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A fun, clear, comprehensive and varied worksheet that works well as starter, plenary or as a chunk of the main body of lesson. Suitable for GCSE or IGCSE Chemistry, Science (Double Award) or Science (Single Award), this resource covers: 1.The meaning of 'group number' and how this relates to reactivity trends 2. How 'group number' relates to outer shell electrons 3.

**The Periodic Table Worksheet (With Answers) by ...**

Name\_\_\_\_\_ Worksheet: Periodic Trends Period\_\_\_\_\_ 1. Which statement best describes Group 2 elements as they are considered in order from top to bottom of the Periodic Table? (A) The number of principal energy levels increases, and the number of valence electrons increases.

**Name Worksheet: Periodic Trends Period - kmacgill.com**

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6.3 Periodic Trends. Prentice Hall Chemistry, 2005. STUDY. PLAY. Atomic radii generally \_\_\_\_\_ as you move from left to right in a period. (energy levels, increases, charge, decreases) decreases.

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Periodic trends: Electronegativity answers. Name \_\_\_\_\_ 1. What is the electronegativity of an element? It is the tendency of a bonded atom to accept or to donate an electron. ... groups, and periods will be graphed to answer this worksheet. On the graph to the right of your screen, go to the drop-down menus below x and y (this

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