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3—3 Review and Reinforcement. Modern Atomic Theory. Complete the following table. 1. 2. 3. If the statement is true, write "true." If it is false, change the underlined word or words to make the statement true. Write your answer on the line provided.

Name Date Class 3—3 Review and Reinforcement
11—2 Review and Reinforcement Solving Stoichiometry Problems If the statement is true, write "true." If it is false, change the underlined word or words to make it true. Write your answer on the line provided. 1. 2. 5. The major types of stoichiometry problems are mass-mass, mass-volume, and volume-volume.

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16—3 Review.nd Reinforcement, pages 20-2J 5. endothermic 6. true 7. increases 8. true 9. The Concentrations of reactants and products, pressure, and temperatuÈe affect the equilibrium 10. Le Chatelier's principle states that if a change is imposed on a system at equilibrium, the equilibrium position will shift in the

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10—2 Review and Reinforcement (continued) 9. 0.0621 mol SS = (Q $ gK2S 10. 65.8gMg= l. atoms Mg Solve each of the following problems as directed. Show all your work. 11. Calculate the number Of atoms in 0.40 mol of sulfur- 12. Calculate the number of atoms in 2.30 mol of silver. G.oux 13. You have a tank of NOZ gas.

10-2 Review and Reinforcements

Write your answer on the line provided s. 10. 12. The term stoichiometrv is derived from the words stoicheion, meaning element, and metron, meaning measure. You can determine the number of moles Of any substance produced in a reaction if vou know the number Of moles of at least of the reactants.

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Il—I Review and Reinforcement (continued) Solve each of the following problems as directed. Show all your work. 13. How many moles Of magnesium are required to react with 2.0 mol of hydrochloric acid? The equation for this reaction is Mg + 2HCI -+ MgC12 + H '2 Omo\ 14. Aluminum reacts with HCl to produce aluminum chloride (AIC13) and hvdrogen gas.

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SECTION 11-1 REVIEW AND REINFORCE The Water Cycle ^ Understanding Main ideas ... L 11. sludge 1, 12. sewage Science Explorer Earth Science I Unit 4 Resources 39 .

11 1 Review And Reinforce Science Answers
Name Date 3—3 Review and Reinforcement (continued) Use the periodic table to determine how many protons, neutrons, and electrons are present in each of the following atoms. Write your answers in the spaces provided. Atom 12. iodine-125 13.

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