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These lecture presentations were designed for my high school Chemistry I Honors class. Students of high school and college general chemistry may find them useful as a supplement to their own class notes or as a review. Teachers, please feel free to use and modify them for your own classes.

Mrs. J's Chemistry Page - Lecture Notes

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provided, write the letter of the term or phrase that best completes each statement or best answers each question. ____ 1.

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26. According to the kinetic-molecular theory, the particles in a liquid can change relative positions but still are influenced by attractive forces.

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40. 7.05 1016 Hz 41. 1.28 s 42. 4.58 10 19 J The Periodic Law, pp. 36–45 TEST A 1. b 2. d 3. b4. 5. d 6. a 7. b8. 9. b 10. a 11. c 12. a 13. c 14. d 15. c 16. b

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Modern Chemistry 136 Chapter Test Name Class Date Chapter Test A, continued ____14. A pH meter measures the pH of a solution by measuring the a. amount of charge on the hydronium ion. b. color of the solution. c. concentration of the solution. d. voltage difference between two electrodes in the meter's probe. ____15. When you conduct an ...

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