

<u>Gameboard</u>

Physics

Waves & Particles

Quantum

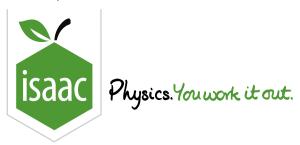
Essential Pre-Uni Physics D7.2

Essential Pre-Uni Physics D7.2



Physical constants which may be necessary to answer the problems on this page can be found within the hint tabs.

A laser diode requires $3.2\,\mathrm{V}$ across it to make it work. This means that its photons will have an energy of $3.2\,\mathrm{eV}$. Calculate the wavelength of the light emitted.



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Essential Pre-Uni Physics D7.7

Essential Pre-Uni Physics D7.7



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Caution - when working with particles, do not use $c = f\lambda$. Question D7.9 shows you why.

Part A Momentum from KE

Calculate the momentum of an electron if its kinetic energy is $10\,\mathrm{keV}$.

Part B Momentum from wavelength

An electron's wavelength is $3.0 \times 10^{-7}\,\mathrm{m}$. What is its momentum?

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Physics

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Essential Pre-Uni Physics D6.5

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A graph of stopping potential (y) against frequency of light (x) is plotted for zinc, and also for aluminium. Without knowing more information, answer the following questions:

Part A Linear graphs?

a) Are the lines straight or not?

No

Yes

Part B Sign of the y-intercepts

b) Are the y-intercepts positive, negative or zero?

Positive

Zero

Negative

Part C	Sign of the gradients
c) Are the gradients positive, negative or zero?	
	Negative
	Positive
	Zero
Part D	Comparing the gradients
d) Are the gradients of the two lines the same or different?	
	Different
	Same
Part E	Comparing the y-intercepts
e)	Are the y -intercepts of the two lines the same or different?
	Different
	Same

Part F The x-intercept

f) What is the significance of the x-intercept?

It is the frequency of the incident light.

It is the threshold frequency.

It is the work function of the material.

Part G Common gradient or intercept

g) If you answered `same' to parts (d) or (e), write down the value of the common gradient or intercept.

Gradient not common

Common intercept = $\frac{e}{h}$

Common gradient = $\frac{h}{e}$

Intercept not common

Common gradient = h

Intercept not common

Common gradient = $\frac{h}{e}$

Common intercept = h

Common gradient = -he

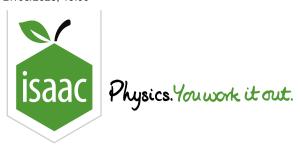
Intercept not common

Gradient not common

Common intercept = $\frac{\text{work function}}{\text{electric charge}}$

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Essential Pre-Uni Physics D6.4

Essential Pre-Uni Physics D6.4



Useful physical constants can be found in the hint tabs.

Part A Maximum speed of photoelectrons

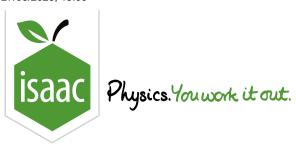
Calculate the maximum speed of the photoelectrons emitted when a material with an $8.4 \times 10^{-20} \, \mathrm{J}$ work function is illuminated by light of frequency $7.0 \times 10^{14} \, \mathrm{Hz}$.

Part B Minimum speed of photoelectrons

What is the minimum speed of the photoelectrons emitted?

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Waves & Particles

Quantum

Essential Pre-Uni Physics D9.1

Essential Pre-Uni Physics D9.1



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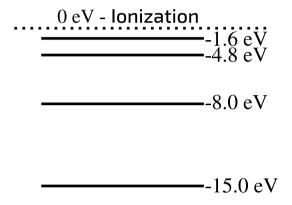
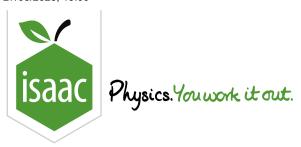


Figure 1: Energy level diagram of the atom this question is concerned with.

How much energy (in eV) is needed to ionize the atom if it is in its ground state?

