



Amino Acids

A Level



Part A General structure

An amino acid is an organic molecule, containing a central atom bound to a hydrogen atom and to three other chemical groups: group (NH_2), group (COOH), and group/side-chain - which is the part that differs in structure among different amino acids.

There are standard amino acids that are coded for by the universal genetic code. Of these standard amino acids, humans can only synthesise 11, and so we have to get the other 9 from our diet. These 9 amino acids are sometimes called amino acids.

Items:

a carboxyl

a nitrate

an amino

an alcohol

oxygen

an R

essential

20

carbon

100

dispensable

Part B Amino acid or not?

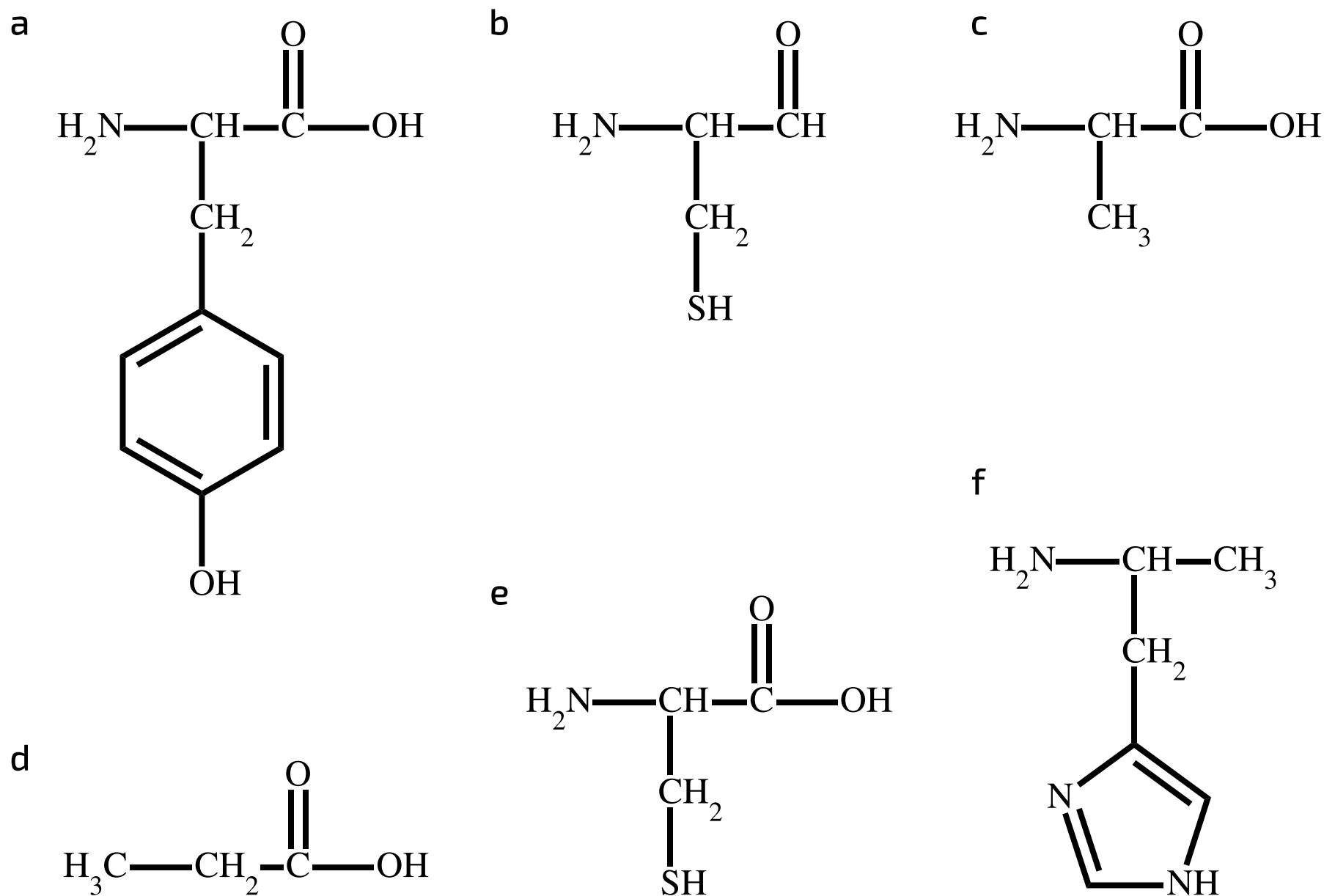


Figure 1: The molecular structures of six organic molecules (a-f).

