

The Radio Frequency Fragment Separator: A Time-of-Flight Filter for Fast Fragmentation Beams



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Introduction

Outline of my presentation:

- Brief overview of rare isotope beam production at NSCL.
- Function of the Radio Frequency Fragment Separator (RFFS).
- Separation of neutron-deficient isotopes based on two recent examples.



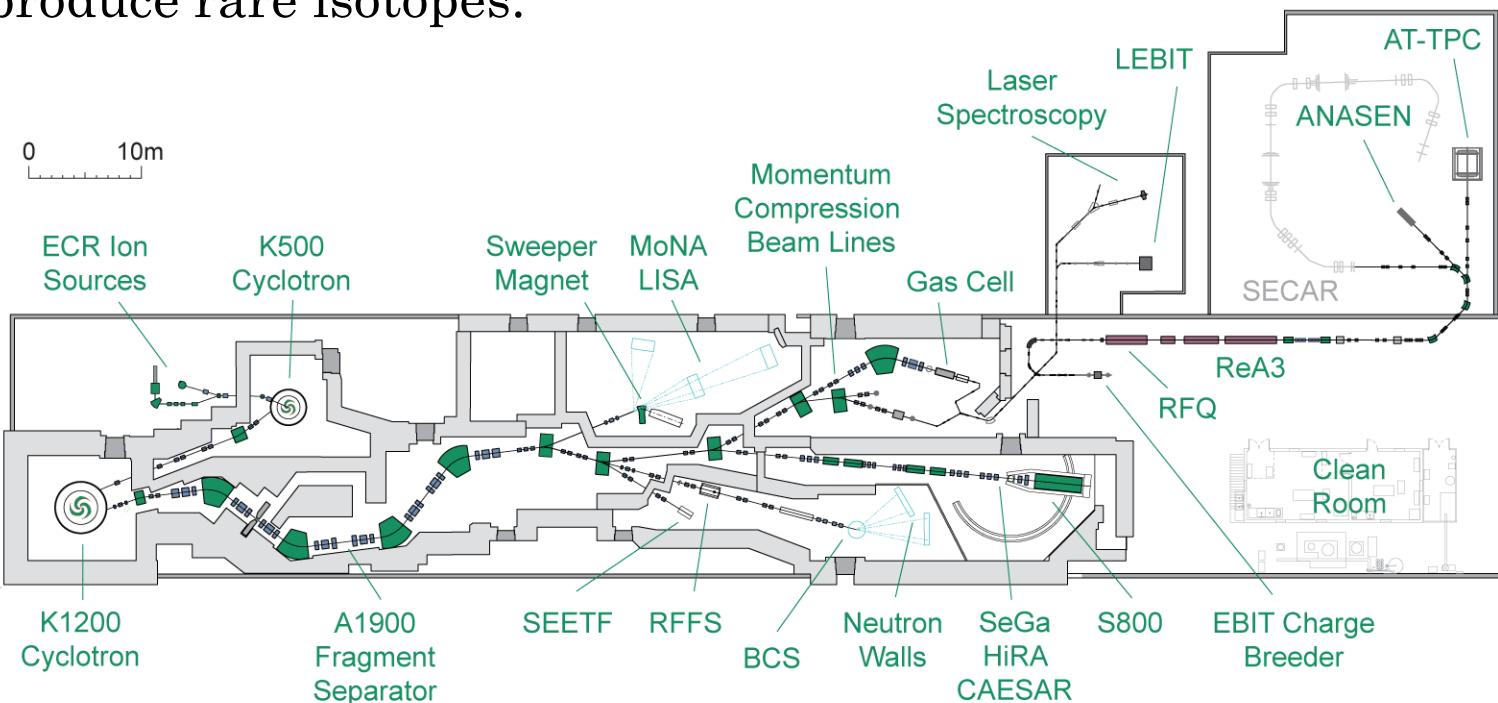
The National Superconducting Cyclotron Laboratory

- Fast fragmentation is the method of choice to produce a large range of rare isotope beams.
- Fragment separators can select the isotope of interest by a combination of magnetic rigidity selection and slowing down in matter (i.e. achromatic wedge).
- The National Superconducting Cyclotron Laboratory (NSCL) at Michigan State University is built around this concept.



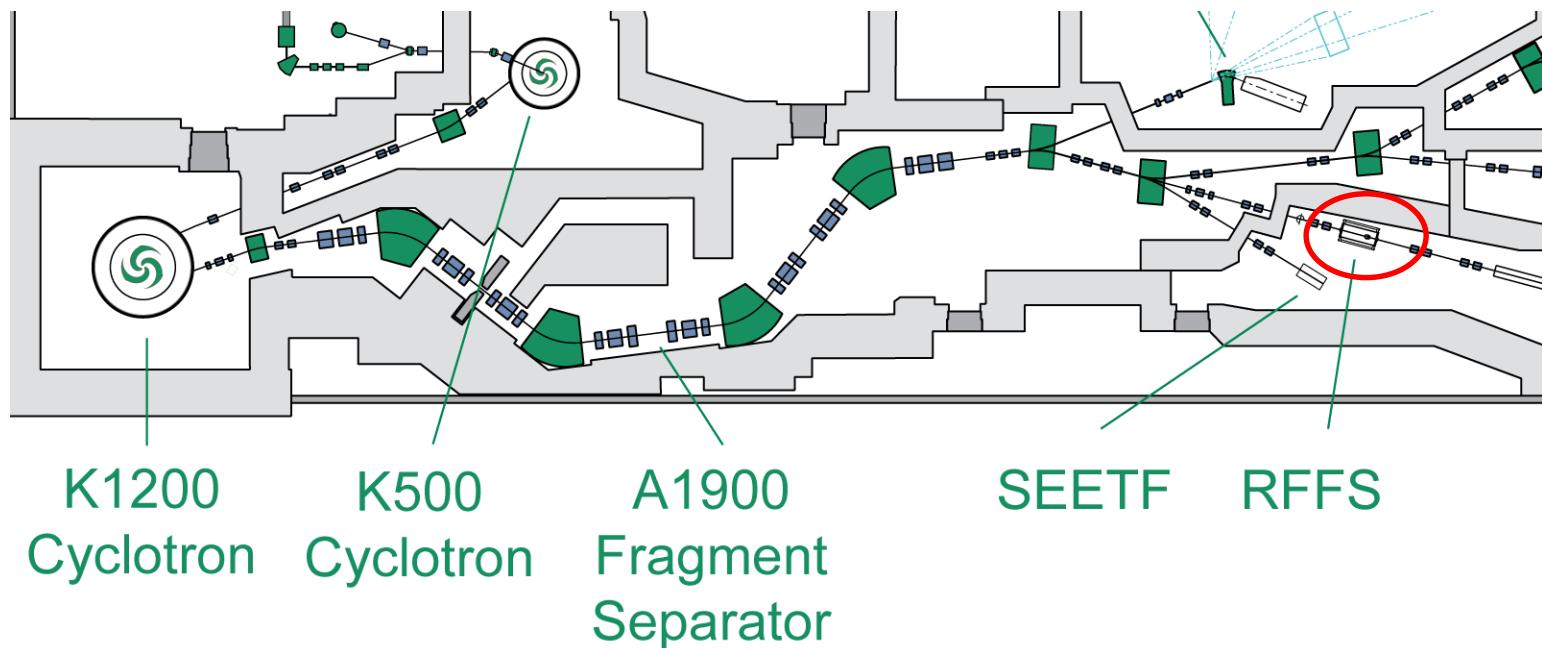
NSCL rare isotope beam facility

- The Coupled Cyclotron Facility utilizes two superconducting cyclotrons to accelerate heavy ion beams to typically 100 to 170 MeV/u.
- These primary beams then undergo a fragmentation reaction to produce rare isotopes.

