

## ***13 3 Review And Reinforcement The Gas Laws Answer Key***

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### 13 3 Review And Reinforcement

3—3 Review and Reinforcement (continued) Use the periodic table to determine how many protons, neutrons, and electrons are present in each of the following atoms. Write your answers in the spaces provided. Atom Protons Neutrons Electrons 12 iodine 125 \_\_\_\_ 13. potassium-39 / / 14. iron-56 \_\_\_\_ 3 Write the chemical symbol for each of the ions ...

### Name Date Class 3—3 Review and Reinforcement

3-1 review and reinforcement. STUDY. PLAY. Matter is neither created or destroyed in chemical reactions. Lavoisier. A given compound always has the same relative numbers and kind of atoms. Dalton. All atoms of a given element are identical, but they differ from those of any other element.

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13. Why might a construction company build a high-rise building with steel alloy beams instead of pure . iron beams? \_\_\_\_ 14. Water and ethanol are miscible in all proportions, whereas oil and water are immiscible. Explain this . statement.

### 13-1 Review and Reinforcement - North Penn School District

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10—2 Review and Reinforcement (continued) 9. 0.0621 mol SS = (Q \$ gK2S 10. 65.8gMg= l. atoms Mg Solve each of the following problems as directed. Show all your work. 11. Calculate the number Of atoms in 0.40 mol of sulfur- 12. Calculate the number of atoms in 2.30 mol of silver. G.oux 13. You have a tank of NOZ gas.

### 10-2 Review and Reinforcements

16—3 Review and Reinforcement, pages 20-2J 5. endothermic 6. true 7. increases 8. true 9. The Concentrations of reactants and products, pressure, and temperature affect the equilibrium 10. Le Chatelier's principle states that if a change is imposed on a system at equilibrium, the equilibrium position will shift in the

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3-1 Review and Reinforcement . Early Models of the Atom . On the line at the left, write the letter of the chemist who proposed each of the ideas listed below. Each letter may be used once, more than once, or not at all. a. Democritus b. Lavoisier c. Proust d. Dalton 1. Matter is neither created nor destroyed . in . chemical reactions. 2.

### **3-1 Review and Reinforcement - MRS. Nairn - HOME**

13—3 Review and Reinforcement The Gas Laws Match each chemist to one of the gas laws 1. The pressure Of a gas is inversely proportional to its volume. 2. The sum of the partial pressures of gases in a mixture is equal to the total pressure of the mixture. 3. Equal volumes of gases contain equal numbers Of particles. 4.

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