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8 Chemistry: Matter and Change • Chapter 9 Teaching Transparency Masters Balancing Chemical EquationsBalancing Chemical Equations TEACHING TRANSPARENCY MASTER Use with Chapter 9, Section 9.1 30 Steps for Balancing Equations 1. Write the skeleton equation for the reaction. 2. Count the atoms of each element in the reactants. 3.

Chapters 9-12 Resources - pgasd.com

TEACHING TRANSPARENCY WORKSHEET Atomic Orbitals 1. What is the shape of an s orbital? Class 16 Use with Chapter 5, Section 5.2 2. What is the relationship between the size of an s orbital and the principal energy level in which it is found? 3. What is the shape of a p orbital? How many p orbitals are there in a ublevel? 4.

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Teaching Transparency Worksheets Chemistry: Matter and Change • Chapter 14 37 1. What variables are plotted on the graph? 2. What is the unit of each variable? 3. Use the graph to complete the table below. 4. At what temperature are sodium chloride and potassium chloride equally soluble in water? 5.

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TEACHING	S TRANSPARI	ENCY 22 Chemis	stry: Matter a	nd Change	Teaching ⁻	ا Transparency 1	Use with
Chapter 7	, Formation	of lons Section	7.1 . Name	Date	Class _	TEACHING	
TRANSPAF	RENCY 22 Ch	emistry: Matter	r and Change	Teaching T	ransparen	cy Worksheet	Chemistry
Matter an	d Change 3	Teacher Guide a	and Answers ⁻	Teaching Tr	ransparenc	y 22	

Name Date Class TEACHING TRANSPARENCY

TEACHING TRANSPARENCY WORKSHEET The Activity Series Use with Chapter 9, Section 9.2 1. For each of the following pairs of elements, underline the one that would replace the other element in a compound. a. calcium, tin e. iron, copper b. bromine, fluorine f. iodine, chlorine c. aluminum, potassium g. silver, lead d. zinc, sodium 2.

TEACHING TRANSPARENCY WORKSHEET The Activity Series

Teaching Transparency Worksheets Chemistry: Matter and Change • Chapter 5 7 1. What kinds of waves have the longest wavelength? What kinds of waves have the short-est wavelength? 2. Which waves have the lowest frequency? 3. Which has a higher frequency: microwaves or X rays? 4. Which waves can be seen by the eye? 5.

TEACHING TRANSPARENCY MASTER 15 The Electromagnetic ...

TEACHING TRANSPARENCY The Electromagnetic Spectrum 1. What kinds of waves have the longest wavelength? What kinds of waves have the shortest w velength? 2. Which waves have the lowest

frequency? 3. Which has a higher frequency: microwaves or X rays? S 4. Which waves can be seen by the eye? 5.

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Explain your answer. 8. Suppose you fill a container with steam and then seal the container. When the steam in the container changes to liquid water at room temperature, will the container still be full? Explain your answer. States of Matter TEACHING TRANSPARENCY WORKSHEET Use with Chapter 3, Section 3.1 7

Name Date Class TEACHING TRANSPARENCY WORKSHEET 7 States ...

TEACHING TRANSPARENCY WORKSHEET Ionic Bonds Use with Chapter 7, Section 7.2 1. How many valence electrons does a neutral magnesium (Mg) atom have? 2. What is the charge on a magnesium ion? What does magnesium have to do to form such an ion, and why does it tend to do so? 3. How many valence electrons does a single neutral chlorine (CD atom have? 4.

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Teaching Transparency Worksheets Chemistry: Matter and Change • Chapter 6 35 1. What are the four sections, or blocks, of the periodic table? 2. What does each block represent? 3. What do elements in the s-block have in common? 4. What is the valence electron configuration of each element in group 1? 5.

TEACHING TRANSPARENCY WORKSHEET 19 The s-, p-, d-, and Use ...

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TEACHING TRANSPARENCY MASTER 45 Temperature Changes of ...

TEACHING TRANSPARENCY WORKSHEET Isotopes 1. What do the following symbols represent? b. no 2. Which subatomic particles are found in an atom's nucleus? 3. Which subatomic particle identifies an atom as that of a particular element? 4. Explain why atoms are neutral even though they contain charged particles. 30 5.

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MATH SKILLS TRANSPARENCY MASTER 16

MATH SKILLS TRANSPARENCY 5 Interpreting Waves 1. Look at the two waves shown. What is the speed of each wave? cc Use with Chapter 5, Section 5.1 syzed 2. Look at the two waves shown. Which wave has a higher frequency? Which wave has a longer wavelength? 3. Assume that wave A has a wavelength of 699 nm. Calculate the frequency of the wave. Show ...

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