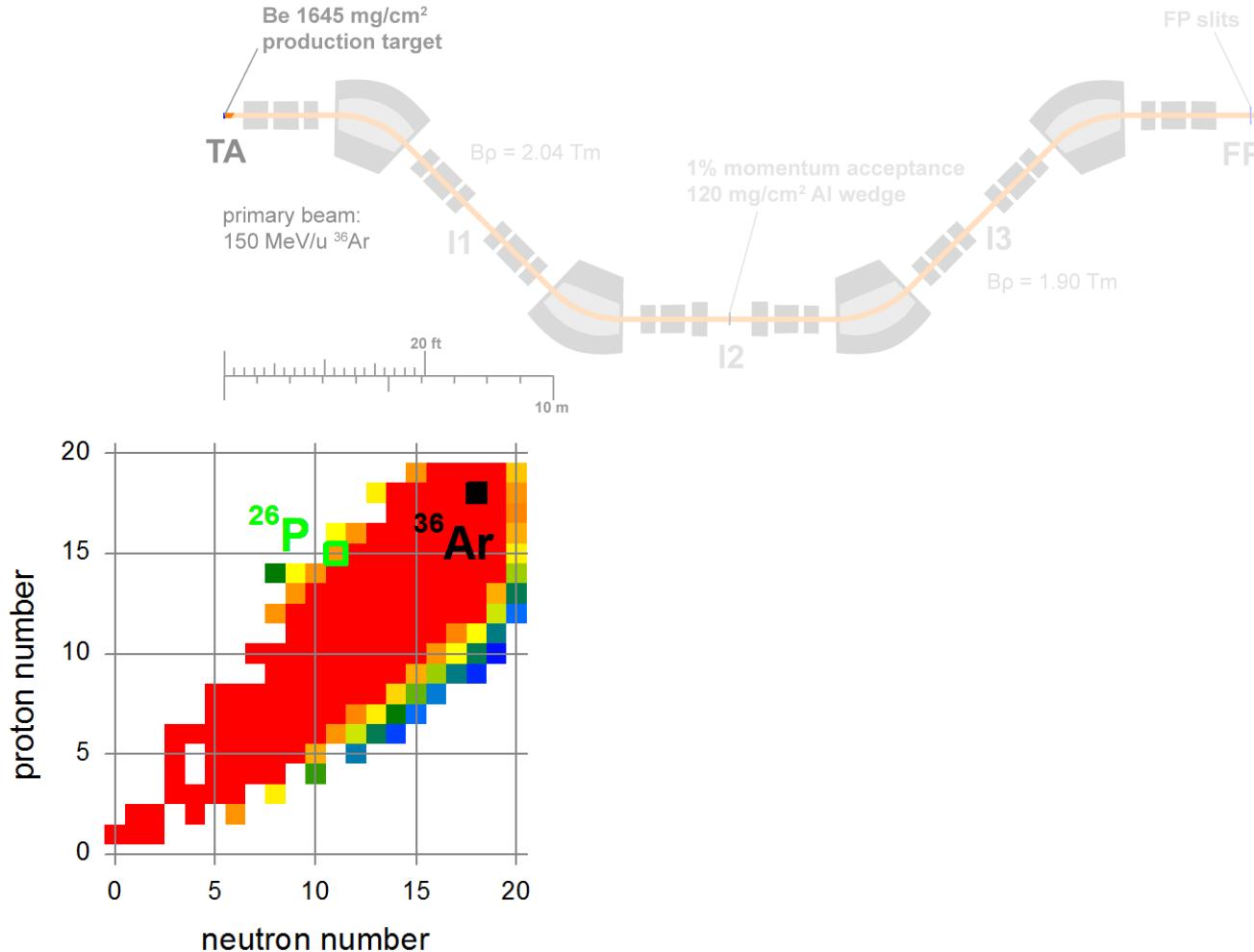


NSCL rare isotope beam facility

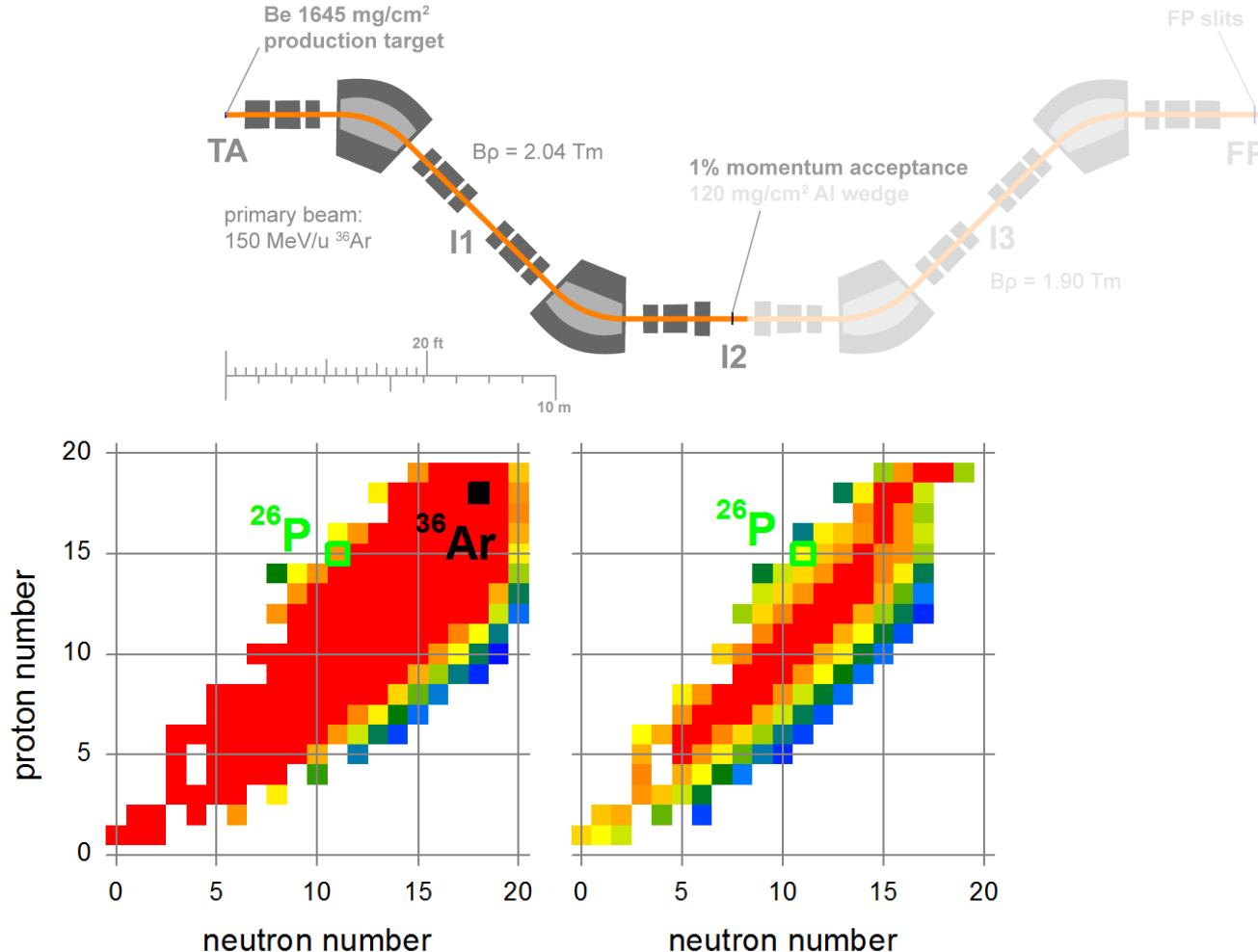
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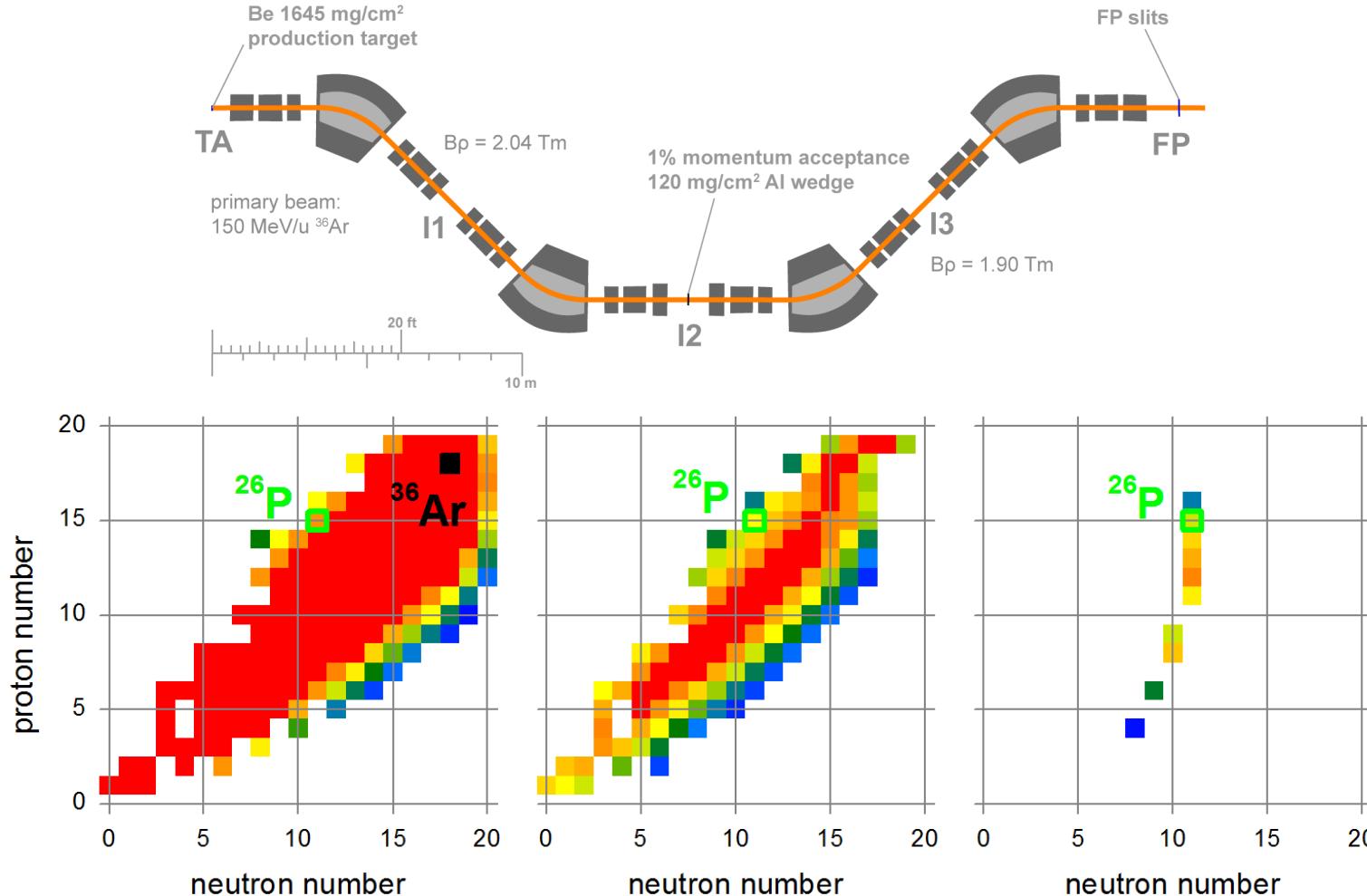
Production of neutron-deficient ^{26}P



