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# First-principles prediction and partial characterization of the vibrational states of water up to dissociation

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#### ABSTRACT

A new, accurate, global, mass-independent, first-principles potential energy surface (PES) is presented for the ground electronic state of the water molecule. The PES is based on 2200 energy points computed at the all-electron aug-cc-pCV6Z IC-MRCI(8,2) level of electronic structure theory and includes the relativistic one-electron mass-velocity and Darwin corrections. For  $\rm H_2^{-16}O$ , the PES has a dissociation energy of  $D_0$  = 41 109 cm<sup>-1</sup> and supports 1150 vibrational energy levels up to 41 083 cm<sup>-1</sup>. The deviation between the computed and the experimentally measured energy levels is below 15 cm<sup>-1</sup> for all the states with energies less than 39 000 cm<sup>-1</sup>. Characterization of approximate vibrational quantum numbers is performed using several techniques: energy decomposition, wave function plots, normal mode distribution, expectation values of the squares of internal coordinates, and perturbing the bending part of the PES. Vibrational normal mode labels, though often not physically meaningful, have been assigned to all the states below 26 500 cm<sup>-1</sup> and to many more above it, including some highly excited stretching states all the way to dissociation. Issues to do with calculating vibrational band intensities for the higher-lying states are discussed.

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#### 1. Introduction

The spectroscopy of the water molecule is fundamental to a wide variety of scientific and engineering applications [1]. High-resolution spectra of the water isotopologues have been extensively studied; indeed, several of us are part of an international task group devoted to developing a definitive information system containing water transitions [2]. Computational techniques to study the rotation–vibration states of water up to dissociation for a given potential energy surface (PES), although computationally demanding, have been available for more than a decade [3–5]. However, as noted in Ref. [3], the available potential energy surfaces used in

those early calculations were not designed for the high energy region approaching dissociation. The results obtained using those preliminary PESs should therefore be treated with caution. Indeed, as described below, the vibrational states predicted by the earlier studies are, in certain aspects, even qualitatively different from those presented here.

In the last decade or so several *ab initio* [6–8] and semi-theoretical [6,9–12] potential energy surfaces have been produced for the water molecule, driven in large part by the needs of spectroscopic measurements at infrared and visible wavelengths. These potentials have generally aimed at covering spectral regions up to the near-ultraviolet, which is as far as water spectra have so far been probed with one-photon spectroscopy [13,14].

Experimentally, higher regions of the water potential have started to be systematically probed by Rizzo et al. using two- [15,16] and three-photon [16–19] excitation

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schemes. These studies give information on some of the vibrational states of  $\rm H_2$   $^{16}\rm O$  all the way to dissociation and beyond, but are only sensitive to states which are accessed by the excitation scheme chosen. In particular, these experiments probe the lowest local mode pair of stretching states within each polyad plus, in many cases, some bending excitation on top of these. The ability to probe the vibrational levels all the way to dissociation represents a major advance; however, so far only a minority of the states have yielded themselves to observation.

The lack of direct observation of higher vibrational states of water does not necessarily mean that such states are unimportant. For example, recent observations of cometary emission spectra suggest that highly excited vibrational states of water are naturally populated in comets [20], although the mechanism for this remains a matter of speculation.

In this paper we present a complete list of computed bound vibrational energy levels for water almost all the way to dissociation obtained using a new, accurate, global, mass-independent, *ab initio* potential energy surface and variational-like nuclear motion treatments employing exact kinetic energy operators. We also include a discussion of issues related to calculating vibrational band intensities at ultraviolet wavelengths. Finally, we give our best estimates for the (approximate) associated vibrational normal mode quantum numbers where possible.

#### 2. Computational details

#### 2.1. Electronic structure calculations

Our new electronic structure calculations were briefly reported in a communication by Grechko et al. [17]. These calculations used an atom-centered, sextuple-zeta, so-called aug-cc-pCV6Z Gaussian basis set from the correlation-consistent family of Dunning [21]. Unlike the quintuple-zeta and lower sets in this series, this extended basis set is not yet completely standardized. For H, we used the standard aug-cc-pV6Z basis. For O, the aug-ccpV6Z part of the basis is also standard [22,23] and the "C" (core-correlation) functions are available via EMSL as "ccpCV6Z(old)" [24]. The basis set employed consists of s, p, d, f, g, h and i functions for oxygen and s, p, d, f, g and h functions for hydrogen, yielding for H<sub>2</sub>O a total of 562/533 uncontracted/contracted primitive gaussian functions. Note that in previous studies [7,8] on the PES of water we found full augmentation of the basis with diffuse functions (aug) to be of particular importance. No basis set extrapolation to the complete basis set (CBS) limit [25,26] was attempted.

Electron correlation was treated at the internally contracted multireference configuration interaction (IC-MRCI) level [27] with a renormalised Davidson correction (+Q) [28,29]. The calculations were performed using the MOLPRO electronic structure package [30]. Unlike during generation of our earlier CVRQD surface [7,8], employing a valence-only treatment at the IC-MRCI level, here all

electrons were directly included in the correlation treatment

A series of test calculations were performed to study the effect of varying the MRCI reference (complete active) space, which included comparison with small basis (ccpVDZ) full configuration interaction (FCI) calculations at several geometries, in particular those toward dissociation, and comparisons with a series of single-reference coupled cluster (CC) treatments, up to quadruple excitation (CCSDTQ), at and around equilibrium. The FCI and CC calculations employed the code MRCC [31]. Our final choice for the reference space for the MRCI computations can be designated as (8,2) in  $C_s$  point-group symmetry, which means 8 A' and 2 A" orbitals were chosen to be freely occupied by the 8 valence electrons. This choice extends the (6,2) complete active space used in a number of previous studies [6-8.32]. Calculations were performed at 2 200 geometries which were chosen to thoroughly sample the PES of the ground electronic state of water up to its first dissociation limit. The IC-MRCI(8,2)+O/aug-ccpCV6Z energies were augmented by relativistic corrections calculated as the sum of one-electron mass-velocity and Darwin (MVD1) terms [33,34]. These energies were fitted to a flexible functional form [35] suitable for generating a dissociative PES; an electronic version of the resulting surface has been given previously [17].

# 2.2. Nuclear motion computations

Nuclear motion calculations were performed on our new potential energy surface using several codes developed either in London or in Budapest, all based on exact kinetic energy operators. The masses (in u) adopted in all computations were  $m_{\rm H}$  = 1.007276 and  $m_{\rm O}$  = 15.990526.

An augmented version of the DVR3D program suite [36,37] employing orthogonal Radau coordinates and a discrete variable representation [38,39] of the Hamiltonian was used to obtain the energy values reported in this paper. Calculations were performed using previously optimized [16] spherical-oscillator [40] functions for the radial and Legendre functions for the angular motions (120 and 70 of them, respectively, during the final run). Increasing the size of the final, contracted Hamiltonian matrix from 15000 to 20000 changed the VBOs by no more than  $0.1 \, \text{cm}^{-1}$ . These calculations converged energy levels to better than 1 cm<sup>-1</sup>, with the exception of an even state at about  $40570\,\mathrm{cm}^{-1}$  which can tentatively be identified as (0260) and which shows considerable sensitivity to the number of angular grid points used. Convergence of the computed energy levels was checked using the latest variant of the code DOPI3 [41], D2FOPI [42]. DVR3D and D2FOPI agreed to better than 1 cm<sup>-1</sup> for all the VBOs reported. Altogether 1 150 VBOs are supported by our computations up to 41 083 cm<sup>-1</sup>. This energy range includes the last observed pair of vibrational state below dissociation; above this the open nature of the PES (see below) should lead to an increasingly diffuse set of vibrational states which we have not attempted to systematically characterize. The last bound state assigned by our present computations of even symmetry is at  $41\,082.75\,\mathrm{cm}^{-1}$ , it is (1900) in normal-mode notation (*vide infra*). The last assigned bound state of odd symmetry, (1801), is at 41082.78 cm<sup>-1</sup>.

Approximate vibrational band intensities were calculated using the Eckart frame [43] and an updated version of the code DIPJO [36]; as a check parallel transitions intensities were also calculated taking three times the  $I = 0 \rightarrow 1$  transition intensity calculated using the code DIPOLE3 [37]. Our calculations initially used the recently developed CVR dipole moment surface (DMS) of water [32]. For vibrational states lying at infrared and visible frequencies, our calculations gave satisfactory agreement with observed band intensities [44]. However, these calculations gave surprisingly strong band intensities 30 000 cm<sup>-1</sup>. Calculations with the dipole moment surface of Schwenke and Partridge (SP2000) [45] gave qualitatively the same results but significantly different intensities for individual states. Previous analyses [32,45] have shown that small imperfections in the fits can lead to the calculation of over-intense transitions for bending overtones at visible wavelengths. We suspected that similar effects were causing our calculations to overestimate the vibrational band intensities to higher vibrational states. Tests performed with low-order polynomial fits to the dipole moment surface supported this conjecture.

As a final test we tried the DMS of Gabriel et al. [46]. This surface, which was demonstrated by the authors to give very good results for the vibrational band intensities of low-lying states, retains no terms higher than fourth-order in the fit. We note that the fit of Gabriel et al. is only valid for a rather limited range of nuclear geometries; however, as here we only consider transitions from the

vibrational ground state this should not be a problem. This calculation gave significantly lower vibrational band intensities than equivalent calculations using the CVR or SP2000 surfaces, see Fig. 1. Our conclusion from these studies is that it is not possible at present to make secure predictions of the vibrational band intensities, or indeed of the intensity of individual rotation–vibration transitions, going from the ground vibrational state directly to states lying in the ultraviolet.

Vibrational energy levels and wave functions were also determined with the DEWE program system [47,48] to help the assignment of normal mode labels. DEWE is based on the DVR of the Eckart–Watson Hamiltonian [49,50] and allows the exact inclusion of a PES represented in arbitrarily chosen coordinates. The DEWE computations employed a direct-product Hermite–DVR grid with 20 and 75 grid points for the bending and stretching vibrational degrees of freedom, respectively.

# 2.3. Wave function plots

Generation of the plots corresponding to real wave functions employed a locally developed code [51]. The input to the code is the file with the wave functions obtained with DVR3D for J=0. All the 1 150 wave function plots are given as PDF files in the Supplementary Material. The plots of real wave functions contain two-dimensional (2D) cuts of the functions presented in the three usual Radau coordinates,  $r_1$ ,  $r_2$ , and  $\theta$ . We note that for water Radau coordinates are very close to the more standard bondlength–bondangle coordinates.

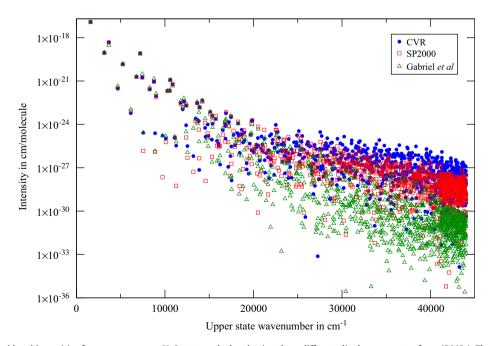


Fig. 1. Vibrational band intensities for even-symmetry  $H_2O$  states calculated using three different dipole moment surfaces (DMSs). The plot shows that for transition energies above  $\approx 30\,000\,\mathrm{cm}^{-1}$  recent DMSs based on high-order polynomial expansions such as CVR and SP2000 considerably overestimate band intensities. See text for details and references.

