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AIEEE-2010 CHEMISTRY, PHYSICS & MATHS 23. The Gibbs energy for the decomposition of Al_2O_3 at 5000C is as follows : $1232 \text{ r } 24 \text{ Al O Al O}$, $G 966 \text{ kJmol}$ 33 m % The potential difference needed for electrolytic reduction of Al_2O_3 at 5000C is at least 1) 4.5 V 2) 3.0 V 3) 2.5 V 4) 5.0 V 24. At 250C, the solubility product of Mg(OH)_2 is 1.0×10^{-10} ...

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