



**NANYANG
TECHNOLOGICAL
UNIVERSITY**

Energy Research Institute @ NTU

Decarbonisation Technical Workshop of Energy Innovation 2021



H2FC Research @ NTU

23 July 2021

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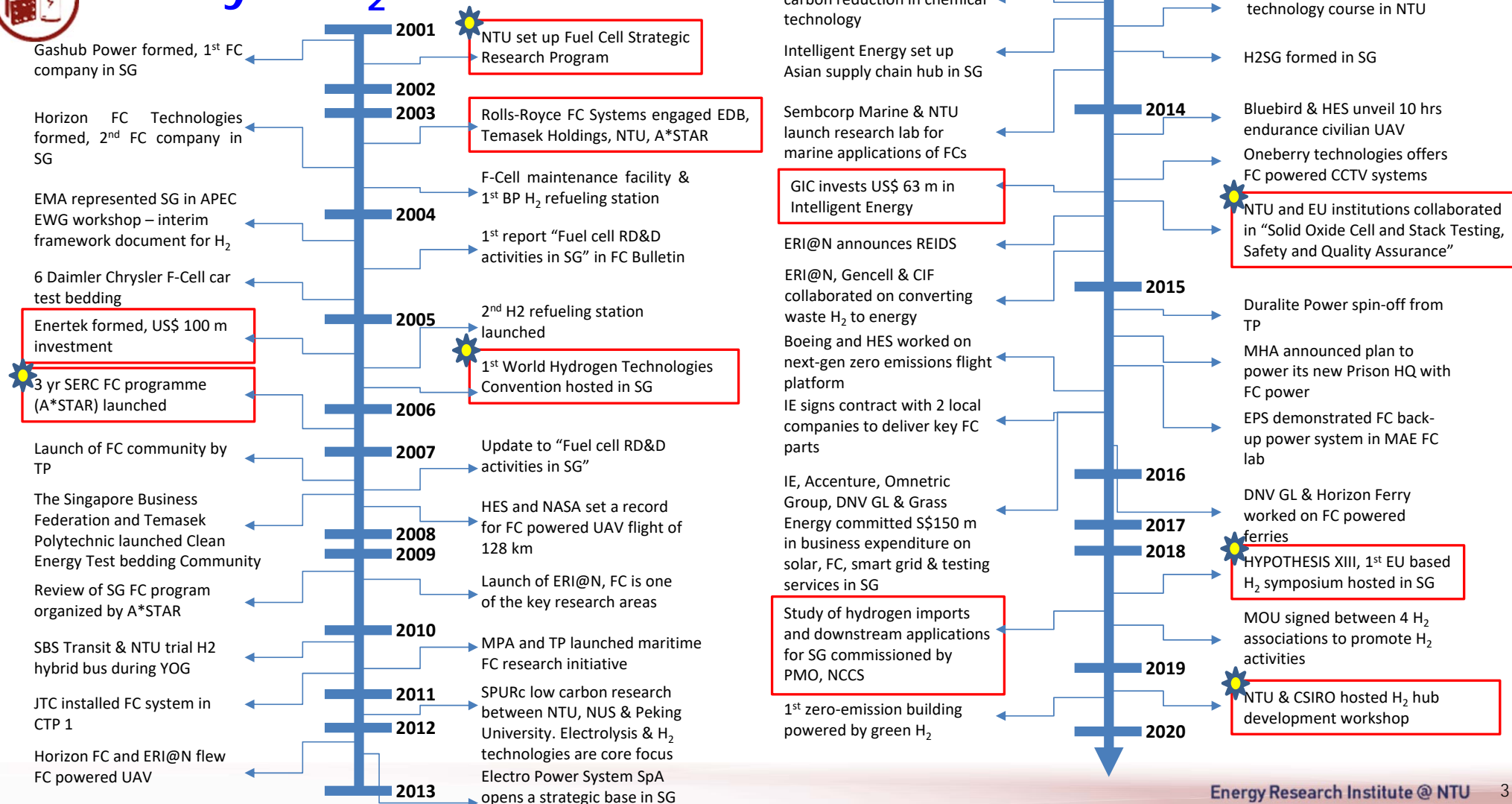


