

## de Broglie's Waves & Uncertainty principle

1. electron →

$$\text{a) } P = \frac{h\nu}{\lambda} = \frac{6.63 \times 10^{-34}}{2 \times 10^{-10}} \text{ kg m/s}$$

$$= 3.31 \times 10^{-24} \text{ kg m/s}$$

This momentum is non-rel. for electron.

$$E_k = \frac{p^2}{2m} = \frac{(3.31 \times 10^{-24})^2}{2 \times 9.1 \times 10^{-31}} \text{ Joule}$$

$$= 0.602 \times 10^{-17} \text{ J}$$

$$= 37.63 \text{ eV}$$

and rest mass energy  $= mc^2$   
 $= 0.51 \text{ MeV}$   
 $= 510000 \text{ eV}$

$$\therefore \text{b) total energy} = 51003 \text{ eV}$$

c) Electron kinetic energy =  $38.63 \text{ eV}$   
 Photon " " " =  $6216 \text{ eV}$ .

2. Let particle velocity =  $v_{\text{particle}}$   
 auf electron "  $\Rightarrow v_{\text{electron}}$

$\Theta, T, P \rightarrow v_{\text{particle}} = 3 v_{\text{ekstrem}}$

again,  $\frac{\lambda_{particle}}{\lambda_{electron}} = 1.813 \times 10^{-4}$

$$\Rightarrow \frac{h}{m_{\text{particle}} \times v_{\text{particle}}} \times \frac{m_{\text{electron}} \times v_{\text{electron}}}{h} = 1.1813 \times 10^{-4}$$

$$\rightarrow \underline{m_{elektron}} = 1.1813 \times 10^{-34}$$

$$\Rightarrow m_{\text{partikel}} = \frac{9.1 \times 10^{-31}}{1.1813 \times 3 \times 10^9} = 2.57 \times 10^{-27} \text{ kg}$$

∴ The particle is neutron.

3. for a relativistic speed  $\rightarrow$

$$\lambda = \frac{hc}{\sqrt{E_k(E_k + 2m_0c^2)}} \quad ; \quad E_k = eV \quad (V = \text{potential diff.})$$

$$\lambda = \frac{hc}{\sqrt{eV(eV + 2m_0c^2)}} = \frac{hc}{\sqrt{eV \cdot 2m_0c^2 \left(1 + \frac{eV}{2m_0c^2}\right)}}$$

$$= \frac{h}{\sqrt{2m_0eV}} \left(1 + \frac{eV}{2m_0c^2}\right)^{-1/2}$$

4. B.T.P  $\rightarrow$

$$\lambda_{\text{electron}} = \lambda_{\text{photon}} = \frac{hc}{E_k} = \frac{6.63 \times 10^{-34} \times 3 \times 10^8}{100 \times 10^3 \times 1.6 \times 10^{-19}} \text{ m}$$

$$\therefore \lambda_{\text{electron}} = 12.43 \times 10^{-12} \text{ m}$$

$$\text{Using, } p = \frac{h}{\lambda_{\text{electron}}} = \frac{6.63 \times 10^{-34}}{12.43 \times 10^{-12}} = 0.533 \times 10^{-22} \text{ kg m/s.}$$

This momentum is non-relativistic momentum for electron.

$$\therefore E_k = \frac{p^2}{2m} = \frac{(0.533 \times 10^{-22})^2}{2 \times 9.1 \times 10^{-31}} \text{ Joule} = \frac{0.0156 \times 10^{-13}}{1.6 \times 10^{-19}} \text{ eV}$$

$$E_k = 9.75 \times 10^3 \text{ eV} = 9.75 \text{ keV.}$$

5.  $(E_k)_{\text{electron}} = (E_k)_{\text{proton}}$

$$\Rightarrow \frac{p_e^2}{2m_e} = \frac{p_{\text{proton}}^2}{2m_p}$$

$$\Rightarrow \frac{p_e^2}{p_{\text{proton}}^2} = \frac{m_e}{m_p} \Rightarrow \frac{\lambda_p}{\lambda_e} = \sqrt{\frac{m_e}{m_p}} = \sqrt{\frac{9.1 \times 10^{-31}}{1.67 \times 10^{-27}}} = 2.33 \times 10^{-2}$$

