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are high-energy, high-speed electrons or positrons emitted by certain types of radioactive nuclei such as potassium-40

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25.2 Nuclear Transformations 25 - schoolwires.henry.k12.ga.us

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Radioactivity •Radioactivity is the process by which nuclei emit particles and rays as they break down. •The name of the penetrating rays emitted by a radioactive source is called radiation. •A radioactive isotope is an unstable atom which breaks down on its own, releasing energy and/or

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806 Chapter 25 Nuclear Chemistry Figure 25-2 Both Pierre and Marie Curie played important roles in founding the field of nuclear chemistry. Marie Curie went on to show that unlike chemical reactions, radioactivity is not affected by changes in physical conditions such as temperature and pressure. She is the only person in history to receive

Chapter 25: Nuclear Chemistry - Jayne Heier

25: NUCLEAR CHEMISTRY. 25.1 PPT-NUCLEAR RADIATION; 25.2 PPT-NUCLEAR TRANSFORMATIONS; 25.3 PPT-NUCLEAR FISSION & FUSION; Unit 1B Nuclear Chemistry Vocabulary; 25 Enrichment; 5: ELECTRON ARRANGEMENT IN ATOMS. Ch 05 Quiz Review

Ramamoorthy, Ramesh — Grace King High School

Chapter 25 Nuclear Chemistry 669 Practice Problems In your notebook, solve the following problems. SECTION 25.1 NUCLEAR RADIATION 1. What happens to the mass number and atomic number of an atom that

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