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Chemistry: Chapter 15 Concentration and Nature of Solutions

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10—2 Review and Reinforcement (continued) 9. $0.0621\text{ mol SS} = (Q \$ \text{ gK}2\text{S } 10. 65.8\text{ gMg} = \text{l. atoms Mg}$ Solve each of the following problems as directed. Show all your work. 11. Calculate the number Of atoms in 0.40 mol of sulfur- 12. Calculate the number of atoms in 2.30 mol of silver. G.oux 13. You have a tank of NO₂ gas.

10-2 Review and Reinforcements - Front Page

2 Answer.Key 2-1 Explore, page 5 ... 2-1 Review and Reinforcement, pages 8-9 . 1. energy 2. radiant 3. chemical potential 4. joule 5. created, destroyed 6. natural gas ... 15. chromatography 16. electrolysis 17. distillation 18. A magnet could be used to separate them

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'16—1A Review and Reinforcement, pages 9-10 6. Answers Will Vary. When the rates Of forward and reverse reactions are equal, equilibrium is established. For the example of ... 15. The forward and reverse reactions are both occurring, but with no net change in concentration. -16—2 Review Reinforcement, pages 15-16

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5-2 Review and Reinforcement - MRS. Nairn - HOME

15. Why is water called the universal solvent? ____ 16. How could you determine experimentally whether a given substance is an electrolyte? ... Title: 13-1 Review and Reinforcement Author: NPSD Last modified by: NPSD Created Date: 9/27/2005 1:14:00 PM Company: NPSD

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