

Solution
Mathematics

$$\begin{aligned}
1. \quad & \int \frac{5x^8 + 7x^6}{(x^2 + 1 + 2x^7)^2} dx \\
&= \int \frac{5x^8 + 7x^6}{x^{14} \left(\frac{1}{x^5} + \frac{1}{x^7} + 2\right)^2} dx \\
&= \int \frac{\frac{5}{x^6} + \frac{7}{x^8}}{\left(\frac{1}{x^5} + \frac{1}{x^7} + 2\right)^2} dx \\
&\text{Put } \frac{1}{x^5} + \frac{1}{x^7} + 2 = t \\
&\Rightarrow \left(-\frac{5}{x^6} - \frac{7}{x^8}\right) dx = dt \\
&\therefore - \int \frac{dt}{t^2} = \frac{1}{t} + c \\
&= \frac{x^7}{x^2 + 1 + 2x^7} + c
\end{aligned}$$

2. $D = 121 - 24\alpha$

for rotational roots D must be perfect square.

$$\therefore \alpha = 3, 4, 5$$

\therefore number of values of $\alpha = 3$

$$\begin{aligned}
3. \quad & \left(\frac{1-t^6}{1-t}\right)^3 = (1-t^6)^3(1-t)^{-3} \\
&= (1-t^{18}-3t^6+3t^{12})(1-t)^{-3} \\
&\therefore \text{Co-efficient of } t^4 = 1 \times {}^{3+4-1}C_4 = {}^6C_4 = 15
\end{aligned}$$

4. Given $\sum_{i=1}^n (x_i^2 + 2x_i + 1) = 9n \dots\dots\dots (i)$

$$\sum_{i=1}^n (x_i^2 - 2x_i + 1) = 5n \dots\dots\dots (ii)$$

For (i) + (ii) we get

$$\sum_{i=1}^n (2x_i^2 + 2) = 14n$$

$$\Rightarrow \sum_{i=1}^n x_i^2 = 6n$$

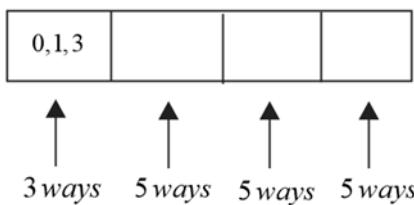
$$\sum_{i=1}^n x_i = n$$

$$\therefore \text{Variance, } r^2 = E(x^2) - (E(x))^2$$

$$= \frac{6n}{n} - \left(\frac{n}{n}\right)^2 = 6 - 1 = 5$$

$$\therefore \text{Standard deviation, } \sigma = \sqrt{5}$$

5.



\therefore total number formed less than 7000 = $3 \times 5 \times 5 \times 5 = 375$

\therefore total natural number formed less than 7000 = $375 - 1 = 374$

(number zero has been excluded)

6. Let A is event that 2nd drawn ball is red.

$$\therefore P(A) = P(R) \times P(A/R) + P(G) \times P(A/G)$$

$$\begin{aligned} & \frac{5}{7} \times \frac{4}{7} + \frac{2}{7} \times \frac{6}{7} \\ &= \frac{32}{49} \end{aligned}$$

$$7. t_n = \frac{3n(1^2 + 2^2 + 3^2 + \dots + n^2)}{2n+1} = \frac{3n \times n(n+1)(2n+1)}{6(2n+1)}$$

$$= \frac{1}{2} (n^3 + n^2)$$

$$\therefore S_{15} = \sum_{n=1}^{15} tn = \sum_{n=1}^{15} \frac{1}{2} (n^3 + n^2)$$

$$= \frac{1}{2} \times \left[\left(\frac{15 \times 16}{2} \right)^2 + \frac{15 \times 16 \times 31}{6} \right]$$

$$= 7200 + 620$$

$$= 7820$$

8. $Z_0 = w$ or w^2

When $Z_0 = w$

$$Z = 3 + 6i \times w^{81} - 3i \times w^{93}$$

$$= 3 + 6i - 3i = 3 + 3i$$

When $Z_0 = w^2$

$$Z = 3 + 6i \times (w^2)^{81} - 3i \times (w^2)^{93}$$

$$= 3 + 6i - 3i = 3 + 3i$$

$$\therefore \arg(z) = \tan^{-1} \left(\frac{3}{3} \right) = \frac{\pi}{4}$$

9. Since $(\vec{a} + \vec{b})$ is perpendicular to \vec{c} , hence $(\vec{a} + \vec{b}) \cdot \vec{c} = 0$

$$\Rightarrow \vec{a} \cdot \vec{c} + \vec{b} \cdot \vec{c} = 0$$

$$\Rightarrow 5 + 1 + 2 + 5b_1 + b_2 + 2 = 0$$

$$\Rightarrow 5b_1 + b_2 = -10 \dots\dots(i)$$

$$\text{Given, } \frac{\vec{b} \cdot \vec{a}}{|\vec{a}|} = |\vec{a}|$$

$$\Rightarrow b_1 + b_2 + 2 = 4$$

$$\Rightarrow b_1 + b_2 = 2 \dots\dots(ii)$$

Solving (i) & (ii), $b_1 = -3, b_2 = 5$

$$\therefore |\vec{b}| = \sqrt{9 + 25 + 2} = 6$$

10. $\int_0^{\pi/3} \frac{\tan x}{\sqrt{2k \sec x}} dx$

$$= \frac{1}{\sqrt{2k}} \int_0^{\pi/3} \frac{\sin x}{\sqrt{\cos x}} dx$$

Put $t = \cos x$

$$dt = -\sin x dx, \text{ also } x = 0, t = 1; x = \frac{\pi}{3}, t = \frac{1}{2}$$

$$\therefore \frac{1}{\sqrt{2k}} \int_1^{1/2} \frac{-dt}{\sqrt{t}}$$

$$= \frac{1}{\sqrt{2k}} \int_{1/2}^1 t^{-\frac{1}{2}} dt = \frac{1}{\sqrt{2k}} \times 2 \left[t^{\frac{1}{2}} \right]_{1/2}^1$$

$$= \frac{\sqrt{2}}{\sqrt{k}} \left(1 - \frac{1}{\sqrt{2}} \right) = \frac{\sqrt{2}-1}{\sqrt{k}}$$

$$\text{Given } \frac{\sqrt{2}-1}{\sqrt{k}} = 1 - \frac{1}{\sqrt{2}} = \frac{\sqrt{2}-1}{\sqrt{2}}$$

