

SCHEMATA

A 3D DIGITAL HAND GESTURE RECOGNITION TOOL
FOR STUDYING 3D CONCEPTS IN CHEMISTRY

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schema |'skēmə|

noun (pl. **schemata** |-mətə| or schemas)

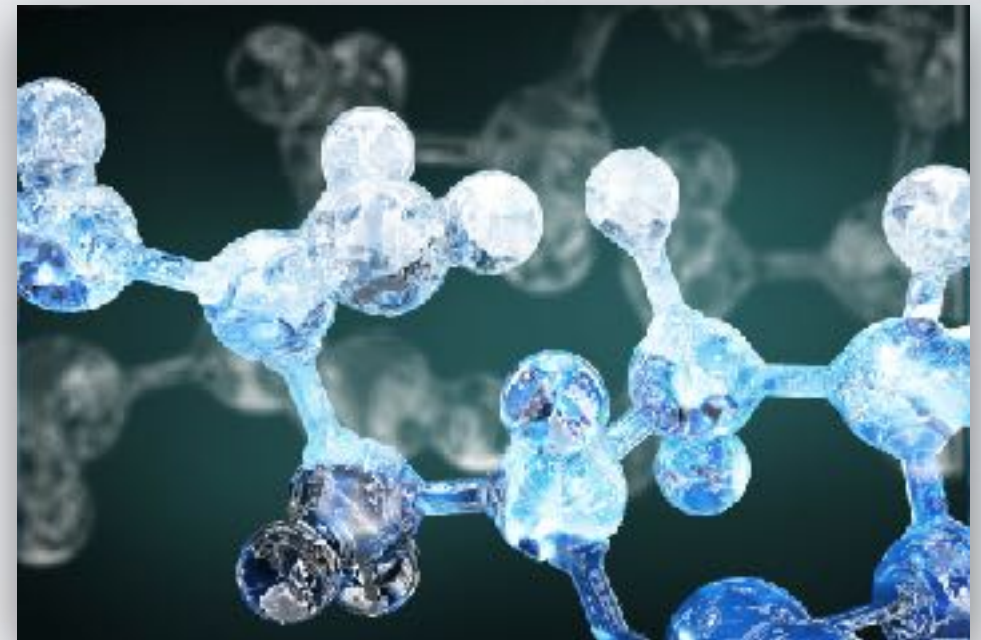
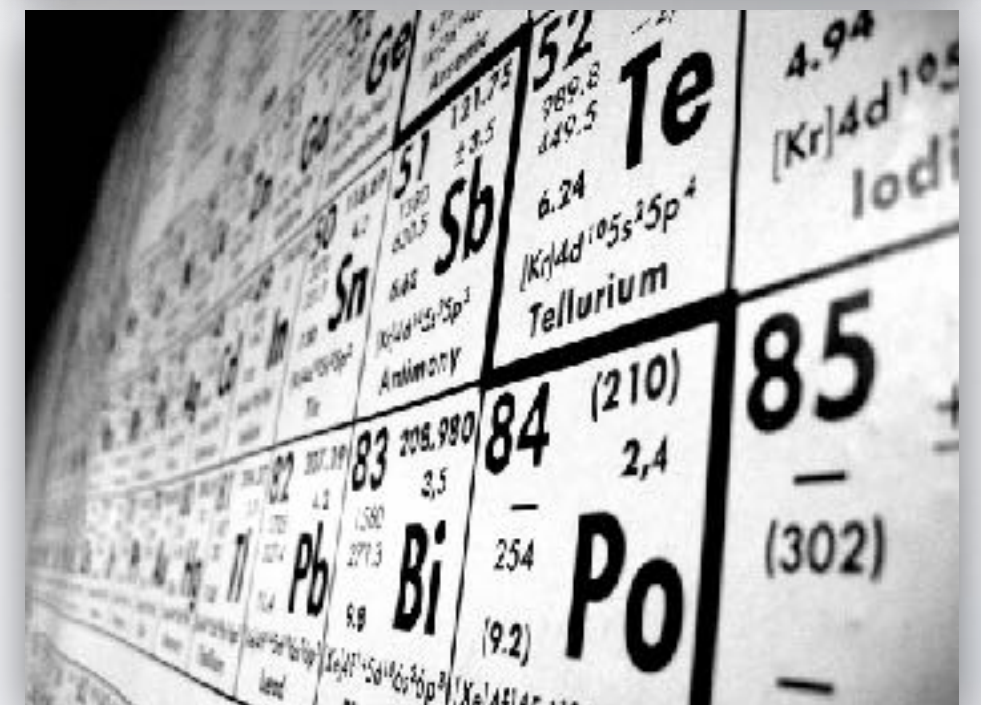
a representation of a plan or theory in the form of an outline or model: *a schema of scientific reasoning.*

