

# OREGON STATE UNIVERSITY CBEE DEPARTMENT OF CHEMICAL ENGINEERING CHE 331 Transport Phenomena I

Dr. Goran Jovanovic

**Mechanical Energy Balance Equation I** 

Please turn-off cell phones



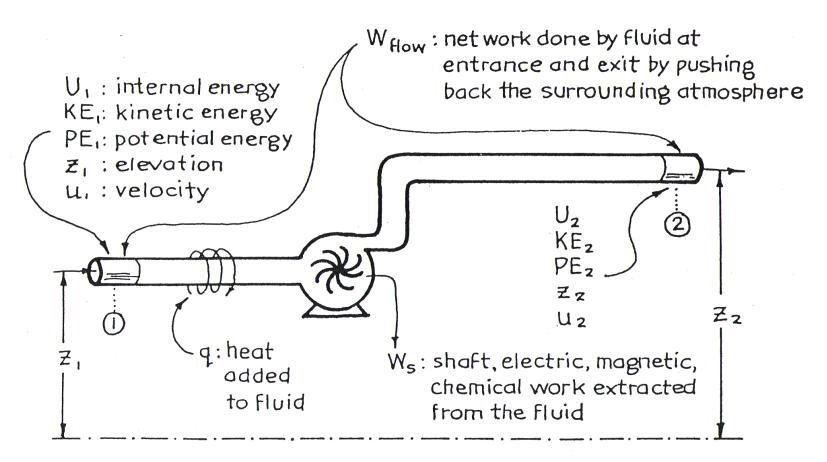
### **MECHANICAL ENERGY BALANCE EQUATION**

INTERNAL ENERGY(*U*) is the energy of a substance associated with the motions, interactions and bonding of its constituent molecules rather than,

EXTERNAL ENERGY, which is associated with the velocity and location of its center of mass which is of primary interest in mechanics.

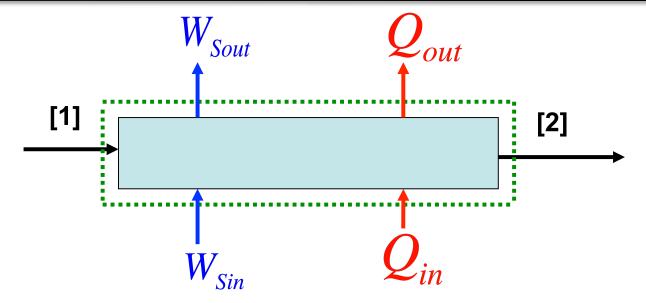


# Flow System



O. Levenspiel, Engineering Flow and Heat Exchange, Plenum Press1988

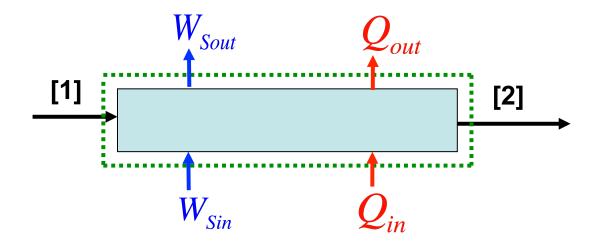




Energy Balance – Conservation of Energy [J] (at steady state):

