

- Q.27 Define Fick's law of diffusion also write its mathematical relation
- Q.28 Describe humidity chart
- Q.29 Write the working of rotary dryer
- Q.30 Discuss the working of spray ponds
- Q.31 Explain gas desorption mass transfer operation with the help of suitable example.
- Q.32 What are the differences in batch dryer and continuous dryer?
- Q.33 Explain Raoult's law and Henry's law
- Q.34 Differentiate between humidification and dehumidification with example.
- Q.35 Discuss any one of the following
- Overall mass transfer coefficient
 - Eddy diffusion

SECTION-D

- Note:** Long answer type questions. Attempt any two questions out of three questions. (2x10=20)
- Q.36 What is drying? Explain the construction and working of tray dryer with the help of neat diagram
- Q.37 Write about difference type tower packing. Explain the different problems encountered in packed towers
- Q.38 What is the basic concept of cooling in cooling towers? Describe the different cooling tower arrangement.

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Roll No.

4th Sem / Chem, P & P Subject:- Mass Transfer I

Time : 3Hrs.

M.M. : 100

SECTION-A

Note: Multiple choice questions. All questions are compulsory (10x1=10)

- Q.1 Diffusion is a process of
- Movement of particles from high concentration to low concentration
 - Movement of particles through semi permeable membrane
 - Accumulation of particles on solid surface
 - None of these
- Q.2 The diffusivity has the same dimensions as
- Density
 - concentration
 - kinematic viscosity
 - Dynamic viscosity
- Q.3 According to film theory mass transfer coefficient is directly proportional to
- $D_{AB}^{2.5}$
 - D_{AB}^2
 - $D_{AB}^{0.5}$
 - D_{AB}
- Q.4 Which of the following is very important to separate a mixture of two volatile liquids by distillation
- Solubility
 - Relative volatility
 - Density difference
 - none of these
- Q.5 The mutual solubility of two liquids
- Decrease with increase in temperature

- b) Increase with increase in temperature
 - c) remains constant
 - d) None of the above
- Q.6 Bound moisture in a solids is that liquid which exert an equilibrium vapour pressure
- a) equal to that of pure liquid at a given temperature
 - b) greater than that of pure liquid at a given temperature
 - c) less than that of pure liquid at a given temperature
 - d) equal to or less than that of pure liquid at a given temperature
- Q.7 The rate of drying during constant-rate period
- a) Increase with increase in air temperature
 - b) Decrease with increase in air temperature
 - c) Unaffected by increase in air temperature
 - d) increase and then decrease with increase in air temperature
- Q.8 Rotary driers are
- a) used to make milk powder
 - b) used to make detergent
 - c) suitable for handling free flowing granular materials
 - d) Suitable for handling sticky materials
- Q.9 At a constant temperature, the gas solubility increases with _____ In pressure.
- a) highly decrease b) Increase
 - c) no change d) slightly decrease

- Q.10 The temperature at which a vapour gas mixture becomes saturated when cooled at constant total pressure out of contact with a liquid is called
- a) Bubble point b) wet bulb temperature
 - c) dry bulb temperature d) dew point

SECTION-B

Note: Objective type questions. All questions are compulsory. (10x1=10)

- Q.11 Define mass transfer.
- Q.12 Write an example of gas absorption.
- Q.13 Expand HETP
- Q.14 Write any one use of humidity chart
- Q.15 Define dry bulb temperature
- Q.16 Write the full form of NTU
- Q.17 Name any two mass transfer theories
- Q.18 What do you understand by equilibrium
- Q.19 Write names of any two packing used in packed towers
- Q.20 Draw any one cooling tower arrangement

SECTION-C

Note: Short answer type questions. Attempt any twelve questions out of fifteen questions. (12x5=60)

- Q.21 Write any two mass transfer operation with one example each
- Q.22 Explain the role of diffusion in mass transfer
- Q.23 Define loading and channeling.
- Q.24 Explain rate of drying curve with its neat diagram
- Q.25 Write about properties of tower packing
- Q.26 Discuss about the factors to be considered before selecting solvent for absorption