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Limiting Reactants Read pages 271 - 276 in your text. Problem: Sodium chloride is prepared by reacting 290 g of sodium with 560 g of chlorine. Determine the mass of sodium chloride produced.

1. Write the balanced the equation. 2 Na + Cl 2 \rightarrow 2 NaCl 2. Write the information given about the reactants.

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Chemistry (1st Semester) - Ms. Johnson's Website

Sometimes during a chemical reaction, one type of reactant will be used up before the other reactants. This reactant is the limiting reactant. Using the Limiting Reactants Gizmo $^{\text{TM}}$, you can determine which reactant is limiting in various scenarios.

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What is the limiting reactant if 2.2 g of Mg is reacted with 4.5 L of oxygen at STP? Both of the following give you the same answer. In the first case, you need to do one or the other. In the second set, you need to do both to determine which reactant makes the least amount of the

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