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AP* Kinetics Free Response Questions page 2 Mechanism 3 is correct. The rate law shows that the slow reaction must involve one Y, consistent with mechanism 3.

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At the half-equivalence point (where the volume of the base added is exactly half its volume at the equivalence point), the concentration [HX] of the weak acid equals the concentration [X⁻] of its anion.

AP* Buffer Equilibrium Free Response Questions KEY

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AP* Buffer Equilibrium Free Response Questions page 6 Problems 1970 (a) What is the pH of a 2.0 molar solution of acetic acid. K_a acetic acid = 1.8×10^{-5} ? (b) A buffer solution is prepared by adding 0.10 liter of 2.0 molar acetic acid solution to 0.1 liter of a

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Judy Battista Judy Battista joined NFL Media in August 2013 after almost 15 years with the New York Times, where she covered the NFL nationally since 2004.

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