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The far-infrared spectrum of ¹⁸O enriched water vapour (40–700 cm⁻¹)



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ABSTRACT

The rotational spectrum of water vapour highly enriched in ¹⁸O has been studied by high resolution (≈ 0.001 cm⁻¹) Fourier transform spectroscopy at the AILES beam line of the SOLEIL synchrotron. The room temperature absorption spectrum has been recorded between 40 and 700 cm⁻¹. The ¹⁸O enrichment of the sample was about 97% while the gas pressure and the absorption pathlength were set to 0.97 mbar and 151.75 m, respectively. The spectrum contains more than 4800 rotational transitions from seven water isotopologues (H2 18O, H2 16O, H2 17O, HD 18O, HD 16O, HD 17O, D2 18O). The assignments were performed using known experimental energy levels as well as calculated line lists based on the results of Schwenke and Partridge. The amount and accuracy of the reported line positions represent an important extension compared to previous works. Overall, lines of about 2570 transitions are observed for the first time and 35, 41, 50, and 16 new energy levels are determined for H2¹⁸O, H2¹⁷O, HD¹⁸O, and HD¹⁷O, respectively. The set of derived energy levels shows a number of important differences from those recommended by an IUPAC-task group. Compared to the HITRAN2016 database, numerous deviations of line positions (up to 0.15 cm⁻¹) are found for the H₂¹⁷O, H₂¹⁸O, HD¹⁷O, and HD¹⁸O species. Incomplete and wrong HITRAN's assignments of more than 90 transitions for H₂¹⁸O, H₂¹⁷O and HD¹⁸O are identified. Overall, the measured line positions will allow to significantly refine and complete the sets of empirical energy levels of H₂¹⁸O, H₂¹⁷O, HD¹⁸O and HD¹⁷O in the ground vibrational state.

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1. Introduction

Water vapour plays a crucial role in the Earth's radiation budget determining our climate [1]. For instance, the water distribution over the globe is permanently monitored by satellite based remote sensing instruments [2]. In this context, a very accurate knowledge of water vapour absorption is required for a wide variety of applications in geoscience [3]. This applies to the main isotopologue, H₂ ¹⁶O, but also to the less abundant H₂ ¹⁸O, H₂ ¹⁷O and HD ¹⁸O species (Note that only CO₂ is more abundant than H₂ ¹⁸O in the Earth atmosphere). Indeed, quantitative information on the natural variation of isotopic abundance ratios, measured by absorption methods, is currently used to trace various chemical and physical atmospheric processes (*e.g.* [4]). Note that a complete and ac-

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curate line list of the water monomer, including line profile parameters, is also a prerequisite for a reliable determination of the atmospheric continuum absorption, which is one of the oldest and the long-standing problem in molecular spectroscopy (*e.g.*, review papers [5–7] and references therein).

In the recent decades, new approaches to the calculation of vibration-rotation energy levels have been developed up to the first dissociation limit (e.g. [8–11]). In general, theoretical line lists show important advantages in terms of completeness, and quality of line intensity predictions. As concerns line positions, theory does not yet compete with the experimental accuracy achieved using high resolution spectrometers. As a result, recommended line lists for water vapour combine the advantages of experimental and theoretical tools, calculated line positions being adjusted according to experimentally determined energy levels. This method is that adopted for a large fraction of the water line list provided by the HITRAN [12] and GEISA [13] databases.

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The present work is devoted to an experimental study of the absorption spectrum of water vapour in the far-IR region, below 700 cm⁻¹. It is commonly believed that the rotational transitions involving the lowest vibrational states of light atmospheric molecules are sufficiently well known and characterized. The results reported in this work indicate that a broad-band high resolution and high sensitivity recording of water vapour spectrum in the far-IR range allows for a substantial extension of the previous knowledge in the region. We believe that the main reason of the revealed information shortage in the THz and far-IR spectral ranges (a region that we define as 1-15 THz or 30-500 cm⁻¹), is the insufficient performances of laboratory spectrometers. While important technological progresses boosted the performances of spectroscopic instruments in the microwave and IR domains in terms of frequency metrology, accessible bandwidth, and sensitivity, the THz/far-IR lacks spectrometers allowing broadband surveys with MHz-level spectral resolution. In the past decades, due to the atmospheric importance of water vapour, several (and often unique) set-ups were applied to the far-IR spectroscopy of water vapour isotopologues [14-40]. They provided valuable although incomplete set of observations with, in general, limited quality in terms of accuracy and sensitivity. Fourier-Transform instruments, with improved spectral resolution over the years, exploited the relatively weak far-IR continuum from globar and/or mercury lamps to record absorption spectra of water isotopologues [14–17,19,21,33,34,38]. These recordings provided important survey data but with limited frequency and intensity accuracy. In addition to absorption spectroscopy, the frequencies of rotation transitions involving the ground and several excited vibrational states of H₂O could be measured using various emission set-ups associated with FT interferometers [27–31,37,39,40].

In order to improve the frequency accuracy of spectroscopic measurements in the far-IR, several original instruments have been developed in late 90's. In particular, Evensons' type tunable far-IR spectrometers [41], whose principle is based on the mixing of two frequency stabilized CO₂ laser lines separated by few THz, allowed to record absorption spectra with few hundred kHz spectral resolution. Using this instrumental approach, the Toyama University group was able to produce several highly accurate line lists of water and its isotopologues in the 1–5 THz range [18,24,36]. In a similar context, Yu et al. [32] reported a combined analysis of highly accurate absorption spectra recorded with frequency multiplication chains (up to 2.7 THz) together with a broadband far-IR FT emission spectra.