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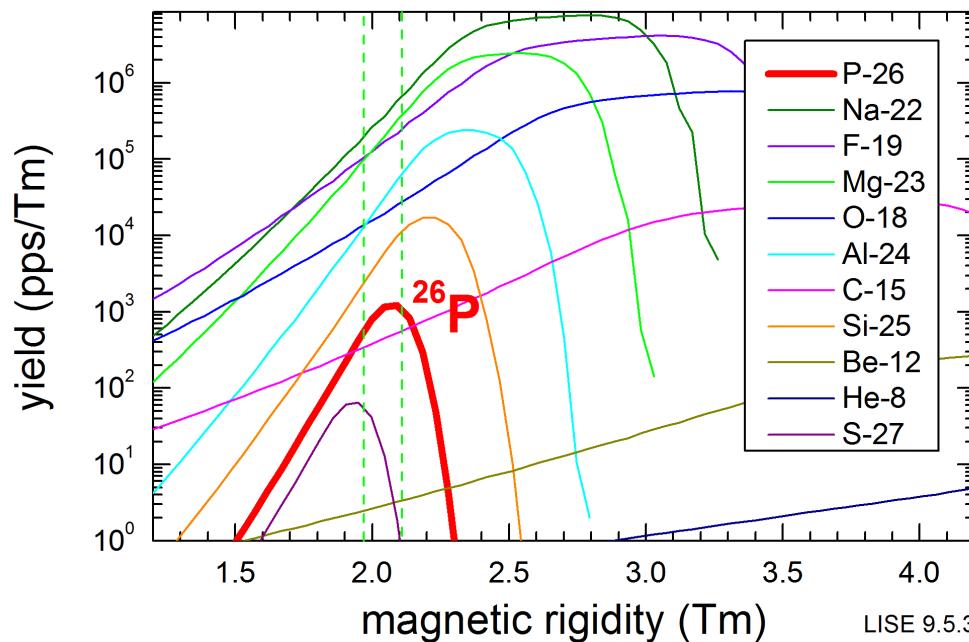
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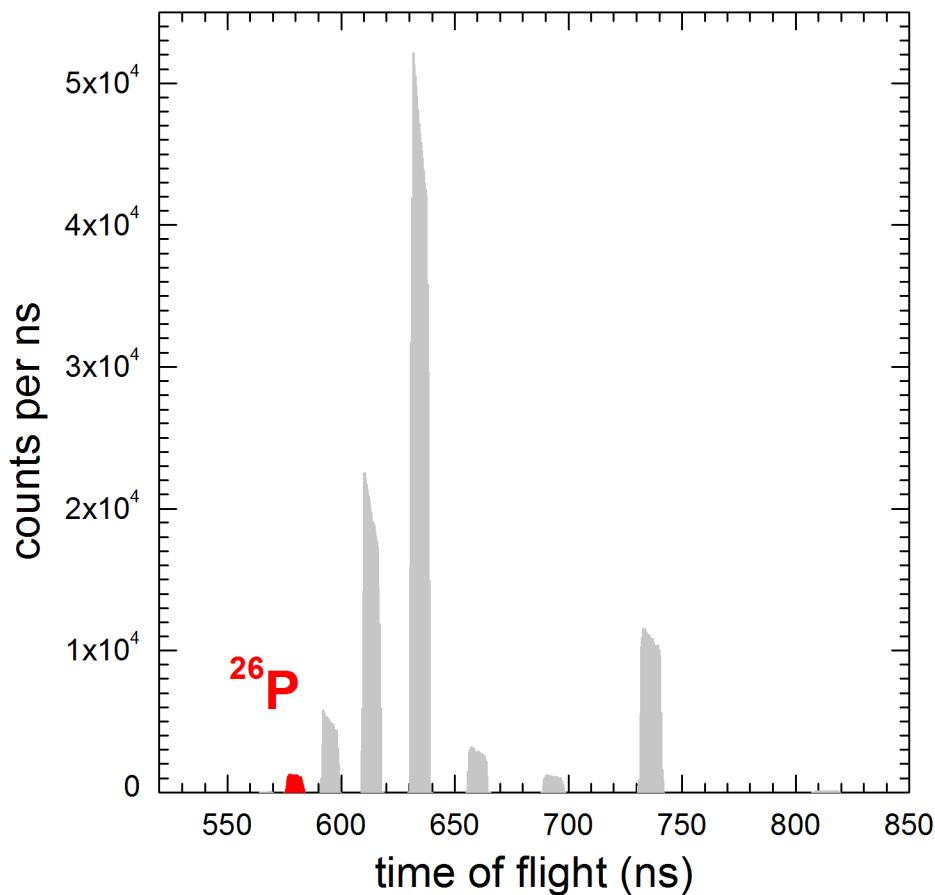
Challenges of neutron-deficient isotopes

Separation of neutron-deficient rare isotopes is challenging:

- At primary beam energies between 50 MeV/u and 200 MeV/u, fragments exhibit a low momentum tail.
- Neutron deficient means lower mass-to-charge ratio or lower rigidity ($B\rho = \frac{mv}{q}$) than isotopes with much higher yields.



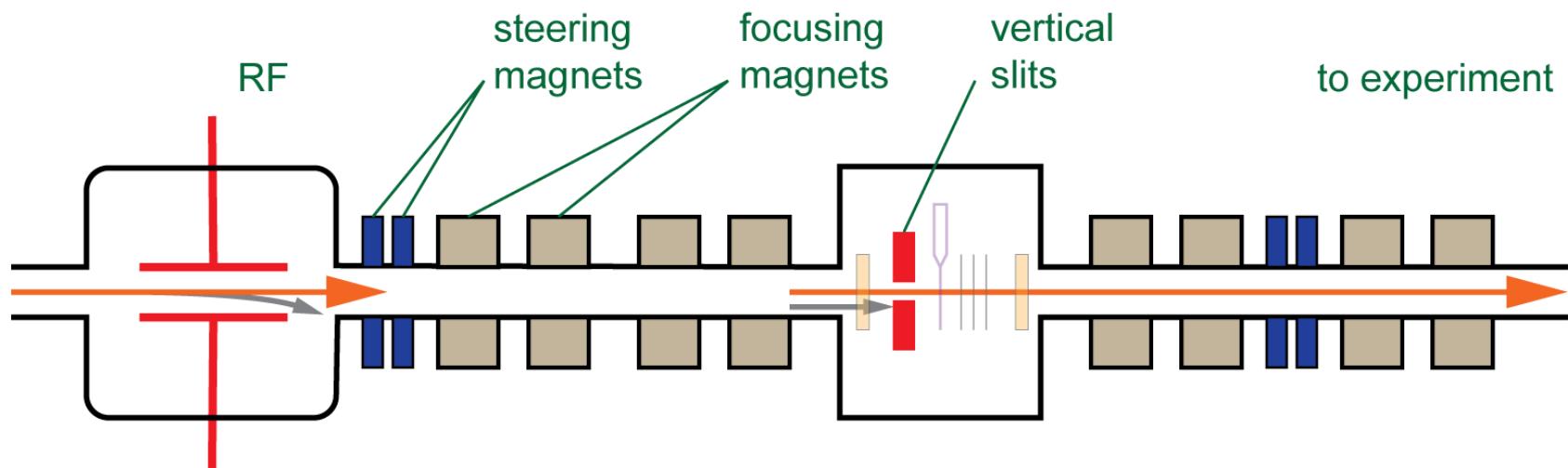
Filtering by time-of-flight



- Isotopes have different velocities and are separated by time of flight.
- A *velocity filter* could separate out the isotope of interest.
- The Radio Frequency Fragment Separator selects fragments based on their phase with respect to the cyclotron RF.