$$\therefore k = 2$$

11. $\sin x - \sin 2x + \sin 3x = 0$

$$\Rightarrow \sin x + \sin 3x - \sin 2x = 0$$

$$\Rightarrow 2 \sin 2x \cdot \cos x - \sin 2x = 0$$

$$\Rightarrow \sin 2x = 0 \text{ or } \cos x = \frac{1}{2}$$

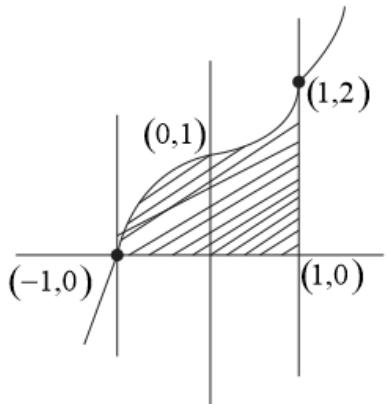
$$\Rightarrow 2x = 0 \text{ or } x = \frac{\pi}{3}$$

$$\Rightarrow x = 0 \text{ or } x = \frac{\pi}{3}$$

$$\therefore \text{number of solutions} = 2$$

12. Area enclosed

$$= \int_{-1}^0 (-x^2 + 1) dx + \int_0^1 (x^2 + 1) dx$$



$$\begin{aligned} &= \left[-\frac{x^3}{3} + x \right]_{-1}^0 + \left[\frac{x^3}{3} + x \right]_0^1 \\ &= 2 \end{aligned}$$

13. $\lim_{x \rightarrow 0^-} \frac{x(|x|) + |x| \sin|x|}{|x|}$

$$\lim_{x \rightarrow 0^-} \frac{x(-1-x) \sin(-1)}{-x}$$

$$\lim_{x \rightarrow 0^-} -(1+x) \sin 1$$

$$-\sin 1$$

14. Given relation is $f(xy) = f(x) \cdot f(y) \quad \forall x, y \in R \dots \text{(i)}$

On putting $x = y = 0$, we get $f(0) = f^2(0) \Rightarrow f(0) = 0, 1$ but $f(0) \neq 0$

$$\Rightarrow f(0) = 1,$$

Now if we put $y = 0$ in (i), we get

$$f(x) = 1$$

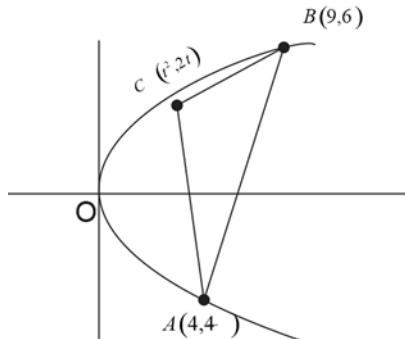
$$\text{Hence } \frac{dy}{dx} = 1 \Rightarrow y = x + c$$

$$\Rightarrow y = x + \frac{1}{2} \text{ (since } y(0) = \frac{1}{2})$$

$$\Rightarrow y\left(\frac{1}{4}\right) + y\left(\frac{3}{4}\right) = \left(\frac{1}{4} + \frac{1}{2}\right) + \left(\frac{3}{4} + \frac{1}{2}\right) = 2.$$

15. Let the coordinates of C be $(t^2, 2t)$

$$\text{Area of } \Delta ABC = \frac{1}{2} \begin{vmatrix} t^2 & 2t & 1 \\ 9 & 6 & 1 \\ 4 & -4 & 1 \end{vmatrix}$$



$$\Delta = 5(6 + t - t^2)$$

$$\frac{d\Delta}{dt} = 1 - 2t = 0 \Rightarrow t = \frac{1}{2}$$

$$\Rightarrow c\left(\frac{1}{4}, 1\right)$$

16. As we know, if two circles intersect each other, then $|r_1 - r_2| < c_1c_2 < r_1 + r_2 \dots\dots\dots (i)$

Now for the first circle $c_1(8, 10)$ and $r_1 = r$

For the second circle $c_2(4, 7)$ and $r_2 = 6$

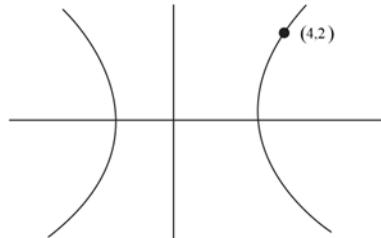
From (i)

$$|r - 6| < 5 < r + 6$$

$$\Rightarrow r \in (1, 11)$$

17. From the given information, equation of hyperbola can be taken as

$$\frac{x^2}{4} - \frac{y^2}{b^2} = 1, a = 2$$



Since it passes through $(4, 2)$,

$$\text{Hence } b^2 = \frac{4}{3}$$

$$\text{Now } e = \sqrt{1 + \frac{b^2}{a^2}} = \sqrt{1 + \frac{1}{3}} = \frac{2}{\sqrt{3}}$$

18. Given $x = 3 \tan t$ and $y = 3 \sec t$

$$\Rightarrow \frac{dx}{dt} = 3 \sec^2 t \text{ and } \frac{dy}{dt} = 3 \sec t \tan t$$

$$\Rightarrow \frac{dy}{dx} = \sin t$$

$$\Rightarrow \frac{d^2y}{dx^2} = \cos t \cdot \frac{dt}{dx} = \cos t \cdot \frac{\cos^2 t}{3} = \frac{\cos^3 t}{3}$$

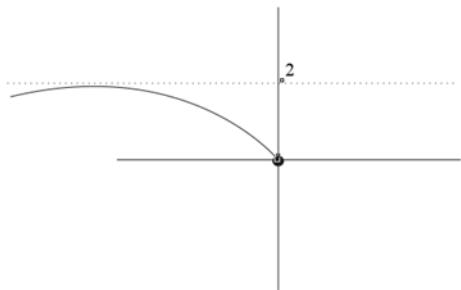
$$\text{At } t = \frac{\pi}{4}$$

$$\frac{d^2y}{dx^2} = \frac{1}{6\sqrt{2}}$$

19. Given function $f(x) = \frac{2x}{x-1}$ and domain of $f(x)$ is $(-\infty, 0)$

$$\text{Now } f'(x) = \frac{-2}{(x-1)^2} < 0$$

$\Rightarrow f$ is always decreasing in the given domain.