Which of the organic molecules in Figure 1 are amino acids? Select all that apply.

- ☐ a
- ☐ b
- ☐ c
- ☐ d
- ☐ e
- ☐ f
- ☐ none of them

Part C Dipeptide identification

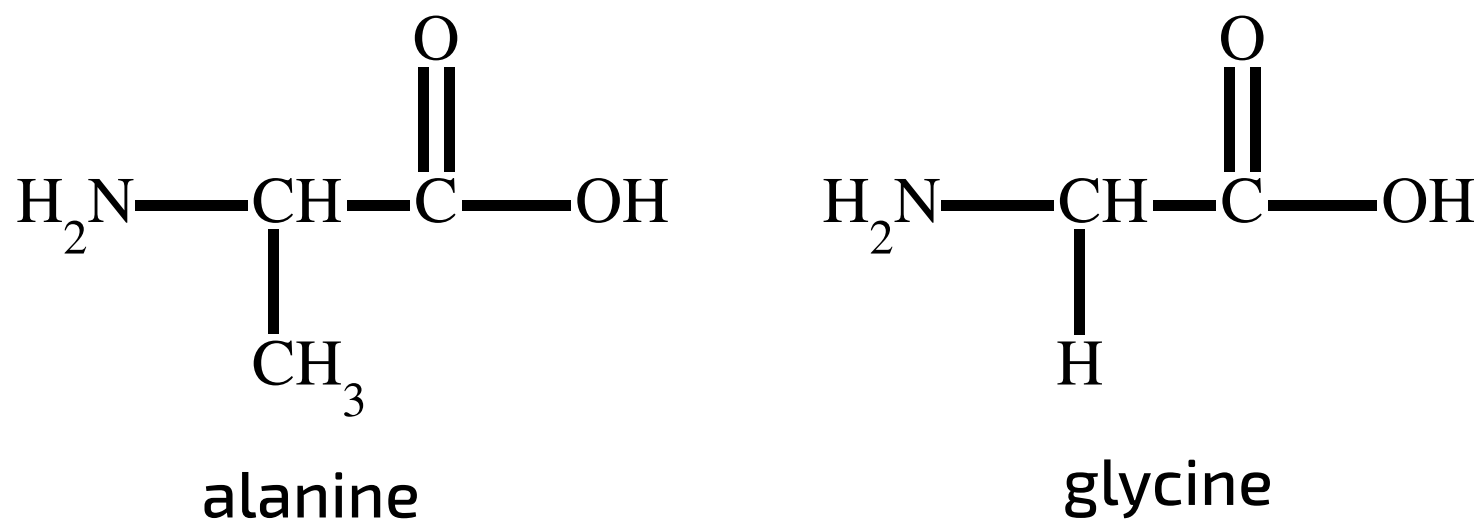
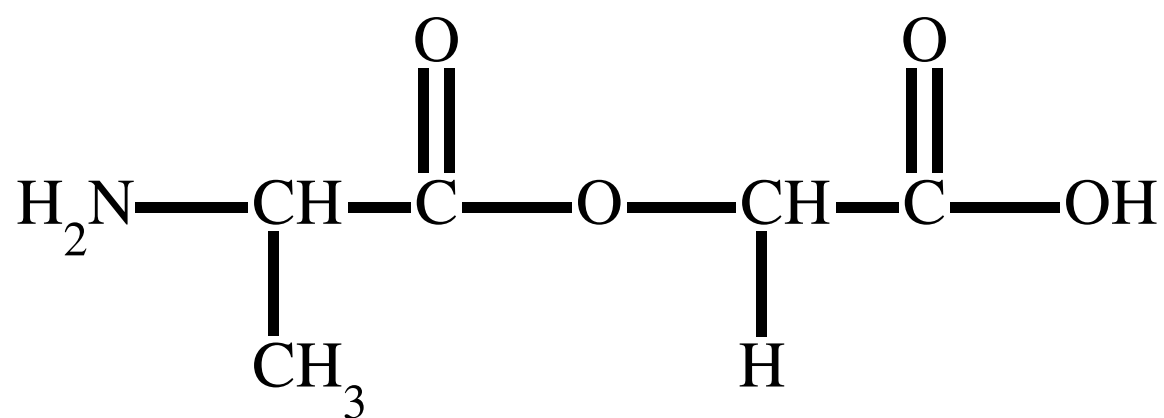
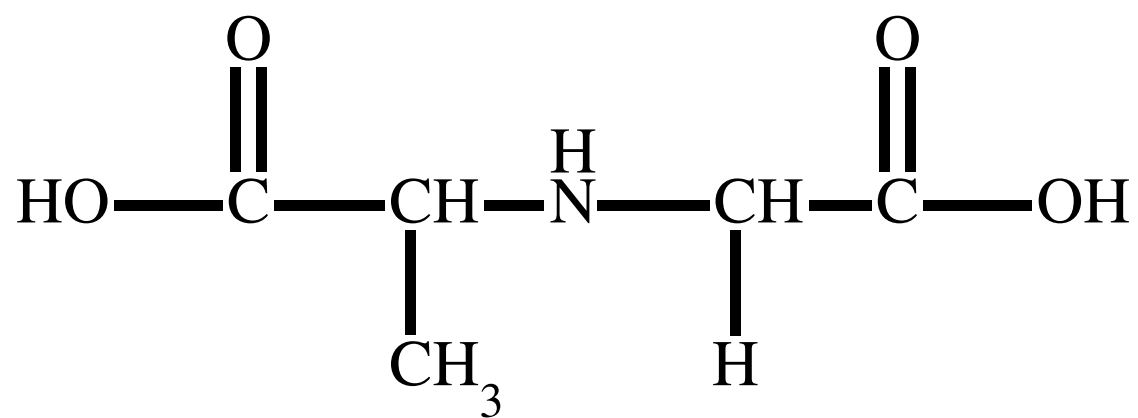


