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Ch 19 Acids Bases and Salts Worksheet - Afrimarine

Acids and bases always come in pairs. HCl is an acid. When it dissolves in water, it gives it's proton to water. Water is a base; makes hydronium ion. Ammonia can be explained as a base by using Brønsted-Lowry: Ammonia is the hydrogen ion acceptor (base), and water is the hydrogen ion donor (acid).

Chapter 19 Acids, Bases, and Salts - Lacey, WA / Welcome!

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Chapter 19 Acids, Bases, and Salts. STUDY. PLAY. acid dissociation constant (K_a) the ratio of the concentration of the dissociated form of an acid to the undissociated form; stronger acids have larger K_a values than weaker acids (19.3) acidic solution.

Chapter 19 Acids, Bases, and Salts Flashcards | Quizlet

Acids, Bases and Salts - Revision Worksheet. 1. Classify each of the following as an acid, base or salt. (1 mark each) a) H_2SO_4 b) $CaCl_2$ c) $Cu(NO_3)_2$ d) HNO_3 ...

Acids Bases and Salts - Worksheet #1 answers - Chemalive - MAFIADOC.COM

19.1.1 Define the properties of acids and bases. 19.1.2 Compare and contrast acids and bases as defined by the theories of Arrhenius, Brønsted-Lowry, and Lewis. Word Parts Explain that the prefixes mono, di, and tri have Greek origins, meaning one, two, and three, respectively.

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