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dissolves. The ions, already present, separate from one another. Ionization is the process of forming ions from the solute molecules by the action of the

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40. 7.05 1016 Hz 41. 1.28 s 42. 4.58 10 19 J The Periodic Law, pp. 36-45 TEST A 1. b 2. d 3. b4. 5. d 6. a 7. b8. 9. b 10. a 11. c 12. a 13. c 14. d 15. c 16. b

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water to produce a sulfuric acid solu-tion that falls to the ground as rain or snow. SO $3(g) H 2O(I) \rightarrow H 2SO 4(aq) 37$. Have a sour taste; change the color of acid-base indicators; some react with

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Calculations in Chemistry 1. b 6. a 2. b 7. a 3. a 8. b 4. b9. 5. a 10. d Chapter Test 1. a11. 2. c 12. b 3. b13. 4. b 14. c 5. b15. 6. d 16. c 7. c 17. d 8. c 18. b 9. d 19. b 10. c 20. b 21. Heat is the total amount of kinetic energy of all the particles in a system, while temperature is a measure of the average kinetic energy of the ...

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Assessment Chapter 5 Pre-Test Chapter: The Periodic Law ... Modern Chemistry 2 The Periodic Law Chapter 5 Pre-Test, continued _____ 3. The periodic law states that a. no two electrons with the same spin can be found in the same place in an atom. b. the physical and chemical properties of the elements are functions of ...

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