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All of these reactions are examples of oxidation and reduction, otherwise known as redox reactions. In this activity you will explore what is happening at the atomic level in redox reactions. Model 1 – Redox Reactions ... from Model 1 to support your answer. 2 POGIL ...

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there any features of the redox reactions that would allow you to identify them as redox reactions? If y, specifi es m Model 1 to support your answer. 6. Identify the following reactions as either redox or non ox using oxidation numbers as evidence. a. 2Na1(aq) Pb12(s) 2NaN03(aq) + 202 d HCI + NaOH -+ HO +

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Oxidation and Reduction 3. Read This! The process of oxidation and reduction can be thought of as a transfer of electrons from one atom to another. Thus, one atom gives up electrons and the other atom gains them. As a result of this process, the oxidation numbers of both atoms change. All redox reactions can be divided up into two reactions—an

# Oxidation and Reduction - Skyline Chemistry

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