNH (and HN<sup>13</sup>CNH) the number of totally symmetric nuclear wavefunctions ( $A_{1s}$ ) is 21 while the number of antisymmetric wavefunctions ( $B_{2s}$ ) is 15. The situation changes for H<sup>15</sup>NC<sup>15</sup>NH and for DNCND, because the total wavefunction preserves its sign under the symmetry operation (12) (34). This leads to total wavefunctions which belong to the representations labelled with  $A_{1s}$  or  $A_{2s}$ . For these two isotopomers the allowed combinations of torsional-rotational and nuclear spin wavefunctions in a vibrational state with  $A_{1s}$  symmetry are also summarized in table 2. H<sup>15</sup>NC<sup>15</sup>NH shows a ratio of symmetric to antisymmetric nuclear wavefunction of 10:6 while this ratio is equal to 45:36 for DNCND. This leads to nuclear spin statistics of 5:3 for H<sup>15</sup>NC<sup>15</sup>NH and of 5:4 for DNCND.

The permanent electric dipole moment of carbodiimide coincides with the  $b$  axis. Therefore a perpendicular  $b$ -type spectrum, obeying the selection rule  $\Delta K_a = +1$ , is observed. This means that pure rotational transitions are not allowed for HNCNH, since the tor-

sional state of the molecule must also change when a rotational transition occurs. This is illustrated in figure 2.

The characteristic features of the rotational torsional spectrum are equally spaced  $Q$  branches throughout the entire millimetre wave (MMW) and FIR region. The identification of these ‘ $Q_{K_a}$ ’ branches is the key for the assignment of carbodiimide lines in the forest of cyanamide absorptions. The ‘aesthetically pleasing’ ‘ $Q_0$ ’ branch of the H<sup>15</sup>NC<sup>15</sup>NH isotopomer is displayed in figure 3. Due to the torsional motion the branch is doubled, with transitions between  $A_{1d} \leftarrow A_{1s}$  and  $A_{2d} \leftarrow A_{2s}$  torsional energy levels. The origins of the two  $Q$  branches are separated by about 435 MHz. The spin statistics of 5:3 can be seen clearly. Transitions with higher  $J$  are shifted to lower frequencies. In this sense the ‘ $Q_0$ ’ branch of H<sup>15</sup>NC<sup>15</sup>NH is very similar to that of the main isotopic species [2]. In contrast, a dramatic change occurs in the shape of the ‘ $Q_0$ ’ branch upon going from these isotopomers to the HN<sup>13</sup>CNH species. This is also shown in figure 3. The ‘ $Q_0$ ’ branch of HN<sup>13</sup>CNH collapses and the  $J$  transitions are only resolved for  $J \geq 27$ . The recorded shape of the strong, unresolved absorption at the subband origin is determined by source modulation with second derivative detection. The torsional splitting is about 455 MHz for HN<sup>13</sup>CNH. A totally different pattern for the ‘ $Q_0$ ’ branch is exhibited by DNCND as is shown in the upper half of figure 4. Due to the deuterium mass, the torsional splitting is reduced by a factor of 100, while the spacing between adjacent  $J$  transitions is much larger so that the characteristic compact form of the  $Q$  branch is lost. Furthermore the transitions with higher  $J$  appear in this case at higher frequencies.

The changes in the shapes of the ‘ $Q_0$ ’ branches described above originate from the near inertial symmetry of the molecule. In this case an explicit expression

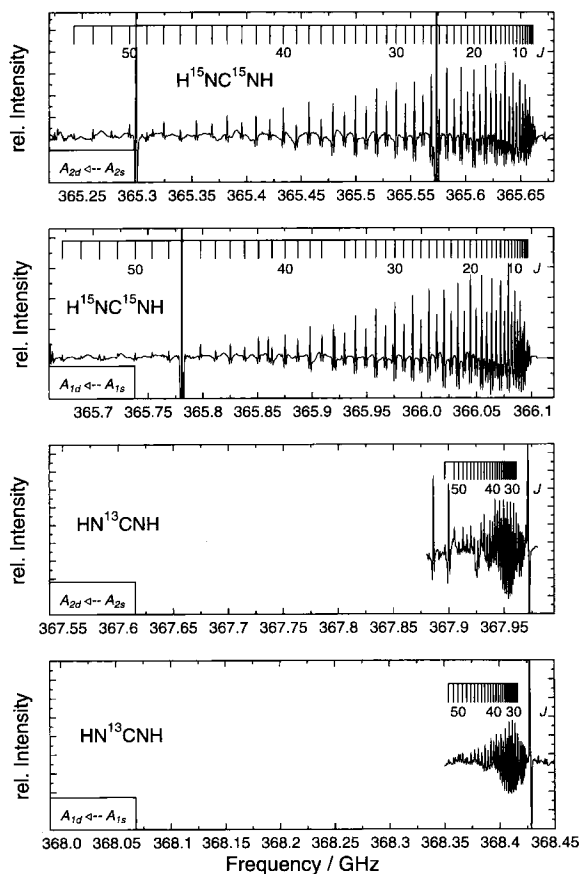


Figure 3.  $rQ_0$  branches of  $\text{H}^{15}\text{NC}^{15}\text{NH}$  and  $\text{HN}^{13}\text{CNH}$ . Due to the torsional motion each  $rQ_0$  branch is doubled. The subbranch at lower frequencies belongs to transitions between  $A_{2d} \leftarrow A_{2s}$  torsional energy levels, the subbranch at higher frequencies belongs to transitions between  $A_{1d} \leftarrow A_{1s}$  torsional energy levels. The assignment combs indicate the  $J$  quantum numbers.

for frequencies  $\nu$  of the  $rQ_0$  branch lines can be given [19]:

$$\nu \approx \left[ A - \frac{1}{2}(B + C) - D_K \right] + \left[ \frac{1}{4}(B - C) - D_{JK} \right] \times J(J + 1) + \dots \quad (1)$$

where  $A$ ,  $B$ , and  $C$  are the rotational constants, and  $D_K$  and  $D_{JK}$  are quartic centrifugal distortion constants.

In the case of  $\text{H}^{15}\text{NC}^{15}\text{NH}$ , which is the most inertially symmetric species of the investigated HNCNH isotopomers, the difference between the rotational constants  $B$  and  $C$  is so small that the  $J(J + 1)$  dependent term is dominated by the negative contribution of the large centrifugal distortion constant  $D_{JK}$ . For the  $\text{HN}^{13}\text{CNH}$  species the two contributions are nearly equal causing the unusually condensed appearance of the observed  $Q$  branch. For DNCND, however, which is the most asymmetric isotopomer, the pattern of the  $rQ_0$  branch is dominated by the inertial asymmetry con-

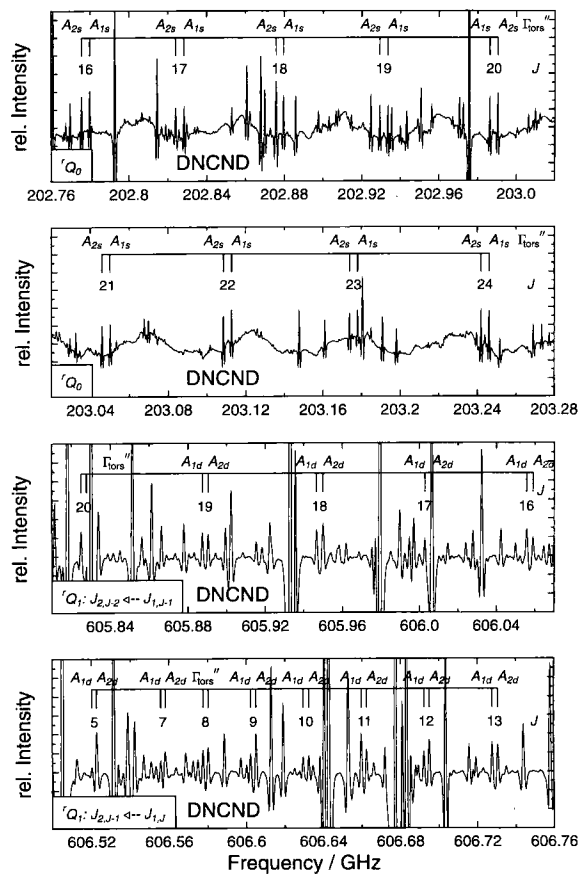


Figure 4. Part of the  $rQ_0$  and  $rQ_1$  branches of DNCND. Compared with the other carbodiimide isotopomers the torsional splitting is decreased by a factor of 100. The two components of the torsional doublet are labelled with the symmetry species  $\Gamma'' \in \{A_{1s}, A_{2s}\}$  of the lower torsional energy level. The  $J$  assignment is indicated.

tribution. The centrifugal distortion constant  $D_{JK}$  of DNCND is an order of magnitude smaller than for the other isotopomers (see below).

The  $J$  assignments of the  $Q$  branches were performed by using an interactive assignment program QBRASS [20], where the transition frequencies belonging to a particular subbranch were fitted to a power series (ps) expression in  $J(J + 1)$ :

$$\nu = \nu_c + \Delta B_{ps} J(J + 1) - \Delta D_{ps} J^2(J + 1)^2 \quad (2)$$

The criteria supporting the correct assignment were the minimal standard deviation together with the information drawn from the nuclear spin statistics. The assignments were then confirmed in each case by measuring  $rP_0$ - and  $rP_1$ -branch transitions in the MMW region, which could be identified among the cyanamide absorptions after the  $Q$  branches had been found. All measured transitions of the isotopomers  $\text{H}^{15}\text{NC}^{15}\text{NH}$  and

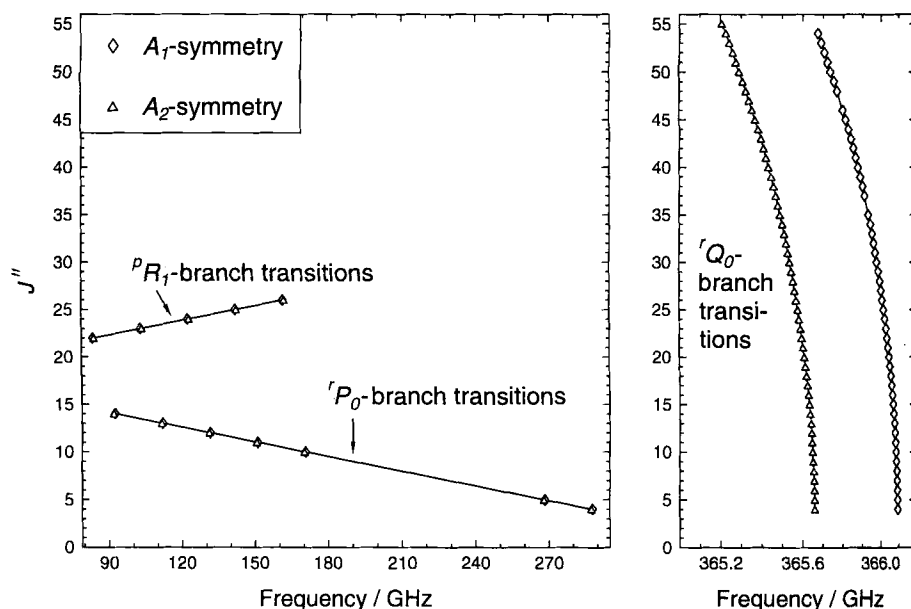


Figure 5. Fortrat diagrams of all measured  $\text{H}^{15}\text{NC}^{15}\text{NH}$  transitions.

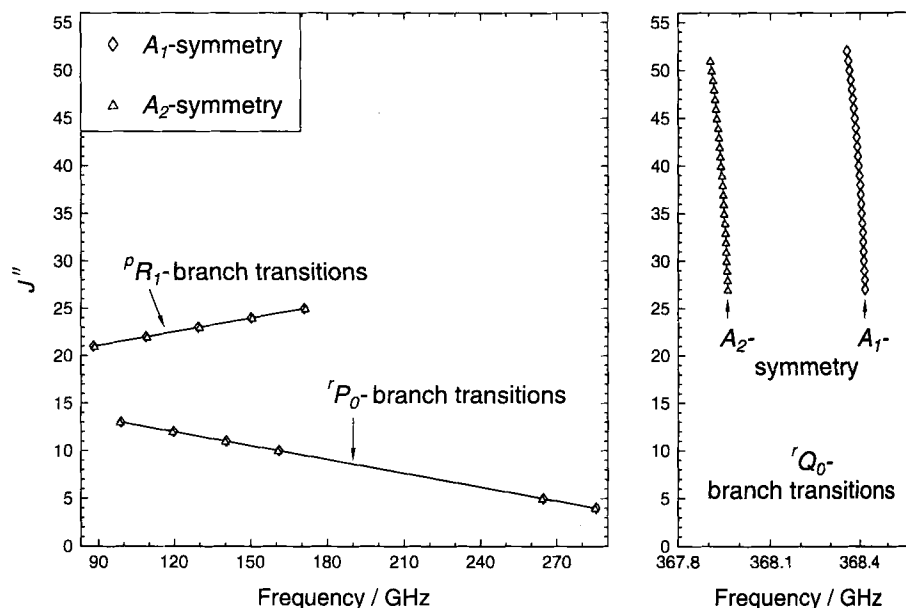


Figure 6. Fortrat diagrams of all measured  $\text{HN}^{13}\text{CNH}$  transitions.

$\text{HN}^{13}\text{CNH}$  are graphically presented in the Fortrat diagrams of figures 5 and 6, and the transition frequencies are listed in tables 3 and 4. Due to the overlapping of strong  $\text{D}_2\text{NCN}$  lines only two  $^{\text{P}}R_1$ -branch transitions could be measured for DNCND. However, additional information was gained by assigning many  $^{\text{r}}Q_1$ -branch transitions at 606.5 GHz. Parts of this branch are shown in the lower half of figure 4. This branch consists of four subbranches, due to the  $K_a$  doubling and the torsional doubling of the energy levels. The serious overlap with

$\text{D}_2\text{NCN}$  lines is obvious and prevented following the series of transitions to high  $J$  values. The Fortrat diagrams of the DNCND  $^{\text{r}}Q_0$  and  $^{\text{r}}Q_1$  branch transitions are given in figure 7. The transition frequencies are presented in table 5.

As mentioned above, the FIR spectrum of DNCND was recorded. The nearly equally spaced  $^{\text{r}}Q_{K_a}$  branches were followed up to  $K_a = 10$ . They are displayed in figure 8. A plot of the subband centres versus  $2K_a + 1$  is shown in figure 9. The almost linear relationship is

Table 3. Assigned transitions of  $\text{H}^{15}\text{NC}^{15}\text{NH}$  together with the centre frequencies of each torsional doublet. The centre frequencies were fitted to the Watson Hamiltonian in  $S$  reduction. The differences between the centre frequencies and those calculated with the constants of table 7 are also listed.