SCHEMATA

MOTIVATIONS

CHEMISTRY

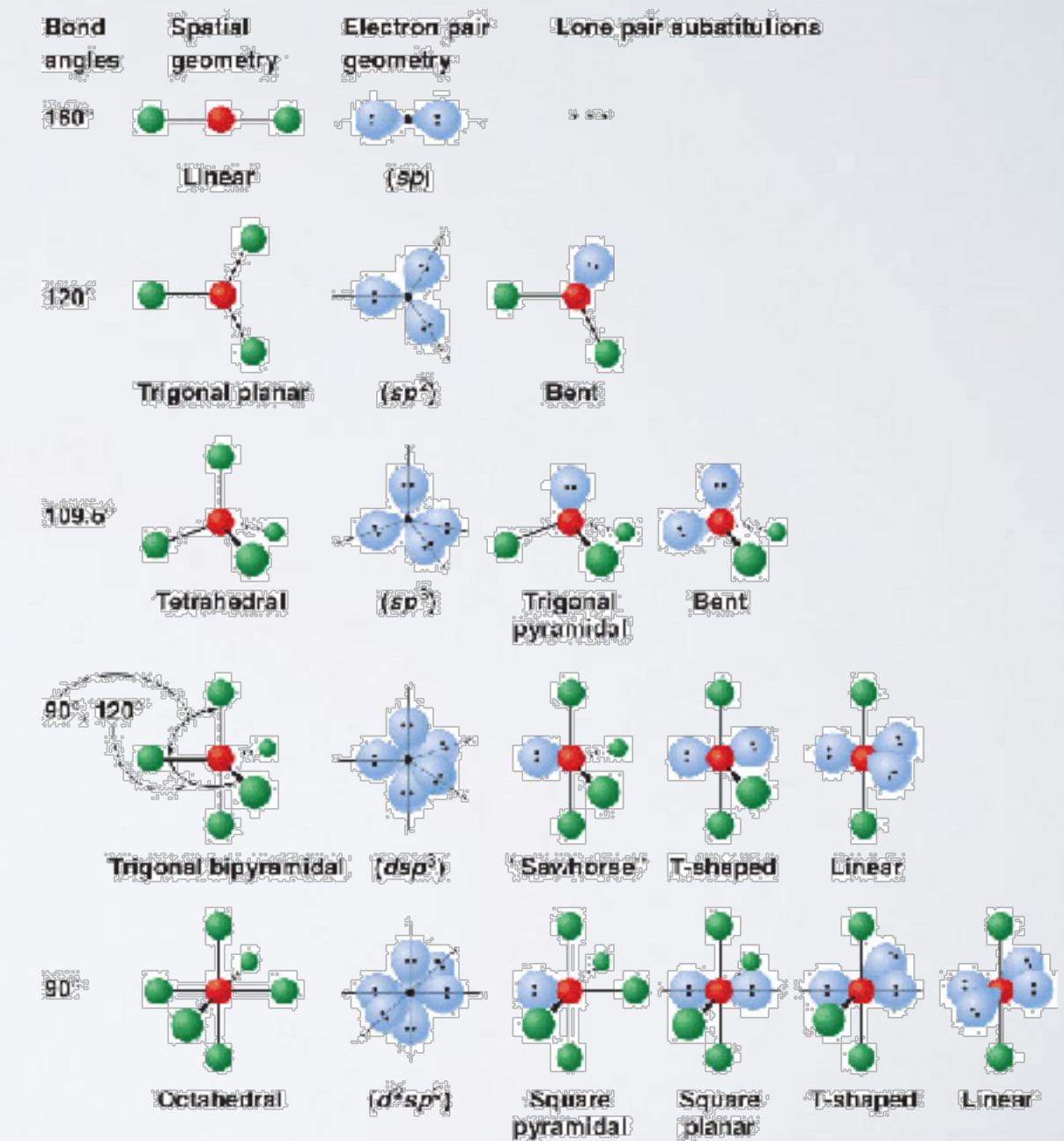
- A physical science!
- But we can't really see teeny tiny molecular and atomic interactions...
- How to create 3D representations of molecules and reactions to visualize otherwise intangible concepts and make them seem REAL?

51 Sb Antimony	52 Te Tellurium	53 I Iodine
83 Bi Bismuth	84 Po Polonium	85 At Astatine

THE LESSON

- VSEPR Model Theory
- VSEPR = Valence Shell Electron Pair Repulsion
- But first, a little background...



