

Fabrication of Hydrophobic Surfaces from Hydrophilic BeO by Alpha-Irradiation-Induced Nuclear Transmutation

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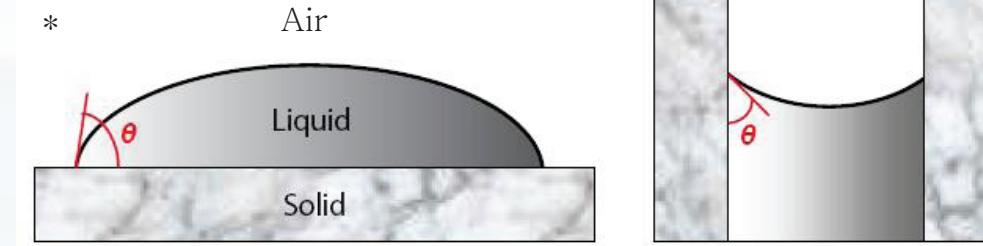
Introduction

1. Wettability

(1) Wetting : process of making contact between a solid and liquid
(adhesion, printing, cleaning, painting, lubrication.....)



(2) Contact angle (CA)



If $\theta < 90^\circ$, the surface is **hydrophilic**. If $\theta > 90^\circ$, the surface is **hydrophobic**.

Introduction

※ Superhydrophobicity ($\theta \geq 150^\circ$)

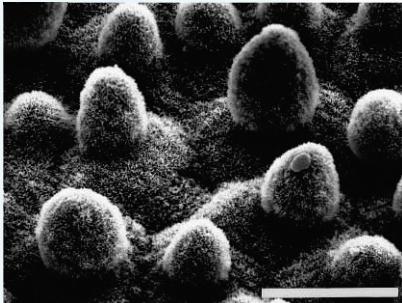
- Superhydrophobicity in nature



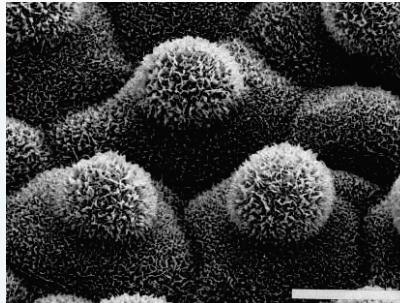
Lotus leaves



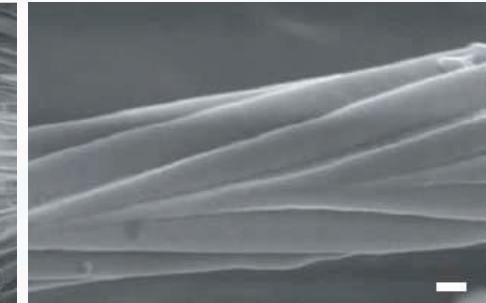
Legs of water strider



Scale bars are 20 μm .



Scale bars are 20 μm and 200 nm, respectively.



* *Planta* 1997, 202, 1. ** *Nature* 2004, 432, 36.

Introduction

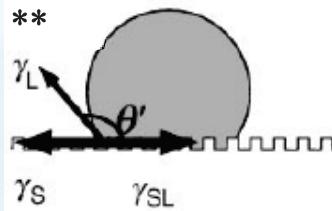
(3) Factors determining the wettability of a surface

① Chemical composition

- Functional groups such as NH_2 , C=O , and OH : hydrophilic
- Functional groups such as CH_x and CF_x : hydrophobic

② Roughness

- Wenzel model *



$$\cos\theta' = \frac{r(\gamma_S - \gamma_{SL})}{\gamma_L} = r\cos\theta$$

$$r = \frac{\text{Real Surface area}}{\text{Apparent Surface area}}$$

r : surface roughness factor ≥ 1

θ' : Contact angle of rough surface

θ : Contact angle of flat surface

If $\theta < 90^\circ$ (hydrophilic)

→ $\theta' < \theta$ (more hydrophilic)

If $\theta > 90^\circ$ (hydrophobic)

→ $\theta' > \theta$ (more hydrophobic)

* *Ind. Eng. Chem.* 1936, 28, 988. ** *Monatsh. Chem.* 2001, 132, 31.