Figure 2: The molecular structures of two amino acids: alanine and glycine.

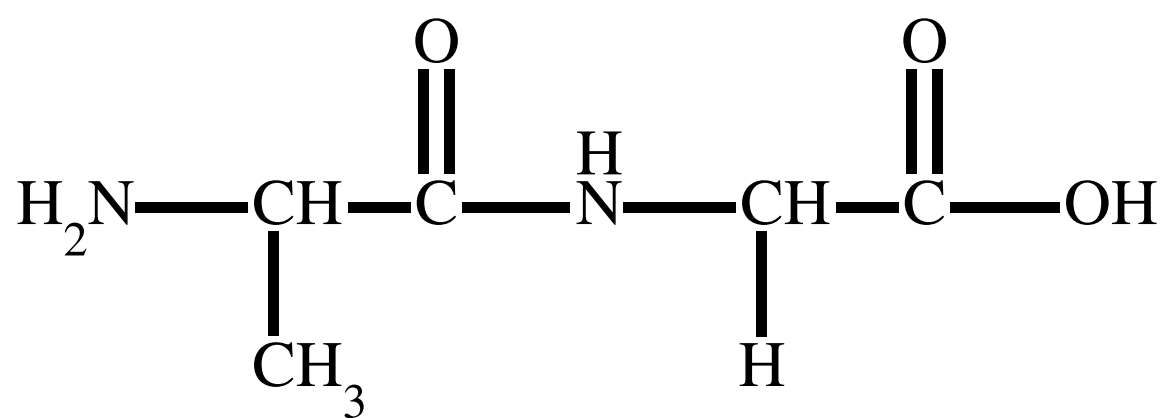
Figure 2 shows the molecular structures of two amino acids: alanine and glycine. Which molecule below represents the dipeptide that would be formed from these amino acids?



