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Metallic Bonds and Metallic Properties 1. Is the following sentence true or false? Metals are made up of cations and valence electrons, not neutral atoms. 2. What are metallic bonds? 3. Name three properties of metals that can be explained by metallic bonding. a. b. c. 4. What happens to an ionic crystal when a force is applied to it? 5.

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Chapter 7 Ionic and Metallic Bonding59 SECTION 7.1 IONS (pages 187–193) This section explains how to use the periodic table to infer the number of valence electrons in an atom and draw its electron dot structure.

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7.1 lons > 15 Draw the electron dot structure for bismuth. Lithium Fluorine Neon Bi Li F Ne

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f. ionic compound g. ionic bonds h. coordination number i. chemical formula j. valence electrons k. metallic bond Column A compound composed of cations and anions the forces of attraction that bind oppositely charged ions together lowest whole-number ratio of ions in an ionic compound a depiction of the valence electrons as dots around the ...

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metallic crystal structures. Metallic Bonds and Metallic Properties (pages 201–202) 1. Is the following sentence true or false? Metals are made up of cations, not neutral atoms. _____ 2. What are metallic bonds? 3. Name three properties of metals that can be explained by metallic bonding. a. b. c. 4. What happens to an ionic crystal when a ...

SECTION 7.1 IONS (pages 187-193)

Chapter 7 Ionic and Metallic Bonding 155 Section Review Objectives • Determine the number of valence electrons in an atom of a representative element • Explain the octet rule • Describe how cations form • Explain how anions form Vocabulary Part ACompletion Use this completion exercise to check your understanding of the concepts and terms

05 CTR ch07 7/9/04 3:27 PM Page 155 IONS 7

The Ionic and Metallic Bonding chapter of this Prentice Hall Chemistry Companion Course helps students learn the essential lessons associated with ionic and metallic bonding.

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lonic and Metallic Bonding 201 ... Metallic Bonds and Metallic Properties helped you learn something new. Metals are made up of closely packed cations rather than neutral atoms. The valence electrons of metal atoms can be modeled as a sea of elec-trons. That is, the valence electrons are mobile and can drift freely from ...

7.3 Bonding in Metals 7 - Henry County School District

Metallic Bonds and Metallic Properties When a metal is subjected to pressure, the metal cations easily slide past one another. If an ionic crystal is struck with a hammer, the blow tends to push the positive ions close together.

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IONIC BONDING AND IONIC COMPOUNDS CHAPTER TEST A 15 ... d. form ionic bonds. _____13. In forming chemical bonds, atoms tend to attain: ... Explain how scientists have used the concept of metallic bonding to account for many of the physical properties of metals, such as electrical conductivity and ...

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