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Solutions and Tests for Exploring Creation With Chemistry SOLUTIONS TO THE EXTRA PRACTICE PROBLEMS FOR MODULE #14 1. Reaction rate has units of M/s. When solving for the rate constant, you divide by the concentrations of the reactants raised to their orders. To get these units for the rate constant, then, the rate must be divided by M^{\wedge} .

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Solutions and Tests for Exploring Creation With Chemistry way, between trials 2 and 3, the F concentration remained constant, but the C4C190H concentration doubled. When that happened, the rate doubled. ... Graph III looks like the one presented in the module, so it is the correct answer. 8. This is a heterogeneous catalyst, because it is in a ...

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A chemistry book lists the rate constants for 2 different reactions. The first rate constant is 0.12 1/s and the second is 34.5 1/(Mxs) . Even though the rate equations for these reactions are not given, an astute student says that the overall order of these 2 chemical reactions must be different from each other.

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Apologia Chemistry module 13 practice problems? My teacher told us to check our work for the practice problems by looking up the answers online. "It'll be easy to find them," she said. :p She has the answer book, I don't understand why she didn't just make a copy of the answers and give them to us.

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