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forming the question, or need help seeing how the lab relates to stoichiometry; performing the stoichiometry; special care should be spent making sure students are using the acetic acid mass, not the mass of the vinegar. To save time I have made this Stoichiometry lab answer key so I can quickly check student work. creating a step-by-step procedure

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128 CH104 Lab 10: Solubility (F15) When solutions of two soluble ionic compounds are mixed, the formation of an insoluble solid precipitate indicates that a chemical reaction has occurred. A solid forms when the positive ion of one substance in solution, and the negative ion in solution are more attractive to each other than they are to water.

LAB 10: S - Chemeketa Community College

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