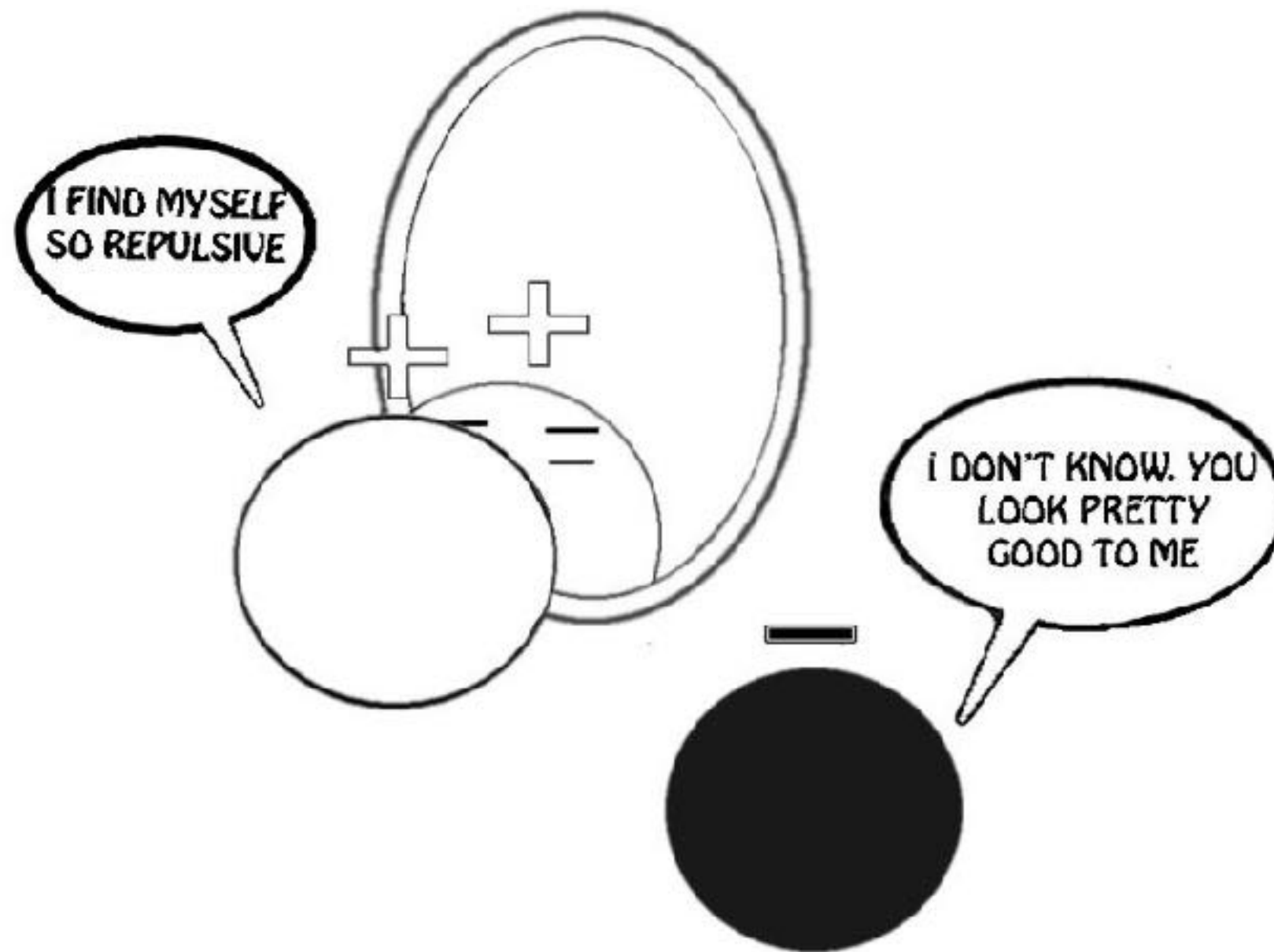
BASICALLY...

- Atoms need full valence (outer) shells (usually 8 electrons) to be considered stable
- Two atoms can share electrons in “bonds”
- Unshared electrons are called “lone pairs”
- These bonds and lone pairs have negative charge

THE MAIN POINT

- In chemistry, opposites attract and like things repel
- So, these negatively charged bonds and lone pairs want to get as far away from each other as possible!
- AKA... Electron Repulsion
- Hence, Valence Shell Electron Pair Repulsion Theory can predict the shape (aka geometry) of molecules