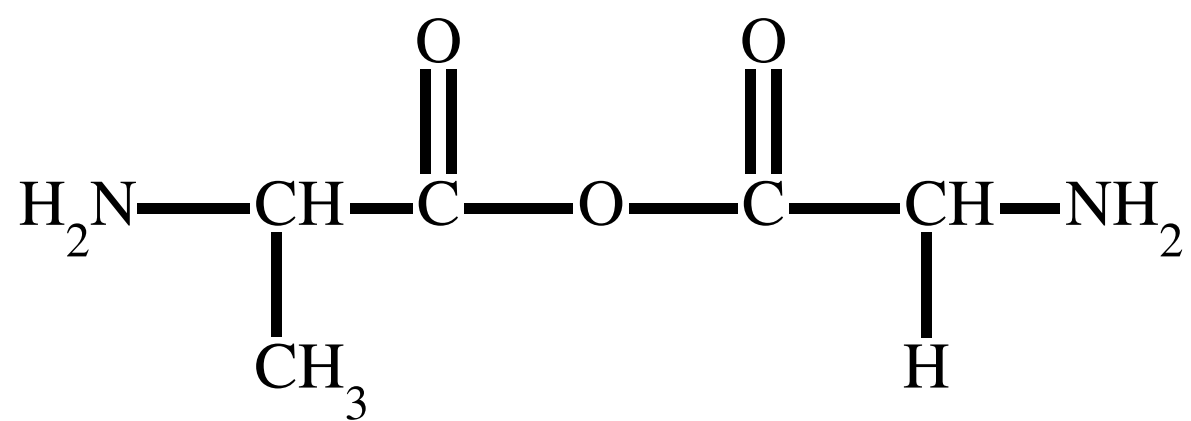
Molecule A



Molecule B



Molecule C



Molecule D

- ☐ A
- ☐ B
- ☐ C
- ☐ D



Protein Primary Structure

Part A The building blocks

Proteins are polymers made up of called . Proteins range in size from just a few amino acids to tens of thousands of amino acids, but most are between 50 and 2 000 amino acids.

A chain of two amino acids is called a . A chain of many amino acids is called a .

The sequence of amino acids is called the of the protein.

Items:

glucose

secondary structure

polypeptide

quaternary structure

monosaccharides

amino acids

dipeptide

monomers

primary structure

diamide

tertiary structure

polyamine

Part B Formation and breakdown

Amino acids join together by the process of . During this process, the group (NH_2) of one amino acid reacts with the group (COOH) of another amino acid to form a bond and .

Proteins are broken down into amino acids by the process of . During this process, is used to break apart the bond.

Items:

amino

hydrolysis

condensation

carboxyl

CO_2

ester

H_2O

peptide

nitrate

Part C Polypeptide possibilities

There are 20 different standard amino acids that are used to build proteins. How many primary structures could be produced for a protein that is 50 amino acids long? Give your answer to 2 significant figures.

Created for isaacphysics.org by Lewis Thomson

Gameboard:

STEM SMART Biology Week 3

All materials on this site are licensed under the **Creative Commons license**, unless stated otherwise.



Levels of Protein Structure



Part A Levels overview

Match the terms to the definitions.

: the association of several polypeptides with each other and (in some cases) with non-protein groups e.g. haemoglobin is made of four polypeptides and four haem groups (iron-containing organic molecules). Not all proteins have this level of structure, as some are only made of one polypeptide.

: the folding of the polypeptide due to hydrogen bonds between the H of one amino acid's NH group (within the peptide bond) and the O of another amino acid's CO group (within the peptide bond). Depending on the amino acid sequence, these hydrogen bonds can cause the polypeptide chain to form a tight coil (α -helix) or a long, snaking chain (β -sheet).

: the sequence of amino acids in the polypeptide (e.g. methionine-alanine-glycine-tyrosine).

: the folding of the polypeptide due to interactions between R side-chains of different amino acids (which are able to interact due to coiling/zig-zagging caused by hydrogen bonds). These interactions include ionic bonds (between carboxyl and amino groups within the R side-chains), disulfide bridges (between the S of one amino acid and the S of another), and hydrophilic/hydrophobic interactions (i.e. the polypeptide will fold such that hydrophilic R side-chains are on the outside, and hydrophobic R side-chains are on the inside).

Items:

Primary structure

Secondary structure

Tertiary structure

Quaternary structure

Part B Primary structure

Which of these describes the primary structure of a protein? Select all that apply.

- ☐ the association between multiple protein-subunits and non-protein groups to form a single, large protein
 - ☐ the snaking of a polypeptide chain to form a β -sheet
 - ☐ the coiling of a polypeptide chain to form an α -helix
 - ☐ the sequence of amino acids in a polypeptide chain e.g. methionine-glycine-alanine-glycine-lysine-alanine-leucine
 - ☐ the 3D folding of a polypeptide chain due to hydrogen bonds, disulfide bridges, ionic bonds, and hydrophobic/hydrophilic interactions
-

Part C Secondary structure

Which of these describes the secondary structure of a protein? Select all that apply.

- ☐ the snaking of a polypeptide chain to form a β -sheet
 - ☐ the sequence of amino acids in a polypeptide chain e.g. methionine-glycine-alanine-glycine-lysine-alanine-leucine
 - ☐ the 3D folding of a polypeptide chain due to hydrogen bonds, disulfide bridges, ionic bonds, and hydrophobic/hydrophilic interactions
 - ☐ the coiling of a polypeptide chain to form an α -helix
 - ☐ the association between multiple protein-subunits and non-protein groups to form a single, large protein
-

Part D Tertiary structure

Which of these describes the tertiary structure of a protein? Select all that apply.