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- H2FC research @ NTU in the last 30 years
- Commercialization efforts
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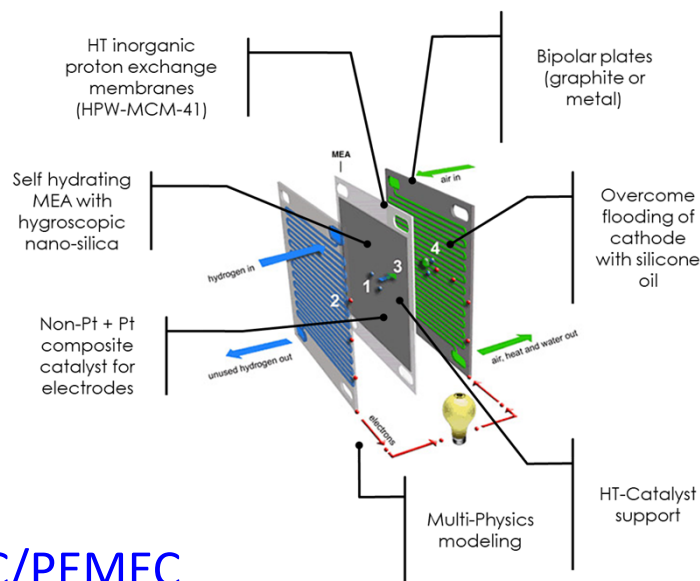


