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In your answer, consider the external and internal pressure data given in Model 1. Read This! Pressure is caused by molecules hitting the sides of a container or other objects. The pressure changes when the molecules change how often or how hard they hit. A nonflexible container is needed if the gas ... 25 Gas Variables-S Created Date:

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POGIL Chemistry Activities Introduction to Chemistry • Safety First ... In your answer, consider the external and internal pressure data given in Model 1. Read This! ... Gas Variables 7 Extension Questions 22. Draw a sample of gas that is colder than all three of the samples in Question 21.

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Gas Variables 1 Gas Variables How are the variables that describe a gas related? Why? Imagine buying a balloon bouquet at a party store. How will the helium gas in the bouquet behave if you carry it outside on a hot summer day? How will it behave if you carry it outside during a snowstorm?

Gas Variables How are the variables that describe a gas related? Why? Imagine buying a balloon bouquet at a party store. How will the helium gas in - Master Your Classes™ | Course Hero

CH301 Worksheet 8—Gases (Answer Key) 1. What do we assume about ideal gases? What is the ideal gas law? Give the units for each variable. Ideal gases are infinitely small, hard spheres that do

not interact with each other. They are essentially "blind" to other gas molecules and will bounce off of each other just as they would bounce off a wall.

CH301 Worksheet 8—Gases (Answer Key) 0.08206 (L*atm)/(K*mol). - University of Texas at Austin

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We will employ various techniques of measuring these variables. Since we will be collecting gas ... (3.25 x 10⁻⁴ moles) Volume of H₂ gas ... Molar Volume Of Hydrogen

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