APÉNDICE 3

Datos termodinámicos a 1 atm y 25°C*

Sustancias inorgánicas

Sustancia	ΔH_f° (kJ/mol)	ΔG_f° (kJ/mol)	S° (J/K · mol)
$Ag(s)$	0	0	42.7
$Ag^+(ac)$	105.9	77.1	73.9
$AgCl(s)$	−127.0	−109.7	96.1
$AgBr(s)$	−99.5	−95.9	107.1
$AgI(s)$	−62.4	−66.3	114.2
$AgNO_3(s)$	−123.1	−32.2	140.9
$Al(s)$	0	0	28.3
$Al^{3+}(ac)$	−524.7	−481.2	−313.38
$Al_2O_3(s)$	−1669.8	−1576.4	50.99
$As(s)$	0	0	35.15
$AsO_4^{3-}(ac)$	−870.3	−635.97	−144.77
$AsH_3(g)$	171.5		
$H_3AsO_4(s)$	−900.4		
$Au(s)$	0	0	47.7
$Au_2O_3(s)$	80.8	163.2	125.5
$AuCl(s)$	−35.2		
$AuCl_3(s)$	−118.4		
$B(s)$	0	0	6.5
$B_2O_3(s)$	−1263.6	−1184.1	54.0
$H_3BO_3(s)$	−1087.9	−963.16	89.58
$H_3BO_3(ac)$	−1067.8	−963.3	159.8
$Ba(s)$	0	0	66.9
$Ba^{2+}(ac)$	−538.4	−560.66	12.55
$BaO(s)$	−558.2	−528.4	70.3
$BaCl_2(s)$	−860.1	−810.66	125.5
$BaSO_4(s)$	−1464.4	−1353.1	132.2
$BaCO_3(s)$	−1218.8	−1138.9	112.1
$Be(s)$	0	0	9.5
$BeO(s)$	−610.9	−581.58	14.1
$Br_2(l)$	0	0	152.3
$Br^-(ac)$	−120.9	−102.8	80.7
$HBr(g)$	−36.2	−53.2	198.48
C(grafito)	0	0	5.69
C(diamante)	1.90	2.87	2.4
$CO(g)$	−110.5	−137.3	197.9
$CO_2(g)$	−393.5	−394.4	213.6
$CO_2(ac)$	−412.9	−386.2	121.3
$CO_3^{2-}(ac)$	−676.3	−528.1	−53.1

* Las cantidades termodinámicas de los iones están basadas en los estados de referencia: $\Delta H_f^\circ[H^+(ac)] = 0$, $\Delta G_f^\circ[H^+(ac)] = 0$, y $S^\circ[H^+(ac)] = 0$ (véase p. 807).

(*continúa*)

Sustancia	ΔH_f° (kJ/mol)	ΔG_f° (kJ/mol)	S° (J/K · mol)
$HCO_3^-(ac)$	−691.1	−587.1	94.98
$H_2CO_3(ac)$	−699.7	−623.2	187.4
$CS_2(g)$	115.3	65.1	237.8
$CS_2(l)$	87.3	63.6	151.0
$HCN(ac)$	105.4	112.1	128.9
$CN^-(ac)$	151.0	165.69	117.99
$(NH_2)_2CO(s)$	−333.19	−197.15	104.6
$(NH_2)_2CO(ac)$	−319.2	−203.84	173.85
$Ca(s)$	0	0	41.6
$Ca^{2+}(ac)$	−542.96	−553.0	−55.2
$CaO(s)$	−635.6	−604.2	39.8
$Ca(OH)_2(s)$	−986.6	−896.8	83.4
$CaF_2(s)$	−1 214.6	−1 161.9	68.87
$CaCl_2(s)$	−794.96	−750.19	113.8
$CaSO_4(s)$	−1 432.69	−1 320.3	106.69
$CaCO_3(s)$	−1 206.9	−1 128.8	92.9
$Cd(s)$	0	0	51.46
$Cd^{2+}(ac)$	−72.38	−77.7	−61.09
$CdO(s)$	−254.6	−225.06	54.8
$CdCl_2(s)$	−389.1	−342.59	118.4
$CdSO_4(s)$	−926.17	−820.2	137.2
$Cl_2(g)$	0	0	223.0
$Cl^-(ac)$	−167.2	−131.2	56.5
$HCl(g)$	−92.3	−95.27	187.0
$Co(s)$	0	0	28.45
$Co^{2+}(ac)$	−67.36	−51.46	155.2
$CoO(s)$	−239.3	−213.38	43.9
$Cr(s)$	0	0	23.77
$Cr^{2+}(ac)$	−138.9		
$Cr_2O_3(s)$	−1 128.4	−1 046.8	81.17
$CrO_4^{2-}(ac)$	−863.16	−706.26	38.49
$Cr_2O_7^{2-}(ac)$	−1 460.6	−1 257.29	213.8
$Cs(s)$	0	0	82.8
$Cs^+(ac)$	−247.69	−282.0	133.05
$Cu(s)$	0	0	33.3
$Cu^+(ac)$	51.88	50.2	−26.4
$Cu^{2+}(ac)$	64.39	64.98	−99.6
$CuO(s)$	−155.2	−127.2	43.5
$Cu_2O(s)$	−166.69	−146.36	100.8
$CuCl(s)$	−134.7	−118.8	91.6
$CuCl_2(s)$	−205.85	?	?
$CuS(s)$	−48.5	−49.0	66.5
$CuSO_4(s)$	−769.86	−661.9	113.39
$F_2(g)$	0	0	203.34
$F^-(ac)$	−329.1	−276.48	−9.6
$HF(g)$	−271.6	−270.7	173.5
$Fe(s)$	0	0	27.2
$Fe^{2+}(ac)$	−87.86	−84.9	−113.39

