

# Relating Microstructure with Rheology in Glassy Colloids

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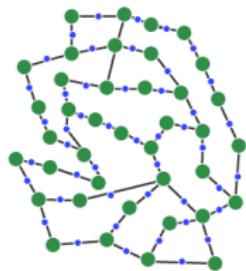
# Outline

- ▶ Glasses and colloids
- ▶ The correlation function
- ▶ Excess entropy
- ▶ Expansion of excess entropy in correlation functions
- ▶ Relate correlation functions with dynamics via excess entropy

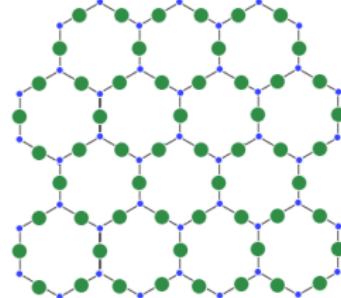
# Glasses and colloids

Glasses are like if a liquid was solid.

- ▶ I.e., they are rigid but isotropic.
- ▶ Their dynamics involve long-timescale *relaxations*.



Amorphous Solids

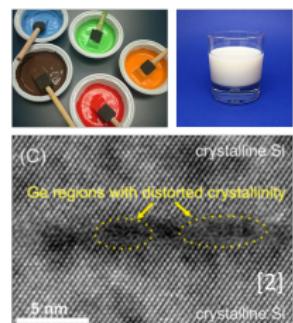
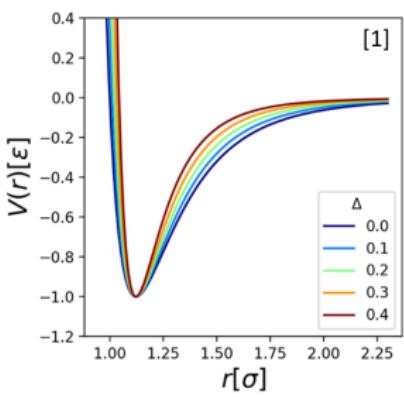
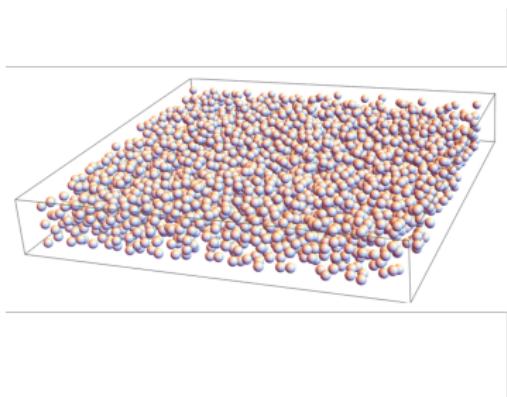


Crystalline Solids

# Glasses and colloids

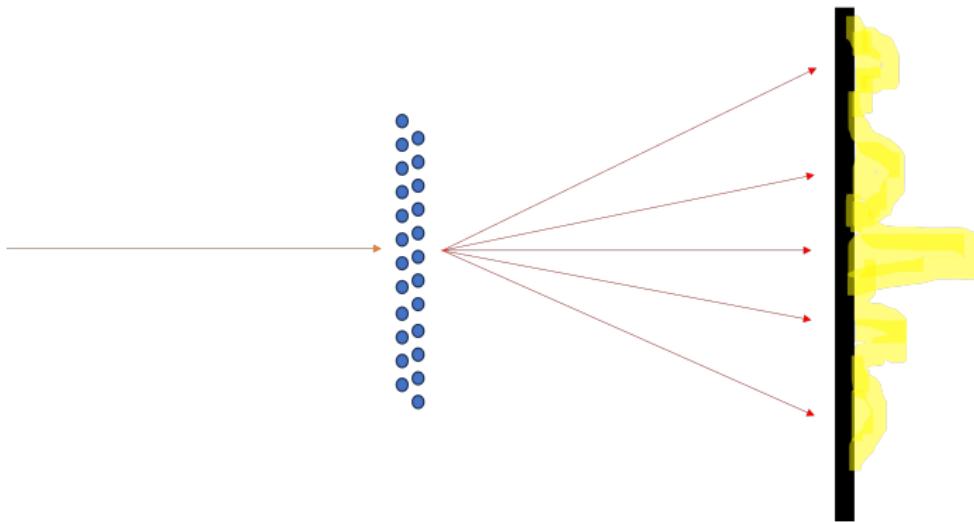
Colloids are suspensions of particles.