$$a) \therefore \frac{\lambda_e}{\lambda_p} = \frac{100}{2.33} \approx 42.9 \quad \checkmark$$

$$b) \frac{\lambda_e}{\lambda_p} = 42.9$$

$$\Rightarrow \frac{m_p v_p}{m_e v_e} = 42.9 \quad (p = \frac{h}{\lambda})$$

$$\Rightarrow \frac{v_p}{v_e} = \frac{42.9 \times 9.1 \times 10^{-31}}{1.67 \times 10^{-27}} = 233.8 \times 10^{-4}$$

$$\therefore \frac{v_e}{v_p} = 42.8 \Rightarrow$$

Particle velocity = Group velocity

$$\therefore \frac{(v_g)_{\text{electron}}}{(v_g)_{\text{proton}}} = 42.8 \quad \checkmark$$

$$c) \frac{(v_e)_{\text{phase}}}{(v_p)_{\text{phase}}} = \frac{v_{e/2}}{v_{p/2}} = \frac{v_e}{v_p} = 42.8 \quad (\text{non-rel. case})$$

6. B.T.P  $\rightarrow v_e = v_p$

$$\therefore \frac{\lambda_e}{\lambda_p} = \frac{m_p v_p}{m_e v_e} = \frac{1.67 \times 10^{-27}}{9.1 \times 10^{-31}} = 1835$$

Similarly,  $\frac{(v_g)_{\text{electron}}}{(v_g)_{\text{proton}}} = \frac{v_e}{v_p} = 1$

and  $\frac{(v_e)_{\text{phase}}}{(v_p)_{\text{phase}}} = 1$

7.  $\lambda_e = \lambda_p$

$$\Rightarrow \frac{h}{m_e v_e} = \frac{h}{m_p v_p} \Rightarrow m_e v_e = m_p v_p$$

Now,  $\frac{E_e}{E_p} = \frac{\frac{1}{2} m_e v_e^2}{\frac{1}{2} m_p v_p^2} = \frac{v_e}{v_p} = \frac{m_p}{m_e} = 1835$

$$\therefore E_e > E_p$$

8. electron  $\rightarrow E_k = 1 \text{ MeV}$  (~~is~~ greater than rest mass energy)

$$\begin{aligned} \therefore \lambda &= \frac{hc}{\sqrt{E_k(E_k + 2m_0c^2)}} \\ &= \frac{6.63 \times 10^{-34} \times 3 \times 10^8}{\sqrt{1 \times 1.51 (\text{MeV})^2}} \\ &= \frac{6.63 \times 10^{-34} \times 3 \times 10^8}{1.23 \text{ MeV}} = \frac{6.63 \times 10^{-34} \times 3 \times 10^8}{1.23 \times 10^6 \times 1.6 \times 10^{-19}} = 10.11 \times 10^{-13} \text{ m} \end{aligned}$$

$[m_0c^2 = 0.51 \text{ MeV}]$



neutron  $\rightarrow$

$$E_k = 1 \text{ MeV} \quad (\text{less than rest mass energy})$$

$$\lambda = \frac{h}{\sqrt{2mE_k}}$$

$$= \frac{6.63 \times 10^{-34}}{\sqrt{2 \times 1.68 \times 10^{-27} \times 1 \times 10^6 \times 1.6 \times 10^{-19}}} \text{ m}$$

$$= \frac{6.63}{2.32} \times 10^{-14} \text{ m} = 2.86 \times 10^{-14} \text{ m}$$