Experimental

1. Materials

- Thermalox995™ (standard BeO), Materion
- diameter = 35 mm, thickness = 2 mm, BeO > 99.5%



2. Irradiation condition

- Alpha (α) particle beam generated from a cyclotron (MC-50, Scanditronix) installed at Korea Institute of Radiological & Medical Sciences (KIRAMS)
- Beam energy = ~ 25 MeV, average beam current = ~ 315 nA
- Fluence = $0, 5.97 \times 10^{14}, 4.53 \times 10^{15} \text{ cm}^{-2}$

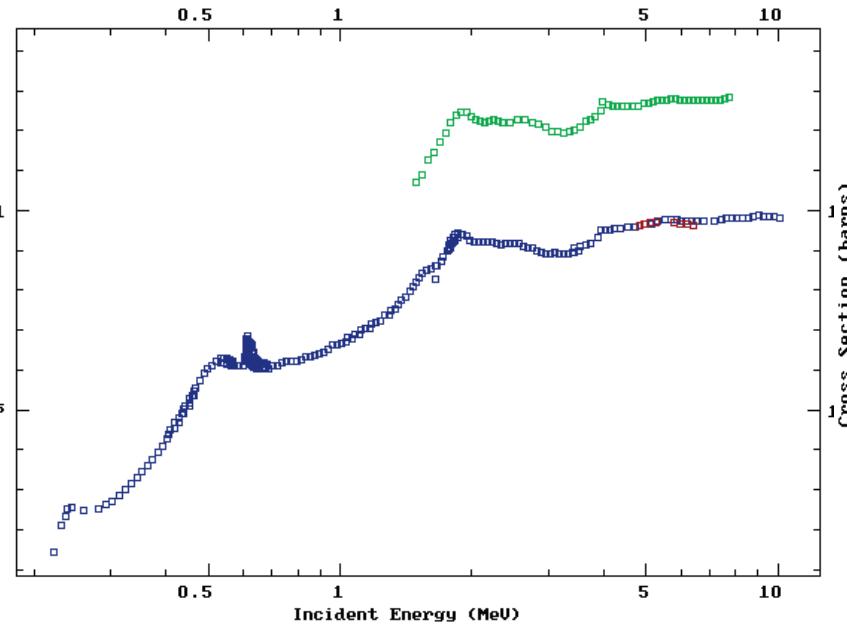
Experimental

- Natural Abundance : ${}^9\text{Be}$ (100%), ${}^{16}\text{O}$ (99.757%)
- Main nuclear reactions probably induced by the alpha irradiation
 - a) ${}^9\text{Be}(\alpha, n){}^{12}\text{C}$
 - b) ${}^{16}\text{O}(\alpha, n){}^{19}\text{F} \rightarrow {}^{19}\text{F}$ (β^+ , $T_{1/2}=17.22$ s),

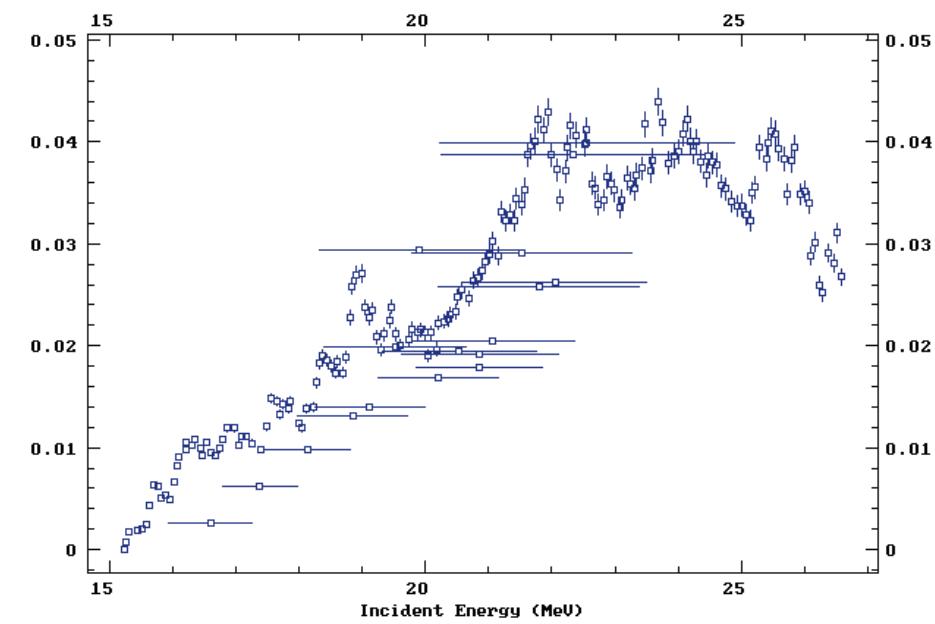
Cross-section data from EXFOR (<http://www.nndc.bnl.gov/nndc/exfor>)

