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5. The average kinetic energy of the gas particles is directly proportional to the Kelvin temperature of the gas. POGIL 2005 Authored by Edited by Linda Padwa and David Hanson, Stony Brook University. 3/3 Kinetic Molecular Theory. Key Questions 1. What causes a gas to exert pressure when confined in a container? 2.

kinetic-molecular-theory original | Gases (1.2K views)

Kinetic Molecular Theory © POGIL – 2005 5/5 Authored by Applications 1. There is a government warning on all aerosol cans that states: Do not store at a temperature above 120° F (50° C). a) Explain why this warning is required in terms of the relationship between temperature and pressure and the kinetic molecular theory.

Kinetic Molecular Theory - Ms. Lynch's Lessons

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© POGIL – 2005 1/1 Authored by Edited by Linda Padwa and David Hanson, Stony Brook University Kinetic Molecular Theory Why? The kinetic-molecular theory is a model or a mental image of how particles of matter behave.

Kinetic Molecular Theory Practice Pt2 (POGIL) - Kinetic ...

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Review Questions Answer KEY 1. Kinetic molecular theory is the theory that explains the motion of solids, liquids, and gases. 2. KMT explains the differences between properties of solids, liquids, and gases by examining how the particles are moving under similar conditions. 3. a. Solid – a phase of matter with a definite shape and volume.

Review Questions Answer KEY - Mrs. Bonanno's Chemistry ...

The Kinetic Molecular Theory states that the pressure of a gas results from the mass of particles. the collisions of particles with the walls of a container. the forces of repulsion between the particles. the energy loss experienced by each particle.

Kinetic Molecular Theory Answer Key - HelpTeaching.com

Modified from Foundations of Chemistry by David Hanson Page 1 of 3 POGIL: Kinetic Molecular Theory Learning Objectives Identify the basic differences between particle behavior in the solid, liquid, and gaseous phases. Develop an understanding of the postulates of the kinetic molecular theory. Identify how temperature affects molecular motion. ...

Kinetic Molecular Theory(POGIL) - PO G IL K inetic M ...

Structure and Dynamics) as needed, and ChemActivity Exercises & Problems Pages 640-644 Chp15 ChemActivity on Nuclear Chemistry Kinetics ... Summary of General Chemistry Redesign Project

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Information (Kinetic Molecular Theory) The observed behavior of gases, as expressed by the

empirical gas laws, can be understood on the basis of the kinetic molecular theory, which developed over many years up to 1857, when Rudolf Clausius (1822-1888) published it in its most complete and elegant form.

Chem 116 POGIL Worksheet - Week 2 Gas Laws - Part 2 introduction to the properties of gases based upon the kinetic molecular theory

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