Photon

$$E_k = 1 \text{ MeV}$$

$$\lambda = \frac{hc}{E_k} = \frac{6.63 \times 10^{-34} \times 3 \times 10^8}{1 \times 10^6 \times 1.6 \times 10^{-19}} \text{ m} = 12.43 \times 10^{-13} \text{ m}$$

9.

de-Broglie wave length  $\leftarrow$

$$\lambda_d = \frac{h}{mv}$$

$$= \frac{h \sqrt{1 - v^2/c^2}}{m_0 v}$$

$$\left[ m = \frac{m_0}{\sqrt{1 - v^2/c^2}} \right]$$

$$= \frac{h \sqrt{1 - 1/2}}{m_0 \cdot c/\sqrt{2}}$$

$$= \frac{h}{m_0 c}$$

$$\boxed{\lambda_d = \lambda_c} \quad \text{proof}$$

$\lambda_c = \text{compton wave length}$

10.

$$v_{\text{phase}} = \sqrt{\frac{2\pi s}{\lambda f}}$$

we know,  $v_g = v_p - \lambda \frac{dv_p}{d\lambda}$

$$= v_p - \lambda \times \sqrt{\frac{2\pi s}{s}} \cdot \left(-\frac{1}{2}\right) \frac{1}{\lambda^{3/2}}$$

$$= v_p + \frac{1}{2} \sqrt{\frac{2\pi s}{\lambda s}}$$

$$= v_p + \frac{1}{2} v_p = \frac{3}{2} v_p$$

11.  $v_p = \sqrt{\frac{8\lambda}{2\pi}}$

$$\begin{aligned} \therefore v_g &= v_p - \lambda \frac{dv_p}{d\lambda} \\ &= v_p - \lambda \sqrt{\frac{8}{2\pi}} \cdot \frac{1}{2} \frac{1}{\sqrt{\lambda}} \\ &= v_p - \frac{1}{2} \sqrt{\frac{8\lambda}{2\pi}} \\ &= v_p - \frac{1}{2} v_p = v_p/2 \end{aligned}$$

12.  $v = 0.900c$

$\therefore v_g$  (group velocity)  $= 0.900c$

again,  $v_p = \frac{c^2}{v} = \frac{c^2}{0.9c} = \frac{c}{0.9} = \frac{3 \times 10^8}{0.9}$   
 $= 3.33 \times 10^8 \text{ m/s.}$

13.  $v_g = \frac{d\omega}{dk}$

$$= \frac{d(2\pi\nu)}{d(2\pi/\lambda)} = \frac{d\nu}{d(1/\lambda)}$$

14. a)  $v_{\text{phase}} = \frac{\omega}{k} = \frac{E}{p} = \frac{\sqrt{p^2 c^2 + m_0^2 c^4}}{p} = \sqrt{c^2 + \frac{m_0^2 c^4}{p^2}}$

$$v_{\text{phase}} = c \sqrt{1 + \left(\frac{m_0 c^2}{h}\right)^2 \lambda^2}$$

b)  $\lambda = 1 \times 10^{-13}$

~~$p = \frac{h}{\lambda}$~~

$\Rightarrow p = \frac{6.63 \times 10^{-34}}{1 \times 10^{-13}} = 6.63 \times 10^{-21}$

$$p = \frac{h}{\lambda} = \frac{6.63 \times 10^{-34}}{1 \times 10^{-13}} = 6.63 \times 10^{-21} \text{ kgm/s.}$$