Wave function plots for selected states are presented in Figs. 2-5. In Figs. 2-5 the positive and negative values are in red and blue, respectively. Change of colors therefore indicates a node in the wavefunction. Each plot also gives a contour corresponding to the classical turning point of the potential at the eigenenergy of the state under consideration. This contour is useful for assessing the chaotic nature of the state: classically chaotic states are ergodic meaning that they sample all available phase space (see Fig. 5). Quantum mechanically an ergodic state would be expected to sample the whole of the available PES [52]. We note that for states near dissociation the potential is in principle open to dissociation, although the molecule has insufficient energy to occupy the OH vibrational ground state and therefore to dissociate. This is a feature of all polyatomic systems and has been found to lead to interesting asymptotic structures in the vibrational states [53].

#### 3. Results

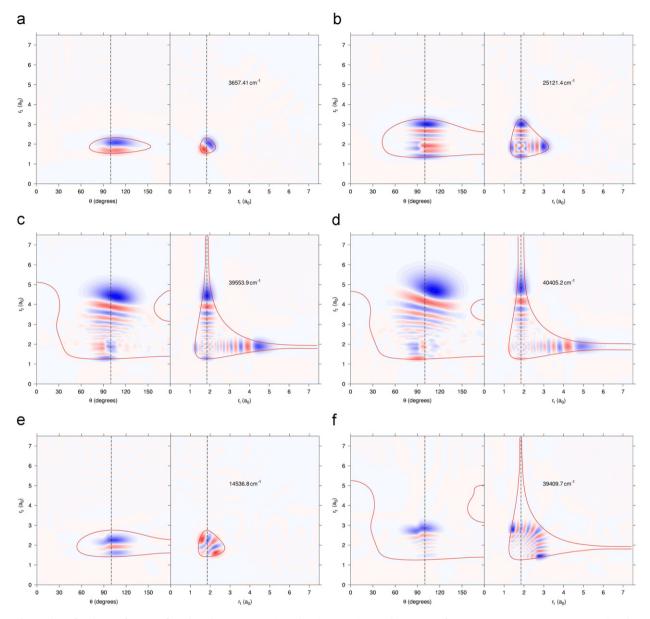
Table 1 presents our calculated vibrational band origins (VBOs) for H<sub>2</sub> <sup>16</sup>O up to 25 200 cm<sup>-1</sup>; where possible, they are compared to available experimental data [16-18,54,55]. It should be noted that as yet transitions to only about 15% of all the VBOs have actually been observed, so the majority of the reported levels represent our predictions for VBOs of water. Table 1 also lists approximate vibrational quantum numbers and information related to several assignment schemes for many of the calculated VBOs. A complete set of VBOs determined in this study are given as Table S1 in the Supplementary Material. Those VBOs lying above 25 200 cm<sup>-1</sup> for which we could determine approximate normal mode labels are given in Table 2; this table covers all those higher-lying VBOs which have experimental counterparts.

#### 3.1. Quality of the PES

In the past we have advocated the use of the composite focal-point analysis (FPA) approach [25,26,56,57] for the computation of the PESs and DMSs of smaller molecules [7,8,32,58]. There were three reasons for deviating slightly from this approach during this study. First, previous electronic structure studies on the ground-state PES of water [7,8] and on its barrier to linearity [59,60] proved that one must use as large of a Gaussian basis set as possible for treating appropriately the large-amplitude excited bending and stretching motions of water. This limits considerably the accessible choices for the level of electronic structure theory applicable to generate the large number of energy points needed for the surfaces. Second, it turned out that high quality treatment of electron correlation is less important than the choice of the Gaussian basis. Third, by far the most important socalled "small effects" [26] are the core and the MVD1 relativistic energy corrections, covered by the present study in a single computation per energy point. The present PES, when used in a variational nuclear motion computation with exact kinetic energy operators, can reproduce excellently the measured vibrational energy levels up to the dissociation limit. The differences between the computed and the experimental levels can be seen in Fig. 6.

There are a number of salient features of the present PES which are worth discussing. The Born-Oppenheimer equilibrium structure,  $r_{\rm e}^{\rm BO}$ , has  $r_{\rm e}({\rm OH})$ =0.95782 Å and  $\theta_e(HOH) = 104.46^{\circ}$ . These values are extremely close to the best available  $r_{\rm e}^{\rm BO}$  estimates, corresponding to the composite CVRQD PES of  $H_2$  <sup>16</sup>O [61], namely  $r_e$ =0.95785 Å and  $\theta_e = 104.50^\circ$ . The independent quadratic force constants, in units of aJ  $\mathring{A}^n$  rad<sup>2-n</sup>, n=0-2, corresponding to the global PES are  $f_{rr}$  = 8.453,  $f_{rr'} = -0.105$ ,  $f_{\theta\theta} = 0.704$ , and  $f_{r\theta} = 0.257$ . These valence coordinate force constants are also very close to those characterizing the CVRQD PES  $(f_{rr}=8.460, f_{rr'}=-0.103, f_{\theta\theta}=0.703, and f_{r\theta}=0.258).$ These structural and force field parameters have been used, along with the masses, to construct the normal coordinates in DEWE and the normal mode decomposition. Finally, we note that the  $D_0$  dissociation energy of  $H_2$ <sup>16</sup>O for the present *ab initio* PES differs from the wellestablished experimental value, 41 145.94(15) cm<sup>-1</sup> [19], by only 37 cm<sup>-1</sup>. This difference comes about as follows. The aug-cc-pCV6Z IC-MRCI+Q(8,2) PES has  $D_e$ =43 951  $cm^{-1}$ . The MVD1 correction to this is  $-50 cm^{-1}$ . The zero-point vibrational energy (ZPVE) correction for H<sub>2</sub>O, corresponding to the present PES, is  $-4638.6 \,\mathrm{cm}^{-1}$ (compare to  $-4638.3 \, \text{cm}^{-1}$  corresponding to the CVRQD PES), while that of OH is  $+1~847~cm^{-1}$ .

The results presented above and below and the benchmark ab initio calculation of  $D_0$  by Ruscic et al. [62] together with the results of previous studies on water built upon the FPA philosophy can be used to provide pointers on how one could improve the present global potential. There are at least four main "effects" not considered in our computational treatment. First, the aug-cc-pCV6Z basis set used, while very large, carries a basis set incompleteness error (BSIE) due to its finite size and especially to the use of fixed exponents and centers [63]. We estimate that the correction to  $D_0$  due to BSIE may be as much as  $+80 \,\mathrm{cm}^{-1}$ . Second, estimation of the correlation energy is based on the IC-MRCI approach which, with respect to full-CI, introduces a noticeable error. As several observations made during this study suggest, this error could be as large as 40-50 cm<sup>-1</sup>. For example, using the cc-pVDZ basis which permits going to the FCI limit, further away from equilibrium the error of IC-MRCI+O using the enlarged active space is of the order of 40 cm<sup>-1</sup>. Third, though scalar relativistic effects, calculated as expectation values of the mass-velocity and one-electron Darwin operators, were included in the calculation, these do not account for all relativistic effects even for the molecule of water [64,65]. Of particular concern when approaching the first dissociation limit, not considered in the present study, is the effect of spin-orbit interaction on  $D_0$ . It was reported by Ruscic et al. [62] to be 38 cm<sup>-1</sup>. Fourth, the electronic structure treatment employed does not go beyond the Born-Oppenheimer separation of nuclear and electronic degrees of freedom. An improvement in the accuracy of the ab initio calcula-



**Fig. 2.** Plots of real wave functions for selected pure symmetric and antisymmetric stretching states of even symmetry. For symmetric stretches the nodes can be observed along the  $r_1$  and  $r_2$  internal coordinates. For antisymmetric stretches the nodes are along  $r_1-r_2$ . Each plot also gives a contour corresponding to the classical turning point of the potential at the eigenenergy of the state under consideration. The plots and thus the states are labelled as  $(v_1 \ v_2 \ v_3)$ , for details see the text. (a) (100), (b) (800), (c) (1600), (d) (1700), (e) (004), (f) (0012).

tions could be made by inclusion of adiabatic and non-adiabatic corrections. For example, the adiabatic correction contributes about  $+40\,\mathrm{cm^{-1}}$  to the dissociation energy. The non-adiabatic contribution is hard to estimate but we would expect it to partially cancel the adiabatic correction. In summary, even at the very high level of electronic structure theory employed in this study it seems that we are gaining accuracy from a fortuitous cancellation of some of the effects mentioned and not taken into account. Further advance in the *ab initio* computation of the complete set of VBOs for water would benefit only from a joint consideration of all these factors

as individual inclusion of any one of them might even make some of our predictions appear worse.

# 3.2. Vibrational band origins

Some estimates of the accuracy of our first-principles predictions can be made by comparison of the computed VBOs with the observed ones. Fig. 6 shows that not only the differences are small but also the trends are systematic. For the lower-energy VBOs there are clear series corresponding to bending overtones with no

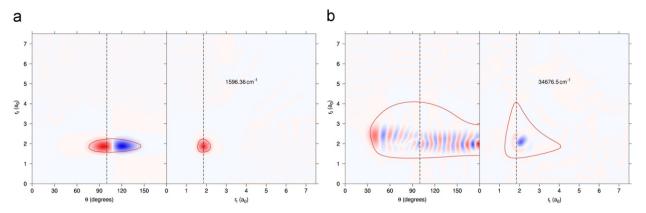


Fig. 3. Wave function plots for pure bending states. For a detailed description see the legend to Fig. 2. (a) (010), (b) (0220).

stretching excitation, one quanta of stretch (two series), two quanta of stretch (two series) and three quanta of stretch (two series). These series show that the errors increase quite rapidly with increasing bending excitations: it is actually well known that this motion is hard to treat correctly ab initio [66,67]. At higher frequencies there are very few observed states with significant bending excitation. In contrast the errors in the calculated pure stretching VBOs remains small, 1 cm<sup>-1</sup> or less, up to about 25 000 cm<sup>-1</sup>; above this value the error becomes positive and then, in the last 2000 cm<sup>-1</sup> below dissociation, sharply negative. At dissociation the error in the highest stretching states, (1700)-(1801), approaches  $40\,\mathrm{cm}^{-1}$ , the amount our PES underestimates  $D_0$ . The systematic nature of these errors suggest that there is scope to improve the calculated VBOs by fitting the PES to the observed levels. Predicted transition frequencies based on the predicted VBOs can be improved either by making manual corrections [68] or by performing a fit to the experimental data starting from our PES.

It can be expected that vibrational band intensities decrease systematically with increasing energy of the VBOs since it is well known that overtone bands become increasingly weaker. However, this decrease is not monotonic with respect to the energy. Instead, some stretching VBOs decrease less rapidly than others as could be anticipated from the fact that direct excitation of states with eight quanta of stretch has already been achieved using cavity ring-down spectroscopy [14].