4-BE-9(A,N)6-C-12
EXFOR Request: 6276/1, 2012-Nov-06 23:21:41

Cross Section (barns)

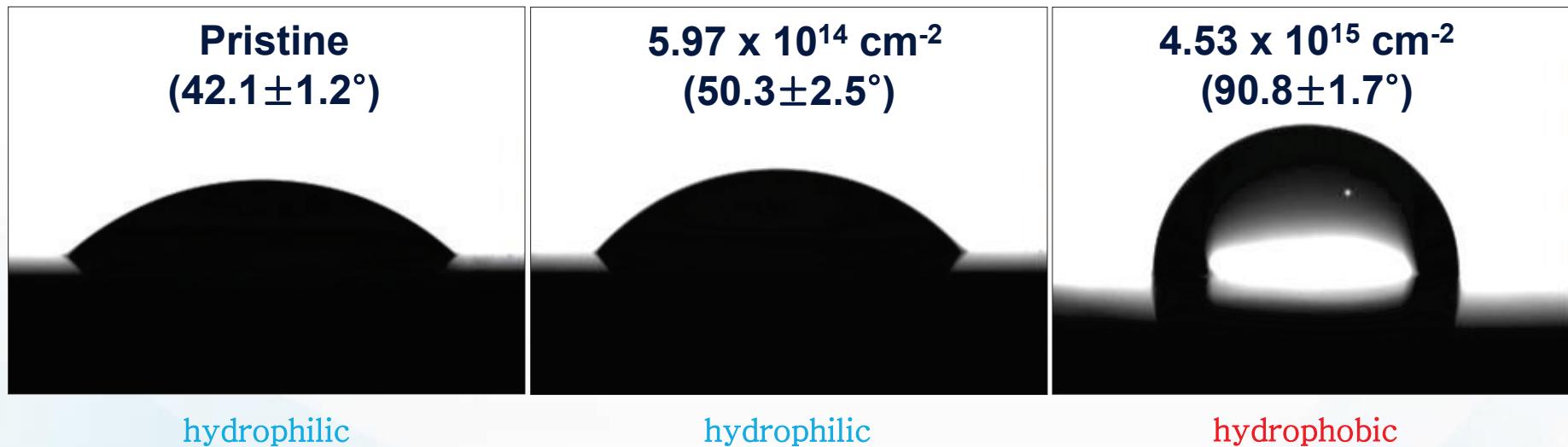


8-O-16(A,N)10-NE-19
EXFOR Request: 5994/1, 2012-Nov-05 19:59:49



Results

1. CA

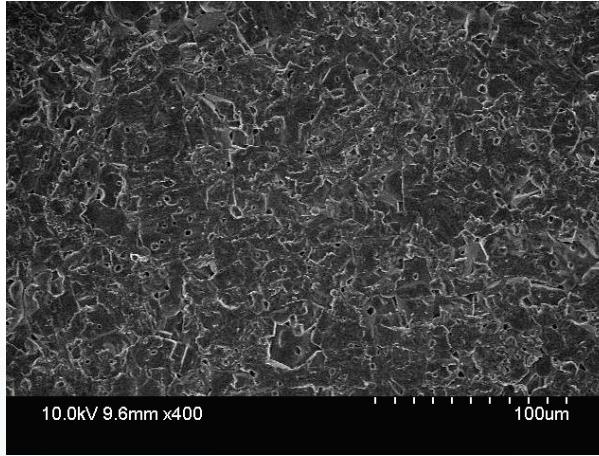


- Hydrophilic BeO surfaces could be converted to hydrophobic surfaces by sufficient alpha irradiation.