$J'_{K'_a, K'_c} \leftarrow J''_{K''_a, K''_c}$	$\Gamma'_{\text{tors}} \leftarrow \Gamma''_{\text{tors}}$	Measured transitions/MHz	Centre frequency /MHz	$\nu_{\text{cent}} - \nu_{\text{calc.}}/\text{MHz}$
$rQ_0$ -branch transitions				
$4_{1,3} \leftarrow 4_{0,4}$	$A_{1d} \leftarrow A_{1s}$	366 096.214	365 878.626	0.0167
	$A_{2d} \leftarrow A_{2s}$	365 661.037		
$5_{1,4} \leftarrow 5_{0,5}$	$A_{1d} \leftarrow A_{1s}$	366 094.905	365 877.299	−0.0175
	$A_{2d} \leftarrow A_{2s}$	365 659.693		
$6_{1,5} \leftarrow 6_{0,6}$	$A_{1d} \leftarrow A_{1s}$	366 093.373	365 875.750	−0.0136
	$A_{2d} \leftarrow A_{2s}$	365 658.127		
$7_{1,6} \leftarrow 7_{0,7}$	$A_{1d} \leftarrow A_{1s}$	366 091.582	365 873.934	−0.0159
	$A_{2d} \leftarrow A_{2s}$	365 656.285		
$8_{1,7} \leftarrow 8_{0,8}$	$A_{1d} \leftarrow A_{1s}$	366 089.555	365 871.867	−0.0074
	$A_{2d} \leftarrow A_{2s}$	365 654.179		
$9_{1,8} \leftarrow 9_{0,9}$	$A_{1d} \leftarrow A_{1s}$	366 087.256	365 869.528	−0.0079
	$A_{2d} \leftarrow A_{2s}$	365 651.800		
$10_{1,9} \leftarrow 10_{0,10}$	$A_{1d} \leftarrow A_{1s}$	366 084.755	365 866.959	0.0256
	$A_{2d} \leftarrow A_{2s}$	365 649.164		
$11_{1,10} \leftarrow 11_{0,11}$	$A_{1d} \leftarrow A_{1s}$	366 081.873	365 864.059	−0.0064
	$A_{2d} \leftarrow A_{2s}$	365 646.246		
$12_{1,11} \leftarrow 12_{0,12}$	$A_{1d} \leftarrow A_{1s}$	366 078.847	365 860.956	0.0255
	$A_{2d} \leftarrow A_{2s}$	365 643.066		
$13_{1,12} \leftarrow 13_{0,13}$	$A_{1d} \leftarrow A_{1s}$	366 075.436	365 857.525	−0.0020
	$A_{2d} \leftarrow A_{2s}$	365 639.615		
$14_{1,13} \leftarrow 14_{0,14}$	$A_{1d} \leftarrow A_{1s}$	366 071.818	365 853.853	−0.0002
	$A_{2d} \leftarrow A_{2s}$	365 635.887		
$15_{1,14} \leftarrow 15_{0,15}$	$A_{1d} \leftarrow A_{1s}$	366 067.930	365 849.907	−0.0002
	$A_{2d} \leftarrow A_{2s}$	365 631.884		
$16_{1,15} \leftarrow 16_{0,16}$	$A_{1d} \leftarrow A_{1s}$	366 063.781	365 845.687	0.0001
	$A_{2d} \leftarrow A_{2s}$	365 627.592		
$17_{1,16} \leftarrow 17_{0,17}$	$A_{1d} \leftarrow A_{1s}$	366 059.355	365 841.189	−0.0013
	$A_{2d} \leftarrow A_{2s}$	365 623.024		
$18_{1,17} \leftarrow 18_{0,18}$	$A_{1d} \leftarrow A_{1s}$	366 054.656	365 836.415	0.0000
	$A_{2d} \leftarrow A_{2s}$	365 618.173		
$19_{1,18} \leftarrow 19_{0,19}$	$A_{1d} \leftarrow A_{1s}$	366 049.676	365 831.358	−0.0005
	$A_{2d} \leftarrow A_{2s}$	365 613.040		
$20_{1,19} \leftarrow 20_{0,20}$	$A_{1d} \leftarrow A_{1s}$	366 044.415	365 826.018	−0.0003
	$A_{2d} \leftarrow A_{2s}$	365 607.620		
$21_{1,20} \leftarrow 21_{0,21}$	$A_{1d} \leftarrow A_{1s}$	366 038.877	365 820.394	0.0022
	$A_{2d} \leftarrow A_{2s}$	365 601.911		
$22_{1,21} \leftarrow 22_{0,22}$	$A_{1d} \leftarrow A_{1s}$	366 033.043	365 814.475	−0.0010
	$A_{2d} \leftarrow A_{2s}$	365 595.906		
$23_{1,22} \leftarrow 23_{0,23}$	$A_{1d} \leftarrow A_{1s}$	366 026.936	365 808.269	0.0010
	$A_{2d} \leftarrow A_{2s}$	365 589.603		

(continued)



Table 3 (*continued*)

$J'_{K'_a, K'_c} \leftarrow J''_{K''_a, K''_c}$	$\Gamma'_{\text{tors}} \leftarrow \Gamma''_{\text{tors}}$	Measured transitions/MHz	Centre frequency /MHz	$\nu_{\text{cent}} - \nu_{\text{calc.}}/\text{MHz}$
24 <sub>1,23</sub> $\leftarrow$ 24 <sub>0,24</sub>	A <sub>1d</sub> $\leftarrow$ A <sub>1s</sub> A <sub>2d</sub> $\leftarrow$ A <sub>2s</sub>	366 020.531 365 582.999	365 801.765	0.0003
25 <sub>1,24</sub> $\leftarrow$ 25 <sub>0,25</sub>	A <sub>1d</sub> $\leftarrow$ A <sub>1s</sub> A <sub>2d</sub> $\leftarrow$ A <sub>2s</sub>	366 013.834 365 576.131	365 794.982	0.0191
26 <sub>1,25</sub> $\leftarrow$ 26 <sub>0,26</sub>	A <sub>1d</sub> $\leftarrow$ A <sub>1s</sub> A <sub>2d</sub> $\leftarrow$ A <sub>2s</sub>	366 006.831 365 568.887	365 787.859	−0.0001
27 <sub>1,26</sub> $\leftarrow$ 27 <sub>0,27</sub>	A <sub>1d</sub> $\leftarrow$ A <sub>1s</sub> A <sub>2d</sub> $\leftarrow$ A <sub>2s</sub>	365 999.529 365 561.373	365 780.451	0.0011
28 <sub>1,27</sub> $\leftarrow$ 28 <sub>0,28</sub>	A <sub>1d</sub> $\leftarrow$ A <sub>1s</sub> A <sub>2d</sub> $\leftarrow$ A <sub>2s</sub>	365 991.922 365 553.535	365 772.728	−0.0035
29 <sub>1,28</sub> $\leftarrow$ 29 <sub>0,29</sub>	A <sub>1d</sub> $\leftarrow$ A <sub>1s</sub> A <sub>2d</sub> $\leftarrow$ A <sub>2s</sub>	365 984.009 365 545.387	365 764.698	−0.0024
30 <sub>1,29</sub> $\leftarrow$ 30 <sub>0,30</sub>	A <sub>1d</sub> $\leftarrow$ A <sub>1s</sub> A <sub>2d</sub> $\leftarrow$ A <sub>2s</sub>	365 975.791 365 536.925	365 756.358	0.0056
31 <sub>1,30</sub> $\leftarrow$ 31 <sub>0,31</sub>	A <sub>1d</sub> $\leftarrow$ A <sub>1s</sub> A <sub>2d</sub> $\leftarrow$ A <sub>2s</sub>	365 967.232 365 528.132	365 747.682	−0.0015
32 <sub>1,31</sub> $\leftarrow$ 32 <sub>0,32</sub>	A <sub>1d</sub> $\leftarrow$ A <sub>1s</sub> A <sub>2d</sub> $\leftarrow$ A <sub>2s</sub>	365 958.372 365 518.997	365 738.684	−0.0057
33 <sub>1,32</sub> $\leftarrow$ 33 <sub>0,33</sub>	A <sub>1d</sub> $\leftarrow$ A <sub>1s</sub> A <sub>2d</sub> $\leftarrow$ A <sub>2s</sub>	365 949.186 365 509.546	365 729.366	−0.0004
34 <sub>1,33</sub> $\leftarrow$ 34 <sub>0,34</sub>	A <sub>1d</sub> $\leftarrow$ A <sub>1s</sub> A <sub>2d</sub> $\leftarrow$ A <sub>2s</sub>	365 939.665 365 499.754	365 719.709	−0.0004
35 <sub>1,34</sub> $\leftarrow$ 35 <sub>0,35</sub>	A <sub>1d</sub> $\leftarrow$ A <sub>1s</sub> A <sub>2d</sub> $\leftarrow$ A <sub>2s</sub>	365 929.812 365 489.616	365 709.714	0.0001
37 <sub>1,36</sub> $\leftarrow$ 37 <sub>0,37</sub>	A <sub>1d</sub> $\leftarrow$ A <sub>1s</sub> A <sub>2d</sub> $\leftarrow$ A <sub>2s</sub>	365 909.078 365 468.290	365 688.684	−0.0047
38 <sub>1,37</sub> $\leftarrow$ 38 <sub>0,38</sub>	A <sub>1d</sub> $\leftarrow$ A <sub>1s</sub> A <sub>2d</sub> $\leftarrow$ A <sub>2s</sub>	365 898.189 365 457.107	365 677.648	−0.0011
39 <sub>1,38</sub> $\leftarrow$ 39 <sub>0,39</sub>	A <sub>1d</sub> $\leftarrow$ A <sub>1s</sub> A <sub>2d</sub> $\leftarrow$ A <sub>2s</sub>	365 886.942 365 445.559	365 666.250	−0.0013
40 <sub>1,39</sub> $\leftarrow$ 40 <sub>0,40</sub>	A <sub>1d</sub> $\leftarrow$ A <sub>1s</sub> A <sub>2d</sub> $\leftarrow$ A <sub>2s</sub>	365 875.359 365 433.647	365 654.503	0.0129
41 <sub>1,40</sub> $\leftarrow$ 41 <sub>0,41</sub>	A <sub>1d</sub> $\leftarrow$ A <sub>1s</sub> A <sub>2d</sub> $\leftarrow$ A <sub>2s</sub>	365 863.374 365 421.339	365 642.356	−0.0041
42 <sub>1,41</sub> $\leftarrow$ 42 <sub>0,42</sub>	A <sub>1d</sub> $\leftarrow$ A <sub>1s</sub> A <sub>2d</sub> $\leftarrow$ A <sub>2s</sub>	365 851.054 365 408.668	365 629.861	0.0052
43 <sub>1,42</sub> $\leftarrow$ 43 <sub>0,43</sub>	A <sub>1d</sub> $\leftarrow$ A <sub>1s</sub> A <sub>2d</sub> $\leftarrow$ A <sub>2s</sub>	365 838.324 365 395.615	365 616.970	−0.0014
44 <sub>1,43</sub> $\leftarrow$ 44 <sub>0,44</sub>	A <sub>1d</sub> $\leftarrow$ A <sub>1s</sub> A <sub>2d</sub> $\leftarrow$ A <sub>2s</sub>	365 825.236 365 382.160	365 603.698	−0.0032
45 <sub>1,44</sub> $\leftarrow$ 45 <sub>0,45</sub>	A <sub>1d</sub> $\leftarrow$ A <sub>1s</sub> A <sub>2d</sub> $\leftarrow$ A <sub>2s</sub>	365 811.764 365 368.320	365 590.042	0.0029
46 <sub>1,45</sub> $\leftarrow$ 46 <sub>0,46</sub>	A <sub>1d</sub> $\leftarrow$ A <sub>1s</sub> A <sub>2d</sub> $\leftarrow$ A <sub>2s</sub>	365 797.871 365 354.075	365 575.973	−0.0062

Table 3 (continued)

$J'_{K'_a, K'_c} \leftarrow J''_{K''_a, K''_c}$	$\Gamma'_{\text{tors}} \leftarrow \Gamma''_{\text{tors}}$	Measured transition/MHz	Centre frequency /MHz	$\nu_{\text{cent}} - \nu_{\text{calc.}}/\text{MHz}$
$48_{1,47} \leftarrow 48_{0,48}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	365 768.899 365 324.362	365 546.630	-0.0107
$49_{1,48} \leftarrow 49_{0,49}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	365 753.815 365 308.903	365 531.359	0.0097
$50_{1,49} \leftarrow 50_{0,50}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	365 738.297 365 292.970	365 515.634	-0.0003
$51_{1,50} \leftarrow 51_{0,51}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	365 722.398 365 276.630	365 499.514	0.0250
$52_{1,51} \leftarrow 52_{0,52}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	365 705.805 365 259.867	365 482.836	-0.0704
$53_{1,52} \leftarrow 53_{0,53}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	365 689.175 365 242.587	365 465.881	0.0014
$54_{1,53} \leftarrow 54_{0,54}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	365 671.927 365 224.956	365 448.442	0.0406
<sup>r</sup> P <sub>0</sub> -branch transitions				
$3_{1,3} \leftarrow 4_{0,4}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	287 920.869 287 485.969	287 703.419	-0.0023
$4_{1,4} \leftarrow 5_{0,5}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	268 374.353 267 939.600	268 156.976	-0.0068
$9_{1,9} \leftarrow 10_{0,10}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	170 633.187 170 199.421	170 416.304	0.0007
$11_{1,11} \leftarrow 12_{0,12}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	131 533.999 131 100.793	131 317.396	0.0153
$12_{1,12} \leftarrow 13_{0,13}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	111 984.067 111 551.180	111 767.624	-0.0093
$13_{1,13} \leftarrow 14_{0,14}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	92 434.067 92 001.515	92 217.791	0.0015
<sup>p</sup> R <sub>1</sub> -branch transitions				
$23_{0,23} \leftarrow 22_{1,22}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	83 509.214 83 937.567	83 723.391	0.0039
$24_{0,24} \leftarrow 23_{1,23}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	103 056.388 103 484.156	103 270.272	0.0029
$25_{0,25} \leftarrow 24_{1,24}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	122 602.901 123 030.017	122 816.459	-0.0005
$26_{0,26} \leftarrow 25_{1,25}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	142 148.617 142 575.150	142 361.884	-0.0017
$27_{0,27} \leftarrow 26_{1,26}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	161 693.541 162 119.404	161 906.473	-0.0021

indeed expected for an unperturbed perpendicular band system. The *Q* branches are rotationally resolved only for  $K_a \geq 9$ , while this is true for  $K_a \geq 4$  in the case of the main species. This is caused by a centrifugal distortion constant  $D_{JK}$  which is an order of magnitude

smaller for DNCND than for the main species (see below) [1].

After the identification of the *Q* branches, the transitions belonging to <sup>r</sup>R<sub>K<sub>a</sub></sub> branches were also found and assigned with the aid of a Loomis–Wood program [20].

Table 4. Assigned transitions of  $\text{HN}^{13}\text{CNH}$  together with the centre frequencies of each torsional doublet. The centre frequencies were fitted to the Watson Hamiltonian in  $S$  reduction. The differences between the centre frequencies and those calculated with the constants of table 7 are also listed.

