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10-2 Review and Reinforcement (continued) 9. 0.0621 mol SS = (Q \$ gK2S 10. 65.8gMg= I. atoms Mg Solve each of the following problems as directed. Show all your work. 11. Calculate the number Of atoms in 0.40 mol of sulfur- 12. Calculate the number of atoms in 2.30 mol of silver. G.oux 13. You have a tank of NOZ gas.

10-2 Review and Reinforcements - Front Page

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10-1 Review and Reinforcement (continued) Find the formula mass of each of the following compôunds. Show all your work. 11. caC03 12. MgS04 13. NaOH 14. KCl 15. BaF2 o JC 04) 137B Find the molar mass of each of the following compounds. Show all your work. 16. nitric acid (HN03) (D

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11—1 Review and Reinforcement Stoichiometry Complete the following sentences by 'filling in the appropriate word or phrase from the list below. Each word or phrase maybe used once, more than once, or not at all reactants molar ratio quantitative actual particles. consdrvption of matter coefficients subscripts mass

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10 2 Review And Reinforcement Answers

2 Answer.Key 2-1 Explore, page 5 The original source of energy for this activity would be the stored chemical energy in the dry cell battery. The chemical energy in the ... 2-1 Review and Reinforcement, pages 8-9 . 1. energy 2. radiant 3. chemical potential 4. joule 5. created, destroyed

6. natural gas 7. kinetic . 8. f . 9. c 10. b 11. a . 12. e .

2 Answer - Weebly

16—3 Review.nd Reinforcement, pages 20-2J 5. endothermic 6. true 7. increases 8. true 9. The Concentrations of reactants and products, pressure, and temperatuÈe affect the equilibrium 10. Le Chatelier's principle states that if a change is imposed on a system at equilibrium, the equilibrium position will shift in the

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10. The molecule NF3 has nitrogen as the central atom with the 3 fluorine atoms bonded to it. Predict the molecular shape and the bond angles in this molecule. What is the morecular shape of phospho- rustrichloridè (PC13)? What are the bond angles in this molecule? (1075) What is the molecular shape of OF? -What is the bond angle in this ...

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