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Figure 24.2.3 A Nuclear Transmutation Reaction Bombarding a target of one element with highenergy nuclei or subatomic particles can create new elements. Electrostatic repulsions normally prevent a positively charged particle from colliding and reacting with a positively charged nucleus.

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Chemistry: Matter ar	nd Change Study Guide 102 24 Section 24.3 Nuclear Reactions In your
textbook, read abou	t the process of induced nuclear transmutation. For each statement below,
write true or false	1. Transmutation is the conversion of an atom of one element to an atom
of another element.	2.

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Answer Key for Nuclear Chemistry Worksheet #1: Nuclear Decay Processes Chem 160 – K. Marr Key Questions 1. What does the symbol " e" tell the reader? (i.e., What does the superscript 0 mean? What can a subscript -1 possibly mean? Why is the beta particle symbolized with an e?)

Answer Key for Nuclear Chemistry Worksheet #1: Nuclear ...

The nuclear binding energy expressed in Joule per nucleus range from about $3.57 \times 10-13$ for deuterium to about $2.89 \times 10-10$ for uranium-238. The larger the nucleus the larger the binding energy. However, nuclear stability may be compared based on binding energy if the latter is expressed in term of Joule per nucleon.

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