(*continúa*)

Sustancia	ΔH_f° (kJ/mol)	ΔG_f° (kJ/mol)	S° (J/K · mol)
$Fe^{3+}(ac)$	−47.7	−10.5	−293.3
$FeO(s)$	−272.0	−255.2	60.8
$Fe_2O_3(s)$	−822.2	−741.0	90.0
$Fe(OH)_2(s)$	−568.19	−483.55	79.5
$Fe(OH)_3(s)$	−824.25	?	?
$H(g)$	218.2	203.2	114.6
$H_2(g)$	0	0	131.0
$H^+(ac)$	0	0	0
$OH^-(ac)$	−229.94	−157.30	−10.5
$H_2O(g)$	−241.8	−228.6	188.7
$H_2O(l)$	−285.8	−237.2	69.9
$H_2O_2(l)$	−187.6	−118.1	?
$Hg(l)$	0	0	77.4
$Hg^{2+}(ac)$		−164.38	
$HgO(s)$	−90.7	−58.5	72.0
$HgCl_2(s)$	−230.1		
$Hg_2Cl_2(s)$	−264.9	−210.66	196.2
$HgS(s)$	−58.16	−48.8	77.8
$HgSO_4(s)$	−704.17		
$Hg_2SO_4(s)$	−741.99	−623.92	200.75
$I_2(s)$	0	0	116.7
$I^-(ac)$	−55.9	−51.67	109.37
$HI(g)$	25.9	1.30	206.3
$K(s)$	0	0	63.6
$K^+(ac)$	−251.2	−282.28	102.5
$KOH(s)$	−425.85		
$KCl(s)$	−435.87	−408.3	82.68
$KClO_3(s)$	−391.20	−289.9	142.97
$KClO_4(s)$	−433.46	−304.18	151.0
$KBr(s)$	−392.17	−379.2	96.4
$KI(s)$	−327.65	−322.29	104.35
$KNO_3(s)$	−492.7	−393.1	132.9
$Li(s)$	0	0	28.0
$Li^+(ac)$	−278.46	−293.8	14.2
$Li_2O(s)$	−595.8	?	?
$LiOH(s)$	−487.2	−443.9	50.2
$Mg(s)$	0	0	32.5
$Mg^{2+}(ac)$	−461.96	−456.0	−117.99
$MgO(s)$	−601.8	−569.6	26.78
$Mg(OH)_2(s)$	−924.66	−833.75	63.1
$MgCl_2(s)$	−641.8	−592.3	89.5
$MgSO_4(s)$	−1 278.2	−1 173.6	91.6
$MgCO_3(s)$	−1 112.9	−1 029.3	65.69
$Mn(s)$	0	0	31.76
$Mn^{2+}(ac)$	−218.8	−223.4	−83.68
$MnO_2(s)$	−520.9	−466.1	53.1
$N_2(g)$	0	0	191.5
$N_3^-(ac)$	245.18	?	?

(*continúa*)

