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4 Chapter 1 • Introduction to Chemistry Section 11.1.1 Figure 1.1 Everything in the universe, including particles in space and things around you, is composed of matter. Objectives Define substance. Explain the formation and

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SECTION 1.3 THINKING LIKE A SCIENTIST (pages 20–25) This section describes the development of an experimental approach to chemistry and steps involved in the scientific method. It also discusses the role of collaboration and communication in science. Alchemy (page 20) 1. Practical alchemy focused on 2.

### **SECTION 1.1 CHEMISTRY (pages 7–11)**

2 Chapter 1 Introduction to Chemistry CHAPTER 1 Visit the Chemistry Web site at chemistrymc.com to find links about chemistry and matter. The four nebulae shown here contain a stew of elements. The red color in two of the nebulae is emitted by hydrogen atoms. The Horsehead Nebula can be seen on the right. The fourth nebula is the bluish ...

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matter and energy, and carbon chemistry. Chemistry affects all aspects of life and most natural events because all living and nonliving things are made of matter. Five traditional areas of study are organic chemistry, inorganic chemistry, biochemistry, analytical chemistry, and physical chemistry. 1.1 The Scope of Chemistry >

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Chapter 1 Introduction to Chemistry 7 Section Review Objectives • Identify five traditional areas of study in chemistry • Relate pure chemistry to applied chemistry • Identify reasons to study chemistry Vocabulary Part A Completion Use this completion exercise to check your understanding of the concepts and terms

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