The absorption of light by water vapor at ultraviolet wavelengths could be of atmospheric importance, particularly at wavelengths below the onset of the main ozone absorptions. The ability to estimate the magnitude of this absorption should be within the scope of our present study but in practice we found that the presently available dipole moment surfaces are not suitable for such calculations. The harmonic model of molecular vibrations suggests that the intensity of a fundamental transition depends principally on the first derivative of the dipole moment at equilibrium taken along the vibrational mode being excited. Within this model, the intensity of the n th overtone then depends on the (n+1)th derivative of the dipole surface. What we have found is that the best dipole

moment surfaces available for water [32,45] do not give reliable tenth or higher derivatives in the region of equilibrium geometry. Resolving this problem will be the subject of future work. Before leaving this topic we should note that our concern about the use of the available dipole moment surfaces for studying high overtones does not invalidate the transition intensity calculations presented by Grechko et al. [17]. In this work intensities were given for individual rotation–vibration transitions which reached states near the dissociation limit. However, these transitions link states differing by 8 or less quanta of stretch, for which no difficulties have been identified.

# 3.3. Vibrational state assignments

The variational-type nuclear motion procedures employed in this study are based on the use of rigorous quantum numbers. For the vibrational states of  $\rm H_2$   $^{16}\rm O$  this means that the states are only distinguished by a single quantum number: the permutational symmetry of the state, which can be viewed as  $(-1)^{\nu_3}$  for states with a normal-mode assignment, where  $\nu_3$  is the vibrational normal-mode quantum number corresponding to the antisymmetric stretch. Other, approximate vibrational quantum number labels have to be assigned by other means.

Previous calculations [3], which used a different PES, suggested that at energies approaching  $D_0$  the vibrational states of water become highly irregular with little systematic underlying structure. This situation is not promising for assigning approximate vibrational labels. Indeed, there is no guarantee that such labels exist [52], although there are clear rules when they can be considered rigorous [69].

In some contrast to this picture, a previous joint experimental-theoretical study of water spectra up to dissociation [17] observed spectra which were strongly structured with a relatively small number of fairly strong lines. This observation was mirrored by the associated transition intensity calculations. For these states, at least, it proved straightforward to assign vibrational labels [17]

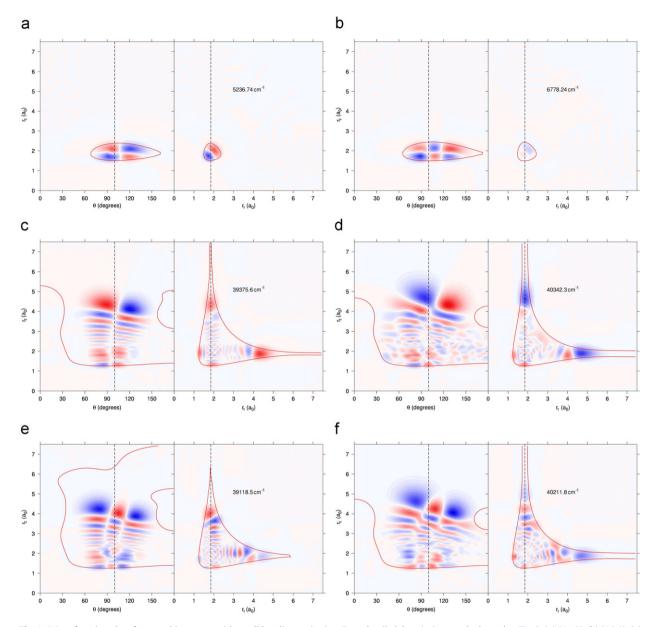


Fig. 4. Wave function plots for stretching states with small bending excitation. For a detailed description see the legend to Fig. 2. (a) (110), (b) (120), (c) (1510), (d) (1610), (e) (1420), (f) (1520).

using the PES described here. These results demonstrate that the VBOs given by our PES are at least partially structured. This motivated us to try and find a set of labels for as many vibrational states as possible, extending a previous study of some of the authors [70] employing a different PES.

To make normal mode quantum number assignments, five strategies were investigated.

 Energy decomposition. The simplest method is to use the energy dependence based on the normal mode quantum numbers. This can be expressed as

$$E(v_1, v_2, v_3) = v_1 v_1 + v_2 v_2 + v_3 v_3, \tag{1}$$

where  $v_1$ ,  $v_2$ ,  $v_3$  ( $v_1$   $v_2$   $v_3$ ) and  $v_1$ ,  $v_2$ ,  $v_3$  are the vibrational quantum numbers and the vibrational fundamentals for the so-called symmetric stretch, bend, and antisymmetric stretch normal modes, respectively. This method works well up to about  $12\,000\,\mathrm{cm^{-1}}$  and no additional information is required to label the VBOs. We note that at the higher end of this region the stretching motions of water are already much closer described by a local mode rather than a normal mode picture [71]. Above this energy, the density of states increases and there are often several candidates for a given set of vibrational normal mode quantum numbers. Furthermore, as the normal mode decomposition (NMD) analysis presented below

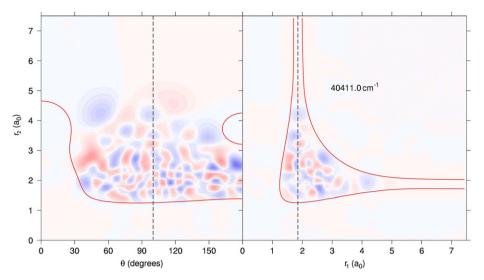


Fig. 5. Wave function plot for an ergodic state. For a detailed description see the legend to Fig. 2.

shows, even when normal mode labels can be allocated based on this scheme, their physical relevance can clearly be questioned.

- 2. Wave function plots. Inspection of 2D cuts of the real wave function plots along appropriately chosen coordinates sometimes gives valuable qualitative information on the level and is particularly useful for states of more or less pure stretching or bending character. However, plots cannot be used to get quantitative information on mixtures without significant manipulation [72], which was not attempted here. It would also be hard to identify multiply excited levels, e.g., (5 5 5), by visual inspection even if such a state was fairly harmonic.
- 3. Mean values of squares of the Jensen coordinates  $\langle s_i^2 \rangle$ i=1,2,3 [73]. These expectation values were calculated by the program XPECT3 [37]. Normal mode-like labels can be created by rounding the expectation values to the nearest integers. For a harmonic oscillator this label should be proportional to the normal mode quantum number  $v_i$ . Indeed, for the lowestlying levels these labels reproduce the normal mode labels of the most dominant harmonic oscillator functions contributing to the "exact" state. However, for a Morse-like real oscillator the well is wider and the mean values grow quicker than  $v_i$ . Thus,  $\langle s_i^2 \rangle$  for the two stretching modes grows too fast. The largest 'xpect' stretching quantum numbers are for states (n00), decreasing for other members of the same polyad. The  $v_2$  'xpect' quantum number, on the other hand, grows slower than the "harmonic" quantum number. By extrapolating the behavior of these 'xpect' quantum numbers from states with known labels to unknown ones one can guess normal mode labels. The resulting labels are given in Table 1 under the heading "xpect".
- 4. Perturbing the bending part of the PES. Assignment of vibrational quantum numbers to levels with large  $\nu_2$  values is particularly difficult. Therefore, we analyzed

the change in the energy of a level with the inclusion of an artificial term in the PES depending only on the angle. For this we actually used a term designed to correct the height of the barrier to linearity of water [60,66,67]. Changes in a level's energy were found to be practically independent of the stretching quantum numbers  $v_1$  and  $v_3$ . For values of the bending quantum number  $v_2$  between 0 and 7, shifts in the levels were found to be directly proportional to  $v_2$ , greatly facilitating the assignment procedure. Consequently, comparing changes observed for unlabelled levels with those of already labelled ones we could determine the values of  $v_2$  for these states. The maximum change occurred near  $v_2$ =8. For higher  $v_2$  values the changes were found to be apparently only indicators that  $v_2$  has a high value.

5. Normal mode decomposition (NMD). Another assignment procedure used in this work was based on the normal mode decomposition [74] of the variational wave function. In the NMD the variationally computed, normalized vibrational wave functions,  $\psi_i$ , are characterized by the square of their overlaps with the normalized harmonic oscillator wave functions,  $\phi_{\nu}^{\rm HO}$ , expressed in terms of normal coordinates corresponding to the actual PES,

$$\chi_{iv} = |\langle \psi_i | \phi_v^{\text{HO}} \rangle|^2, \tag{2}$$

where  $v=(v_1 \ v_2 \ v_3)$  is a composite index. For each vibrational state the sum of the NMD entries equals one,  $\sum_{j=0}\chi_{ij}=1$ , as the wave function can be expressed as a linear combination of harmonic oscillator wave functions. The wave functions obtained with the DEWE program [47,48] are given in terms of normal coordinates, so the computation of their overlap with the actual harmonic oscillator wave functions is straightforward. As the bending modes of water are very anharmonic and higher bends cannot be treated with the DEWE protocol, the method provides meaningful results only for states with  $v_2=0$  values, as could

**Table 1**The lowest 250 vibrational band origins (computed as part of this study and experimental (Expt.)), their symmetries (Sym.), the corresponding polyad number P, defined as  $P = 2\nu_1 + \nu_2 + 2\nu_3$ , and approximate vibrational assignments corresponding to the PES developed in this paper.<sup>a</sup>