- ▶ They are often spherical.
- ▶ They have interactions.
- ▶ They often form glasses and other disordered states.

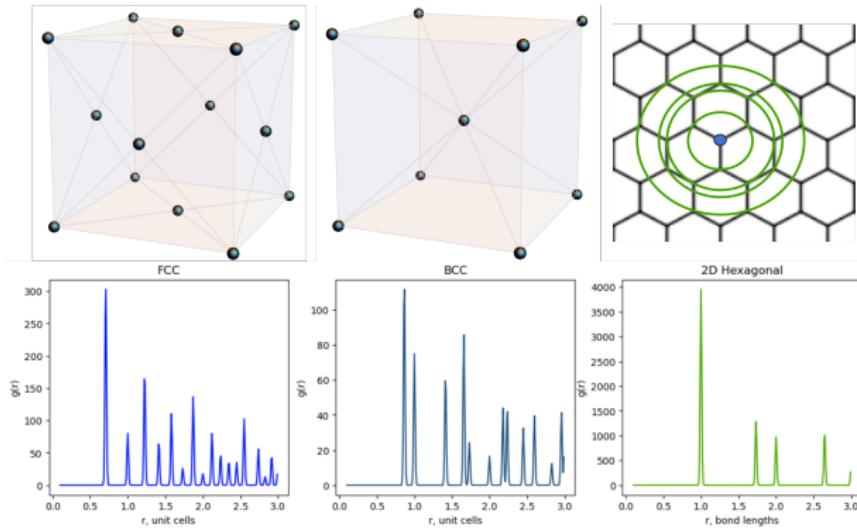


# The correlation function $g(r)$

- ▶  $g(r)$  is the expected (relative) density of particles given that you are a distance  $r$  away from some particle.
- ▶ It can be calculated from scattering experiments.



# $g(r)$ contains information about structure



$g(r)$  is calculated experimentally using a **histogram**.

*Recall:  $g(r)$  is the expected (relative) density of particles given that you are a distance  $r$  away from some particle.*

# What if the lattice has disorder?

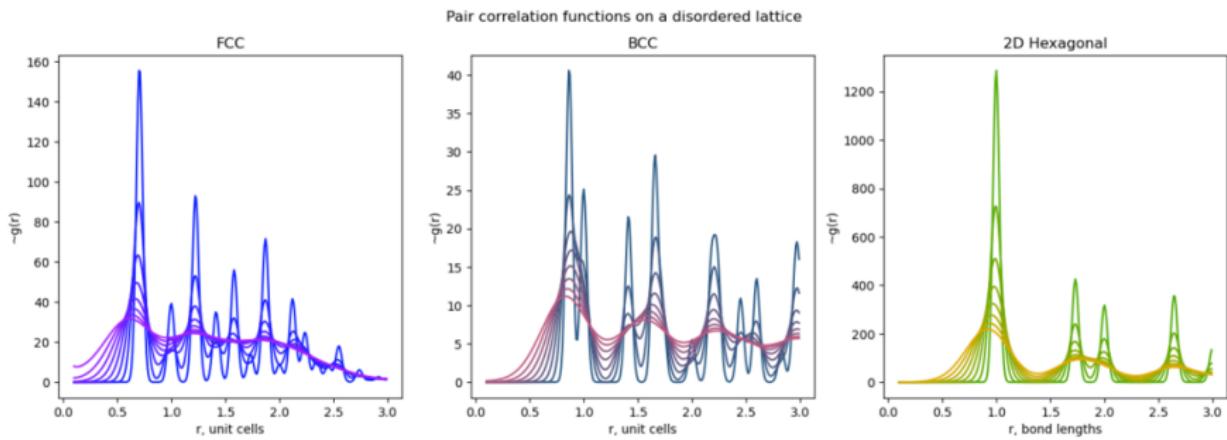
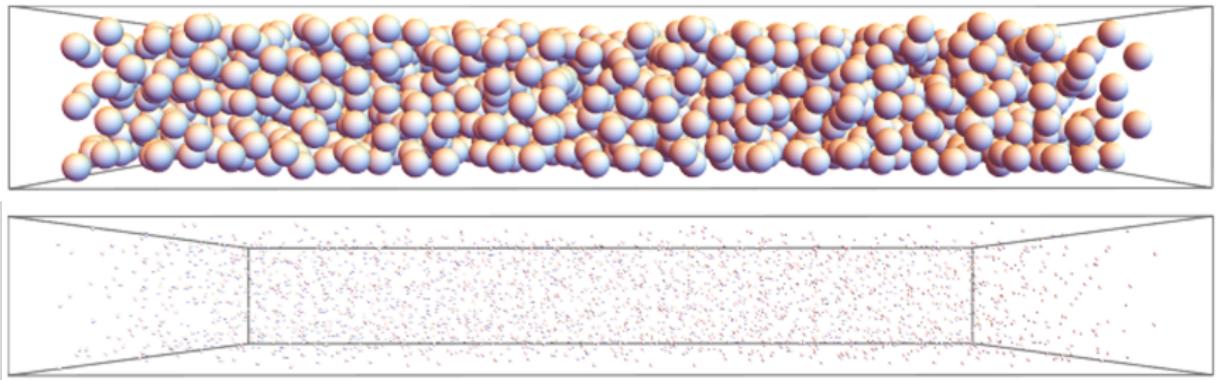