Input – Output = Accumulation



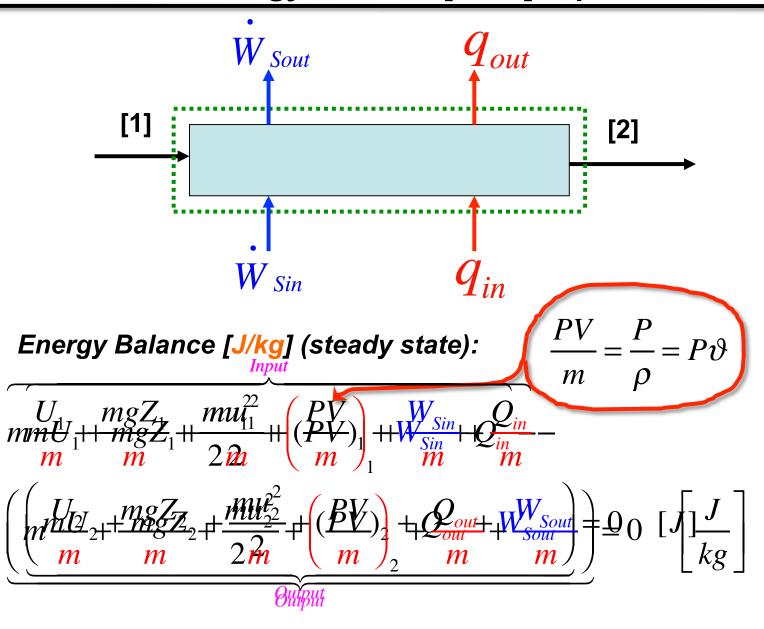


# **Energy Balance: Input – Output = 0**

$$mU_{1} + mgZ_{1} + \frac{mu_{1}^{2}}{2} + (PV)_{1} + W_{Sin} + Q_{in} -$$

$$\left(mU_{2} + mgZ_{2} + \frac{mu_{2}^{2}}{2} + (PV)_{2} + Q_{out} + W_{Sout}\right) = 0 \quad [J]$$
Output





CHE 331 Fall 2022



$$-(U_2-U_1)-g(Z_2-Z_1)-\frac{1}{2}(u_2^2-u_1^2)-[(P\vartheta)_2-(P\vartheta)_1]-(q_{out}-q_{in})-(\dot{W}_{Sout}-\dot{W}_{Sin})=0$$
 Let  $\Delta$  be (=) (2-1) or (out - in) Shaft Work 
$$\Delta U+g\Delta Z+\frac{1}{2}\Delta u^2+\Delta(P\vartheta)+\Delta q+\Delta W=0$$
 Flow Work Kinetic Energy Potential Energy

CHE 331 Fall 2022

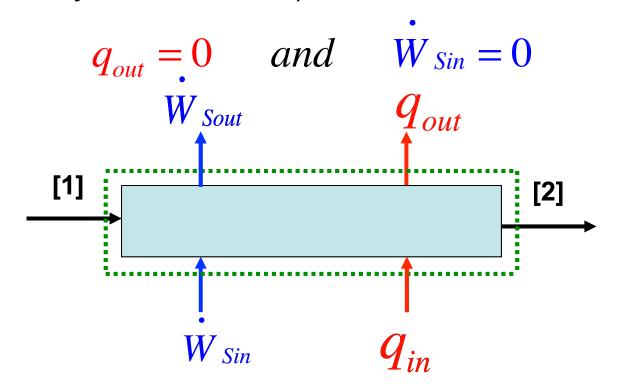
**Internal Energy** 



Remember:  $\Delta q = q_{out} - q_{in}$  and  $\Delta W = W_{Sout} - W_{Sin}$ 

$$\Delta U + g\Delta Z + \frac{1}{2}\Delta u^2 + \Delta(P\vartheta) = -\Delta q - \Delta \dot{W}$$

Often people consider (in the spirit of an old thermodynamic tradition) that:



CHE 331 Fall 2022



Which converts the above equation into:

$$\Delta U + g\Delta Z + \frac{1}{2}\Delta u^2 + \Delta(P\vartheta) = q_{in} - \dot{W}_{Sout}$$

We could rewrite the last equation:

$$[\Delta U + \Delta(P\vartheta)] + g\Delta Z + \frac{1}{2}\Delta u^2 = q_{in} - \dot{W}_{Sout}$$

$$\Delta H + g\Delta Z + \frac{1}{2}\Delta u^2 = q_{in} - \dot{W}_{Sout}$$

Because: 
$$\Delta H = \Delta U + \Delta (P \vartheta)$$

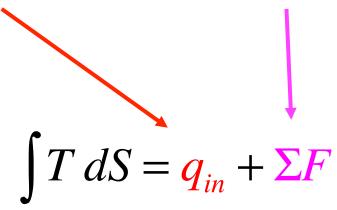


# **Second Law of Thermodynamics**

From the Second Law of Thermodynamics we know

$$\Delta U = \int T \, dS - \int P \, d\vartheta + [all \ other \ forms \ of \ energy]$$

