

An Aqueous Antifreeze Solution Is 40 Ethylene Glycol

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An Aqueous Antifreeze Solution Is

An aqueous antifreeze solution is 40.0% ethylene glycol ($\text{C}_2\text{H}_6\text{O}_2$) by mass. The density of the solution is 1.05 g/cm³. Calculate the molality, molarity, and mole fraction of the ethylene glycol.

Solved: An aqueous antifreeze solution is 40.0% ethylene ...

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Submitted by Teresa M on Tue, 2012-03-20 13:20. An aqueous antifreeze solution is 40.0% ethylene glycol ($\text{C}_2\text{H}_6\text{O}_2$) by mass. The density of the solution is 1.05 g/cm³. Calculate the molality, molarity, and mole fraction of the ethylene glycol.

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