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Oxidative Phosphorylation



Oxidative phosphorylation is the final stage of aerobic respiration.

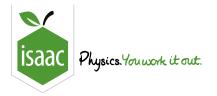
't A	The process
Du	uring oxidative phosphorylation, NADH (reduced NAD) and FADH ₂ (reduced FAD) are,
rel	leasing $ m H^+$ ions and electrons, and regenerating NAD $^+$ and FAD. The electrons move along a
se	ries of protein complexes embedded in the called the . The energy released
	uring this allows these protein complexes to actively transport \mathbf{H}^+ ions from the mitochondrial atrix to the
Th	nese H^+ ions then flow back into the mitochondrial matrix down their electrochemical gradient
thr	rough a transmembrane protein called providing the energy needed to synthesise ATP
fro	om ADP & Pi (inorganic phosphate). The maximum number of ATP molecules produced during this
sta	age (per molecule of glucose) is $pprox 34$, but the actual number varies depending on how much of the
en	nergy produced is used to make ATP, and how much is lost in the form of
Aft	ter reaching the final part of the electron transport chain, the electrons react with oxygen (the
) and the ${ m H}^+$ ions to form water (${ m H}_2{ m O}$).
Ite	ems:
_	
	ATP synthase cytoplasm electron transport chain outer mitochondrial membrane reduced
٦	final electron acceptor heat intermembrane space oxidised inner mitochondrial membrane

	oxygen
	water
	NAD ⁺
	NADH (reduced NAD)
	FAD
	FADH ₂ (reduced FAD)
	ADP
	ATP
	carbon dioxide
Pro	oducts
/hich o	of the following molecules are produced during oxidative phosphorylation?
Vhich o	of the following molecules are produced during oxidative phosphorylation?
/hich o	of the following molecules are produced during oxidative phosphorylation? oxygen water
/hich o	of the following molecules are produced during oxidative phosphorylation? oxygen water NAD ⁺
/hich o	of the following molecules are produced during oxidative phosphorylation? oxygen water
/hich o	of the following molecules are produced during oxidative phosphorylation? oxygen water NAD+ NADH (reduced NAD)
/hich o	of the following molecules are produced during oxidative phosphorylation? oxygen water NAD+ NADH (reduced NAD) FAD
Vhich o	of the following molecules are produced during oxidative phosphorylation? oxygen water NAD+ NADH (reduced NAD) FAD FADH ₂ (reduced FAD)

Part B

Reactants

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<u>Home</u> <u>Gameboard</u> <u>Biology</u> <u>Biochemistry</u> <u>Respiration</u> <u>Adenosine Triphosphate (ATP)</u>

Adenosine Triphosphate (ATP)



Adenosine triphosphate (ATP) is the main "energy transfer" molecule used in cells. Energy (from organic molecules in respiration, or from sunlight in photosynthesis) is used to synthesise ATP from adenosine diphosphate (ADP) and inorganic phosphate (P_i). This energy is stored within the ATP molecule and can then be released during ATP hydrolysis.

Figure 1: The molecular structure of adenosine triphosphate (ATP).

Part A X and Y

What is the name of the nitrogenous base labelled **X** in Figure 1?

What is the name of the sugar labelled Y in Figure 1?

Part B Inorganic phosphate

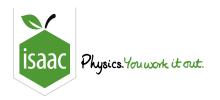
(inorga	respiration and photosynthesis, ATP is synthesised from ADP (adenosine diphosphate) and P _i nic phosphate).
What is	s meant by "inorganic phosphate"?
	a single phosphate ion $(PO_4^{\ 3^-})$ which is not bound to anything else
	a compound consisting of one or more phosphate groups but which is not bound to anything else
	any compound that contains one or more phosphate groups
\bigcirc	any compound that contains one or more phosphate groups but does not contain any carbon atoms
	emical energy
	pes the hydrolysis of ATP to ADP and P _i release energy?
	ATP is much more stable than ADP, which means that energy will be released when ATP is converted to ADP.
0	
0	ATP is much more stable than ADP, which means that energy will be released when ATP is converted to ADP.
	ATP is much more stable than ADP, which means that energy will be released when ATP is converted to ADP. ATP hydrolysis is a spontaneous reaction that requires no energy input.
	ATP is much more stable than ADP, which means that energy will be released when ATP is converted to ADP. ATP hydrolysis is a spontaneous reaction that requires no energy input. Breaking chemical bonds is exothermic, whereas forming chemical bonds is endothermic. The bonds between phosphate groups within ATP are "high-energy" bonds, which means they are very strong and
	ATP is much more stable than ADP, which means that energy will be released when ATP is converted to ADP. ATP hydrolysis is a spontaneous reaction that requires no energy input. Breaking chemical bonds is exothermic, whereas forming chemical bonds is endothermic. The bonds between phosphate groups within ATP are "high-energy" bonds, which means they are very strong and therefore release a lot of energy when they are broken.

