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Answers to exam-style questions The volume of carbon dioxide produced is measured every 10 s. [3] b and d b[2] d[2] Topic 6 1 C 2 A 3 B 4 C 5 C 6 C 7 A 8 C 9 D 10 B 11 There are two factors that contribute to the increase in rate of reaction with increase in temperature. The

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9 A (The question should have referred to the wavelength in air) 10 C 11 a In simple harmonic motion the acceleration is opposite to and proportional to the displacement from the

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Answers to exam-style questions 437.0 mol of Mn 3O 4 will react with 437.0 \times 8/3, i.e. 1165 mol. The number of moles of Al is greater than this, so Al is present in excess and Mn 3O 4 is the limiting reactant. Mn 3O 4 must be used in all calculations. 3 mol Mn 3O 4 produces 9 mol Mn. Therefore 437.0 mol

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2 Correct answer = 2 marks. Correct working but incorrect answer = 1 mark. (ii) provides glucose and oxygen and removes carbon dioxide; increases respiration/increases ATP yield; 2 Many candidates forget that blood removes carbon dioxide from respiring cells. 3 (a) (i) 75 beats per minute; 1 The question states a complete cardiac

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(b) any 2 from: bacteria, viruses, fungi; 2 The question states organisms rather than microorganisms so fungi is an acceptable answer, as well as some other

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Best Answer: A counterexample is an example where the assertion is not true. In this case $8^3 - 7^3 = 512 - 343 = 169$ is a counterexample because $169 = 13^2$ is not a prime number. b You just have to work it out

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words. The same words may appear in the NCLEX question and in the correct answer. It may be the same word or a synonym of the word.

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of the question. The pH at the equivalence point is greater than 7, as sodium butanoate is formed in the titration. The <code>_</code> nal pH will approach the pH of the sodium hydroxide solution used. iv The most suitable indicator for a strong base-weak acid titration is phenolphthalein. The range of the indicator (8.3–10.0) comes

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