The RFFS concept

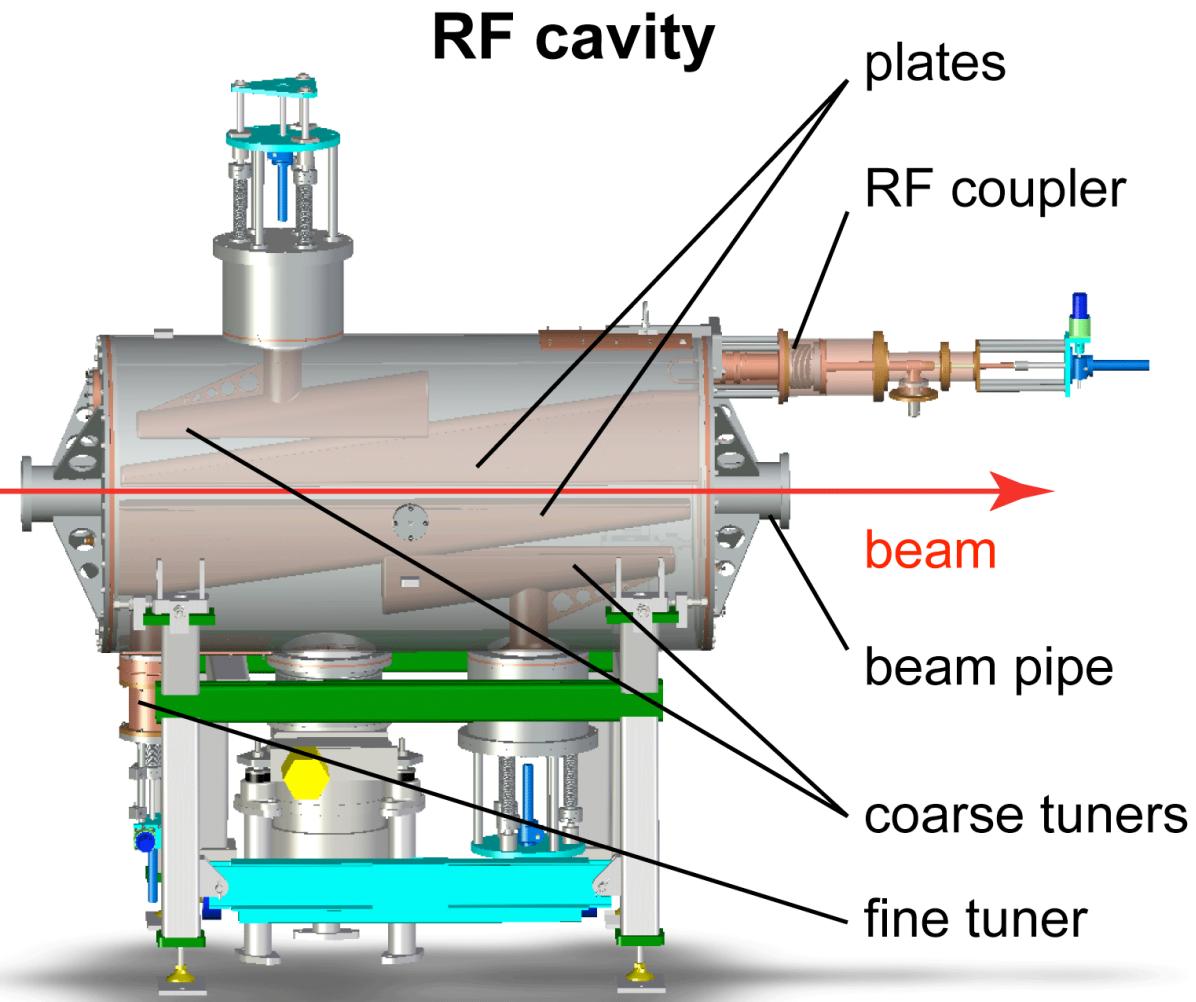
- A radio frequency separator for rare isotope beams was first realized at the RIPS facility in RIKEN (Japan).
K. Yamada, T. Motobayashi, I. Tanihata, Nucl. Phys. A 746 (2004) 156c
- The RFFS is a combination of an RF driven electric deflector and a set of slits.



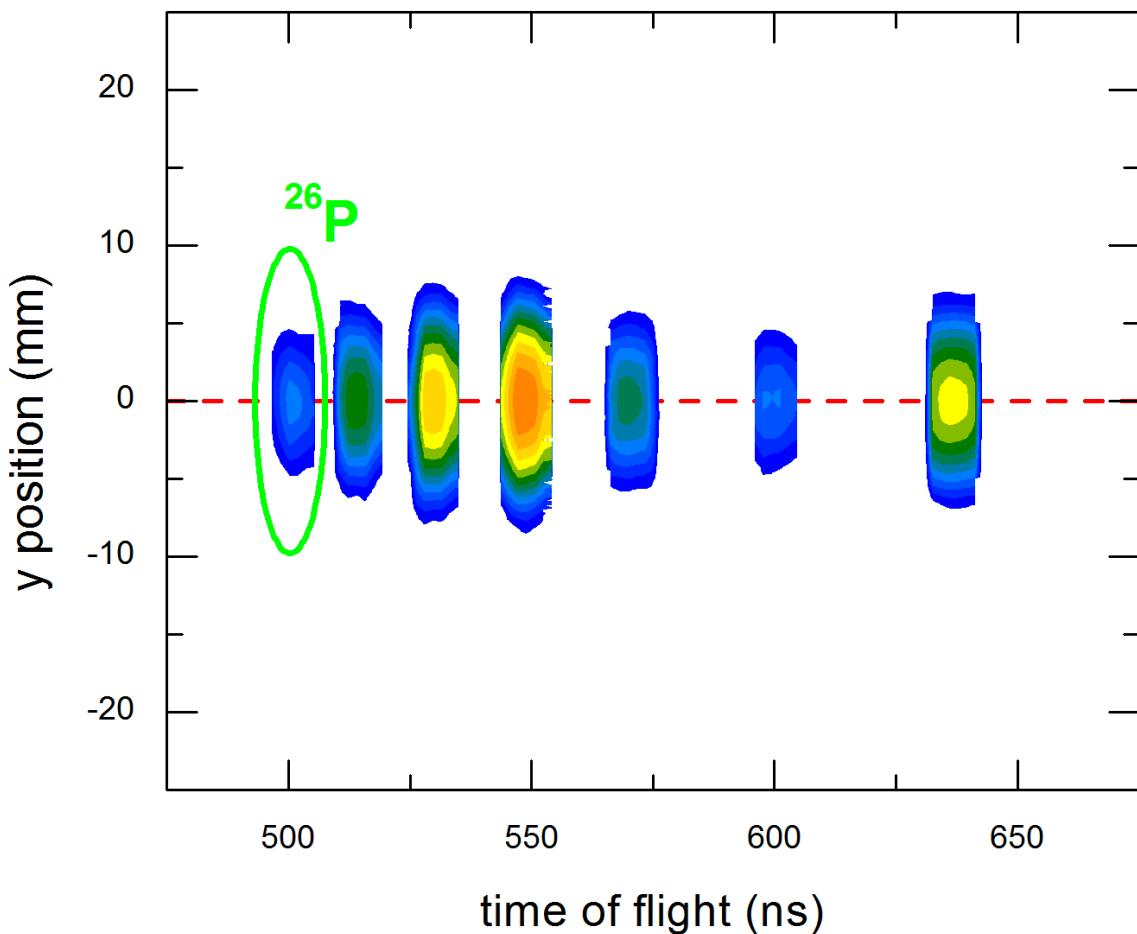
The RFFS cavity

RF cavity design parameters

electrode length	1.5 m
electrode gap	5 cm
RF voltage	100 kV
frequency	21–28 MHz
phase length	~140°

