

## *15 4 Practice Problems Prentice Hall Answers*

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15-4 Practice Problems 1) What is the boiling point elevation when 11.4 g of ammonia (NH<sub>3</sub>) is dissolved in 200g of water? K<sub>b</sub> for water is 0.52 °C/m. 2) How many grams of benzoic acid (C<sub>7</sub>H<sub>6</sub>O<sub>2</sub>) must be dissolved in 78.1g of ethanol to raise the boiling point by 4.00 °C? K<sub>b</sub> for ethanol is 1.2 °C/m.

### 15-4prac - 15-4 Practice Problems 1 What is the boiling point elevation when 11.4 g of ammonia(NH<sub>3</sub>) is dissolved in 200g of water K<sub>b</sub> for water is 0.52 - Master Your Classes™ | Course Hero

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4.75 . L of solution, what is the molarity of the solution? 8. To prepare 10.5 L of a 2.50 . M solution of potassium hydroxide (KOH), how many grams of potassium hydroxide must be used? 9. What is the molality of a solution containing 75.2 g of silver perchlorate (AgClO<sub>4</sub>) dissolved in 885 g of benzene? 10.

### 15-2 Practice Problems - Weebly

Chapter 15 Practice Problems Basically, all problems at the end of chapter 15 (pages 373-374) are worth considering. I have provided answers to these below. 1. A central bank plays a vital role in any nation's economy. By controlling the rate at which it creates paper money, the central bank is able to affect inflation and economic growth. 2.

### **Chapter 15 Practice Problems with answers**

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Practice Problems In your notebook, solve the following problems. SECTION 6.1 ORGANIZING THE ELEMENTS 1. Which element listed below should have chemical properties similar to fluorine (F)? a. Li b. Si c. Br d. Ne 2. Identify each element as a metal, metalloid, or nonmetal. a. fluorine b. germanium c. zinc d. phosphorus e. lithium 3.

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10—2 Practice Problems, pages 14-15 Answers marked with an asterisk denote additional practice problems that appear ... 0.15 g 0.30 g 10—2 Review and Reinforcement, Here are the answers to the practice problems from sections ...

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15—4 Practice Problems (continued) 11. How many grams of silver would have to be dissolved in 1120 g of ethanol to lower the freezing point by 0.25 °C? K<sub>f</sub> for ethanol is 1.99 °C/m. 12. What is the freezing point depression when 85.3 g of oxygen is dissolved in 1500 g of water? K<sub>f</sub> for water is 1.86 °C/m. 13. Ethylene glycol (C<sub>2</sub>H<sub>6</sub>O<sub>2</sub>) is the ...

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pressure is 762.4 mm Hg. If the mercury level on the side open to the atmosphere is 3.6 mm higher than on the side attached to the cabinet, what is the pressure inside the cabinet in units of kPa? The U-tube of a manometer is 26.4 cm tall. With both ends open, it is filled until the mercury level in each tube is 13.2 cm from the top.

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238 Unit Practice Problem Answers Additional Practice Problem Answers 14 1) tenths 2) hundredths 3) thousandths 1) tenths 2) hundredths 3) thousandths 4) ten thousandths 5) hundred thousandths 4) ten thousandths 5) hundred thousandths 6) millionths 7) four-tenths 8) forty-three hundredths 6) millionths 7) six-tenths 8) sixty-three hundredths 9) four ...

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