

## ***14 3 Review And Reinforcement Answers***

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### 14 3 Review And Reinforcement

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### 14 3 REVIEW AND REINFORCEMENT ANSWERS

14 3 Review And Reinforcement Answers Name date class 3—3 review and reinforcement, date 3—3 review and reinforcement modern atomic theory complete the following table 1 2 3 if the statement is true, write “true” if it is false, change the underlined word or words to make.

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### 14 3 REVIEW AND REINFORCEMENT ANSWER KEY

3—3 Review and Reinforcement (continued) Use the periodic table to determine how many protons, neutrons, and electrons are present in each of the following atoms. Write your answers in the spaces provided. Atom Protons Neutrons Electrons 12 iodine 125 \_\_\_\_ 13. potassium-39 // 14. iron-56 \_\_\_\_ 3 Write the chemical symbol for each of the ions ...

### Name Date Class 3—3 Review and Reinforcement

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14 3 Review And Reinforcement Answer Key 2 answer weebly, 2 answerkey 2 1 explore, page 5 the original source of energy for this activity would be the stored chemical energy in the dry cell battery the chemical

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14 3 Review And Reinforcement Abstract: The framework of reinforcement learning or optimal control provides a mathematical formalization of intelligent decision making that is powerful and broadly applicable.

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16—3 Review and Reinforcement, pages 20-2] 5. endothermic 6. true 7. increases 8. true 9. The Concentrations of reactants and products, pressure, and temperature affect the equilibrium 10. Le Chatelier's principle states that if a change is imposed on a system at equilibrium, the equilibrium position will shift in the

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24—1 Review and Reinforcement (continued) 13. There are 3.29 g of iodine-126 remaining in a sample originally containing 26.3 g of iodine-126. The half-life of iodine-126 is 13 days. How old is the sample? 14. A sample that originally contained 2.5 g of rubidium-87 now contains 1.25 g. The half-life of rubidium-87 is  $6 \times 10^{10}$  years.

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