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# Chemical Compounds Practice Quiz | Mr. Carman's Blog

This quiz can be used as a pre-test for a unit on atoms, molecules, elements, and compounds. The quiz will along with a post-test will serve as a method to determine how much the students have learned after the science unit has been taught.

#### Atoms, Elements, Molecules...pre-test - ProProfs Quiz

Holt Chemistry 14 Atoms and Moles Section: Electron Configuration In the space provided, write the letter of the term or phrase that best answers the ... Quiz—Section: Substances Are Made of Atoms 1. d 6. a 2. a 7. b 3. c 8. b 4. c 9. b 5. b 10. c Quiz—Section: Structure of Atoms 1. b 6. c 2. c7. 3. c 8. b 4. a 9. c

# **Assessment Quiz - Woodland Hills School District**

Some ionic compounds do not dissolve in water because a. the attractions between the water

molecules and the ions are not as strong as the attractions between the ions. b. they are less dense than water and float on the surface. c. water is a covalent compound and covalent compounds cannot dissolve ionic compounds.

# Assessment Chemical Bonding - clarkchargers.org

3. Molecules are stable when they achieve a noble gas configuration with an outer shell of 8 electrons. Hydrogen is the exception to this rule since it achieves a noble gas configuration when it has two electrons in its outer shell. 4. In a covalent bond, atoms share one or more pairs of electrons. In an ionic bond, one atom transfers electrons to another. 5.

#### Skills Worksheet Concept Review - Home - Default

Chapter Test A, continued \_\_\_\_\_ 7. The melting points of ionic compounds are higher than the melting points of molecular compounds because a. ionic substances tend to vaporize at room temperature. b. ionic substances are brittle. c. attractive forces between ions are greater than the attractive forces between molecules.

# Assessment Chapter Test A - Kettering City School District

Chemistry is the study of matter, and all matter is made up of atoms. We will learn about elements, atomic number and mass, isotopes, moles (chemistry moles, not the animal), and compounds. Learn for free about math, art, computer programming, economics, physics, chemistry, biology, medicine, finance, history, and more.

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