Clears given function is one-one.

Since codomain $(-\infty, \infty) \neq$ Range $(0, 2)$, hence f is into.

20. Let the first terms of A.P. be A and common difference be d , then

$$a = A + 6d$$

$$b = A = 10d$$

$$c = A + 12d$$

Given a, b, c and in G. P. $\Rightarrow b^2 = ac$

$$\Rightarrow (A + 10d)^2 = (A + 6d)(A + 12d)$$

$$\Rightarrow A^2 + 100d^2 + 20Ad = A^2 + 18Ad + 72d^2$$

$$\Rightarrow A = -14d$$

$$\text{Now } \frac{a}{c} = \frac{A+6d}{A+12d} = \frac{-14d+6d}{-14d+12d} = \frac{-8d}{-2d} = 4$$

21. Since one vertex of given triangle is o (origin) and other two A & B lies on coordinate axes, hence triangle must be a right angled triangle.

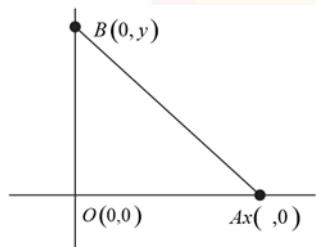
$$\text{Area} = \frac{1}{2}x \cdot y, \quad x, y \in I$$

As per given condition

$$\frac{1}{2}x \cdot y = 50 \Rightarrow x \cdot y = 100$$

Possible ordered pairs of (x, y) when Δ formed in 1st quadrant are 9 i. e. $(1, 100), (2, 50), (4, 25), (5, 20), (10, 10), (20, 5), (25, 4), (50, 2), (100, 1)$.

Δ can be formed in any of the quadrants, hence total number of cases $9 \times 4 = 36$.



22. Given roots of $x^2 - mx + 4 = 0$ are distinct and lies in $(1, 5)$, Following conditions must be true

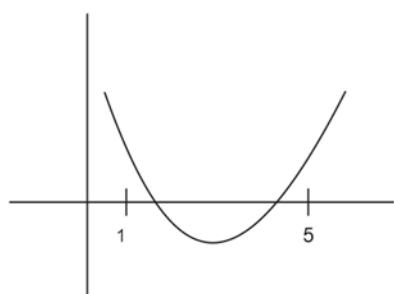
(i) $D > 0 \Rightarrow M \in (-\infty, -4) \cup (4, \infty)$

(ii) $f(1) > 0 \Rightarrow M < 5$

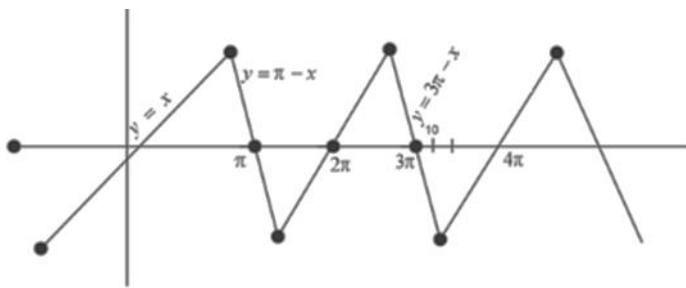
(iii) $f(5) > 0 \Rightarrow M < \frac{29}{5}$

(iv) $1 < -\frac{B}{2A} < 5 \Rightarrow 2 < m < 10$

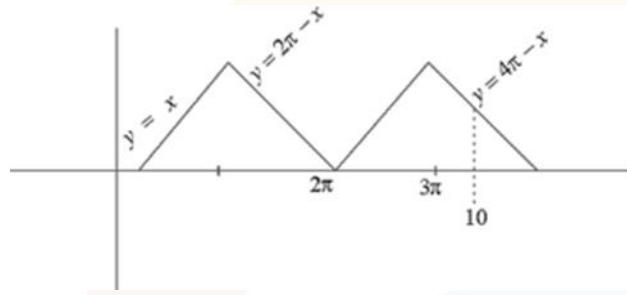
Taking intersection of all conditions, we get $m \in (4, 5)$



23.



From the graph of $y = \sin^{-1}(\sin x)$, it is clear that $\sin^{-1}(\sin 10) = 3\pi - 10$



From the graph of $y = \cos^{-1}(\cos x)$

It is clear that $\cos^{-1}(\cos 10) = 4\pi - 10$

Hence $y - x = (4\pi - 10) - (3\pi - 10) = \pi$

24. Since given matrix A is invertible $\Rightarrow |A| \neq 0$

$$\begin{aligned} \text{Now } |A| &= \begin{vmatrix} e^t & e^{-t}(\sin t - 2 \cos t) & e^{-t}(-2 \sin t - \cos t) \\ e^t & -e^{-t}(2 \sin t + \cos t) & e^{-t}(\sin t - 2 \cos t) \\ e^t & e^{-t} \cos t & e^{-t} \sin t \end{vmatrix} \\ &= e^t \cdot e^{-t} \cdot e^{-t} \begin{vmatrix} 1 & \sin t - 2 \cos t & -2 \sin t - \cos t \\ 1 & 2 \sin t + \cos t & \sin t - 2 \cos t \\ 1 & \cos t & \sin t \end{vmatrix} \\ &= 2e^{-t} \begin{vmatrix} 1 & \sin t - 2 \cos t & -2 \sin t - \cos t \\ 0 & \sin t & -\cos t \\ 1 & \cos t & \sin t \end{vmatrix} \end{aligned}$$

$$R_1 \rightarrow R_1 - R_2$$

$$= 2e^{-t} \begin{vmatrix} 1 & -2 \cos t & -2 \sin t \\ 0 & \sin t & -\cos t \\ 1 & \cos t & \sin t \end{vmatrix}$$

$$R_1 \rightarrow R_1 + 2R_3$$

$$= 2e^{-t} \begin{vmatrix} 3 & 0 & 0 \\ 0 & \sin t & -\cos t \\ 1 & \cos t & \sin t \end{vmatrix}$$

$$|A| = 6e^{-t} \neq 0 \forall t \in \mathbb{R}$$

Hence given matrix is always invertible.

25. DR's of the first line

$$\begin{vmatrix} \hat{i} & \hat{j} & \hat{k} \\ 1 & -a & 0 \\ 0 & c & -1 \end{vmatrix} = a\hat{i} + \hat{j} + c\hat{k}$$

Hence DR's of the first line are $a, 1, c$.