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Physics

Waves & Particles

Quantum

Essential Pre-Uni Physics D9.3

Essential Pre-Uni Physics D9.3



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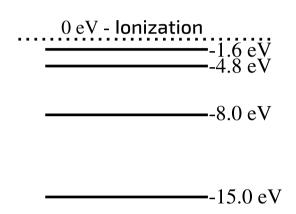
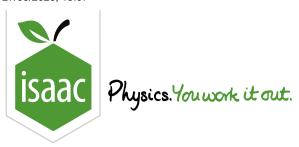


Figure 1: Energy level diagram of the atom this question is concerned with.

If the atom is in the $-1.6\,\mathrm{eV}$ state, and the electron descends to the ground state in three separate stages, what is the wavelength of the least energetic photon emitted?

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Essential Pre-Uni Physics D9.5

Essential Pre-Uni Physics D9.5



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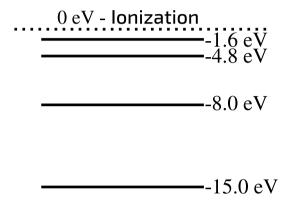


Figure 1: Energy level diagram of the atom this question is concerned with.

Part A Ground state

A $10\,\mathrm{eV}$ photon passes through this atom when it is in the ground state. How much energy is the atom likely to absorb from the photon? Give your answer in electron volts.

Part B First excited state

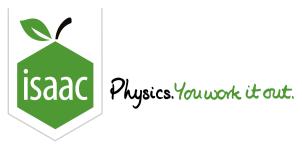
Repeat the question if the atom starts in the $-8.0\,\mathrm{eV}$ state. You should give a different answer to Part A if at all possible.

Part C Incident electron

For the situation where a $10\,\mathrm{eV}$ electron passes through the atom when it is in its ground state, how much energy is the atom likely to absorb from the electron? You should have a different answer to Part A.

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Waves & Particles

Nuclear

Essential Pre-Uni Physics J1.1

Essential Pre-Uni Physics J1.1



Complete the nuclear equation. Don't forget the neutrino / antineutrino if it is a beta decay! A periodic table is available here, however you shouldn't need it.

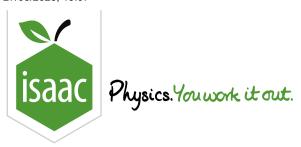
$$^{241}_{95}\mathrm{Am}
ightarrow \dots$$
 (Alpha decay)

$$\qquad \qquad ^{241}_{96}{\rm Np} \quad + \quad ^{0}_{-1}{\rm e} \quad + \quad ^{0}_{0}\overline{\nu} \\$$

$$\bigcirc \qquad ^{237}_{93}\mathrm{Np} \quad + \quad ^{4}_{2}\mathrm{He}$$

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Physics

Waves & Particles

Nuclear

Essential Pre-Uni Physics J1.2

Essential Pre-Uni Physics J1.2

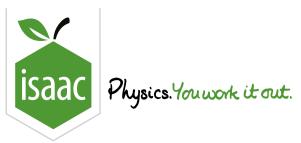


Complete the nuclear equation. Don't forget the neutrino / antineutrino if it is a beta decay! A periodic table is available here, however you shouldn't need it.

$$\bigcirc \qquad {}^3_2\mathrm{He} \quad + \quad {}^0_{-1}\mathrm{e} \quad + \quad {}^0_0\overline{\nu}$$

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Physics

Waves & Particles

Nuclear

Essential Pre-Uni Physics J1.9

Essential Pre-Uni Physics J1.9



Complete the nuclear equation. Don't forget the neutrino / antineutrino if it is a beta decay! A periodic table is available here, however you shouldn't need it.

$$^{23}_{13}{
m Al}
ightarrow \dots$$
 (Beta $+$ decay)

$$\qquad \qquad ^{23}_{12}\mathrm{Mg} \quad + \quad ^{0}_{1}\mathrm{e} + \quad ^{0}_{0}\overline{\nu}$$