History of H₂ Activities in SG

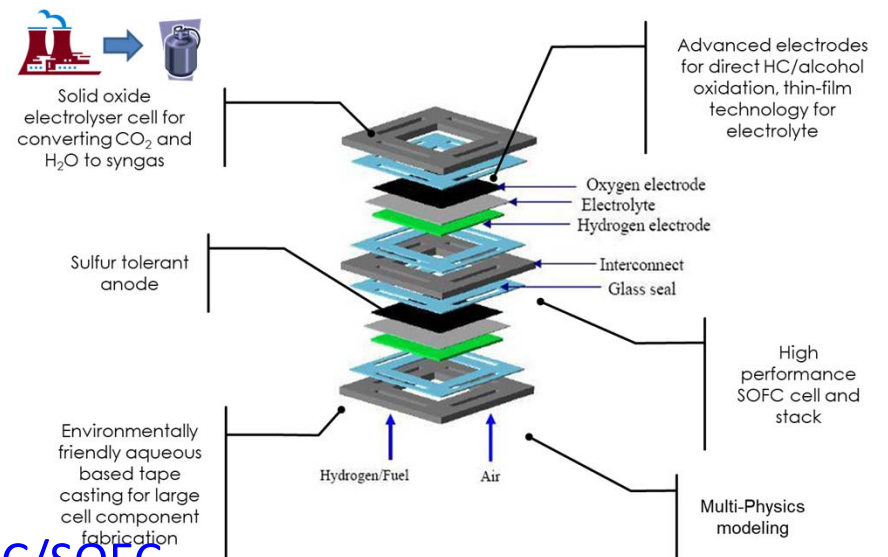




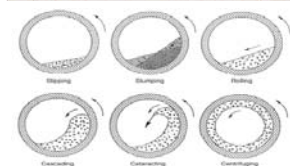
PEMFC/PEMEC



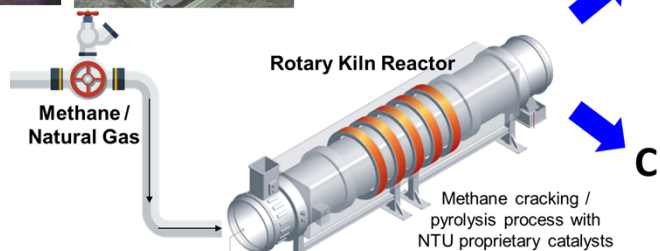
SOFC/SOEC



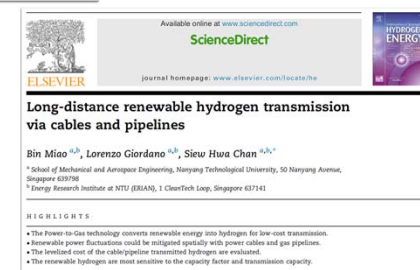
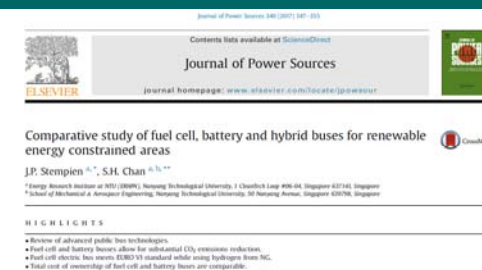
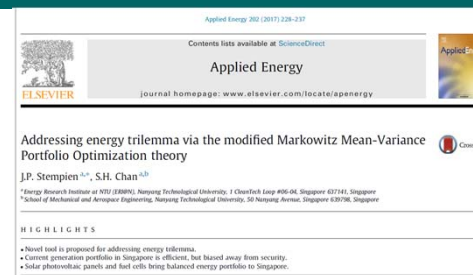
Reforming



CDM



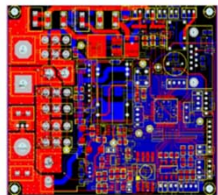
Feasibility study



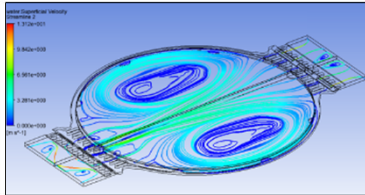


Snapshot of H2FC Research @ NTU over the last ~30 years

| PEMFC + EC | SOFC + EC | Hydrogen Technologies |



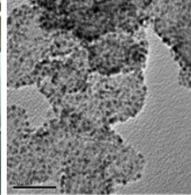
Control circuit board
& programming



Thermo fluids & thermodynamic
analysis



Stack
development



Catalyst development



FC charger



H2 based range
extender



100W-1kW



Mini fuel cell

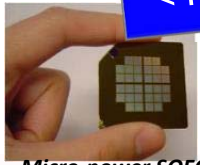
< 1 W

1-10 W

10-100 W



UAVs



Micro-power SOFC



H2-On-Demand



SOFC/EC



Methanol based
range extender



FC based backup power



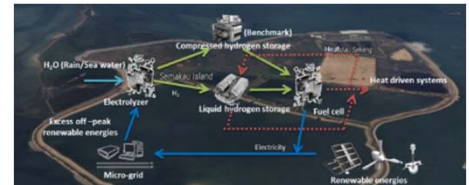
FC hybrid bus for YoG



Regenerative fuel cell system



Distributed Power
Generation



H₂ demo @ Semakau Island



Remote power generator



Four Major Laboratories *for H₂FC Research and Development*



MAE

(Fabrication & Characterization)



CSIJRI @ GKC
(Commercialization)



FC Fab
& Demo



CBE

(Power to Gas)

Labs

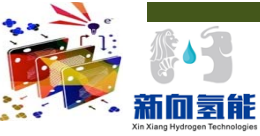
Catalysts

P2G



ERI@N
(Novel Catalyst)





Xin-Xiang (GZ) Hydrogen Technologies

Commercialization through China-Singapore International Joint Research Institute @ GKC

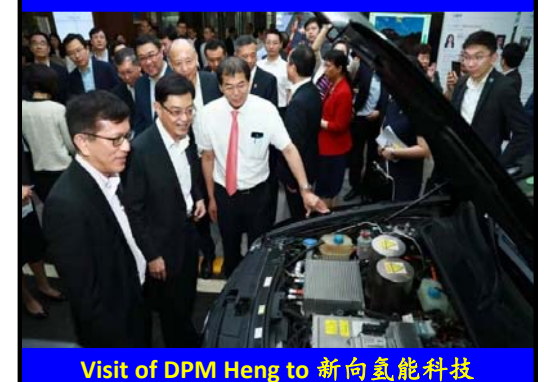
- Catalyst mass production
- CCM/MEA mass production



Visit of Guangdong Governor Ma Xingrui



Visit of DPM Heng to CSIJRI



Visit of DPM Heng to 新向氢能科技



NTU-Sydney Energy Partnership

- Sydney Energy: a JV company of Nanofilm and Temasek
- Sydney Energy Core Activity: Manufacture of Fuel Cells
- Scope of collaboration: PEMFC technology, key components design and manufacturing, characterization.

