

Assignment 3: solutions:

1. Problem 2, p. 226: Show that

$$\Theta_{20}(\theta) = \frac{\sqrt{10}}{4} (3 \cos^2 \theta - 1)$$

is a solution of the Eq.

$$\frac{1}{\sin \theta} \frac{d}{d\theta} \left(\sin \theta \frac{d\Theta}{d\theta} \right) + \left[l(l+1) - \frac{m_l^2}{\sin^2 \theta} \right] \Theta = 0$$

and that it is normalized.

Ans : To show that Θ_{20} is a solution, we need not worry about the constant factor. Let us consider $f(\theta) = 3 \cos^2 \theta - 1$. In the L.H.S of the differential equation

1st term :

$$\frac{df}{d\theta} = -6 \cos \theta \sin \theta$$

$$\sin \theta \frac{df}{d\theta} = -6 \cos \theta \sin^2 \theta$$

$$\frac{1}{\sin \theta} \frac{d}{d\theta} \sin \theta \frac{df}{d\theta} = -\frac{6}{\sin \theta} [-\sin^3 \theta + 2 \cos^2 \theta \sin \theta] = 6[\sin^2 \theta - 2 \cos^2 \theta] = -6f$$

2nd term:

$$\left[l(l+1) - \frac{m_l^2}{\sin^2 \theta} \right] f = 2(2+1)f = 6f$$

$$\therefore \text{L.H.S} = -6f + 6f = 0 = \text{R.H.S}$$

$\Rightarrow \Theta_{20}$ is a solution .

$$\text{And } \int |\Theta_{20}|^2 \sin \theta d\theta = \frac{10}{16} \int_0^\pi (3 \cos^2 \theta - 1)^2 \sin \theta d\theta$$

$$= \frac{5}{8} \int_{-1}^{+1} (3X^2 - 1)^2 dX = \frac{5}{8} \int_{-1}^{+1} (9X^4 - 6X^2 + 1) dX = \frac{5}{8} \left[\frac{9}{5} X^5 \Big|_{-1}^{+1} - 2X^3 \Big|_{-1}^{+1} + X \Big|_{-1}^{+1} \right]$$

$$\Rightarrow \frac{5}{8} \left[\frac{18}{5} - 4 + 2 \right] = \frac{5}{8} \cdot \frac{8}{5} = 1$$

Hence Θ_{20} is normalised in θ - space.

2. Problem 3, p. 226: Show that

$$R_{10}(r) = \frac{2}{a_0^2} e^{-\frac{r}{a_0}}$$

$$\frac{1}{r^2} \frac{d}{dr} \left(r^2 \frac{dR}{dr} \right) + \left[\frac{2m}{\hbar^2} \left(\frac{e^2}{4\pi\epsilon_0 r} + E \right) - \frac{l(l+1)}{r^2} \right] R = 0$$

and that it is normalized.

Ans. Consider

$$f(r) = \exp\left(\frac{-r}{a_0}\right)$$

1st term

$$\frac{1}{r^2} \frac{d}{dr} r^2 \frac{df}{dr} = -\frac{a_0}{r^2} \frac{d}{dr} [r^2 \exp \frac{-r}{a_0}] = -\frac{1}{a_0 r^2} [2r - \frac{1}{a_0} r^2] \exp \frac{-r}{a_0} = -\frac{1}{a_0} \left[-\frac{1}{a_0} + \frac{2}{r} \right] e^{\frac{-r}{a_0}}$$

2nd term: using $a_0 = \frac{\hbar^2}{m} \frac{4\pi\epsilon_0}{e^2}$; $E = \frac{e^2}{8\pi\epsilon_0 a_0}$; and $\frac{2m}{\hbar^2} E = \frac{1}{a_0^2}$

$$\left[\frac{2m}{\hbar^2} \left(\frac{e^2}{4\pi\epsilon_0 r} + E \right) - \frac{l(l+1)}{r^2} \right] e^{\frac{-r}{a_0}} \xrightarrow{l=0} \frac{1}{a_0} \left[-\frac{1}{a_0} + \frac{2}{r} \right] e^{-\frac{r}{a_0}}$$

$$\therefore \text{L.H.S} = -\frac{1}{a_0} \left[-\frac{1}{a_0} + \frac{2}{r} \right] e^{-\frac{r}{a_0}} + \frac{1}{a_0} \left[-\frac{1}{a_0} + \frac{2}{r} \right] e^{-\frac{r}{a_0}} = 0 = \text{R.H.S}$$

