

```
In [9]: import numpy as np

import pandas as pd
```

```
In [4]: from metas_unclib import *
```

```
In [6]: water_tap = ufloatfromsamples(
    (289.5, 281.1, 278.5),
    desc="water tap"
)
water_tap
```

Out[6]: 283.03333333333336 ± 7.2867724653006665

```
In [7]: water_dei= ufloatfromsamples(
    (2.84, 2.9, 2.8),
    desc="water deionized"
)
water_dei
```

Out[7]: 2.8466666666666662 ± 0.0637931056763051

```
In [8]: water_pure = ufloatfromsamples(
    (1, 1.2, 0.97),
    desc="water purified, degased"
)
water_pure
```

Out[8]: 1.0566666666666666 ± 0.15847230355837524

```
In [14]: print(pd.DataFrame(np.array(((1,2),(3,4)))).to_latex())
```

```
\begin{tabular}{lrr}
\toprule
{} & 0 & 1 \\
\midrule
0 & 1 & 2 \\
1 & 3 & 4 \\
\bottomrule
\end{tabular}
```

Molar

K₂CO₃

```
In [20]: kalium_M = ufloat(138.205, desc="molar mass potassium carbonate") # g / mol
kalium_m = ufloat(1.3895, 0.0001, desc="mass potassium carbonate")
kalium_c = kalium_m / kalium_M / 0.1 / 10
kalium_c
```

Out[20]: 0.010053905430338989 ± 7.235628233421367e-07

```
In [24]: kalium_lambda = ufloatfromsamples(
          (2.163, 2.190, 2.188),
          desc="molar conductivity K2CO3"
        ) / kalium_c
          kalium_lambda # S cm2 / mol
```

Out[24]: 216.86431690056378 ± 1.8966262009802657

Na2CO3

```
In [25]: natrium_M = ufloat(105.99, desc="molar mass sodium carbonate") # g / mol
          natrium_m = ufloat(1.3895, 0.0001, desc="mass sodium carbonate")
          natrium_c = natrium_m / natrium_M / 0.1 / 10
          natrium_c
```

Out[25]: 0.01310972733276724 ± 9.434852344560808e-07

```
In [26]: natrium_lambda = ufloatfromsamples(
          (1.715, 1.720, 1.720),
          desc="molar conductivity Na2CO3"
        ) / natrium_c
          natrium_lambda # S cm2 / mol
```

Out[26]: 131.073155811443 ± 0.2792487421399839

Exercises from the book

$$\epsilon_i = \frac{E}{6 \pi \eta} \frac{z_i}{r_i}$$

Ex 1

```
In [28]: from math import pi
```

```
In [33]: e = 1.602176e-19 # C
          E = 100 # V / m
          eta = 1e-3 # Pa s
          r = 1e-9 # m

          z_mg = 2
          z_essig = 1
```

```
In [34]: e * E / (6 * pi * eta) * z_mg / r # m/s
```

Out[34]: 1.6999615340213366e-06

```
In [35]: e * E / (6 * pi * eta) * z_essig / r # m/s
```

Out[35]: 8.499807670106683e-07