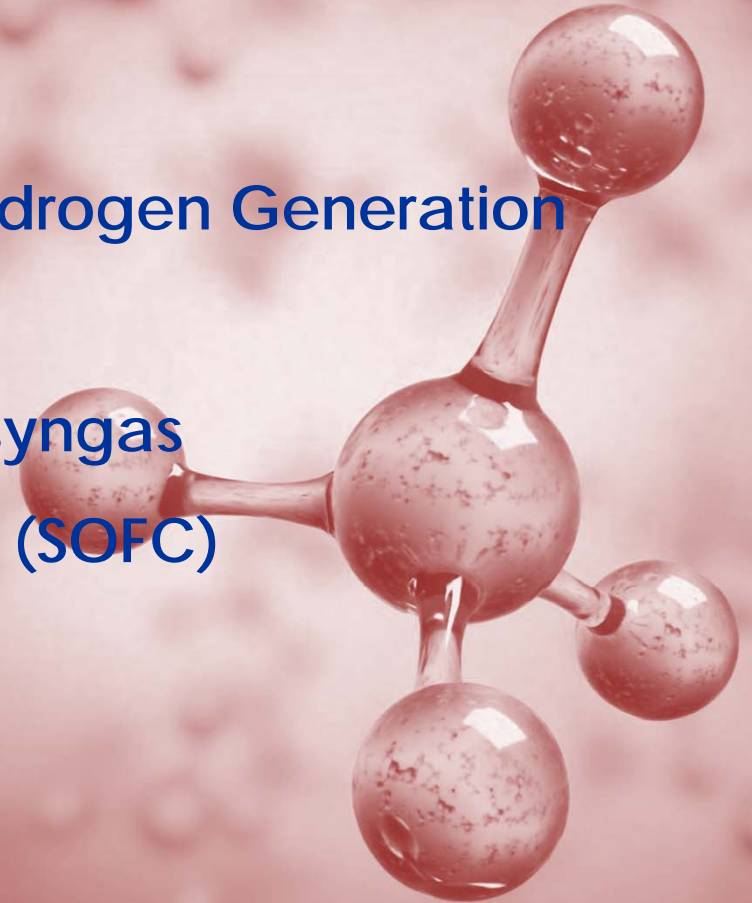


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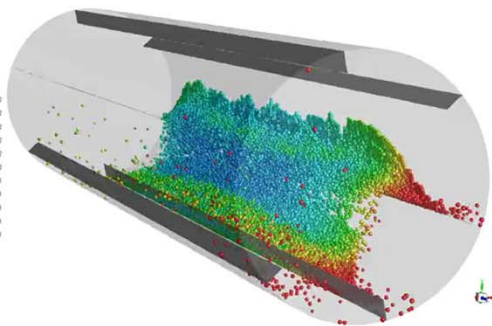
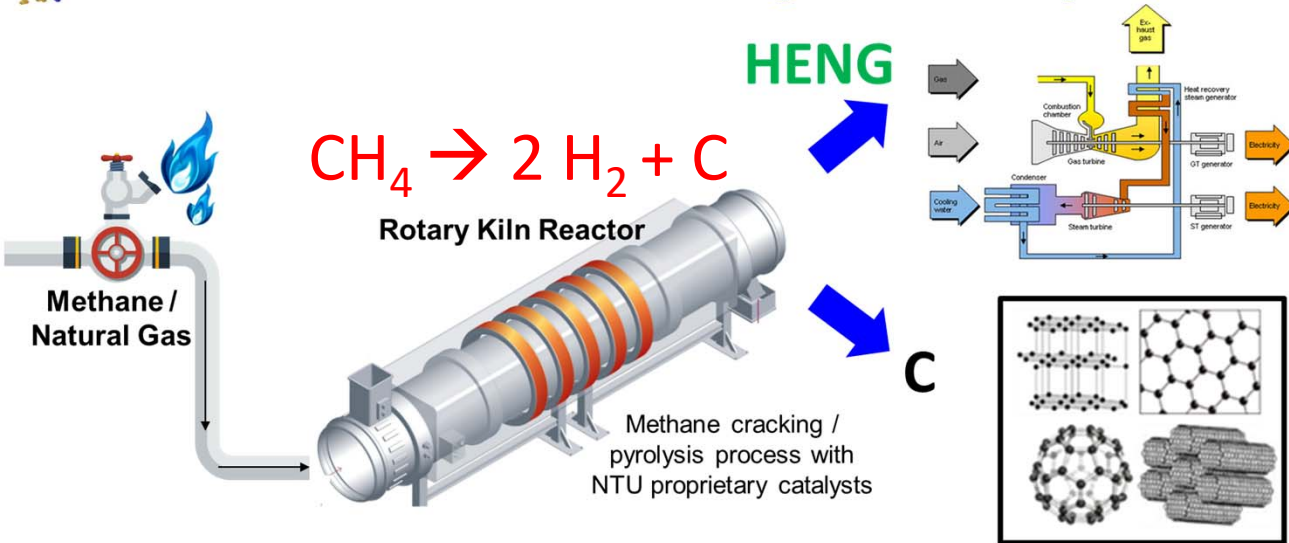
Research projects:

