



# Triglycerides

A Level



## Part A Triglyceride formation

A triglyceride is a particular type of lipid formed by a condensation reaction between one  molecule and   molecules. During condensation, the carboxyl groups of the  molecules react with the hydroxyl groups of the  molecule to form  bonds. Therefore, this condensation reaction is also called  reaction.

Items:

two

an esterification

ester

a phosphorylation

glycerol

three

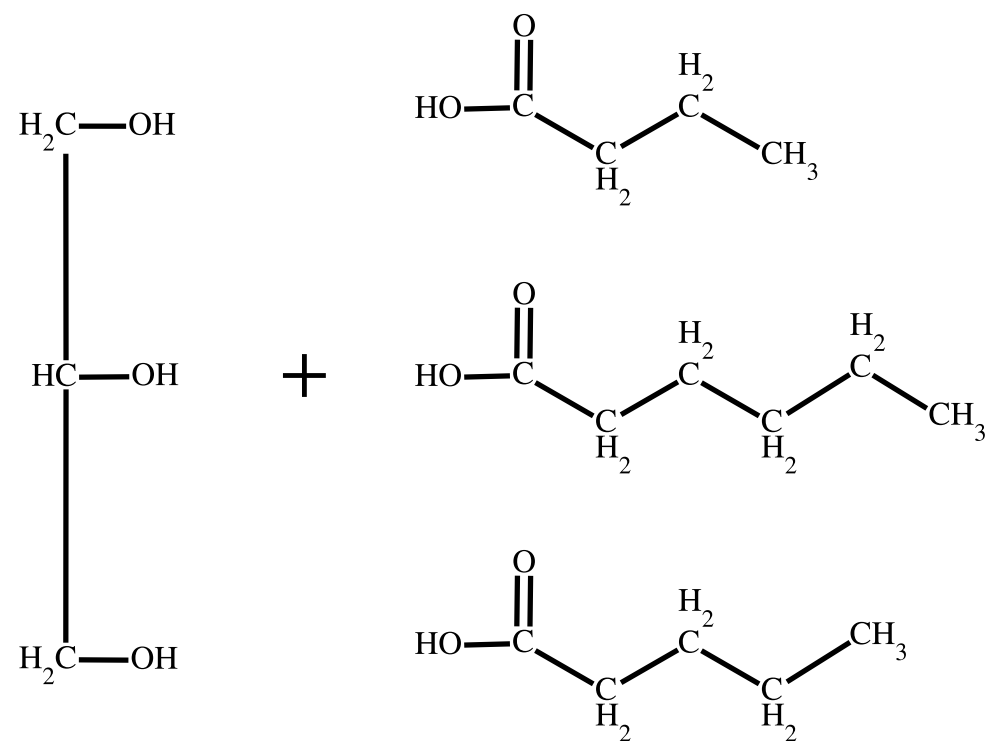
four

glycine

fatty acid

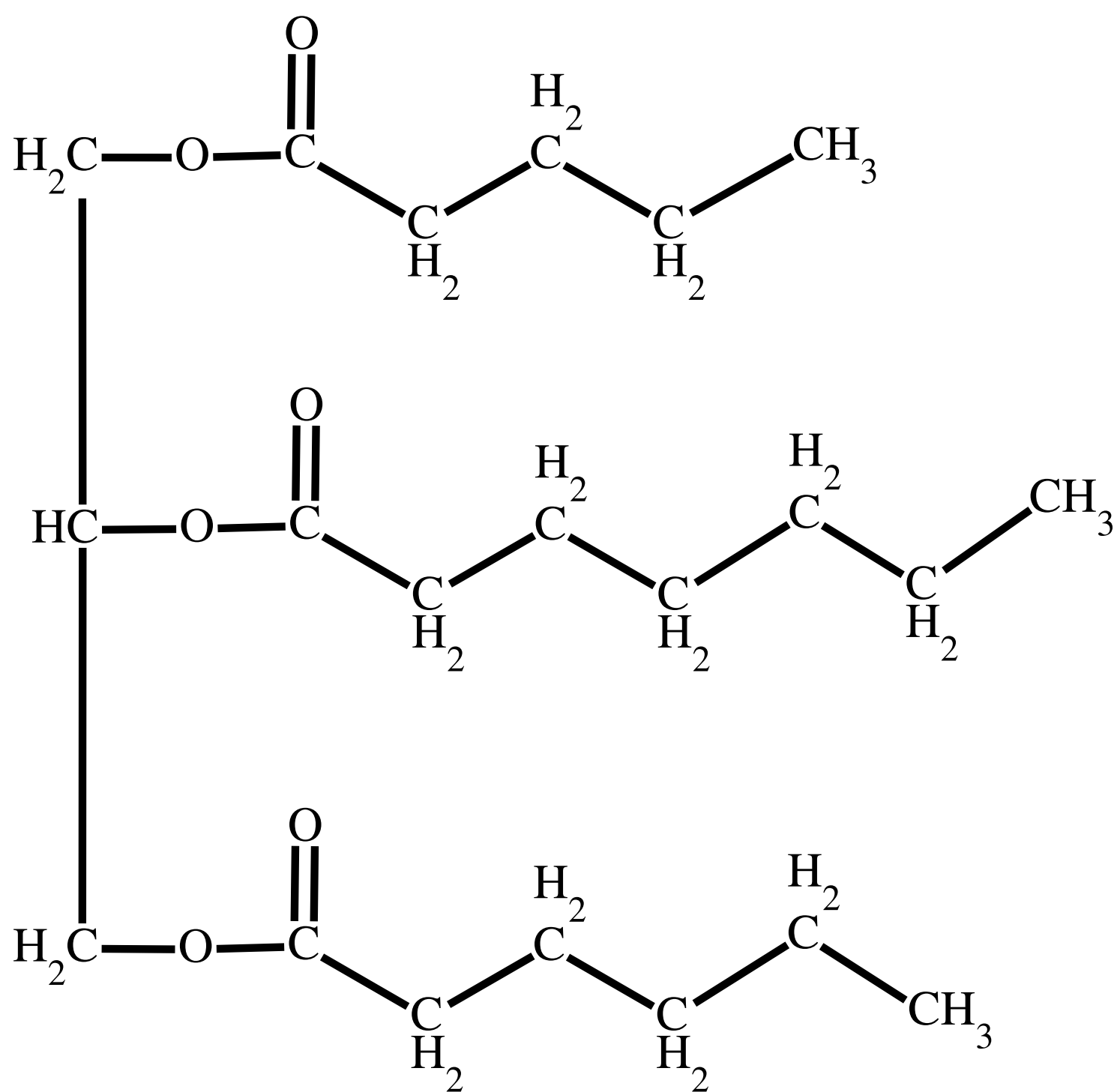
phosphate

## Part B Condensation consequences



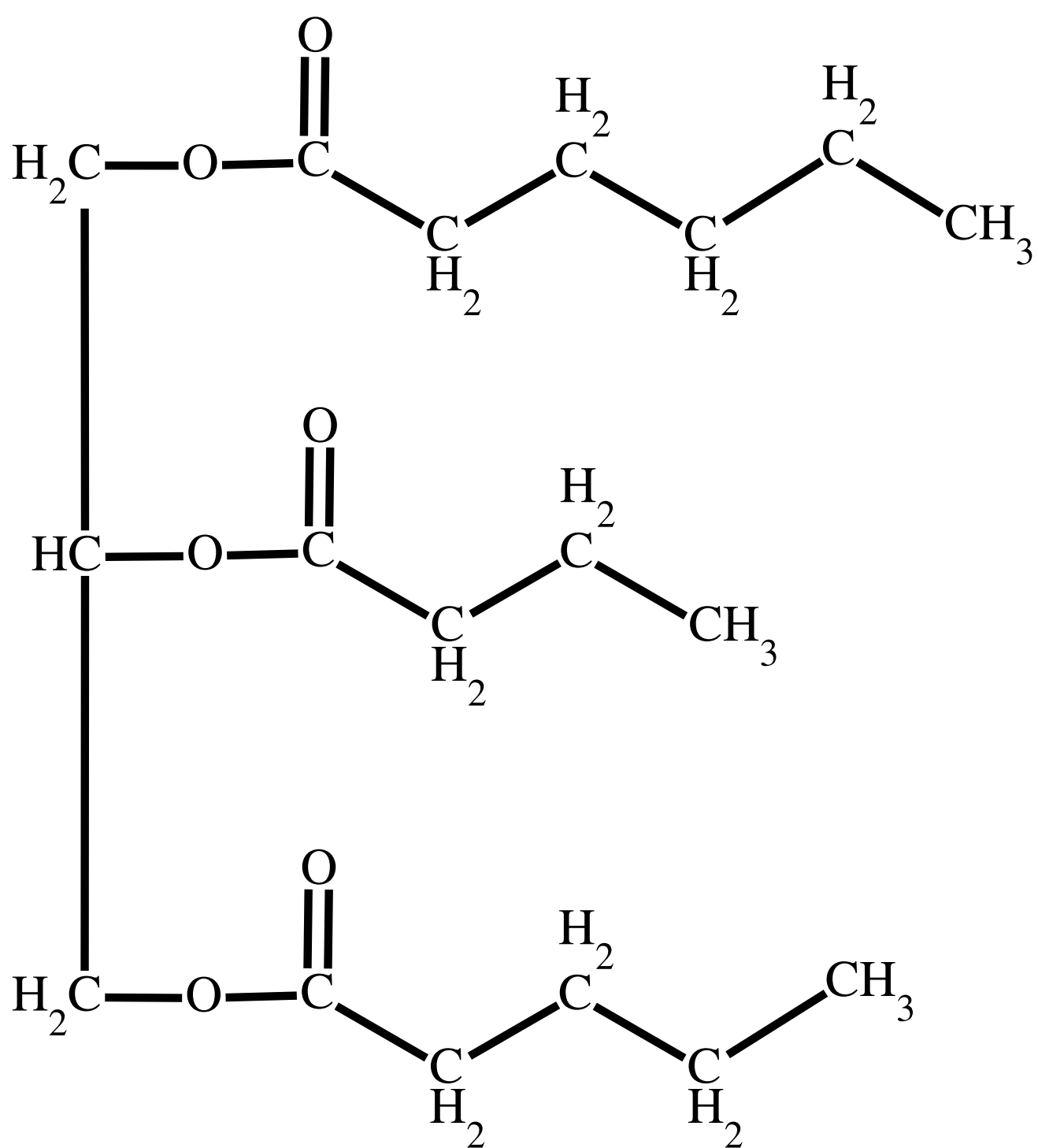
**Figure 1:** A condensation reaction between one glycerol molecule and three fatty acids.

Which of the images below represent triglycerides that could be formed in the condensation reaction shown in Figure 1? Select all that apply.