- ☐ the sequence of amino acids in a polypeptide chain e.g. methionine-glycine-alanine-glycine-lysine-alanine-leucine
 - ☐ the coiling of a polypeptide chain to form an α -helix
 - ☐ the association between multiple protein-subunits and non-protein groups to form a single, large protein
 - ☐ the snaking of a polypeptide chain to form a β -sheet
 - ☐ the 3D folding of a polypeptide chain due to hydrogen bonds, disulfide bridges, ionic bonds, and hydrophobic/hydrophilic interactions
-

Part E Quaternary structure

Which of these describes the quaternary structure of a protein? Select all that apply.

- ☐ the coiling of a polypeptide chain to form an α -helix
 - ☐ the snaking of a polypeptide chain to form a β -sheet
 - ☐ the association between multiple protein-subunits and non-protein groups to form a single, large protein
 - ☐ the 3D folding of a polypeptide chain due to hydrogen bonds, disulfide bridges, ionic bonds, and hydrophobic/hydrophilic interactions
 - ☐ the sequence of amino acids in a polypeptide chain e.g. methionine-glycine-alanine-glycine-lysine-alanine-leucine
-

Created for isaacphysics.org by Lewis Thomson

Gameboard:

STEM SMART Biology Week 3

All materials on this site are licensed under the **Creative Commons license**, unless stated otherwise.



Types of Proteins

Part A Definitions

Most proteins can be categorized as either fibrous or globular, based on their structures.

Fill in the table below, comparing fibrous and globular proteins.

	Fibrous	Globular
Primary structure	<div></div>	<div></div>
Tertiary/quaternary structure	<div></div>	<div></div>
Solubility in water	<div></div>	<div></div>
Function(s)	<div></div>	<div></div>

Items:

- repetitive sequence

insoluble

non-repetitive sequence

simple: long & linear

structural
- chemical interactions

soluble

complex: highly folded

Part B Fibrous proteins

Which of the following are examples of fibrous proteins? Select all that apply.

- ☐ insulin (the hormone that causes cells to increase their uptake of glucose and convert it to glycogen and/or triglycerides)
 - ☐ elastin (the protein that gives elasticity to blood vessel walls)
 - ☐ keratin (a type of protein found in hair, nails, and skin)
 - ☐ α -amylase (the enzyme that breaks down starch into disaccharides and trisaccharides)
 - ☐ collagen (a major component of tendons, ligaments, bones, and skin)
 - ☐ haemoglobin (the protein that transports oxygen through the bloodstream)
-

Part C Globular proteins

Which of the following are examples of globular proteins? Select all that apply.

- ☐ insulin (the hormone that causes cells to increase their uptake of glucose and convert it to glycogen and/or triglycerides)
 - ☐ collagen (a major component of tendons, ligaments, bones, and skin)
 - ☐ haemoglobin (the protein that transports oxygen through the bloodstream)
 - ☐ elastin (the protein that gives elasticity to blood vessel walls)
 - ☐ keratin (a type of protein found in hair, nails, and skin)
 - ☐ α -amylase (the enzyme that breaks down starch into disaccharides and trisaccharides)
-

Part D Conjugated proteins

Conjugated proteins are proteins that have a non-protein component, which is called a .

is a conjugated protein, as it is made of four polypeptides - each bound to a haem group (an organic molecule containing).

Examples of other non-protein components include carbohydrates and lipids.

Items:

Haemoglobin

prosthetic group

an Fe^{2+} ion

fibrous

globular

a Ca^{2+} ion

Insulin

Collagen

Created for isaacphysics.org by Lewis Thomson

Gameboard:

STEM SMART Biology Week 3

All materials on this site are licensed under the **Creative Commons license**, unless stated otherwise.



Physics. *You work it out.*

[Home](#) [Gameboard](#) [Biology](#) [Biochemistry](#) [Proteins](#) [Protein Practicals](#)

Protein Practicals



Part A Testing for proteins

What is the name of the test used to determine if proteins are present in a solution?

Fill in the blanks to explain how this test works.

The protein solution is added to (e.g. sodium hydroxide solution), and dilute copper(II) sulfate solution is added to this. The solution will change colour from to , due to a reaction between the copper(II) ions and the .

