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1) $3.500 \text{ kg x } 0.0362 \text{ mole} = 0.1267 \text{ moles. kg. 2}) 0.1267 \text{ moles x } 35.45 \text{ g} = 4.49g. \text{ mole. Answer each of the following questions in the space provided. Describe how a chemist can accurately prepare a solution of precise molarity. Measure the grams of a solute. Divide by the molar mass to determine the number of moles. Then add the solute to <math>1 \dots$

15-2 Review and Reinforcement - Mr. Fischer

15. Why is water called the universal solvent? _____ 16. How could you determine experimentally whether a given substance is an electrolyte? ... Title: 13-1 Review and Reinforcement Author: NPSD Last modified by: NPSD Created Date: 9/27/2005 1:14:00 PM Company: NPSD

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Estimated weight of peach seed: 1-2 grams; peach tree: 20-22 kg. 2. Yes. 3. The tree uses water and nutrients from the ground and carbon dioxide from the air to build new cells, increasing the mass of the peach tree.

2 Answer - Weebly

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15. The forward and reverse reactions are both occurring, but with no net change in concentration. -16—2 Review Reinforcement, pages 15-16 (H2J[12] - [NH3JIHC1] 9. Keg - [Asd[C016 10. = ICO [C02] 12. 13. No; toward the reactants. 14. No; toward the products: 15. No; toward the reactants. 16. a 18. b 19. There is no effect. 20.

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15. Even though water and carbon dioxide both have polar bonds, they have different shapes. Carbon dioxide is linear, while water is bent. This makes water polar and carbon diox- ide nonpolar. 16. Polar sidechains are on the outside of the protein molecule because they interact with cytoplasm, which contains a large amount of water, a polar ...

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Name II—I Review and Reinforcement (continued) Solve each of the following problems as directed. Show all your work. 13. How many moles Of magnesium are required to react with 2.0 mol of hydrochloric acid?

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