The term  $\int T dS$  accounts for heat exchange with surrounding and heat generated by friction losses.





$$\Delta U + g\Delta Z + \frac{1}{2}\Delta u^2 + \Delta(P\vartheta) = \mathbf{q}_{in} - \dot{W}_{Sout}$$

$$\left[\int T \, dS - \int P \, d\vartheta\right] + g\Delta Z + \frac{1}{2}\Delta u^2 + \Delta(P\vartheta) = \frac{\dot{q}_{in}}{2} - \dot{W}_{Sout}$$

$$\left[q_{in} + \sum F - \int P \, d\vartheta\right] + g\Delta Z + \frac{1}{2}\Delta u^2 + \Delta(P\vartheta) = q_{in} - \dot{W}_{Sout}$$

We can differentiate the above expression:

$$\delta\left(\sum F\right) - Pd\vartheta + gdZ + udu + Pd\vartheta + \vartheta dP = -\delta\left(\dot{W}_{Sout}\right)$$



And we can obtain the *differential form* of the Mechanical Energy Balance Equation:

$$gdZ + udu + \vartheta dP + \delta \left(\sum_{i} F\right) + \delta \left(\dot{W}_{Sout}\right) = 0$$

$$gdZ + udu + \frac{dP}{\rho} + \delta \left(\sum_{i} F\right) + \delta \left(\dot{W}_{Sout}\right) = 0$$

Or in the *difference form* and for constant density:

$$g\Delta Z + \frac{\Delta u^2}{2} + \frac{\Delta P}{\rho} + \sum F + \dot{W}_{Sout} = 0$$



$$g\Delta Z + \frac{\Delta u^2}{2} + \frac{\Delta P}{\rho} + \sum F + \dot{W}_{Sout} = 0$$

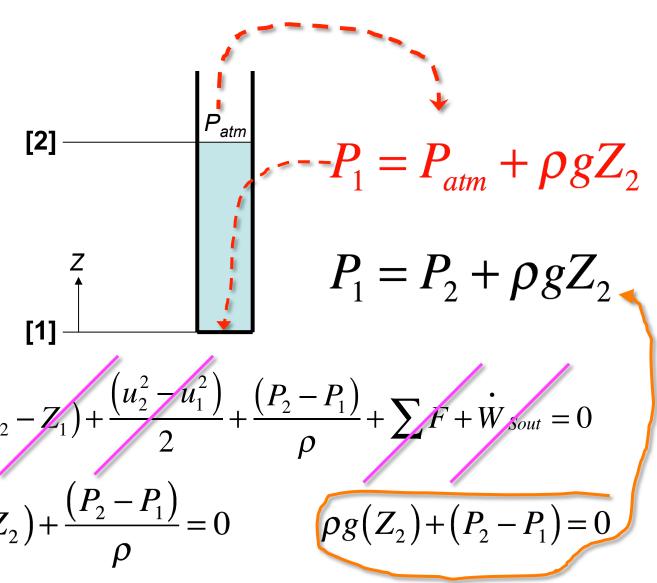


$$g(Z_2 - Z_1) + \frac{(u_2^2 - u_1^2)}{2} + \frac{(P_2 - P_1)}{\rho} + \sum_{i=1}^{n} F + \dot{W}_{Sout} = 0$$

**Mechanical Energy balance Equation** 



### **Consider the following Static Fluid Situation:**



CHE 331 Fall 2022

(Dr. Levenspiel Book contains all equations for MEB)





People. Ideas. Innovation.

Thank you for your attention!