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17-2 Kinetic Theory The kinetic theory relates microscopic quantities (position, velocity) to macroscopic ones (pressure, temperature). Assumptions: • N identical molecules of mass m are inside a container of volume V; each acts as a point particle. • Molecules move randomly and always obey Newton's laws.

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Summary of Chapter 26 •A wave front is a surface along which the wave phase is constant. Rays, perpendicular to the wave fronts, indicate the direction of propagation. •The angle of incidence equals the angle of reflection. •The image formed by a plane mirror is upright, but appears reversed left to right; appears to be

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James S. Walker James Walker obtained his Ph.D. in theoretical physics from the University of Washington in 1978. He subsequently served as a post-doc at the University of Pennsylvania, the Massachusetts Institute of Technology, and the University of California at San Diego before joining the physics faculty at Washington State University in 1983.

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