Items:

carboxyl groups

an alkaline solution

an acidic solution

blue

red

peptide bonds

R side-chains

purple

Part BThin-layer chromatography

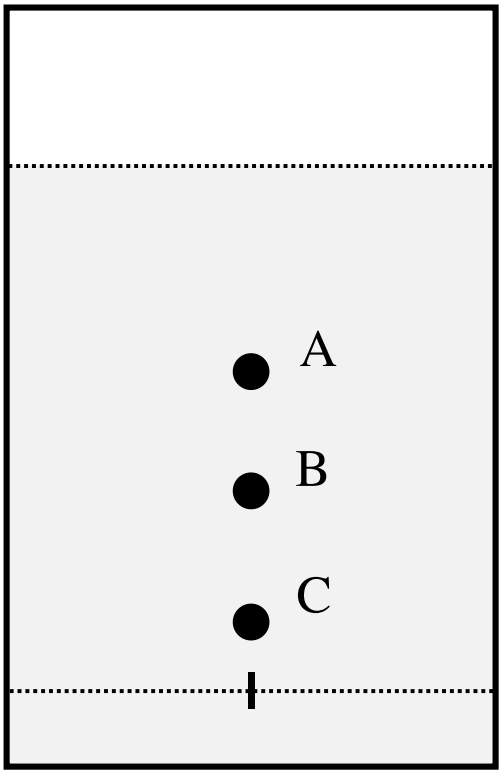


Figure 1: A thin-layer chromatography (TLC) plate was prepared for one solution ("Solution X"), which contained three amino acids. After 10 minutes, the plate was removed and sprayed with ninhydrin spray to visualise the amino acids (labelled A-C). The solvent front (top dotted line) had moved a distance of 18 cm from the starting position (bottom dotted line). "A" moved 10.98 cm. "B" moved 6.84 cm. "C" moved 2.34 cm. The retention factor (R_f) values for some amino acids are given in the table below.

Amino acid	R_f value
alanine	0.38
arginine	0.20
cysteine	0.40
glutamine	0.13
methionine	0.55
phenylalanine	0.68
serine	0.27
threonine	0.35
valine	0.61

Which amino acids are present in solution X?

- ☐ alanine
- ☐ arginine
- ☐ cysteine
- ☐ glutamine

- ☐ methionine
 - ☐ phenylalanine
 - ☐ serine
 - ☐ threonine
 - ☐ valine
-

Created for isaacphysics.org by Lewis Thomson

Gameboard:

STEM SMART Biology Week 3

All materials on this site are licensed under the **Creative Commons license**, unless stated otherwise.



Enzymes

Part A Enzyme overview

Enzymes are proteins that function as in biological reactions. Enzymes are essential for life because they the activation energy of fundamental reactions i.e. they allow these reactions to happen much . The complex tertiary/quaternary structure of enzymes ensures that enzymes are highly in terms of the reactions they catalyse.

Items:

globular

faster

lower

general

substrates

catalysts

specific

increase

slower

fibrous

Part B Enzyme reactions

An enzyme catalyses a reaction by binding to the to form an . The part of the enzyme that binds is called the . After the reaction is complete, the releases the . The enzyme is then free to catalyse another reaction.

Items:

product(s)

enzyme-substrate complex

enzyme-product complex

substrate(s)

active site

Part C Models of enzyme action

: the of the enzyme perfectly matches the shape of the , which ensures complete specificity.

: the of the enzyme changes shape in response to the binding, and only matches the shape after this initial binding.

Items:

Induced-fit model

substrate

product

active site

cofactor

Lock-and-key model

Part D Limiting factors

Which of the following are limiting factors in all enzyme-controlled reactions? Select all that apply.

- ☐ pH
 - ☐ concentration of product
 - ☐ concentration of substrate
 - ☐ temperature
 - ☐ concentration of enzyme
-

Part E Enzyme examples

Enzyme	Reactant(s)	Product(s)
Amylases	<div></div>	<div></div>
Proteases	<div></div>	<div></div>
Lipases	<div></div>	<div></div>
Catalase	<div></div>	<div></div>

Items:

- proteins

starch

fats

fatty acids & glycerol

hydrogen peroxide

maltose

peptides/amino acids

water & oxygen

Part F Stopping an enzyme from working

What is the name given to a non-substrate molecule that binds to the active site of an enzyme, and thus prevents the substrate from binding?

What is the name given to a non-substrate molecule that binds to part of the enzyme that is **not** the active site, but in doing so causes the shape of the active site to change, such that the substrate can no longer bind?



