

- Q.29 Write about mechanism of crystallization/crystal/formation.
- Q.30 Write about Rectification.
- Q.31 Write about the McCabe Theile Method.
- Q.32 Discuss the leaching through stationary solid beds.
- Q.33 With diagram, discuss about plate column.
- Q.34 Discuss ultra-filtration with diagram.
- Q.35 Difference between Leaching and Extraction.

SECTION-D

Note: Long answer type questions. Attempt any two questions out of three questions. (2x10=20)

- Q.36 Describe the agitated tower extractor with complete details and diagram.
- Q.37 Explain the construction and working of Fractionation column in details.
- Q.38 Write short note on any three:-
1. Effect of temp. on adsorption
 2. Plate column
 3. Flash or Equilibrium Distillation
 4. Liquid extraction

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**5th Sem / Branch : Chemical Engg. (P&P)
Sub.: Mass Transfer-II**

Time : 3Hrs.

M.M. : 100

SECTION-A

Note: Multiple choice questions. All questions are compulsory (10x1=10)

- Q.1 We calculate many type of reflux ratios like
- a) Total reflux ratio
 - b) Minimum reflux ratio
 - c) Optimum reflux ratio
 - d) All of above
- Q.2 Oil extracted from soil seeds is
- a) Extraction b) Distillation
 - c) Drying d) None of above
- Q.3 The higher difference in pressure inside the tray tower cause
- a) Flooding b) Loading
 - c) Weeping d) Dumping
- Q.4 Solvent lean phase are known as
- a) Extract b) Residue
 - c) Raffinate d) None of above

- Q.5 The membrane separation processes are
a) RO b) Dialysis
c) Ultrafiltration d) All of above
- Q.6 Leaching generally used for ore removal purposes
a) True b) False
- Q.7 According to Raoult's law, for a pure component solution the partial pressure is equals to
a) Total pressure
b) Vapour pressure
c) Atmospheric pressure
d) Mole fraction of respective phase
- Q.8 The ratio of number of moles of species A to the total number of moles of the mixture is known as
a) Mole fraction b) Mass fraction
c) Partial pressure d) Mass density
- Q.9 What is the unit of diffusion coefficient?
a) m^2 b) s
c) m^2s d) m^2/s
- Q.10 According to Hendry's law,
a) $Y = (p/P)$ b) $Y = (p^*P)$
c) $p = (Y/P)$ d) $p = (Y^*P)$

SECTION-B

Note: Objective type questions. All questions are compulsory. $(10 \times 1 = 10)$

- Q.11 What is azeoptrope?

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- Q.12 Define mass transfer?
- Q.13 Write one equipment used for crystallization.
- Q.14 Write formula for total reflux?
- Q.15 What is separation?
- Q.16 Write formula of Relative Volatility.
- Q.17 Name any one distillation plant in India.
- Q.18 What is flooding?
- Q.19 Write one use of distillation?
- Q.20 What is Adsorption?

SECTION-C

Note: Short answer type questions. Attempt any twelve questions out of fifteen questions. $(12 \times 5 = 60)$

- Q.21 Discuss about packed extraction towers?
- Q.22 Classify distillation.
- Q.23 Difference between Leaching and extraction.
- Q.24 Describe about the construction and working of agitated tank crystallizer.
- Q.25 Write about the Membrane & Types of Membrane.
- Q.26 Explain the Concept of flooding, Weeping, Entrainment and loading in distillation columns.
- Q.27 Explain draft tube crystallizer.
- Q.28 Write about Concept of Adsorption operation & types of adsorption.

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