

---

**1<sup>st</sup> Semester**  
**Branch: Common**  
**Subject Name: Applied Chemistry**

## **Time Allowed : 3 Hrs.**

**MM: 60**

## **Section -A**

**Note: Multiple Choice questions. All questions are compulsory.**

$$6 \times 1 = 6$$



## **Section-B**

**Note: Objective/Completion type questions. All questions are compulsory.**  $6 \times 1 = 6$

- Q.7 Define modern periodic law.

Q.8 The earthy impurities present in the ore are known as-----.

Q.9 Define normality.

Q.10 The full form of LPG is -----.

Q.11 SI unit of viscosity is -----.

Q.12 Define carbon based nano materials.

## **Section –C**

**Note: Short answer type Questions. Attempt any eight questions out of ten questions.**  $8 \times 4 = 32$

- Q.13 Write a short note on metallic bond.
  - Q.14 Define alloys. What are the purpose of making alloys?
  - Q.15 Define pH. What are industrial applications of pH?
  - Q.16 Define scale & sludge. Write 2 disadvantages of scale & sludge.
  - Q.17 What are the characteristics of an ideal fuel?( 4 only)
  - Q.18 Briefly explain the properties and uses of CNG.
  - Q.19 Write a short note on cementation.
  - Q.20 Write 4 applications of nano materials.
  - Q.21 State 4 points of difference between orbit and orbital.
  - Q.22 Define the following terms---- Ductility, Elasticity

## **Section-D**

**Note: Long answer questions. Attempt any two questions out of three questions.**  
**2x8=16**

- Q.23 Define permanent hardness of water. Discuss the ion exchange method used for the removal of permanent hardness of water.

Q.24 Define the following terms--- oiliness, viscosity, viscosity index, flash point

Q.25 Define polymer. State difference between addition and condensation polymers. (6 points)