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Best answer: Chloroform - CHCl3 Chloroform is a clear liquid with an ether-like odor and a slightly sweet taste. It is a naturally-occurring chemical, but most of the chloroform in the environment is man-made. ... Chloroform evaporates quickly.

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Modern Chemistry 7 Chemical Equations & Reactions CHAPTER 8 STUDY GUIDE Chemical Equations and Reactions SECTION 8-3 SHORT ANSWER Answer the following questions in the space provided. 1. List four metals that will not replace hydrogen in an acid.

CHAPTER 8 Chemical Equations and Reactions

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Worked Chemistry Problem Examples - ThoughtCo

Chemistry Section Review 5.2 Multiple Choice Identify the choice that best completes the statement or answers the question. Put the LETTER of the correct answer in the blank. ____ 1. Argon, krypton, and xenon are a. alkaline earth metals. c. actinides. b. noble gases. d.

Chemistry Section Review 5 - Weebly

A chemistry student working in the lab might be asked to calculate how much 1-bromo-2-methylpropane, C 4H ... The answer is 2.50×104 kg P 4O 10. We used the following steps: Balance chemical equations. (Section 4.1.) ... = 9.52×103 kg H 2O 1 kg 103 g objeCtive 3 objeCtive 4

Chapter 10 ChemiCal alCulations and equations - Mark Bishop

of study in the field of chemistry. Chemistry is usually divided into five branches, as summarized in the table below. 1.3 Scientific Methods A scientific methodis a systematic approach used to answer a ques-tion or study a situation. It is both an organized way for scientists to do research and a way for scientists to verify the work of other sci-

Solving Problems: A Chemistry Handbook - MARRIC

answer with justification. (b) As the mass of the Sn electrode decreases, where does the mass go? The atoms on the Sn electrode are going into the solution as Sn2+ ions. 1 point is earned for the correct answer. (c) In the expanded view of the center portion of the salt bridge shown in the diagram below, draw and label

AP Chemistry 2014 Scoring Guidelines - College Board

General Chemistry 5.2 Review Worksheet: What are the charges fo ions formed from elements of the following groups? a) Group 1 1+ b) Group 2 2+ c) Group 16 2-d) Group 17 1- ... Answer: Sodium Bromide. Ionic compounds are generally hard and brittle. Motor oil is a liquid thus it is neither.

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