# 2012 Heat And Thermo 1 Answer Key

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# 2012 Heat And Thermo 1

A. \* How much heat (in J) is necessary to raise 1 kg of steam 1 degree Celsius? B. \* How much heat is necessary to raise 1 kg of steam 2 degrees Celsius? 5. Ice has a specific heat of 2090 J/kg $\,^{\circ}$ C $^{\circ}$ . How much heat is necessary to raise 1 kg of ice 1 degree Celsius? 6. Water has a specific heat of 4186 J/kg $\,^{\circ}$ C $^{\circ}$ .

# **Heat and Thermo 1 - Mr Murray's Science and Music**

2012 Heat And Thermo 1 Thermoregulation is the ability of an organism to keep its body temperature within certain boundaries, even when the surrounding temperature is very different. A thermoconforming organism, by contrast, simply adopts the surrounding temperature as its own

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#### Ch 10 11 Heat. Thermo 1 Heat

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Due Fri., Mar 2 2012 Heat and Thermo 6 1. A piston moves in, so that 3500J of work is done to compress the gas. A. Since the gas is compressed, the work done on the gas = B. If the gas gives up 1200 J of heat to the surroundings, Q = C. \* What is the change of energy of the gas? ... 3/1/2012 1:59:47 PM ...

# 2012 Heat and Thermo 6 - Mr Murray's Science and Music

Thermo Heat is the #1 Maximum Strength Fat Loss Supplement that boosts your metabolism, helps manage your appetite, enhances your energy, and mental alertness. The key to fat loss is burning more calories than you consume. However, the longer you sustain a calorie deficit, the more likely you are to degrade your metabolism, which can slow your ...

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THERMOCHEMISTRY CALCULATIONS WORKSHEET 1! 1. What is the heat change when 6.44 g of Sulfur reacts with excess O 2 according to the following equation?  $2S + 3O 2 \rightarrow 2SO 3 \Delta H^{\circ} = -791.4 kJ$  Is this Endothermic or Exothermic 2. What is the heat change when 4.72 g of Carbon reacts with excess O 2 according to the following equation?  $C + O 2 \rightarrow CO$ 

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