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26. According to the kinetic-molecular theory, the particles in a liquid can change relative positions but still are influenced by attractive forces.

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40. 7.05 1016 Hz 41. 1.28 s 42. 4.58 10 19 J The Periodic Law, pp. 36–45 TEST A 1. b 2. d 3. b4. 5. d

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SECTION 1 continued 6. Some binary compounds are ionic, others are covalent. The type of bond favored partially depends on the position of the elements in the periodic table.

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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Write formulas for the following compounds: CuCO 3 a. copper(II) carbonate Na 2SO 3 b. sodium sulfite (NH 4) 3PO 4 c. ammonium phosphate SnS 2 d. tin(IV) sulfide HNO 2 e. nitrous acid 2.

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## Answers

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