Recently, the advent of few beam lines extracting the far-IR radiation from synchrotron facilities allowed for recording Fourier transform spectra with much higher SNR than with the conventional blackbody sources [23,32,42], despite a spectral resolution inherently limited to about 30 MHz for commercial instruments. The increased sensitivity of these set-ups allowed for the detection of relatively weak signals as shown in our recent measurements of the far-IR water continuum absorption [43,44], or the pure rotation spectrum of O₂ [45]. In the present study, we used the experimental set-up developed on the AILES beam line of the synchrotron SOLEIL to extensively improve the far-IR dataset of water isotopic species providing new frequency standards for these important molecules.

The paper is organized as follows. In Section, II we present experimental details including the spectrum acquisition and line list construction. The assignment of observed transitions is presented in Section III. Comparison of our data with previously known information and closing remarks are given in Section IV and V, respectively.

2. Experiment

The Fourier-transform absorption spectrum of ¹⁸O enriched water vapour was recorded in the 40 – 700 cm⁻¹ region on the AILES beam line of SOLEIL synchrotron facility. The recordings were performed during a measurement campaign devoted to the study of the far-IR water vapour self-continuum absorption [43]. The continuum measurements were obtained using both water in natural isotopic abundance and with a highly enriched ¹⁸O sample (the impact of such isotopic substitution upon the water self-continuum might give insights on the physical nature of the water continuum). Note that the continuum retrieval from the measured absorption requires the subtraction of the monomer contribution and thus an accurate and complete line list of water monomer lines in the region.

In the present work, the cell was filled with ¹⁸O enriched water vapour (97% enrichment from Eurisotop) at a pressure of 0.98 mbar measured by a capacitance gauge (Pfeiffer 10 mbar full range with corresponding accuracy of 0.01 mbar). The resolution (defined as 0.9/MOPD where MOPD = 882 cm is the maximum optical path difference) was set to 0.00102 cm⁻¹ in Bruker' definition and no apodization of the interferogram was used (boxcar option of the Bruker software). The spectra were recorded in the 40–700 cm⁻¹ spectral range using the SOLEIL synchrotron radiation in standard mode, together with a 4K cooled Si bolometer detector and a 6 µm mylar-composite beam splitter. The absorption spectra were obtained using a multipass cell in White-type configuration. The total absorption path length was set to 151.75 \pm 1.5 m corresponding to 60 passes between mirrors separated by 2.52 m and about 0.5 m of space between the 50 µm thick polypropylene films windows. The spectrum was recorded at room temperature (298.0(3) K), that was monitored by a pair of platinum sensors mounted on the cell external surface. Two hundred spectra corresponding to a recording time of 10 h were co-added. An overview of the transmittance spectrum is displayed on the upper panel of Fig. 1, which includes successive zooms. The absorption coefficient was determined as $\alpha_{total} = 1/L \ln (I_0(\nu)/I(\nu))$, where $I(\nu)$ and $I_0(\nu)$ correspond to the spectrum recorded with the cell filled with water vapour and evacuated, respectively.

The absolute frequency calibration was performed by frequency matching of 380 measured line centers of intermediate intensities with accurate $\rm H_2^{16}O$ line positions from Refs. [17–19,23,24,26,32]. As a result of the calibration procedure, the statistical uncertainty of the line centers is estimated to vary from $\rm 3\times 10^{-5}$ to $\rm 1\times 10^{-4}~cm^{-1}$ over the 50–700 cm⁻¹ range of the recordings. Note that, at the recording pressure of 0.98 mbar, the amplitude of the water line pressure shifts may reach 0.1 cm⁻¹/atm for the lowest $\it J=0$ and 1 transitions. For these lines we recommend increasing the evaluated uncertainty up to $\rm 2\times 10^{-4}~cm^{-1}$ until accurate data on line shift coefficients become available. For larger $\it J$ values, the shift magnitude decreases and thus a total uncertainty of $\rm 1\times 10^{-4}~cm^{-1}$ is believed to be a conservative estimate.

The line parameters retrieval was performed using a homemade multiline fitting program in LabVIEW and C++. In our pressure conditions, the pressure broadening (about 4×10^{-4} cm⁻¹ HWHM at 1 mbar [12]), the Doppler broadening (on the order of 1.5×10^{-4} cm⁻¹ near 100 cm⁻¹) and the apparatus function (about 3.5×10^{-4} cm⁻¹ HWHM) contribute significantly to the line profile. The present study being mainly focused on line positions, the FTS transmittance spectrum was fitted assuming the standard Voigt line profile as line shape. As starting point of the fit, an empirical line list including line profile parameters was prepared mainly on the basis of the HITRAN database [12], intensity values being scaled according to estimated values of the isotopic abun-

