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### **mc06se cFMsr i-vi**

REVIEW Chemical Bonding MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. a.

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CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

### **10 States of Matter - Ms. Agostine's Chemistry Page**

CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. The element boron, B, has an atomic mass of 10.81 amu according to the periodic table.

### **3 Atoms: The Building Blocks of Matter - Ms. Sheehan's ...**

CHAPTER 5 REVIEW The Periodic Law MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Consider the neutral atom with 53 protons and 74 neutrons to answer the following questions. 53 a. What is its atomic number? 127 b. What is its mass number? atomic number c. Is the element's position in a modern periodic table ...

### **5 The Periodic Law - Jefferson Township Public Schools**

CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Write formulas for the following compounds: CuCO<sub>3</sub> 3 a. copper(II) carbonate Na<sub>2</sub>SO<sub>3</sub> 3 b. sodium sulfite (NH<sub>4</sub>)<sub>3</sub>PO<sub>4</sub> 4 c. ammonium phosphate SnS<sub>2</sub> 2 d. tin(IV) sulfide HNO<sub>2</sub> 2 e. nitrous acid 2.

### **7 Chemical Formulas and Chemical Compounds**

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### **Holt McDougal Modern Chemistry Chapter 17: Reaction ...**

Modern Chemistry 1 The Periodic Law CHAPTER 5 The Periodic Law SECTION 1 History of the Periodic Table OBJECTIVES 1. Explain the roles of Mendeleev and Moseley in the development of the ... MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. ...

### **CHAPTER 5 The Periodic Law - mchsapchemistry.com**

SECTION 3 continued 4. Table 6 on page 485 of the text lists several amphoteric species, but only one other than water is neutral. NH<sub>3</sub> 3 a. Identify that neutral compound. b. Write two equations that demonstrate this compound's amphoteric properties. (Answers will vary, but one equation should form NH<sub>4</sub><sup>+</sup>)

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