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<u>Home</u> <u>Gameboard</u> Biology Biochemistry Respiration NAD and FAD

NAD and **FAD**



Part A	Function of NAD and FAD
Wh	at is the main function of NAD ⁺ and FAD in aerobic respiration?
	To act as the final electron acceptors during oxidative phosphorylation.
	To react with coenzyme A (CoA) to form acetyl CoA.
	To accept electrons from organic molecules and transfer them to the electron transport chain - where they are used to synthesise ATP.
	To be transported across the inner mitochondrial membrane, driving an electrochemical gradient that is then used to synthesise ATP.
	To act as energy transfer molecules which transfer energy by donating a phosphate group.
	To be broken down into smaller organic molecules in order to provide the energy needed for other reactions.

Part B Oxidation and reduction

In the table below, match the reaction type to the equation.

Equation	Reaction type
$NAD^{+} + H^{+} + 2e^{-} \longrightarrow NADH$	
$NADH \longrightarrow NAD^{+} + H^{+} + 2 e^{-}$	
$FAD + 2 H^{+} + 2 e^{-} \longrightarrow FADH$	
$FADH \longrightarrow FAD + 2 H^{+} + 2 e^{-}$	

Items:

oxidation reduction

Part C NADH and FADH2 production

Fill in the table below to give the number of molecules produced during each stage of aerobic respiration **per molecule of glucose**.

Stage	Molecule	
	NADH	FADH ₂
Glycolysis		
Link reaction		
Krebs cycle		
Oxidative phosphorylation		

Items:

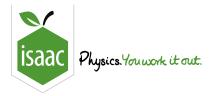


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Home Gameboard Biology Biochemistry Respiration Oxygen Levels

Oxygen Levels



Liver cells are frequently used as a source of mitochondria. These cells are homogenised in a sucrose solution and the mitochondria isolated. The suspended mitochondria are then placed in an oxygen electrode where the oxygen uptake of these organelles can be measured over a given time period.

The results of one of these experiments are shown in Figure 1.

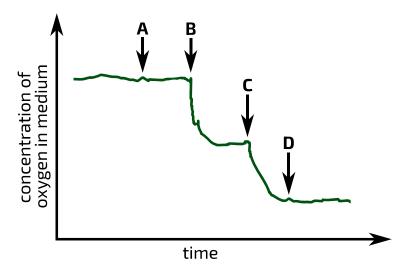


Figure 1: Oxygen concentration over time in a medium containing isolated mitochondria. At point A, glucose was added. At points B, C, and D, equal quantities of ADP were added.

Part A A to B

vvnicn	of the following explains the results between points A and B? Select all that apply.
	There is not enough glucose present for aerobic respiration to occur.
	There is not enough oxygen present for aerobic respiration to occur.
	The earlier stages of aerobic respiration do not use up oxygen, so the oxygen concentration remains constant during this period.
	Photosynthesis and respiration are occurring at equal rates, so the oxygen concentration remains constant during this period.
	The mitochondria are carrying out the process of fermentation, so the oxygen concentration remains constant during this period.
	Not enough ADP is present for the later stages of aerobic respiration to occur.
rt B B	to C
	to C of the following explains the results between points B and C? Select all that apply.
	of the following explains the results between points B and C? Select all that apply.
	of the following explains the results between points B and C? Select all that apply. The decrease in oxygen concentration slows down as ADP is used up (in making ATP).
	of the following explains the results between points B and C? Select all that apply. The decrease in oxygen concentration slows down as ADP is used up (in making ATP). Glycolysis is occurring during this period, so the oxygen concentration decreases.
	of the following explains the results between points B and C? Select all that apply. The decrease in oxygen concentration slows down as ADP is used up (in making ATP). Glycolysis is occurring during this period, so the oxygen concentration decreases. The rate of aerobic respiration slows down as oxygen begins to run out.