Results

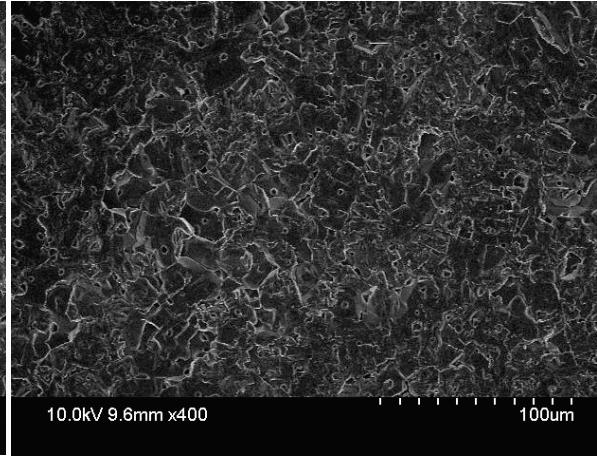
2. Surface morphology

Pristine



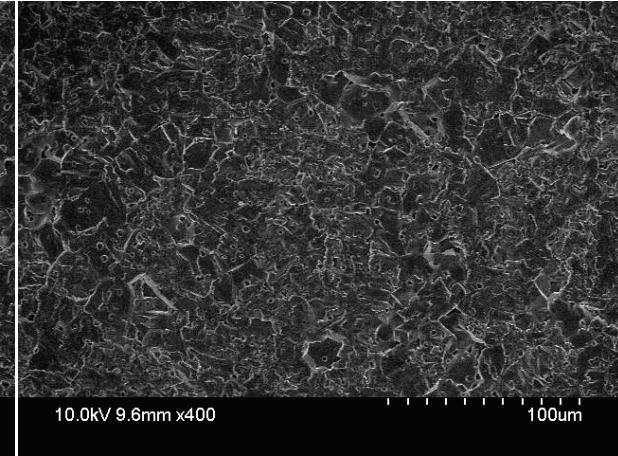
10.0kV 9.6mm x400

$5.97 \times 10^{14} \text{ cm}^{-2}$

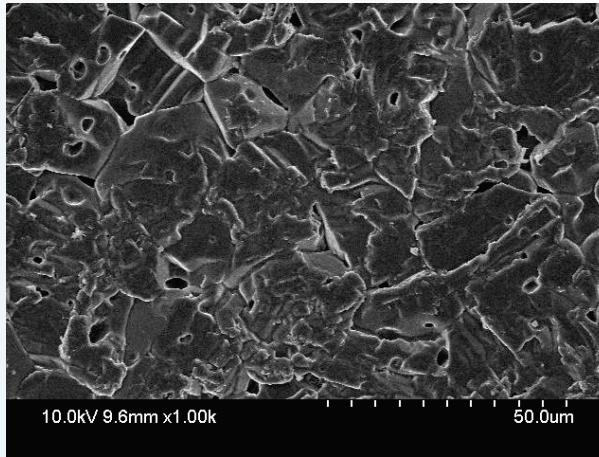


10.0kV 9.6mm x400

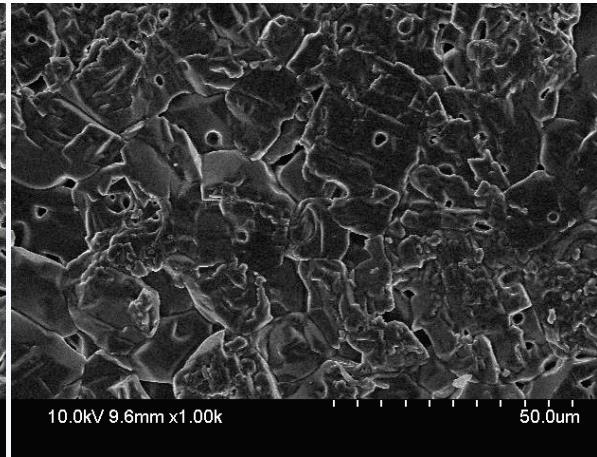
$4.53 \times 10^{15} \text{ cm}^{-2}$



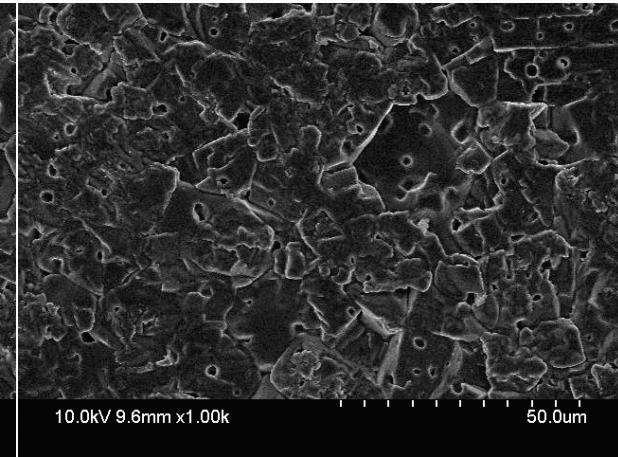