For DR's of the second line

$$\begin{vmatrix} \hat{i} & \hat{j} & \hat{k} \\ 1 & 0 & -a' \\ 0 & 1 & -c' \end{vmatrix} = a'\hat{i} + c'\hat{j} + \hat{k}$$

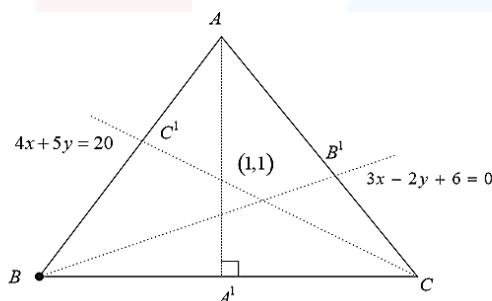
Hence DR's of the 2^{nd} line = $a', c', 1$

Since both the lines are perpendicular,

$$a \cdot a' + c' + c = 0$$

26. To find equation of BC , first we will find coordinates of B and C .

Equation of BB^1



$$y - 1 = \frac{-2}{3}(x - 1)$$

$$\Rightarrow 2x + 3y = 5 \dots\dots(i)$$

Point of intersection of AB and $BB^1 \Rightarrow B \left(\frac{35}{2}, -10 \right)$

Similarly point $c \left(-13, -\frac{33}{2} \right)$

Equation of BC $y + 10 = \frac{13}{61} \left(x - \frac{35}{2} \right)$

27. Given functional inequality is

$$|f(x) - f(y)| \leq 2|x - y|^{3/2}$$

Put $x = y + h$, we get

$$|f(y + h) - f(y)| \leq 2|h|^{3/2}$$

$$\Rightarrow \left| \frac{f(y + h) - f(y)}{h} \right| \leq 2|h|^{1/2}$$

$$\Rightarrow \lim_{h \rightarrow 0} \left| \frac{f(y+h) - f(y)}{h} \right| \leq \lim_{h \rightarrow 0} 2|h|^{1/2}$$

$$\Rightarrow |f'(y)| \leq 0$$

$$\Rightarrow |f'(y)| = 0$$

$$\Rightarrow f'(y) = 0$$

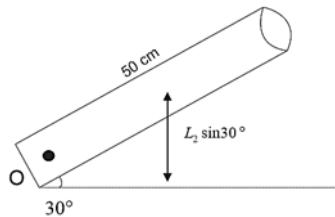
$$\Rightarrow f(y) = c$$

$$\Rightarrow f(y) = 1 \text{ (since } f(0) = 1)$$

$$\text{Now } \int_0^1 f^2(x) dx = \int_0^1 1 dx = 1$$

Physics

1. From conservation of mechanical energy



$$mg \frac{l}{2} \sin 30 = \frac{1}{2} I \omega^2$$

$$mg \frac{l}{2} \sin 30 = \frac{1}{2} \frac{ml^2}{3} \omega^2$$

$$\omega = \sqrt{\frac{3g}{2l}} = \sqrt{\frac{3 \times 10}{2 \times 50 \times 10^{-2}}} = \sqrt{30} \text{ rad/s}$$

2. 75 sec, $\frac{4}{10}$, Application (i) Velocity

$$x = A \cos \omega t$$

$$y = A \sin \omega t$$

$$z = A \omega t$$

$$v_x = -A\omega \sin \omega t$$

$$v_y = A\omega \cos \omega t$$

$$v_z = A\omega$$

$$v = \sqrt{v_x^2 + v_y^2 + v_z^2}$$

$$= \sqrt{(-A\omega \sin \omega t)^2 + (A\omega \cos \omega t)^2 + (A\omega)^2}$$

$$= A\omega\sqrt{2}$$

3. $60 \text{ sec}, \frac{3}{10}$, Application (i) Transformer

(ii) Efficiency of a transformer

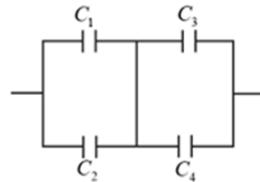
Efficiency $y = 90\%$

$\Rightarrow 0.9 \times \text{power in} = \text{Power out}$

$$0.9 \times 2300 \times 5 = 230 \times I_{\text{out}}$$

$$I_{\text{out}} = 45 \text{ A}$$

4. $75 \text{ sec}, \frac{4}{10}$, Application, (i) Capacitance with dielectric



$$C_1 = \frac{\epsilon_0 \frac{A}{2} K_1}{\frac{d}{2}} = \frac{\epsilon_0 A K_1}{d}$$

$$C_2 = \frac{\epsilon_0 A K_2}{d}$$

$$C_3 = \frac{\epsilon_0 A K_3}{d}$$

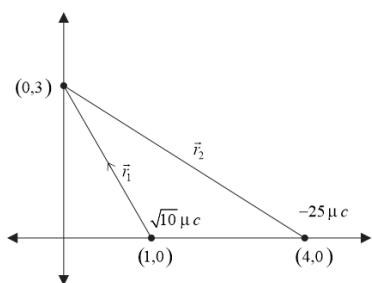
$$C_4 = \frac{\epsilon_0 A K_4}{d}$$

$$C_{\text{eq}} = \frac{(C_1 + C_2)(C_3 + C_4)}{C_1 + C_2 + C_3 + C_4}$$

$$\frac{\epsilon_0 A K_{\text{eq}}}{d} = \frac{\epsilon_0 A}{d} \frac{(K_1 + K_2)(K_3 + K_4)}{K_1 + K_2 + K_3 + K_4}$$

$$K_{\text{eq}} = \frac{(K_1 + K_2)(K_3 + K_4)}{K_1 + K_2 + K_3 + K_4}$$

5. Electric field due to $\sqrt{10} \mu c$



$$\vec{E}_1 = \frac{1}{4\pi \epsilon_0} \frac{\sqrt{10} \times 10^{-6}}{r_1^3} \vec{r}_1$$

$$= \frac{1}{4\pi \epsilon_0} \frac{\sqrt{10} \times 10^{-6}}{(\sqrt{10})^3} (-\hat{i} + 3\hat{j})$$