$J'_{K'_a, K'_c} \leftarrow J''_{K''_a, K''_c}$	$\Gamma'_{\text{tors}} \leftarrow \Gamma''_{\text{tors}}$	Measured transitions/MHz	Centre frequency /MHz	$\nu_{\text{cent}} - \nu_{\text{calc.}}/\text{MHz}$
$^rQ_0$ -branch transitions				
$27_{1,26} \leftarrow 27_{0,27}$	$A_{1d} \leftarrow A_{1s}$	368 416.539	368 188.695	0.0129
	$A_{2d} \leftarrow A_{2s}$	367 960.850		
$28_{1,27} \leftarrow 28_{0,28}$	$A_{1d} \leftarrow A_{1s}$	368 415.422	368 187.595	0.0150
	$A_{2d} \leftarrow A_{2s}$	367 959.767		
$29_{1,28} \leftarrow 29_{0,29}$	$A_{1d} \leftarrow A_{1s}$	368 414.230	368 186.399	−0.0024
	$A_{2d} \leftarrow A_{2s}$	367 958.568		
$30_{1,29} \leftarrow 30_{0,30}$	$A_{1d} \leftarrow A_{1s}$	368 412.967	368 185.139	−0.0032
	$A_{2d} \leftarrow A_{2s}$	367 957.311		
$31_{1,30} \leftarrow 31_{0,31}$	$A_{1d} \leftarrow A_{1s}$	368 411.625	368 183.800	0.0016
	$A_{2d} \leftarrow A_{2s}$	367 955.975		
$32_{1,31} \leftarrow 32_{0,32}$	$A_{1d} \leftarrow A_{1s}$	368 410.166	368 182.356	−0.0098
	$A_{2d} \leftarrow A_{2s}$	367 954.546		
$33_{1,32} \leftarrow 33_{0,33}$	$A_{1d} \leftarrow A_{1s}$	368 408.634	368 180.835	−0.0051
	$A_{2d} \leftarrow A_{2s}$	367 953.035		
$34_{1,33} \leftarrow 34_{0,34}$	$A_{1d} \leftarrow A_{1s}$	368 406.993	368 179.213	−0.0039
	$A_{2d} \leftarrow A_{2s}$	367 951.433		
$35_{1,34} \leftarrow 35_{0,35}$	$A_{1d} \leftarrow A_{1s}$	368 405.244	368 177.489	−0.0024
	$A_{2d} \leftarrow A_{2s}$	367 949.733		
$36_{1,35} \leftarrow 36_{0,36}$	$A_{1d} \leftarrow A_{1s}$	368 403.389	368 175.657	−0.0021
	$A_{2d} \leftarrow A_{2s}$	367 947.924		
$37_{1,36} \leftarrow 37_{0,37}$	$A_{1d} \leftarrow A_{1s}$	368 401.401	368 173.709	−0.0060
	$A_{2d} \leftarrow A_{2s}$	367 946.017		
$38_{1,37} \leftarrow 38_{0,38}$	$A_{1d} \leftarrow A_{1s}$	368 399.292	368 171.640	−0.0141
	$A_{2d} \leftarrow A_{2s}$	367 943.988		
$39_{1,38} \leftarrow 39_{0,39}$	$A_{1d} \leftarrow A_{1s}$	368 397.081	368 169.472	0.0008
	$A_{2d} \leftarrow A_{2s}$	367 941.863		
$40_{1,39} \leftarrow 40_{0,40}$	$A_{1d} \leftarrow A_{1s}$	368 394.724	368 167.170	0.0088
	$A_{2d} \leftarrow A_{2s}$	367 939.615		
$41_{1,40} \leftarrow 41_{0,41}$	$A_{1d} \leftarrow A_{1s}$	368 392.222	368 164.719	0.0004
	$A_{2d} \leftarrow A_{2s}$	367 937.215		
$42_{1,41} \leftarrow 42_{0,42}$	$A_{1d} \leftarrow A_{1s}$	368 389.574	368 162.129	−0.0088
	$A_{2d} \leftarrow A_{2s}$	367 934.684		
$43_{1,42} \leftarrow 43_{0,43}$	$A_{1d} \leftarrow A_{1s}$	368 386.815	368 159.416	0.0028
	$A_{2d} \leftarrow A_{2s}$	367 932.017		
$44_{1,43} \leftarrow 44_{0,44}$	$A_{1d} \leftarrow A_{1s}$	368 383.817	368 156.546	0.0071
	$A_{2d} \leftarrow A_{2s}$	367 929.274		
$45_{1,44} \leftarrow 45_{0,45}$	$A_{1d} \leftarrow A_{1s}$	368 380.724	368 153.528	0.0190
	$A_{2d} \leftarrow A_{2s}$	367 926.331		
$47_{1,46} \leftarrow 47_{0,47}$	$A_{1d} \leftarrow A_{1s}$	368 374.001	368 146.972	0.0140
	$A_{2d} \leftarrow A_{2s}$	367 919.942		
$48_{1,47} \leftarrow 48_{0,48}$	$A_{1d} \leftarrow A_{1s}$	368 370.364	368 143.408	−0.0162
	$A_{2d} \leftarrow A_{2s}$	367 916.452		

Table 4 (continued)

$J'_{K'_a, K'_c} \leftarrow J''_{K''_a, K''_c}$	$\Gamma'_{\text{tors}} \leftarrow \Gamma''_{\text{tors}}$	Measured transitions/MHz	Centre frequency /MHz	$\nu_{\text{cent}} - \nu_{\text{calc.}}/\text{MHz}$
$49_{1,48} \leftarrow 49_{0,49}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	368 366.506 367 912.885	368 139.696	-0.0137
$50_{1,49} \leftarrow 50_{0,50}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	368 362.537 367 909.124	368 135.831	0.0233
$51_{1,50} \leftarrow 51_{0,51}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	368 358.272 367 905.119	368 131.696	-0.0156
$rP_0$ -branch transitions				
$3_{1,2} \leftarrow 4_{0,4}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	285 487.205 285 031.414	285 259.310	0.0045
$4_{1,4} \leftarrow 5_{0,5}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	264 749.249 264 293.733	264 521.491	-0.0130
$11_{1,11} \leftarrow 12_{0,12}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	119 563.478 119 109.604	119 336.541	0.0127
$12_{1,12} \leftarrow 13_{0,13}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	98 820.604 98 367.099	98 593.852	-0.0032
$rR_1$ -branch transitions				
$22_{0,22} \leftarrow 21_{1,21}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	87 867.496 88 316.522	88 092.009	0.0090
$23_{0,23} \leftarrow 22_{1,22}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	108 609.407 109 057.802	108 833.604	0.0011
$24_{0,24} \leftarrow 23_{1,23}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	129 350.763 129 798.539	129 574.651	-0.0053
$25_{0,25} \leftarrow 24_{1,24}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	150 091.563 150 538.597	150 315.080	0.0007
$26_{0,26} \leftarrow 25_{1,25}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	170 831.626 171 277.952	170 054.789	-0.0018

Figure 7. Fortrat diagrams of the measured DNCND  $Q$ -branch transitions in the MMW region. Due to the decreased torsional splitting and the higher asymmetry of DNCND compared with the other isotopomers, the torsional splitting is not resolved.

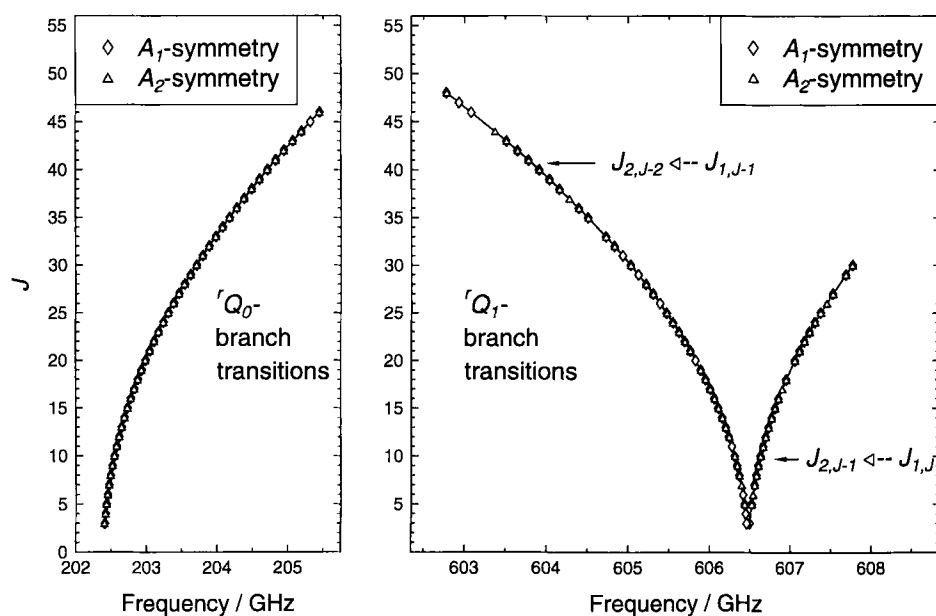


Table 5. Assigned transitions of DNCND in the MMW and SubMMW region. The centre frequencies of the torsional doublets were fitted to the Watson Hamiltonian in  $S$  reduction together with the observed IR transition wavenumbers given in table 6. The differences between the centre frequencies and those calculated with the constants of table 7 are also listed.

$J'_{K'_a, K'_c} \leftarrow J''_{K''_a, K''_c}$	$\Gamma'_{\text{tors}} \leftarrow \Gamma''_{\text{tors}}$	Measured transition/MHz	Centre frequency/MHz	$\nu_{\text{cent}} - \nu_{\text{calc.}}/\text{MHz}$
$^rQ_0$ -branch transitions				
$3_{1,2} \leftarrow 3_{0,3}$	$A_{1d} \leftarrow A_{1s}$	202 409.184	202 407.127	0.0645
	$A_{2d} \leftarrow A_{2s}$	202 405.070		
$4_{1,3} \leftarrow 4_{0,4}$	$A_{1d} \leftarrow A_{1s}$	202 420.608	202 418.508	0.0260
	$A_{2d} \leftarrow A_{2s}$	202 416.409		
$5_{1,4} \leftarrow 5_{0,5}$	$A_{1d} \leftarrow A_{1s}$	202 434.526	202 432.637	-0.1180
	$A_{2d} \leftarrow A_{2s}$	202 430.748		
$6_{1,5} \leftarrow 6_{0,6}$	$A_{1d} \leftarrow A_{1s}$	202 451.984	202 449.936	0.0550
	$A_{2d} \leftarrow A_{2s}$	202 447.889		
$7_{1,6} \leftarrow 7_{0,7}$	$A_{1d} \leftarrow A_{1s}$	202 471.634	202 469.742	-0.1170
	$A_{2d} \leftarrow A_{2s}$	202 467.850		
$8_{1,7} \leftarrow 8_{0,8}$	$A_{1d} \leftarrow A_{1s}$	202 494.810	202 492.741	0.0539
	$A_{2d} \leftarrow A_{2s}$	202 490.672		
$9_{1,8} \leftarrow 9_{0,9}$	$A_{1d} \leftarrow A_{1s}$	202 520.474	202 518.384	0.0192
	$A_{2d} \leftarrow A_{2s}$	202 516.295		
$10_{1,9} \leftarrow 10_{0,10}$	$A_{1d} \leftarrow A_{1s}$	202 549.005	202 546.887	-0.0034
	$A_{2d} \leftarrow A_{2s}$	202 544.769		
$11_{1,10} \leftarrow 11_{0,11}$	$A_{1d} \leftarrow A_{1s}$	202 580.382	202 578.270	0.0078
	$A_{2d} \leftarrow A_{2s}$	202 576.159		
$12_{1,11} \leftarrow 12_{0,12}$	$A_{1d} \leftarrow A_{1s}$	202 614.612	202 612.529	0.0506
	$A_{2d} \leftarrow A_{2s}$	202 610.447		
$13_{1,12} \leftarrow 13_{0,13}$	$A_{1d} \leftarrow A_{1s}$	202 651.656	202 649.580	0.0430
	$A_{2d} \leftarrow A_{2s}$	202 647.505		
$14_{1,13} \leftarrow 14_{0,14}$	$A_{1d} \leftarrow A_{1s}$	202 691.557	202 689.475	0.0391
	$A_{2d} \leftarrow A_{2s}$	202 687.392		
$15_{1,14} \leftarrow 15_{0,15}$	$A_{1d} \leftarrow A_{1s}$	202 734.293	202 732.212	0.0393
	$A_{2d} \leftarrow A_{2s}$	202 730.131		
$16_{1,15} \leftarrow 16_{0,16}$	$A_{1d} \leftarrow A_{1s}$	202 779.860	202 777.787	0.0421
	$A_{2d} \leftarrow A_{2s}$	202 775.713		
$17_{1,16} \leftarrow 17_{0,17}$	$A_{1d} \leftarrow A_{1s}$	202 828.275	202 826.140	-0.0100
	$A_{2d} \leftarrow A_{2s}$	202 824.005		
$19_{1,18} \leftarrow 19_{0,19}$	$A_{1d} \leftarrow A_{1s}$	202 933.571	202 931.483	0.0357
	$A_{2d} \leftarrow A_{2s}$	202 929.395		
$20_{1,19} \leftarrow 20_{0,20}$	$A_{1d} \leftarrow A_{1s}$	202 990.460	202 988.362	0.0287
	$A_{2d} \leftarrow A_{2s}$	202 986.265		
$21_{1,20} \leftarrow 21_{0,21}$	$A_{1d} \leftarrow A_{1s}$	203 050.160	203 048.064	0.0241
	$A_{2d} \leftarrow A_{2s}$	203 045.967		
$22_{1,21} \leftarrow 22_{0,22}$	$A_{1d} \leftarrow A_{1s}$	203 112.682	203 110.579	0.0154
	$A_{2d} \leftarrow A_{2s}$	203 108.475		
$23_{1,22} \leftarrow 23_{0,23}$	$A_{1d} \leftarrow A_{1s}$	203 178.045	203 175.921	0.0202
	$A_{2d} \leftarrow A_{2s}$	203 173.797		
$24_{1,23} \leftarrow 24_{0,24}$	$A_{1d} \leftarrow A_{1s}$	203 246.184	203 244.074	0.0264
	$A_{2d} \leftarrow A_{2s}$	203 241.964		
$25_{1,24} \leftarrow 25_{0,25}$	$A_{1d} \leftarrow A_{1s}$	203 317.132	203 315.014	0.0140
	$A_{2d} \leftarrow A_{2s}$	203 312.895		

Table 5 (continued)

$J'_{K'_a, K'_c} \leftarrow J''_{K''_a, K''_c}$	$\Gamma'_{\text{tors}} \leftarrow \Gamma''_{\text{tors}}$	Measured transition/MHz	Centre frequency/MHz	$\nu_{\text{cent}} - \nu_{\text{calc.}}/\text{MHz}$
$26_{1,25} \leftarrow 26_{0,26}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	203 390.836 203 386.628	203 388.732	-0.0220
$27_{1,26} \leftarrow 27_{0,27}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	203 467.455 203 463.178	203 465.317	0.0118
$28_{1,27} \leftarrow 28_{0,28}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	203 546.774 203 542.505	203 544.640	-0.0091
$29_{1,28} \leftarrow 29_{0,29}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	203 628.895 203 624.658	203 626.777	-0.0040
$30_{1,29} \leftarrow 30_{0,30}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	203 713.832 203 709.531	203 711.682	-0.0141
$31_{1,30} \leftarrow 31_{0,31}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	203 801.534 203 797.229	203 799.381	-0.0085
$32_{1,31} \leftarrow 32_{0,32}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	203 892.019 203 887.690	203 889.855	-0.0010
$33_{1,32} \leftarrow 33_{0,33}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	203 985.232 203 980.907	203 983.070	-0.0202
$34_{1,33} \leftarrow 34_{0,34}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	204 081.212 204 076.883	204 079.047	-0.0395
$35_{1,34} \leftarrow 35_{0,35}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	204 179.992 204 175.651	204 177.822	-0.0175
$36_{1,35} \leftarrow 36_{0,36}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	204 281.507 204 277.136	204 279.321	-0.0221
$37_{1,36} \leftarrow 37_{0,37}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	204 385.750 204 381.378	204 383.564	-0.0273
$38_{1,37} \leftarrow 38_{0,38}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	204 492.743 204 488.371	204 490.557	-0.0210
$39_{1,38} \leftarrow 39_{0,39}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	204 602.487 204 598.071	204 600.279	-0.0177
$40_{1,39} \leftarrow 40_{0,40}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	204 714.941 204 710.514	204 712.727	-0.0139
$41_{1,40} \leftarrow 41_{0,41}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	204 830.105 204 825.669	204 827.887	-0.0169
$42_{1,41} \leftarrow 42_{0,42}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	204 948.013 204 943.545	204 945.779	0.0002
$43_{1,42} \leftarrow 43_{0,43}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	205 068.573 205 064.162	205 066.367	0.0086
$44_{1,43} \leftarrow 44_{0,44}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	205 191.927 205 187.398	205 189.663	0.0275
$46_{1,45} \leftarrow 46_{0,46}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	205 446.548 205 442.051	205 444.300	0.0477
<sup>p</sup> R <sub>1</sub> -branch transitions				
$19_{0,19} \leftarrow 18_{1,18}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	142 682.381 142 686.421	142 684.401	0.0314
$20_{0,20} \leftarrow 19_{1,19}$	$A_{1d} \leftarrow A_{1s}$ $A_{2d} \leftarrow A_{2s}$	160 864.911 160 869.323	160 867.117	-0.0200

(continued)

Table 5 (*continued*)