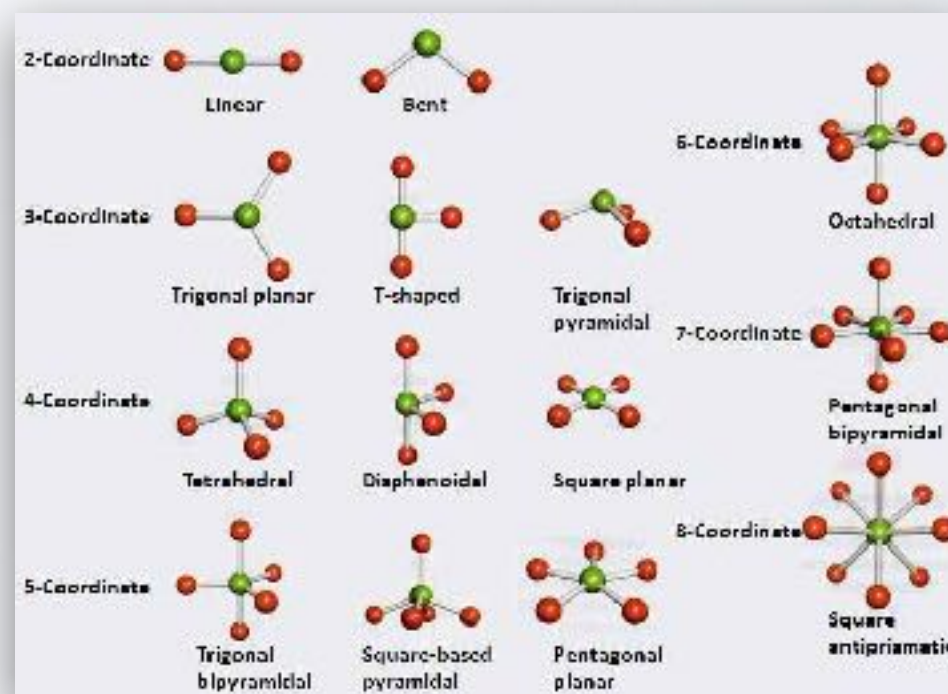


EXISTING APPROACHES

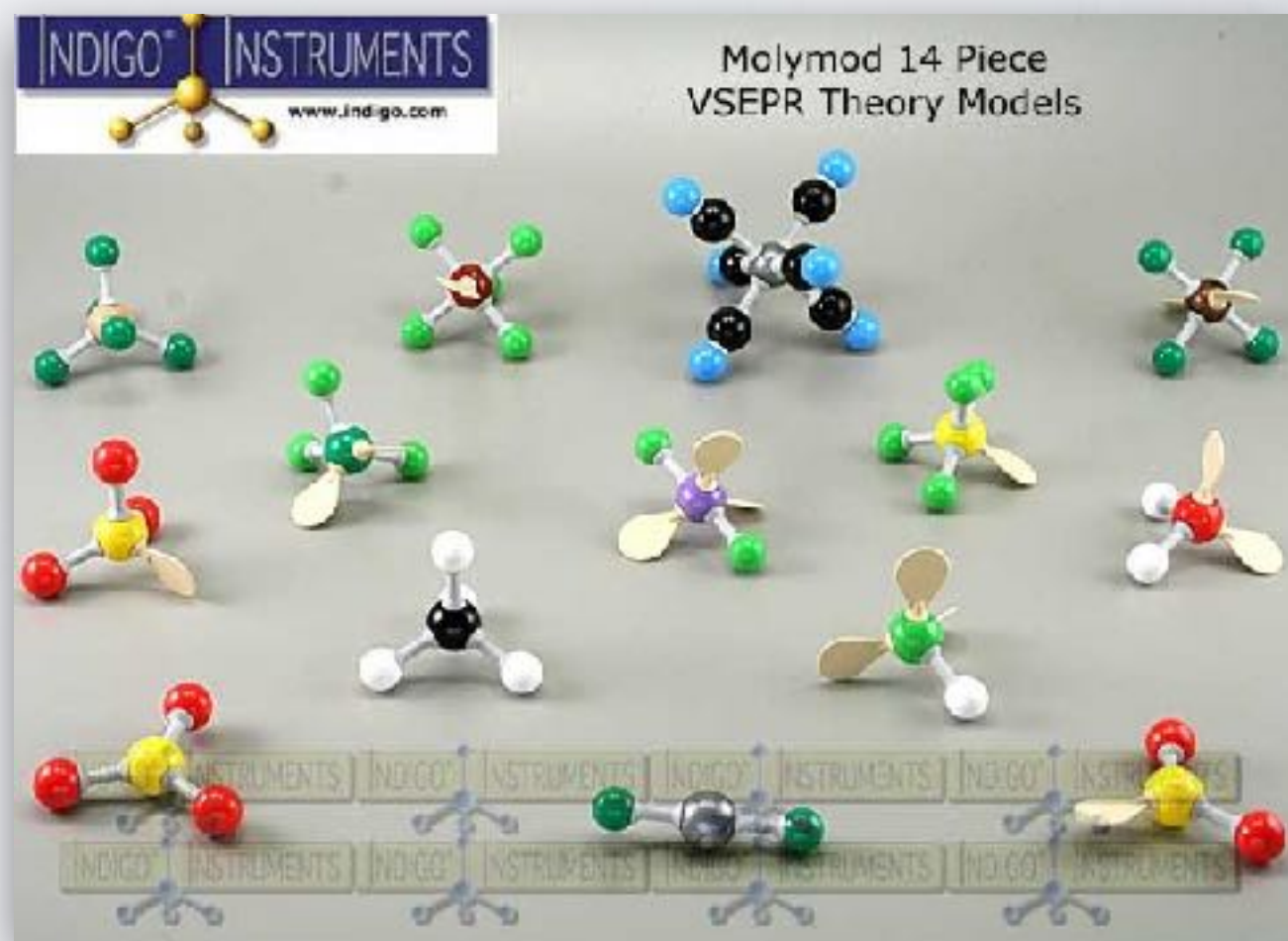
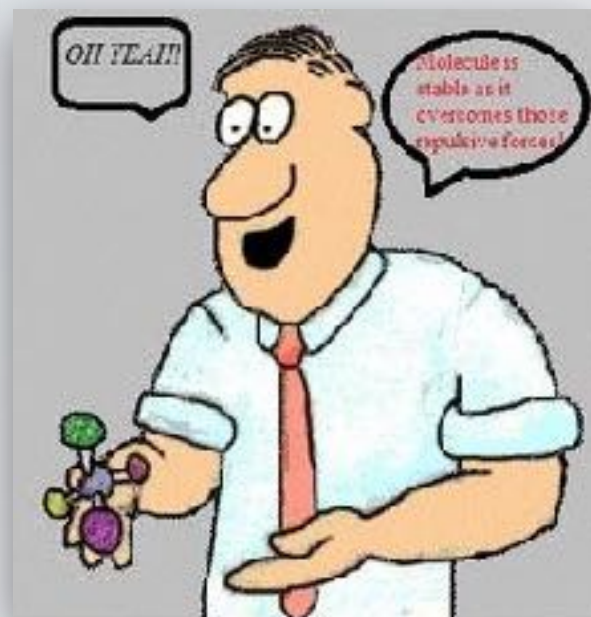
TYPICAL CLASSROOM METHODS