Similarly electric field due to $-25 \mu C$

$$\vec{E}_2 = \frac{1}{4\pi \epsilon_0} \frac{25 \times 10^{-6}}{(5)^3} (4\hat{i} - 3\hat{j})$$

net electric field

$$\vec{E} = 9 \times 10^9 \times 10^{-6} \left(\frac{-\hat{i}}{10} + \frac{3\hat{j}}{10} + \frac{4}{5}\hat{i} - \frac{3}{5}\hat{j} \right)$$

$$= 9 \times 10^3 \left(\frac{7\hat{i}}{10} - \frac{3\hat{j}}{10} \right)$$

$$= (63\hat{i} - 27\hat{j}) \times 10^2 \frac{N}{C}$$

6. $E_1 + \left(\frac{-GMm}{R} \right) = \frac{-GMm}{R+h}$

$$E_2 = \frac{1}{2} m V^2 \text{ and } \frac{mv^2}{R+h} = \frac{GMm}{(R+h)^2}$$

$$\Rightarrow E_2 = \frac{1}{2} \left(\frac{GMm}{R+h} \right)$$

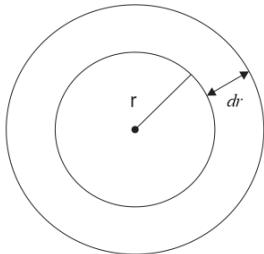
$$E_1 = E_2 \Rightarrow \frac{GMm}{R} - \frac{GMm}{R+h} = \frac{GMm}{2(R+h)}$$

$$\frac{h}{R(h+R)} = \frac{1}{2(R+h)}$$

$$h = \frac{R}{2} = 3.2 \times 10^3 \text{ km}$$

7. $B = B_0 [\sin(3.14 \times 10^7 Ct) + \sin(6.28 \times 10^7 Ct)]$

8. $\rho(r) = \frac{A}{r^2} e^{-\frac{2r}{a}}$



Charge enclosed between r and $r + dr$ is

$$dq = \rho(r) 4\pi r^2 dr$$

To get total charge ' Q' ,

$$Q = \int dq = \int_0^R \frac{A}{r^2} e^{-2r/a} 4\pi r^2 dr$$

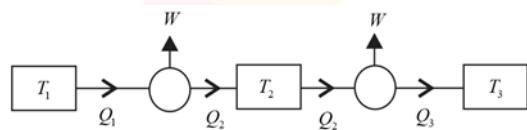
$$= -4\pi A \frac{a}{2} [e^{-2r/a}]^R$$

$$Q = -4\pi A \frac{a}{2} [e^{-2R/a} - 1]$$

$$R = -\ln \left[1 - \frac{Q}{4\pi A a} \right] \frac{a}{2}$$

$$= \frac{a}{2} \ln \left[\frac{2\pi Q A}{2\pi A a - Q} \right]$$

9.



$$W = Q_1 - Q_2 = Q_2 - Q_3$$

$$\frac{Q_1}{Q_2} + \frac{Q_3}{Q_2} = 2$$

$$\frac{T_1}{T_2} + \frac{T_3}{T_2} = 2$$

$$T_2 = \frac{T_1 + T_3}{2} = 500K$$

10. $S = 3t^2 + 5$

$$\text{Velocity } V = 6t + 0$$

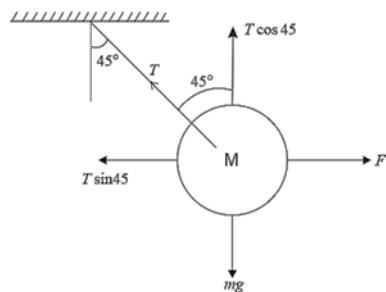
From work energy theorem

Work done = ΔKE

$$= \frac{1}{2} \times 2 \times (6 \times 5)^2 - \frac{1}{2} \times 2 \times (6 \times 0)^2$$

$$= 900 J$$

11.



$$T \cos 45 = mg$$

$$T \sin 45 = F$$

$$\tan 45 = \frac{F}{mg}$$

$$F = mg = 100 N$$

12. Current through R_4 (as well as R_3)

$$\text{is } i_1 = \frac{V}{R} = \frac{5}{500} = 10 mA$$

Potential drop across 400Ω is

$$V_{400} = 18 - 10 \times 10^{-3} \times 600$$

$$= 12 V$$

Current through battery

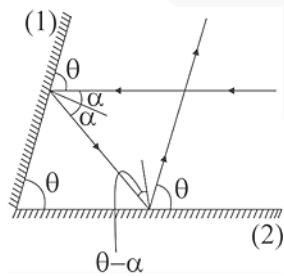
$$I = \frac{12}{400} = 30 mA$$

$$\therefore \text{Current through } R_2 i_2 = 30 - 10$$

$$= 20 mA$$

$$\therefore R_2 = \frac{6}{20 \times 10^{-3}} = 300 \Omega$$

- 13.



$$90 - \alpha + \theta + x$$

$$x = 90 + \alpha - \theta$$

$$90 + \alpha - \theta = \theta$$

$$\text{and } 90 - \alpha = \theta$$

$$180 - \theta = 2\theta$$

$$\theta = 60^\circ$$

14. $\frac{1}{2} a_1 t_0^2 = \frac{1}{2} a_2 (t_0 + t)^2$

$$v_1 = a_1 t_0 \quad v_2 = a_2 (t_0 + t)$$

$$v = v_1 - v_2 = (a_1 - a_2)t_0 - a_2 t$$

$$\sqrt{\frac{a_1}{a_2}} \quad t_0 = t_0 + t$$

$$t_0 = \frac{t}{\sqrt{\frac{a_1}{a_2} - 1}}$$

$$v = (a_1 - a_2) \frac{t}{\sqrt{\frac{a_1}{a_2} - 1}} - a_2 t$$

$$= t \left(\frac{a_1 \sqrt{a_2}}{\sqrt{a_1} - \sqrt{a_2}} - \frac{a_2 \sqrt{a_2}}{\sqrt{a_1} - \sqrt{a_2}} - a_2 \right)$$

$$= t \left(\frac{a_1 \sqrt{a_2} - a_2 \sqrt{a_1}}{\sqrt{a_1} - \sqrt{a_2}} \right)$$

