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Chapter 9 Stoichiometry. Adapted from notes by Stephen Cotton ©. Section 9.1 The Arithmetic of Equations. XOBJECTIVES: • Calculate the amount of reactants required, or product formed, in a nonchemical process. • Interpret balanced chemical equations in terms of interacting moles, representative particles, masses, and gas volume at STP.

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12.1 The Arithmetic of Equations > Using Equations. Balanced Chemical Equations. The calculations of quantities in chemical reactions is a subjyect of chemistry called. stoichiometry. • For chemists stoichiometry is a form ofFor chemists, stoichiometry is a form of bookkeeping.

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### Chapter 12

tions apply to both chemistry and everyday situations. 12.1.2 Interpret balanced chemical equations in terms of moles, representative particles, mass, and gas volume at STP. 12.1.3 Identify the quantities that are always conserved in chemical reactions. Guide for Reading Build Vocabulary Paraphrase Introduce the term sto-ichiometry in your own words.

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the following reaction, how many atoms are on each side of the equation?  $2\text{Al (s)} + 3\text{CuSO}_4 \rightarrow 2\text{Al}_2\text{O}_3 + 3\text{Cu}$  ...

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