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25.1 Nuclear Radiation Guide for Reading Key Concepts • How does an unstable nucleus release energy? • What are the three main types of nuclear radiation? Vocabulary radioactivity radiation radioisotopes alpha particle beta particle gamma ray Reading Strategy Relating Text and Visuals As you read about types of radiation, look at Figure 25 ...

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Radioactivity •Radioactivity is the process by which nuclei emit particles and rays as they break down. •The name of the penetrating rays emitted by a radioactive source is called radiation. •A radioactive isotope is an unstable atom which breaks down on its own, releasing energy and/or

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Key Concepts Unlike chemical reactions, nuclear reactions are not affected by changes in temperature, pressure, or the presence of catalysts. Also, nuclear reactions of a ... Chapter 25 Nuclear Chemistry 25.1 Nuclear Radiation 25.2 Nuclear Transformations 25.3 Fission and Fusion 25.4 Radiation in Your Life.

Chapter 25

Chapter 25: Nuclear Chemistry Vocab. Chapter 25 of Prentice Hall Chemistry Vocabulary and other vocab relating to nuclear chemistry. STUDY. PLAY. radioactivity. the process by which materials give off rays emitted by uranium atoms. radiation. the penetrating rays and particles emitted by a radioactive source.

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SECTION 25.3 FISSION AND FUSION OF ATOMIC NUCLEI (pages 810–813) This section describes nuclear fission and nuclear fusion. It discusses their potential as sources of energy, methods used to control them, and issues involved in containment of nuclear waste. Nuclear Fission (pages 810–811) 1.

SECTION 25.1 NUCLEAR RADIATION (pages 799-802)

806 Chapter 25 Nuclear Chemistry Figure 25-2 Both Pierre and Marie Curie played important roles in founding the field of nuclear chemistry. Marie Curie went on to show that unlike chemical reactions, radioactivity is not affected by changes in physical conditions such as temperature and pressure. She is the only person in history to receive

Chapter 25: Nuclear Chemistry - Jayne Heier

Section 25.2 Nuclear Transformations 805 Table 25.3 Half-Lives and Radiation of Some Naturally Occurring Radioisotopes Half-lives can be as short as a fraction of a second or as long as billions of years. Table 25.3 shows the half-lives of some radioisotopes that occur in nature. Scientists use the half-lives of some radioisotopes found in nature

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