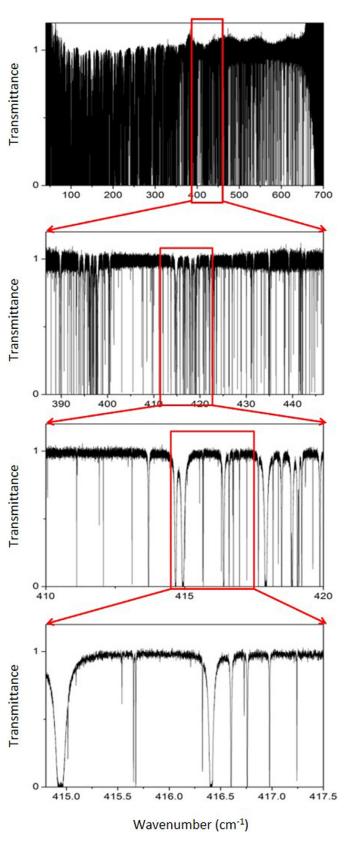


Fig. 1. Successive zooms of the room temperature FTS spectrum of water vapour highly enriched in 18 O recorded at SOLEIL synchrotron ($P\!=\!0.98$ mbar) between 40 and 700 cm $^{-1}$.

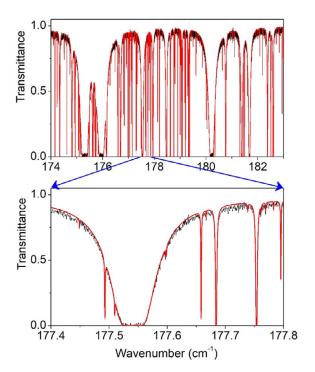


Fig. 2. Comparison of the FTS spectrum of water vapour highly enriched in 18 O near 175 cm $^{-1}$ to a simulation using a Voigt profile as line shape.

Table 1Line-by-line statistics of water transitions assigned between 44 and 678 cm⁻¹.

Molecule	Abundance	NT ^a	NT _{new} b	$J_{\max} K_{a \max}$	Range, cm ⁻¹
H ₂ ¹⁶ O	0.0523	816	37	19 12	44.099 - 673.284
$H_2^{18}O$	0.9345	1206	762	21 14	44.218 - 677.744
$H_2^{17}O$	9.4×10^{-3}	675	318	18 12	44.218 - 665.558
$HD^{16}O$	2.06×10^{-4}	528	133	18 10	46.206 - 558.092
$HD^{18}O$	3.51×10^{-3}	1126	926	21 12	44.866 - 645.736
$HD^{17}O$	3.72×10^{-5}	363	324	16 10	46.023 - 433.483
$D_2^{18}O$	3.75×10^{-6}	110	38	15 10	59.554 - 271.653
Total	1.000	4824	2538	21 14	44.099 - 677.744

Notes.

dances. The water isotopologue identification included in this preliminary list was used to fix the Doppler broadening according to the mass of the involved isotopologue. We intended to adjust the position, area and Lorentzian width of the different lines but in the case of strongly "saturated" lines, due to the high opacity, the line profile parameters were generally constrained to their default values and only the position and area were fitted. Fig. 2 illustrates the achieved spectrum reproduction. In the case of intermediate intensity lines, (obs. – calc.) residuals corresponding to the noise level ($\sim 2\%$ of the empty cell transmittance) was generally achieved. This value corresponds to a noise equivalent absorption on the order of $10^{-6}~{\rm cm}^{-1}$ and a detectivity threshold of about $10^{-25}~{\rm cm/molecule}$ for the line intensities.

Overall, line parameters of 5262 absorption features were retrieved. This large dataset includes about 1000 "saturated" water lines too strong to be accurately measured in our experimental conditions. Indeed, the investigated spectral region involves very strong transitions with intensity larger than $10^{-18}\,\text{cm/molecule}$ while the pressure and absorption pathlength conditions of the studied spectrum are suitable for derivation of line intensities in the 5×10^{-25} - $8\times 10^{-23}\,\text{cm/molecule}$ range (see below).

^a NT – Number of transitions contributing to the spectrum (including "saturated" lines too strong to be accurately measured).

 $^{^{\}rm b}$ NT $_{\rm new}$ – Number of transitions which were previously unobserved in absorption or in emission.

New energy levels of H₂ ¹⁸O, H₂ ¹⁷O, HD¹⁸O, and HD¹⁷O determined from the analysis of the FTS spectrum of water vapour highly enriched in ¹⁸O between 44 and 678 cm⁻¹.

Energy^a K_{α} K_c $HD^{17}O$ 1286.95798 1286.95806 1077.48245 1237.64510 1137.95227 1273.25513 1282.27792 1400.64291 1476.06777 1687.62496 1623.95330 1623.95890 1807.63052 1807.86174 1832.90619 1832.90869 HD180 2727.92355 2727.92356 2625.80073 2625.80071 2907.45459 2907.45459 3211.69276 3211.69275 2821.45430 2821.45437 3101.84662 3101.84664 3404.72928 3404.72928 2776.11040 2776.11032 3032.03855 3032.03955 3311.06726 3311.06719 2390.77152 2567.59499 2772.56672 2772.56729 3003.01555 3003.01461 3257.51792 3257.51800 2639.47084 2813.03401 3016.06013 3016.07130 3244.89230 3244.90173 2691.62971 2746.38188 2779.97472 2904.61371 3073.91144 2868.61008 2759.48253 2759.49883 2967.05298 2967.43987 2782.54227 2782.54232 3029.29136 3029.29996 3051.90805 3051.90800 H₂¹⁷O 3493.27691 3493.27731 3744.31027

