

## *Answer Key For Chem Quest*

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X = chemical symbol of the element. Z = "atomic number" A = "mass number", , and are notations that represent . isotopes. of carbon., and are notations that represent . isotopes. of hydrogen. The part of the atom where the protons and neutrons are is called the . nucleus. Critical Thinking Questions. 1.

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Critical Thinking Questions 1. Write the electron configurations for Li, Na and K. (Remember for electron configurations, arrows are not necessary.)  $\text{Li } 2= 1s^2 2s^1$   $\text{Na } 2= 2s^2 2p^6 3s^1$   $\text{K } 2= 2s^2 2p^6 3s^2 3p^4 4s^1$  2. What is similar about the electron configurations of all of the elements in question 1? Look at the very end of their configurations.

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