$$= t(\sqrt{a_1} \sqrt{a_2})$$

15. $KE = \frac{1}{2}kx^2 \quad PE = \frac{1}{2}K(A^2 - x^2)$

$$\frac{1}{2}Kx^2 = \frac{1}{2}K(A^2 - x^2)$$

$$x^2 = \frac{A^2}{2}$$

$$x = \frac{A}{\sqrt{2}}$$

16. $T = 2\pi \sqrt{\frac{I}{c}}$

$$f = \frac{1}{2\pi} \sqrt{\frac{C}{I}}$$

$$f_{\text{initial}} = \frac{1}{2\pi} \sqrt{\frac{C_3}{MC^2}}$$

$$f_{\text{final}} = \frac{1}{2\pi} \sqrt{\frac{C}{\left(\frac{MC^2}{3} + 2m\frac{L^2}{4}\right)}}$$

$$0.2 \frac{1}{2\pi} \sqrt{\frac{3C}{ML^2}} = \frac{1}{2\pi} \sqrt{\frac{C}{\frac{MC^2}{3} + \frac{2MC^2}{4}}}$$

$$= 0.04 \frac{3}{M} = \frac{12}{4M + 6M}$$

$$4M = 0.16M + 0.24M$$

$$\frac{m}{M} = \frac{3.84}{0.24} = 16$$

17. $V_{rms} = \sqrt{\frac{3RT}{M}}$

T must be increased 4 times to get V_{rms} doubled.

$$\Delta Q = nC_v\Delta T$$

$$= \frac{15}{28} \times \frac{5}{2} \times 8.314 \times (1200 - 300)$$

$$= 10021.3J = 10 kJ$$

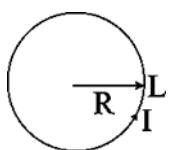
19. Frequency of flute $f = 2 \times \frac{V}{2l}$
 $= 660 \text{ Hz}$

From Doppler effect, observed frequency $f = f_0 \left(\frac{V + V_0}{V} \right)$
 $= 660 \left(\frac{\frac{330 + 10 \times \frac{5}{18}}{330}}{1} \right)$
 $= 665 \text{ Hz}$

20. For circular loop, $r = \frac{L}{2\pi}$

Magnetic field at centre $B_2 = \frac{\mu_0 i}{2r}$

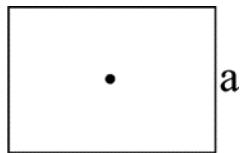
$$B_C = \frac{\mu_0 i}{2L} 2\pi$$



For square, magnetic field

$$B_3 = \frac{4\mu_0 i}{4\pi \left(\frac{2}{8}\right)} \sqrt{2}$$

$$\frac{B_2}{B_3} = \frac{2\pi \pi \left(\frac{2}{8}\right)}{2L \sqrt{2}}$$



$$= \frac{\pi^2}{8\sqrt{2}}$$

21.

Color	Value
--------------	--------------

Green	5
Orange	3
Yellow	4

$$R = 53 \times 10^4 \pm 58$$

22. For particle to move in circle under magnetic field

$$qVB = \frac{mv^2}{R} \quad \dots(i)$$

For particle to move in straight line

In both magnetic field and electric field

$$qVB = 2E$$

$$\Rightarrow V = \frac{E}{B} \quad \dots(ii)$$

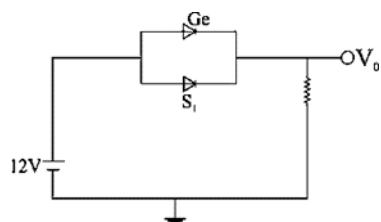
From (i) to (ii)

$$qB = \frac{mv}{R}$$

$$= \frac{m \cdot E}{BR}$$

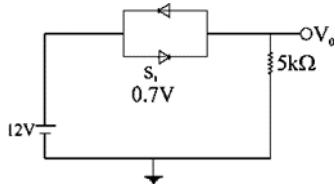
$$\Rightarrow m = \frac{qE^2R}{E} = \frac{1.6 \times 10^{-13} \times 0.5^2 \times 10^{-2}}{0.15} kg = \frac{4}{3} \times 10^{-20} kg$$

23.



$$V_0 = 12 V - V_{Ge}$$

$$= (12 - 0.3)V$$



$$= 11.7 V$$

$$V_0 = 12V - 0.7V$$

$$= 11.3V$$

$$\therefore \Delta V_0 = 11.7 - 11.3 = 0.4V$$

25. Let $[T] = [G]^{\alpha}[L]^{\beta}[C]^{\gamma}$

$$= [M^{-1}L^3T^{-2}]^9 \cdot [ML^2T^{-1}]^{\beta}[LT^{-1}]^{\gamma}$$

$$\Rightarrow -\alpha + \beta = 0$$

$$3 \propto +2\beta + \gamma = 0$$

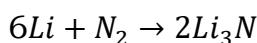
$$-2\alpha - \beta - \gamma = 1$$

On solving we get

$$\alpha = \frac{1}{2}, \beta = \frac{1}{2}, \gamma = \frac{5}{2}$$

Chemistry

1. Among the alkali metals, only lithium is able to form a stable nitride.



Lithium nitride

2. According to molecular orbital theory

$$\text{Bond order} = \frac{(e^- \text{ in bonding molecular orbital}) - (e^- \text{ in Anti bonding Molecular orbital})}{2}$$

Molecular orbital configuration.

$$(A) O_2 \Rightarrow \sigma 1s^2 < \sigma^* 1s^2 < \sigma 2s^2 < \sigma^* 2s^2 < \sigma 2p_x^2 < \pi 2p_y^2 = \pi 2p_z^2 < \sigma^* 2p_y^1 = \pi^* 2p_y^1$$

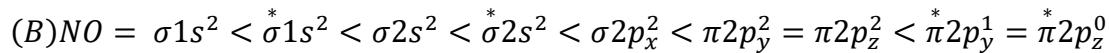
$$\text{B.O.} = \frac{6 - 2}{2} = 2$$

$$O_2^+ \Rightarrow \sigma 1s^2 < \sigma^* 1s^2 < \sigma 2s^2 < \sigma^* 2s^2 < \sigma 2p_x^2 < \pi 2p_y^2 = \pi 2p_z^2 < \pi^* 2p_y^1$$

$$\text{B.O.} = \frac{6 - 3}{2} = 2.5$$

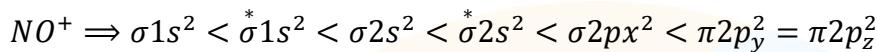
Both are paramagnetic because having unpaired e^-

bond order increase $O_2 \rightarrow O_2^+$



$$\text{B.O.} = \frac{6 - 1}{2} = 2.5$$

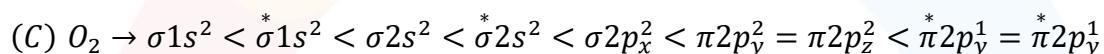
1 Unpaired e^- so paramagnetic



$$\text{B.O.} = \frac{6 - 0}{2} = 3$$

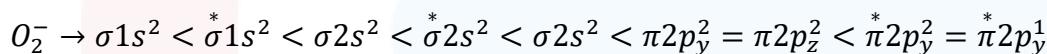
doesn't have unpaired e^- so diamagnetic

