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10—2 Review and Reinforcement (continued) 9. $0.0621 \text{ mol SS} = (Q \text{ } \$ \text{ gK2S } 10. 65.8 \text{ gMg} = \text{I. atoms Mg}$ Solve each of the following problems as directed. Show all your work. 11. Calculate the number of atoms in 0.40 mol of sulfur- 12. Calculate the number of atoms in 2.30 mol of silver. G.oux 13. You have a tank of NO₂ gas.

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radiant 3. chemical potential 4. joule 5. created, destroyed 6. natural gas 7. kinetic ... 11. filtration
12. filtration 13. crystallization 14. distillation 15. chromatography

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11—1 Review and Reinforcement Stoichiometry Complete the following sentences by 'filling in the appropriate word or phrase from the list below. Each word or phrase maybe used once, more than once, or not at all reactants molar ratio quantitative actual particles. consdrvption of matter coefficients subscripts mass

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