1. Methane Cracking for Turquoise Hydrogen Generation
2. Seawater Electrolysis
3. $\text{CO}_2 + \text{H}_2\text{O}$ (flue gas) converted to syngas
4. Ammonia-fed Solid Oxide Fuel Cell (SOFC)



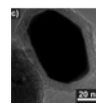
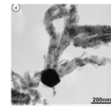
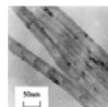
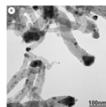


Methane Cracking for Turquoise Hydrogen Generation

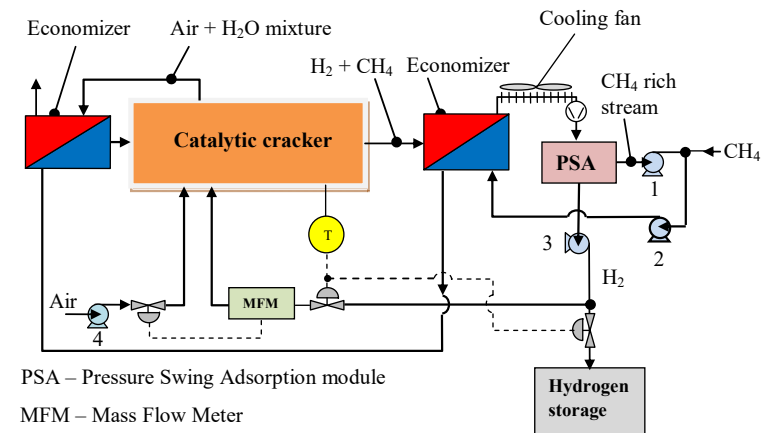


Catalyst beads in reactor

	Ni	Co	Fe
Catalytic activity	Best	Moderate	Relatively poor
Temperature range	500-650°C	500-900°C	700-900°C
Oxide form	NiO	CoO & Co ₂ O ₃	Fe ₂ O ₃

