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Dependence of Electromotive Force on Concentrations Calculate the electromotive force (in volts) registered by an electrode immersed in a solution containing the following mixtures of NAD and NADH at pH 7.0 and 25 C, with reference to a half-cell of E 0.00 V. (a) 1.0 mM NAD and 10 mM NADH (b) 1.0 mM NAD and 1.0 mM NADH (c) 10 mM NAD and 1.0 mM NADH Answer The relevant equation for calculating E for this system is $E = E^\circ - \frac{RT}{n} \ln \frac{[NAD]}{[NADH]}$ At 25 C, the RT/n term simplifies to 0.026 V/n.

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