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SUBSTANCES VS. MIXTURES A substance is matter for which a chemical formula can be written. Elements and compounds are substances. Mixtures can be in any proportion, and the parts are not chemically bonded. Classify the following as to whether it is a substance or a mixture by writing S or M in the space provided. 1. sodium 2. water 3. soil 4 ...

SUBSTANCES VS. MIXTURES - Mendham Borough Schools

HALF-LIFE PROBLEMS Name Block 1. An isotope of cesium (cesium-137) has a half-life of 30 years. If 1,0 g of cesium-137 disintegrates over a period of 90 years, how many g of cesium-137 would remain?

HALF-LIFE PROBLEMS

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II. Answer these: ____ 1. the part of the atom that carries no electric charge ____ 2. the part of the atom that carries a positive charge ____ 3. the part of the atom that carries a negative charge ____ 4. the number of electrons that can be held in the first orbit (closest to the nucleus)

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