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of solution. Molality is the number of moles of solute per kilogram of solvent. 27. Solutions are homogeneous mixtures. Suspensions and colloids are heterogeneous mixtures. Solutions have the smallest particle size, followed by the size of colloid particles and suspension particles. Solutions and colloids do not settle out on standing, but ...

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A solution that contains less than the maximum amount of dissolved solute. A solution that contains more than the maximum amount of dissolved solute than a saturated solution under the same conditions. A solution that contains the maximum amount of a dissolved solute under different conditions.

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Chapter 13 Holt Chemistry. Vocabulary. STUDY. PLAY. solution. a homogeneous mixture of two or more substances uniformly dispersed throughout a single phase. ... a solution that contains less solute than a saturated solution and that is able to dissolve additional solute. supersaturated solutions.

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