Physics. *You work it out.*

Enzyme Reactions

A Level

C

C

C

Part A Product formation over time

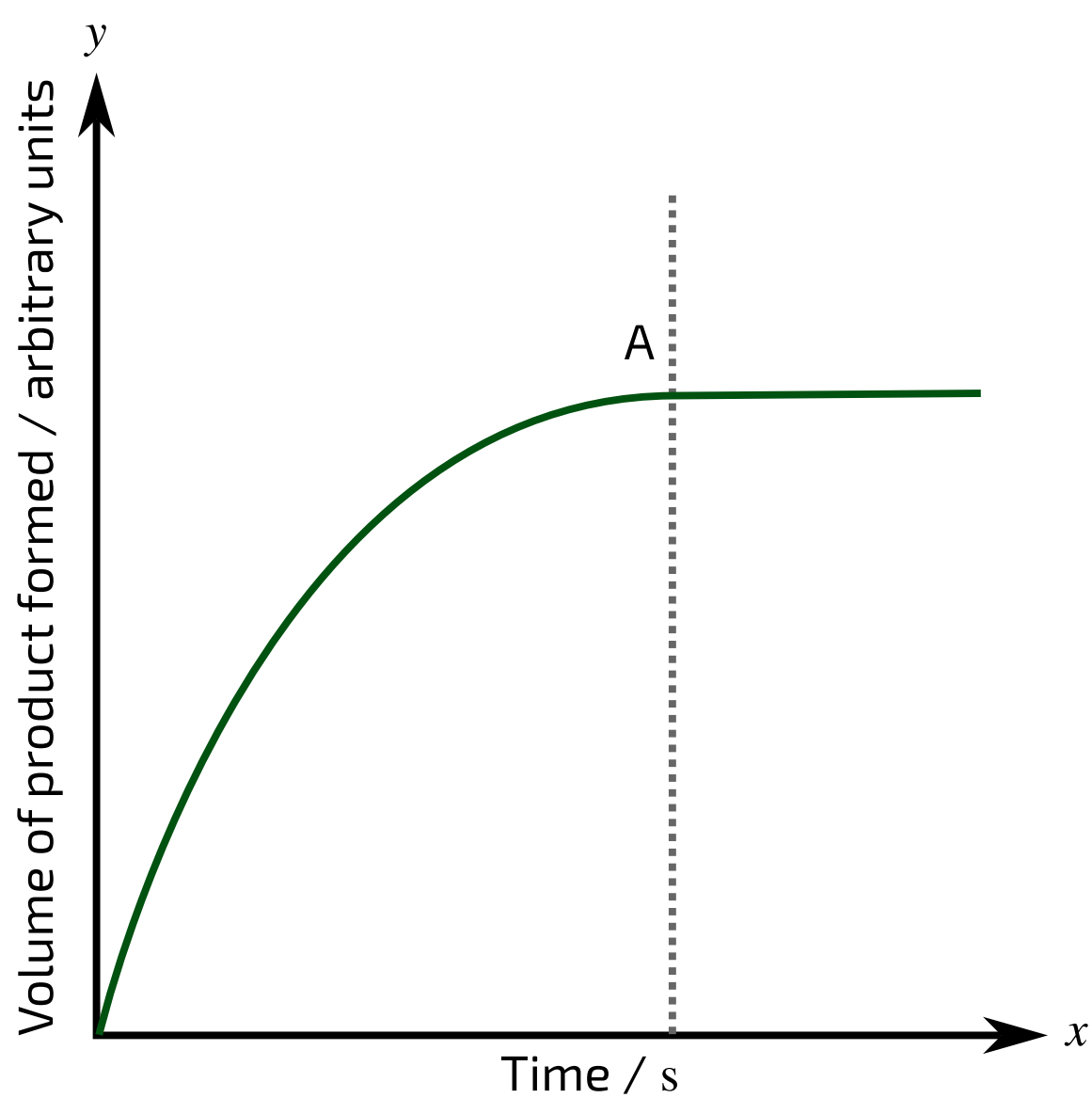


Figure 1: The volume of product formed over time for an enzyme-controlled reaction. Temperature and pH were kept constant throughout the reaction.

Which of the following could explain why no more product is being formed after point A in Figure 1?
Select all that apply.

