

Theoretical Astrophysics: Physics of Sun and Stars

Homework 4

P. Käpylä, I. Milić

July 5, 2024

Deadline for this homework is 01/07 23:59

Problem 1: Following the numerical exercise that we did on the handson number 9, related to stellar atmospheres, calculate the emergent flux from a given atmospheric model. You can assume that the optical depth given in the atmospheric model is representative of the average optical depth over all wavelengths (so - called gray atmosphere). This means that you do not have to worry about wavelengt dependence of opacity and thus optical depth.

For the same atmosphere, check whether the flux is constant with height/depth, and what is the ratio of the K-integral and the mean intensity with depth.

For this you will need to solve radiative transfer equation numerically in different directions (you specify a grid of directions) and to integrate, numerically, intensity over the angle.

Problem 2:

For a gas of given pressure and temperature, assuming it only consists of hydrogen in neutral and ionized state, find the electron density (you will need Saha equation for this, follow exercise 8).

Assuming that the number density of negative ion of Hydrogen is negligible compared to the neutral and ionized hydrogen, use Saha equation to estimate the number density of negative ions of hydrogen.

Now, using Boltzmann equation, estimate the number density of neutral Hydrogen in state $i = 3$ of neutral Hydrogen. Recall that the energy of the hydrogen states scales like:

$$E_i = \frac{-E_{ion}}{i^2} \quad (1)$$

where i is the principal quantum number, E_i is the energy of the level (negative, because we need to add the energy to break up the bound state), and E_{ion} is the ionization energy of Hydrogen, which is equal to 13.6 eV. Ioniozation energy of the negative ion of Hydrogen is 0.75 eV.

Finally, compare the number densities of neutral hydrogen in state $i = 3$ and the negative ion of hydrogen in the model atmosphere you used in the problem 1. For this problem you will need temperature (third column) and total gas pressure (fourth column). Keep in mind the model is given in CGS units! :-)

Problem 3:

Useful physical constants

- $R_{\odot} = 696 \times 10^6 \text{ m}$
 - $M_{\odot} = 1.989 \times 10^{30} \text{ kg}$
 - $L_{\odot} = 3.83 \times 10^{26} \text{ W}$
 - $T_{\odot}^{\text{eff}} = 5777 \text{ K}$
 - $1 \text{ AU} = 1.496 \times 10^8 \text{ km}$
 - $c = 2.997 \times 10^8 \text{ m/s}$
 - $G = 6.674 \times 10^{-11} \text{ Nm}^2/\text{kg}^2$
 - $k = 1.38 \cdot 10^{-23} \text{ J/K}$
 - $m_{\text{H}} = 1.67 \cdot 10^{-27} \text{ kg}$
 - $h = 6.626 \times 10^{-34} \text{ J s.}$
 - $k = 1.38 \times 10^{-23} \text{ J/K.}$
-