2	2	0	Linear	Linear	$\text{BeF}_2, \text{CO}_2, \text{F}-\text{B}-\text{F}$	
sp	1	1	Linear	Linear	$\text{CO}, \text{N}_2, \text{H}-\text{C}\equiv\text{N}$	
3	3	0	Trigonal planar	Trigonal planar	$\text{BF}_3, \text{CO}_3^{2-}$	
sp^2	2	1	Bent	Bent	O_3, SO_2	
	1	2	Linear	Linear	$\text{O}_2, \text{C}\equiv\text{O}$	
4	4	0	Tetrahedral	Tetrahedral	$\text{CH}_4, \text{SO}_4^{2-}$	
sp^3	3	1	Trigonal pyramidal	Trigonal pyramidal	$\text{NH}_3, \text{H}_3\text{O}^+$	
	2	2	Bent	Bent	$\text{H}_2\text{O}, \text{ClO}_2^-$	
	1	3	Linear	Linear	HCl, OH^-	
5	5	0	Trigonal bipyramidal	Trigonal bipyramidal	PF_5	
sp^3d	4	1	Sawtooth	Sawtooth	$\text{SF}_4, \text{TeCl}_4, \text{IF}_5$	
	3	2	T-shaped	T-shaped	ClF_3	
	2	3	Linear	Linear	$\text{I}_3^-, \text{XeF}_2$	
6	6	0	Octahedral	Octahedral	$\text{SF}_6, \text{PF}_6^-, \text{PtCl}_6^{2-}$	
sp^3d^2	5	1	Square pyramidal	Square pyramidal	$\text{BrF}_5, \text{ClF}_5$	

Handout



PHYSICAL MANIPULATIVES: *INDIGO INSTRUMENTS*



SMALLAB



LIMITATIONS

- Standard: Difficult to translate 3D concepts from 2D means
- Indigo: Lack of digital feedback or meaningful connections
- SMALLab: Limited space in classroom or lab settings

DESIGN PRINCIPLES

DESIGN PRINCIPLES

- Embodied learning & body syntonicity
- Partner collaboration
- Constructionist learning
- Learning narratives

BENEFITS OF EMBODIED LEARNING

Is the technology much better than physical representations of the molecules? What are the benefits here?

- “learner’s **physical actions** enacted in synchrony with **computational feedback** can influence the developmental trajectories of learners’ cognitive schemes”
 - Charoenying, Gaysinksy, Ryokai (2012): The Choreography of Conceptual Development in Computer Supported Instructional Environments

BENEFITS OF EMBODIED LEARNING

- “Recent research supports the idea that **body activity** can be an important catalyst for generating learning, and new technologies are being developed that use natural **human physicality and gesture as input.**”
 - Lindgren (2013): Emboldened by Embodiment: Six Precepts for Research on Embodied Learning and Mixed Reality

BENEFITS OF COLLABORATION

Would they be able to play together to accomplish a task with a lab partner?