To prove the normalisation, we show using $\int x^2 e^{ax} dx = \left(\frac{x^2}{a} - \frac{2x}{a^2} + \frac{2}{a^3} \right) e^{ax}$,

$$\int_0^\infty \left(\frac{2}{a_0^{\frac{3}{2}}} e^{-\frac{r}{a_0}} \right)^2 r^2 dr = \frac{4}{a_0^3} \left[\left(-\frac{r^2}{2} - \frac{2r}{\left(\frac{2}{a_0}\right)^2} - \frac{2}{\left(\frac{2}{a_0}\right)^3} \right) e^{-\frac{2r}{a_0}} \right]_0^\infty = 1$$

3. Problem 21, p. 227: The probability of finding an atomic electron whose radial wave function is $R(r)$ outside a sphere of radius r_0 centered on the nucleus is

$$\int_{r_0}^\infty |R(r)|^2 r^2 dr$$

(a) Calculate the probability of finding a 1s electron in a hydrogen atom at a distance greater than a_0 from the nucleus.

(b) When a 1s electron in a hydrogen atom is $2a_0$ from the nucleus, all its energy is potential energy. According to classical physics, the electron therefore cannot ever exceed the distance $2a_0$ from the nucleus. Find the probability $r > 2a_0$ for a 1s electron in a hydrogen atom.

Ans.

$$p_{r>a_0} = \int_{a_0}^\infty |R(r)|^2 r^2 dr = \int_{a_0}^\infty \left(\frac{2}{a_0^{\frac{3}{2}}} \right)^2 \left(e^{-\frac{r}{a_0}} \right)^2 r^2 dr = \frac{4}{a_0^3} \int_{a_0}^\infty e^{-\frac{2r}{a_0}} r^2 dr$$

$$\text{using } \int x^2 e^{ax} dx = \left(\frac{x^2}{a} - \frac{2x}{a^2} + \frac{2}{a^3} \right) e^{ax}$$

we get

$$\begin{aligned} p_{r>a_0} &= \frac{4}{a_0^3} \left[\left(-\frac{a_0}{2} \cdot r^2 - 2 \left(\frac{a_0}{2} \right)^2 \cdot r - \left(\frac{a_0}{2} \right)^3 \cdot 2 \right) e^{-\frac{2r}{a_0}} \right]_{a_0}^\infty \\ &= \frac{4}{a_0^3} \left[\frac{a_0^3}{2} + \frac{a_0^3}{2} + \frac{a_0^3}{4} \right] e^{-2} \\ &= 5 \cdot e^{-2} = 0.677 \end{aligned}$$

(b)

$$\begin{aligned} p_{r>2a_0} &= \frac{4}{a_0^3} \left[2^2 \cdot \frac{a_0^3}{2} + 2 \cdot \frac{a_0^3}{2} + \frac{a_0^3}{4} \right] e^{-2 \times 2} \\ &= 13e^{-4} = 0.238 \end{aligned}$$

4. Problem 22, p. 227: According to Fig. 6.11 (p. 214), a 2s electron in a hydrogen atom is more

likely than a 2p electron to be closer to the nucleus than $r = a_0$ (that is, to be between $r = 0$ and $r = a_0$). Verify thLs by calculating the relevant probabilities,

Ans. Using the results from above ,

$$\begin{aligned} p_{r < a_0}^{1s} &= \frac{4}{a_0^3} \left[- \left(\frac{a_0}{2} \cdot r^2 + \frac{a_0^2}{2} r + \frac{a_0^3}{4} \right) e^{-\frac{2r}{a_0}} \right]_0^{a_0} \\ &= \frac{4}{a_0^3} \left[\frac{a_0^3}{4} - a_0^3 \left(\frac{1}{2} + \frac{1}{2} + \frac{1}{4} \right) e^{-2} \right] \\ &= 1 - 0.677 = 0.323 \end{aligned}$$

