



$$T_0 = \text{ConvertTemp} \left(\text{C}, \text{K} \mid T(\text{Water}, P = 80 \text{ [kPa]}, x = 0) \right)$$

$$T_{\text{inf}} = \text{ConvertTemp} \left(\text{C}, \text{K} \mid 20 \text{ [C]} \right) \text{ Ambient Temp}$$

$$h = 10 \text{ [W/m}^2\text{-K]} \text{ heat transfer coeff}$$

$$D = 7 \text{ [cm]} \cdot \left| 0.01 \cdot \frac{\text{m}}{\text{cm}} \right|$$

$$L = 7 \text{ [cm]} \cdot \left| 0.01 \cdot \frac{\text{m}}{\text{cm}} \right|$$

$$A = \pi \cdot D \cdot L \text{ area}$$

$$\rho = \rho(\text{Water}, P = 80 \text{ [kPa]}, x = 0)$$

$$m = \rho \cdot \frac{\pi}{4} \cdot D^2 \cdot L$$

$$C_V = C_v(\text{Water}, P = 80 \text{ [kPa]}, x = 0) \cdot \left| 1000 \cdot \frac{\text{J}}{\text{kJ}} \right| \text{ Heat Cap}$$

$$\frac{dT}{dt} = \frac{h \cdot A}{m \cdot C_V} \cdot (T_0 - T_{\text{inf}}) \text{ solving eq}$$

SOLUTION

Unit Settings: SI C kPa kJ mass deg

$$A = 0.01539 \text{ [m}^2\text{]}$$

$$dTdt = 0.01147 \text{ [K/s]} \{0.01147 \text{ [C/s]}\}$$

$$m = 0.2594 \text{ [kg]}$$

$$T_{inf} = 293.2 \text{ [K]}$$

$$C_v = 3802 \text{ [J/kg-K]}$$

$$h = 10 \text{ [W/m}^2\text{-K]}$$

$$\rho = 962.9 \text{ [kg/m}^3\text{]}$$

$$D = 0.07 \text{ [m]}$$

$$L = 0.07 \text{ [m]}$$

$$T_0 = 366.6 \text{ [K]}$$

No unit problems were detected.