- Piaget: “among peers, **difference in perspective** leads to cognitive conflict, which, in turn, leads to cognitive restructuring and **growth**”
- Cassell (2004): Towards a model of technology and literacy development: Story listening systems

BENEFITS OF CONSTRUCTIONIST LEARNING

- Piaget: “**in order to know** objects, **the subject must act** upon them: he must displace, connect, combine, take apart, and reassemble them”
- Charoenying, Gaysinsky, Ryokai (2012): The Choreography of Conceptual Development in Computer Supported Instructional Environments

BENEFITS OF LEARNING NARRATIVES

- “The best learning experiences, for most people, come when they are actively engaged in designing and creating things, especially things that are **meaningful** to them or around them.”
 - Resnick & Silverman (2005): Some Reflections on Designing Construction Kits for Kids

DESIGN EVOLUTION

- mapEdge: Interactive knowledge maps
- Focus on Embodied Learning
- Added collaboration and narratives based on class discussions and feedback
- ...sChemata!

DESIGN

Introducing... sChemata!

THE TECHNOLOGY

- 3D hand gesture recognition screens
- Storytelling adventure framework to encourage mnemonics and meaningful connections
- Two person mission-like labs
- Quizzes, aka “Quests” to evaluate understanding and give feedback (*‘using this as an assessment tool would be fascinating’*)

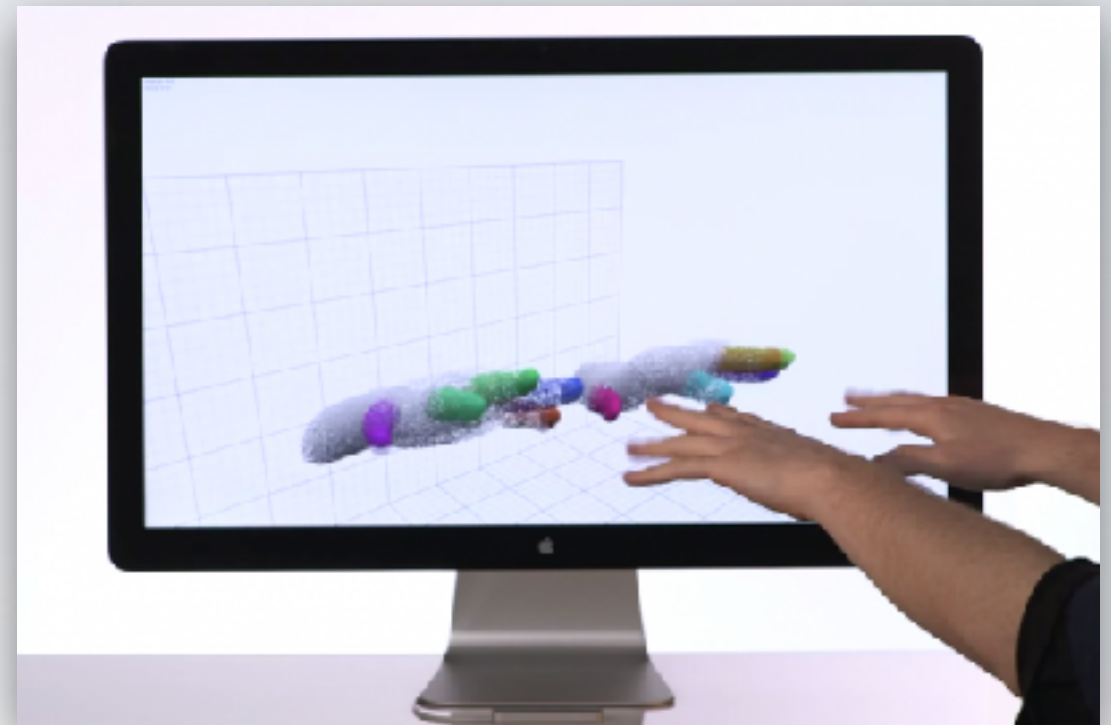
THE SETUP

- High school AP Chemistry class (or college general chemistry lab) (*‘Is the chemistry classroom setting the ideal setting or just what you think is the realistic space it will be used in?’*)
- Two students per sChemata screen work together
- Teacher can walk around for tips and guidance (*‘Does the teacher need to demonstrate? Wouldn’t the kids be potentially more engaged if they tried first?’*)



THE DEVICE

- Large, thin screen
- No keyboard or mouse required
- Easy to learn, intuitive hand gestures with tutorial (i.e. make a fist to grab, release to let go; point twice to select option; grab and rotate both hands to rotate molecule)



3D HAND GESTURE RECOGNITION



WHY NOT JUST MAKE AN APP?

It sounds like this could be an app with 3D images.

- Encourages greater engagement in the learning process through physical interaction in a digital feedback environment, i.e. body syntonicity!
- Removes distractions of touchscreen, keyboard, mouse, or other browser window tabs, texts, notifications, etc. on a multi-use device

WHY NOT JUST MAKE AN APP?