(continued on next column)

Table 2 (continued)

J	K_a	K_c	Energy ^a	δE^{b}
12	12	0	3744.31067	73
13	7	7	2918.03362	46
13	8	6	3116.43172	54
13	9	5	3333.91518	58
13	9	4	3333.91904	44
13	10	4	3566.86888	63
13	10	3	3566.86799	50
13	11	3	3811.88536	71
13	11	2	3811.88481	65
14	4	10	2876.88740	34
14	5	10	2912.22127	44
14	5	9	2979.51743	41
14	6	9	3077.62393	51
14	6	8	3095.29248	40
14	7	8	3255.17281	46
14	7	7	3257.54686	45
14	8	7	3453.37869	51
14	9	6	3671.37911	60
15	1	14	2625.42191	48
15	2	14	2625.43495	51
15	2	13	2866.13932	46
15	3	13	2866,41578	47
15	3	12	3074.19168	49
15	4	11	3239.69214	51
15	6	9	3467.06349	51
15	7	8	3620.21357	64
16	1	15	2945.81632	65
16	2	15	2945.82275	61
16	3	14	3204.28827	61
16	4	13	3432.23621	59
17	0	17	2974.64780	77
17	1	17	2974.65177	89
17	1	16	3283.81730	76
17	2			76 74
		16	3283.82057	
17	2	15	3559.46121	71
18	0	18	3311.98363	97
18 18	1 2	18 17	3311.97966 3639.33094	86 84
	2	17	3639.33094	84
H₂¹⁸O 15	5	11	2257 12217	8
16	1	11 15	3257.12317	10
			2939.95141	
16	3	13	3424.23366	11
16	4	12	3612.34609	11
16	5	11	3749.54756	12
16	6	10	3859.76914	10
16	8	8	4185.60745	13
17	2	16	3277.27565	12
17	2	15	3552.56928	9
17	3	15	3552.63498	12
17	3	14	3796.19838	9
17	4	14	3797.10678	14
17	5	13	4012.43988	12
18	0	18	3305.31389	13
18	1	18	3305.31308	11
18	1	17	3632.07157	13
18	2	17	3632.07254	11
18	2	16	3924.35214	13
18	3	16	3924.38536	11
18	3	15	4184.79228	16
18	4	15	4185.28138	11
18	5	14	4416.22560	11
18	15	3	6806.97939	120
19	0	19	3659.47265	13
19	1	19	3659.47360	15
19	1	18	4004.24664	13
19	2	18	4004.24772	15
19	2	17	4313.21925	12
19	3	17	4313.23774	15
20	0	20	4031.03352	17
20	1	20	4031.03257	16
20	1	20 19	4393.69352	17
	2			
20		19 21	4393.69394	14 17
21	0	21	4419.88437	17 10
21	1	21	4419.88532	19
Notes.				
iotes.				

Notes.

^a Term energy in cm^{−1}.

^b Statistical uncertainty in 10⁻⁵ cm⁻¹.

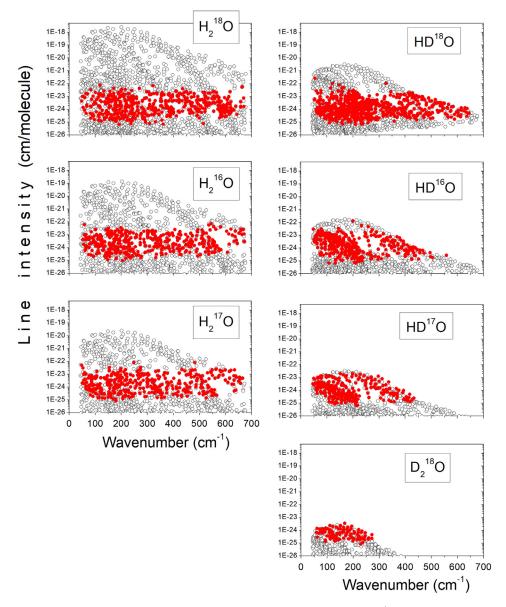


Fig. 3. Overview of the presently assigned transitions for the different water isotopologues between 44 and 678 cm $^{-1}$. The plotted observations (red points) are limited to "unsaturated" lines for which line parameters could be reliably determined. Open circles correspond to the HITRAN2016 line list except for $D_2^{18}O$, absent in the HITRAN database, for which the predictions based on Schwenke and Partridge calculations [8,9] were used. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

3. Rovibrational assignments

The rovibrational assignments were performed using literature results as well as calculated line lists based on the results of Schwenke and Partridge (SP) [8,9]. 4610 lines were assigned to 4824 transitions of seven water isotologues (some absorption features correspond to several transitions-see **Table 1**). In addition, on the basis of the HITRAN2016 database [12], 100 lines were assigned to transitions of HF, NH₃, $^{12}\text{C}^{16}\text{O}_2$, and $^{12}\text{C}^{18}\text{O}_2$ present as impurities in the sample, with estimated relative concentration of 1, 2, 10 and 7 ppm, respectively (Note the strong ^{18}O enrichment of CO₂ probably due to oxygen atom exchanges between ^{18}O enriched water vapour and natural carbon dioxide). 552 lines with intensity below 8.3×10^{-25} cm/molecule remain unassigned. The relative abundances of the different water isotopologues were estimated from the ratio of the measured line intensities to their calculated values [46–51].

