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Stoichiometry: Mixed Problems (KEY) 1) $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$ What volume of NH_3 at STP is produced if 25.0 of N_2 is reacted with an excess of H_2 ? 3 3 3 2 3 2 2 40.0L NH_3 1mol NH_3 22.4L NH_3 1mol N 2mol NH_3 28.0g N 25.0g N 1mol N $\times \times \times =$ 2) $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$ If 5.0g of KClO_3 is decomposed, what volume of O_2 is produced at STP? 2

Stoichiometry: Mixed Problems (KEY)

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Chemistry IF8766 9 OInstructional Fair, Inc. 59. ... SIGNIFICANT FIGURES Name ? When multiplying and dividing, limit and round to the least number of significant figures in any of the factors. 11 Example 1 : $23.0 \text{ cm} \times 432 \text{ cm} \times 19 \text{ cm} = 188,784 \text{ cm}^3$ 11 The answer is expressed as $190,000 \text{ cm}^3$ since 19 cm has only two significant fiaures.

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