

SAFETY DATA SHEET

Ferric Sulfate 60%

NFPA diamond and HMIS ratings

for this product may be found in

section 16 of this Safety Data

1. Product and Company Identification

Product Name Ferric Sulfate

Synonyms Iron (III Sulfate), Iron Persulfate

MSDS Number D24740

Product Use Water and wastewater treatment

Company Identification Aquachem of America Inc.

PO Box 129

Little Chute, WI 54140

Telephone Aquachem of America Inc. – 920-687-5238

CHEMTREC - 800.424.9300

2. Hazards Identification

Form Liquid solution
Color Reddish brown
Odor Slight odor

OSHA/HCS Status Material is considered hazardous by the OSHA Hazard Communication

Standard (29 CFR 1910.1200). Corrosive to metals (Category 1) Acute toxicity, oral (Category 4)

Skin corrosion/irritation (Category 4)
Serious eye damage (Category 1)

Pictogram

Danger

Signal Word

GHS Classification

Hazard Statement(s)

H290 May be corrosive to metals.
H302 Harmful if swallowed.
H315 Causes skin irritation.

H318 Causes serious eye damage.

Precautionary Statement(s)

P234 Keep only in original container.
P264 Wash skin thoroughly after handling.

P270 Do not eat, drink, or smoke when using this product.

P280 Wear protective gloves/ protective clothing/ eye protection/ face protection.

P301 + P312 + P330 IF SWALLOWED: Call a POISON CENTER or doctor/ physician if you feel unwell.

Rinse mouth.



P302 + P352 IF ON SKIN: Wash with plenty of soap and water.

P305 + P351 + P338 IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if

present and easy to do. Continue rinsing. Immediately call a POISON CENTER or

doctor/ physician.

P332 + P313 If skin irritation occurs: Get medical advice/ attention.

P362 + P364 Take off contaminated clothing and wash it before reuse.

P391 Absorb spillage to prevent material damage.

P501 Dispose of contents/ container to an approved waste disposal plant.

Potential Acute Health Effects

P+310

Inhalation May cause irritation to the respiratory tract.

Ingestion Harmful is swallowed. Can irritate or burn digestive tract.

Skin May cause irritation.

Eyes Can cause severe irritation or burns.

See section 11 for more detailed information on health effects and symptoms

3. Composition/Information on Ingredients

 Ingredient Name
 CAS Number
 WT %

 Ferric Sulfate
 10028-22-5
 60.0

 As Iron (FE)
 12.0 - 12.5

 Sulfuric Acid
 7664-93-9
 < 0.5</td>

4. First Aid Measures

Eye Contact Immediately flush eyes with water for 15 minutes. Get medical attention immediately.

Skin Contact Flush with plenty of water for at least 15 minutes while removing contaminated

clothing and shoes. Get medical attention if symptoms develop.

Inhalation If inhaled, remove to fresh air. If not breathing, give artificial respiration. If breathing is

difficult, give oxygen. Get medical attention.

Ingestion Do not induce vomiting. Drink large amounts of water. Get medical attention immediately.

Protection of First Aid

Personnel

No action shall be taken involving any personal risk or without suitable training. It may be dangerous to the person providing aid to give mouth-to-mouth resuscitation. Wash contaminated clothing thoroughly with water before removing it, or wear gloves. If it is suspected that dust, vapor, mist, or gas are still present, the rescuer should wear an appropriate mask or self-contained breathing apparatus.

5. Fire-fighting Measures

Flammability of the Product Not flammable Flash Point (Method) Not applicable

Auto Ignition Temperature Not applicable



Extinguishing Media

Suitable Water spray, foam, carbon dioxide, or dry chemical. Not Suitable Not available

Special Fire-fighting
Procedures & Hazards

Use water spray or foam in large fires. Wear self-contained breathing apparatus.

Unusual Fire & Explosion Hazards

Small Spill

Dangerous and irritating sulfur dioxide fumes may be present in fire involving this

substance.

6. Accidental Release Measures

Personal Precautions Environmental Precautions Large Spill Avoid contact with skin, eyes, and clothing. Avoid breathing vapors or mists.

Prevent run-off onto public land and waterways.

Dike with inert absorbent and mop or pump into clean containers.

Absorb small spills in clay or other inert and neutralize residue with soda ash or

sodium bicarbonate.

7. Handling and Storage

Handling Keep container tightly closed when not in use. Avoid contact with skin, eyes, and clothing.

Avoid breathing vapors or mists. Remove contaminated clothing and wash thoroughly

after handling.

Storage Keep storage container tightly closed. Store in a cool, dry, well-ventilated area. Isolate from

incompatible substances. Store and ship in plastic or rubber-lined containers.

8. Exposure Controls/Personal Protection

Ingredient NameACGIH TLVOSHA PELFerric SulfateNot availableNot availableSulfuric Acid0.2 mg/m³TWA1 mg/m³TWA

Engineering Measures Hygiene Measures Use local exhaust to keep airborne concentrations below the permissible exposure limits. Handle in accordance with good industrial hygiene and safety practice. Wash hands

before breaks and at the end of workday.

Respiratory Where risk assessment shows air-purifying respirators are appropriate, use a full-

face respirator with multi-purpose combination (US) or type ABEK (EN 14387) respirator cartridges as a backup to engineering controls. If the respirator is the sole

means of protection, use a full-face supplied air respirator.

Eyes and Face Wear safety glasses with side shields (or goggles) and a face shield.

Skin Wear appropriate personal protective clothing to prevent skin contact. If prolonged or

repeated contact is anticipated, all clothing should be impervious to liquid.