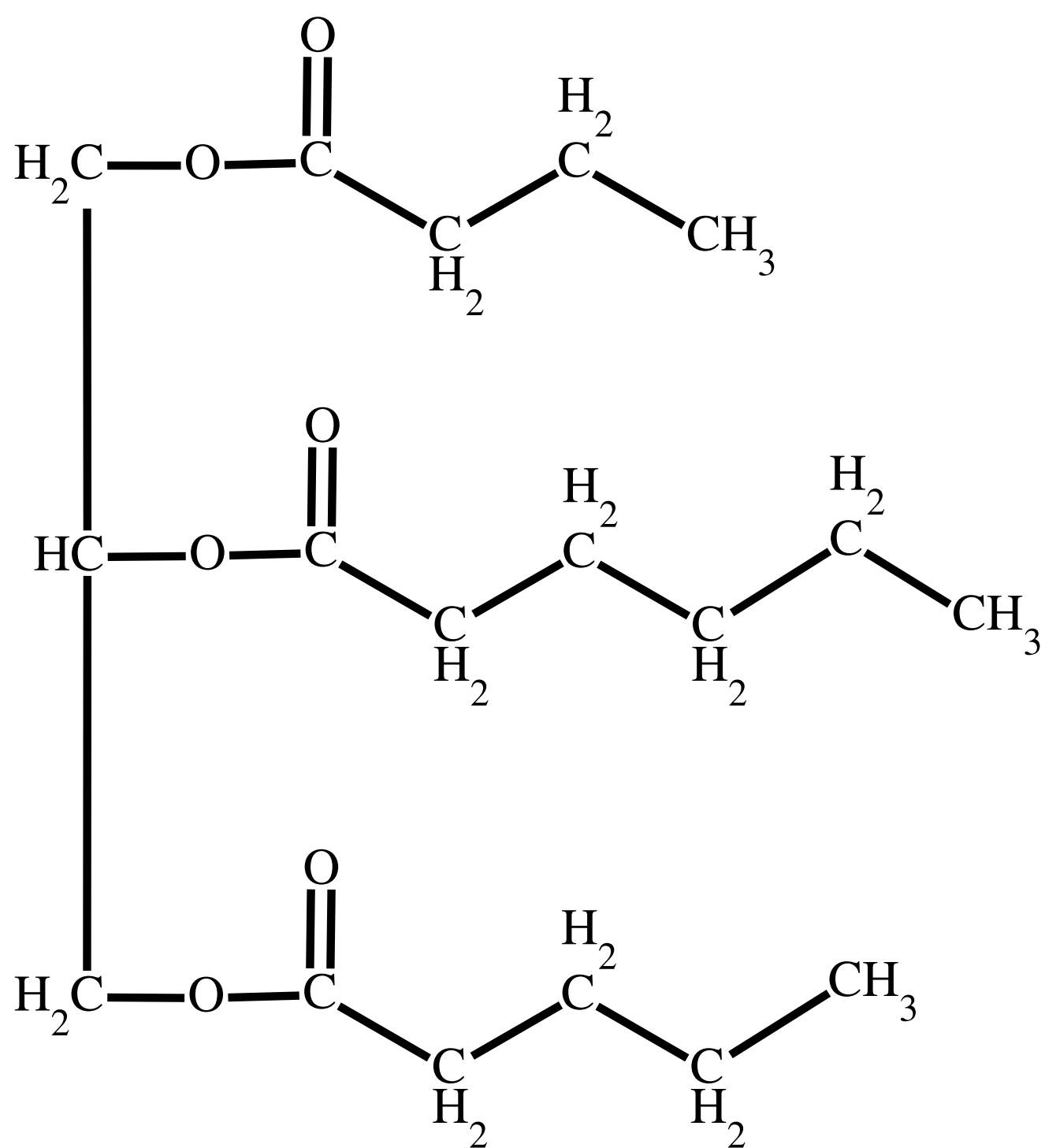


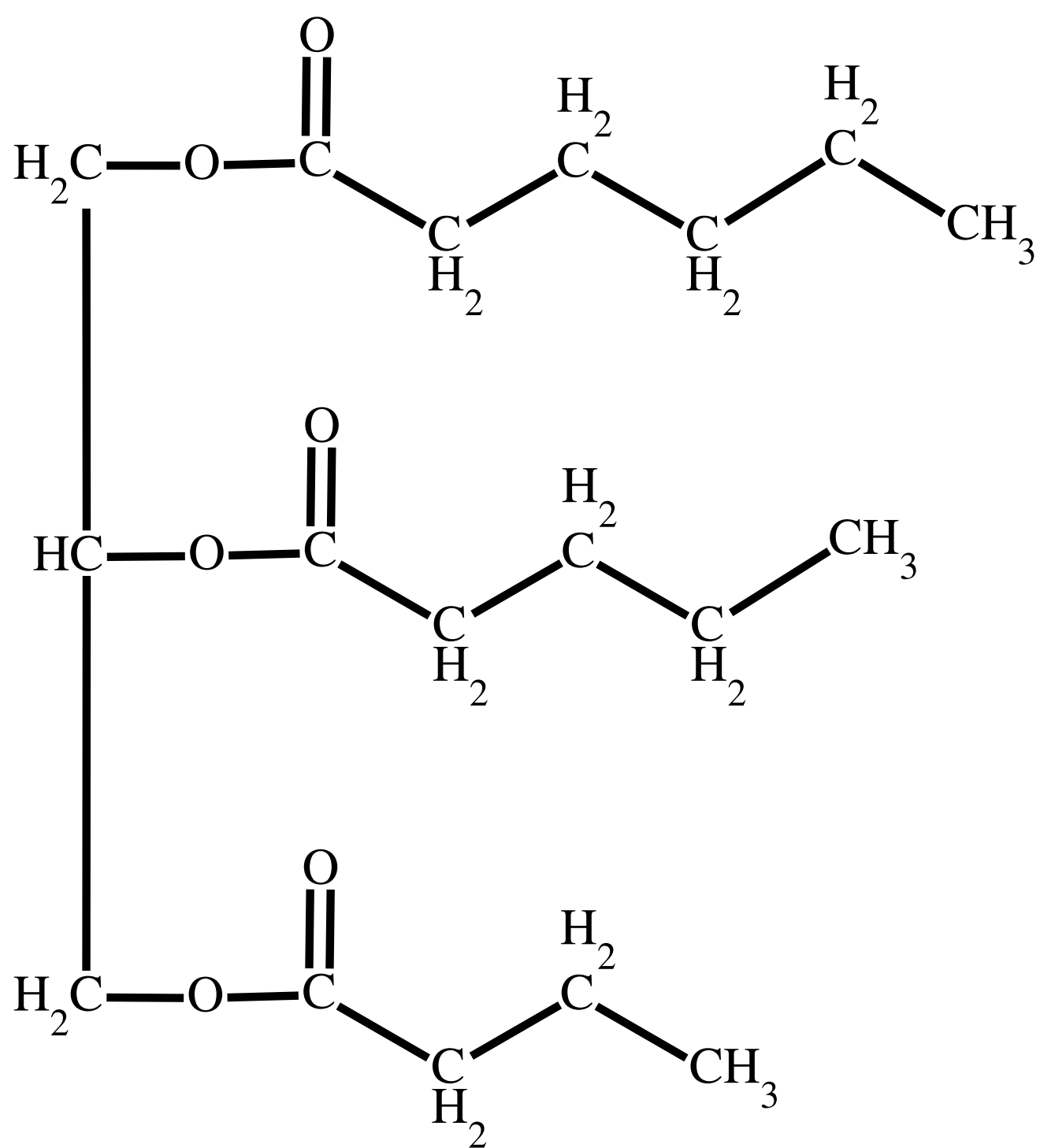
---

A



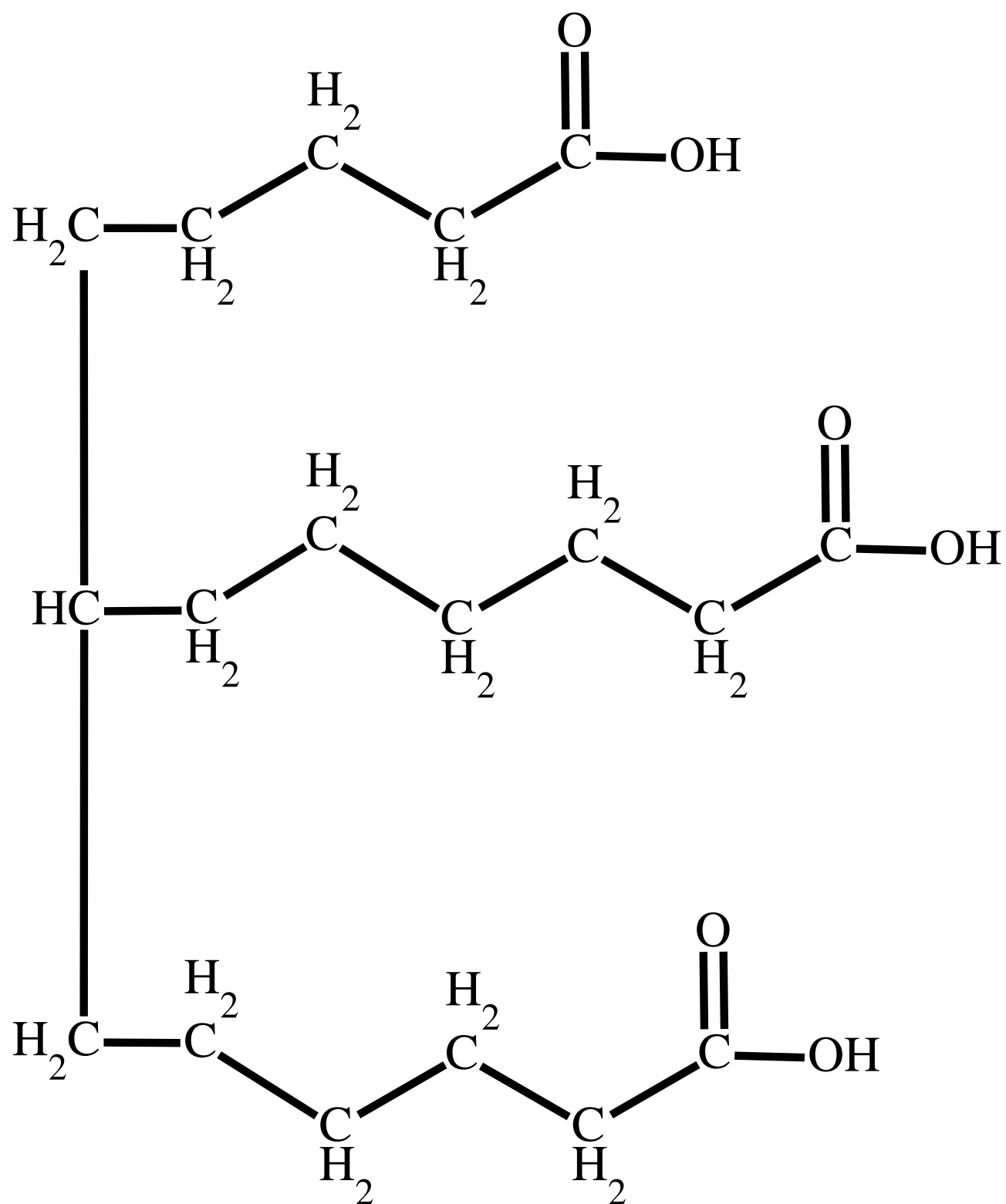






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E



F

- ☐ A
- ☐ B
- ☐ C
- ☐ D
- ☐ E
- ☐ F



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## Part C    Triglyceride functions

Which of the following are functions of triglycerides? Select all that apply.

- ☐ insulation & protection
  - ☐ act as biological catalysts
  - ☐ primary component of cell membranes
  - ☐ precursor for steroid hormones
  - ☐ energy storage
- 

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# Phospholipids

A Level



## Part A Phospholipid structure

A phospholipid is a particular type of lipid that contains one  molecule,  , and one phosphate group. It is, therefore, very similar in structure to a triglyceride, except that one of the  have been replaced by a .

Items:

