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Exploring Periodic Data - TrekNorth Math

Use the periodic table to write the electron configuration for silicon. Explain your thinking. 2. Use the periodic table to write the electron configuration for iodine. Explain your thinking. 3. Which group of elements is characterized by an s2p3 configuration? 4. Name the element that matches

the following description.

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Possible answers:Magnesium plays a key role in the process that uses sunlight to produce sugar in plants. Mixtures of magnesium and other metals are used in transportation because they can be as strong as steel,but much lighter. Aluminum is the most abundant metal in Earth's crust. It is strong,lightweight,

Chapter 5The Periodic Table Section 5.3 Representative Groups

Circle the letter of the correct answer. When chlorine occurs in nature, there are three atoms of chlorine-35 for every one atom of chlorine-37. Which atomic mass number is closer to the average atomic mass of chlorine? a. 35 amu b. 37 amu CHAPTER 5, Atomic Structure and the Periodic Table(continued)

5 ATOMIC STRUCTURE AND THE PERIODIC TABLE - Bergen

Prentice Hall Chemistry (Wilbraham) - Louisiana Edition ... element based only on its location in the periodic table (PS-H-C2) SE: 160-166, 167-173, 174-184 16. Predict the stable ion(s) an element is likely to form when it reacts with other specified elements (PS-H-C2) SE: 174-182,194-200, 201-208 , 219

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