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accept a positive hydrogen ion. Thus, an acid-base reaction in the reverse direc-tion can occur. The acid H 3 O can react with the base X to form water and HX and the following equilibrium is established. 598 Chapter 19 Acids and Bases HX(aq) H 2 O(1) 3 H X (aq) 3 O (aq) Acid Base Conjugate acid Conjugate base

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Chapter 19"Acids, Bases, and Salts ... Figure 19.22, page 615. Neutralization. The solution of known concentration is called the standard solution. added by using a buret. Continue adding until the indicator . changes color.

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Acids: taste sour, react with certain metals to produce hydrogen gas, react with carbonates/bicarbonates to produce carbon dioxide gas.Corrode metals, pH is less than 7, turns blue litmus paper red. Bases: taste bitter, have a slippery feeling.React with acids to form salts and water, pH is greater than 7.

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Chapter(11:(Acids(and(Bases((Forourpurposes,(an(acid(isasubstancethatproduceshydrogenion(H+)when(dissolved(in(water.((Abase(isasubstancethatproduceshydroxideion ...

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634 Chapter 18 • Acids and Bases Section 118.18.1 Figure 18.1 Rhododendrons flourish in rich, moist soil that is moderately acidic, whereas semper-vivum, commonly called hen and chicks, grows best in drier, slightly basic soil. Introduction to Acids and Bases-!).)DEADifferent models help describe the behavior of acids and bases.

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