ab initio 1596.37 3154.45 3657.41 3755.93 4670.86 5236.75 5332.87	Expt.  1594.75 3151.63 3657.05 3755.93 4666.79	+ + + +	P 1 2	Normal	xpect	$W_{ m stre}$	Plot
3154.45 3657.41 3755.93 4670.86 5236.75	3151.63 3657.05 3755.93	+		(0.1.0)			
3657.41 3755.93 4670.86 5236.75	3657.05 3755.93	•	2	(010)	[010]	_	G
3755.93 4670.86 5236.75	3755.93	+	2	(020)	[020]	-	G
4670.86 5236.75			2	(100)	[100]	0.98	G
5236.75	4666.70	-	2	(001)	[001]	0.99	G
	4000.73	+	3	(030)	[030]	-	G
5332.87	5234.97	+	3	(110)	[110]	-	G
3332.07	5331.27	_	3	(011)	[011]	-	G
6139.62	6134.01	+	4	(040)	[040]	-	G
6778.25	6775.09	+	4	(120)	[120]	_	G
6874.32	6871.52	_	4	(021)	[021]	_	G
		+	4			0.96	G
		_	4			0.97	G
		+	4				G
		+	5			_	G
		+				_	G
		_				_	G
		+				_	G
		_				_	Ğ
		+				_	G
		+				_	G
		+				_	G
		_				_	G
		+				_	G
		+				_	G
		_				_	G
		+				_	G
		+				0.93	G
		_					G
		_					G
		·					G
						0.30	G
		•					G
		_					G
						_	G
		T				-	G
		_				_	G
		•				_	G
						_	
						_	G
	6874.32 7202.45 7250.44 7250.44 72444.91 7550.09 8278.89 8377.92 8763.71 8808.85 8808.85 9001.59 9731.26 9839.13 10 100.36 10 287.81 10 331.65 10 524.53 10 600.38 10 613.82 10 869.62 11 032.04 11 109.00 11 250.15 11 267.44 11 772.53 11 817.51 12 012.03 12 140.91 12 152.60 12 392.96	7202.45         7201.54           7250.44         7249.82           7444.91         7445.05           7550.09         7542.44           8278.89         8273.98           8377.92         8373.85           8763.71         8761.58           8808.85         8869.95           9001.59         9000.14           9731.26         9724.3°           9839.13         9833.59           10100.36         10086.05           10287.81         10284.37           10331.65         10328.73           10524.53         10521.76           10600.38         10599.69           10613.82         10613.35           10869.62         10868.88           11032.04         11032.40           1109.00         -           11250.15         11242.78           11267.44         11254.00           11772.53         11767.39           11817.51         11813.20           12012.03         12007.78           12140.91         12139.32           12152.60         12151.25	7202.45         7201.54         +           7250.44         7249.82         -           7444.91         7445.05         +           7550.09         7542.44         +           8278.89         8273.98         +           8377.92         8373.85         -           8763.71         8761.58         +           8808.85         8807.00         -           8880.85         8869.95         +           9001.59         9000.14         +           9731.26         9724.3°         +           9839.13         9833.59         -           10 100.36         10 086.05         +           10 287.81         10 284.37         +           10 331.65         10 328.73         -           10 524.53         10 521.76         +           10 600.38         10 599.69         +           10 613.82         10 613.35         -           10 869.62         10 868.88         +           11 109.00         -         +           11 250.15         11 242.78         -           11 267.44         11 254.00         +           11 772.53         11 767.39         +	7202.45       7201.54       +       4         7250.44       7249.82       -       4         7444.91       7445.05       +       4         7550.09       7542.44       +       5         8278.89       8273.98       +       5         8377.92       8373.85       -       5         8763.71       8761.58       +       5         8808.85       8807.00       -       5         8880.85       8869.95       +       6         9001.59       9000.14       +       5         9731.26       9724.3°       +       6         9839.13       9833.59       -       6         1010.36       10086.05       +       7         10287.81       10284.37       +       6         10331.65       10 328.73       -       6         10524.53       10 521.76       +       6         10600.38       10 599.69       +       6         10613.82       10 613.35       -       6         10869.62       10 868.88       +       6         11 109.00       -       +       7         11 267.44	7202.45         7201.54         +         4         (200)           7250.44         7249.82         -         4         (101)           7444.91         7445.05         +         4         (002)           7550.09         7542.44         +         5         (050)           8278.89         8273.98         +         5         (130)           8377.92         8373.85         -         5         (031)           8763.71         8761.58         +         5         (210)           8808.85         8807.00         -         5         (111)           8808.85         8869.95         +         6         (060)           9001.59         9000.14         +         5         (012)           9731.26         9724.3°         +         6         (041)           9839.13         9833.59         -         6         (041)           10100.36         10 086.05         +         7         (070)           10287.81         10 284.37         +         6         (121)           10524.53         10 521.76         +         6         (022)           10600.38         10 599.69         +	7202.45         7201.54         +         4         (200)         [200]           7250.44         7249.82         -         4         (101)         [101]           7444.91         7445.05         +         4         (002)         [102]           7550.09         7542.44         +         5         (050)         [050]           8278.89         8273.98         +         5         (130)         [130]           8377.92         8373.85         -         5         (031)         [031]           8763.71         8761.58         +         5         (210)         [210]           8808.85         8869.95         +         6         (060)         [060]           9001.59         9000.14         +         5         (012)         [112]           9731.26         9724.3°         +         6         (140)         [140]           9839.13         9833.59         -         6         (041)         [041]           1010.36         10086.05         +         7         (070)         [070]           10287.81         10284.37         +         6         (220)         [221]           10600.38         1059.	7202.45         7201.54         +         4         (200)         [200]         0.96           7250.44         7249.82         -         4         (101)         [101]         0.97           7444.91         7445.05         +         4         (002)         [102]         0.98           7550.09         7542.44         +         5         (050)         [050]         -           8278.89         8273.98         +         5         (050)         [130]         -           8377.92         8373.85         -         5         (031)         (031]         -           8763.71         8761.58         +         5         (210)         [210]         -           8808.85         8807.00         -         5         (111)         [111]         -           8808.85         8869.95         +         6         (060)         [060]         -           9001.59         900.14         +         5         (012)         [112]         -           931.26         9724.3¹         +         6         (140)         [140]         -         -           9839.13         9833.59         -         6         (120)

Table 1 (continued)

	Energy		Sym.	Approxima	te assignments			
	ab initio	Expt.		P	Normal	xpect	$W_{ m stre}$	Plot
40	12 409.80	12 407.66	+	7	(112)	[211]	_	G
41	12 545.97	12 533.72	+	9	(090)	[170]	_	_
42	12 566.13	12 565.01	_	7	(013)	[1 1 3 ]	_	G
43	12 596.07	12 586*	_	8	(061)	[061]	_	G
44	13 211.95	13 205.1*	+	8	(240)	[240]	_	G
45	13 262.09	13 256.2*	_	8	(141)	[241]	_	G
46	13 459.42	13 453.7°	+	8	(042)	[142]	_	_
47	13 643.57	13 640.5*	+	8	(320)	[322]	_	G
48	13 654.79	13 652.66	_	8	(221)	[322]	_	G
49	13 673.82	13 661*	+	9	(170)	[160]	_	_
50	13 828.00	13 828.28	+	8	(400)	[303]	0.88	G
51	13 830.61	13 830.94	_	8	(301)	[303]	0.86	G
52	13 848.43	13 835.37	_	9	(071)	[171]	-	G
53	13 869.65	-	+	10	(0100)	[170]	_	G
54	13 914.40	13 910.88	+	8	(122)	[321]	_	G
55	14 068.53	14 066.20		8	(023)	[123]	_	G
56	14222.08	14 221.16	+	8	(202)	[401]	0.90	G
57	14318.68	14318.81	· _	8	(103)	[303]	0.92	G
58	14536.82	14537.50	+	8	(004)	[204]	0.93	-
59	14 588.61	-	+	9	(250)	[250]	-	
60	14 655.88	14 647.98	'	9	(151)	[251]	_	G
61	14830.83	14047.58	+	9	(052)	[161]	=	G
62	14892.78	_	+	10	(180)	[161]	_	- G
63	14 999.08	_	т	10	(081)	[171]	_	G
64	15 111.86	15 108.24	+	9	(330)	[332]	_	G
65	15 122.37	15 119.03	т	9	(231)	[332]	_	G
66	15 305.69	-	+	11	(0110)	[170]	_	G
67	15 345.07	- 15 344.50	+	9		[413]	_	G
68				9	(410)		=	G
69	15 348.16	15 347.96 15 377.7 <sup>*</sup>	_	9	(311)	[413]	-	G
	15 382.63		+		(132)	[331]	<del>-</del>	
70 71	15 538.43	15 534.71	+	9 9	(033)	[133]	-	G G
71 72	15 744.91	15 742.80	+	9	(212)	[411]	-	G
73	15 833.74	15 832.78	_		(113)	[313]	-	_
	15 882.40	15 871 <sup>*</sup>	+	10	(260)	[260]	-	-
74	15 979.45	15 969 <sup>*</sup>	<del>-</del>	10	(161)	[261]	-	G
75	16 047.67	16 046 <sup>*</sup>	+	9	(014)	[214]	=	G
76	16 085.44	-	+	11	(190)	[271]	-	_
77	16 173.80	-		11	(091)	[171]	-	G
78	16225.40	16215*	+	10	(062)	[162]	-	G
79	16 539.55	16 534.5°	+	10	(340)	[342]	-	G
80	16551.02	16 546.32	_	10	(241)	[342]	-	G
81	16 802.10	16 795.9*	+	10	(142)	[332]	-	G
82	16 822.18	16821.63	-	10	(321)	[423]	-	G
83	16 824.94	16823.32	+	10	(420)	[423]	-	G
84	16 833.78	<del>-</del>	+	12	(0 12 0)	[170]	<del>-</del>	G
85	16897.29	16 898.4°	+	10	(500)	[414]	0.68	G
86	16897.70	16 898.84	-	10	(401)	[414]	0.68	G
87	16 972.71	16 967.6*	_	11	(043)	[243]	_	G

								_
88	17 151.43	17 139 <sup>*</sup>	+	11	(270)	[260]	_	G
89	17 230.92	17 227.38	+	10	(222)	[421]	-	G
90	17 242.04	-	_	11	(171)	[261]	-	G
91	17 314.59	17 312.55	-	10	(123)	[323]	-	G
92	17 396.09	-	+	12	(1100)	[171]	_	G
93	17 456.97	_	_	12	(0101)	[171]	_	G
94	17 458.36	17 458.21	+	10	(302)	[502]	0.86	G G
95	17 494.79	17 495.53	_	10	(203)	[403]	0.86	G
96	17 504.05	_	+	11	(072)	[172]	_	G
97	17 528.34	17 526.3°	+	10	(024)	[224]	_	G
98	17 748.00	17 748.11	+	10	(104)	[403]	0.88	G
99	17 918.45	17 911.6°	+	11	(350)	[341]	-	G
100	17 934.32	17 928 <sup>*</sup>		11	(251)	[342]		G G G
101	17 947.30	-		10	(005)	[205]	0.91	G
102	18 170.61	18 162*	+	11	(152)	[351]	-	-
102	18 267.50	18 265.82	т	11	(331)	[433]	_	G
			_		, ,		<del>-</del>	G
104	18 268.81	18 267.1°	+	11	(430)	[433]	-	G
105	18 346.16	-	+	12	(280)	[270]	_	G
106	18 357.81	18 350.3	_	11	(053)	[253]	_	-
107	18 391.95	18 392.78	+	11	(510)	[524]	_	G
108	18 392.51	18 393.31	_	11	(411)	[524]	-	G
109	18 435.01	<del>-</del>	+	13	(0 13 0)	[180]	-	G
110	18 436.95	-	_	12	(181)	[272]	-	G
111	18 657.57	_	+	12	(082)	[172]	-	G
112	18 680.15	_	+	11	(232)	[431]	-	G G
113	18 761.97	18 758.63	-	11	(133)	[333]	-	G
114	18 817.64	_	+	13	(1110)	[280]	-	G
115	18830.31	-	_	13	(0111)	[181]	_	G
116	18 957.16	18 955.7 <sup>*</sup>	+	11	(312)	[512]	_	G G
117	18 980.19	18 977.2 <sup>*</sup>	+	11	(034)	[334]	_	G
118	18 990.12	18 989.96	_	11	(213)	[413]	-	G
119	19 232.48	19 223.5°	+	12	(360)	[351]	_	G
120	19 243.31	_	+	11	(114)	[413]	_	G G
121	19258.70	19 250 <sup>*</sup>	_	12	(261)	[352]	_	G
122	19 432.72	_	_	11	(015)	[315]	_	G
123	19 451.40	_	+	12	(162)	[362]	_	G
124	19 564.31	_	_	13	(191)	[272]	_	G G
125	19 587.41	_	+	13	(290)	[371]	_	G
126	19 680.63	19 677.8°	+	12	(440)	[433]		G G
127	19 682.08	19677.8		12	(341)	[433]		G
127	19 729.44	19 720.2°	_	12	(063)	[253]	_	G G G
129	19 779.54	19 720.2	+	12	(600)		- 0.87	G
			+		` '	[605]	0.87	G
130	19 779.78	19781.10	_	12	(501)	[605]	0.88	G
131	19 793.51	- 10.004 T*	+	13	(092)	[182]	-	G
132	19864.02	19 864.7°	+	12	(520)	[524]	_	G
133	19 864.47	19 865.28	-	12	(421)	[524]	-	G
134	20 039.28	-	+	14	(0 14 0)	[170]	-	G
135	20 085.94	-	+	12	(242)	[541]	-	G
136	20 173.47	-	-	12	(143)	[343]	=	G
137	20 288.91	-	-	14	(0121)	[181]	-	G
138	20 321.93	-	+	14	(1120)	[280]	-	G
139	20 390.73	-	+	12	(044)	[334]	-	G
140	20 421.53	-	+	12	(322)	[522]	-	G