(we could have obtained this as $p_{r < a_0}^{1s} = 1 - p_{r > a_0}^{1s}$)

$$\begin{aligned} p_{r < a_0}^{2p} &= \int_0^{a_0} \left(\frac{1}{2\sqrt{6}a_0^{\frac{3}{2}}} \cdot \frac{r}{a_0} e^{-\frac{r}{2a_0}} \right)^2 r^2 dr \\ &= \frac{1}{24a_0^5} \int_0^{a_0} r^4 e^{-\frac{r}{a_0}} dr \end{aligned}$$

using $\int x^4 e^{ax} dx = \left[\frac{x^4}{a} - \frac{4}{a^2} x^3 + \frac{12}{a^3} x^2 - \frac{24}{a^4} x + \frac{24}{a^5} \right] e^{ax}$

$$\begin{aligned} p_{r < a_0}^{2p} &= \left(\frac{1}{24a_0^5} \left[-a_0 r^4 - 4a_0^2 r^3 - 12a_0^3 r^2 - 24a_0^4 r - 24a_0^5 \right] e^{-\frac{r}{a_0}} \right)_0^{a_0} \\ &= 1 - \frac{1}{24} [1 + 4 + 12 + 24 + 24] e^{-1} \\ &= 1 - \frac{65}{24} e^{-1} = 1 - 0.996 = 0.004 \end{aligned}$$

5.. Problem 23, p. 227: Unsöld's theorem states that for any value of the orbital quantum number l , the probability densities summed over all possible states from $m_l = -l$ to $m_l = +l$ yield a constant independent of angles θ or ϕ ; that is,

$$\sum_{m_l = -l}^{m_l = +l} |\Theta|^2 |\Phi|^2 = \text{constant}$$

This theorem means that every closed subshell atom or ion has a spherically symmetric distribution of electric charge. Verify Unsöld theorem for $l = 0$, $l = 1$, and $l = 2$ with the help of Table 6.1.

Ans:

$$S = \sum_{m_l = -l}^{+l} |\Theta|^2 |\Phi|^2$$

For $l = 0$

$$S = |\Theta_{00}|^2 |\Phi_{00}|^2 = \frac{1}{4\pi}$$

For $l = 1$

$$\begin{aligned} S &= |\Theta_{1,0}|^2 |\Phi_0|^2 + 2|\Theta_{1,\pm 1}|^2 |\Phi_{\pm 1}|^2 \\ &= \frac{6}{2} \cos^2 \theta \cdot \frac{1}{2\pi} + 2 \cdot \frac{3}{2} \sin^2 \theta \cdot \frac{1}{2\pi} \end{aligned}$$

$$= \frac{3}{2\pi}(\cos^2 \theta + \sin^2 \theta) = \frac{3}{2\pi}$$

For $l = 2$

$$\begin{aligned}
S &= |\Theta_{2,0}|^2 |\Phi_0|^2 + 2 \cdot |\Theta_{2,\pm 1}|^2 |\Phi_{\pm}|^2 + 2 |\Theta_{2,\pm 2}|^2 |\Phi_{\pm 2}|^2 \\
&= \frac{10}{16} (3 \cos^2 \theta - 1)^2 \cdot \frac{1}{2\pi} + 2 \cdot \frac{15}{4} \sin^2 \theta \cos^2 \theta \frac{1}{2\pi} + 2 \cdot \frac{15}{16} \sin^4 \theta \cdot \frac{1}{2\pi} \\
&= \frac{5}{16\pi} [(3 \cos^2 \theta - 1)^2 + 12 \sin^2 \theta \cos^2 \theta + 3 \sin^4 \theta] \\
&= \frac{5}{16\pi} [(3 - 3 \sin^2 \theta - 1)^2 + 12 \sin^2 \theta (1 - \sin^2 \theta) + 3 \sin^4 \theta] \\
&= \frac{5}{16\pi} [4 - 12 \sin^2 \theta + 9 \sin^4 \theta + 12 \sin^2 \theta - 12 \sin^4 \theta + 3 \sin^4 \theta] \\
&= \frac{5}{16\pi} [4] = \frac{5}{4\pi}
\end{aligned}$$