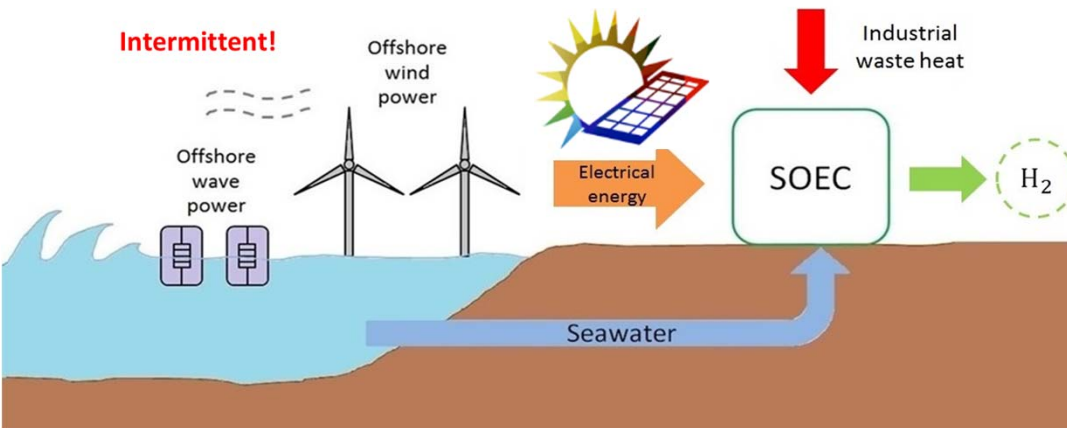


- A technology can potentially produce hydrogen at a cost like SMR but without CO₂ emissions.
- By-product graphite powder can be valorized to offset the cost of H₂.
- Compared to other reactors, rotary kiln reactor can potentially be scaled up for mass production.
- The challenging issues for research includes (1) catalyst fouling, (2) catalyst regeneration, (3) continuous operation.

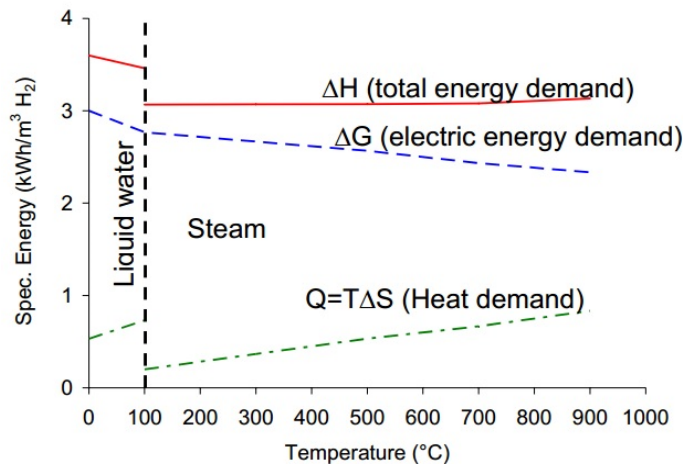
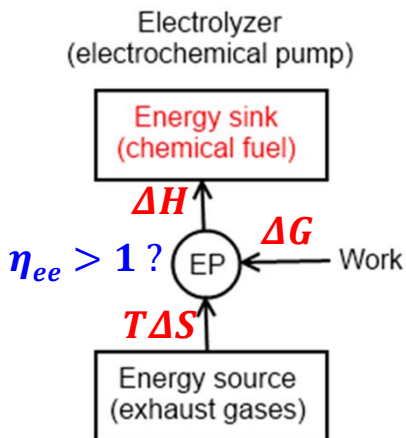




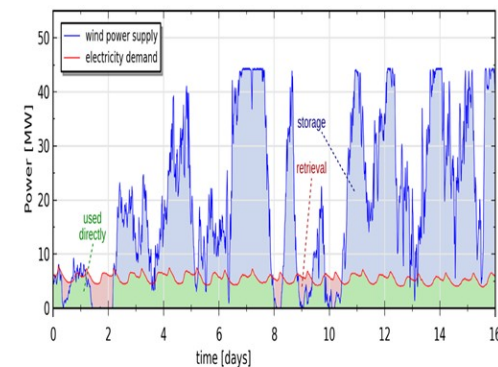
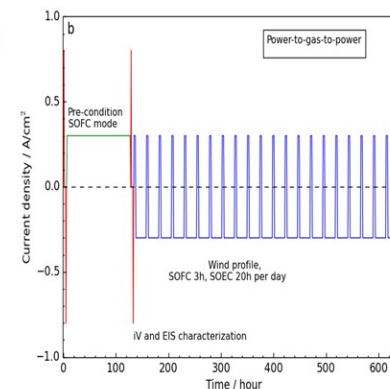
Seawater Electrolysis



Overall concept for seawater electrolysis using an SOEC to store offshore renewable energy in the form of H₂