three

four

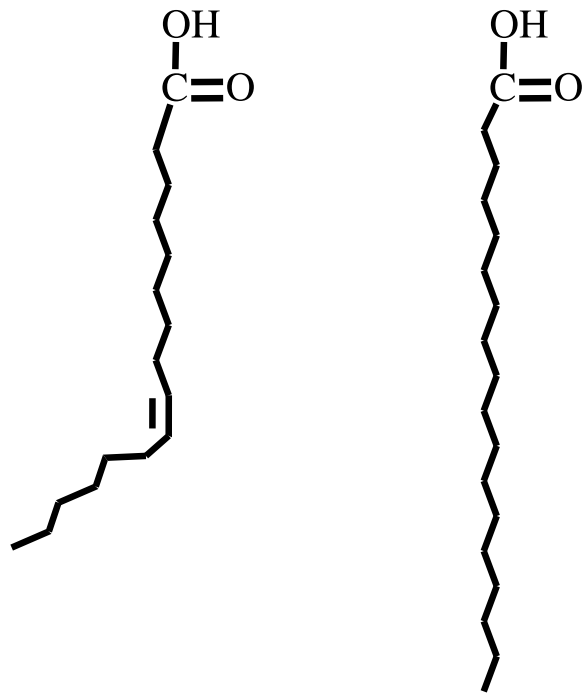
glycerol

two

fatty acids

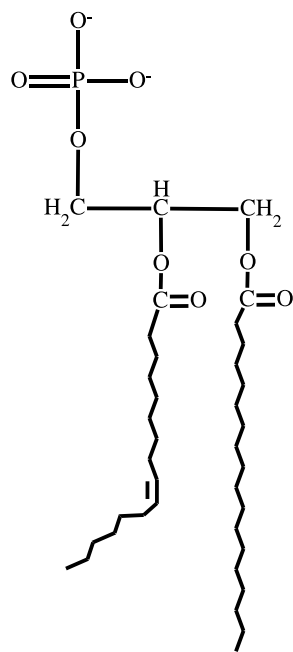
phosphate group

## Part B Phospholipid possibilities

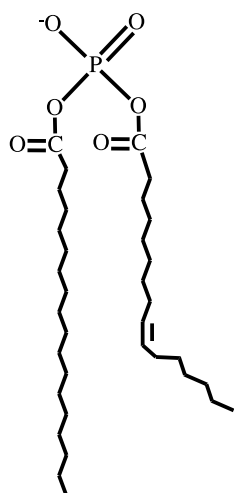


**Figure 1:** The simplified structures of two fatty acids are shown.

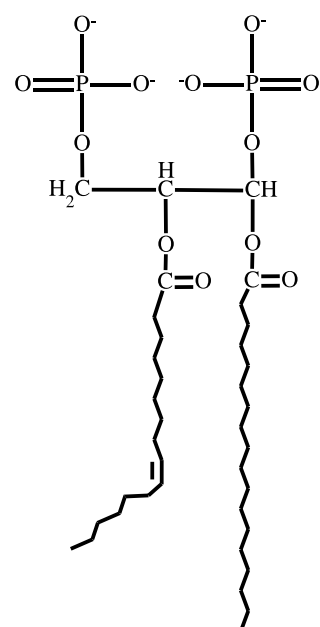
Which of the images below represent a phospholipid that could be formed from the fatty acids shown in Figure 1? Select all that apply.



A

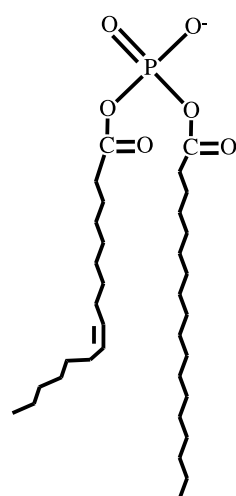


B



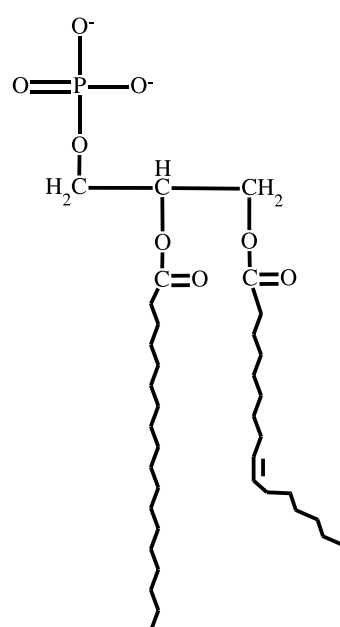

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C



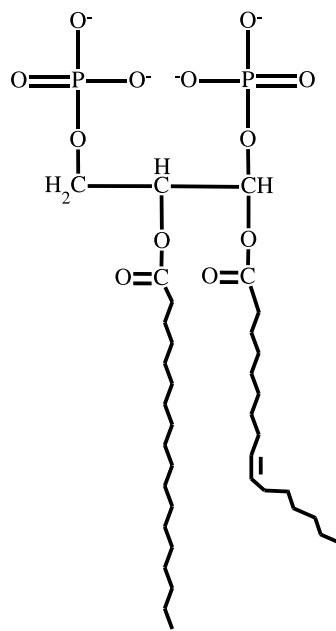

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D




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E



F

- ☐ A
- ☐ B
- ☐ C
- ☐ D
- ☐ E
- ☐ F

### Part C Phospholipid properties

The "head" of a phospholipid (composed of the ) is charged and therefore . The "tail" (composed of the ) is non-polar and therefore . Phospholipids are therefore described as  (molecules that contain both a hydrophilic and a hydrophobic part).

Items:

hydrophilic

phosphate group

fatty acids

hydrophobic

amphipathic

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## Part D    Phospholipid function

Which of the following is the main function of phospholipids?

- ☐ insulation & protection
  - ☐ energy storage
  - ☐ act as biological catalysts
  - ☐ forming cell and organelle membranes
  - ☐ precursor for steroid hormones
- 

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# Sterols

A Level



## Part A Sterol structure

Sterols are a type of lipid that have very different structures from other lipids (triglycerides and ). However, they are also classed as lipids because they are not water-soluble.

A sterol is a molecule with a core of  carbon-rings. This core is connected to a  at one end, and to a hydrocarbon chain at the opposite end. The  is polar and therefore , whereas the rest of the molecule is non-polar and therefore .

Sterols (e.g. cholesterol) are found in eukaryotic cell membranes, with the  facing out of the membrane (like the phospholipid heads) and the rest of the sterol facing into the membrane (like the phospholipid tails).