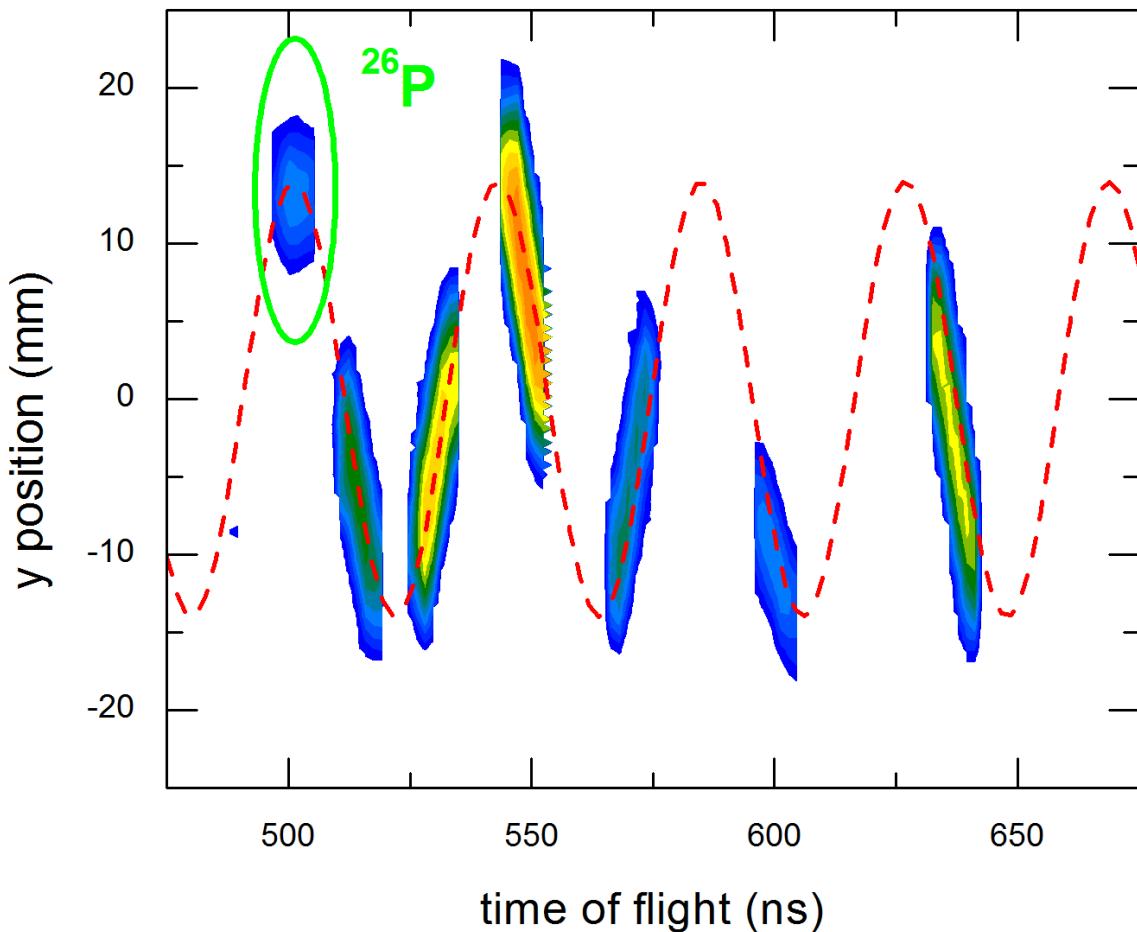


How to separate ^{26}P



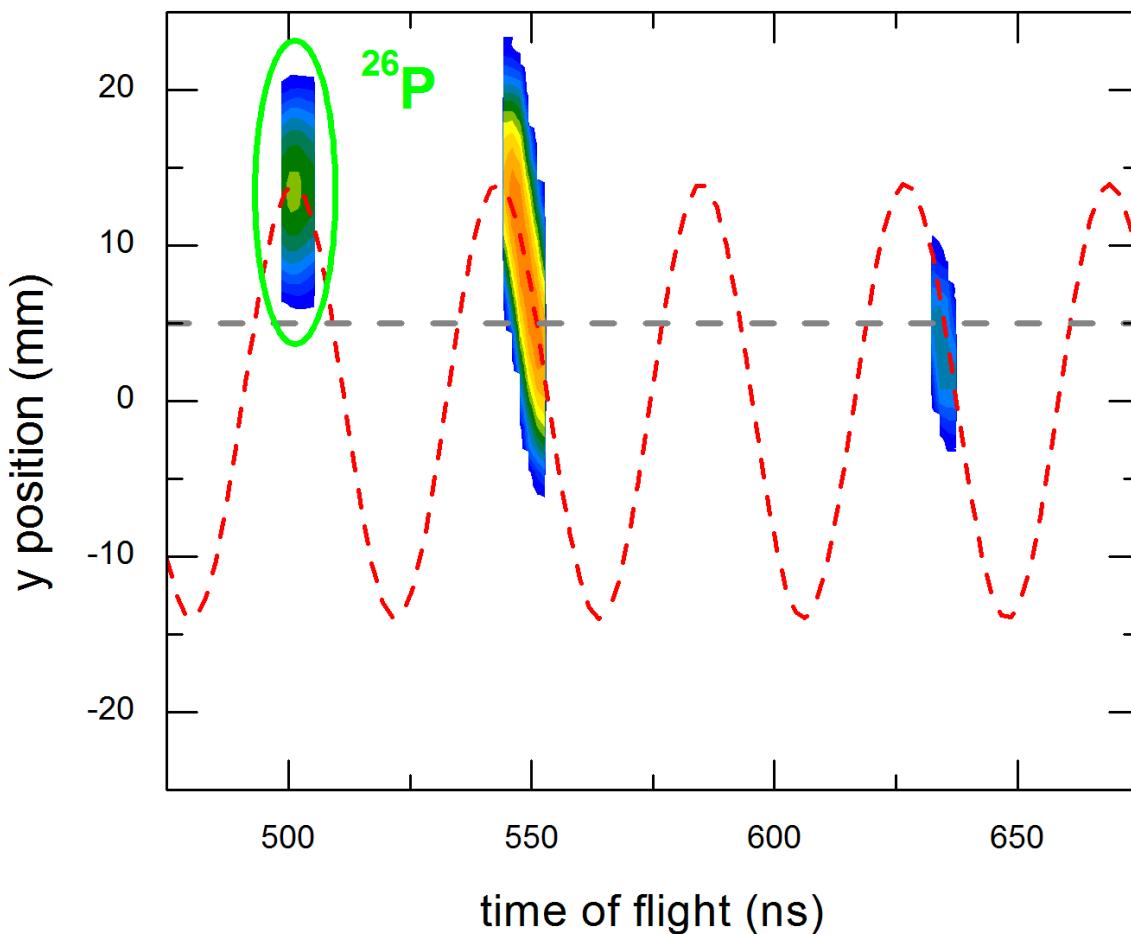
- A LISE simulation is used to calculate the optimum RFFS settings.
- ^{26}P is a drip-line nucleus (lightest P isotope).
- It was discovered in 1983.
- Produced by fragmentation of ^{36}Ar at 150 MeV/u.
- 55 MeV/u beam energy.