10.0kV 9.6mm x400



10.0kV 9.6mm x1.00k



10.0kV 9.6mm x1.00k

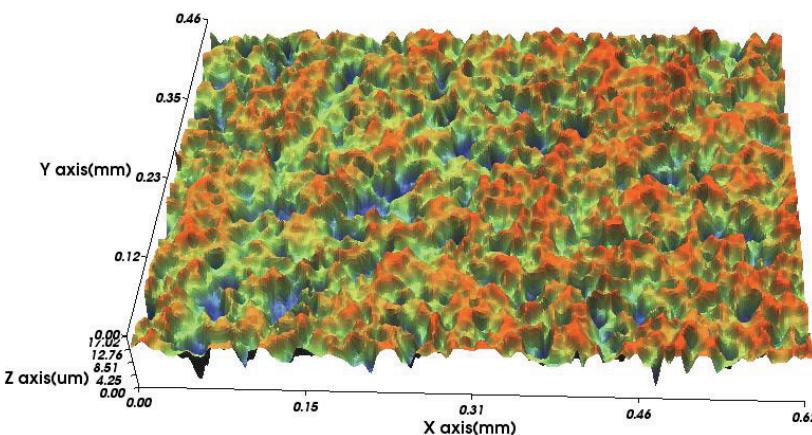


10.0kV 9.6mm x1.00k

- Microstructure of BeO surfaces was not so much influenced by the alpha irradiation.

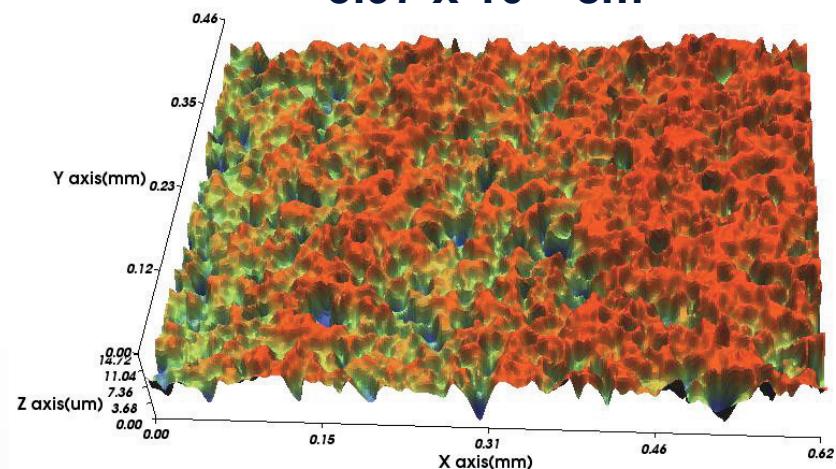
Results

Pristine



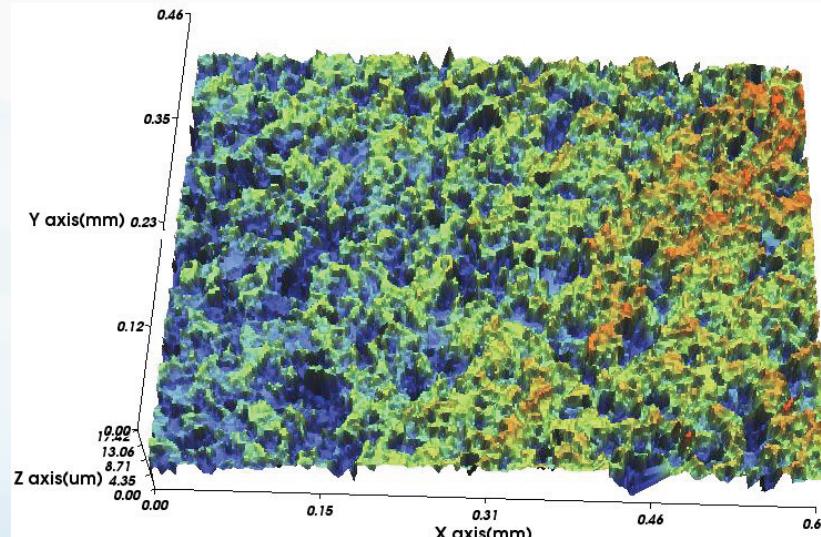
RMS roughness: $1.083 \pm 0.046 \mu\text{m}$

