21.01

What is Physics?

- **Physics** = the study of physical phenomenon and the mechanisms that cause them
 - This class is focused on **electromagnetism**
- Physics first discovered by the Greek philosophers
 - Rub piece of amber and it "pulls" straw near it
- Hans Christian Oersted = physicist who discovered the *unity* of electricity and magnetism(~1820)
 - Prior to this, the two phenomenon were assumed to be independent of one another
- **Michael Faraday** = physicist that pioneered much of the properties of electromagnetism
 - Didn't break these laws down into concrete formulas
 - * Physicists *love* them some formulas
- James Clerk Maxwell = physicist who made Faraday's discoveries more conrete
 - Created Maxwell's Equations
 - * Gauss's Law

$$\nabla \cdot \mathbf{E} = \frac{\rho}{\epsilon_0}$$

* Gauss's Law for Magnetism

$$\nabla \cdot \mathbf{B} = 0$$

* Faraday's Law of Induction

$$\nabla \times \mathbf{E} = -\frac{\delta \mathbf{B}}{\delta t}$$

* Ampere's Circuital Law

$$\nabla \times \mathbf{B} = \mu_0 \left(\mathbf{J} + \epsilon_0 \frac{\delta \mathbf{E}}{\delta t} \right)$$

- Benjamin Franklin = American scientist
 - Credited with describing the two electric charges as "positive" and "negative"
 - Also introduced the idea of excess charge
 - * Excess charge = an imbalance in positive and negative charge

Conductors and Insulators

- Conductor = a material that permits electrons to move relatively freely
 - **Superconductors** = a material that acts as a *perfect* conductor
 - * Offers no resistance to electric flow
 - Generally, conductors have electrons on the outer shells whose ionization energy is small
- **Insulators** = a material that blocks electrons from moving freely
 - Also called nonconductors
- **Semiconductor** = a material whose capacity to permit electrons to move freely is in between **insulators** and **conductors**
- Induced charge = a phenomenon in which an object's internal electric field will orient itself to align with an external electric field and create a dipole
 - $\mathbf{Dipole} =$ an object that is positively charged on one side and negatively charged on the other

Coulomb's Law

- Electrostatic force = the force that charged particles exert on one another
 - Vector quantity—the direction always points towards or away from the particle in question
- Coulomb's Law = an equation that describes the magnitude of the electrostatic force
 - Named after Charles-Augustin de Coulomb
 - Discovered in 1785

$$\vec{F} = k \frac{q_1 q_2}{r^2} \hat{r}$$

$$\vec{F} = \frac{1}{4\pi\epsilon_0} \frac{q_1 q_2}{r^2} \hat{r}$$

$$k = \frac{1}{4\pi\epsilon_0} = 9.99 \times 10^9 \frac{Nm^2}{C^2}$$

$$\epsilon_0 = 8.85 \times 10^{-12} \frac{C^2}{Nm^2}$$

- $-\epsilon_0$ is called the **permittivity constant**
 - * Mostly used for historical reasons, as it made simplifying other equations easier
- Very similar to equation for Newtonian gravity

$$\vec{F_g} = G \frac{m_1 m_2}{r^2} \hat{r}$$

Electric Charge and Current

- SI unit for charge is Coulomb(C)
 - For practical reasons, the **Coulomb** is defined using the **Ampere**
- Ampere = SI unit for current
 - Current is defined as the following, where i is current and q is charge

$$i = \frac{dq}{dt}$$

* Rearranging, we can see where the units come from

$$i \times dt = dq$$

$$(A) \times (sec) = (C)$$

- * So 1 Ampere is equal to $1\frac{C}{sec}$, and 1 Coulomb is equal to $1A \times sec$
- Electrically charged objects (which we treat as particles for simplicity) exhibit many of the same characteristics as do particles in newtonian physics
 - Force vectors add together to create a **net force** $(\sum \mathbf{F})$
 - * Net force is also called resultant force

Shell Theories

- Shell theories = models that are used to simplify problems by assuming that bodies can be treated as particles
- Shell Theory 1
 - a charged particle placed outside of a shell that is uniformly charged about its surface will experience an electrostatic force equal to the circumstance in which the shell is a particle located at the center of the shell with the same charge
- Shell Theory 2
 - a charged particle placed *inside* of a shell that is uniformly charged about its surface will experience no electrostatic force from the shell
 - * Why would that be the case???
- Spherical conductors
 - in a spherical conductor with an excess charge, particles seek to maximize the distance between all other similarly charged particles

21.02

Charge Is Quantized

- Originally, scientists thought that electric charge was a continuous phenomenon
 - But, because matter is a discrete phenomenon, charge is as well
- Elementary charge = the essential "stepping size" that the net charge of an object or particle can increment or decrement by
 - All charges can be represented as

$$n\mathbf{e}, n \in \mathbb{Z}$$

- where

$$e = 1.602 \times 10^{-19} C$$

- Quantized = a word that is meant to convey the discrete nature of a phenomenon
 - things come in "packets" rather than continuously flowing like a substance

21.03

Charge Is Conserved

- In physics, almost everything is conserved
 - the same holds true for **charge**
- Charge acts like mass in classical Newtonian physics
- Conservation of Charge = a hypothesis put forth by Benjamin Franklin that asserts the prior is the case
- Examples
 - Nuclear decay
 - * Whenever a decay event occurs, the newly created particles always have an overall charge that is equal to that of the parent particle

$$\cdot~^{238}\mathrm{U} \rightarrow ^{234}\mathrm{Th} + {}^{4}\mathrm{He}$$

- * **Annihilation** = two particles colliding and destroying each other
 - · Charge is perserved
- * **Pair Production** = the spontaneous creation of particles from existing energy

$$\gamma \rightarrow e^- + e^+$$