How to separate ^{26}P



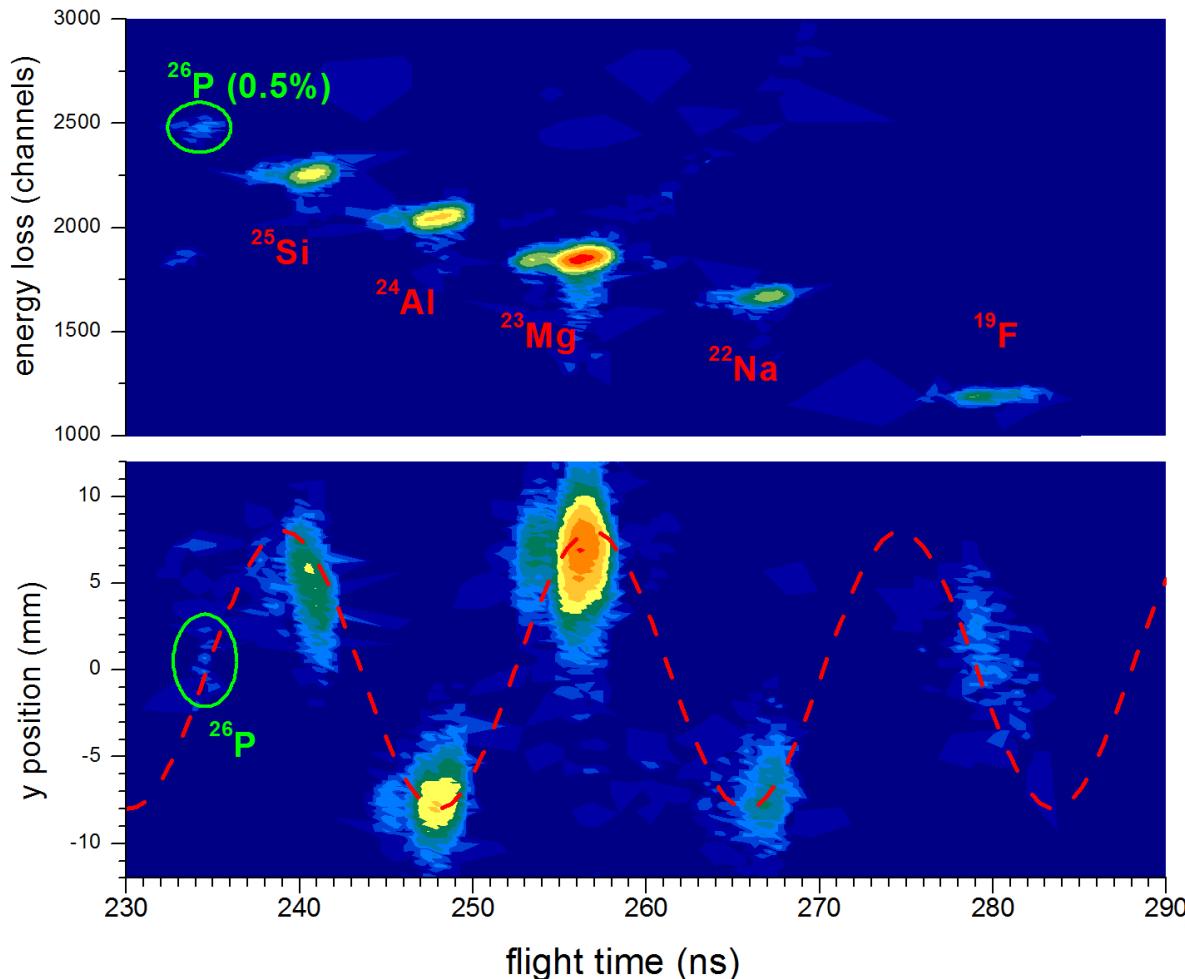
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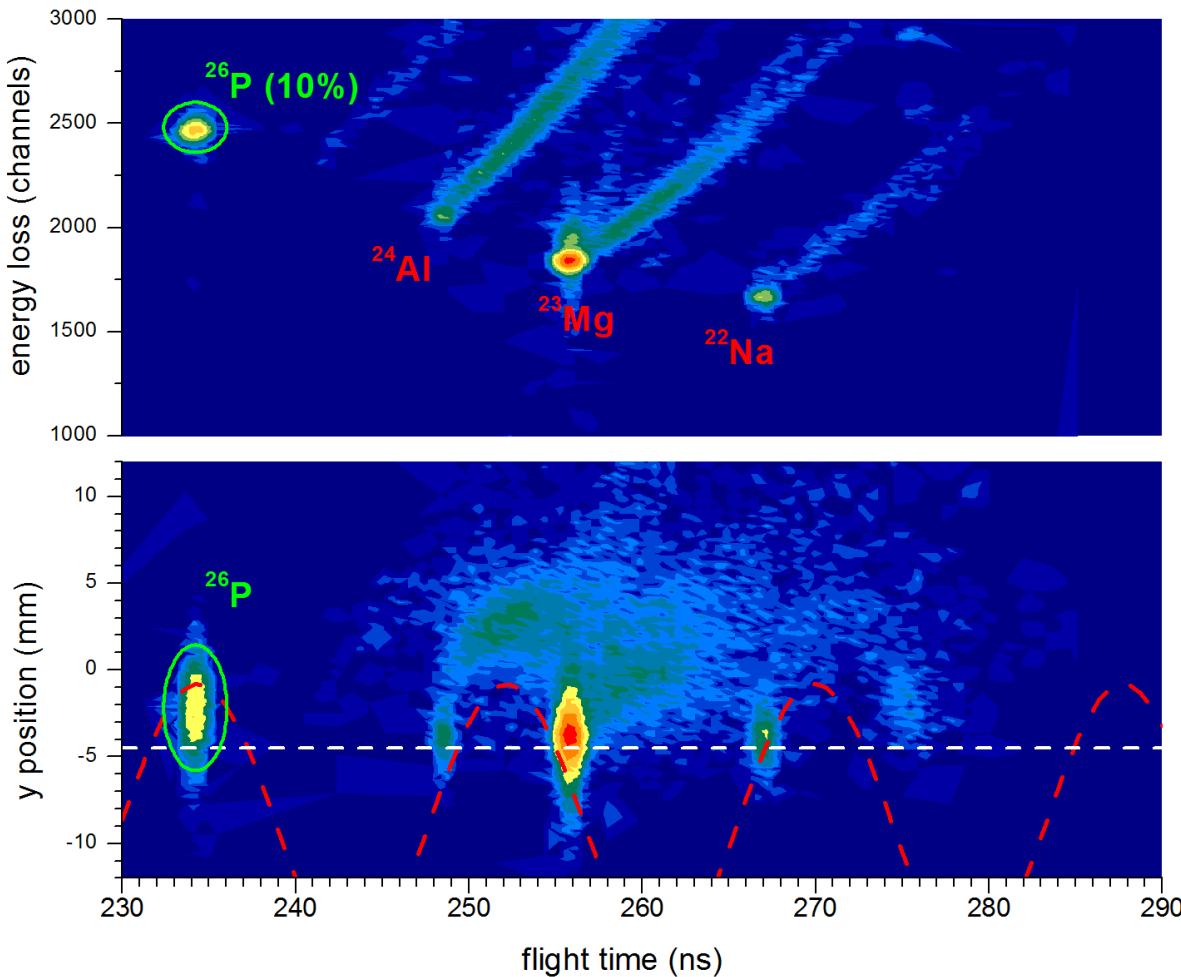
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Measured ^{26}P spectra



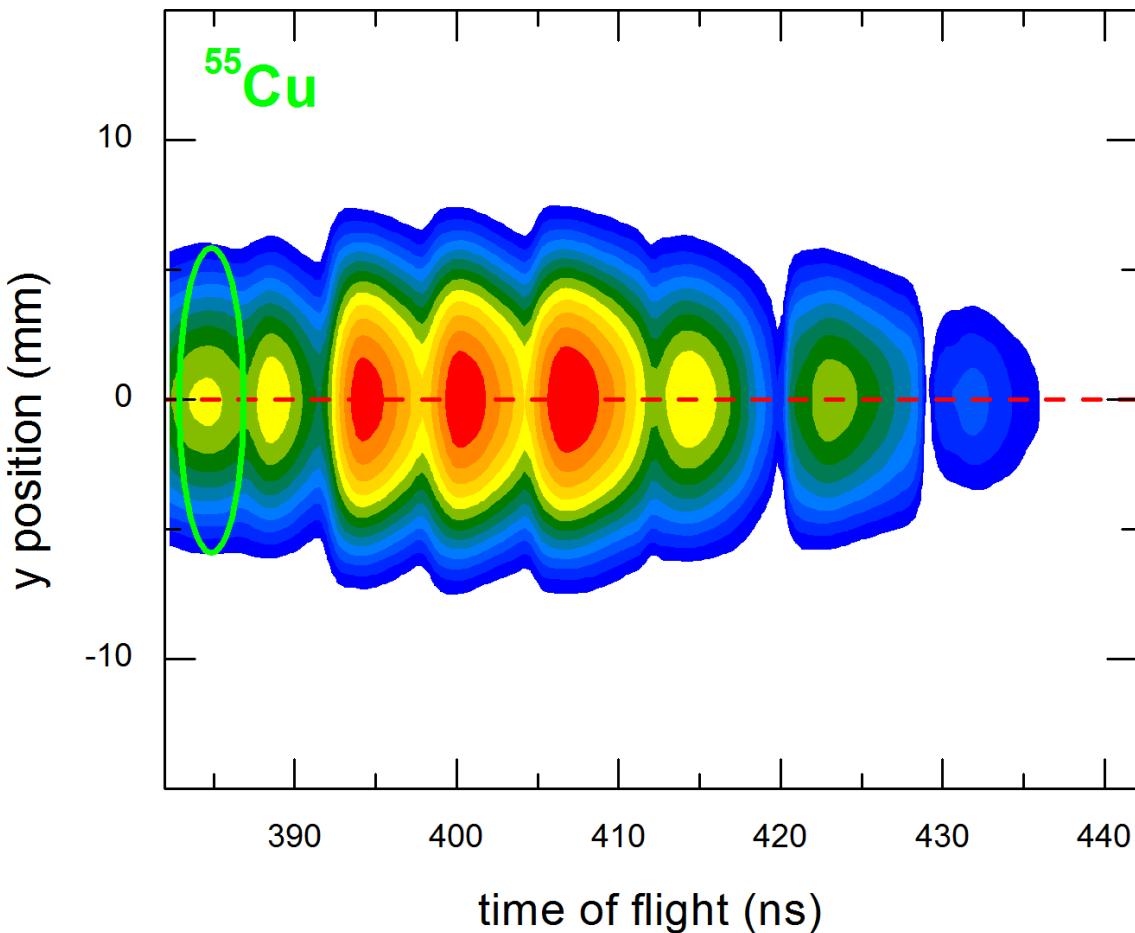
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Measured ^{26}P spectra