$J'_{K'_a, K'_c} \leftarrow J''_{K''_a, K''_c}$	$\Gamma'_{\text{tors}} \leftarrow \Gamma''_{\text{tors}}$	Measured transition/MHz	Centre frequency/MHz	$\nu_{\text{cent}} - \nu_{\text{calc.}}/\text{MHz}$
<sup>r</sup> Q <sub>1</sub> -branch transitions				
5 <sub>2,3</sub> ← 5 <sub>1,4</sub>	A <sub>1s</sub> ← A <sub>1d</sub>	606 432.652	606 434.222	0.0582
	A <sub>2s</sub> ← A <sub>2d</sub>	606 435.792		
8 <sub>2,6</sub> ← 8 <sub>1,7</sub>	A <sub>1s</sub> ← A <sub>1d</sub>	606 367.438	606 368.891	0.0159
	A <sub>2s</sub> ← A <sub>2d</sub>	606 370.344		
10 <sub>2,8</sub> ← 10 <sub>1,9</sub>	A <sub>1s</sub> ← A <sub>1d</sub>	606 308.279	606 309.911	0.1370
	A <sub>2s</sub> ← A <sub>2d</sub>	606 311.543		
12 <sub>2,10</sub> ← 12 <sub>1,11</sub>	A <sub>1s</sub> ← A <sub>1d</sub>	606 236.241	606 238.189	−0.0045
	A <sub>2s</sub> ← A <sub>2d</sub>	606 240.136		
13 <sub>2,11</sub> ← 13 <sub>1,12</sub>	A <sub>1s</sub> ← A <sub>1d</sub>	606 196.130	606 197.528	−0.1890
	A <sub>2s</sub> ← A <sub>2d</sub>	606 198.925		
14 <sub>2,12</sub> ← 14 <sub>1,13</sub>	A <sub>1s</sub> ← A <sub>1d</sub>	606 152.515	606 154.143	0.0315
	A <sub>2s</sub> ← A <sub>2d</sub>	606 155.770		
16 <sub>2,14</sub> ← 16 <sub>1,15</sub>	A <sub>1s</sub> ← A <sub>1d</sub>	606 055.701	606 057.341	−0.1630
	A <sub>2s</sub> ← A <sub>2d</sub>	606 058.981		
17 <sub>2,15</sub> ← 17 <sub>1,16</sub>	A <sub>1s</sub> ← A <sub>1d</sub>	606 002.820	606 004.655	0.1600
	A <sub>2s</sub> ← A <sub>2d</sub>	606 006.490		
18 <sub>2,16</sub> ← 18 <sub>1,17</sub>	A <sub>1s</sub> ← A <sub>1d</sub>	605 946.695	605 948.354	0.0100
	A <sub>2s</sub> ← A <sub>2d</sub>	605 950.013		
19 <sub>2,17</sub> ← 19 <sub>1,18</sub>	A <sub>1s</sub> ← A <sub>1d</sub>	605 887.453	605 888.958	−0.0894
	A <sub>2s</sub> ← A <sub>2d</sub>	605 890.463		
21 <sub>2,19</sub> ← 21 <sub>1,20</sub>	A <sub>1s</sub> ← A <sub>1d</sub>	605 759.490	605 760.946	0.0552
	A <sub>2s</sub> ← A <sub>2d</sub>	605 762.403		
23 <sub>2,21</sub> ← 23 <sub>1,22</sub>	A <sub>1s</sub> ← A <sub>1d</sub>	605 618.695	605 620.388	0.0653
	A <sub>2s</sub> ← A <sub>2d</sub>	605 622.081		
24 <sub>2,22</sub> ← 24 <sub>1,23</sub>	A <sub>1s</sub> ← A <sub>1d</sub>	605 543.626	605 545.166	−0.0694
	A <sub>2s</sub> ← A <sub>2d</sub>	605 546.706		
25 <sub>2,23</sub> ← 25 <sub>1,24</sub>	A <sub>1s</sub> ← A <sub>1d</sub>	605 465.378	605 466.925	−0.0515
	A <sub>2s</sub> ← A <sub>2d</sub>	605 468.471		
27 <sub>2,25</sub> ← 27 <sub>1,26</sub>	A <sub>1s</sub> ← A <sub>1d</sub>	605 299.304	605 300.870	−0.0554
	A <sub>2s</sub> ← A <sub>2d</sub>	605 302.436		
28 <sub>2,26</sub> ← 28 <sub>1,27</sub>	A <sub>1s</sub> ← A <sub>1d</sub>	605 211.446	605 213.114	−0.0095
	A <sub>2s</sub> ← A <sub>2d</sub>	605 214.782		
30 <sub>2,28</sub> ← 30 <sub>1,29</sub>	A <sub>1s</sub> ← A <sub>1d</sub>	605 026.271	605 027.951	0.0080
	A <sub>2s</sub> ← A <sub>2d</sub>	605 029.630		
32 <sub>2,30</sub> ← 32 <sub>1,31</sub>	A <sub>1s</sub> ← A <sub>1d</sub>	604 828.383	604 829.984	0.0234
	A <sub>2s</sub> ← A <sub>2d</sub>	604 831.586		
33 <sub>2,31</sub> ← 33 <sub>1,32</sub>	A <sub>1s</sub> ← A <sub>1d</sub>	604 724.559	604 726.170	0.0136
	A <sub>2s</sub> ← A <sub>2d</sub>	604 727.780		
35 <sub>2,33</sub> ← 35 <sub>1,34</sub>	A <sub>1s</sub> ← A <sub>1d</sub>	604 507.264	604 508.910	0.0124
	A <sub>2s</sub> ← A <sub>2d</sub>	604 510.556		
36 <sub>2,34</sub> ← 36 <sub>1,35</sub>	A <sub>1s</sub> ← A <sub>1d</sub>	604 393.892	604 395.487	0.0535
	A <sub>2s</sub> ← A <sub>2d</sub>	604 397.082		

Table 5 (continued)

$J'_{K'_a, K'_c} \leftarrow J''_{K''_a, K''_c}$	$\Gamma'_{\text{tors}} \leftarrow \Gamma''_{\text{tors}}$	Measured transition/MHz	Centre frequency/MHz	$\nu_{\text{cent}} - \nu_{\text{calc.}}/\text{MHz}$
$38_{2,36} \leftarrow 38_{1,37}$	$A_{1s} \leftarrow A_{1d}$ $A_{2s} \leftarrow A_{2d}$	604 157.075 604 160.348	604 158.711	-0.1010
$39_{2,37} \leftarrow 39_{1,38}$	$A_{1s} \leftarrow A_{1d}$ $A_{2s} \leftarrow A_{2d}$	604 034.252 604 037.610	604 035.931	0.2840
$40_{2,38} \leftarrow 40_{1,39}$	$A_{1s} \leftarrow A_{1d}$ $A_{2s} \leftarrow A_{2d}$	603 907.420 603 910.695	603 909.058	-0.1800
$41_{2,39} \leftarrow 41_{1,40}$	$A_{1s} \leftarrow A_{1d}$ $A_{2s} \leftarrow A_{2d}$	603 777.974 603 781.068	603 779.521	-0.0613
$43_{2,41} \leftarrow 43_{1,42}$	$A_{1s} \leftarrow A_{1d}$ $A_{2s} \leftarrow A_{2d}$	603 508.900 603 512.190	603 510.545	0.0300
$8_{2,7} \leftarrow 8_{1,8}$	$A_{1s} \leftarrow A_{1d}$ $A_{2s} \leftarrow A_{2d}$	606 577.525 606 580.192	606 578.858	-0.0990
$9_{2,8} \leftarrow 9_{1,9}$	$A_{1s} \leftarrow A_{1d}$ $A_{2s} \leftarrow A_{2d}$	606 602.127 606 604.972	606 603.550	0.0220
$10_{2,9} \leftarrow 10_{1,10}$	$A_{1s} \leftarrow A_{1d}$ $A_{2s} \leftarrow A_{2d}$	606 629.451 606 632.357	606 630.904	0.0628
$11_{2,10} \leftarrow 11_{1,11}$	$A_{1s} \leftarrow A_{1d}$ $A_{2s} \leftarrow A_{2d}$	606 659.643 606 662.423	606 661.033	0.1330
$12_{2,11} \leftarrow 12_{1,12}$	$A_{1s} \leftarrow A_{1d}$ $A_{2s} \leftarrow A_{2d}$	606 692.257 606 695.048	606 693.653	-0.0566
$13_{2,12} \leftarrow 13_{1,13}$	$A_{1s} \leftarrow A_{1d}$ $A_{2s} \leftarrow A_{2d}$	606 727.826 606 730.742	606 729.284	0.0104
$14_{2,13} \leftarrow 14_{1,14}$	$A_{1s} \leftarrow A_{1d}$ $A_{2d} \leftarrow A_{2s}$	606 766.074 606 768.995	606 767.535	-0.0619
$15_{2,14} \leftarrow 15_{1,15}$	$A_{1s} \leftarrow A_{1d}$ $A_{2s} \leftarrow A_{2d}$	606 807.204 606 810.084	606 808.644	-0.0409
$16_{2,15} \leftarrow 16_{1,16}$	$A_{1s} \leftarrow A_{1d}$ $A_{2s} \leftarrow A_{2d}$	606 850.978 606 853.994	606 852.486	-0.0570
$18_{2,17} \leftarrow 18_{1,18}$	$A_{1s} \leftarrow A_{1d}$ $A_{2s} \leftarrow A_{2d}$	606 947.035 606 949.883	606 948.459	-0.1340
$20_{2,19} \leftarrow 20_{1,20}$	$A_{1s} \leftarrow A_{1d}$ $A_{2s} \leftarrow A_{2d}$	607 054.728 607 057.189	607 055.958	0.1610
$21_{2,20} \leftarrow 21_{1,21}$	$A_{1s} \leftarrow A_{1d}$ $A_{2s} \leftarrow A_{2d}$	607 112.227 607 114.942	607 113.585	-0.0140
$22_{2,21} \leftarrow 22_{1,22}$	$A_{1s} \leftarrow A_{1d}$ $A_{2s} \leftarrow A_{2d}$	607 172.655 607 175.652	607 174.154	-0.0563
$23_{2,22} \leftarrow 23_{1,23}$	$A_{1s} \leftarrow A_{1d}$ $A_{2s} \leftarrow A_{2d}$	607 236.106 607 239.685	607 237.895	0.2570
$24_{2,23} \leftarrow 24_{1,24}$	$A_{1s} \leftarrow A_{1d}$ $A_{2s} \leftarrow A_{2d}$	607 302.395 607 305.201	607 303.798	-0.0933
$25_{2,24} \leftarrow 25_{1,25}$	$A_{1s} \leftarrow A_{1d}$ $A_{2s} \leftarrow A_{2d}$	607 371.521 607 374.212	607 372.867	-0.1100



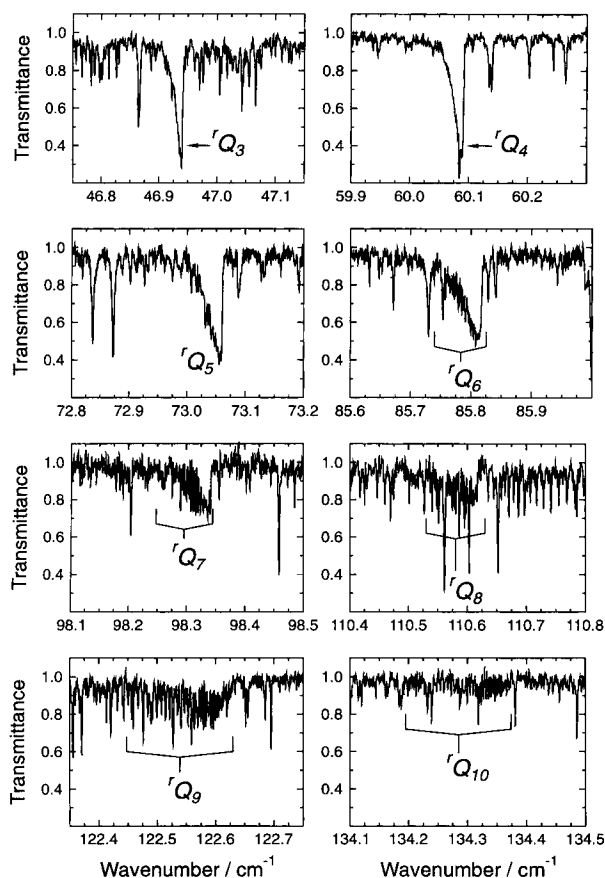


Figure 8.  ${}^rQ_{K_a}$  branches of DNCND in the far infrared region. Due to their characteristic appearance they play a key role in the identification of DNCND transitions among the dense transitions of the  $D_2NCN$  spectrum.

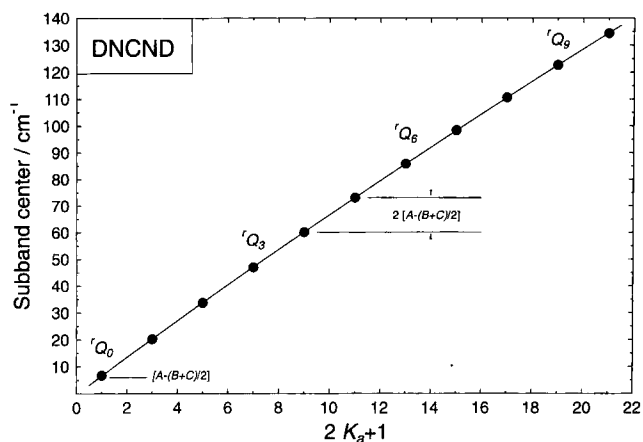


Figure 9. Plot of the subband centres of DNCND sub-branches versus  $(2K_a + 1)$ , which shows the regular linear behaviour of an unperturbed perpendicular band system.

An example for these transitions is presented in figure 10, and the assigned line positions are given in table 6. Overlaps with strong  $D_2NCN$  transitions are again frequent. Some  ${}^rR_2$ -branch lines were also observed. However, they are omitted in table 6 due to their low intensities and large errors caused by the overlap with cyanamide transitions.

Neither asymmetry splitting nor torsional splittings are observed for DNCND lines assigned in the FIR region. The asymmetry splitting collapses, of course, with increasing  $K_a$ . The torsional splitting also collapses, because the rotation about the  $a$  axis produces a centrifugal force which effectively increases the torsional barrier [3].

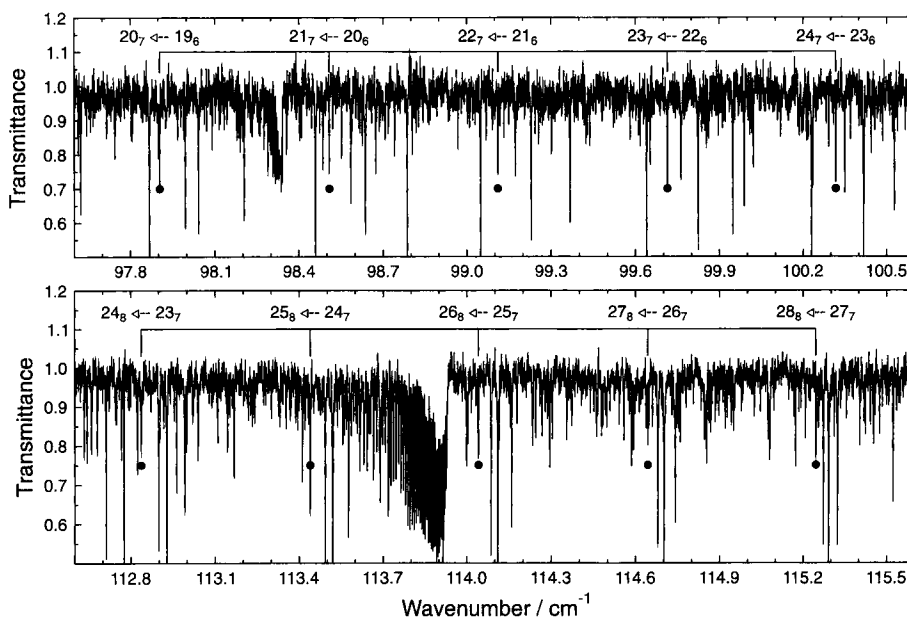


Figure 10. Part of two  ${}^rR_{K_a}$  branches of DNCND in the far infrared region. Nearly all of the strong lines belong to the  $D_2NCN$  molecule.

Table 6. Assigned transitions of DNCND in the FIR region. The measured wavenumbers, which show no asymmetry or torsional splitting, were fitted together with the data of table 5 to the Watson Hamiltonian in  $S$  reduction. The wavenumber differences between the observed transitions and those calculated with the constants of table 7 are also listed.

