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In your answer, consider the external and internal pressure data given in Model 1. Read This! Pressure is caused by molecules hitting the sides of a container or other objects. The pressure changes when the molecules change how often or how hard they hit. A nonflexible container is needed if the gas ... 25 Gas Variables-S

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Gas Variables 5 13. Complete the following table for the three experiments in Model 2. Experiment C Experiment D Experiment E Independent Variable Dependent Variable Controlled Variable(s) 14. Provide a molecular level explanation for the increase in volume among the balloons in Experiment C

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Consider the following changes imposed upon a sample of gas, assuming the variables ... increases from 25 C to 323 C? (Note: in the answers ... If the pressure on the ...

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Dependent Variable Controlled Variable(s) 5. Of the variables that were controlled in both Experiment A and Experiment B in Model 1, one requires a nonflexible container. Name this variable, and explain why a nonflexible container is necessary. In your answer, consider the external and internal pressure data given in Model 1. Read This!

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For an ideal gas, which two variables are directly proportional to each other (if all other conditions remain constant)? P and V P and n P and T 2 only 2 and 3 only 1 and 2 only 1 only 3 only A given mass of gas occupies a volume of 4.00 L at 60 degree C and 550 mmHg.

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