Table 1 (continued)

	Energy		Sym.	Approxima	te assignments			
	ab initio	Expt.		P	Normal	xpect	$W_{ m stre}$	Plot
141	20 443.66	20 442.78	_	12	(223)	[523]	-	G
142	20 502.98	-	+	13	(370)	[361]	-	G
143	20 526.34	-	_	13	(271)	[362]	-	G
144	20 532.45	20 534.5°	+	12	(402)	[604]	0.74	G
145	20 541.96	20 543.13	_	12	(303)	[614]	0.70	G
146	20702.48	_	+	12	(124)	[423]	_	G
147	20731.38	_	+	13	(172)	[362]	_	G
148	20813.03	_	_	14	(1101)	[282]	_	G
149	20846.69	_	+	14	(2100)	[371]	_	G
150	20 888.28	_	_	12	(025)	[325]	_	Ğ
151	20 906.28	_	+	12	(204)	[602]	0.82	Ğ
152	21 016.05			13	(073)	[263]	-	G
153	21 016.42	_	+	14	(0102)	[282]	_	G
154	21 040.68	-	т	12		[504]	0.83	G
		- 21.052*	<del>-</del>		(105)			
155	21 056.29	21 052*	+	13	(450)	[443]	-	G
156	21 057.74	21 053*	_	13	(351)	[443]	-	G
157	21 221.16	21 221.57	+	13	(610)	[625]	_	G
158	21 221.48	21 221.83	_	13	(511)	[625]	-	G
159	21 272.94	-	+	12	(006)	[306]	0.87	G
160	21 312.99	21 312 <sup>*</sup>	+	13	(530)	[535]	-	F
161	21 313.69	21 314.45	_	13	(431)	[535]	_	G
162	21 438.30	-	+	13	(252)	[551]	_	G
163	21 539.81	-	_	13	(153)	[353]	_	G
164	21 639.05	-	+	15	(0150)	[270]	-	G
165	21 715.72	<del>-</del>	+	14	(380)	[371]	_	G
166	21 720.50	_	_	14	(281)	[372]	_	G
167	21 770.36	_	+	13	(054)	[344]	_	G
168	21 820.43	_	_	15	(0131)	[281]	_	G
169	21 847.82	_	+	13	(332)	[532]	_	G
170	21 869.17	21 867 <sup>*</sup>	_	13	(233)	[533]	_	G
171	21 926.33	21007	+	15	(1130)	[281]		G
172	21 984.51	_	+	14	(182)	[371]		G
173	22 006.94		+	13	(412)	[624]		G
174	22 000.94	22 015 <sup>*</sup>	т	13	(313)	[624]	_	G
174			_				=	
	22 131.40	-	+	13	(134)	[533]	_	G
176	22 142.51	-	_	15	(1111)	[282]	-	G
177	22 176.51	-	+	15	(2110)	[281]	_	G
178	22 234.85	-	-	14	(083)	[273]	-	G
179	22 315.57	-	-	13	(035)	[335]	-	G
180	22 338.62	-	+	15	(0112)	[281]	-	G
181	22 382.53	22 376	+	14	(460)	[553]	-	G
182	22 383.89	22 377*	-	14	(361)	[453]	-	G
183	22 387.51	-	+	13	(214)	[622]	-	G
184	22 507.61	22 508 <sup>*</sup>	-	13	(115)	[514]	-	G
185	22 528.67	22 529.29	+	14	(700)	[807]	0.92	G
186	22 528.80	22 529.44	_	14	(601)	[807]	0.92	G
187	22 628.47	22 626*	+	14	(620)	[635]	=	G

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188	22 630.29	22 629 <sup>*</sup>	_	14	(521)	[635]	-	G
189	22 708.35	-	+	14	(262)	[552]	-	G
190	22 732.30	-	+	13	(016)	[416]	_	G
191	22 738.51	-	_	14	(441)	[635]	-	G
192	22 744.61	_	+	14	(540)	[545]	_	G
193	22810.73	_	_	15	(291)	[373]	_	G
194	22 885.31	_	+	15	(390)	[471]	_	G
195	22 925.68	_	_	14	(163)	[362]	_	_
196	23 067.43	_	+	15	(192)	[363]	_	_
197	23 157.70	_	+	14	(064)	[363]	_	_
198	23 236.60	_	+	14	(342)	[642]	_	G
199	23 257.08		·	14	(243)	[543]		G
200	23 263.52	_	+	16	(1140)	[370]	_	G
201	23 322.45	_	т	15			<u>-</u> -	-
		=	_		(093)	[283]		
202 203	23 366.38	- 23 401*	+	16	(0141)	[282]	- 0.74	G G
	23 400.04		+	14	(502)	[715]	0.74	
204	23 404.07	23 405.4°		14	(403)	[715]	0.76	G
205	23 466.78	_	+	14	(422)	[624]	-	G
206	23 474.45	-	_	14	(323)	[625]	_	G
207	23 524.78	-	+	14	(144)	[543]	-	G
208	23 559.74	-	+	16	(2120)	[381]	-	G
209	23 605.07	-	_	16	(1 12 1)	[382]	-	G
210	23 652.29	-	+	16	(0 16 0)	[372]	-	-
211	23 663.89	-	+	15	(470)	[372]	-	-
212	23 665.77	-	_	15	(371)	[463]	-	G
213	23 713.51	-	_	14	(045)	[345]	-	G
214	23 787.05	-	+	16	(0122)	[281]	_	-
215	23 825.74	-	+	14	(224)	[722]	_	G
216	23 935.12	23 934 <sup>*</sup>	_	14	(125)	[625]	_	G
217	23 944.48	23 942 <sup>*</sup>	+	15	(710)	[726]	_	G
218	23 947.68	23 947 <sup>*</sup>	_	15	(611)	[7 1 7]	_	G
219	23 960.85	_	+	15	(272)	[553]	_	_
220	23 977.25	_	+	14	(304)	[813]	0.73	G
221	23 998.57	_	_	15	(451)	[554]	_	_
222	24015.44	_	+	15	(550)	[645]	_	G
223	24033.57	_	_	16	(2101)	[463]	_	_
224	24038.37	_	_	14	(205)	[714]	0.69	G
225	24091.22	_	+	16	(3100)	[482]	_	G
226	24 142.76	_	_	15	(531)	[646]	_	G
227	24 143.05	_	+	15	(630)	[646]	_	F
228	24 162.05	_	+	14	(026)	[426]	<del>-</del>	G
229	24 177.98	_	_	15	(173)	[463]	_	-
230	24177.56	_	+	14	(106)	[605]	0.78	- G
231	24 2 3 1 2 . 6 5	_	+	16	(1102)	[372]	-	F
232	24 435.42	_	+	15	(074)	[363]	<del>-</del>	F
233	24 466.99	_	т.					F
233		=	_	16	(0103)	[283]	- 0.02	
	24508.22	-	_	14	(007)	[407]	0.83	G
235	24 582.36	-	+	15	(352)	[642]	-	F
236	24 609.05	-	-	15	(253)	[543]	=	G
237	24798.22	-	+	15	(512)	[734]	-	-
238	24807.86	-	-	15	(413)	[725]	-	_
239	24826.40	-	+	17	(2130)	[461]	-	F
240	24 856.94	-	_	16	(381)	[463]	-	-
241	24 869.48	-	+	15	(154)	[543]	-	-

	Energy		Sym.	Approximate	Approximate assignments			
	ab initio	Expt.		Ь	Normal	xpect	$W_{ m stre}$	Plot
242	24 883.66	1	+	16	(480)	[472]	1	H
243	24910.71	1	1	15	(333)	[634]	I	1
244	24 915.51	1	+	15	(432)	[634]	1	1
245	24 925.35	1	1	17	(1131)	[372]	ı	ш
246	25 077.92	I	1	15	(055)	[345]	ı	ی
247	25 094.22	I	+	17	(1150)	[282]	I	ш
248	25 121.48	25120*	+	16	(800)	[10010]	0.89	ц
249	25 121.50	25120.28	ı	16	(701)	[10010]	0.89	ڻ
250	25 156.46	I	+	17	(0132)	[471]	ı	ı

<sup>a</sup> Sym., permutation symmetry characterized also by  $(-1)^{1/2}$  for states with a normal mode assignment. Normal, normal mode assignment  $(v_1, v_2, v_3)$  based on the normal mode quantum numbers  $v_1, v_2$ , and  $v_3$  for the symmetric stretch, bend, and antisymmetric stretch modes. xpect, expectation values of the squares of the Jensen coordinates rounded to the nearest integer (see text).  $W_{\rm stre}$ , stretching weight in the wave function, calculated according to Eq. (3). The band centers marked with an asterisk were estimated from levels with J > 0. Under "Plot", G, good and F, fair, -, plot does not yield labels

be seen when comparing the eigenvalues obtained with DVR3D and DEWE (the latter are not presented here).

The mean values of the squares of the Jensen coordinates provide meaningful labels for the lowest vibrational levels. The labels obtained agree with the chosen normal mode labels up to (002) at 7441.91 cm<sup>-1</sup>. Above this state there are still a lot of meaningful labels but they increasingly start to break down. Of course, the meaningful labels are the same as those produced by the energy decomposition scheme.

According to the NMD analysis, the lowest-lying vibrational energy levels are characterized by a single, dominant harmonic oscillator function, and thus their assignment is absolutely unambiguous. For example, the vibrational state at  $4670.86 \,\mathrm{cm}^{-1}$  and described as (030)with a polyad number P = 3, where  $P = 2v_1 + v_2 + 2v_3$ , is the following mixture of the harmonic oscillator basis states: 0.88(030) + 0.05(110) + 0.03(040). At the same time, for higher-lying levels, with polyad numbers P =4-6, i.e., the second and third stretching overtone and combination levels, the identification of a single, dominant harmonic oscillator wave function is not possible. Already for these states the attachment of a single  $(v_1, v_2)$  $v_3$ ) normal mode label cannot be carried out without overwhelming and steadily increasing ambiguities. For example, it seems completely natural to give the labels (400) and (202) to the states computed at 13828 and 14 222 cm<sup>-1</sup>. Nevertheless, the NMD analysis clearly indicates that the harmonic oscillator basis functions (400) and (202) have basically zero weight in both exact wave functions. Both wave functions are extremely heavy mixtures of the basis states, the largest NMD contributions are 0.20(102), 0.12(302), and 0.11(300) for the state at  $13\,828\,\text{cm}^{-1}$ , and  $0.21(5\,0\,0)$ ,  $0.18(3\,0\,0)$ , and 0.11(102) for the state at  $14222 \,\mathrm{cm}^{-1}$ . It is of particular interest to note that Choi and Light [75], based on their wave function analysis, reversed the order of the (4 0 0) and (202) states. The reverse order is supported by the 'xpect' values of this study (Table 1). Nevertheless, due to the indications of the NMD analysis, we decided to keep the original, intuitively appealing ordering.