$J_{K'_d} \leftarrow J_{K''_d}$	Measured transitions /cm <sup>-1</sup>	$\tilde{\nu}_{\text{obs.}} - \tilde{\nu}_{\text{calc}}$ $\times 10^3$ /cm <sup>-1</sup>	$J_{K'_d} \leftarrow J_{K''_d}$	Measured transitions /cm <sup>-1</sup>	$\tilde{\nu}_{\text{obs.}} - \tilde{\nu}_{\text{calc}}$ $\times 10^3$ /cm <sup>-1</sup>	$J_{K'_d} \leftarrow J_{K''_d}$	Measured transitions /cm <sup>-1</sup>	$\tilde{\nu}_{\text{obs.}} - \tilde{\nu}_{\text{calc}}$ $\times 10^3$ /cm <sup>-1</sup>
<sup>r</sup> R <sub>3</sub> -branch transitions, $A_s \leftarrow A_d$								
4 <sub>4</sub> ← 3 <sub>3</sub>	49.360 888	-0.416	5 <sub>4</sub> ← 4 <sub>3</sub>	49.965 848	-0.376	6 <sub>4</sub> ← 5 <sub>3</sub>	50.570 951	-0.163
7 <sub>4</sub> ← 6 <sub>3</sub>	51.176 040	0.067	8 <sub>4</sub> ← 7 <sub>3</sub>	51.780 777	-0.021	9 <sub>4</sub> ← 8 <sub>3</sub>	52.385 525	-0.063
10 <sub>4</sub> ← 9 <sub>3</sub>	52.990 037	-0.302	11 <sub>4</sub> ← 10 <sub>3</sub>	53.595 092	0.041	12 <sub>4</sub> ← 11 <sub>3</sub>	54.199 671	-0.049
13 <sub>4</sub> ← 12 <sub>3</sub>	54.804 387	0.042	14 <sub>4</sub> ← 13 <sub>3</sub>	55.408 722	-0.201	15 <sub>4</sub> ← 14 <sub>3</sub>	56.013 427	-0.025
16 <sub>4</sub> ← 15 <sub>3</sub>	56.618 022	0.091	17 <sub>4</sub> ← 16 <sub>3</sub>	57.222 289	-0.067	18 <sub>4</sub> ← 17 <sub>3</sub>	57.826 786	0.060
19 <sub>4</sub> ← 18 <sub>3</sub>	58.431 067	0.028	20 <sub>4</sub> ← 19 <sub>3</sub>	59.035 321	0.029	22 <sub>4</sub> ← 21 <sub>3</sub>	60.243 528	-0.082
23 <sub>4</sub> ← 22 <sub>3</sub>	60.847 610	-0.060	24 <sub>4</sub> ← 23 <sub>3</sub>	61.451 541	-0.121	25 <sub>4</sub> ← 24 <sub>3</sub>	62.055 438	-0.146
26 <sub>4</sub> ← 25 <sub>3</sub>	62.659 493	0.060	27 <sub>4</sub> ← 26 <sub>3</sub>	63.263 013	-0.194	28 <sub>4</sub> ← 27 <sub>3</sub>	63.866 894	-0.010
29 <sub>4</sub> ← 28 <sub>3</sub>	64.470 537	0.016	30 <sub>4</sub> ← 29 <sub>3</sub>	65.074 168	0.111	31 <sub>4</sub> ← 30 <sub>3</sub>	65.677 471	-0.038
32 <sub>4</sub> ← 31 <sub>3</sub>	66.280 628	-0.247	33 <sub>4</sub> ← 32 <sub>3</sub>	66.884 160	0.006	34 <sub>4</sub> ← 33 <sub>3</sub>	67.487 083	-0.259
35 <sub>4</sub> ← 34 <sub>3</sub>	68.090 298	-0.139	36 <sub>4</sub> ← 35 <sub>3</sub>	68.693 100	-0.338	39 <sub>4</sub> ← 38 <sub>3</sub>	70.501 874	0.023
40 <sub>4</sub> ← 39 <sub>3</sub>	71.104 084	-0.367	41 <sub>4</sub> ← 40 <sub>3</sub>	71.706 652	-0.294	43 <sub>4</sub> ← 42 <sub>3</sub>	72.911 650	0.040
<sup>r</sup> R <sub>4</sub> -branch transitions, $A_d \leftarrow A_s$								
5 <sub>4</sub> ← 4 <sub>4</sub>	63.117 351	0.015	6 <sub>5</sub> ← 5 <sub>4</sub>	63.722 215	0.060	7 <sub>5</sub> ← 6 <sub>4</sub>	64.326 997	0.063
8 <sub>5</sub> ← 7 <sub>4</sub>	64.931 693	0.022	9 <sub>5</sub> ← 8 <sub>4</sub>	65.536 482	0.119	10 <sub>5</sub> ← 9 <sub>4</sub>	66.141 094	0.086
11 <sub>5</sub> ← 10 <sub>4</sub>	66.745 700	0.095	12 <sub>5</sub> ← 11 <sub>4</sub>	67.350 285	0.134	13 <sub>5</sub> ← 12 <sub>4</sub>	67.954 794	0.151
14 <sub>5</sub> ← 13 <sub>4</sub>	68.559 154	0.074	15 <sub>5</sub> ← 14 <sub>4</sub>	69.163 574	0.115	16 <sub>5</sub> ← 15 <sub>4</sub>	69.767 873	0.094
17 <sub>5</sub> ← 16 <sub>4</sub>	70.372 201	0.164	18 <sub>5</sub> ← 17 <sub>4</sub>	70.976 269	0.039	19 <sub>5</sub> ← 18 <sub>4</sub>	71.580 644	0.287
20 <sub>5</sub> ← 19 <sub>4</sub>	72.184 698	0.282	21 <sub>5</sub> ← 20 <sub>4</sub>	72.788 530	0.125	22 <sub>5</sub> ← 21 <sub>4</sub>	73.392 501	0.181
23 <sub>5</sub> ← 22 <sub>4</sub>	73.996 227	0.067	24 <sub>5</sub> ← 23 <sub>4</sub>	74.599 726	-0.198	25 <sub>5</sub> ← 24 <sub>4</sub>	75.203 509	-0.099
26 <sub>5</sub> ← 25 <sub>4</sub>	75.807 134	-0.076	27 <sub>5</sub> ← 26 <sub>4</sub>	76.410 765	0.036	28 <sub>5</sub> ← 27 <sub>4</sub>	77.013 952	-0.210
29 <sub>5</sub> ← 28 <sub>4</sub>	77.617 411	-0.096	31 <sub>5</sub> ← 30 <sub>4</sub>	78.824 046	0.122	32 <sub>5</sub> ← 31 <sub>4</sub>	79.427 123	0.132
33 <sub>5</sub> ← 32 <sub>4</sub>	80.030 026	0.064	34 <sub>5</sub> ← 33 <sub>4</sub>	80.632 963	0.129	35 <sub>5</sub> ← 34 <sub>4</sub>	81.235 584	-0.021
36 <sub>5</sub> ← 35 <sub>4</sub>	81.838 457	0.184	37 <sub>5</sub> ← 36 <sub>4</sub>	82.441 048	0.213	38 <sub>5</sub> ← 37 <sub>4</sub>	83.043 525	0.236
39 <sub>5</sub> ← 38 <sub>4</sub>	83.645 097	-0.537	40 <sub>5</sub> ← 39 <sub>4</sub>	84.247 733	-0.134	41 <sub>5</sub> ← 40 <sub>4</sub>	84.849 798	-0.187
42 <sub>5</sub> ← 41 <sub>4</sub>	85.451 490	-0.497	43 <sub>5</sub> ← 42 <sub>4</sub>	86.054 216	0.345	44 <sub>5</sub> ← 43 <sub>4</sub>	86.655 547	-0.087
<sup>r</sup> R <sub>5</sub> -branch transitions, $A_s \leftarrow A_d$								
6 <sub>6</sub> ← 5 <sub>5</sub>	76.690 714	0.161	7 <sub>6</sub> ← 6 <sub>5</sub>	77.295 523	0.298	8 <sub>6</sub> ← 7 <sub>5</sub>	77.900 007	0.164
9 <sub>6</sub> ← 8 <sub>5</sub>	78.504 722	0.318	10 <sub>6</sub> ← 9 <sub>5</sub>	79.109 100	0.192	11 <sub>6</sub> ← 10 <sub>5</sub>	79.713 379	0.027
12 <sub>6</sub> ← 11 <sub>5</sub>	80.317 774	0.041	13 <sub>6</sub> ← 12 <sub>5</sub>	80.922 218	0.169	14 <sub>6</sub> ← 13 <sub>5</sub>	81.526 404	0.105
15 <sub>6</sub> ← 14 <sub>5</sub>	82.130 513	0.034	16 <sub>6</sub> ← 15 <sub>5</sub>	82.734 808	0.220	17 <sub>6</sub> ← 16 <sub>5</sub>	83.338 759	0.135
18 <sub>6</sub> ← 17 <sub>5</sub>	83.942 717	0.132	19 <sub>6</sub> ← 18 <sub>5</sub>	84.546 760	0.293	20 <sub>6</sub> ← 19 <sub>5</sub>	85.150 369	0.099
21 <sub>6</sub> ← 20 <sub>5</sub>	85.754 226	0.235	22 <sub>6</sub> ← 21 <sub>5</sub>	86.357 808	0.181	23 <sub>6</sub> ← 22 <sub>5</sub>	86.961 429	0.252
25 <sub>6</sub> ← 24 <sub>5</sub>	88.168 126	0.117	26 <sub>6</sub> ← 25 <sub>5</sub>	88.771 731	0.444	27 <sub>6</sub> ← 26 <sub>5</sub>	89.374 722	0.252
28 <sub>6</sub> ← 27 <sub>5</sub>	89.977 574	0.019	29 <sub>6</sub> ← 28 <sub>5</sub>	90.580 669	0.128	30 <sub>6</sub> ← 29 <sub>5</sub>	91.183 573	0.148
31 <sub>6</sub> ← 30 <sub>5</sub>	91.786 608	0.403	32 <sub>6</sub> ← 31 <sub>5</sub>	92.388 824	-0.055	33 <sub>6</sub> ← 32 <sub>5</sub>	92.991 512	0.067
34 <sub>6</sub> ← 33 <sub>5</sub>	93.593 251	-0.650	35 <sub>6</sub> ← 34 <sub>5</sub>	94.196 547	0.303	36 <sub>6</sub> ← 35 <sub>5</sub>	94.798 271	-0.202
37 <sub>6</sub> ← 36 <sub>5</sub>	95.400 346	-0.238	38 <sub>6</sub> ← 37 <sub>5</sub>	96.002 493	-0.084	39 <sub>6</sub> ← 38 <sub>5</sub>	96.604 431	-0.017
40 <sub>6</sub> ← 39 <sub>5</sub>	97.205 928	-0.268	41 <sub>6</sub> ← 40 <sub>5</sub>	97.807 528	-0.290	42 <sub>6</sub> ← 41 <sub>5</sub>	98.409 320	0.007
43 <sub>6</sub> ← 42 <sub>5</sub>	99.010 458	-0.219	44 <sub>6</sub> ← 43 <sub>5</sub>	99.611 828	-0.082	45 <sub>6</sub> ← 44 <sub>5</sub>	100.212 332	-0.676
46 <sub>6</sub> ← 45 <sub>5</sub>	100.814 096	0.127	47 <sub>6</sub> ← 46 <sub>5</sub>	101.414 887	0.095	48 <sub>6</sub> ← 47 <sub>5</sub>	102.015 110	-0.364
<sup>r</sup> R <sub>6</sub> -branch transitions, $A_d \leftarrow A_s$								
7 <sub>7</sub> ← 6 <sub>6</sub>	90.052 909	-0.006	9 <sub>7</sub> ← 8 <sub>6</sub>	91.261 734	-0.034	10 <sub>7</sub> ← 9 <sub>6</sub>	91.865 992	-0.094
11 <sub>7</sub> ← 10 <sub>6</sub>	92.470 399	0.070	12 <sub>7</sub> ← 11 <sub>6</sub>	93.074 414	-0.081	13 <sub>7</sub> ← 12 <sub>6</sub>	93.678 355	-0.226
14 <sub>7</sub> ← 13 <sub>6</sub>	94.282 654	0.068	15 <sub>7</sub> ← 14 <sub>6</sub>	94.886 309	-0.199	16 <sub>7</sub> ← 15 <sub>6</sub>	95.490 149	-0.194
18 <sub>7</sub> ← 17 <sub>6</sub>	96.697 742	-0.005	19 <sub>7</sub> ← 18 <sub>6</sub>	97.301 202	-0.110	20 <sub>7</sub> ← 19 <sub>6</sub>	97.904 691	-0.091
21 <sub>7</sub> ← 20 <sub>6</sub>	98.508 114	-0.041	22 <sub>7</sub> ← 21 <sub>6</sub>	99.111 159	-0.271	23 <sub>7</sub> ← 22 <sub>6</sub>	99.714 346	-0.257
24 <sub>7</sub> ← 23 <sub>6</sub>	100.317 703	0.030	25 <sub>7</sub> ← 24 <sub>6</sub>	100.920 504	-0.133	26 <sub>7</sub> ← 25 <sub>6</sub>	101.523 381	-0.113
28 <sub>7</sub> ← 27 <sub>6</sub>	102.728 877	0.002	29 <sub>7</sub> ← 28 <sub>6</sub>	103.331 551	0.155	30 <sub>7</sub> ← 29 <sub>6</sub>	103.933 858	0.057

(continued)

Table 6 (*continued*)

