

## ***2 3 Carbon Compounds Answers***

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### 2 3 Carbon Compounds Answers

Section 2-3 Carbon Compounds (pages 44-48) This section explains how the element carbon is able to form millions of carbon, or organic, compounds. It also describes the four groups of organic compounds found in living things. The Chemistry of Carbon (page 44) 1. How many valence electrons does each carbon atom have? Each carbon atom has four electrons. 2.

### Macromolecules The Chemistry of Carbon

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### Biology 2.3 Carbon Compounds Worksheet Answers | Semesprit

SAMPLE ANSWER: Carbon is the primary element found in living things. SAMPLE ANSWER: Organisms use carbon compounds to form four types of molecules: lipids, carbohydrates, nucleic acids, and proteins. SAMPLE ANSWER: A lot of what that happens in an organism is based on chemical reactions.

### The Chemistry of Life - Weebly

Section 2-3 Carbon Compounds (pages 44-48) Key Concept • What are the functions of each group of organic compounds? The Chemistry of Carbon (page 44) 1. How many valence electrons does each carbon atom have? Each carbon atom has four electrons. 2. What gives carbon the ability to form chains that are almost unlimited in length?

### SC06 GRSW CH02 5/23/06 3:13 PM Page 19 Section 2-3 Carbon ...

Carbon Compounds Ch 2.3. 2. 2.3 Carbon Compounds A. The Chemistry of Carbon Organic any substance that contains carbon except CO and CO<sub>2</sub>. Carbon is very important to living things 1. Carbon has 4 Valence electrons Makes it very versatile Will easily bond with H, O, N, P, S..all important for living things 2.

### Carbon Compounds Ch 2 - Northern Highlands

Section 2-3 Carbon Compounds This section explains how the element carbon able to form millions of carbon, or organic, compounds. It describes the four groups of organic compounds found in living things. The Chemistry of Carbon How many valence electrons does each carbon atom have? Each carbon atom has four electrons.

### Section 2-3 Carbon Compounds - Teachers.Henrico Webserver

in Figure 2-11. Carbon has the ability to form millions of different large and complex structures .

No other element even comes close to matching carbon's versatility. 2-3 Carbon Compounds Key Concept • What are the functions of each group of organic compounds? Vocabulary monomer polymer carbohydrate monosaccharide polysaccharide lipid ...

### Section 2-3 2-3 Carbon Compounds

20 Name Class Date. Macromolecules. Amino acids are the monomers of proteins. Each amino acid has three distinct parts: an amino group, an R group, and a carboxyl group. An amino group has the formula  $\text{-NH}_2$ , a carboxyl group is  $\text{-COOH}$ , and the R group varies from one amino acid to another.

### 2.3 Carbon Compounds - isd2135.k12.mn.us

Name Class Date Section 2-3 Carbon Compounds (pages 44-48) Key Concept • What are the functions of each group of organic compounds? The Chemistry of Carbon (page 44) 1. How many valence electrons does each carbon atom have? 2.

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