$NO \rightarrow NO^+$ {Bond order increases and paramagnetic to diamagnetic character change}



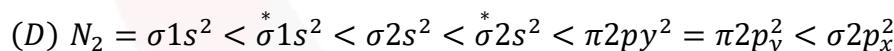
$$\text{B.O.} = \frac{6 - 2}{2} = 2$$

Having 2 - unpaired e^- so paramagnetic



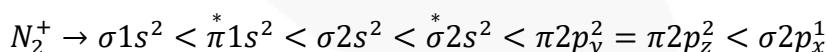
$$\text{B.O.} = \frac{6 - 3}{2} = 1.5$$

Having 1- unpaired e^- so paramagnetic bond order decrease



$$\text{B.O.} = \frac{6 - 0}{2} = 3$$

No unpaired e^- so diamagnetic



$$\text{B.O.} = \frac{5 - 0}{2} = 2.5$$

Having one unpaired e^- so paramagnetic

Bond angle increases N_2 to N_2^+

3.

$\lambda_{\text{emission}}$ Blue < Green < Red

$$L_1 > L_2 > L_3$$

Δ_{emission} $L_1 > L_2 > L_3$

$\Delta_{\text{Absorption}}$ $L_1 > L_2 > L_3$

Strength of ligand $\propto \Delta_{\text{Absorption}}$

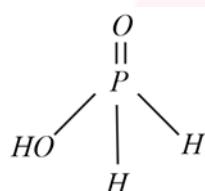
4.

Elements	<i>V</i>	<i>Cu</i>	<i>Zn</i>	<i>Fe</i>
Enthalpy of atomization	515 KJ/mol	339 KJ/mol	216 KJ/mol	416 KJ/mol

5. According to Hardy's -schulz rule, greater the valency of the active ion or flocculating ion greater will be its coagulating power. The cation having more +ve charge more will coagulating power.

Hence Al^{+3} having maximum coagulating power

6.



The hydrogen atoms in OH bond are ionizable and are acidic whereas the $P - H$ bonds have reducing property. There are two such $P - H$ bonds in Hypophosphours acid.

7. Crystal structure is *FCC* so number of atom per unit cell (N) = 4

$$\text{Edge length} = x \cdot A^\circ = x \cdot 10^{-8} \text{ cm}^3$$

$$\begin{aligned} \text{Density } \frac{zm}{N_A a^3} &= \frac{4 \times 63.5}{(6.023 \times 10^{23})(x \times 10^{-8})^3} \text{ gm/cm}^3 \\ &= \frac{4 \times 63.5}{(6.023) \times x^3 \times 10^{-1}} = \frac{421.716}{x^3} \text{ gm/cm}^3 \end{aligned}$$

8. $\Delta G = \Delta G^\circ + RT \ln Q$

At equilibrium $\Delta G = 0, Q = k$

$$0 = \Delta G^\circ + RT \ln K$$

$$\Delta G^\circ = -RT \ln K \quad (\because \Delta G^\circ = -nFE^\circ)$$

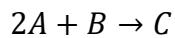
$$\text{So } -nFE^\circ = -RT \ln k$$

$$'n' \text{ factor for the above cell} = 2$$

$$\text{So } -2 \times 96500 \times 2 = -8 \times 300 \ln k$$

$$\ln k = 160$$

$$k = e^{160}$$

9. $2A$ 

$$\text{Initial } r_1 = k[A]^x[B]^y = 0.3 \text{ m/sec} \quad \dots(\text{i})$$

$$r_2 = k[2A]^x[2B]^y = 2.4 \quad \dots(\text{ii})$$

$$r_3 = k[2A]^x[B]^y = 0.6 \quad \dots(\text{iii})$$

From (i) \div (iii)

$$\left(\frac{1}{2}\right)^x (1)^y = \frac{0.3}{0.6}$$

$$\left(\frac{1}{2}\right)^x = \frac{1}{2}$$

$$x = 1$$

(ii) \div (iii)

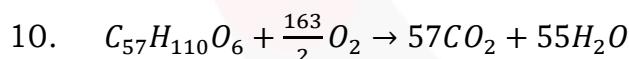
$$\frac{r_2}{r_3} = \frac{2.4}{0.6} = (2)^y$$

$$4 = (2)^y$$

$$y = 2$$

The order of reaction w.r.t. $A = 1$ and w.r.t. $B = 2$

Total order of reaction = $1 + 2 = 3$



Molecular weight of $C_{57}H_{110}O_6 = 890$

$$\text{Moles of } C_{57}H_{110}O_6 = \frac{\text{weight(gm)}}{\text{Molecular weight}}$$

$$= \frac{445}{890} = \frac{1}{2}$$

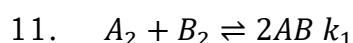
According to stoichiometry

$\because 1 \text{ mole of } C_{57}H_{110}O_6$ gives $= 55 \text{ mole of } O_2$

$\therefore \frac{1}{2} \text{ mole of } C_{57}H_{110}O_6$ gives $= 55 \times \frac{1}{2} \text{ mole of } H_2O$

Wt. of H_2O = mole of $H_2O \times M$ Wt. of H_2O

$$= \frac{1}{2} \times 55 \times 18 = 495 \text{ gm}$$



$$k_1 = \frac{[AB]^2}{[A_2][B_2]}$$



$$k_2 = \frac{[A_2]^3[B_2]^3}{[AB]^6}$$

$$\text{So, } \frac{1}{k_1^3} = k_2$$

12. Cl^- ions is cause of permanent hardness & HCO_3^- is cause of temporary Hardness.
13. Strong field ligands have more *CFSE* value while weak field ligand have less *CFSE* value.

CN^- is strongest ligand among all so $[CO(CN)_6]^{3-}$ has maximum *CFSE* & value.

