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19. 1; 1 ; in the nucleus 20. 1; 0; in the nucleus 21. 0; 1 ; outside the nucleus (in the elec-tron cloud) 22. Answers should include three of the

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Modern Chemistry 171 Nuclearchemistry CHAPTER 21 REVIEW Nuclear Chemistry SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. ____ The nuclear equation is an example of an equation that represents (a) alpha emission. (b) beta emission. (c) positron emission. (d) electron capture. 2.

CHAPTER 21 REVIEW Nuclear Chemistry

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These lecture presentations were designed for my high school Chemistry I Honors class. Students of high school and college general chemistry may find them useful as a supplement to their own class notes or as a review. Teachers, please feel free to use and modify them for your own classes.

Mrs. J's Chemistry Page - Lecture Notes

Nuclear Chemistry Kinetics of Radioactive Decay A wooden object from an archeological site is subjected to radiocarbon dating. The activity of the sample that is due to ^{14}C is measured to be 11.6 disintegrations per second.

Chapter 21 Nuclear Chemistry - University of Massachusetts ...

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