

23 Mole Ratio Pogil Answers

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is the correct answer. SHOW the math below. 37. Given the following equation, how many L of H_2 gas are produced from 3.5 g HCl at STP? $\text{Ca (s)} + 2\text{HCl (aq)} \rightarrow \text{CaCl}_2\text{(aq)} + \text{H}_2\text{(g)}$ 38. Look at the purpose of the POGIL on the front page and answer the question. Molar Mass Mole Ratio Molar Volume Molar Mass NH_3 $14.01 \text{ g/mol} + (3 \times 1.01 \text{ g/mol}) = 17.04 \text{ g/mol}$

POGIL- Stoichiometry

Chem 115 POGIL Worksheet - Week 4 Moles & Stoichiometry Answers Key Questions & Exercises 1. The atomic weight of carbon is 12.0107 u, so a mole of carbon has a mass of 12.0107 g.

Moles & Stoichiometry Answers Key Questions & Exercises

Mole Ratios 1 Mole Ratios How can the coefficients in a chemical equation be interpreted? Why? A balanced chemical equation can tell us the number of reactant and product particles (ions, atoms, molecules or formula units) that are necessary to conserve mass during a chemical reaction. Typically when we balance the chemical equation we think in terms of individual particles.

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