$J_{K'_a} \leftarrow J_{K''_a}$	Measured transitions /cm <sup>-1</sup>	$\tilde{\nu}_{\text{obs.}} - \tilde{\nu}_{\text{calc.}}$ $\times 10^3$ /cm <sup>-1</sup>	$J_{K'_a} \leftarrow J_{K''_a}$	Measured transitions /cm <sup>-1</sup>	$\tilde{\nu}_{\text{obs.}} - \tilde{\nu}_{\text{calc.}}$ $\times 10^3$ /cm <sup>-1</sup>	$J_{K'_a} \leftarrow J_{K''_a}$	Measured transitions /cm <sup>-1</sup>	$\tilde{\nu}_{\text{obs.}} - \tilde{\nu}_{\text{calc.}}$ $\times 10^3$ /cm <sup>-1</sup>
31 <sub>7</sub> ← 30 <sub>6</sub>	104.536 091	0.005	33 <sub>7</sub> ← 32 <sub>6</sub>	105.740 033	-0.261	34 <sub>7</sub> ← 33 <sub>6</sub>	106.342 121	-0.090
35 <sub>7</sub> ← 34 <sub>6</sub>	106.943 889	-0.112	36 <sub>7</sub> ← 35 <sub>6</sub>	107.545 818	0.157	37 <sub>7</sub> ← 36 <sub>6</sub>	108.146 935	-0.255
38 <sub>7</sub> ← 37 <sub>6</sub>	108.748 704	0.119	39 <sub>7</sub> ← 38 <sub>6</sub>	109.349 906	0.061	40 <sub>7</sub> ← 39 <sub>6</sub>	109.950 725	-0.241
41 <sub>7</sub> ← 40 <sub>6</sub>	110.551 514	-0.432	43 <sub>7</sub> ← 42 <sub>6</sub>	111.753 625	0.147	44 <sub>7</sub> ← 43 <sub>6</sub>	112.354 172	0.147
45 <sub>7</sub> ← 44 <sub>6</sub>	112.954 357	-0.066	46 <sub>7</sub> ← 45 <sub>6</sub>	113.554 806	0.136	47 <sub>7</sub> ← 46 <sub>6</sub>	114.155 040	0.277
<sup>r</sup> R <sub>7</sub> -branch transitions, $A_s \leftarrow A_d$								
8 <sub>8</sub> ← 7 <sub>7</sub>	103.181 903	-0.090	9 <sub>8</sub> ← 8 <sub>7</sub>	103.785 908	-0.256	10 <sub>8</sub> ← 9 <sub>7</sub>	104.390 197	-0.046
11 <sub>8</sub> ← 10 <sub>7</sub>	104.994 452	0.223	12 <sub>8</sub> ← 11 <sub>7</sub>	105.597 847	-0.272	13 <sub>8</sub> ← 12 <sub>7</sub>	106.201 699	-0.212
14 <sub>8</sub> ← 13 <sub>7</sub>	106.805 445	-0.158	15 <sub>8</sub> ← 14 <sub>7</sub>	107.409 143	-0.050	16 <sub>8</sub> ← 15 <sub>7</sub>	108.012 665	-0.013
17 <sub>8</sub> ← 16 <sub>7</sub>	108.615 927	-0.129	18 <sub>8</sub> ← 17 <sub>7</sub>	109.219 045	-0.281	18 <sub>8</sub> ← 18 <sub>7</sub>	109.822 160	-0.325
20 <sub>8</sub> ← 19 <sub>7</sub>	110.425 481	-0.049	21 <sub>8</sub> ← 20 <sub>7</sub>	111.028 469	0.008	22 <sub>8</sub> ← 21 <sub>7</sub>	111.631 183	-0.090
24 <sub>8</sub> ← 23 <sub>7</sub>	112.836 776	0.238	25 <sub>8</sub> ← 24 <sub>7</sub>	113.438 985	0.000	26 <sub>8</sub> ← 25 <sub>7</sub>	114.041 353	0.047
27 <sub>8</sub> ← 26 <sub>7</sub>	114.643 399	-0.099	28 <sub>8</sub> ← 27 <sub>7</sub>	115.245 382	-0.178	30 <sub>8</sub> ← 29 <sub>7</sub>	116.449 168	-0.116
33 <sub>8</sub> ← 32 <sub>7</sub>	118.253 975	0.139	34 <sub>8</sub> ← 33 <sub>7</sub>	118.855 458	0.389	35 <sub>8</sub> ← 34 <sub>7</sub>	119.456 147	-0.009
36 <sub>8</sub> ← 35 <sub>7</sub>	120.057 306	0.211	37 <sub>8</sub> ← 36 <sub>7</sub>	120.657 956	0.072	38 <sub>8</sub> ← 37 <sub>7</sub>	121.258 538	0.017
39 <sub>8</sub> ← 38 <sub>7</sub>	121.859 340	0.337	40 <sub>8</sub> ← 39 <sub>7</sub>	122.459 502	0.174	41 <sub>8</sub> ← 40 <sub>7</sub>	123.059 797	0.302
42 <sub>8</sub> ← 41 <sub>7</sub>	123.659 866	0.365	43 <sub>8</sub> ← 42 <sub>7</sub>	124.259 675	0.332	44 <sub>8</sub> ← 43 <sub>7</sub>	124.859 492	0.472
45 <sub>8</sub> ← 44 <sub>7</sub>	125.458 890	0.360						
<sup>r</sup> R <sub>8</sub> -branch transitions, $A_d \leftarrow A_s$								
10 <sub>9</sub> ← 9 <sub>8</sub>	116.664 469	0.007	11 <sub>9</sub> ← 10 <sub>8</sub>	117.267 803	-0.320	12 <sub>9</sub> ← 11 <sub>8</sub>	117.871 419	-0.246
13 <sub>9</sub> ← 12 <sub>8</sub>	118.474 787	-0.299	14 <sub>9</sub> ← 13 <sub>8</sub>	119.078 214	-0.170	15 <sub>9</sub> ← 14 <sub>8</sub>	119.681 240	-0.317
16 <sub>9</sub> ← 15 <sub>8</sub>	120.284 423	-0.179	17 <sub>9</sub> ← 16 <sub>8</sub>	120.887 349	-0.169	20 <sub>9</sub> ← 19 <sub>8</sub>	122.695 413	-0.052
21 <sub>9</sub> ← 20 <sub>8</sub>	123.297 642	-0.199	22 <sub>9</sub> ← 21 <sub>8</sub>	123.899 924	-0.152	23 <sub>9</sub> ← 22 <sub>8</sub>	124.501 777	-0.391
24 <sub>9</sub> ← 23 <sub>8</sub>	125.104 079	-0.036	25 <sub>9</sub> ← 24 <sub>8</sub>	125.705 829	-0.087	26 <sub>9</sub> ← 25 <sub>8</sub>	126.307 550	-0.017
27 <sub>9</sub> ← 26 <sub>8</sub>	126.909 149	0.082	28 <sub>9</sub> ← 27 <sub>8</sub>	127.510 476	0.063	29 <sub>9</sub> ← 28 <sub>8</sub>	128.111 523	-0.080
30 <sub>9</sub> ← 29 <sub>8</sub>	128.712 401	-0.235	31 <sub>9</sub> ← 30 <sub>8</sub>	129.313 595	0.087	32 <sub>9</sub> ← 31 <sub>8</sub>	129.914 445	0.227
34 <sub>9</sub> ← 33 <sub>8</sub>	131.115 078	-0.065	35 <sub>9</sub> ← 34 <sub>8</sub>	131.714 653	-0.700	36 <sub>9</sub> ← 35 <sub>8</sub>	132.315 545	0.152
37 <sub>9</sub> ← 36 <sub>8</sub>	132.915 432	0.173	38 <sub>9</sub> ← 37 <sub>8</sub>	133.515 311	0.361	39 <sub>9</sub> ← 38 <sub>8</sub>	134.114 517	0.054
40 <sub>9</sub> ← 39 <sub>8</sub>	134.714 032	0.235	42 <sub>9</sub> ← 41 <sub>8</sub>	135.912 454	0.537	43 <sub>9</sub> ← 42 <sub>8</sub>	136.511 177	0.478
<sup>r</sup> R <sub>9</sub> -branch transitions, $A_s \leftarrow A_d$								
10 <sub>10</sub> ← 9 <sub>9</sub>	128.676 823	0.019	11 <sub>10</sub> ← 10 <sub>9</sub>	129.279 997	-0.065	12 <sub>10</sub> ← 11 <sub>9</sub>	129.883 165	-0.008
13 <sub>10</sub> ← 12 <sub>9</sub>	130.486 050	-0.085	15 <sub>10</sub> ← 14 <sub>9</sub>	131.691 651	0.048	16 <sub>10</sub> ← 15 <sub>9</sub>	132.294 058	-0.047
17 <sub>10</sub> ← 16 <sub>9</sub>	132.896 491	0.042	18 <sub>10</sub> ← 17 <sub>9</sub>	133.498 565	-0.069	19 <sub>10</sub> ← 18 <sub>9</sub>	134.100 566	-0.090
20 <sub>10</sub> ← 19 <sub>9</sub>	134.702 387	-0.127	21 <sub>10</sub> ← 20 <sub>9</sub>	135.304 216	0.010	22 <sub>10</sub> ← 21 <sub>9</sub>	135.905 464	-0.265
23 <sub>10</sub> ← 22 <sub>9</sub>	136.507 059	-0.023	24 <sub>10</sub> ← 23 <sub>9</sub>	137.108 185	-0.077	25 <sub>10</sub> ← 24 <sub>9</sub>	137.709 033	-0.233
26 <sub>10</sub> ← 25 <sub>9</sub>	138.310 216	0.122	27 <sub>10</sub> ← 26 <sub>9</sub>	138.910 741	-0.000	28 <sub>10</sub> ← 27 <sub>9</sub>	139.511 241	0.033
29 <sub>10</sub> ← 28 <sub>9</sub>	140.111 701	0.210	31 <sub>10</sub> ← 30 <sub>9</sub>	141.311 567	0.071	32 <sub>10</sub> ← 31 <sub>9</sub>	141.911 527	0.313
33 <sub>10</sub> ← 32 <sub>9</sub>	142.510 736	-0.004	34 <sub>10</sub> ← 33 <sub>9</sub>	143.110 251	0.180	35 <sub>10</sub> ← 34 <sub>9</sub>	143.709 387	0.181
36 <sub>10</sub> ← 35 <sub>9</sub>	144.308 199	0.058	37 <sub>10</sub> ← 36 <sub>9</sub>	144.907 240	0.365	38 <sub>10</sub> ← 37 <sub>9</sub>	145.505 617	0.211
<sup>r</sup> Q <sub>9</sub> -branch transitions, $A_s \leftarrow A_d$								
13 <sub>10</sub> ← 13 <sub>9</sub>	122.625 920	0.468	14 <sub>10</sub> ← 14 <sub>9</sub>	122.624 073	0.341	15 <sub>10</sub> ← 15 <sub>9</sub>	122.621 966	0.077
16 <sub>10</sub> ← 16 <sub>9</sub>	122.619 928	0.005	17 <sub>10</sub> ← 17 <sub>9</sub>	122.617 538	-0.296	18 <sub>10</sub> ← 18 <sub>9</sub>	122.615 399	-0.223
19 <sub>10</sub> ← 19 <sub>9</sub>	122.613 122	-0.166	20 <sub>10</sub> ← 20 <sub>9</sub>	122.610 838	0.007	21 <sub>10</sub> ← 21 <sub>9</sub>	122.607 984	-0.266
22 <sub>10</sub> ← 22 <sub>9</sub>	122.605 620	0.073	23 <sub>10</sub> ← 23 <sub>9</sub>	122.603 003	0.282	24 <sub>10</sub> ← 24 <sub>9</sub>	122.600 083	0.310
25 <sub>10</sub> ← 25 <sub>9</sub>	122.596 476	-0.225	26 <sub>10</sub> ← 26 <sub>9</sub>	122.592 735	-0.771			
<sup>r</sup> R <sub>10</sub> -branch transitions, $A_d \leftarrow A_s$								
11 <sub>11</sub> ← 10 <sub>10</sub>	141.023 180	0.223	12 <sub>11</sub> ← 11 <sub>10</sub>	141.625 828	0.287	13 <sub>11</sub> ← 12 <sub>10</sub>	142.228 330	0.387
14 <sub>11</sub> ← 13 <sub>10</sub>	142.830 491	0.332	15 <sub>11</sub> ← 14 <sub>10</sub>	143.432 258	0.069	16 <sub>11</sub> ← 15 <sub>10</sub>	144.034 337	0.308
17 <sub>11</sub> ← 16 <sub>10</sub>	144.635 834	0.155	18 <sub>11</sub> ← 17 <sub>10</sub>	145.237 066	-0.068	19 <sub>11</sub> ← 18 <sub>10</sub>	145.838 383	-0.011
20 <sub>11</sub> ← 19 <sub>10</sub>	146.439 649	0.192	21 <sub>11</sub> ← 20 <sub>10</sub>	147.040 887	0.568	23 <sub>11</sub> ← 22 <sub>10</sub>	148.241 815	0.381

Table 6 (continued)

$J_{K'_a} \leftarrow J_{K''_a}$	Measured transitions /cm <sup>-1</sup>	$\tilde{\nu}_{\text{obs.}} - \tilde{\nu}_{\text{calc.}} \times 10^3$ /cm <sup>-1</sup>	$J_{K'_a} \leftarrow J_{K''_a}$	Measured transitions /cm <sup>-1</sup>	$\tilde{\nu}_{\text{obs.}} - \tilde{\nu}_{\text{calc.}} \times 10^3$ /cm <sup>-1</sup>	$J_{K'_a} \leftarrow J_{K''_a}$	Measured transitions /cm <sup>-1</sup>	$\tilde{\nu}_{\text{obs.}} - \tilde{\nu}_{\text{calc.}} \times 10^3$ /cm <sup>-1</sup>
25 <sub>11</sub> ← 24 <sub>10</sub>	149.441 642	-0.082	26 <sub>11</sub> ← 25 <sub>10</sub>	150.041 538	-0.015	27 <sub>11</sub> ← 26 <sub>10</sub>	150.640 741	-0.428
28 <sub>11</sub> ← 27 <sub>10</sub>	151.240 395	-0.175	29 <sub>11</sub> ← 28 <sub>10</sub>	151.839 846	0.092	30 <sub>11</sub> ← 29 <sub>10</sub>	152.438 507	-0.211
31 <sub>11</sub> ← 30 <sub>10</sub>	153.037 243	-0.217	32 <sub>11</sub> ← 31 <sub>10</sub>	153.635 715	-0.264	33 <sub>11</sub> ← 32 <sub>10</sub>	154.233 970	-0.301
34 <sub>11</sub> ← 33 <sub>10</sub>	154.832 087	-0.248	35 <sub>11</sub> ← 34 <sub>10</sub>	155.429 894	-0.274	36 <sub>11</sub> ← 35 <sub>10</sub>	156.027 551	-0.218
37 <sub>11</sub> ← 36 <sub>10</sub>	156.624 799	-0.336	38 <sub>11</sub> ← 37 <sub>10</sub>	157.222 606	0.342	39 <sub>11</sub> ← 38 <sub>10</sub>	157.819 036	-0.117
40 <sub>11</sub> ← 39 <sub>10</sub>	158.415 512	-0.290	41 <sub>11</sub> ← 40 <sub>10</sub>	159.012 085	-0.122	42 <sub>11</sub> ← 41 <sub>10</sub>	159.608 432	0.066
43 <sub>11</sub> ← 42 <sub>10</sub>	160.204 642	0.365	44 <sub>11</sub> ← 43 <sub>10</sub>	160.799 369	-0.569			
<sup>r</sup> Q <sub>10</sub> -branch transitions, $A_d \leftarrow A_s$								
11 <sub>11</sub> ← 11 <sub>10</sub>	134.373 351	0.581	12 <sub>11</sub> ← 12 <sub>10</sub>	134.371 351	0.459	13 <sub>11</sub> ← 13 <sub>10</sub>	134.369 368	0.511
14 <sub>11</sub> ← 14 <sub>10</sub>	134.367 094	0.428	15 <sub>11</sub> ← 15 <sub>10</sub>	134.364 408	0.090	16 <sub>11</sub> ← 16 <sub>10</sub>	134.361 952	0.139
17 <sub>11</sub> ← 17 <sub>10</sub>	134.359 071	-0.081	18 <sub>11</sub> ← 18 <sub>10</sub>	134.356 446	0.111	19 <sub>11</sub> ← 19 <sub>10</sub>	134.352 993	-0.368
20 <sub>11</sub> ← 20 <sub>10</sub>	134.350 542	0.312	21 <sub>11</sub> ← 21 <sub>10</sub>	134.347 241	0.298	22 <sub>11</sub> ← 22 <sub>10</sub>	134.343 548	0.048
23 <sub>11</sub> ← 23 <sub>10</sub>	134.339 815	-0.085	24 <sub>11</sub> ← 24 <sub>10</sub>	134.336 014	-0.129	25 <sub>11</sub> ← 25 <sub>10</sub>	134.332 108	-0.122
26 <sub>11</sub> ← 26 <sub>10</sub>	134.328 088	-0.072	27 <sub>11</sub> ← 27 <sub>10</sub>	134.323 867	-0.067	28 <sub>11</sub> ← 28 <sub>10</sub>	134.318 966	-0.585
29 <sub>11</sub> ← 29 <sub>10</sub>	134.314 675	-0.337	30 <sub>11</sub> ← 30 <sub>10</sub>	134.310 129	-0.187	31 <sub>11</sub> ← 31 <sub>10</sub>	134.304 709	-0.755

In order to determine the rotational constants for the isotopomers of HNCNH, a separation of the torsional from the rotational information is needed. The unperturbed position of the pure rotational transition is given by the centre of each torsional-rotational doublet, since both components have the same statistical weight. The resulting line positions were then fitted to the constants of the Watson Hamiltonian in *S* reduction (*I'* representation) [21, 22]. The matrix elements in the basis of the prolate symmetric top wavefunctions  $|JK_a\rangle$ , including the constants relevant to this work, are as follows:

$$\begin{aligned} \frac{E_{K_a, K_a}}{h} = & \frac{1}{2}(B + C)J(J + 1) + [A - \frac{1}{2}(B + C)]K_a^2 \\ & - D_J J^2(J + 1)^2 - D_{JK} J(J + 1)K_a^2 \\ & - D_K K_a^4 + H_{KJ} J(J + 1)K_a^4 + H_K K_a^6 \\ & - L_{KJ} J(J + 1)K_a^6 - L_K K_a^8 + S_{KJ} J(J + 1)K_a^8 \\ & + S_K K_a^{10} - T_{KJ} J(J + 1)K_a^{10} \\ & - T_K K_a^{12} + U_K K_a^{14}, \end{aligned} \quad (3)$$

$$\begin{aligned} \frac{W_{K_a, K_a \pm 2}}{h} = & \frac{1}{4}[(B - C) + d_1 J(J + 1)] \\ & \times \{f(J, K_a, 0)f(J, K_a, 1)\}, \end{aligned} \quad (4)$$

$$\frac{V_{K_a, K_a \pm 4}}{h} = \{d_2 + h_2 J(J + 1)\} \prod_{l=0}^3 f(J, K_a, l), \quad (5)$$

with

$$f(J, K_a, l) = [J(J + 1) - (K_a \pm l)(K_a \pm l)(K_a \pm l \pm 1)]^{1/2}. \quad (6)$$

*A*, *B*, *C* are the rotational constants, while all the other constants are centrifugal distortion constants.

