

Acid Bases And Salts Practice Problems Answers

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Unit 4: Acids & Bases - Ms. Popov - jpopov.weebly.com

An Arrhenius acid is therefore any substance that ionizes when it dissolves in water to give the H^+ , or hydrogen, ion.. An Arrhenius base is any substance that gives the OH^- , or hydroxide, ion when it dissolves in water.. Arrhenius acids include compounds such as HCl , HCN , and H_2SO_4 that ionize in water to give the H^+ ion. Arrhenius bases include ionic compounds that contain the OH^- ion ...

Definitions of Acids and Bases, and the Role of Water

An acid dissociation constant, K_a , (also known as acidity constant, or acid-ionization constant) is a quantitative measure of the strength of an acid in solution. It is the equilibrium constant for a chemical reaction known as dissociation in the context of acid-base reactions. $HA \rightleftharpoons H^+ + A^-$. The chemical species HA , A^- , and H^+ are said to be in equilibrium when their concentrations ...

Acid dissociation constant - Wikipedia

An oxyacid, or oxoacid, is an acid that contains oxygen. Specifically, it is a compound that contains hydrogen, oxygen, and at least one other element, with at least one hydrogen atom bond to oxygen that can dissociate to produce the H^+ cation and the anion of the acid.

Oxyacid - Wikipedia

Recommendations for Students and Parents. Chemistry can be a very challenging class for some of our students. We have a larger proportion of the student body taking chemistry than any other public school in the area.

Chemistry Homepage - ScienceGeek.net

**Take note that qualitative analysis is a very useful and important technique used by chemists to find out the identity of an aqueous substance by testing for the type of cation and anion found in the aqueous solution of the substance.

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