14. Molecular weight of ethylene glycol

$$(C_2H_6O_2) = 62$$

$$\text{Moles of ethylene glycol} = \frac{Wt}{M.Wt}$$

$$= \frac{62}{62} = 1$$

$$\Delta T_f = i \times k_f \times m$$

$i = 1$ for ethylene glycol

$$\Delta T_f = 1 \times 8.6 \times \frac{1}{\frac{250}{1000}}$$

$$\Delta T_f = 1.86 \times 4 = 7.44$$

As $\Delta T_f = 10$ it implies that some amount of water has frozen and this has led a greater depression of freezing point

$$10 = 1.86 \times \frac{\frac{62}{62}}{\frac{Wt \text{ of } H_2O}{1000}}$$

$$W_{H_2O} = 186$$

So weight of water freeze = $250 - 186$

$$= 64 \text{ gm}$$

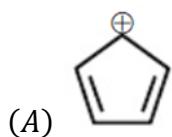
15. $\Delta G = \Delta H - T\Delta S$

For reaction feasibility $\Delta G = -ve$

So Zn can be extracted by ZnO

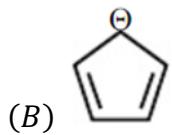
By using *Al* at $500^\circ C$

16.



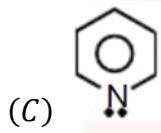
Cyclic conjugation and by Huckel rule = $4\pi e^-$

So Anti aromatic



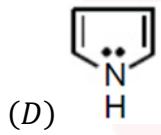
Cyclic conjugation and by Huckel rule = $6\pi e^-$

So aromatic



Cyclic conjugation 1p of $-N$ is not involved in delocalization

Having = $6\pi e^-$ Aromatic

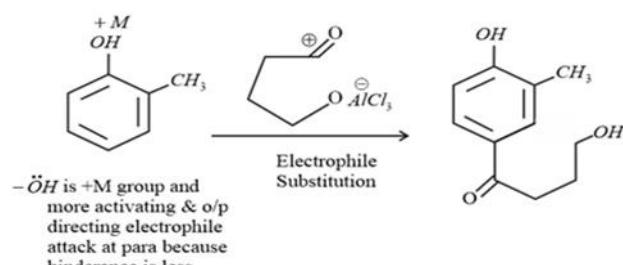
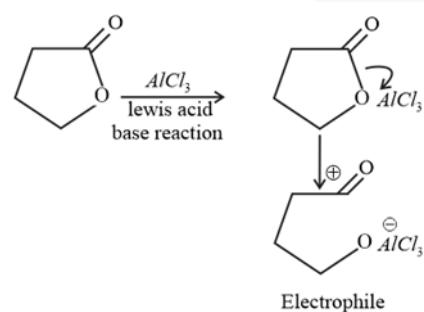


Cyclic conjugation

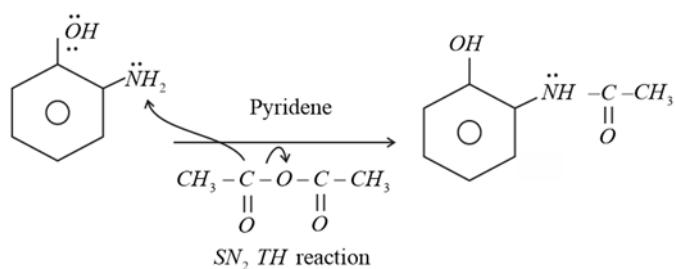
Lone pair involved in Aromatically

Having = $6\pi e^-$ Aromatic

17.



18.



$\ddot{\text{N}}\text{H}_2 > \ddot{\text{O}}\text{H}$ is the order of nucleophilicity

19. Basic character $\propto +M \propto +H \propto +I$

In case of polar protic solvent two factors decide the basic character according to $+I$ effect of $-CH_3$

$3^\circ > 2^\circ > 1^\circ$ Amine

& According to H -bonding the basic character is $1^\circ > 2 > 3^\circ$

& the resultant basic character is determined by taking the above two factors into account.

So order will be $2^\circ > 1^\circ > 3^\circ$

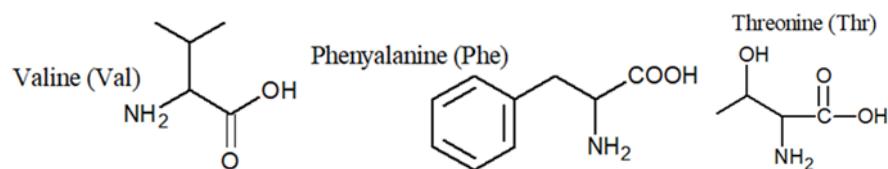
In aromatic & Aliphatic: Aliphatic Amines are more basic than aromatic because the lone pair of N is involved in delocalization.

So basic character is in the order $2 > 1^\circ > 3^\circ >$ Aromatic Amine

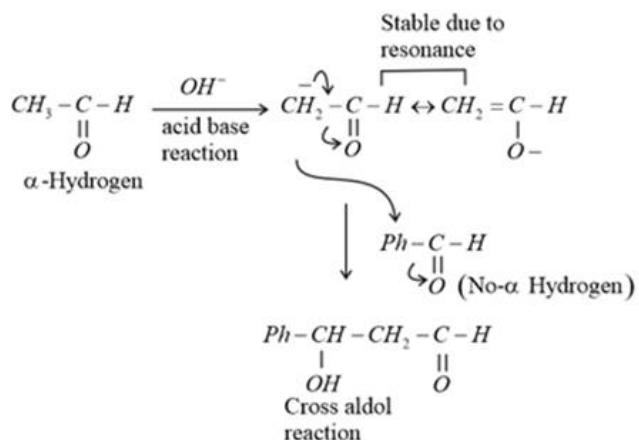
So (iii) > (i) > (iv) > (ii)

20. Normal clean rain has a pH value of around 5.6, which is slightly acidic.

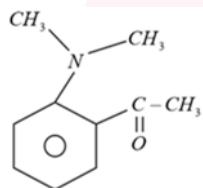
21.


22. Methemoglobinemia disease occurs due to NO_3^- , when its concentration is higher than 50 ppm in water.

23.



24.



Ketones with a methyl group gives positive iodoform test as well as 2,4

DNP: Test which is an identification for carbonyl groups. Azo dye: For Azo dye H should be present with $-N$, or there should be a primary and secondary amines

25.