The adjusted constants are given in table 7 for four isotopomers of HNCNH. The data for the fit of the main species were taken from [1, 4, 5]. ([5] reports only spectroscopic constants for each torsional symmetry species separately.) Due to the limited data available for the H<sup>15</sup>NC<sup>15</sup>NH and HN<sup>13</sup>CNH, species, the constants *D<sub>K</sub>*, *H<sub>K</sub>*, *H<sub>KJ</sub>* and *d<sub>2</sub>* were held fixed in the fits at the values found for the main species. Ray's asymmetry parameter *κ* is also listed in table 8 (see section 5) for the four isotopomers.

#### 4. Discussion of the rotational analysis, torsional splitting and *K<sub>a</sub>* doubling

The most outstanding feature observed upon comparing the spectroscopic constants of the four HNCNH isotopomers, going from the main species to DNCND, is the decreasing trend of the *A* rotational constants and of the *D<sub>JK</sub>* and *D<sub>K</sub>* centrifugal distortion constants which describe the *K<sub>a</sub>*-dependent parts in the Hamiltonian. Lines with higher *K<sub>a</sub>* values are included in the DNCND fit compared with HNCNH (DNCND: *K<sub>a</sub>*<sup>max</sup> = 11, HNCNH: *K<sub>a</sub>*<sup>max</sup> = 10). However, higher order constants are necessary in the fit of the main species.

It is interesting to follow the changes in the torsional splittings and their *J* dependence for the various isotopomers of HNCNH, as plotted in figure 11. With higher mass, compared with the main species, the splitting decreases. For the deuterium isotopomer, the splitting decreases by two orders of magnitude, as was predicted

Table 7. Spectroscopic ground state constants of the Watson Hamiltonian in  $S$  reduction for carbodiimide isotopomers.

Constant	HNCNH <sup>a</sup>	HN <sup>13</sup> CNH	H <sup>15</sup> NC <sup>15</sup> NH	DNCND
$A/\text{MHz}$	379 244.402(21) <sup>b</sup>	378 735.042(19)	375 820.944 5(49)	211 517.168(25)
$B/\text{MHz}$	10 366.940 2(11)	10 367.373 7(20)	9 771.933 0(11)	9 072.434 0(50)
$C/\text{MHz}$	10 366.088 7(12)	10 366.120 4(30)	9 771.184 6(18)	9 066.601 9(60)
$D_J/\text{kHz}$	3.246 24(82)	3.240 8(28)	2.882 5(24)	2.716 9(57)
$D_{JK}/\text{kHz}$	325.790(16)	323.89(23)	315.83(13)	29.988(47)
$D_K/\text{MHz}$	168.625 0(44)	168.625 0 <sup>c</sup>	168.625 0 <sup>c</sup>	57.792 2(53)
$d_1/\text{Hz}$	−5.678(26)	−5.734 2(67)	−5.665 0(30)	−10.812(37)
$d_2/\text{Hz}$	−16.825(12)	−16.825 <sup>c</sup>	−16.825 <sup>c</sup>	−25.52(23)
$H_{KJ}/\text{Hz}$	−268.8(25)	−268.8 <sup>c</sup>	−268.8 <sup>c</sup>	−369.86(43)
$H_K/\text{kHz}$	430.92(40)	430.92 <sup>c</sup>	430.92 <sup>c</sup>	76.00(14)
$h_2/\text{mHz}$	—	—	—	0.83(11)
$L_{KJ}/\text{Hz}$	5.66(12)	—	—	—
$L_K/\text{kHz}$	1.773(17)	—	—	0.141 6(15)
$S_{KJ}/\text{mHz}$	32.3(23)	—	—	—
$S_K/\text{Hz}$	8.42(35)	—	—	0.193 9(51)
$T_{KJ}/\text{mHz}$	0.468(13)	—	—	—
$T_K/\text{mHz}$	31.7(34)	—	—	—
$U_K/\mu\text{Hz}$	61.(12)	—	—	—
$\sigma^d/\text{kHz}$	168	11	15	261
No. of lines	449	33	60	397
$\kappa^e$	−0.999 995 383(12)	−0.999 993 195(20)	−0.999 995 911(12)	−0.999 942 385(70)

<sup>a</sup> Data for the fit of the main species were taken from [1, 4, 5]<sup>b</sup> The numbers in parentheses are one standard deviation in units of the last digit quoted.<sup>c</sup> Constant held fixed at the value of the main species.<sup>d</sup> Standard deviation of the fit.<sup>e</sup> Ray's asymmetry parameter.

by Pracna *et al.* [3]. The higher mass and the resulting decrease of the zero point energy level are responsible for this effect.

Not only the magnitude of the torsional splitting changes upon isotopic substitution, but also the  $J$  dependence: the splitting of the  ${}^7Q_0$ -branch transitions of HN<sup>13</sup>CNH decreases with increasing  $J$ , while for all other isotopomers the splitting increases with  $J$ . The strong cyanamide absorptions in the region of the  ${}^7Q_1$  branch of DNCND perturb in many cases at least one component of the torsional doublet, so that the errors in the values obtained for the torsional splitting are larger than the  $J$  dependence. From the diagram in figure 11, one can only see that the torsional splitting of the  $J_{2,J-2} \leftarrow J_{1,J-1}$  branch transitions is somewhat larger than the splitting of the  $J_{2,J-1} \leftarrow J_{1,J}$  transitions, which is also true for the main species [5]. The  ${}^7Q_0$ - and  ${}^7Q_1$ -branch transitions of DNCND show a much smaller  $J$  dependence than was found for the other isotopomers.

DNCND is more asymmetric than the other isotopomers, but it still shows an anomalous  $K_a$  doubling of the  $K_a = 2$  energy levels, which means that the  $J_{2,J-2}$  energy

levels are below the  $J_{2,J-1}$  energy levels. This can be clearly seen in figure 12, where the reduced term values  $F_{\text{red}}$  of these states are plotted against  $J$ . The reduced term values are defined below.

Winnewisser [19, 23] showed that two contributions to the  $K_a$ -type doubling are important for accidentally nearly symmetric top molecules: the inertial asymmetry splitting which can be considered to be a  $\Delta K_a = \pm 2$  interaction, and the centrifugal-distortion splitting, represented by the  $\Delta K_a = \pm 4$  interaction matrix element in the asymmetric top Hamiltonian. The anomalous  $K_a$ -type doubling has been discussed by applying a second-order perturbation treatment to Watson's  $S$ -reduced Hamiltonian [24]. The different contributions to the observed splitting can be visualized by calculating reduced term values  $F_{\text{red}}$  defined as  $F_{\text{red}} = F - F_{\text{mod}}$  [5]. The term values  $F$  were calculated with all constants given in table 7, reproducing the observed spectrum, and the term values  $F_{\text{mod}}$  were obtained according to three different models:

- $F_{\text{mod}}$  is calculated with both the  $\Delta K_a = \pm 2$  and  $\Delta K_a = \pm 4$  matrix elements fixed to zero, so that  $F_{\text{mod}}$  represents the term value of DNCND as a

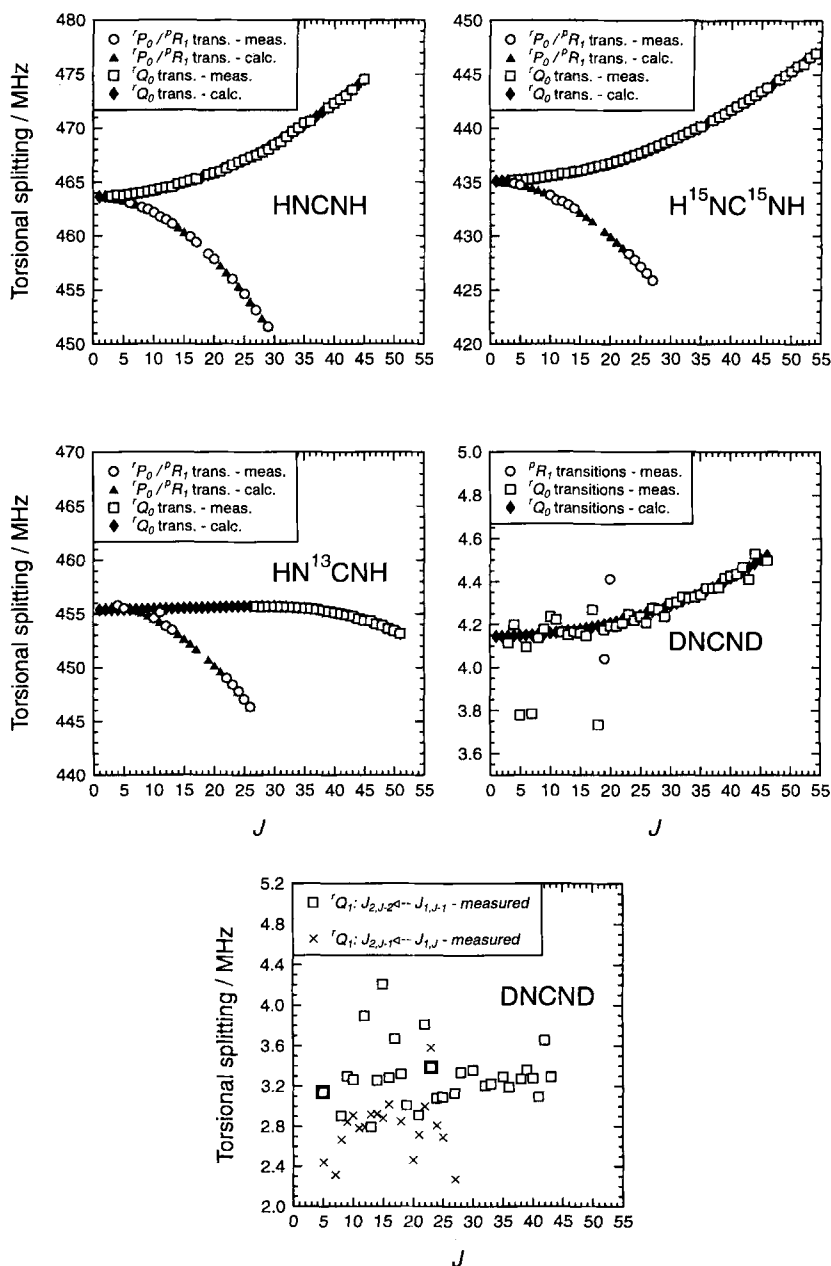


Figure 11. The  $J$  dependence of the torsional splittings of the  $Q_0$  branches of five HNCNH isotopomers. The calculated values of the splittings for the first four species were obtained either with the aid of rotational constants of the Watson Hamiltonian or with the aid of power series constants. Additionally, the measured torsional splittings of the  $Q_1$ -branch transitions of DNCND are displayed. Overlaps with strong cyanamide lines are so serious in this region, that the errors of the splitting are as large as  $\pm 1$  MHz, and therefore a dependence of the torsional splitting on  $J$  cannot be determined.

symmetric top molecule. In this case the difference between two  $F_{\text{red}}$  values with the same  $J$  and  $K_a$  quantum numbers is equal to the total observable  $K_a$ -type doubling contribution to the energy levels.

- $F_{\text{mod}}$  is calculated with the  $\Delta K_a = \pm 2$  matrix elements fixed to zero; it contains, however, the  $\Delta K_a = \pm 4$  centrifugal distortion contribution to the  $K_a$ -type doubling of the transitions. In this case  $F_{\text{red}}$  reflects the inertial asymmetry contribution to the doubling.

- $F_{\text{mod}}$  is calculated with the  $\Delta K_a = \pm 4$  matrix elements fixed to zero, but retaining the  $\Delta K_a = \pm 2$  inertial asymmetry contribution to the  $K_a$ -type doubling of the transitions. In this case  $F_{\text{red}}$  reflects the centrifugal-distortion contribution to the splitting.

It becomes clear from figure 12 that even for DNCND the  $\Delta K_a = \pm 4$  interaction contribution is the dominant term for the  $K_a$  doubling of the  $K_a = 2$  levels, and that the  $\Delta K_a = \pm 2$  and  $\Delta K_a = \pm 4$  interactions are of oppo-

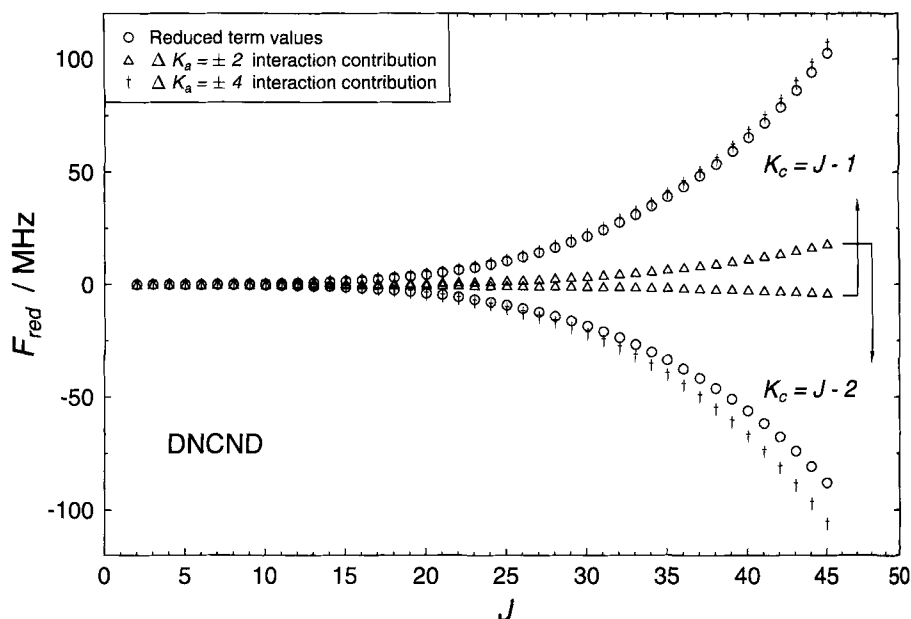


Figure 12. Reduced term values of the  $K_a = 1$  and  $K_a = 2$  rotational levels of DNCND. The reduced term values are defined as  $F_{\text{red}} = F - F_{\text{mod}}$ , where  $F_{\text{mod}}$  is a term value calculated according to the three different models discussed in the text.

site sign. A comparison with figure 8 in [5] shows that the centrifugal distortion contribution is very similar in HNCNH and DNCND, while the asymmetry contribution is 100 times larger in DNCND.