This momentum is relativistic momentum for electron,

$\therefore mv = 6.63 \times 10^{-21}$

~~$\frac{h}{\lambda} = mv$~~

$$\frac{m_0 v}{\sqrt{1-v^2/c^2}} = 6.63 \times 10^{-21}$$

$$\Rightarrow \frac{v}{\sqrt{1-v^2/c^2}} = \frac{6.63 \times 10^{-21}}{9.1 \times 10^{-31}} = 0.73 \times 10^{10}$$

$$\therefore \frac{v^2}{1-v^2/c^2} = 0.53 \times 10^{20}$$

$$\Rightarrow c^2 v^2 = 0.53 \times 10^{20} (c^2 - v^2)$$

$$\Rightarrow v^2 (9 \times 10^{16} + 0.53 \times 10^{20}) = 9 \times 10^{16} \times 0.53 \times 10^{20}$$

$$\Rightarrow v^2 \times 5.309 \times 10^{19} = 4.77 \times 10^{36}$$

$$\Rightarrow v^2 = 0.898 \times 10^{17} = 8.98 \times 10^{16}$$

$$\Rightarrow v = 2.99 \times 10^8 \text{ m/s.}$$

$$\therefore v_g = v = 2.99 \times 10^8 = 0.99 c$$

$$\text{and } v_p = \frac{c^2}{v} = \frac{c^2}{0.99 c} = \frac{c}{0.99} = 1.01 c$$

14.

$$v = 1 \text{ km/s} = 1000 \text{ m/s.}$$

$$\Delta v = 1000 \times \frac{0.05}{100} = 0.5$$

$\therefore$  Using Heisenberg Uncertainty principle —

$$\Delta x \cdot \Delta p \geq \hbar$$

$$\therefore \Delta x \cdot m \cdot \Delta v \geq \hbar$$

$$\Rightarrow \Delta x \geq \frac{\hbar}{m \cdot \Delta v} = \frac{6.63 \times 10^{-34}}{2 \times 10^{-27} \times 0.5} = 0.232 \times 10^{-3} \text{ m.}$$

15.

$$\Delta x = 10^{-9} \text{ m.}$$

$\therefore$  Using H.U.P.

$$\Delta v = \frac{\hbar}{\Delta x \cdot m} = \frac{6.63 \times 10^{-34}}{10^{-9} \times 2 \times 10^{-27} \times 9.1 \times 10^{-31}} = 0.116 \times 10^6 \text{ m/s.}$$

Using H.U.P  $\rightarrow$

$$\Delta E \Delta t \geq \hbar$$

$$\Delta E \geq \frac{6.63 \times 10^{-34}}{2\pi \times 10^{-8}}$$

$$\therefore (\Delta E)_{\min} = \frac{6.63 \times 10^{-34}}{2\pi \times 10^{-8}}$$

$$= 1.056 \times 10^{-26} \text{ Joule}$$

$$= 6.59 \times 10^{-8} \text{ eV.}$$

17.

$$\Delta x = \lambda$$

$$\therefore \Delta x = \frac{h}{mv}$$

Using H.U.P  $\rightarrow$

$$\Delta x \Delta p \geq \hbar$$

$$\therefore \Delta x \cdot m \Delta v \geq \hbar$$

$$\Rightarrow \Delta v \geq \frac{\hbar}{2\pi \times \Delta x \cdot m}$$

$$\geq \frac{\hbar \cdot mv}{2\pi \times \hbar \cdot m}$$

$$\boxed{\Delta v \geq \frac{v}{2\pi}}$$

if we consider,  $\Delta x \Delta p \geq \hbar$

$$\lim \boxed{\Delta v \geq v.}$$

$$\Delta v = 300 \times \frac{0.01}{10} = 0.03 \therefore \Delta x = \frac{\hbar}{2\pi \times m \Delta v} = \frac{6.63 \times 10^{-34}}{2\pi \times 50 \times 10^{-3} \times 0.03}$$

$$= 7.07 \times 10^{-32} \text{ m,}$$

$$\Delta L = 2\hbar \times \frac{5}{10} = \frac{\hbar}{10}$$

Using H.U.P  $\rightarrow$

$$\Delta L \Delta \theta \geq \hbar \Rightarrow \Delta \theta = 10 \text{ radian,}$$

$$\therefore (\Delta \theta)_{\min} = 2\pi \text{ radian,}$$

$\therefore$  cannot be specified at all.