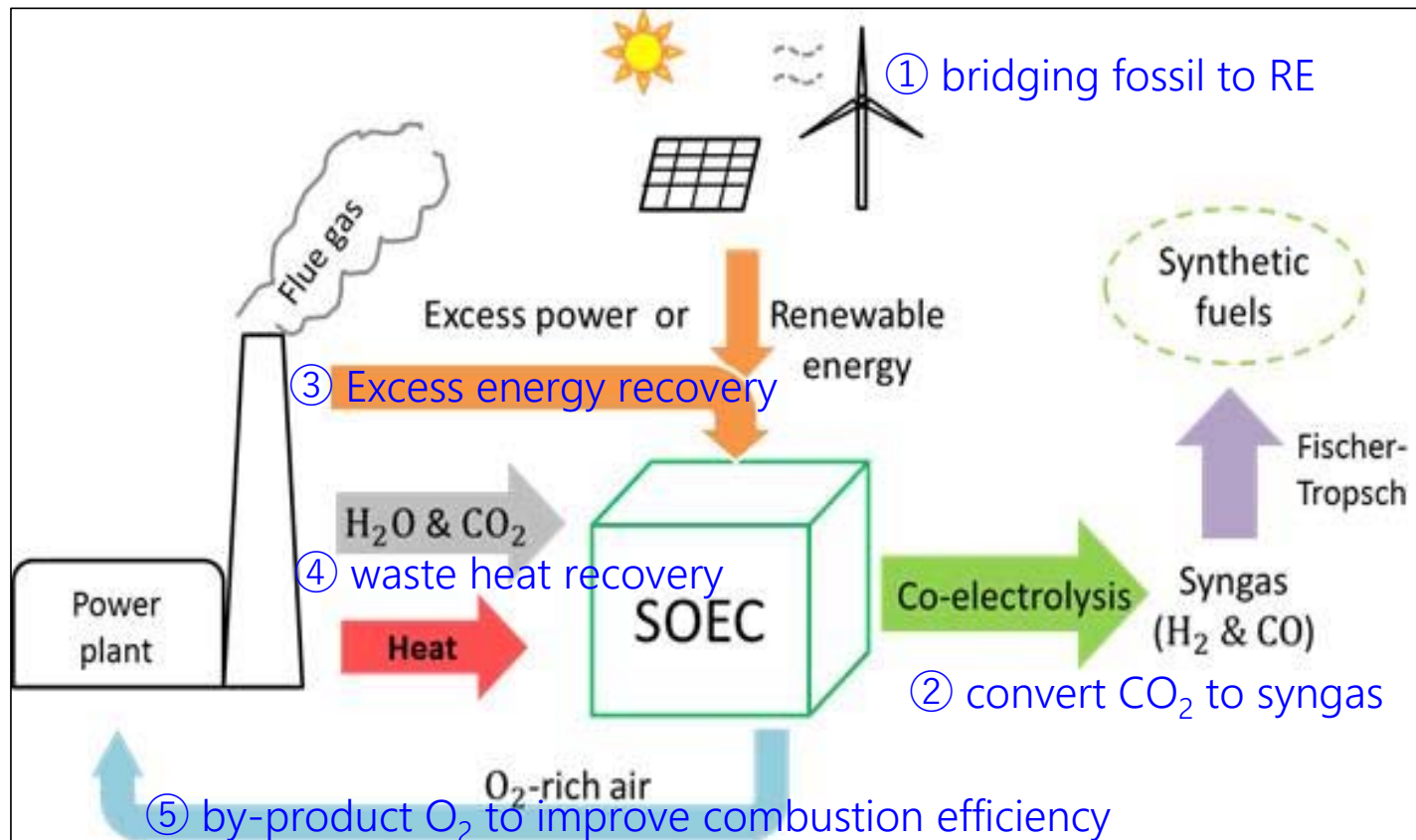
- Solid oxide cell is the ONLY technology that can operate in both fuel cell & electrolyzer modes.
- The electricity to H₂ heating value can be more than 100% if waste heat is available.
- SOCTESQA (2014-2017) is a consortium led by DLR. Testing conducted under reversible SOFC/SOEC mode following a wind profile recorded in a Danish island (3 h SOFC, 20 h SOEC per day).
- Challenges: (1) oxidation of fuel electrode, (2) long term effect of impurities (Na⁺, Ca⁺⁺, etc.) on the fuel electrode.