Fig. 3 shows an overview of the "unsaturated" assigned lines for the different isotopologues. The number of assigned transitions, maximum values of rotational numbers and spectral range are given in Table 1. The number of newly observed transitions included in the table indicates that new transitions are observed for the seven isotopologues, in particular the main one, H₂¹⁶O, although its relative abundance was limited to about 5% in the used water sample. For the H₂¹⁸O molecule, the observed lines are rotational transitions within the ground vibrational state or within the (010) and (020) bending states (vibrational states are labelled $(V_1V_2V_3)$ where V_1 , V_2 and V_3 are the quantum numbers corresponding to the symmetric stretch, bending and antisymmetric stretch, respectively). For the $H_2^{16}O$, $H_2^{17}O$, $HD^{16}O$ and $HD^{18}O$ isotopologues, only rotational transitions within the (000) and (010) states are detected while for the least abundant HD¹⁷O and D₂ ¹⁸O species, only ground state rotational transitions were assigned.

We provide as supplementary material the fitted parameter values of 3625 lines, which were reliably determined, thus excluding

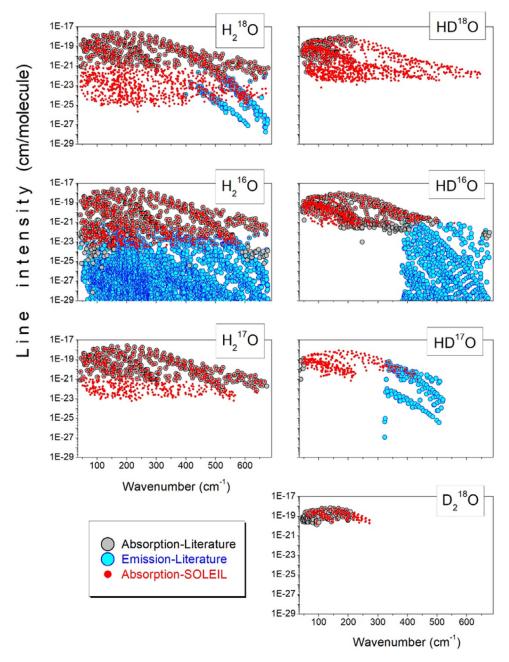


Fig. 4. Overview comparison of the present observations (red points) to previous literature data. Transitions from the literature observed by absorption and emission spectroscopy are indicated by grey and blue circles, respectively. For the sake of the comparison, we have associated an absorption line intensity to the emission data. Line intensities correspond to the pure species (100% abundance). (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

many "saturated" lines. The position comparison presented below applies to this set of transitions.

Following the exhaustive review of the literature presented in the next paragraph, 35, 41, 50, and 16 new energy levels are determined for $\rm H_2^{18}O$, $\rm H_2^{17}O$, $\rm HD^{18}O$, and $\rm HD^{17}O$, respectively. They are listed in Table 2.

4. Comparison to literature

4.1. Previous experimental studies

Some years ago, an IUPAC task group performed an exhaustive review of the experimental investigations of the absorption

and emission spectra of the various water isotopologues [52–55]. Each of the IUPAC-TG report includes a table listing the studies available at that time and for each work, the spectral range, the number of reported transitions and the main experimental characteristics. Most of the available experimental data in our region of interest have been obtained by FTS. In particular, several FTS emission spectra provided a considerable amount of observations which should be used with caution because emission line positions are in general less accurate as they suffer from frequent line overlapping and significant Doppler broadening in the case of hot spectra. Overall, most of the previous absorption studies are not recent and limited to less than one hundred lines. This is due to the fact that, all literature works but Ref. [36] used natural water or a

Table 3 HITRAN2016 water contents between 44 and 678 cm^{-1} .

Isotopologue	Abundance	Number of transitions	Intensity cut-off	References	
			(cm/molecule)	Position	Intensity
H ₂ ¹⁶ O	0.997317	6551	1.026×10^{-32}	[54,58,59]	[46-49]
H ₂ ¹⁸ O	0.00199983	2261	1.350×10^{-31}	[52,60,61]	[50]
$H_2^{-17}O$	0.000371884	1832	2.020×10^{-31}	[50,61]	[50]
HD ¹⁶ O	0.000310693	4363	1.473×10^{-31}	[51]	[51]
HD ¹⁸ O	0.000000623003	1824	1.500×10^{-31}	[51,62]	[51]
HD ¹⁷ O	0.000000115853	1403	1.576×10^{-31}	[51,62]	[51]
$D_2^{16}O$	0.000000024197	3534	2.301×10^{-34}	[51,62]	[51]
Total	1.000	21,768			

Table 4 Line position comparison with HITRAN2016 [12].