- ☐ All of the substrate has been used up in the reaction
 - ☐ All of the enzyme has been used up in the reaction
 - ☐ The enzyme has become denatured
 - ☐ The product is acting as a cofactor
 - ☐ The product is acting as a competitive inhibitor
-

Part B Reaction rate over temperature

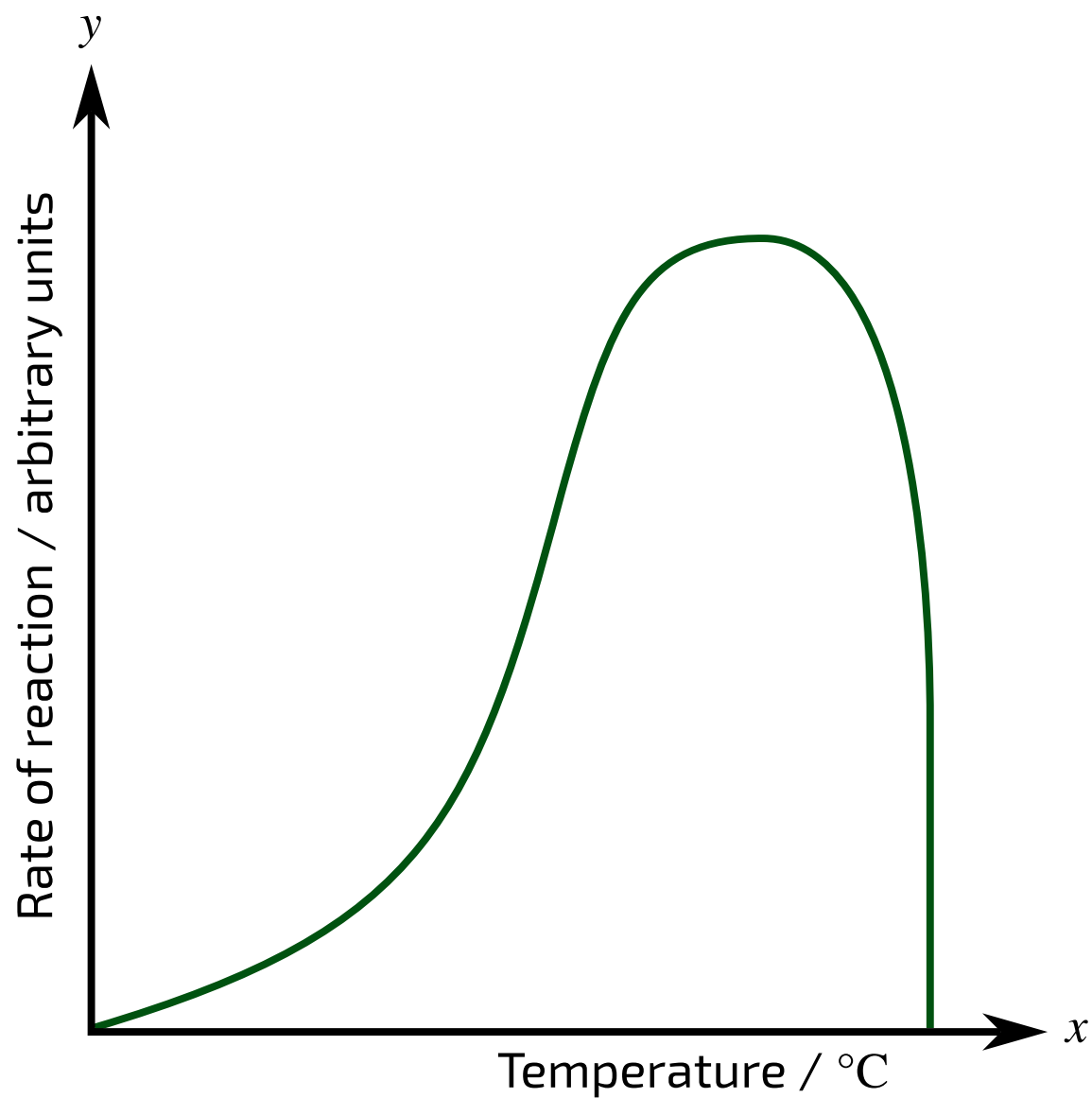


Figure 2: The reaction rate at different temperatures for an enzyme-controlled reaction.

Which of the following correctly describe the relationship shown in Figure 2?

- ☐ The rate of reaction increases over time until all the substrate is used up, after which point the reaction stops.
- ☐ The rate of reaction increases over time until the product begins to act as a competitive-inhibitor, which slows the reaction down.
- ☐ The rate of reaction positively correlates with temperature across the range of temperatures measured.
- ☐ The rate of reaction increases with temperature up to the enzyme's optimal temperature, beyond which the enzyme is denatured and the reaction will not work.

Question elements adapted with permission from NSAA 2021 Section 1 Q67

All materials on this site are licensed under the [Creative Commons license](#), unless stated otherwise.