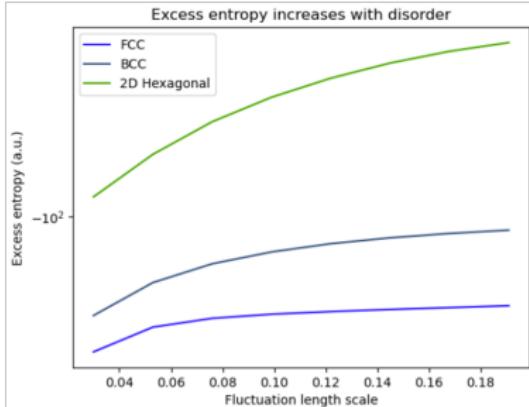
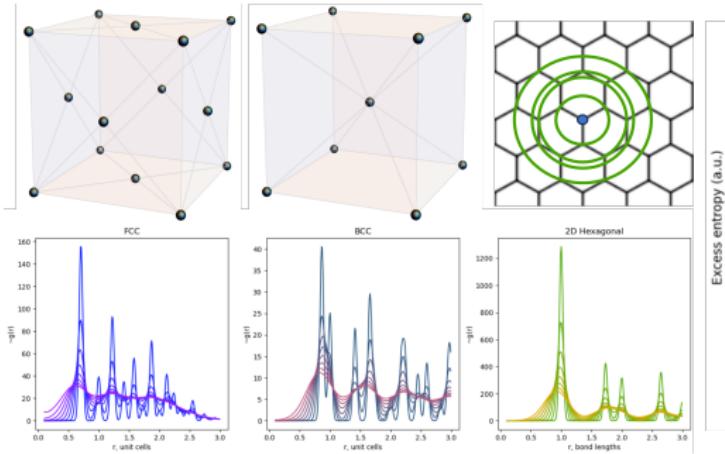
Figure:  $g(r)$  on lattice with gaussian fluctuations, which increase in length scale from 0.03 (less red) to 0.19 (more red) units.

# Configurational entropy

- ▶ The ideal gas has maximal configurational entropy.
- ▶ In terms of entropy, colloidal spheres and an equivalent ideal gas differ principally in the number of available (spatial) configurations.
- ▶ This difference is called **excess entropy**.



# Excess entropy measures disorder



# Excess entropy in terms of correlation functions

- We can write the entropy of a colloid in terms of correlation functions.
- The bulk of the **excess** entropy is accounted for by a term involving only  $g_2(r) = g(r)$ .

[3]

$$S_N = -\frac{k_B}{h^{3N} N!} \int \cdots \int P_N \ln P_N d\mathbf{r}_1 d\mathbf{p}_1 \cdots d\mathbf{r}_N d\mathbf{p}_N$$
$$= -\frac{k_B}{N!} \int \cdots \int f_N^{(N)} \ln h^{3N} f_N^{(N)} d\mathbf{r}_1 d\mathbf{p}_1 \cdots d\mathbf{r}_N d\mathbf{p}_N$$

$$g_N^{(n)}(\mathbf{r}_1, \dots, \mathbf{r}_n)$$
$$= \frac{\rho_N^{N-n}}{(N-n)!} \int \cdots \int g_N^{(N)}(\mathbf{r}_1, \dots, \mathbf{r}_N) d\mathbf{r}_{n+1} \cdots d\mathbf{r}_N$$

