

- ### SECTION-D
- Note:** Long answer type questions. Attempt any two questions out of three questions. (2x10=20)
- Q.36 Explain the principle and use of first and second law of thermodynamics.
- Q.37 Explain with a neat sketch, working, uses of the shell and tube heat exchanger?
- Q.38 Explain principle, working and use of filter press with neat sketch.

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Roll No.

Time : 3Hrs. M.M. : 100

Note: Multiple choice questions. All questions are compulsory (10x1=10)

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- a) Jaw crusher b) Gyratory crusher
c) Fine crusher d) Tramp crusher
- Q.5 The S.I unit of temperature is _____.
- a) Centigrade b) Celsius
c) Fahrenheit d) Kelvin
- Q.6 What is the unit of diffusion coefficient?
- a) m^2 b) s
c) $m^2 s$ d) m^2/s
- Q.7 Humidification is a
- a) Mass transfer operation
b) Heat transfer operation
c) Simultaneous heat and mass transfer
d) Neither mass and heat transfer operation
- Q.8 Joule was the first scientist to prove that heat is a type of energy, which work on the fundamental law of thermodynamics.
- a) False b) True
- Q.9 Which of the following laws is applicable for the behavior of perfect gas _____
- a) Boyle's law b) Charles's law
c) Gas-Lussac law d) All of the above
- Q.10 Which is not a Mechanical operation
- a) Agitation b) Filtration
c) Humidification d) Size enlargement

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SECTION-B

Note: Objective type questions. All questions are compulsory. (10x1=10)

- Q.11 Write conduction.
Q.12 Define insulator.
Q.13 Define isolated system
Q.14 Define work
Q.15 Define mesh number.
Q.16 Define thermal conductivity.
Q.17 Define crushing.
Q.18 Define isothermal process.
Q.19 Define heat transfer.
Q.20 Define Entropy.

SECTION-C

Note: Short answer type questions. Attempt any twelve questions out of fifteen questions. (12x5=60)

- Q.21 Explain Fourier's law of heat transfer.
Q.22 Describe the first law of thermodynamics.
Q.23 Define the principle of ball mill and give its use.
Q.24 State bond's law and Kick's law of crushing.
Q.25 Write the difference between crushing and grinding.
Q.26 Explain homogeneous and heterogeneous system with example.
Q.27 Explain Isothermal process of thermodynamics.

(3) 182241/122241/032241
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