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the rates of effusion of gases at the same temperature and pressure are inversely proportional to the square roots of their molar masses Rate of effusion of A / B = $\sqrt{M_b} / \sqrt{M_a}$

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Modern Chemistry 1 Solutions CHAPTER 12 REVIEW Solutions Teacher Notes and Answers Chapter 12 SECTION 1 SHORT ANSWER 1. c 2. a 3. b 2. a. alcohol b. water c. the gels 3. The mixture is a colloid. The properties are consistent with those reported in Table 3 on page 404 of the text. The particle size is small, but not too small, and the mixture

CHAPTER 12 REVIEW Solutions - Weebly

182 Study Guide for An Introduction to Chemistry Section Goals and Introductions Section 11.1 Gases and Their Properties Goals To describe the particle nature of both real and ideal gases. To describe the properties of gases that can be used to explain their characteristics: volume, number of particles, temperature, and pressure.

Chapter 11 - Gases - An Introduction to Chemistry

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mc06se_cFMs_r_i-vi.qxd 11/22/04 12:07 PM Page vi. CHAPTER 1 REVIEW Matter and Change SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. ... SHORT ANSWER Answer the following questions in the space provided. 1.

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