Items:

hydroxyl group

two

carboxyl group

four

three

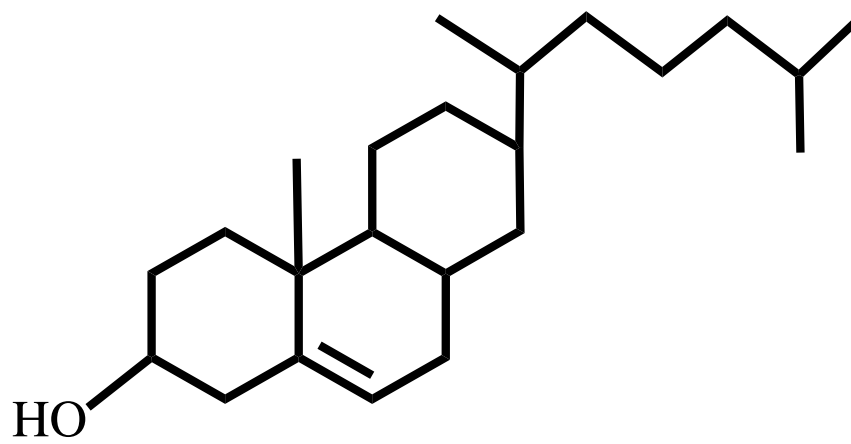
hydrophobic

hydrophilic

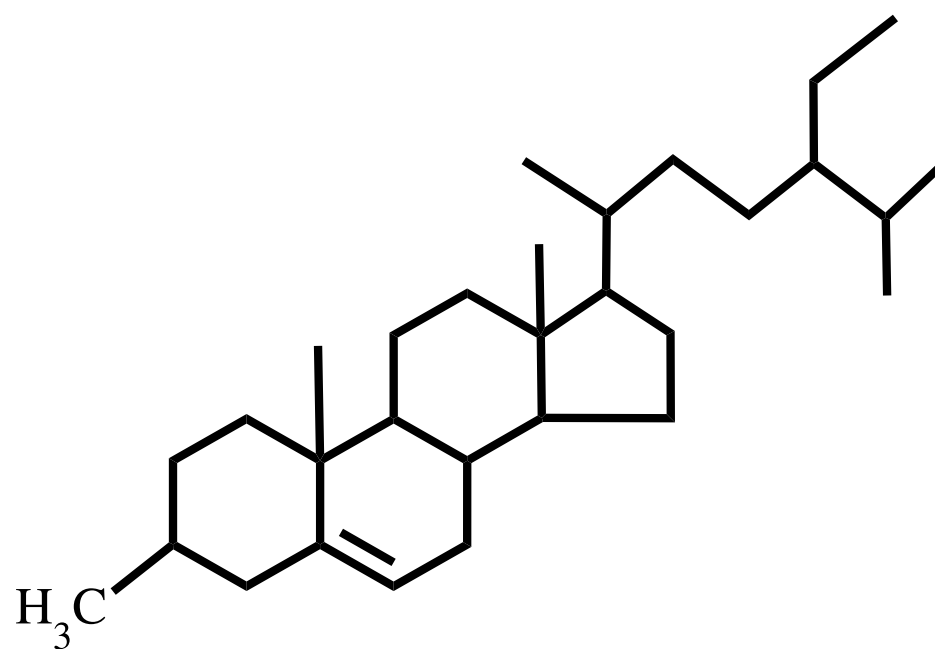
phospholipids

**Part B** Identify the sterols!

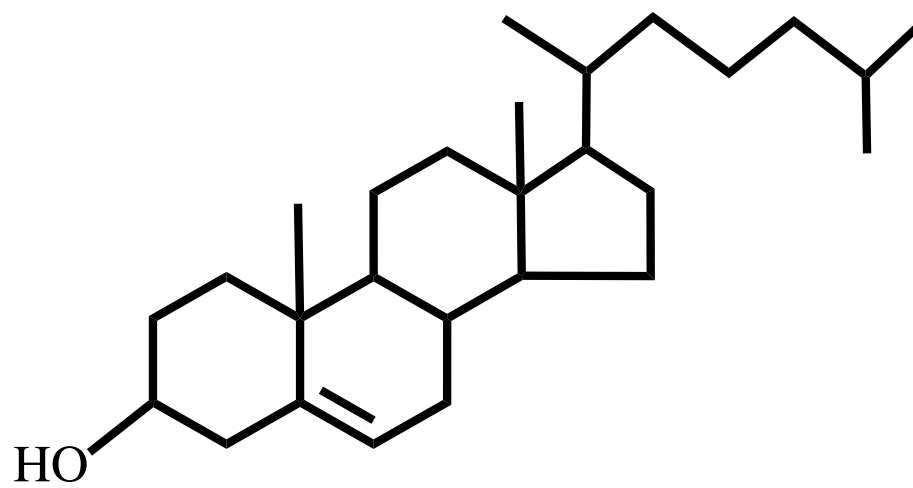
Which of the images below are sterols?