Thus, it is important to emphasize that the distribution of the possible normal mode labels  $(v_1 \ v_2 \ v_3)$  is not rooted in firm theory but simply in tradition. Apart from the fundamentals and maybe low-lying overtones and combination levels, the linear combination of the "exact" variational wave function in terms of harmonic oscillator basis functions is typically not dominated by a single harmonic oscillator function. Despite the strong mixing of harmonic oscillator wave functions in excited vibrational levels, we attempted to employ the NMD analysis to justify the distribution of some normal mode labels. The intention was to identify "pure" stretching states,  $(v_1 \ 0 \ v_3)v_1, v_3 = 0, 1, \ldots$  In order to do this, the "stretching weight",  $W_{\rm stre}$ , and the "number of stretching quanta",  $N_{\rm stre}$ , were introduced as

$$W_{\text{stre}} = \sum_{\nu_1, \nu_3} \chi_{i, \nu = (\nu_1, 0, \nu_3)}$$
 (3)

 Table 2

 High-energy vibrational states with vibrational quantum number assignments.

323     27 504.57     +     (53 2)     27 502.       324     27 539.87     -     (80 1)     0.51     27 536.       325     27 544.20     +     (90 0)     0.50     27 540.       326     27 573.10     -     (51 3)     27 569.       327     27 576.89     +     (61 2)     27 574.       328     27 601.62     -     (1103)       329     27 603.94     +     (2102)       330     27 659.15     +     (008)     wfp     0.79	No.	$\tilde{v}_{\mathrm{theo}}$	par	Label	Plot	$W_{\rm stre}$	$\tilde{v}_{\mathrm{Exp}}$
253   25 265.78   *   (28 2)					wfp		
254   2529729							
255			+				
256							
257							
258							
259					wfp		
260			_				
262	260	25 372.71	+		wfp		
263					wfp		
264					-		
265   25519.30					wfp		
266         25,519,70         -         (54 1)           267         25,565,47         +         (0.84)         wfp           268         25,606,61         +         (0.84)         wfp           270         25,712,28         -         (0.113)         wfp           271         25,735,97         +         (1.16)         wfp           272         25,860,53         +         (36,2)         wfp           273         25,891,48         -         (26,3)           274         25,939,41         -         (0.17)           275         26037,33         -         (391)         wfp           276         26037,33         -         (391)         wfp           277         26133,81         +         (602)         0.78           278         2614,83         -         (503)         wfp         0.80           279         26150,66         +         (164)         w         0.80           281         2620,72         +         (442)         o         c           282         2631,74         +         (292)         wfp         c           283         26357,01         +							
267			+		wip		
268         25 606 61         +         (084)         wfp           269         25 688 97         +         (1112)         -           270         25 712 28         -         (0113)         wfp           271         25 735 597         +         (116)         wfp           272         25 891.48         -         (263)           274         25 939.41         -         (017)           275         26033.48         +         (490)         wfp           276         26037.33         -         (391)         wfp           277         26133.81         +         (602)         0.78           278         26 141.83         -         (503)         wfp         0.80           280         26 197.10         -         (343)         -         (442)         -           281         26 267.2         +         (442)         -			+		wfp		
269   25889.7					•		
271         25735.97         +         (116)         wfp           272         2580.03         +         (362)         wfp           273         25891.48         -         (263)           275         26032.48         +         (490)         wfp           276         26037.33         -         (391)         wfp           277         26133.81         +         (602)         wfp         0.78           278         26150.66         +         (164)         -         26207.72         -         4813.83         -         (503)         wfp         0.80         -         28120.067.72         +         (442)         -         48222.02         -         4233         ok         -         4823         ok			+		•		
272         2580.53         +         (362)         wfp           274         2599.941         -         (017)           275         26033.48         +         (490)         wfp           276         26037.33         -         (391)         wfp           278         26141.83         -         (503)         wfp         0.80           279         26150.66         +         (164)         wfp         0.80           281         26 197.10         -         (343)         -         (343)         -           281         26 521.24         +         (442)         - <t< td=""><td>270</td><td>25 712.28</td><td>_</td><td>(0113)</td><td>wfp</td><td></td><td></td></t<>	270	25 712.28	_	(0113)	wfp		
273         25891.48         —         (263)         —         2274         2593941         —         (017)         —         275         26033.48         +         (490)         wfp         —         277         26133.81         +         (602)         Wfp         0.78         —         0.71         —         0.71         0.80         —         0.71         —         0.71         —         0.71         —         0.71         —			+	(116)	wfp		
274         25939.41         —         (017)         wfp         276         26037.33         —         (391)         wfp         277         26133.81         +         (602)         0.78         278         26141.83         —         (503)         wfp         0.80         279         26150.66         +         (164)         —         0.80         279         26150.66         +         (164)         —         0.80         279         26150.66         +         (164)         —         0.80         281         26206.72         +         (442)         —         4231         —         4282         26321.84         +         (522)         283         26321.92         —         (423)         ok         —         4292         wfp         —         4282         283         26321.92         —         (193)         wfp         —         4282         283         26321.92         —         (193)         wfp         —         4282         283         26517.91         +         (292)         wfp         —         4282         285         26393.20         —         (191)         Wfp         —         287         26432.58         —         (0141)         —         4882         26517.			+		wfp		
275         26033.48         +         (490)         wfp           276         26037.33         -         (391)         wfp         0.78           277         26133.81         +         (602)         wfp         0.80           279         26150.66         +         (164)         .         .         .           280         26197.10         -         (343)         .			_				
276							
277         26 133.81         +         (602)         0.78           278         26 1418.3         -         (503)         wfp         0.80           279         26 150.66         +         (164)         -         (343)         -         -           280         26 197.10         -         (343)         - <t< td=""><td></td><td></td><td>+</td><td></td><td></td><td></td><td></td></t<>			+				
278       26 141.83       -       (50 3)       wfp       0.80         279       26 150.66       +       (16 4)       -       (34 3)         281       26 2006.72       +       (44 2)       -       -       (28 2)       -       -       (42 3)       ok       -			+		WIP	0.78	
279					wfp		
280			+	, ,			
282       26321.84       +       (522)         283       26321.92       -       (423)       0k         284       26357.01       +       (292)       wfp         285       26393.20       -       (193)       wfp         287       26432.58       -       (0141)         287       26432.58       -       (0141)         288       26517.96       -       (711)       wfp         290       26517.96       -       (711)       wfp         291       26598.25       +       (3120)         292       26603.02       +       (0142)         293       26712.53       -       (2121)         299       26736.33       +       (730)         300       2674.80       +       (094)         301       26825.82       -       (0161)       wfp         302       26832.12       +       (404)       0.47         303       26832.12       +       (160)       wfp         304       2866.40       -       (305)       wfp       0.61         308       2694.74       +       (324)       wfp         30	280		_				
283       26 321.92       -       (42 3)       ok         284       26 357.01       +       (29 2)       wfp         285       26 393.20       -       (19 3)       wfp         286       26 428.83       +       (2140)       ***         287       26 432.58       -       (0141)       ***         288       26 517.91       +       (8 10)       wfp         290       26 517.96       -       (7 11)       wfp         291       26 598.25       +       (3120)         292       26 603.02       +       (0142)         293       26 712.53       -       (2121)         297       26 736.36       -       (6 31)         299       26 736.34       +       (7 30)         300       26 774.80       +       (094)         301       26 825.82       -       (0161)       wfp         302       26 839.27       +       (1160)       wfp         304       26 866.40       -       (305)       wfp       0.61         305       26 905.74       +       (324)       *         309       26 942.74       +       <		26 206.72	+	(442)			
284			+				
285			_				
286					•		
287					wip		
288							
290			+		wfp		
292			_		-		
293		26 598.25	+	(3120)			
297			+				
299			_				
300							
301							
302					wfp		
303			+		wip	0.47	
305					wfp		
305	304		_			0.61	
309			+	(324)			
310       27 099.79       -       (0123)         313       27 105.02       +       (1122)         314       27 213.93       +       (4100)       wfp         316       27 243.07       -       (3101)       wfp         317       27 263.00       +       (206)       wfp       0.66         318       27 279.09       +       (0180)       wfp         319       27 338.95       -       (027)       wfp         320       27 422.86       -       (107)       wfp       0.69         322       27 499.54       -       (433)       27 497.         323       27 504.57       +       (532)       27 502.         324       27 539.87       -       (801)       0.51       27 536.         325       27 544.20       +       (900)       0.50       27 540.         326       27 573.10       -       (513)       27 569.         327       27 576.89       +       (612)       27 574.         328       27 601.62       -       (1103)         329       27 603.94       +       (2102)         330       27 659.15       + <td< td=""><td></td><td></td><td>-</td><td></td><td></td><td></td><td></td></td<>			-				
313       27 105.02       +       (1122)         314       27 213.93       +       (4100)       wfp         316       27 243.07       -       (3 10 1)       wfp         317       27 263.00       +       (206)       wfp       0.66         318       27 279.09       +       (0180)       wfp         319       27 338.95       -       (027)       wfp         320       27 422.86       -       (107)       wfp       0.69         322       27 499.54       -       (433)       27 497.         323       27 504.57       +       (532)       27 502.         324       27 539.87       -       (801)       0.51       27 536.         325       27 544.20       +       (900)       0.50       27 540.         326       27 573.10       -       (513)       27 569.         327       27 576.89       +       (612)       27 574.         329       27 601.62       -       (1103)         329       27 603.94       +       (2102)         330       27 659.15       +       (008)       wfp       0.79			+		wfp		
314       27213.93       +       (4100)       wfp         316       27243.07       -       (3101)       wfp         317       27263.00       +       (206)       wfp       0.66         318       27279.09       +       (0180)       wfp         319       27338.95       -       (027)       wfp         320       27422.86       -       (107)       wfp       0.69         322       27499.54       -       (433)       27497.         323       27504.57       +       (532)       27502.         324       27539.87       -       (801)       0.51       27536.         325       27544.20       +       (900)       0.50       27540.         326       27573.10       -       (513)       27569.         327       27576.89       +       (612)       27574.         329       27601.62       -       (1103)         329       27603.94       +       (2102)         330       27659.15       +       (008)       wfp       0.79			_				
316       27243.07       -       (3101)       wfp         317       27263.00       +       (206)       wfp       0.66         318       27279.09       +       (0180)       wfp         319       27338.95       -       (027)       wfp         320       27422.86       -       (107)       wfp       0.69         322       27499.54       -       (433)       27497.         323       27504.57       +       (532)       27502.         324       27539.87       -       (801)       0.51       27536.         325       27544.20       +       (900)       0.50       27540.         326       27573.10       -       (513)       27569.         327       27576.89       +       (612)       27574.         328       27601.62       -       (1103)         329       27603.94       +       (2102)         330       27659.15       +       (008)       wfp       0.79					wfp		
317       27263.00       +       (206)       wfp       0.66         318       27279.09       +       (0180)       wfp         319       27338.95       -       (027)       wfp         320       27422.86       -       (107)       wfp       0.69         322       27499.54       -       (433)       27497.         323       27504.57       +       (532)       27502.         324       27539.87       -       (801)       0.51       27536.         325       27544.20       +       (900)       0.50       27540.         326       27573.10       -       (513)       27569.         327       27576.89       +       (612)       27574.         328       27601.62       -       (1103)         329       27603.94       +       (2102)         330       27659.15       +       (008)       wfp       0.79			_		-		
318 27279.09 + (0180) wfp 319 27338.95 - (027) wfp 320 27422.86 - (107) wfp 322 27499.54 - (433) 27497. 323 27504.57 + (532) 27502. 324 27539.87 - (801) 0.51 27536. 325 27544.20 + (900) 0.50 27540. 326 27573.10 - (513) 27569. 327 27576.89 + (612) 27574. 328 27601.62 - (1103) 329 27603.94 + (2102) 330 27659.15 + (008) wfp 0.79			+	, ,	-	0.66	
319       27338.95       -       (027)       wfp         320       27422.86       -       (107)       wfp       0.69         322       27499.54       -       (433)       27497.         323       27504.57       +       (532)       27502.         324       27539.87       -       (801)       0.51       27536.         325       27544.20       +       (900)       0.50       27540.         326       27573.10       -       (513)       27569.         327       27576.89       +       (612)       27574.         328       27601.62       -       (1103)         329       27603.94       +       (2102)         330       27659.15       +       (008)       wfp       0.79					-	2.30	
320       27422.86       -       (107)       wfp       0.69         322       27499.54       -       (433)       27497.         323       27504.57       +       (532)       27502.         324       27539.87       -       (801)       0.51       27536.         325       27544.20       +       (900)       0.50       27540.         326       27573.10       -       (513)       27569.         327       27576.89       +       (612)       27574.         328       27601.62       -       (1103)         329       27603.94       +       (2102)         330       27659.15       +       (008)       wfp       0.79			-		-		
323     27 504.57     +     (53 2)     27 502.       324     27 539.87     -     (801)     0.51     27 536.       325     27 544.20     +     (900)     0.50     27 540.       326     27 573.10     -     (513)     27 569.       327     27 576.89     +     (612)     27 574.       328     27 601.62     -     (1103)       329     27 603.94     +     (2102)       330     27 659.15     +     (008)     wfp     0.79			_		wfp	0.69	
324     27539.87     -     (801)     0.51     27536.       325     27544.20     +     (900)     0.50     27540.       326     27573.10     -     (513)     27569.       327     27576.89     +     (612)     27574.       328     27601.62     -     (1103)       329     27603.94     +     (2102)       330     27659.15     +     (008)     wfp     0.79			_				27 497.2°
325     27544.20     +     (900)     0.50     27540.       326     27573.10     -     (513)     27569.       327     27576.89     +     (612)     27574.       328     27601.62     -     (1103)       329     27603.94     +     (2102)       330     27659.15     +     (008)     wfp     0.79			+			0.51	27 502.66
326       27 573.10       -       (513)       27 569.         327       27 576.89       +       (612)       27 574.         328       27 601.62       -       (1103)         329       27 603.94       +       (2102)         330       27 659.15       +       (008)       wfp       0.79			-				27 536.3
327 27 576.89 + (612) 27 574. 328 27 601.62 - (1103) 329 27 603.94 + (2102) 330 27 659.15 + (008) wfp 0.79			+			0.50	27 540.69
328 27 601.62 - (1103) 329 27 603.94 + (2102) 330 27 659.15 + (008) wfp 0.79			-				
329 27 603.94 + (2 10 2) 330 27 659.15 + (0 0 8) wfp 0.79							2/3/4,31
330 27659.15 + (008) wfp 0.79							
					wfp	0.79	
331 27864.31 + (0104)	331	27 864.31	+	(0104)			