Molecule	NT a	$\delta v_{ m max}$	$\delta \nu (10^{-3} \text{ cm}^{-1})^b$				
		$(10^{-3} \text{ cm}^{-1})$	$\delta \nu < 0.1$	$0.1 \le \delta v < 0.5$	$0.5 \le \delta \nu < 1.0$	$\delta v \ge 1.0$	
H ₂ ¹⁶ O	604	0.57	399	201	4		
$H_2^{18}O$	734	43.96	191	295	103	145	
$H_2^{17}O$	619	35.51	241	202	52	124	
$HD^{16}O$	504	3.04	275	211	14	4	
HD ¹⁸ O	984	152.54	411	369	101	103	
HD ¹⁷ O	356	117.21	43	180	66	67	
Total	3801	152.54	1560	1458	340	546	

Notes

- ^a NT number of transitions.
- b Absolute value of the $(\nu_{meas}\text{--}\nu_{IUPAC})$ position difference in $10^{-3}~cm^{-1}$ unit.

moderated isotopic enrichment in 18 O while a water vapour highly enriched in $\mathrm{H_2}^{18}$ O and a long absorption pathlength are presently used. The large amount of newly measured transitions is illustrated in **Fig. 4** where previous observations by absorption and emission spectroscopy are distinguished. In order to give an overview of the available transition wavenumbers, we have associated a line intensity (HITRAN values scaled to 100% abundance for each isotopologue) to all the wavenumbers reported in the literature both in absorption and in emission. Compared to previous absorption studies, a considerable gain is noted for all the isotopologues but $\mathrm{H_2}^{16}\mathrm{O}$ and $\mathrm{HD}^{16}\mathrm{O}$. In particular, the present measurements lower by four orders of magnitude (from $\mathrm{10}^{-21}$ to $\mathrm{10}^{-25}$ cm/molecule), the detectivity threshold of $\mathrm{H_2}^{18}\mathrm{O}$ lines and a gain of two orders of magnitude is noted for $\mathrm{H_2}^{17}\mathrm{O}$ and $\mathrm{HD}^{18}\mathrm{O}$. Previous $\mathrm{HD}^{17}\mathrm{O}$ measurements were practically absent in this spectral region.

Below, we give a brief updated summary of the present knowledge for the various isotopologues in the region of interest (44–678 cm⁻¹).

The most extensive study of ${\rm H_2}^{16}{\rm O}$ absorption lines in the region was reported by Kauppinen et al. [14] in 1978. 382 pure rotational transitions and 17 rotational transitions of the ν_2 - ν_2 band were reported. Later, 233 additional rotational transitions in absorption were reported in Refs. [15–25]. Besides, hot emission spectra provided a high number of rotational and rotation-vibration transitions in the ground and 15 excited vibrational states [26–32]. Overall, 2868 different rotational transitions were reported in Refs. [14–32] within the ground and first excited (V_2 =1) vibrational states.

4.1.1. $H_2^{18}O$ and $H_2^{17}O$

The literature data concerning these minor isotopologues is limited. In 1977, Winther measured 121 and 48 pure rotational transitions of $\rm H_2^{18}O$ and $\rm H_2^{17}O$, respectively, by absorption FTS in natural water vapour below 501 cm⁻¹ [33]. Later, 255 absorption transitions were reported for $\rm H_2^{18}O$ in Refs. [14,15,17,21,34–36]. Note that only one transition (3_{2 2} - 3_{1 3} at 73.31376 cm⁻¹) was re-

ported by Johns [17] for the first excited state (V_2 =1). Finally, Mikhailenko et al. [37] assigned 127 H₂¹⁸O emission lines to 144 transitions in the 399 – 677 cm⁻¹ region. 36 of these 144 transitions belong to the v_2 - v_2 band. Overall, 498 rotational absorption and emission transitions in the (000) and (010) states were reported in Refs. [14,15,17,21,33–37] for H₂¹⁸O. As for H₂¹⁷O, all the 364 transitions known in the far-IR range are pure rotational transitions [14,15,17,21,33–36].

4.1.2. HD¹⁶O

Only four high resolution studies of HD¹⁶O absorption were reported in the studied frequency range. Sixty pure rotational transitions were reported in 1978 by Kauppinen et al. [14] between 152 and 419 cm⁻¹. Later, Johns [17], Paso & Horneman [19] and Toth [38] expanded the studied range to 45 – 677 cm⁻¹ leading to a total number of 528 transitions. More recently, Janca et al. reported the HD¹⁶O emission spectrum above 381 cm⁻¹ [39]. Overall, 1168 transitions within the ground and V_2 =1 vibrational states were reported in our region in Refs. [14,17,19,38,39].

4.1.3. $HD^{18}O$ and $D_2^{18}O$

182 and 136 pure rotational transitions between 44 and 220 cm $^{-1}$ were reported by Johns [17] for HD 18 O and D $_2$ 18 O, respectively. More recently, Yu et al. [56] reported accurate measurements of 31 transitions between 50 and 134 cm $^{-1}$ for HD 18 O.

4.1.4. HD¹⁷O

Up to now, only six transitions of HD¹⁷O were measured in absorption [57]. They are located between 46.0 and 50.4 cm⁻¹ and show a resolved hyperfine structure due to the nuclear spin of ¹⁷O. In a recent FTS emission study, Mellau et al. [40] assigned 189 transitions to 169 emission lines between 320 and 520 cm⁻¹. These measurements provided an extended set of the energy levels of the ground and first excited $(V_2=1)$ states up to $J_{max}=17$ and $K_{a max}=13$.

 Table 5

 Line position comparison with values calculated using the IUPAC-TG energy levels.

