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AP* Solution Chemistry Free Response KEY page 6 1994 (a) volume of water decreases while the concentration of sugar solution decreases. Pure water has a higher vapor pressure than does the 10% sugar solution and when equilibrium is reached

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At the half-equivalence point (where the volume of the base added is exactly half its volume at the equivalence point), the concentration $[HX]$ of the weak acid equals the concentration $[X^-]$ of its anion.

AP* Buffer Equilibrium Free Response Questions KEY

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AP* Kinetics Free Response Questions page 2 Mechanism 3 is correct. The rate law shows that the slow reaction must involve one Y, consistent with mechanism 3. Mechanisms 1 and 2 would involve both [X] and [Y] in the rate law, not consistent with the rate law. 1987 a) three points; one each for form of rate law, HgCl_2 exponent, $\text{C}_2\text{O}_4^{2-}$ exponent

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Practicing free-response questions is one of the best things you can do to improve your AP Chemistry score. Not only will you excel on the free-response section, but you'll also know the material so well that the multiple-choice questions will be a piece of cake.

AP Chemistry FRQ: How to Ace the Free Response Questions

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- Graphing calculators have been required on the AP Calculus Exams since 1995. From 1995-1999, the calculator could be used on all 6 free-response questions. Since the 2000 Exams, the free-response section has consisted of two parts -- Part A (questions 1-3) requires a graphing calculator and Part B (questions 4-6) does not allow the use of a

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