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4th Sem / Branch : MECH. ENGG.
Subject:- THERMODYNAMICS-I

Time : 3Hrs.

M.M. : 100

SECTION-A

Note: Multiple choice questions. All questions are compulsory (10x1=10)

- Q.1 Bomb Calorimeter is an example of (CO-1)
a) Closed system
b) Isolated system
c) Open system
d) Homogeneous system
- Q.2 The value of characteristics gas constant (R) for atmospheric air is (CO-2)
a) 287 J/Kg K b) 0.287 J/Kg K
c) 2.87J/Kg K d) 28.7 J/Kg K
- Q.3 Constant pressure process is also known as (CO-3)
a) Isochoric process b) Isobaric process
c) Isothermal process d) Throttling process
- Q.4 First law of thermodynamics deals with (CO-4)
a) Conservation of energy
b) Conservation of force
c) Conservation of mass
d) Conservation of momentum

- Q.5 Kelvin planck's statement deals with (CO-4)
a) Conservation of heat
b) Conservation of momentum
c) Conservation of mass
d) Conservation of heat into work
- Q.6 The point at which all the three phases-solid, liquid and vapour co-exist in equilibrium is called (CO-5)
a) Critical Point
b) Point of Contraflexure
c) Triple point
d) Ideal point
- Q.7 Mollier diagram is a plot of (CO-6)
a) Temperature and entropy
b) Temperature and enthalpy
c) Entropy and pressure
d) Enthalpy and entropy
- Q.8 Which of the following is a boiler mounting (CO-7)
a) Pressure gauge b) Economiser
c) Superheater d) Air preheater
- Q.9 Otto cycle is also known as (CO-8)
a) Constant pressure cycle
b) Constant volume cycle
c) Constant entropy cycle
d) Constant temperature cycle
- Q.10 For same compression ratio, the efficiency of otto cycle is ____ Diesel cycle (CO-8)
a) Greater than b) Less than
c) Equal to d) None of the above

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SECTION-B

Note: Objective type questions. All questions are compulsory.
 (10x1=10)

- Q.11 In a closed system, there is no transfer of _____ across the system boundary. (CO-1)
 Q.12 Name any two intrinsic properties. (CO-1)
 Q.13 In the Throttling process Workdone= _____. (CO_3)
 Q.14 A machine which violates the first law thermodynamics is called _____. (CO-4)
 Q.15 Define Entropy (CO-6)
 Q.16 For dry steam, dryness fraction= _____ (CO-6)
 Q.17 Give any two examples of fire tube boilers. (CO-7)
 Q.18 A thermodynamic cycle using air as the working substance is known as _____. (CO-8)
 Q.19 Write the function of intercooler. (CO-9)
 Q.20 The gas which obeys the law $Pv=RT$ at all temperature and pressure is known as _____. (CO-5)

SECTION-C

Note: Short answer type questions. Attempt any twelve questions out of fifteen questions. (12x5=60)

- Q.21 Define thermodynamic property. what are its different types. Explain ? (CO-1)
 Q.22 Find the mass of a gas occupying 5.5m^3 at 7 bar absolute pressure and 200°C . Take gas constant $R=287 \text{ J/Kg K}$. (CO-2)
 Q.23 Explain Throttling process (CO-3)
 Q.24 Explain any two applications of general steady flow equation (CO-4)
 Q.25 Explain Vanderwall's Equation of state (CO-5)

- Q.26 If the dryness fraction of a given sample of steam is 0.85, find the mass of water particles contained in 2Kg of this steam (CO-6)
 Q.27 Write any five difference between Fire tube and water Tube boilers (CO-7)
 Q.28 Find the compression ratio of an otto cycle , if the efficiency of otto cycle is 55% and $\gamma=1.5$ (CO-8)
 Q.29 Write any five difference between Reciprocating and Rotary air compressor (CO-9)
 Q.30 Define specific heat Give the relation between C_p and C_v (CO-2)
 Q.31 A carnot engine working between 620 K and 300 K gives 200 KJ of work. Calculate the thermal efficiency of engine and heat added during the process (CO-4)
 Q.32 Explain P-V-T surface of an Ideal gas (CO-5)
 Q.33 Explain the working of separating calorimeter (CO-6)
 Q.34 What are the objectives of Boiler accessories. Name any four boiler accessories (CO-7)
 Q.35 Draw P-V and T-S diagrams of a Diesel cycle (CO-8)

SECTION-D

Note: Long answer type questions. Attempt any two questions out of three questions. (2x10=20)

- Q.36 Explain the construction and working of Babcock Wilcox Boiler with a neat sketch (CO-7)
 Q.37 Explain the construction and working of single stage Reciprocating Compressor with a neat sketch (CO-9)
 Q.38 Explain Kelvin-Planck's and Clausius statements. (CO-4)