Molecule	NT ^a	NTO ^b	δv_{max} (10 ⁻³ cm ⁻¹)	$\delta \nu (10^{-3} \delta \nu < 0.1)$	cm^{-1}) $\frac{d}{0.1 < \delta \nu < 0.5}$	0.5<δν<1.0	δυ>1.0
			(10 cm)	01 <0.1	0.1 _0 1 < 0.5	0.5_0 / < 1.0	
$H_2^{16}O$	604	0	7.43	208	286	76	34
$H_2^{18}O$	734	43	39.61	189	294	102	106
$H_2^{17}O$	619	49	34.98	248	202	49	71
HD ¹⁶ O	504	0	1.06	275	212	13	4
$HD^{18}O$	984	277	29.61 ^c	105	284	131	187
$HD^{17}O$	356	174	36.24	45	81	30	26
$D_2^{18}O$	110	0	2.83	31	63	7	9
Total	3911	543	39.61 ^{c)}	1101	1422	408	437

Notes.

- ^a NT number of transitions.
- ^b NTO number of transitions that could not be calculated for IUPAC energy levels [52,53].
- ^c Excluding the 14 $_{3.11}$ 13 $_{2.12}$ rotation transition (ν_{OBS} = 438.44685 cm⁻¹ while ν_{IUPAC} = 195.91293 cm⁻¹ (see Text).
- ^d Absolute value of the (ν_{OBS} ν_{IUPAC}) position difference in 10^{-3} cm⁻¹ unit.

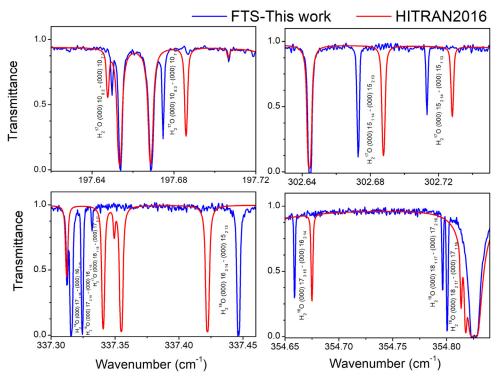


Fig. 5. Comparison of the recorded spectrum to a simulation based on the HITRAN2016 database in the same experimental conditions.

4.2. HITRAN database

Table 3 summarizes the contents of the HITRAN2016 water line list [12] in the 44-678 cm⁻¹ region. 21,768 transitions of seven isotopologues H₂¹⁶O, H₂¹⁸O, H₂¹⁷O, HD¹⁶O, HD¹⁸O, HD¹⁷O, and D₂¹⁶O in natural abundance are listed with intensity cut-offs ranging between 2.3×10^{-34} and 2.0×10^{-31} cm/molecule depending on the species. The last column of the table gives the source of the HI-TRAN2016 line positions and line intensities. For all but the main isotopologue, line intensities are ab initio values while line positions were calculated from IUPAC-TG energy levels when available or, otherwise, from ab initio energy levels. In the case of H₂¹⁶O, 2507 rotation and rotation-vibration transitions in the five lowest vibration states - (000), (010), (020), (100), and (001) - originate from an effective operator approach [46,49,58,59]. For transitions involving other vibrational states, HITRAN line parameters were obtained in the same way as for the minor isotopologues (see Section 2.1 of Ref. [12]).

The systematic comparison of our observations to HITRAN line list revealed that six rotational transitions of $\rm H_2^{18}O$ with J>17, observed in our spectrum, are missing in the HITRAN list. Ninety-two rotational transitions of $\rm H_2^{18}O$, $\rm H_2^{17}O$ and $\rm HD^{18}O$ are provided with incomplete or erroneous assignments. The list of $\rm H_2^{18}O$ missing lines and the corrected assignments are provided as Supplementary Material.

Table 4 summarizes the results of the systematic comparison of our measured line positions to HITRAN values. The comparison applies to lines "unsaturated" in our spectrum (see **Fig. 3**). Due to an extremely small natural relative abundance, $D_2^{18}O$ transitions are not included in the HITRAN list. **Table 4** gives the number of transitions corresponding to different ranges of deviations. Overall, a very good agreement is observed for $H_2^{16}O$ and to a less extend for $HD^{16}O$. This is not the case for the $H_2^{18}O$, $H_2^{17}O$, $HD^{18}O$ and $HD^{17}O$ species which show a significant fraction of positions deviating by more than 10^{-3} cm⁻¹. Maximum deviations correspond to *ab initio* values from Refs. [50,51] and exceed 0.03 cm⁻¹ for $H_2^{18}O$

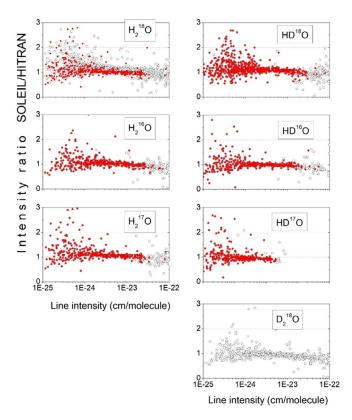


Fig. 6. Ratios of measured and HITRAN line intensities *versus* experimental intensity values (for D₂¹⁸O, SP calculated values [8,9] are used instead of HITRAN values [12]). The average intensity ratio was fixed close to unity, by scaling the HITRAN (or SP) intensity values according to the abundance values given in Table 1. The more reliable ratios corresponding to "unsaturated" lines are highlighted (red dots). (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

and $\rm H_2^{17}O$ and 0.11 cm $^{-1}$ for HD ^{18}O and HD ^{17}O . Fig. 5 shows a comparison of our FTS spectrum to an HITRAN simulation in the experimental conditions of the recordings for several spectra intervals where HITRAN data deviate importantly from the observations.