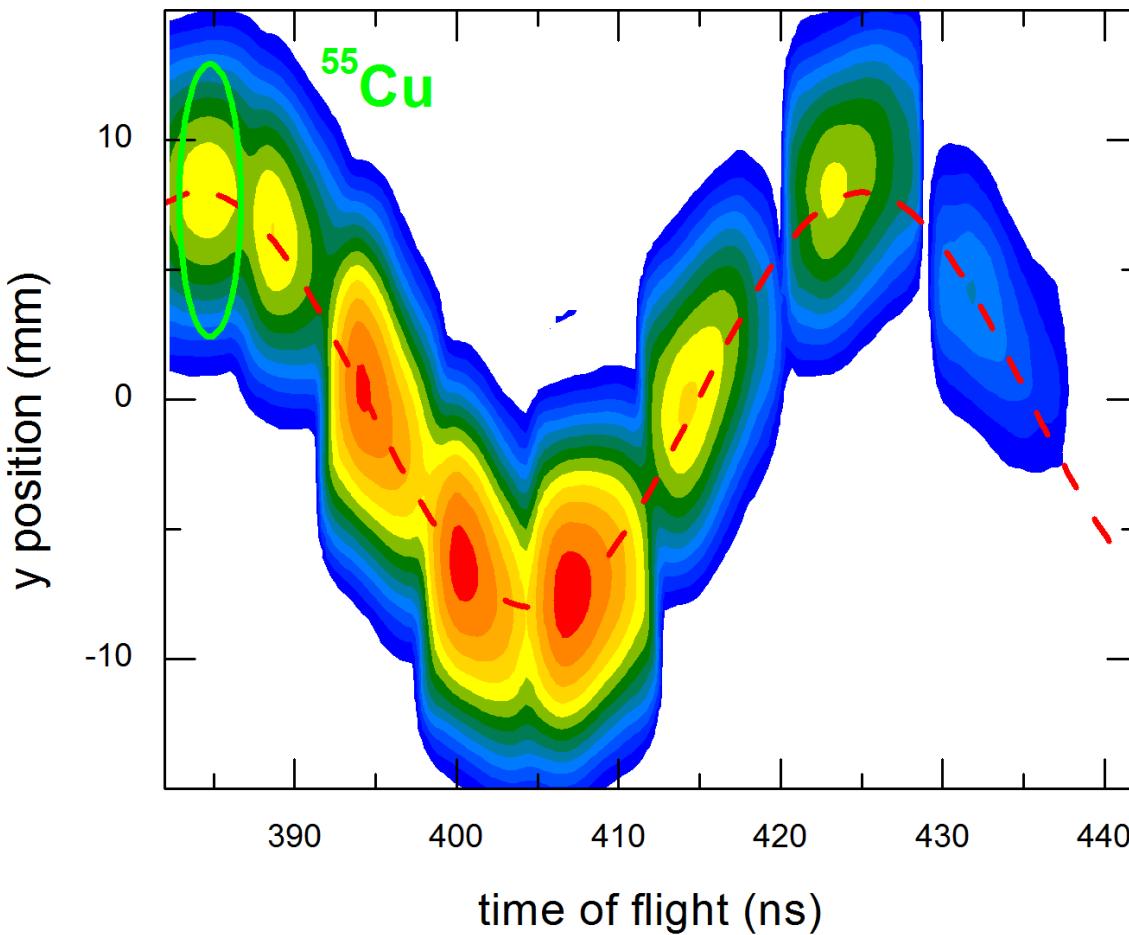
- ^{26}P beam has only 0.5% purity.
- Using the RFFS this was improved to 10%.
- ^{23}Mg contaminant can not be removed at this beam energy because it is almost in phase with ^{26}P .

Another example: ^{55}Cu



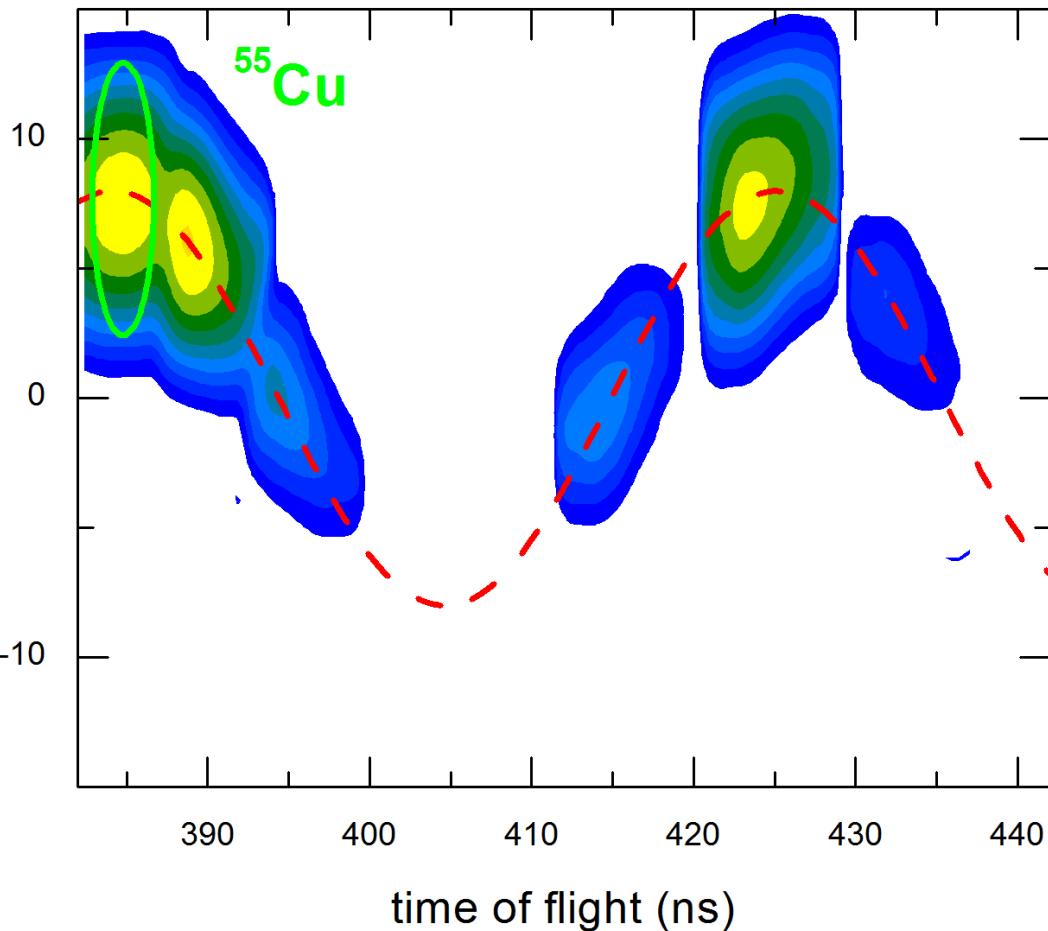
- Also a drip-line nucleus (lightest Cu isotope).
- Discovered in 1987.
- Produced using a ^{58}Ni primary beam at 160 MeV/u.
- 100 MeV/u beam energy.