### 5. Calculation of the experimental structure

Due to the  $C_2$  symmetry of the HNCNH, the rotational constants of the four isotopomers considered here are sufficient to derive an  $r_0$ ,  $r_s$  and a  $r_m^o$  structure. For calculating the  $r_0$  structure the least squares fitting program STRFTQ3 written by Schwendeman [25] was used. This program could also calculate pseudo-Kraitchman structures (p-Kr), in which the five independent, geometrically defined coordinates (the HN and CN distances and the HNC, NCN and HN...NH angles) are fitted to differences of either the planar moments ( $\Delta P$ ) or moments of inertia ( $\Delta I$ ). All rotational constants listed in table 7 are included in the fit with the same weight. A generalized correction of the rovibrational contributions to the experimentally determined moments of inertia  $I_0$  was performed following the relation given by Demaison and Nemes [26]:

$$\log \epsilon = \log (I_0 - I_e) = 1.247 \log (I_0 - 2.651). \quad (7)$$

The so-called Laurie correction [25, 27] was also applied, which takes into account the shrinkage of a bond by substituting an atom with its heavier isotope. The HN distance was reduced by 0.003 Å when a H/D substitution was considered and the CN distance by 0.00008 Å when one of the heavier atoms was substituted. The results of all the structural fits are reported in table 8.

The high inertial symmetry of carbodiimide, with  $B \approx C$ , correlated with small  $b$  and  $c$  coordinates for

several atoms, leads to a major problem in the investigation of its  $r_s$  structure. The equations given by Nygaard [28] for double substitution produce several imaginary atomic coordinates and are therefore not applicable. However, an  $r_s$  structure was successfully calculated with the fitting program RU238 written by Rudolph [29]. The fitting procedure, introduced by Typke [30], provides the advantage of including the centre of mass and the 2nd moment conditions in the fit of the atomic coordinates in the principal axis system (PAS) of the main isotopomer. The program also offers the option of considering the correlation between rotational constants and their experimental accuracies. The Demaison/Nemes correction as well as the Laurie correction have also been applied in the  $r_s$  fits. The results are presented in table 8.

Harmony and co-workers [31–34] introduced a method for obtaining near equilibrium structures by using scaled moments of inertia,  $I_{mg}^o$ , which are calculated according to the formula

$$[I_{mg}^o]_\alpha = (2\rho_g - 1)[I_0^g]_\alpha. \quad (8)$$

The scaling factor  $\rho_g$  is obtained by

$$\rho_g = \frac{[I_s^g]_1}{[I_0^g]_1}. \quad (9)$$

$[I_0^g]_1$  is the experimentally determined ground state moment of inertia of the parent species about the  $g$  axis,  $g \in \{a, b, c\}$ , and  $I_s^g$  is the corresponding substitution moment of inertia computed from the atomic substitution coordinates resulting from the  $r_s$  fit. The index  $\alpha$  numbers the available isotopomers. After the determi-

Table 8. Internuclear distances and bond angles of HNCNH, calculated using different models.

Nr.	Method	$r_{\text{(NH)}}/\text{\AA}$	$r_{\text{(CN)}}/\text{\AA}$	$\angle\text{HNC}/^\circ$	$\angle\text{NCN}/^\circ$	$\angle\text{HN}\cdots\text{NH}/^\circ$
1	$r_0$	1.003 9(d) <sup>d</sup>	1.224 7(2)	119.10(7)	171.6(2)	89.35(9)
2	$r_0$	DEM./NEM. <sup>e</sup>	1.221 2(3)	119.0(1)	171.6(4)	89.3(1)
3	$r_0$	LAURIE <sup>f</sup>	1.223 7(1)	119.53(7)	172.7(2)	89.75(8)
4	$r_0$	DE./NE. <sup>e</sup> +LA. <sup>f</sup>	1.220 3(2)	119.38(8)	172.6(3)	89.7(1)
5	$r_0$	p – Kr, $\Delta P$	1.223(1)	119.2(9)	172.0(2)	89.6(4)
6	$r_0$	p – Kr, $\Delta I$	1.223(1)	119.3(7)	172.0(2)	89.6(4)
7	$r_s^a$	1.007 416(2)	1.224 136(9)	118.634(4)	170.66(1)	89.10(1)
8	$r_s^b$	1.007 41(2)	1.224 15(1)	118.638(2)	170.631(2)	88.99(3)
9	$r_s^c$	1.007 37(9)	1.224 22(4)	118.635(8)	170.630(4)	88.986(5)
10	$r_s^c$	DEM./NEM. <sup>e</sup>	1.004 86(9)	1.219 75(4)	118.550(8)	170.602(4)
11	$r_s^c$	LAURIE <sup>f</sup>	1.012 7(1)	1.225 33(5)	118.630(9)	170.806(5)
12	$r_s^c$	DE./NE. <sup>e</sup> +LA. <sup>f</sup>	1.010 14(9)	1.220 86(4)	118.544(8)	170.778(4)
13	$r_m^\rho$	1.013 5(6)	1.222 8(2)	118.28(9)	171.2(3)	88.8(1)
14	$r_m^\rho$	BER./HARM. <sup>g</sup>	1.014 6(3)	1.222 8(1)	118.41(4)	170.7(2)
15	$r_e$	ab + initio <sup>h</sup>	1.009 2	1.223 1	118.26	170.62

<sup>a</sup>  $r_s$  fit without considering centre of mass and 2nd moment conditions. Rotational constants are assumed to be uncorrelated.

<sup>b</sup>  $r_s$  fit including centre of mass and 2nd moment conditions. Rotational constants are assumed to be uncorrelated.

<sup>c</sup>  $r_s$  fit including centre of mass and 2nd moment conditions. Correlations between the rotational constants are considered.

<sup>d</sup> The numbers in parentheses are one standard deviation in units of the last digit quoted.

<sup>e</sup> Empirical correction of rovibrational effects according to Demaison and Nemes [26].

<sup>f</sup> Laurie correction [27].

<sup>g</sup> Correction of large vibration–rotation effects of deuterium according to Berry and Harmony [32, 35].

<sup>h</sup> [6].

nation of the  $I_{mg}^\rho$  values for all four isotopomers the data were then used to calculate the  $r_m^\rho$  structure with the aid of Schwendeman's [25] program. For molecules containing hydrogen, the  $I_m^\rho$  values obtained for the isotopomers involving H/D substitution are generally unreliable, and their utilization along with the  $I_m^\rho$  values of the heavy atom isotopomers lead to poor X–H parameters and degrade the heavy atom parameters [32]. To account for the imperfect scaling in these cases the  $I_{mg}^\rho$  values for a D-substituted species are corrected by an amount  $\Delta_g$  equivalent to an elongation  $\delta r$  of the C–D internuclear distance by an amount  $\delta r = 0.0028 \text{ \AA}$  [35]. If two equivalent hydrogen atoms are replaced by two deuterium atoms, the correction for  $I_m^\rho$  about the  $a$  axis is

$$(I_{ma}^\rho)_{\text{corr}}^D - (I_{ma}^\rho)^D = \Delta_a = \pm 2m_D \sum_{i=1}^2 (b_i \delta b_i + c_i \delta c_i). \quad (10)$$

$b_i$ ,  $c_i$  and  $\delta b_i$ ,  $\delta c_i$  are, respectively, the coordinates of the  $i$ th deuterium atom and the components of  $\delta \mathbf{r}$ . Similar equations apply for the other axes, and the plus or minus sign is selected according to whether  $\rho_g < 1$  or  $\rho_g > 1$ . For calculating the  $I_{mg}^\rho$  values the  $I_s$  values were used, which were obtained taking into account the centre of mass, the 2nd moment condition, and correlation between the rotational constants. This fit is

designated by  $r_s^c$  (number 9) in table 8. The resulting  $r_m^\rho$  structures, with and without the described correction, are included in table 8 (numbers 13 and 14).

For comparison with all these approaches to an experimental structure, the *ab initio*  $r_e$ –structure [6], extrapolated for an infinitely large basis set, is also listed in table 8 (number 15).

## 6. Discussion of the HNCNH structures

As mentioned above, the most serious problems in determining the experimental structure of HNCNH are encountered for the small coordinates of the nitrogen and carbon nuclei in the principal axis system of the main species. The carbon nucleus, lying on the  $C_2$ –symmetry axis, shows a  $b$  coordinate of  $0.0431 \text{ \AA}$ , and the absolute values of the  $b$  and  $c$  coordinates of the nitrogen nuclei are  $0.0568 \text{ \AA}$  and  $0.0565 \text{ \AA}$ , respectively. The critical limit for the determination of a substitution coordinate with still significant accuracy is given as  $\approx 0.12 \text{ \AA}$  by Rudolph [29]. This means that three of the seven coordinates which have to be determined for the HNCNH structure are below this limit. It has been observed that small coordinates can lead to non-converging  $r_s$  fits [30]. Nevertheless, in the case of carbodiimide all  $r_s$  fits converged. The corrections of the atomic coordinates in the last iteration step were smaller than



$10^{-14}$  Å even for the fit designated  $r_s^a$  in table 8 (number 7), where the centre of mass and 2nd moment conditions were not considered. The effect in the small coordinates can only be seen directly in the pseudo-Kraitchman fits. The structural parameters of those fits show statistical errors which are approximately one order of magnitude larger than for the other fits.

Because the number of equations is equal to the number of variables to be determined in the  $r_s^a$  fit, the errors given result from propagation of the experimental error. For all other fits the number of equations is larger than the number of variables. However, the  $r_s$  structural parameters do not change substantially upon including the centre of mass and 2nd moment conditions. This indicates that the small atomic coordinates are not critical for the determination of an  $r_s$  structure by the fitting method in this specific case. Nevertheless, the Costain errors  $\sigma_{(x)}$ , which are given by  $\sigma_{(x)} = K/x$ , where  $x$  is the coordinate in the PAS of the main species and  $K$  a 'constant' given by van Eijck [36], are large. Propagation of  $\sigma_x$  to the internuclear distances and angles result in:  $\delta(r_{\text{NH}}) = 0.0168$  Å,  $\delta(r_{\text{NC}}) = 0.0025$  Å,  $\delta(\angle\text{HNC}) = 1.65^\circ$ ,  $\delta(\angle\text{NCN}) = 2.50^\circ$  and  $\delta(\angle\text{HN}\cdots\text{NH}) = 3.75^\circ$ . These errors are in most cases much larger than the differences between the structures resulting from all applied models. However, the Costain formula for the uncertainty in the coordinate  $x$  is obtained under the assumption that one factor in Kraitchman's equations dominates the uncertainty of  $x$ , namely, the factor which does not depend on a difference of planar moments in the denominator. This is not necessarily true in the case of small coordinates.

The basic assumption of the  $r_m^o$  method is that the ratio between the substitution moment of inertia  $I_s^g$  about the  $g$  axis and the corresponding experimentally determined moment of inertia  $I_0^g$  is nearly constant within a few parts in  $10^{-4}$  for all isotopomers [31]. Table 9 gives the described ratio for all investigated HNCNH isotopomers. The numbers in the table indicate that the condition for the  $r_m^o$  method is fulfilled for all but the  $[I_s^a]_{\text{D}}/[I_0^a]_{\text{D}}$  ratio. After applying the correction given by equation (13) the heavy atom distance, i.e. the CN distance, should therefore approximate the true  $r_e$  value within 0.002 Å [34]. This accuracy is supported

by the excellent agreement between the  $(r_m^o)_{\text{corr}}$  value and the *ab initio* calculation. According to Helgaker *et al.* [37] the  $r_e$  distances obtained by the cc-pVQZ CCSD(T) *ab initio* method should also be in error of about 0.0022 Å. The agreement between the  $(r_m^o)_{\text{corr}}$  and the *ab initio* structure in the case of the second heavy atom parameter, the NCN angle, is also satisfying. On the other hand, it was stated that the  $(r_m^o)_{\text{corr}}$  X–H distances of polyatomic organic molecules cannot be relied upon as good approximations to  $r_e$  values. They remain sensitive to the inadequacies of the scaling and correction procedures which probably cannot account for the large vibration–rotation effects in light atoms, especially those undergoing large-amplitude, low-frequency motions [33], as in carbodiimide. Therefore, the relatively large difference of 0.0054 Å between the  $(r_m^o)_{\text{corr}}$  and *ab initio* value of the HN distance is not surprising. In this case the agreement between the  $r_s$  structure, with the application of the Demaison/Nemes and Laurie correction (number 12 in table 8), and the *ab initio* calculation is within the stated error of 0.0022 Å, which is also true for the CN distance.

However, Rudolph [29] emphasized that the Demaison/Nemes relation was not meant to be directly applicable to large molecules. The Laurie corrections suffer from the fact that rules for the effects of isotopic substitution on bond angles have not been given and probably do not exist [25]. In addition, the effect of using corrections for average internuclear distances has not been assessed. For these reasons the Laurie correction is most useful for assessing uncertainties [25]. We therefore chose the  $r_s$  structure which takes into account the centre of mass, the second moment condition, and the correlation of the rotational constants, as our best experimental approximation to the  $r_e$  structure (number 9 in table 8):  $r_{\text{NH}} = 1.0074$  Å,  $r_{\text{CN}} = 1.2242$  Å,  $\angle\text{HNC} = 118.63^\circ$ ,  $\angle\text{NCN} = 170.63^\circ$ ,  $\angle\text{HN}\cdots\text{NH} = 88.99^\circ$ . The number of digits quoted allow for errors with two significant figures according to the recommendations of Taylor *et al.* [38]. Considering the differences in the various  $r_s$  structural parameters of table 8 as a basis for assessing the errors, we conclude that the errors of the internuclear distances are in the range from 0.002 to 0.01 Å and the errors of the bond angles in the range

Table 9. Ratio  $I_s^g : I_0^g$  for all investigated HNCNH isotopomers.

Isotopomer $\alpha$	$\frac{[I_s^s]_\alpha}{[I_0^s]_\alpha}$	$\frac{[I_s^s]_\alpha}{[I_0^s]_\alpha} - \frac{[I_s^a]_\alpha}{[I_0^a]_\alpha}$	$\frac{[I_s^s]_\alpha}{[I_0^s]_\alpha}$	$\frac{[I_s^s]_\alpha}{[I_0^s]_\alpha} - \frac{[I_s^b]_\alpha}{[I_0^b]_\alpha}$	$\frac{[I_s^s]_\alpha}{[I_0^s]_\alpha}$	$\frac{[I_s^s]_\alpha}{[I_0^s]_\alpha} - \frac{[I_s^c]_\alpha}{[I_0^c]_\alpha}$
HNCNH	1.017871	0	0.998 163	0	0.998 319	0
HN <sup>13</sup> CNH	1.017873	$-0.02/10^4$	0.998 205	$-0.42/10^4$	0.998 359	$-0.40/10^4$
H <sup>15</sup> NC <sup>15</sup> NH	1.017710	$1.61/10^4$	0.998 274	$-1.11/10^4$	0.998 424	$-1.05/10^4$
DNCND	1.009910	$79.61/10^4$	0.998 393	$-2.30/10^4$	0.998 533	$-2.14/10^4$

Table 10. Comparison of molecular structures of related imines.

Structural parameter		HN=CH <sub>2</sub> [39]	HN=C=CH <sub>2</sub> <i>ab initio</i> [40]	HN=C=NH This work	HN=C=O [41]
$r_{\text{NH}}$	/Å	1.021	1.017	1.0074	0.9946
$r_{\text{CN}}$	/Å	1.273	1.234	1.2242	1.2140
$\Delta\text{HNC}$	/°	110.4	115.4	118.63	123.9
$\Delta\text{XCN}$	/°	—	174.6	170.63	172.6

from 0.1 to 0.2°. Errors of this magnitude are normally expected for an  $r_s$  structure [29].

In order to understand the structure of HNCNH in the context of simple models we collected the structural parameters of related molecules with an imine group in table 10. They are listed in a sequence showing a shrinkage of the HN bond and also a shrinkage of the CN bond, while the HNC angle is opened. Going from methanimine (H<sub>2</sub>CNH) to ketenimine (H<sub>2</sub>C<sub>2</sub>NH) a second  $\pi$  system is introduced with which the lone pair of the nitrogen can interact. Because of this interaction the CCN skeleton is not linear, which might have been expected for a system which is isoelectronic with the allene molecule. The nitrogen moves towards the second  $\pi$ -bond system, forming a CCN angle smaller than 180° and a larger HNC angle. Substitution with the more electronegative NH or O groups intensifies the interaction, leading to even stronger NH and CN bonds in carbodiimide and isocyanic acid, both of which exhibit spectra showing indications of large amplitude motions.

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