- “finger touch leaves fingerprints on the screen, some on-screen content is occluded by the fingers, and for large displays some display regions might be inaccessible. For 3d applications the biggest disadvantage is the limitation to two dimensions. Here, 3d gesture tracking allows a much **more direct interaction** with 3d objects.”

- Caputo, Denker, Dums, Umlauf (2012): 3D Hand Gesture Recognition Based on Sensor Fusion of Commodity Hardware

SCENARIOS

VSEPR Tool Kit

Grab, drag, and drop into
Valence Land!

Single Bond

Double Bond

Triple Bond

Electron Pair



Add Atoms



Random

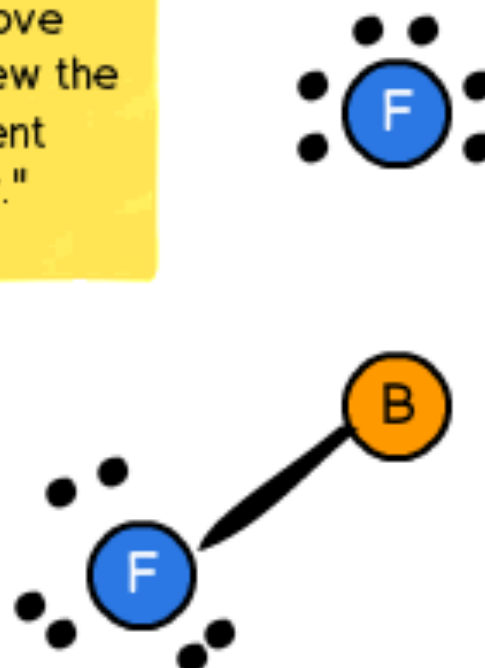
Pop-up of Periodic
Table to select
from.

Generates random
molecule or geometry
for student to build.

Valence Land

Experiment with building your own molecules. Valence Land needs your help!

Note: Molecule and tools are
3D. Students can also use hand
gestures to rotate and move
the molecule around to view the
3D geometry from different
perspectives in their "city."



? Need a Hint?

Fluorine is the "Electron Hog!" Looks like you have two unstable Fluorines right now causing chaos in the city. How might our Sheriff Boron save the day to establish peace over Valence Land again?

Atoms with unstable
valence shells should
be vibrating to show
unrest.

Test it!

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Add Atoms

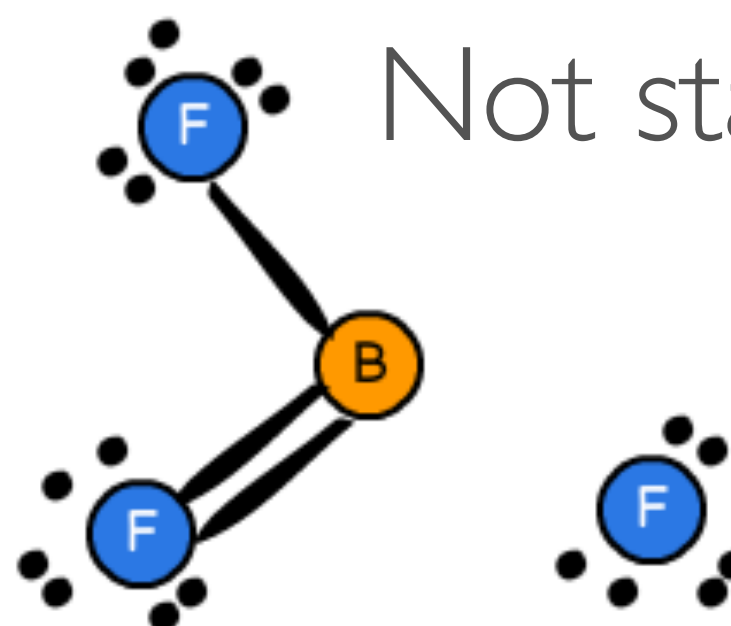


Random

Valence Land

Experiment with building your own molecules. Valence Land needs your help!

[? Need a Hint?](#)



Uh oh! Looks like you have two unstable Fluorines. Keep experimenting!

Try Again

VSEPR Tool Kit

Grab, drag, and drop into
Valence Land!

