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the mercury level in each tube is 13.2 cm from the top. What is the largest difference in pressure this manometer can measure in units of mm Hg? A manometer contains a sample of nitrogen gas at a pressure of 88.3 kPa. The level of mercury in the U-tube is 12.8 mm lower on the end open to the atmosphere. What is the atmospheric pressure in kPa?

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At 75°C, a gas has a volume of 3.22 dm³. What volume will it occupy at 75 K? A discarded spray paint can contains only a small volume of the propellant gas at a pressure of 34,470 Pa. The volume of the can is 473.18 mL. If the can is run over by the garbage truck and flattened to a volume of 13.16 mL, what is the pressure in atm?

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6.13.2. Practice Problems - Chemistry LibreTexts

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Practice Problem Solutions SOLUTIONS TO THE PRACTICE PROBLEMS FOR MODULE I. The chemical equation contains no phases, so we must use bond energies to solve this problem. This means we start with Lewis structures: o Using this picture and Table 13.1, we know that two Cl-Cl bonds (each worth 240 kJ/mole) and four

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