A

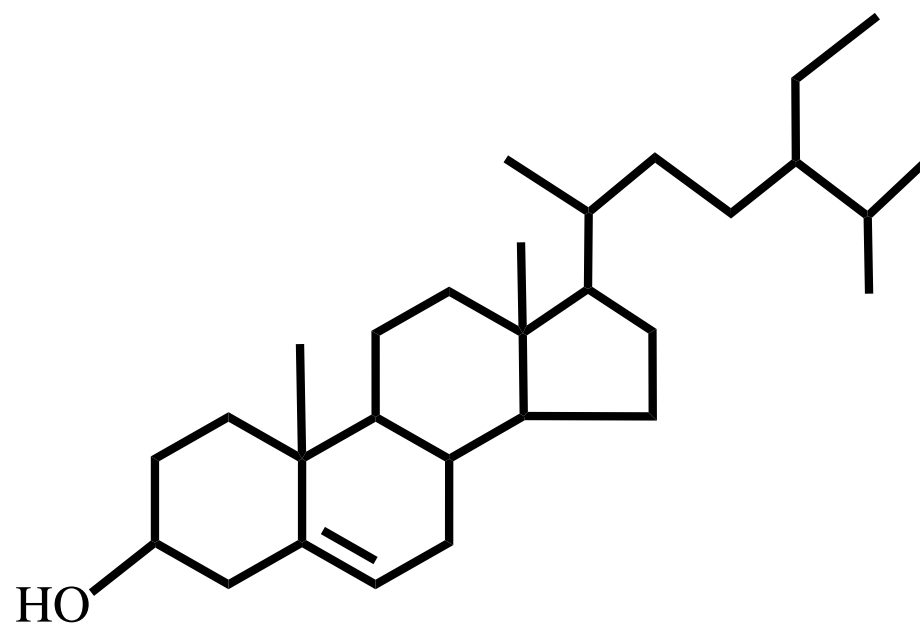


B

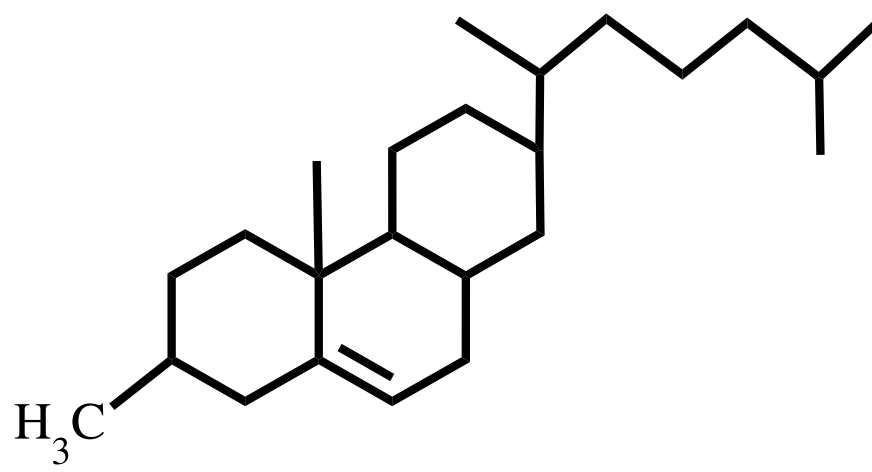


C

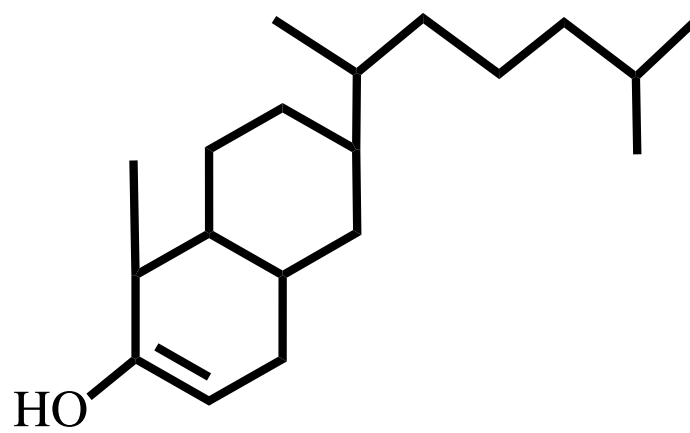




D



E



F

- ☐ A
- ☐ B
- ☐ C
- ☐ D

☐ E

☐ F

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### Part C    Cholesterol

Cholesterol is an important sterol in animals. Which of the following are functions of cholesterol?  
Select all that apply.

- ☐ act as biological catalysts
  - ☐ precursor for steroid hormones
  - ☐ insulation & protection
  - ☐ energy storage
  - ☐ regulates membrane fluidity
- 

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# Fatty Acid Saturation

Stearic acid is a fatty acid. It contains 18 carbon atoms and zero carbon–carbon double bonds. It can be represented by the notation C18 : 0, where 18 is the number of carbons and 0 is the number of carbon–carbon double bonds present.

Oleic acid can be represented by C18 : 1.

Linoleic acid can be represented by C18 : 2.

A triglyceride was formed using one of each of the three fatty acids.

## Part A   Saturation and melting points

Match the saturation type and relative melting point to the fatty acid.

Fatty acid	Saturation type	Melting point (relative)
Stearic acid	<div></div>	<div></div>
Oleic acid	<div></div>	<div></div>
Linoleic acid	<div></div>	<div></div>

Items:

- lowest
- unsaturated
- intermediate
- polyunsaturated
- saturated
- highest

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**Part B    Hydrogen numbers**

Within the triglyceride, how many hydrogen atoms does the stearic acid chain have?

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Within the triglyceride, how many hydrogen atoms does the oleic acid chain have?

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Within the triglyceride, how many hydrogen atoms does the linoleic acid chain have?

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**Part C    Oxygen numbers**

How many oxygen atoms does the triglyceride have?

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Part D Identify the fatty acids