Single Bond

Double Bond

Triple Bond

Electron Pair



Add Atoms



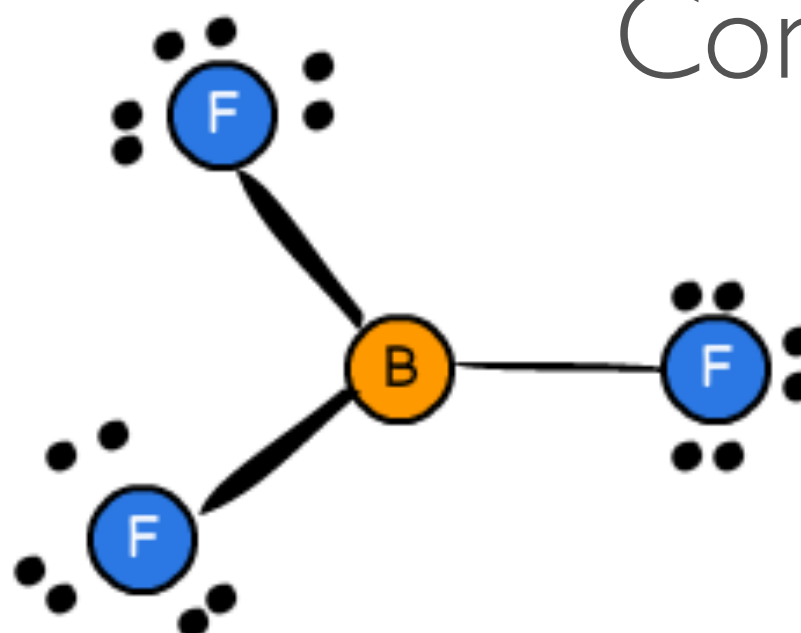
Random

Valence Land

Experiment with building your own molecules. Valence Land needs your help!

Congrats!

Animation of
atoms
congratulating
player!



Success! You made a trigonal planar molecular geometry! Now all your Fluorines are in bonds and Valence Land is peaceful once again. 😊

Keep Experimenting

Try a Quest?

ISN'T THIS SUPPOSED TO BE 3D?

- Yes!
- The molecules should look a bit more like this:
- Perfect example of an octahedral molecular geometry
- Images are nice, but now, imagine you could rotate, spin it around, and build one yourself!



EVALUATION

VSEPR Tool Kit

Grab, drag, and drop into
Valence Land!

Single Bond

Double Bond

Triple Bond

Electron Pair



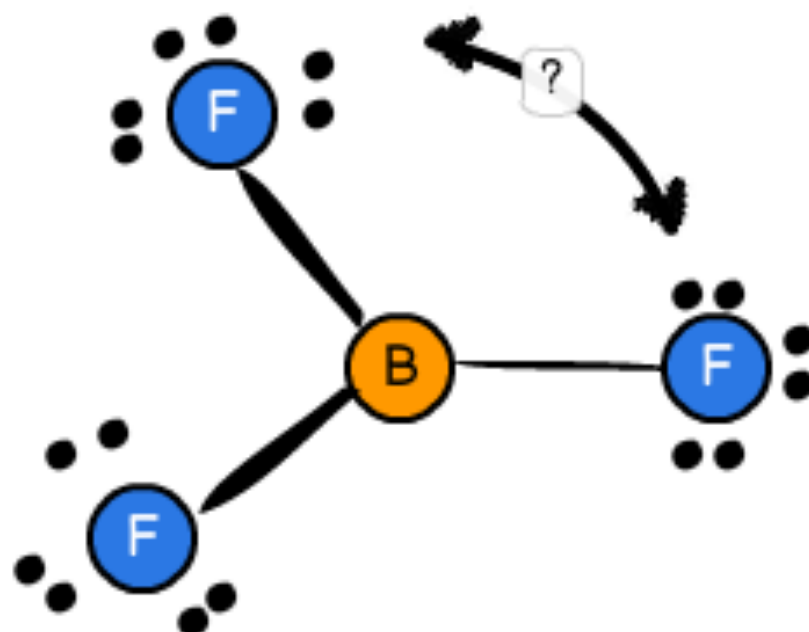
Add Atoms



Random

Quest

Discuss with your partner or try quizzing each other!



1) Rotate the BF₃ molecule around to see it from different perspectives. What is the bond angle between each Fluorine atom?

? Hint: Imagine you and your partner are arch enemy villains in bonds like electrons. That sounds like a negative situation! Which angle allows you two to get as far away from each other as possible?



Success!!

Next

120 degrees



FUTURE DIRECTIONS

Can you integrate physical manipulatives? Can you integrate real chemicals and their reactions? Can you use the lab equipment as input? i.e. does turning on a bunsen burner add “heat” to the virtual environment?

Might not be super relevant to VSEPR
theory, but...

MORE CHEMISTRY TOPICS!

- Physical input could be helpful for learning...
 - chemical reactions
 - gas laws
 - acids & bases
 - thermodynamics
 - organic chemistry

OTHER SUBJECTS

- 3D Gesture Technology could be helpful for learning...
 - physics
 - biology
 - math
- sChemata could also potentially be simplified for younger ages, or developed for more complex concepts for research experimentation

THE ORIGINAL PLAN...

- Eventually, this could be one small module of a massive interactive 3D knowledge map! Fancy that, mapEdge returns!

THANK YOU

Questions? Comments? Feedback?