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Figure 3 When an ionic compound dissolves in water, the charged ends of water molecules surround the oppositely charged ions. 1 2 3. H2O Figure 2 A stainless steel pot or pan is a solution of chromium and nickel in iron. In a solution, the solvent is the substance in the greatest quantity.

Section 8.1 8.1 Formation of Solutions - PC\|MAC

8.1 Formation of Solutions. Chapter 8: Solutions, Acids, and Bases. Dissolving. Any state of matter can become part of a solution. For a solution to form, one substance must dissolve in another (definition of solution) Solution. Substances can dissolve in three ways.

8.1 Formation of Solutions by Rachel Ward on Prezi

8.1 Formation of Solutions Notes Solutions ... Solution process is described as exothermic or endothermic. In order for a solution to form: o Attractions among solute particles and the attractions among solvent particles must be _____, which _____ energy, ...

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- 8.1.1 Describe how a substance can dissolve in water by dissociation, dispersion, or ionization.
- 8.1.2 Describe how the physical properties of a solution can differ from those of its solute and solvent. 8.1.3 Identify energy changes that occur during the formation of a solution.

Section 8.1 8.1 Formation of Solutions

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Chapter 8 Solutions, Acids, and Bases. Section 8.1 Formation of Solutions. (pages 228–234) This section explains the parts of a solution, the processes that occur when compounds dissolve, and how the properties of a solution compare with those of its solvent and solute. Reading Strategy (page 228)

Chapter 8 Solutions, Acids, and Bases Section 8.1 ...

8.1 FORMATION OF SOLUTIONS. SOLUTIONS A homogeneous mixture of two or more substances Homogeneous? Mixture? PARTS OF A SOLUTION Every solution has two parts: ... 8.1 Formation of Solutions Created Date: 2/10/2014 3:55:14 PM ...

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8.1 Formation of Solutions Reading Strategy Comparing and Contrastin Copy the Venn ... occur during the formation of a solution. 8.1.4 Describe factors affecting the Chapter 8 Solutions, Acids, and Bases Section 8.2 Solubility ...

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1. A solution is a mixture of two or more substances. 2. Circle the letter that identifies a substance whose particles are dissolved in a solution. a. solvent b. solute c. ion 3. The process in which an ionic compound separates into ions as it dissolves is called . Circle the correct answer. dispersion dissociation ionization 4.

Chapter 8 Solutions, Acids, and Bases Section 8.1 ...

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These often occur when a reaction happens Formation of a solid Formation of water Formation of a gas Transfer of electrons Can you give examples of each of these? B. Precipitation Reactions What Happens When an Ionic Compound Dissolves in Water? ... Section 8.1 Understanding Reactions in Aqueous Solutions * * * * * * * * * * * * Section 8.1 ...

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