9. Physical and Chemical Properties

Appearance Reddish brown solution

Odor Slight odor pH 0.1 – 1.50
Water Solubility Very soluble Vapor Density (air = 1) Not determined Evaporation rate (butyl acetate = 1) Not determined

Boiling Point (°F) 220-235 °F (104-113 °C)

Freezing Point (°F) -50 °F (-46 °C)

Specific Gravity $(H_20 = 1 @ 70 °F)$ 1.580

Vapor Pressure Not applicable Volatile Organic (VOC) Content Not applicable

10. Stability and Reactivity

Stable: X Unstable: Hazardous Polymerization: Occurs: Does Not Occur: X

Conditions to Avoid None known

Materials to Avoid Corrosive to cast iron, bronze, copper and its alloys, galvanized steel

Decomposition Products When heated to decomposition, toxic sulfur dioxide, sulfur trioxide, and iron oxide

fumes are produced.

11. Toxicological Information

Eye Can cause severe irritation or burns.

Ferric Sulfate No data available Sulfuric Acid No data available

Dermal May cause irritation or burns.

Ferric Sulfate LD50 (intraperitoneal, mouse) – 168 mg/kg

Sulfuric Acid No data available

Inhalation May cause irritation to the respiratory tract.

Ferric Sulfate No data available

Sulfuric Acid LC50 (rat) $-510 \text{ mg/m}^3 - 2 \text{ hr.}$

Oral Harmful is swallowed. Can irritate or burn digestive tract.

Ferric Sulfate No data available

LD50 (rat) - 2140 mg/kg

Potential Chronic Health Effects

Sulfuric Acid

Chronic Effects Erosion of teeth, mouth inflammation, lesions of the skin, tracheo-bronchitis,

conjunctivitis, and gastritis.

Carcinogenicity IARC and NTP have classified "strong inorganic acid mists containing sulfuric

acid" as a known human carcinogen.

MutagenicityNo data availableTeratogenicityNo data availableDevelopmental EffectsNo data availableFertility EffectsNo data available



12. Ecological Information

Biodegradability **Ecotoxicity**

No information available

Toxicity to fish:

Ferric Sulfate (LC50 – Gambusia affinis – 37.2 mg/l - 96 h) Sulfuric Acid (LC50 - Brachydanio rerio - > 500 mg/l - 96 h)

Toxicity to aquatic invertebrates:

Ferric Sulfate (no data available)

Sulfuric Acid (EC50 - Water flea - 29 mg/l - 24 h)

13. Disposal Considerations

Waste Disposal

Product should be neutralized with alkalis. Empty containers should be taken

for local recycling, recovery, or waste disposal.

RCRA

This material is not listed as a hazardous waste if and when it is discarded. **Note:** If this product is altered, it is the responsibility of the user to determine whether the material meets the criteria for hazardous waste at the time of

disposal.

14.Transportation

The data provided in this section is for information only and may not be specific to your package size or mode of transport. You will need to apply the appropriate regulations to properly classify your shipment for transportation.

US DOT 49 CFR 172.101	Non-bulk Shipments (Drums/Totes)	Bulk Shipments (Tank Trucks/Rail cars)
Proper Shipping Name	Corrosive Liquid, Acidic, Inorganic, N.O.S. (Ferric Sulfate Solution)	Same
Hazard Class	8	Same
Identification Number	UN3264	Same
Packing Group	III	Same
Reportable Quantities	RQ=1000 lbs.	Same
Placards/Labels	Corrosive	Same

15. Regulatory Information

CERCLA / SARA

A spill or release of this material may trigger the emergency release reporting requirements under CERCLA (40 CFR Part 300) and/or SARA Title III (40 CFR Part 355). State or local reporting requirements may differ from federal requirements. Consult counsel for further guidance on your responsibilities under these laws.

Ferric Sulfate - CAS# 10028-22-5 - 1000 lbs. (CERCLA) Sulfuric Acid – CAS# 7664-93-9 – 1000 lbs. (CERCLA)

SARA Title III Section 313

This product is known to contain the following chemicals which are listed in 40 CFR 372.65 as toxic chemicals requiring notification. This information must be included in all SDS's that are copied and distributed for this product.

No component listed



Clean Water Act (CWA) Section 311 The following chemicals are listed under Section 311 as hazardous substance requiring the submission of a National Pollutant Discharge Elimination System (NPDES) permit application to EPA.

No component listed

TSCA – Toxic Substances Control Act All components of this product are listed as "Active" on the Toxic Substances (TSCA) 8(b) Inventory.

RCRA – Resource Conservation and Recovery Act The requirements of the federal hazardous waste regulations do not apply unless the waste fails to pass any of EPA's four tests for determining hazardous wastes. Note: If this product is altered, it is the responsibility of the user to determine whether the material meets the criteria for hazardous waste at the time of disposal.

State Regulations

Massachusetts New Jersey Pennsylvania California RTK Substances: The following components are listed: Ferric Sulfate, Sulfuric Acid RTK Substances: The following components are listed: Ferric Sulfate, Sulfuric Acid RTK Substances: The following components are listed: Ferric Sulfate, Sulfuric Acid Proposition 65: WARNING! The State of California has listed "strong inorganic acid mists containing sulfuric acid" as a cancer-causing agent. Sulfuric Acid – CAS# 7664-93-9



16.Other Information

Date of Issue 3/27/2013 | 8/31/2015 | 8/20/2019 -updated TSCA statement, section 15 (RP) | 8/13/2020- updated address, section 1 (ST) | 8/4/2023-accuracy review (ST)

NFPA



HMIS



Caution: NFPA and HMIS ratings are based on a 0-4 rating scale, with 0 representing minimal hazards or risks, and 4 representing significant hazards or risks. Although these ratings are not required on SDSs under29 CFR 1910.1200, the preparer may choose to provide them.

The customer is responsible for determining the PPE code for this material.

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