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Essential Understanding Ionic compounds are the result of ionic bonds forming between oppositely charged ions. Formation of Ionic Compounds An ionic compound is made up of anions and cations and has an overall charge of 0. ►►The electrostatic attraction between an anion and a cation is an ionic bond.

BONDING AND INTERACTIONS

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Ionic Compounds Molecular compounds tend to have relatively lower melting and boiling points than ionic compounds.

Covalent Bonding - Pittsfield High School

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Chapter 6 Chemical Bonds Section 6.1 Ionic Bonding

Physical Science Reading and Study Workbook Chapter 6 59 Section 6.1 Ionic Bonding (pages 158–164) This section describes the formation of ionic bonds and the properties of ionic compounds. Reading Strategy (page 158) Sequencing As you read, complete the concept map to show what happens to atoms during ionic bonding. For more information on this

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- In contrast to ionic compounds, which are formed from a metal combined with a nonmetal, most molecular compounds are composed of atoms of two or more nonmetals. Comparing Molecular and Ionic Compounds Molecular compounds tend to have relatively lower melting and boiling points than ionic compounds.

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