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2)  $2 \text{ KClO}_3 \rightarrow 2 \text{ KCl} + 3 \text{ O}_2$  3)  $2 \text{ NaCl} + 1 \text{ F}_2 \rightarrow 2 \text{ NaF} + 1 \text{ Cl}_2$  4)  $2 \text{ H}_2 + 1 \text{ O}_2 \rightarrow 2 \text{ H}_2\text{O}$  5)  $1 \text{ Pb(OH)}_2 + 2 \text{ HCl} \rightarrow 2 \text{ H}_2\text{O} + 1 \text{ PbCl}_2$  6)  $2 \text{ AlBr}_3 + 3 \text{ K}_2\text{SO}_4 \rightarrow 6 \text{ KBr} + 1 \text{ Al}_2(\text{SO}_4)_3$

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Naming Ionic Compounds Directions: Name the following compounds using the Stock Naming System. 1.  $\text{CaCO}_3$  calcium carbonate\_\_\_\_\_ 2.  $\text{KCl}$  potassium chloride\_\_\_\_\_ 3.  $\text{FeSO}_4$  iron (II) ... Chemistry IF8766 Page 45 Instructional Fair, Inc. Title: Name: Author: ASD Created Date:

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