

## *Answer Key To Modern Chemistry*

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unit of measure that is a combination of other measurements. volume. a measure of the size of a body or region in three-dimensional space. density. the ratio of the mass of a substance to the volume of a substance; often expressed as grams per cubic centimeter for solids and liquids and as grams per liter for gases.

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Modern Chemistry 88 Chapter Test Chapter: States of Matter PART I In the space provided, write the

letter of the term or phrase that best completes each statement or best answers each question.

\_\_\_\_ 1. According to the kinetic-molecular theory, particles of an ideal gas a. attract each other but do not collide. b. repel each other and collide.

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Modern Chemistry • CHAPTER 9 HOMEWORK 9-1 (pp. 275-279) ... Modern Chemistry • CHAPTER 9 HOMEWORK 9-2 (pp. 280-282) ... Circle the letter of the best answer. 1. For the general chemical equation  $2A + 3B \rightarrow 4C + D$ , how many moles of A are required to produce 18.4 mol of C?

### **HOMEWORK 9-1**

CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

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Modern Chemistry 33 Chapter Test Name Class Date Chapter Test B, continued 15. The energy state of an atom is called its ground state. 16. The number of waves that pass a point in one second is called. 17. ... Modern Chemistry 224 Answer Key TEACHER RESOURCE PAGE.

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### **Modern chemistry chapter 9 3 review stoichiometry answers**

Modern Chemistry 84 Chapter Test Chapter: States of Matter In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. \_\_\_\_ 1. The kinetic-molecular theory of gases assumes that the particles of an ideal gas are separated by great distances. This implies that the gas

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