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of either +1 or -1 and thus produce a charged meson. Only these four arrangements are possible. a meson with a charge of +1 requires either (ud

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Answers Marks Examiner's tips. 1 (a) sinoatrial node/SAN; 1 SAN can be used to represent the s-noatrial node. (b) (i) to allow the contraction/emptying of. the atrium before the ventricle contracts/allows ventricles to fill before contraction; 1 Remember both atria contract simultaneously, followed by simultaneous contraction of the ventricles.

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Answers to exam-style questions Topic 5 d 1 B 2 C 3 B 4 D 5 C 6 B 7 D 8 D 9 B 10 A 11 a The a verage amount of energy required to break one mole of covalent bonds in a gaseous molecule under standard conditions. 'Average' refers to the fact that the bond enthalpy is di^{\sim} erent in di^{\sim} erent

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If you get the right answer you get all 3 marks. If it is not correct then the examiner goes back to find how many marks you can be awarded. CH 3 CH 3 C Br H :NH 3 CH 3 CH 3 C H + H N H H (:NH 3) - + + C Br or C Br H CH 3 C Br H C H H H CH 3 C CH 2:OH-

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End of Unit 4 Answers to examination-style questions AQA Physics A A2 Level © Nelson Thornes Ltd 2009 3 Answers Marks Examiner's tips (iii) Longitudinal stress in ...

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Answers to exam-style questions 437.0 mol of Mn 3O 4 will react with 437.0 \times 8/3, i.e. 1165 mol. The number of moles of Al is greater than this, so Al is present in excess and Mn 3O 4 is the limiting reactant. Mn 3O 4 must be used in all calculations. 3 mol Mn 3O 4 produces 9 mol Mn. Therefore 437.0 mol

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Answers to examination-style questions - Lagan College Physics

Answers Marks Examiner's tips. 1 (a) pathogens; 1 (b) any 2 from: bacteria, viruses, fungi; 2 The question states organisms rather than microorganisms so fungi is an acceptable answer, as well as some other types of organism that you do not need to know about. (c) digestive system/gut; respiratory system/lungs;

Answers to examination-style questions - VBIOLOGY

Answers to exam-style questions The volume of carbon dioxide produced is measured every 10 s. [3] b and d b[2] d[2] Topic 6 1 C 2 A 3 B 4 C 5 C 6 C 7 A 8 C 9 D 10 B 11 There are two factors that contribute to the increase in rate of reaction with increase in temperature. The

Answers to exam-style questions - Cambridge University Press

of the question. The pH at the equivalence point is greater than 7, as sodium butanoate is formed in the titration. The $_{\cdot}$ nal pH will approach the pH of the sodium hydroxide solution used. iv The most suitable indicator for a strong base–weak acid titration is phenolphthalein. The range of the indicator (8.3–10.0) comes

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