Another example: ^{55}Cu



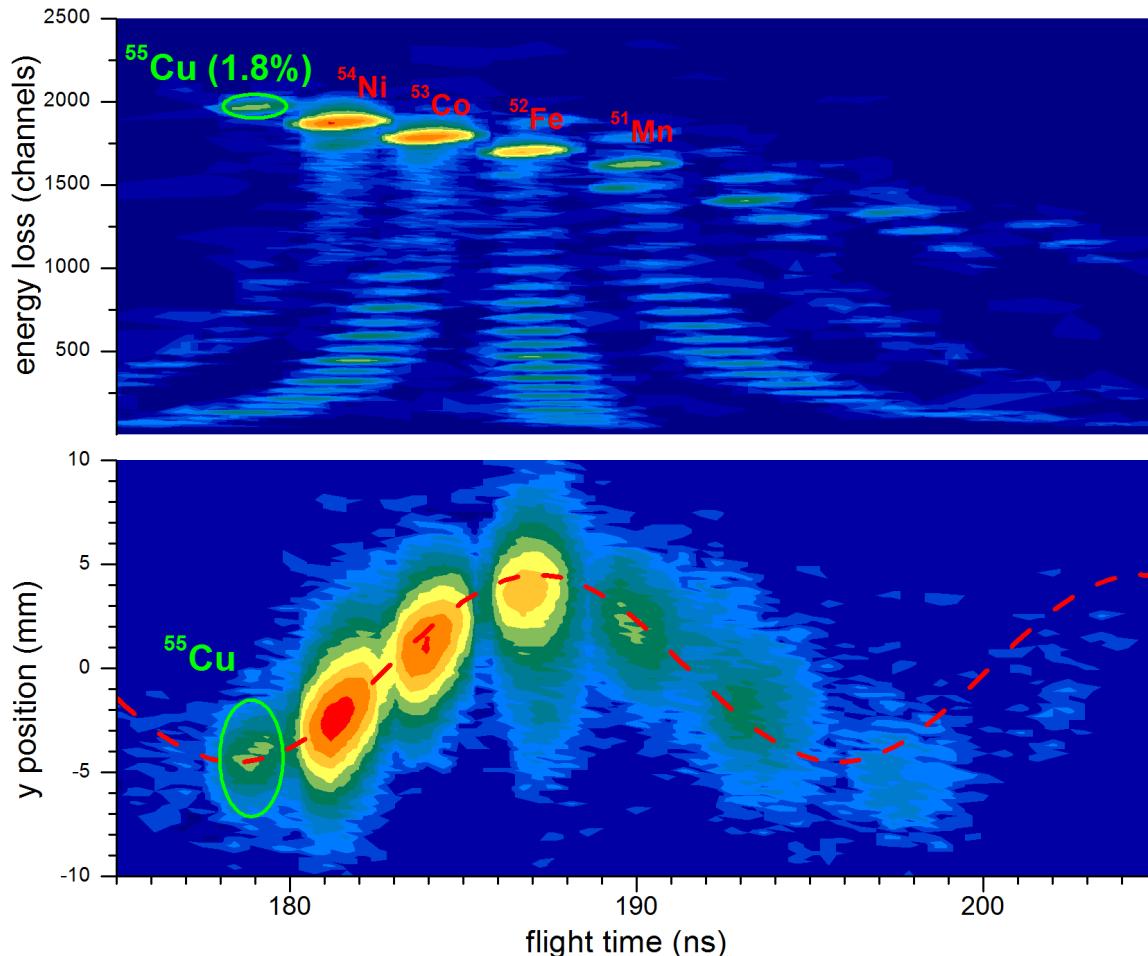
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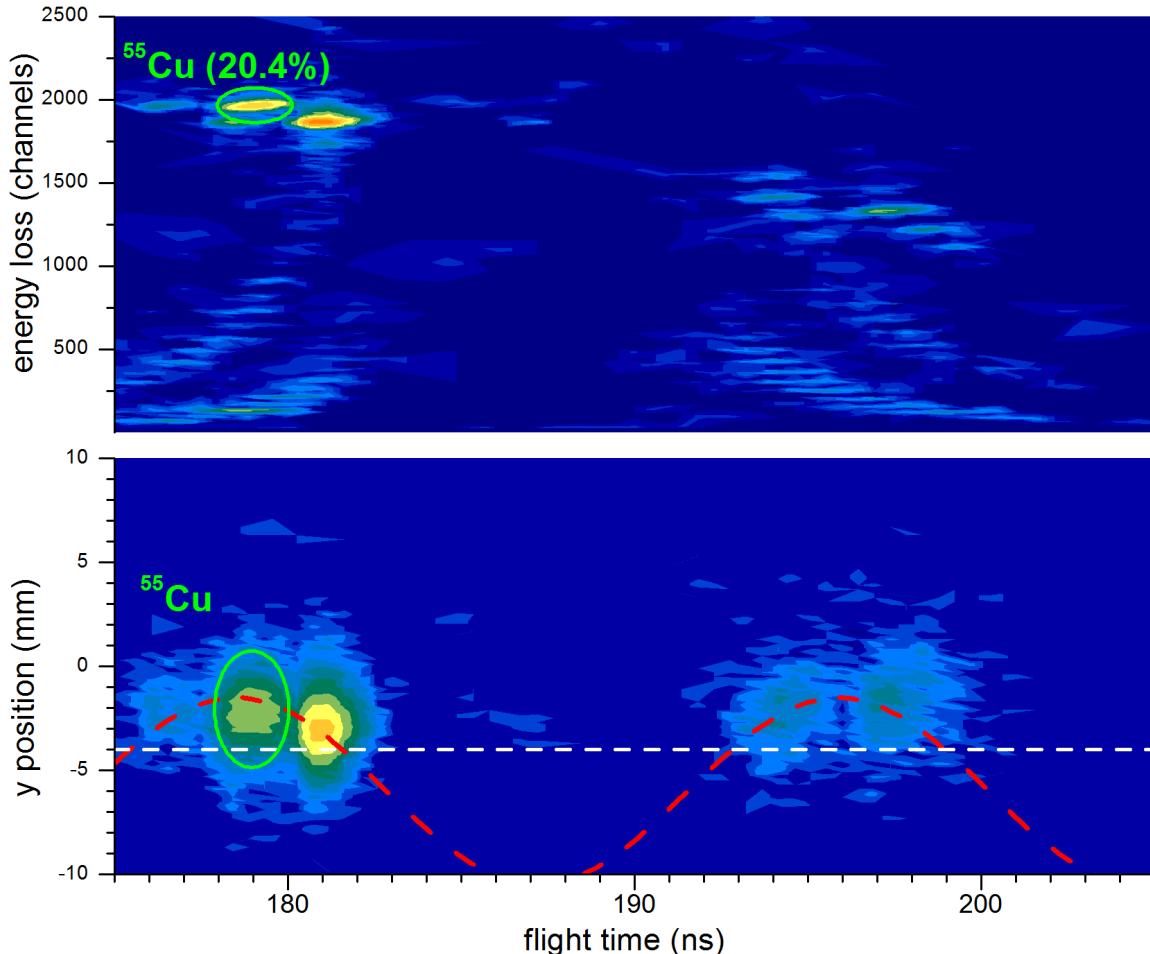
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Measured ^{55}Cu spectra



- ^{55}Cu beam has not even 2% purity.

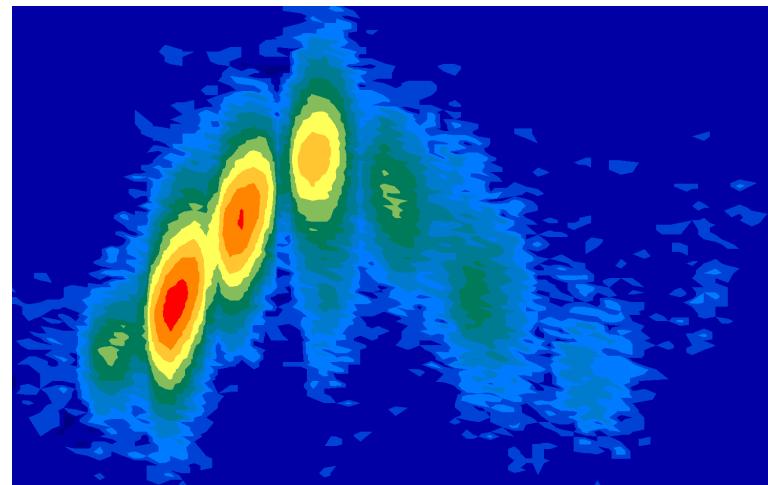
Measured ^{55}Cu spectra



- ^{55}Cu beam has not even 2% purity.
- Using the RFFS this was improved to over 20%.
- ^{54}Ni contaminant can not be completely removed because it is too close in phase with ^{55}Cu .

Conclusion

- The RFFS is a very useful tool to produce neutron-deficient rare isotopes at improved purity.
- The interplay of beam velocity, RF phase, an slit setting requires careful planning of these beams.
- Even in less than optimal situations the purity can be improved by a factor 10 to 20.



Acknowledgment

The beams I showcased were planned and prepared by the NSCL beam physicist group:

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M. Doléans, P. F. Mantica,
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