Table 2 (continued)

No.	$\tilde{v}_{ ext{theo}}$	par	Label	Plot	$W_{\rm stre}$	$\tilde{v}_{\rm Exp}$
337	27 892.06	+	(820)	wfp		
338	27 892.11	_	(722)	wfp		
339	27 942.95	+	(2150)	wfp		
341	27 949.51	-	(1151)	wfp		
342	28 072.52	+	(740)			
344	28 073.18	_	(641)	wfp		
345	28 107.85	+	(0152)	wfp		
346	28 206.95	+	(334)			
350	28 243.29	_	(235)			
352	28 288.20	+	(056)	wfp		
353	28 337.68	+	(414)			
356	28 355.33	_	(315)			
357	28 521.13	+	(136)	wfp		
362	28 548.19	_	(0171)	wfp		
363	28 629.96	+	(1170)			
364	28 695.77	+	(216)			
365	28 713.75	-	(037)			
367	28719.39	_	(603)		0.69	
368	28719.63	+	(702)		0.74	
369	28 837.71	-	(117)	wfp		
373	28 893.69	+	(622)			28 890.1
375	28 894.10	_	(523)			28 890.6°
376	28 937.73	+	(910)	wfp		28 934.14
377	28 938.05	_	(811)			28 934.4°
378	29 061.60	+	(018)	wfp		
383	29 107.45	+	(0190)	wfp		
386	29 242.09	_	(731)	wfp		
388	29 242.09	+	(830)	wfp		
389	29 372.33	+	(750)	wfp		
391	29 416.59	+	(2160)			
393	29 418.69	_	(1161)	wfp		
394	29 620.12	+	(504)	-	0.51	
403	29 626.92	_	(405)		0.45	
404	29 707.71	+	(0162)			
406	29 735.62	+	(424)			
407	29 751.17	_	(325)			
408	29817.04	+	(1000)	wfp	0.84	29 810.85
411	29 817.06	_	(901)	wfp	0.84	29810.87
412	30 063.49	+	(712)	wfp		
420	30 069.57	_	(613)			
421	30 082.30	+	(226)	wfp		
423	30 170.74	+	(306)	p	0.46	
426	30 195.28	_	(207)	wfp	0.50	
429	30 245.18	+	(632)	Wip	0.50	
430	30246.13	_	(533)			
431	30 260.15		(0181)	wfp		
432	30 282.81	+	(920)	wip		
434	30 285.03			wfp		
434 435		- +	(821)	wfp		
	30 314.96		(1 18 0)			
136	30 430.56	+	(028)	wfp		
140	30 501.35	+	(108)	wfp		
142	30 562.79	+	(840)	<b>C</b>		
144	30 562.92	_	(741)	wfp	0.75	
145	30722.49	_	(009)	wfp	0.75	
450	30893.89	+	(354)			
456	30 964.32	+	(0200)	wfp		
459	30 970.29	+	(514)			
460	30 979.20	_	(415)			
461	31 077.13	+	(1010)	wfp		31 071.57
462	31 078.62	_	(911)	wfp		31 070.7°
463	31 105.59	+	(434)			
466	31 120.52	_	(335)			
467	31 161.54	+	(2170)			
468	31 211.96	+	(802)	wfp	0.6	31 207.09
472	31 212.02	_	(703)	wfp	0.6	31 212.3°
473	31 287.70	-	(1171)			
475	31 386.89	+	(722)			
478	31 392.14	_	(623)			
480	31 510.67	+	(0172)			
484	31 585.29	+	(316)			

Tabl	<u>~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ </u>	(continued)	

No.	$\tilde{v}_{\mathrm{theo}}$	par	Label	Plot	$W_{\rm stre}$	$\tilde{v}_{\rm Exp}$
490	31 607.88	-	(831)	wfp		
491	31 654.75	-	(217)			
494	31 778.63	+	(038)	wfp		
498 504	31 892.22 31 917.22	+	(118) (1001)	wfp	0.81	31 905.3 <sup>*</sup>
506	31 917.23	+	(1100)	wfp	0.81	31 909.68
507	31 996.00	_	(0191)	wfp		
509	32 094.02	_	(019)	wfp		
513	32 160.43	<del>-</del>	(505)	c	0.68	
515 516	32 169.87 32 190.74	+	(604) (1190)	wfp wfp	0.37	
519	32 298.84	+	(524)	wip		
523	32 361.87	+	(1020)	wfp		
525	32 364.45	_	(921)	wfp		
526	32 564.47	+	(812)			
534	32 564.56	_	(713)			
535 538	32 676.74 32 677.03	+	(633) (732)			
539	32 832.36	+	(0210)	wfp		
547	32 884.73	+	(406)	p	0.69	
549	32 903.53	+	(940)			
551	32 922.45	_	(307)	wfp	0.55	
554	32 927.07	<del>-</del>	(941)	wfp		
556 563	33 102.82	+	(048) (1110)	wfp		33 144.71
566	33 152.66 33 152.70	<u>-</u>	(1011)	wfp wfp		33 144.71 33 144.7°
567	33 182.07	+	(2180)	wfp		33 144.7
570	33 228.07	+	(208)	r.	0.43	
571	33 249.83	-	(1181)	wfp		
572	33 421.02	_	(029)			
580	33 427.19	+	(902)	wfp	0.69	
581 582	33 428.88 33 476.08	_ _	(803) (109)		0.63 0.37	
584	33 488.04	+	(614)		0.57	
585	33 489.77	_	(515)			
586	33 533.51	+	(0182)			
587	33 621.49	-	(931)			
591	33 629.32	+	(1030)	wfp	0.71	
592 597	33 702.68 33 757.10	+	(0010) (0201)	wfp wfp	0.71	
598	33 842.41	_	(1101)	wfp	0.78	33 835.22
601	33 842.42	+	(1200)	wfp	0.78	33 835.25
602	33 891.54	+	(822)	wfp		
605	33 891.64	_	(723)	wfp		
606	33 973.74	+	(1200)	wfp		
613 630	34 344.87 34 368.60	+	(416) (1021)	wfp		
631	34368.94	+	(1120)	wfp		
632	34572.39	_	(605)	•	0.61	
645	34 572.74	+	(704)	wfp	0.58	
646	34 676.53	+	(0220)	wfp		
649	34 687.36 34 735.04	+	(218)	wfp		
650 652	34 832.01	<u>-</u>	(724) (119)	wfp		
660	35 025.35	_	(1111)	wfp		
670	35 025.36	+	(1210)	wfp		
671	35 043.00	+	(0110)	wfp		
672	35 055.57	+	(1040)		0.44	
673 693	35 448.85 35 460.76	+	(506)		0.44 0.63	
695	35 460.76	+	(407) (1002)		0.63	35 507°
698	35 516.88	_	(903)	wfp	0.79	35 509.68
699	35 519.58	-	(0211)	wfp		
700	35 555.90	-	(1103)			
704	35 562.73	+	(1130)	wfp	0.00	35 554°
705 700	35 591.84	+	(1300)	wfp	0.68	35 585.96 35 586.01
709 710	35 592.40 35 836.28	+	(12 1 0) (1 21 0)	wfp wfp	0.66	35 586.01
710	35 851.89	- -	(823)	wfp wfp		
723	35 857.44	+	(922)			
724	35 972.24	+	(308)	wfp	0.52	
727	36 030.54	-	(049)	wfp		

Table 2 (continued)

