Alkaline Solution Definition Chemistry

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Updated March 08, 2017. Alkaline Definition: Alkaline refers to an aqueous solution having a pH greater than 7 or having a [OH-] greater than 10-7. Also Known As: basic. Know the Definition of pH in Chemistry. Definition of Alkalinity in Chemistry. Basic Solution Definition.

Definition of Alkaline in Chemstry - ThoughtCo

An alkaline solution is a mixture of base solids dissolved in water. The potential of hydrogen, also known as the pH scale, measures the alkalinity or acidity level of a solution. The scale ranges from zero to 14. The midpoint 7 represents a neutral pH. A neutral solution is neither an acid nor alkaline.

What is an Alkaline Solution? | Livestrong.com

Two Meanings. Sometimes, chemists use the phrase "alkaline solution" more broadly to refer to any base solution. Bases measure higher than a neutral 7 on the pH scale, and cause the solution to be high in OH- ions. Examples of bases are the kitchen cleaners ammonia and sodium hypochlorite, or bleach.

What Is an Alkaline Solution? | Sciencing

Acidic Solution Definition. An acidic solution is any aqueous solution which has a pH < 7.0 ([H +] $> 1.0 \times 10$ -7 M). While it's never a good idea to taste an unknown solution, acidic solutions are sour, in contrast to alkaline solutions, which are soapy. Examples: Lemon juice, vinegar, 0.1 M HCl, or any concentration...

Acidic Solution Definition in Chemistry - ThoughtCo

Alkali. In chemistry, an alkali (/ˈælkəlaɪ/; from Arabic: al-qaly "ashes of the saltwort ") is a basic, ionic salt of an alkali metal or alkaline earth metal chemical element. An alkali also can be defined as a base that dissolves in water. A solution of a soluble base has a pH greater than 7.0.

Alkali - Wikipedia

Acidic solutions are any solution that has a higher concentration of hydrogen ions than water; solutions that have a lower concentration of hydrogen ions than water are called basic or alkaline solutions. Acidity is measured on a scale known as pH which sets water at 7; all acidic solutions have pH less than 7 and bases have pHs greater than 7.

Definition of Acidic Solution | Sciencing

An alkaline solution is formed when an alkali is dissolved in water. Alkali refers to any basic hydroxide or a salt of alkali metals or alkaline earth metals. The term alkaline is derived from the metal elements of group 1 and group 2 in the periodic table of elements.

Difference Between Acid and Alkaline | Definition ...

alkaline - relating to or containing an alkali; having a pH greater than 7; "alkaline soils derived from chalk or limestone". alkalic. chemical science, chemistry - the science of matter; the branch of the natural sciences dealing with the composition of substances and their properties and reactions.

Alkaline - definition of alkaline by The Free Dictionary

Alkaline buffer solutions. An alkaline buffer solution has a pH greater than 7. Alkaline buffer solutions are commonly made from a weak base and one of its salts. A frequently used example is a mixture of ammonia solution and ammonium chloride solution.

BUFFER SOLUTIONS - chemquide

The same holds true for pH values above 7, each of which is ten times more alkaline (another way to say basic) than the next lower whole value. For example, pH 10 is ten times more alkaline than pH 9 and 100 times (10 times 10) more alkaline than pH 8. Pure water is neutral.

pH Scale - Department of Chemistry

Acidity and alkalinity are characteristics of substances in an acid-base reaction. Generally, acidic substances can turn purple litmus test solution into red, and alkaline substances can make them blue. Later, with the development of acid-base theory, people give a more accurate and complete definition gradually touches on the nature of acid and alkali.

Chemistry, Acidity and Alkalinity

Acidic solution. A solution that has a pH below 7.0 (neutral) ... Inorganic chemistry. ... A pH below 7 is an acidic solution; a pH above 7 is an alkaline solution. Physical change. A change in the form or physical properties of a substance, without a chemical reaction or the creation of a new substance.

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