$$g_N^{(N)}(1, \dots, N) = g_N^{(2)}(1, 2) \cdots g_N^{(2)}(N-1, N)$$
$$\times \delta g_N^{(3)}(1, 2, 3) \cdots \delta g_N^{(3)}(N-2, N-1, N)$$
$$\times \cdots \times \delta g_N^{(n)}(1, \dots, n).$$

$$H_N = \sum_{s=1}^N \frac{\mathbf{p}_s^2}{2m} + \Phi_N$$

$$f_N^{(N)}(\mathbf{r}_1, \mathbf{p}_1, \dots, \mathbf{r}_N, \mathbf{p}_N) = \frac{e^{-\beta H_N}}{h^{3N} Z_N}$$

$$= f_N^{(1)}(\mathbf{p}_1) \cdots f_N^{(1)}(\mathbf{p}_N) g_N^{(N)}(\mathbf{r}_1, \dots, \mathbf{r}_N)$$

$$f_N^{(1)}(\mathbf{p}) = \rho_N (\beta / 2\pi m)^{3/2} \exp(-\beta \mathbf{p}^2 / 2m)$$

$$f_N^{(n)}(\mathbf{r}_1, \mathbf{p}_1, \dots, \mathbf{r}_n, \mathbf{p}_n)$$
$$= \frac{1}{(N-n)!} \int \cdots \int f_N^{(N)}(\mathbf{r}_1, \mathbf{p}_1, \dots, \mathbf{r}_N, \mathbf{p}_N)$$
$$\times d\mathbf{r}_{n+1} d\mathbf{p}_{n+1} \cdots d\mathbf{r}_N d\mathbf{p}_N.$$

$$f_N^{(n)}(\mathbf{r}_1, \mathbf{p}_1, \dots, \mathbf{r}_n, \mathbf{p}_n) = f_N^{(1)}(\mathbf{p}_1) \cdots f_N^{(1)}(\mathbf{p}_n) g_N^{(n)}(\mathbf{r}_1, \dots, \mathbf{r}_n)$$

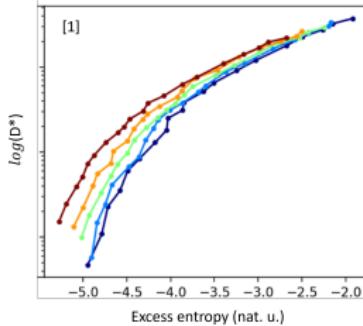
$$S_2 = -2\pi\rho \int_0^\infty \{g(r)\ln[g(r)] - [g(r) - 1]\} r^2 dr \quad [4]$$

# Connecting entropy with dynamics

Recall that the dynamics of glassy states consist of long-timescale rearrangements.

- ▶ Idea: The frequency of rearrangements is proportional to the number of available microstates [5].
- ▶ This holds at the ensemble level [1], and is predicted to hold at the **single-particle level**.

$$D^* \propto e^S \quad (1)$$



# Summary

- ▶  $g(r)$  is the expected (relative) density of particles given that you are a distance  $r$  away from some particle. It characterizes the structure in a system.
- ▶ Excess entropy is the entropy minus that of an equivalent ideal gas. It characterizes the degree of disorder in a system.
- ▶ Intuitively, systems with higher structure will change less over time.
  - ▶ Excess entropy quantifies the connection between  $g(r)$  and particle dynamics.

# Citations

1. Graham, I.R., Arratia, P.E. and Riggleman, R.A. (2023) 'Exploring the relationship between softness and excess entropy in glass-forming systems', *The Journal of Chemical Physics*, 158(20). doi:10.1063/5.0143603.
2. Grydlik, M. *et al.* (2016) 'Lasing from glassy ge quantum dots in Crystalline Si', *ACS Photonics*, 3(2), pp. 298–303. doi:10.1021/acsphotonics.5b00671.
3. Wallace, D.C. (1987) 'On the role of density fluctuations in the entropy of a fluid', *The Journal of Chemical Physics*, 87(4), pp. 2282–2284. doi:10.1063/1.453158.
4. Baranyai, A. and Evans, D.J. (1989) 'Direct entropy calculation from computer simulation of liquids', *Physical Review A*, 40(7), pp. 3817–3822. doi:10.1103/physreva.40.3817.
5. Dzugutov, M. (1996) 'A universal scaling law for atomic diffusion in condensed matter', *Nature*, 381(6578), pp. 137–139. doi:10.1038/381137a0.