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Experiment 10 Stoichiometry- Gravimetric Analysis

Lab 9 Stoichiometry And Gravimetric Gravimetric Analysis Escience Lab 11. Gravimetric Analysis CHEM 200-03 Fall Semester 2012 September 19, 2012 Jeff Marasigan Abstract Gravimetric analysis is based on the quantitative isolation of the desired constituent the analyte of interest from the

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Lab: 9-2 Stoichiometryand Gravimetric Analysis. A.Write a balanced equation for the reactions (Na2CO3 andCaCl2). Na2Co3(aq)+CaCl2(aq)--->CaCO3(s) + 2NaCl(aq) What is the precipitate?Na2CO3 and CaCl2 Writeits empirical formula? a. How many molesof Na2Co3 were present in the 15mL sample?

Solved: Lab: 9-2 Stoichiometryand Gravimetric Analysis A.W ...

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Stoichiometry And Gravimetric Analysis Lab Answers

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Chemistry 2202 Lab Activity 3: Gravimetric Stoichiometry I Page 3 of 4 Day 2 1. Record visual observations describing the changes that have occurred in the colour of the solution, the appearance of the copper wire, and the formation of a new solid. 2. Gently shake the solid crystals from the copper wire. Use the wash bottle and the

Lab Activity 3: Gravimetric Stoichiometry I

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