The ratios of measured and HITRAN line intensities were used to determine the isotopologue abundance in our sample except for the $D_2^{\,18}O$, absent in the HITRAN list, for which the calculated SP intensities [8,9] were used. The obtained ratios are plotted in Fig. 6 versus the measured line intensities. In order to get an average intensity ratio of unity, the reference intensity values (HITRAN or SP) were scaled according to the abundance values given in Table 1. The more reliable sets of ratios corresponding to "unsaturated" lines are highlighted. Overall, the achieved agreement is reasonable. Large errors in the retrieval of the experimental line intensities are believed to be responsible of the observed outliers.

4.3. IUPAC task group

The energy levels recommended by the IUPAC-TG [52–55] have been used to calculate the line centers of our measured transitions. The comparison (**Table 5**) applies to the same set of unsaturated lines as for the HITRAN comparison (**Table 4**). Similarly to the HITRAN comparison, **Table 5** gives the number of transitions corresponding to different ranges of deviations. The considered observed transitions of H₂¹⁸O, H₂¹⁷O, HD¹⁸O, and HD¹⁷O involve 503, 329, 415 and 187 energy levels, respectively. For these four species, the maximum position differences are on the order of 0.03 cm⁻¹ and a total of 543 IUPAC positions could not be calculated because

they involve levels absent in the IUPAC-TG datasets. More than 10% of the compared line positions deviate by more than 10^{-3} cm⁻¹ *i.e.* by more than ten times the accuracy of our reported line positions, suggesting that the IUPAC values of the involved energy levels should be significantly corrected. Note that for HD¹⁸O, we excluded from **Table 5**, the largest position difference corresponding to the $14_{3 \ 11}$ – $13_{2 \ 12}$ transition (ν_{OBS} = 438.44685 cm⁻¹ while ν_{IUPAC} = 195.91293 cm⁻¹). This error is due to the fact that the upper energy level at 1592.7268 cm⁻¹ in the IUPAC energy data set [53] is assigned to the (000) $14_{3 \ 11}$ level and not to the level (000) $14_{1 \ 13}$. In fact, this is a result of erroneous assignments of three ν_3 transitions in Ref. [63]. The transitions at 3457.1689, 3482.6769, 3483.1638 cm⁻¹ assigned in Ref. [63] as $13_{1 \ 12}$ – $14_{1 \ 13}$, $13_{3 \ 10}$ – $14_{3 \ 11}$ and $13_{2 \ 12}$ – $14_{3 \ 11}$, respectively, should be assigned to $13_{3 \ 10}$ – $14_{3 \ 11}$, $13_{1 \ 12}$ – $14_{1 \ 13}$ and $13_{2 \ 12}$ – $14_{1 \ 13}$, respectively.

5. Concluding remarks

The knowledge of the absorption spectrum of water vapour, in particular of the minor isotopologues, has been significantly improved in the range of the rotational band, on the basis of a FTS spectrum highly enriched in ¹⁸O recorded at the AILES beam line of the SOLEIL synchrotron. The quality of the recorded high resolution spectrum benefitted from the high photon flux and large spectral coverage provided by the synchrotron source. The use of a 151.75 m absorption pathlength allows for the detection of 2538 new transitions of seven water isotopologues (H2¹⁸O, H2¹⁶O, H₂¹⁷O, HD¹⁸O, HD¹⁶O, HD¹⁷O, D₂¹⁸O). Overall, one hundred and forty-two energy levels were newly determined for H₂¹⁸O, H₂¹⁷O, HD¹⁸O, and HD¹⁷O. In addition, for a large fraction of previously measured transitions (in particular those derived from emission spectra), the line position accuracy (about 10^{-4} cm⁻¹) is significantly improved. The systematic comparison to the HITRAN2016 database and to the position values derived from the IUPAC-task energy levels reveals a significant number of inaccuracies. Considering that the observed transitions involve the lowest vibrational states of most of the water vapour transitions, the energy levels corrections evidenced in the present study will propagate to a large number of energy values of excited vibrational levels.

Note added to the proofs: We have become aware of the (effective Hamiltonian) modelling of the $\rm H_2^{18}O$ spectrum up to the first triad and J=17 published by Coudert and Chélin [J Mol Spectrosc 2016;326;130–135] thus including our region of interest. The comparison of our experimental positions to those reported in this work has been performed. Overall, positions calculated by Coudert and Chélin show a significantly better agreement to our measurements than those of the HITRAN2016 list. Nevertheless, a number of deviations (up to 0.01 cm $^{-1}$) are noted in particular for the highest J levels of the list of Coudert and Chélin. The comparison will be presented and discussed in details in a future contribution.

Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

CRediT authorship contribution statement

S.N. Mikhailenko: Formal analysis. **S. Béguier:** Formal analysis. **T.A. Odintsova:** Investigation, Writing - review & editing. **M.Yu. Tretyakov:** Investigation, Writing - review & editing. **O. Pirali:** Investigation, Writing - review & editing. **A. Campargue:** Formal analysis, Writing - review & editing.

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Supplementary materials

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