Sustancia	ΔH_f° (kJ/mol)	ΔG_f° (kJ/mol)	S° (J/K · mol)
$NH_3(g)$	−46.3	−16.6	193.0
$NH_4^+(ac)$	−132.80	−79.5	112.8
$NH_4Cl(s)$	−315.39	−203.89	94.56
$NH_3(ac)$	−80.3	−26.5	111.3
$N_2H_4(l)$	50.4		
$NO(g)$	90.4	86.7	210.6
$NO_2(g)$	33.85	51.8	240.46
$N_2O_4(g)$	9.66	98.29	304.3
$N_2O(g)$	81.56	103.6	219.99
$HNO_2(ac)$	−118.8	−53.6	
$HNO_3(l)$	−173.2	−79.9	155.6
$NO_3^-(ac)$	−206.57	−110.5	146.4
$Na(s)$	0	0	51.05
$Na^+(ac)$	−239.66	−261.87	60.25
$Na_2O(s)$	−415.9	−376.56	72.8
$NaCl(s)$	−411.0	−384.0	72.38
$NaI(s)$	−288.0		
$Na_2SO_4(s)$	−1 384.49	−1 266.8	149.49
$NaNO_3(s)$	−466.68	−365.89	116.3
$Na_2CO_3(s)$	−1 130.9	−1 047.67	135.98
$NaHCO_3(s)$	−947.68	−851.86	102.09
$Ni(s)$	0	0	30.1
$Ni^{2+}(ac)$	−64.0	−46.4	−159.4
$NiO(s)$	−244.35	−216.3	38.58
$Ni(OH)_2(s)$	−538.06	−453.1	79.5
$O(g)$	249.4	230.1	160.95
$O_2(g)$	0	0	205.0
$O_3(ac)$	−12.09	16.3	110.88
$O_3(g)$	142.2	163.4	237.6
P(blanco)	0	0	44.0
P(rojo)	−18.4	13.8	29.3
$PO_4^{3-}(ac)$	−1 284.07	−1 025.59	−217.57
$P_4O_{10}(s)$	−3 012.48		
$PH_3(g)$	9.25	18.2	210.0
$HPO_4^{2-}(ac)$	−1 298.7	−1 094.1	−35.98
$H_2PO_4^-(ac)$	−1 302.48	−1 135.1	89.1
$Pb(s)$	0	0	64.89
$Pb^{2+}(ac)$	1.6	−24.3	21.3
$PbO(s)$	−217.86	−188.49	69.45
$PbO_2(s)$	−276.65	−218.99	76.57
$PbCl_2(s)$	−359.2	−313.97	136.4
$PbS(s)$	−94.3	−92.68	91.2
$PbSO_4(s)$	−918.4	−811.2	147.28
$Pt(s)$	0	0	41.84
$PtCl_4^{2-}(ac)$	−516.3	−384.5	175.7
$Rb(s)$	0	0	69.45
$Rb^+(ac)$	−246.4	−282.2	124.27
S(rómbico)	0	0	31.88

(*continúa*)

Sustancia	ΔH_f° (kJ/mol)	ΔG_f° (kJ/mol)	S° (J/K · mol)
S(monoclínico)	0.30	0.10	32.55
$SO_2(g)$	−296.4	−300.4	248.5
$SO_3(g)$	−395.2	−370.4	256.2
$SO_3^{2-}(ac)$	−624.25	−497.06	43.5
$SO_4^{2-}(ac)$	−907.5	−741.99	17.15
$H_2S(g)$	−20.15	−33.0	205.64
$HSO_3^-(ac)$	−627.98	−527.3	132.38
$HSO_4^-(ac)$	−885.75	−752.87	126.86
$H_2SO_4(l)$	−811.3	?	?
$SF_6(g)$	−1 096.2	?	?
$Si(s)$	0	0	18.70
$SiO_2(s)$	−859.3	−805.0	41.84
$Sr(s)$	0	0	54.39
$Sr^{2+}(ac)$	−545.5	−557.3	−39.33
$SrCl_2(s)$	−828.4	−781.15	117.15
$SrSO_4(s)$	−1 444.74	−1 334.28	121.75
$SrCO_3(s)$	−1 218.38	−1 137.6	97.07
$Zn(s)$	0	0	41.6
$Zn^{2+}(ac)$	−152.4	−147.2	−106.48
$ZnO(s)$	−348.0	−318.2	43.9
$ZnCl_2(s)$	−415.89	−369.26	108.37
$ZnS(s)$	−202.9	−198.3	57.7
$ZnSO_4(s)$	−978.6	−871.6	124.7

Sustancias orgánicas

Sustancia	Fórmula	ΔH_f° (kJ/mol)	ΔG_f° (kJ/mol)	S° (J/K · mol)
Acetaldehído (*g*)	CH_3CHO	−166.35	−139.08	264.2
Acetileno (*g*)	C_2H_2	226.6	209.2	200.8
Acetona (*l*)	CH_3COCH_3	−246.8	−153.55	198.7
Ácido acético (*l*)	CH_3COOH	−484.2	−389.45	159.8
Ácido fórmico (*l*)	$HCOOH$	−409.2	−346.0	129.0
Benceno (*l*)	C_6H_6	49.04	124.5	172.8
Butano (*g*)	C_4H_{10}	−124.7	−15.7	310.0
Etano (*g*)	C_2H_6	−84.7	−32.89	229.5
Etanol (*l*)	C_2H_5OH	−276.98	−174.18	161.0
Etileno (*g*)	C_2H_4	52.3	68.1	219.5
Glucosa (*s*)	$C_6H_{12}O_6$	−1 274.5	−910.56	212.1
Metano (*g*)	CH_4	−74.85	−50.8	186.2
Metanol (*l*)	CH_3OH	−238.7	−166.3	126.8
Propano (*g*)	C_3H_8	−103.9	−23.5	269.9
Sacarosa (*s*)	$C_{12}H_{22}O_{11}$	−2 221.7	−1 544.3	360.2