$5.97 \times 10^{14} \text{ cm}^{-2}$



RMS roughness: $1.065 \pm 0.034 \mu\text{m}$

$4.53 \times 10^{15} \text{ cm}^{-2}$



RMS roughness: $1.090 \pm 0.014 \mu\text{m}$

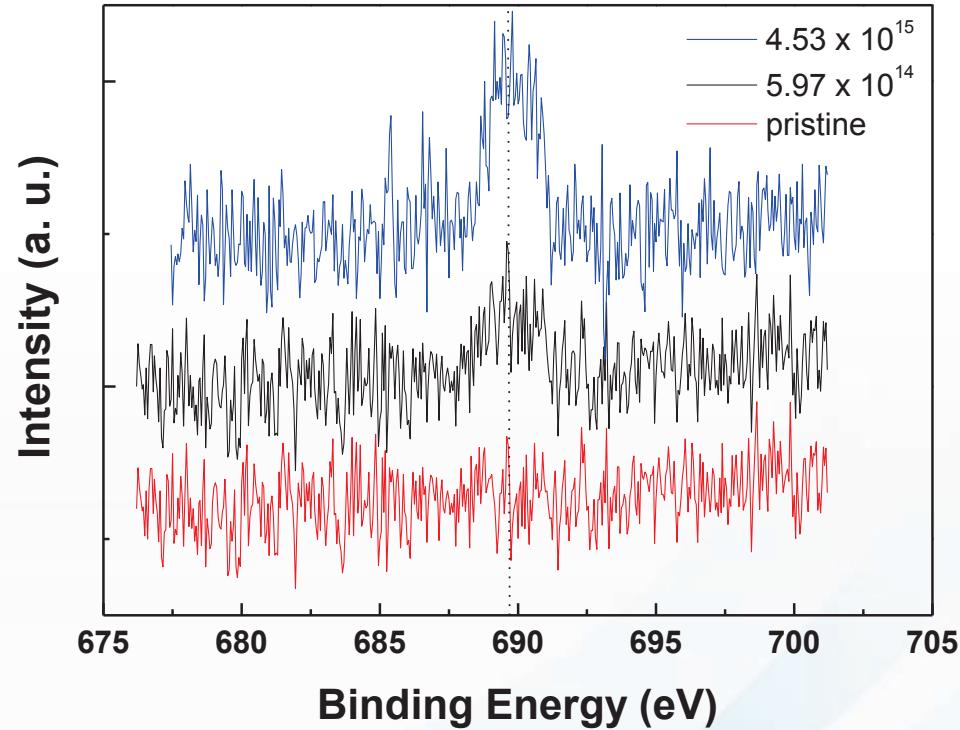
Results

3. XPS

| | Be (At. %) | O (At. %) | C (At. %) | F (At. %) |
|---------------------------------------|------------|-----------|-----------|-----------|
| Pristine | 43.12 | 48.43 | 8.45 | 0 |
| $5.97 \times 10^{14} \text{ cm}^{-2}$ | 39.62 | 47.94 | 12.33 | 0.11 |
| $4.53 \times 10^{15} \text{ cm}^{-2}$ | 35.86 | 47.51 | 16.36 | 0.27 |

decreased

increased



688.9 eV : CF_2 molecular bonds

Summary

1. A facile route to fabricate hydrophobic surfaces from hydrophilic BeO was presented on the base of the alpha irradiation.
2. When Be and O atoms were irradiated with alpha particle beam, C and F atoms were successfully created.
3. CF_2 functional groups, providing hydrophobic property, were formed by the combination of produced C and F atoms.

Thank you for
attention!