Part C D onwards

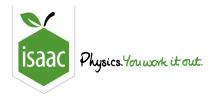
Which	of the following explains the results from point D onwards? Select all that apply.
	Only glycolysis is occurring from point D onwards, so the oxygen concentration remains constant.
	All of the glucose has been used up, so respiration cannot proceed.
	The rate of photosynthesis increased between points C and D, resulting in oxygen levels being kept constant.
	Not enough ADP is present for respiration to occur.
	All of the oxygen has been used up, so aerobic respiration cannot proceed.
	The lactic acid produced by the mitochondria has changed the pH of the solution such that the respiratory enzymes can no longer function.

 $Adapted\ with\ permission\ from\ OCR\ A\ Level\ June\ 2003,\ Science\ Modular\ Central\ Concepts\ in\ Biology,\ Question\ 3$

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<u>Home</u> <u>Gameboard</u> <u>Biology</u> <u>Biochemistry</u> <u>Respiration</u> Respiratory Quotients

Respiratory Quotients



The respiratory quotient (RQ) of a respiratory substrate is the number of CO_2 molecules produced divided by the number of O_2 molecules consumed during the aerobic respiration of one molecule of substrate, i.e.

$$RQ = \frac{CO_2 \, produced}{O_2 \, consumed}$$

Because the respiratory quotient differs between different respiratory substrates, it is possible to estimate the relative proportions of macronutrients being metabolised by measuring an individual's CO_2 production and O_2 consumption.

Part A Carbohydrates

Carbohydrates are the main type of respiratory substrate. Larger carbohydrates (polysaccharides) are broken down into glucose, which can then enter the glycolysis pathway.

Balance the equation below to calculate the respiratory quotient of glucose.

$$C_6H_{12}O_6 + O_2 \longrightarrow CO_2 + H_2O$$

What is the respiratory quotient of glucose?

Part B Fats

Carbohydrates are not the only type of respiratory substrate. Fats (triglycerides) are also used in aerobic respiration, after being broken down into glycerol and fatty acids. Glycerol can then enter the glycolysis pathway (after being converted into one of the intermediate molecules), and fatty acids can enter Krebs cycle (after being converted into acetyl CoA).

Balance the equation below to calculate the respiratory quotient of a particular fatty acid.

$$C_{18}H_{36}O_2+O_2 \longrightarrow CO_2+H_2O$$

What is the respiratory quotient of this fatty acid? Give your answer to 2 decimal places.

Part C Energy storage

Cells store energy in the form of triglycerides and polysaccharide carbohydrates.

What is the name of the energy-storage carbohydrate found in plant cells?

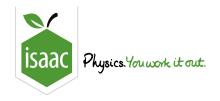
What is the name of the energy-storage carbohydrate in animal cells?

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<u>Home</u> <u>Gameboard</u> <u>Biology</u> <u>Biochemistry</u> <u>Respiration</u> Respiration Summary

Respiration Summary



Part A Respiration processes

Match the processes and cellular locations to the descriptions in the table below.

Process	Description	Location
	NADH and ${\rm FADH_2}$ are oxidised and the energy released is used to produce ATP. Water is also produced.	
	Pyruvate is reduced to lactate or ethanol in order to regenerate NAD ⁺ from NADH.	
	Pyruvate is used to make acetyl CoA. NADH and CO_2 are produced.	
	Glucose is broken down into two pyruvate molecules. NADH and ATP are produced.	
	Acetyl CoA goes through a series of reactions. NADH, FADH $_2$, ATP, and CO_2 are produced	
tems: the link reaction oxidative phosphorylation fermentation cytoplasm glycolysis Krebs cycle		
Line Innic reacti	Since the spinor procedure of the proced	Tarobb Cycle

Part B Molecule functions in aerobic respiration

Match the molecules to the functions in the table below.

Molecule	Function
	respiratory substrates
	electron carriers (accept electrons from organic molecules and donate them to the electron transport chain)
	final electron acceptor (accepts electrons from the electron transport chain)
	primary energy transfer molecule
	reacts with oxaloacetate (4C) to form citrate (6C) and restart the Krebs cycle
Items:	ohydrates NAD ⁺ and FAD oxygen acetyl CoA

Part C Aerobic and anaerobic respiration Which of the following processes are involved in aerobic respiration of glucose? glycolysis fermentation the link reaction Krebs cycle oxidative phosphorylation Which of the following processes are involved in anaerobic respiration of glucose? glycolysis fermentation the link reaction Krebs cycle oxidative phosphorylation

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