No.	$\tilde{v}_{ ext{theo}}$	par	Label	Plot	$W_{\rm stre}$	$\tilde{v}_{\rm Exp}$
733	36 093.53	_	(209)		0.46	
740	36 185.49	+	(1220)	wfp		36 179.32
746	36 185.89	_	(1121)	•		36 179°
747	36 334.69	+	(1010)	wfp	0.45	
753	36 388.85	+	(0210)			
756	36 597.17	_	(0011)	wfp	0.66	
772	36 687.96	+	(1310)	wfp		36 684.05
776	36 688.90	-	(1211)	wfp		36 684.88
777	36 739.03	+	(0230)	wfp		
780	36 745.30	_	(913)			36 739.78
782	36 745.63	+	(1012)	wfp		36 740.6
783	36 915.67	+	(804)		0.36	
795	36 919.29	-	(705)		0.46	
796	37 122.21	_	(1301)	wfp	0.71	37 122.72
807	37 122.24	+	(1400)	wfp	0.70	37 122.7
808	37 256.54	_	(0221)	wfp		
815	37 312.63	_	(1131)	wfp		37 309.85
821	37 314.57	+	(1230)	wfp		37 311.28
822	37 437.55	-	(1201)		0.73	
832	37 437.55	+	(1102)	wfp	0.72	
833	37 717.05	+	(0111)	wfp		
851	37 742.45	+	(1 22 0)	wfp		*
854	37 767.33	-	(1221)	wfp		37 765*
855	37 767.33	+	(1320)	wfp		37 765.65
856	37 824.75	+	(606)		0.39	
861	37 826.13	-	(507)		0.25	
862	37 904.14	<del>-</del>	(0111)	wfp		2045225
867	38 149.89	+	(1410)	wfp		38 153.25
883	38 149.94	<del>-</del>	(1311)	wfp		38 153.31
884	38 214.65	+	(814)			
887	38 219.07	_	(715)		0.00	20.462.54
888	38 453.59	_	(1401)	wfp	0.69	38 462.54
905	38 453.60	+	(1500)	wfp	0.69	38 462.52
906	38 534.79	+	(408)	wfp	0.25	
911	38 882.06	+	(0410)	wfp		
940 944	38 937.90	_	(0231)	wfp		39 123°
961	39 118.55 39 118.57	+	(1321) (1420)	wfp wfp		39 123.77
962	39 145.48	_	(1011)	wfp	0.34	35 123.77
963	39 176.43	+	(1202)	wib	0.34	
966	39 176.99	_	(1103)	wfp	0.40	
967	39 375.61	_	(1411)	wfp	0.40	39 390.22
986	39 375.61	+	(1510)	wfp		39 390.26
987	39 409.71	+	(0012)	wfp	0.62	33 330,20
988	39 553.99	+	(1600)	wfp	0.02	39 574.54
1001	39 554.01	_	(1501)	wfp		33374.34
1002	39 722.21	+	(1230)	wip		
1012	40 022.79	_	(1113)			40 044.72
1037	40 022.81	+	(1212)			40 044.57
1038	40 21 1.83	+	(1520)	wfp		40 225
1057	40 212.26	_	(1421)	wfp		40 226.31
1058	40 256.54	_	(1331)	wip		10 220.51
1061	40 342.35	+	(1610)	wfp		40 370.55
1071	40 342.68	_	(1511)	wfp		40 370.83
1072	40 405.24	+	(1700)	wfp		40 437.23
1080	40 405.40	_	(1601)	wfp		40 437.26
1081	40 698.03	_	(1203)	p		10 157.20
1106	40 698.10	+	(1302)			
1107	40 887.34	_	(1701)			
1125	40 889.03	+	(1800)			
1126	40 959.15	_	(1611)			
1132	40 960.40	+	(1710)			
1133	41 055.83	_	(1521)			
1145	41 056.27	+	(1620)			
		+	(1900)			
1146	41 002 73					
1146 1149	41 082.75 41 082.78	_	(1801)			

A complete set of vibrational levels is given in the supplementary material. wfp, labelling supported by wave function plot. Label, normal mode assignment ( $v_1$   $v_2$   $v_3$ ) based on the normal mode quantum numbers  $v_1$ ,  $v_2$ , and  $v_3$  for the symmetric stretch, bend, and antisymmetric stretch modes.  $W_{\text{stre}}$ , stretching weight in the wave function, calculated according to Eq. (3). The experimental band centers,  $\tilde{v}_{\text{Exp}}$ , marked with an asterisk were estimated from levels with J > 0.

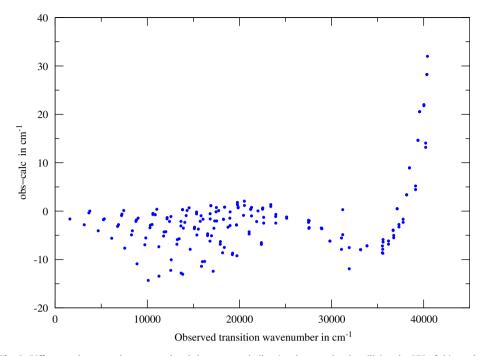


Fig. 6. Differences between the measured and the computed vibrational energy levels utilizing the PES of this study.

and

$$N_{\text{stre}} = \sum_{\nu_1, \nu_3} (\nu_1 + \nu_3) \chi_{i, \nu = (\nu_1, 0, \nu_3)}. \tag{4}$$

In the case of  $W_{\rm stre} > 0.6$ , the level was assigned to be a "pure" stretching state. The remaining  $1-W_{\rm stre}$  contribution is due to harmonic oscillator basis functions in which mixed bending and stretching basis functions or pure bending ones are excited.

Interestingly, above  $7445\,\mathrm{cm}^{-1}\,N_{\mathrm{stre}}$  did not turn out to be a quantitative measure for the number of exciting quanta on the strictly stretching degrees of freedom (similarly to the Jensen-type 'xpect' values). Furthermore, above  $25\,000\,\mathrm{cm}^{-1}$  certain levels with  $0.3 < W_{\mathrm{stre}} < 0.6$  had to be accepted as "pure" stretching states in order to obtain a logically complete labelling. These quantitative defects apart, regularities can be observed in the values of  $W_{\mathrm{stre}}$  and  $N_{\mathrm{stre}}$ . Most importantly, the lowest- and highestlying stretching levels in a polyad have the largest stretching weight,  $W_{\mathrm{stre}}$ , and the largest number of stretching quanta,  $N_{\mathrm{stre}}$ .

Table 1 gives vibrational labels for VBOs below  $25\,200\,\mathrm{cm}^{-1}$ . When all the available different procedures agree in the labels, such assignments are considered to be secure. Above  $26\,500\,\mathrm{cm}^{-1}$  the proportion of labelled VBOs, apart from the "pure" stretching VBOs which are labelled up to about  $35\,000\,\mathrm{cm}^{-1}$  with  $W_{\mathrm{stre}}$  and  $N_{\mathrm{stre}}$ , drops steadily with energy (Table 2). It may well be possible to assign further labels to these high-lying states but it also true that for these states, apart from isolated cases, visual inspection of each wave function plot yields little insight into the nature of the state.

The following can be observed about the excited pure stretching states. Similarly to many other "trends"

observed in the spectroscopy of water, due to the occasional and unpredictable occurrence of resonances these observed "trends" break down. One such trend is the steady alteration of + and - symmetry labels in the sequence of pure stretching states. According to the present PES the first breakdown happens at around 27540 cm<sup>-1</sup>, where the (801) state has a lower energy than the (900) state. The trend of the formation of more and more close-lying + and - states, such as  $(n,0)^+0$  and  $(n,0)^-0$  in local-mode notation, also breaks down. The steady, exponential-like decrease in the computed differences between the  $(n,0)^-0$  and  $(n,0)^+0$  energy levels, all in  $cm^{-1}$ , 98.52 (P=2), 47.99 (P=4), 13.44 (P=6), 2.61 (P=8), 0.41 (P=10), 0.24 (P=12), 0.13 (P=14), and 0.02 (P=16), jumps to -4.34 for P=18, while it is back to 0.01 for P=20. The same can be said about the state pairs  $(n,1)^+0$  and  $(n,1)^-0$ ,  $(n,2)^+0$  and  $(n,2)^-0$ , etc., all in local-mode notation. Pure stretching states which correspond to different polyad numbers are in different ranges up to about 27500 cm<sup>-1</sup>, where it happens that the first pure stretching member of the P=18 polyad, the by now celebrated (801) state at 27539.86 cm<sup>-1</sup>, has a lower energy than the last pure stretching member of the P=16polyad occurring at 27 659.15 cm<sup>-1</sup>. As one moves up in energy, increasing overlap of the polyads is observed.

Despite the drop in the number of assigned VBOs with energy, we are still able to label local mode pairs of the form  $(n,0)^{\pm}v_2$  for low values of  $v_2$  all the way to dissociation. Many of these were observed in a recent experimental study [17]. This suggests that our new potential is actually qualitatively different in the near dissociation region to those used in the previous studies which did not appear to support this systematic series of states [3].

#### 4. Conclusions

The following are the main conclusions that can be drawn based on the present computational study:

- 1. Based principally on 2200 energy points computed at the all-electron, aug-cc-pCV6Z IC-MRCI(8,2) level of electronic structure theory, a new global, mass-independent potential energy surface has been determined for the ground electronic state of the water molecule. The energy points were corrected for relativistic effects via one-electron mass-velocity and Darwin corrections obtained at the same level of theory and in the same calculation. The energy points were fitted to a suitable analytic form allowing for a good global description of the PES toward the first dissociation limit. The PES thus obtained supports 1 150 vibrational energy levels below an energy cutoff value of 41 083 cm<sup>-1</sup>. The average deviation between the computed and the limited number of experimentally measured energy levels is surprisingly small. Even the maximum deviation is less than 15 cm<sup>-1</sup> for all the states with energies below 39500 cm<sup>-1</sup> (see Fig. 6). Therefore, the use of the present ab initio PES for global studies of water is recommended. It is expected that this surface will provide excellent results for the resonance states of water allowing their improved characterization.
- 2. The following can be said about the virtues and weaknesses of the five wave function assignation protocols analyzed. The simple energy decomposition scheme is rather weak, it can be used only for the lowest eigenstates, and it does not provide labels for all but the simplest cases, which can be investigated by other means, as well. Wave function plots, independent of the actual basis representation but dependent upon the coordinate choice, are highly useful and they are able to provide labels based on node counting. This is particularly true for localized high-energy states where all the energy is in a single mode, see Figs. 1–3 for examples. While for many highly excited states such node counting becomes impossible (see, e.g., Fig. 5), for several states the plots provide information not obtainable by other means. One example is the identification of the highly excited bending state (0220), see Fig. 3. The NMD analysis, though basis and coordinate representation dependent, and the corresponding "stretching weight" as well as the mean values of the squares of the Jensen coordinates provide useful and quantitative measures about the composition of the wave functions. Unfortunately, these measures are also incomplete, break down, or just simply indicate the exceedingly complex character of the wave functions at higher excitations.
- 3. The distribution of approximate labels referring to normal mode quantum numbers becomes ambiguous above about 26 500 cm<sup>-1</sup> for all but a few selected states having low or no bending excitation or of pure bends. This is due to the fact that the exact vibrational wave functions become strong mixtures of several basis states.

4. More work is required to obtain completely smooth representations of the dipole moment surface which would allow the wavefunctions given here to be used to estimate vibrational band intensities for states of water lying in the ultraviolet.

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# Appendix A. Supplementary data

Supplementary data associated with this article can be found in the online version at doi:10.1016/j.jqsrt.2010.02.009.

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