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e. electrolysis of molten potassium chloride decomposition; $2KCI(I) \rightarrow 2K(s) CI2(g) 4.)$ (I 2 4((()) (((() (((() + (()

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provided, write the letter of the term or phrase that best completes each statement or best answers each question. _____ 1.

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Modern Chemistry 2 Chemical Equations and Reactions 6. Gold has a low reactivity and therefore does not corrode over time. 7. In single-displacement reactions, if the activity of the free element is greater than that of the element in the compound, the reaction will take place. 8. Yes; because aluminum is above copper in

CHAPTER 8 REVIEW Chemical Equations and Reactions

Modern Chemistry 32 Chapter Test Name Class Date Chapter Test B, continued _____ 6. When electrons change energy states, the amount of energy given off or absorbed is equal to a. hc. b. hv. c. n. d. cv. _____ 7. Max Planck proposed that "hot" objects radiate energy in small, specific amounts called a. quanta. b. waves. c. hertz. d ...

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