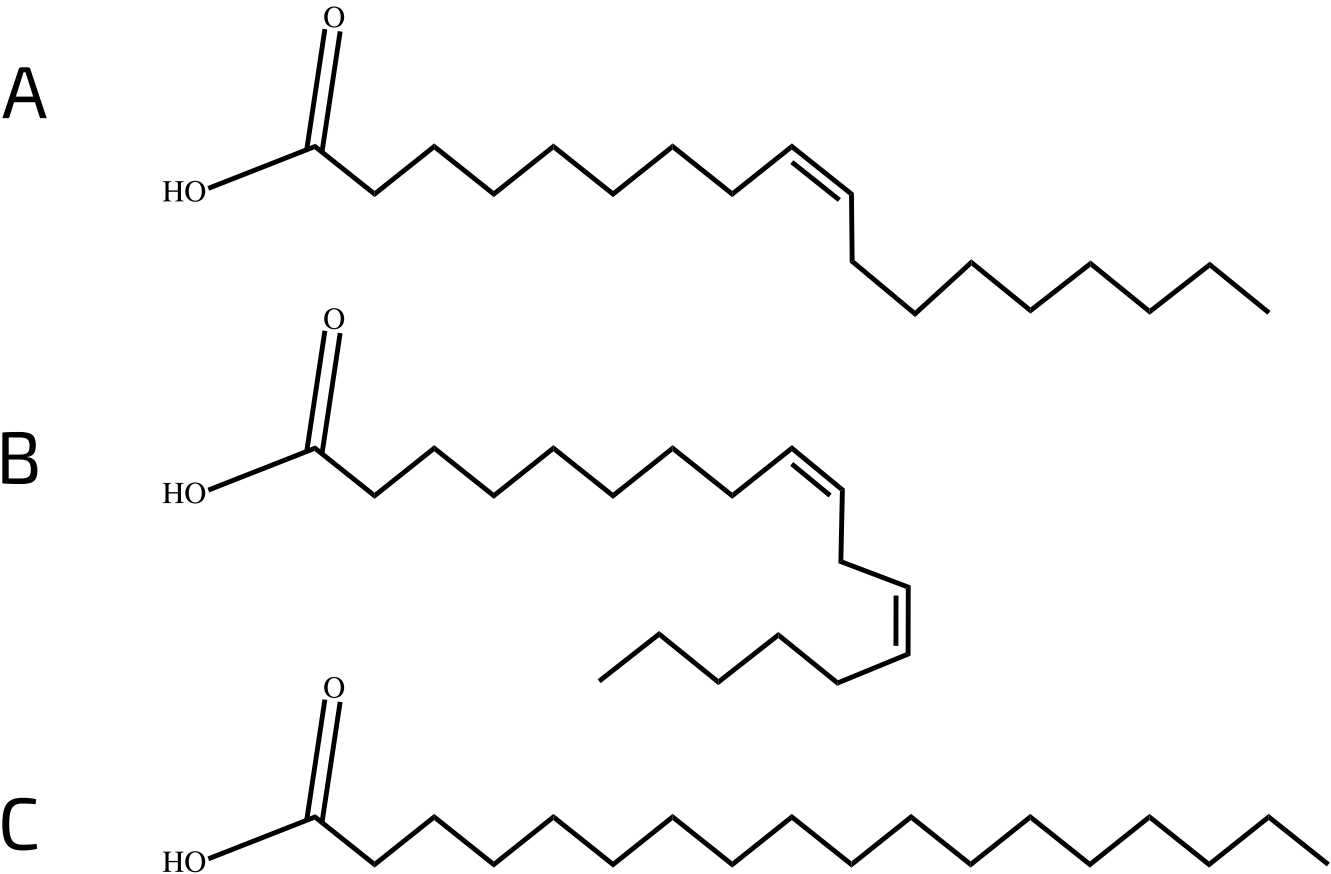


Figure 1: Structures of three fatty acids.

Match the fatty acids to the labels in Figure 1.

Letter	Fatty acid
A	<div></div>
B	<div></div>
C	<div></div>

Items:

- linoleic acid
- stearic acid
- oleic acid

Question elements adapted with permission from NSAA 2022 Specimen Paper Section 2 Q21



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# Testing For Lipids

A Level  
P P P

The test used to determine if lipids are present in a sample is . This involves adding  and water to the sample and shaking. If the solution remains clear then there are  in the sample. If  forms, then there are  in the sample.

Items:

no lipids

Benedict's reagent

the emulsion test

a red colour

ethanol

lipids

Benedict's test

a white emulsion

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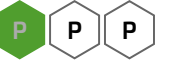
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# Lipids Overview

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A Level



## Part A Lipid definition

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What is the defining characteristic of a lipid?

- ☐ form part of cell membranes
  - ☐ insoluble in water
  - ☐ contains a hydrophilic region and a hydrophobic region
  - ☐ form bilayers
  - ☐ composed of amino acids
  - ☐ soluble in water
  - ☐ composed of monosaccharides
  - ☐ contains glycerol and one or more fatty acid
-

Part B    Lipid properties

Match the lipids to their properties.

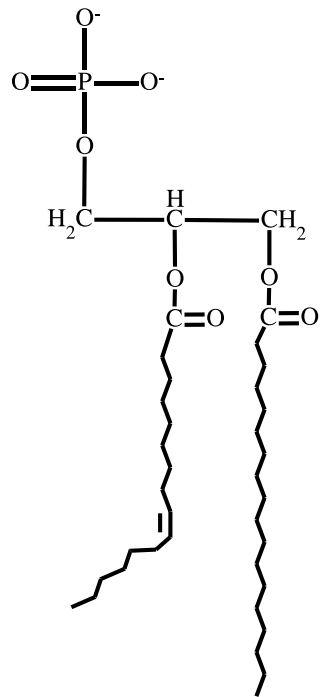
Lipid	Components	Polarity	Functions
<div></div>	glycerol, 3 fatty acids	<div></div>	energy storage, insulation, protection
<div></div>	glycerol, 2 fatty acids, phosphate group	<div></div>	<div></div>
<div></div>	4 carbon-rings, hydrocarbon chain, hydroxyl group	<div></div>	regulate membrane fluidity, precursor for steroid hormones

Items:

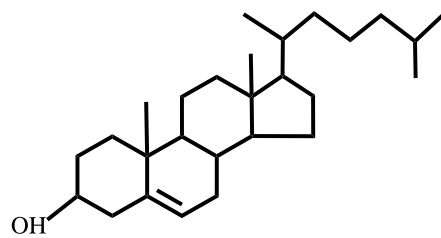
- hydrophobic
- form membranes
- amphipathic
- Phospholipids
- Sterols
- Triglycerides



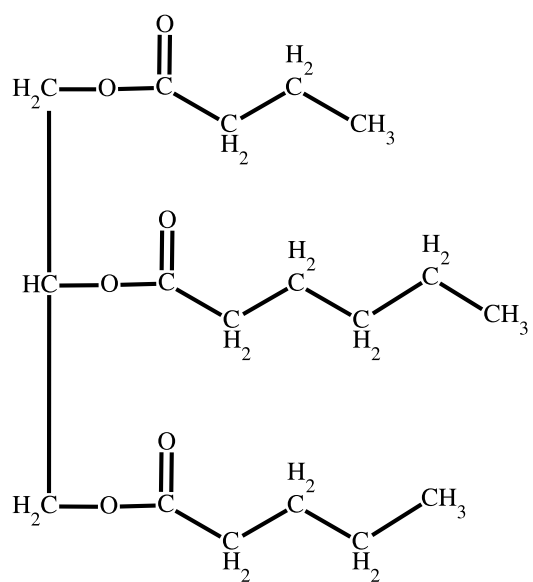
## Part C Lipid structures



A



B



C

Match the type of lipid to the image above.

A:

B:

C:

Items:

sterol

diglyceride

phospholipid

glycolipid

proteolipid

triglyceride

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