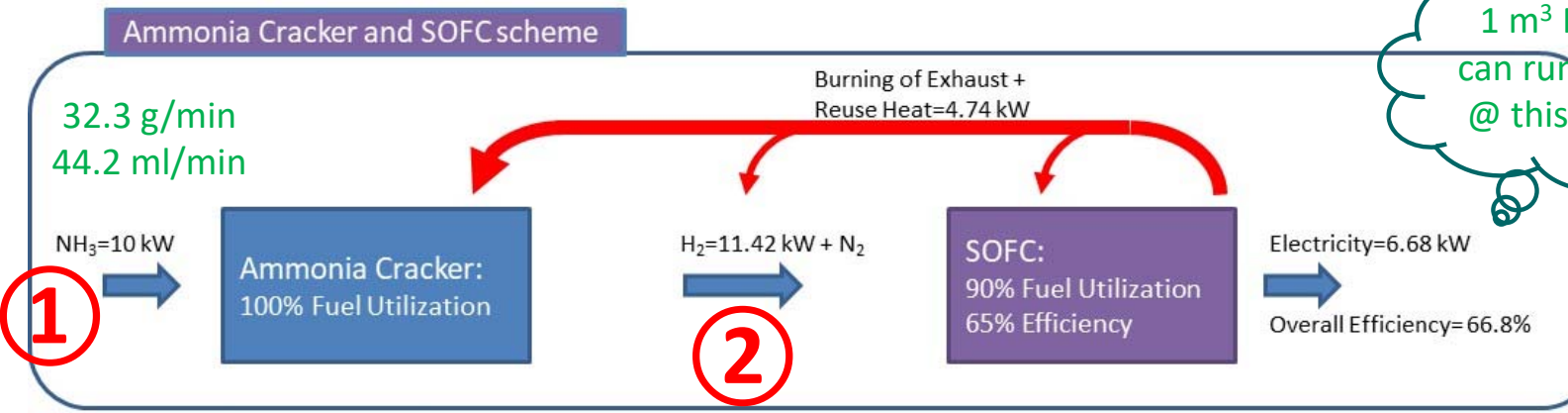
CO₂ Capture from Oxy-Power Plant using SOEC

Flue gas from oxy-combustion plants (metal smelting, glass industries, etc.) contain rich CO₂ and H₂O.





Ammonia-fed Solid Oxide Fuel Cell (SOFC)

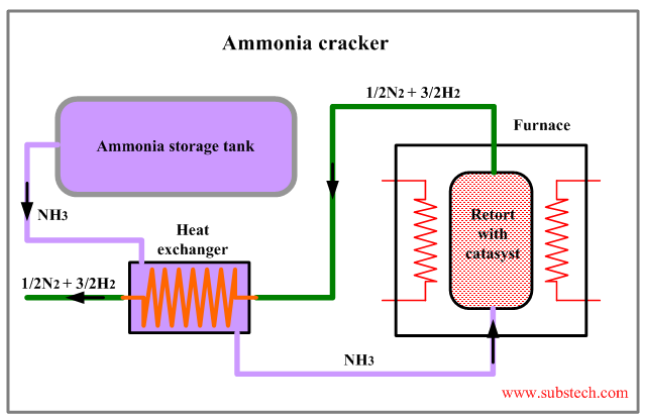


3

1 m³ NH₃ (l)
can run 377 hr
@ this power

1. Using ammonia as H₂ carrier
2. H₂ is easily regenerated with increased HV of 14.2%
3. An energy storage with high volumetric energy density
4. NO_x issues can be solved.

Ammonia Cracker



*NO_x can be reduced chemically by either H₂ or NH₃

4

	dH [kJ/mol reaction]	dH [kJ/mol NH3]
2 NH ₃ + 3 N ₂ O = 4 N ₂ + 3 H ₂ O	dH=-1010	-506
4 NH ₃ + 6 NO = 5 N ₂ + 6 H ₂ O	dH=-2071	-517
8 NH ₃ + 6 NO ₂ = 7 N ₂ + 12 H ₂ O	dH=-3259	-407
H ₂ + N ₂ O = N ₂ + H ₂ O	dH=-368.4	
5 H ₂ + 2 NO = 2 NH ₃ + 2 H ₂ O	dH=-844	
3.5 H ₂ + NO ₂ = NH ₃ + 2 H ₂ O	dH=-651	



Thank you